



Chemical Constituents from the Root Bark and Leaves of
Artocarpus elasticus

Prakit Chaithada

A Thesis Submitted in Partial Fulfillment of the Requirements
for the Degree of Master of Science in Organic Chemistry
Prince of Songkla University

2010

Copyright of Prince of Songkla University

Thesis Title Chemical Constituents from the Root Bark and Leaves of
Artocarpus elasticus
Author Mr. Prakit Chaithada
Major Program Organic Chemistry

Major Advisor:

.....
(Assoc. Prof. Dr. Wilawan Mahabusarakam)

Co-advisor:

.....
(Dr. Suda Chakthong)

Examining Committee:

.....Chairperson
(Asst. Prof. Dr. Orasa Pancharoen)

.....
(Assoc. Prof. Dr. Wilawan Mahabusarakam)

.....
(Dr. Suda Chakthong)

.....
(Assoc. Prof. Dr. Chatchanok Kalarai)

The Graduate School, Prince of Songkla University, has approved this thesis as partial fulfillment of the requirements for the Master of Science Degree in Organic Chemistry.

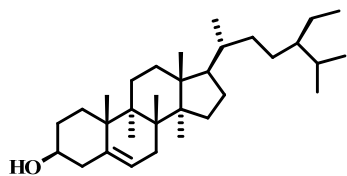
.....
(Prof. Dr. Amornrat Phongdara)
Dean of Graduate School

ชื่อวิทยานิพนธ์	องค์ประกอบทางเคมีจากเปลือกกรากและใบกะออก (<i>Artocarpus elasticus</i>)
ผู้เขียน	นายประกิต ไชยธาดา
สาขาวิชา	เคมีอินทรีย์
ปีการศึกษา	2553

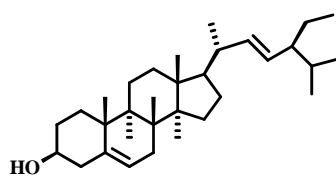
บทคัดย่อ

การศึกษากองศ์ประกอบทางเคมีของเปลือกกรากและใบกะออก (*Artocarpus elasticus*) แยกได้สารกลุ่ม prenylated dihydrochalcones ที่ยังไม่มีรายงานการวิจัย 5 สาร ได้แก่ 1-(2,4-dihydroxyphenyl)-3-(8-hydroxy-2,2-dimethyl-7-(3-methylbut-2-enyl)-2*H*-chromen-6-yl)propan-1-one (**PK13**), 1-(2,4-dihydroxyphenyl)-3-(3,4-dihydroxy-2,5-bis(3-methylbut-2-enyl)phenyl)propan-1-one (**PK14**), 1-(2,4-dihydroxyphenyl)-3-(7-((3,3-dimethyloxiran-2-yl)methyl)-8-hydroxy-2,2-dimethyl-2*H*-chromen-6-yl)propan-1-one (**PK15**), 1-(2,4-dihydroxyphenyl)-3-(4-hydroxy-2,2-dimethyl-5-(3-methylbut-2-enyl)-2,7*b*-dihydro-1*aH*-oxireno[2,3-*c*]chromen-6-yl)propan-1-one (**PK17**) และ 1-(2,4-dihydroxyphenyl)-3-(7-hydroxy-6-(3-methylbut-2-enyl)benzofuran-5-yl)propan-1-one (**PK18**) นอกจากนี้ยังได้สารที่มีรายงานวิจัยแล้ว 13 สาร ได้แก่ สารผสมของ β -sitosterol และ stigmasterol (**PK1**), (*E*)-4-(3',4'-dimethoxyphenyl)-3-butenyl acetate (**PK2**), 5*a*,6-dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-9-(3-methyl-2-buten-1-yl)-5*H*,7*H*,11*H*-benzofuro[3,4-*bc*]pyrano[3,2-*h*]xanthen-7-one (**PK3**), 4-hydroxybenzaldehyde (**PK4**), 2,3,8-trihydroxy-11,11-dimethyl-13-(3-methyl-2-butenyl)-6-(2-methyl-1-propenyl)-6*H*,7*H*,11*H*-bis[1]benzopyrano[4,3-*b*:6',7'-*e*]pyran-7-one (**PK5**), (*E*)-4-(3',4'-dimethoxyphenyl)but-3-en-1-ol (**PK6**), 2-(2,4-dihydroxyphenyl)-5-hydroxy-8,8-dimethyl-3-(3-methylbut-2-enyl)pyrano[3,2-*g*]chromen-4(8*H*)-one (**PK7**), 5*a*,6-dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-5*H*,7*H*,11*H*-benzofuro[3,4-*bc*]pyrano[3,2-*h*]xanthen-7-one (**PK8**), 6,7-dihydro-5,9,11,14-tetrahydroxy-3,3-dimethyl-6-(1-methylethenyl)-(-)-3*H*,8*H*-pyrano[3',2':4,5]benzo[1,2-*c*]xanthen-8-one (**PK9**), 8,9-dihydro-6,10,11,13-tetrahydroxy-3,3-dimethyl-9-(1-methylethenyl)-3*H*,7*H*-benzo[*c*]pyrano[3,2-*h*]xanthen-7-one (**PK10**), (*E*)-3-(4'-hydroxy-3'-methoxyphenyl)-2-propenoic acid (**PK11**), 5-hydroxy-8,8-dimethyl-3-(3-methyl-2-butenyl)-2-(2,4,5-trihydroxyphenyl)-4*H*,8*H*-benzo[1,2-*b*:3,4-*b'*]dipyran-4-one (**PK12**) และ (*S*)-2-(2,4-dihydroxyphenyl)-5-hydroxy-7-methoxychroman-

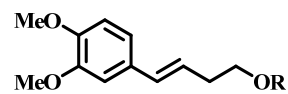
4-one (PK16) โครงสร้างของสารประกอบเหล่านี้วิเคราะห์โดยใช้ข้อมูลทางสเปกโทรสโกปี UV IR NMR MS และ เปรียบเทียบกับสารที่มีรายงานการวิจัยแล้ว



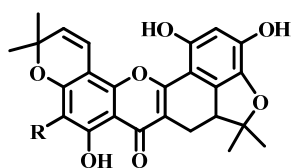
PK1



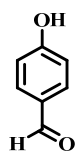
PK2 : R = CH₃CO



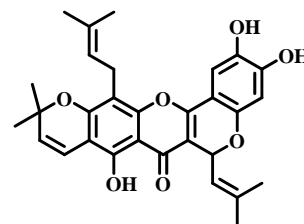
PK6 : R = H



PK3 : R = prenyl

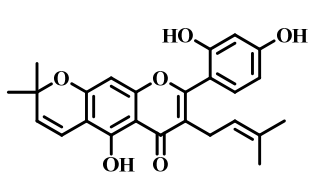


PK4

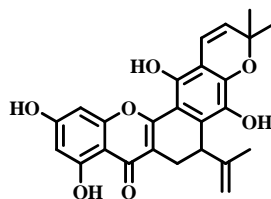


PK5

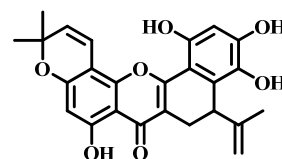
PK8 : R = H



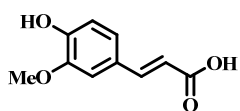
PK7



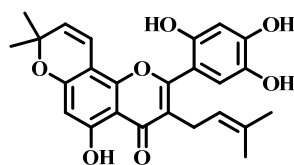
PK9



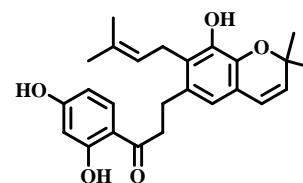
PK10



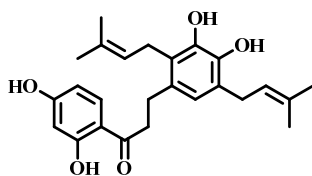
PK11



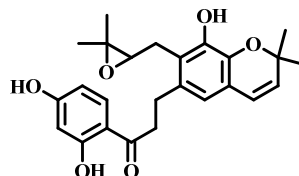
PK12



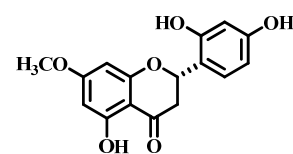
PK13



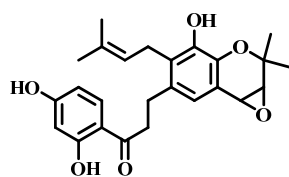
PK14



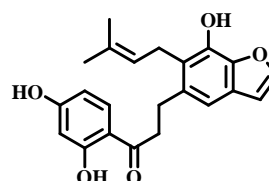
PK15



PK16



PK17



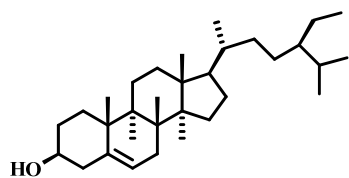
PK18

Thesis Title	Chemical Constituents from the Root Bark and Leaves of <i>Artocarpus elasticus</i>
Author	Mr. Prakrit Chaithada
Major Program	Organic Chemistry
Academic Year	2010

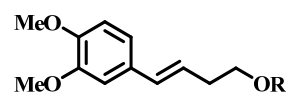
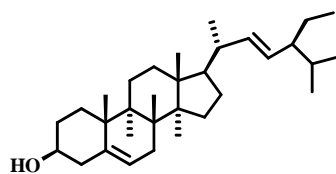
ABSTRACT

Investigation of the chemical constituents from the root bark and leaves of *Artocarpus elasticus* yielded five new prenylated dihydrochalcones: 1-(2,4-dihydroxyphenyl)-3-(8-hydroxy-2,2-dimethyl-7-(3-methylbut-2-enyl)-2*H*-chromen-6-yl)propan-1-one (**PK13**), 1-(2,4-dihydroxyphenyl)-3-(3,4-dihydroxy-2,5-bis(3-methylbut-2-enyl)phenyl)propan-1-one (**PK14**), 1-(2,4-dihydroxyphenyl)-3-(7-((3,3-dimethyloxiran-2-yl)methyl)-8-hydroxy-2,2-dimethyl-2*H*-chromen-6-yl)propan-1-one (**PK15**), 1-(2,4-dihydroxyphenyl)-3-(4-hydroxy-2,2-dimethyl-5-(3-methylbut-2-enyl)-2,7*b*-dihydro-1*aH*-oxireno[2,3-*c*]chromen-6-yl)propan-1-one (**PK17**) and 1-(2,4-dihydroxyphenyl)-3-(7-hydroxy-6-(3-methylbut-2-enyl)benzofuran-5-yl)propan-1-one (**PK18**). Thirteen known compounds were also obtained: a mixture of β -sitosterol and stigmasterol (**PK1**), (*E*)-4-(3',4'-dimethoxyphenyl)-3-butenyl acetate (**PK2**), 5*a*,6-dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-9-(3-methyl-2-buten-1-yl)-5*H*,7*H*,11*H*-benzofuro[3,4-*bc*]pyrano[3,2-*h*]xanthen-7-one (**PK3**), 4-hydroxybenzaldehyde (**PK4**), 2,3,8-trihydroxy-11,11-dimethyl-13-(3-methyl-2-butenyl)-6-(2-methyl-1-propenyl)-6*H*,7*H*,11*H*-bis[1]benzopyrano[4,3-*b*:6',7'-*e*]pyran-7-one (**PK5**), (*E*)-4-(3',4'-dimethoxyphenyl)but-3-en-1-ol (**PK6**), 2-(2,4-dihydroxyphenyl)-5-hydroxy-8,8-dimethyl-3-(3-methylbut-2-enyl)pyrano[3,2-*g*]chromen-4(8*H*)-one (**PK7**), 5*a*,6-dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-5*H*,7*H*,11*H*-benzofuro[3,4-*bc*]pyrano[3,2-*h*]xanthen-7-one (**PK8**), 6,7-dihydro-5,9,11,14-tetrahydroxy-3,3-dimethyl-6-(1-methylethenyl)-(-)-3*H*,8*H*-pyrano[3',2':4,5]benzo[1,2-*c*]xanthen-8-one (**PK9**), 8,9-dihydro-6,10,11,13-tetrahydroxy-3,3-dimethyl-9-(1-methylethenyl)-3*H*,7*H*-benzo[*c*]pyrano[3,2-*h*]xanthen-7-one (**PK10**), (*E*)-3-(4'-hydroxy-3'-methoxyphenyl)-2-propenoic acid (**PK11**), 5-hydroxy-8,8-dimethyl-3-(3-methyl-2-butenyl)-2-(2,4,5-trihydroxyphenyl)-4*H*,8*H*-benzo[1,2-*b*:3,4-*b'*]dipyran-4-one (**PK12**) and (*S*)-2-(2,4-dihydroxyphenyl)-5-hydroxy-7-methoxychroman-4-one (**PK16**). Their structures

were determined on the basis of UV, IR, NMR, MS and by comparison their spectroscopic data with those reported.

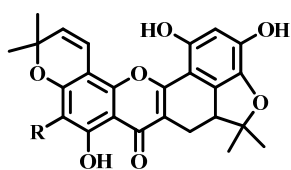


PK1



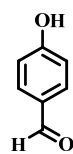
PK2 : R = CH₃CO

PK6 : R = H

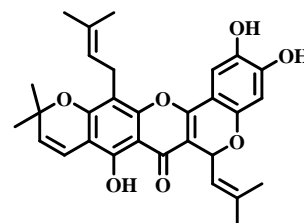


PK3 : R = prenyl

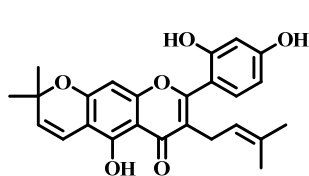
PK8 : R = H



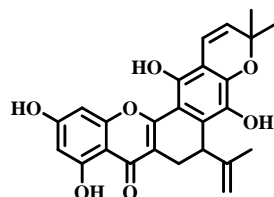
PK4



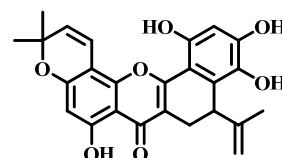
PK5



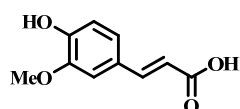
PK7



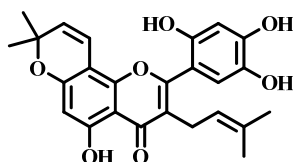
PK9



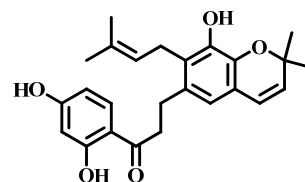
PK10



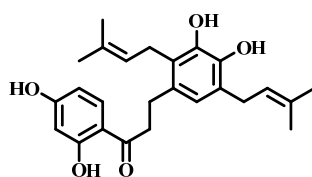
PK11



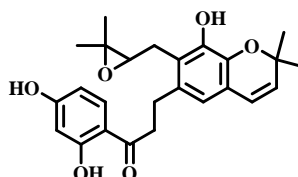
PK12



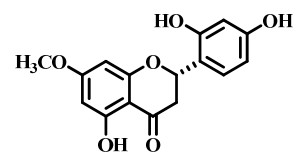
PK13



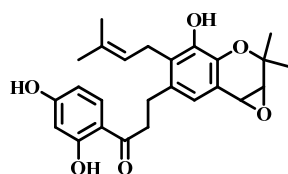
PK14



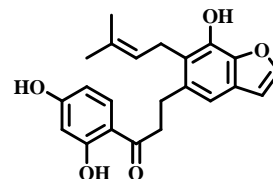
PK15



PK16



PK17



PK18

ACKNOWLEDGEMENTS

I wish to express my deepest and sincere gratitude to my advisor, Associate Professor Dr. Wilawan Mahabusarakam, for her valuable instruction, expert guidance and excellent suggestion. I would also like to express my appreciation to her for correction of my thesis. My sincere thanks are expressed to Dr. Suda Chakthong my co-advisor, for her kindness and valuable advice.

My sincere thanks are expressed to Associate Professor Dr. Saowalak Phongpaijit and Associate Professor Dr. Nongyao Sawangjaroen for bioactivity testing and Mr. Charernsak Saewai for plant identification. In addition, I would like to sincerely thank Mr. Kreangsak Wicheandang for plant material.

I would like to extend my appreciation to the staffs of the Department of Chemistry, Faculty of Science, Prince of Songkla University for making the thesis possible.

The research was made possible by a scholarship from Center for Innovation in Chemistry (PERCH-CIC), Commission on Higher Education, the Graduate School, Prince of Songkla University and Natural Products Research Center, Prince of Songkla University.

Finally, none of this would have been possible without love and encouragement of my family and colleagues. I thank them for their understanding during all of the times when I could not be with them and their steady love that supports me.

Prakit Chaithada

THE RELEVANCE OF THE RESEARCH WORK TO THAILAND

The purpose of this research is to investigate the chemical constituents from the root bark and leaves of *Artocarpus elasticus*. It is a part of the basic research on the utilization of Thai medicinal plants. This research will contribute significantly to scientific basis of traditional medicine. Seven prenylated flavones, five prenylated dihydrochalcones, two phenylbutenoids, a phenylpropanoids, a benzaldehyde derivatives, a flavanones and a mixture of triterpenoids were isolated from this plant. Some of the compounds showed strong antibacterial activity. Moreover, some compounds of these have been reported to show cytotoxicity, anti-inflammatory and antioxidation activities. So further study on the biological activity of the isolated compounds should be performed which can lead to active compounds. Therefore Thai plant can be utilized as a natural resource of potential drugs.

CONTENTS

	Page
Abstract (in Thai)	iii
Abstract (in English)	vi
Acknowledgements	ix
The relevance of the research work to Thailand	x
Contents	xi
List of Tables	xiii
List of Illustrations	xv
List of Abbreviations and Symbols	xvii
Chapter	
1. Introduction	1
1.1 Introduction	1
1.2 Review of Literatures	2
1.2.1 The Chemical Constituents of <i>Artocarpus</i> genus	2
1.2.2 Biogenetic relationship of some flavonoid compounds	19
1.2.3 The Biological Activity of <i>Artocarpus</i> genus	26
1.2.4 <i>Artocarpus elasticus</i>	26
1.3 Objective	27
2. Experimental	28
2.1 Instruments and Chemicals	28
2.2 Plant Material	28
2.3 Extraction and Isolation	29
A. Root bark	
2.3.1 Purification of dichloromethane extract	29
B. Leaves	
2.3.2 Purification of dichloromethane extract	33
3. Results and Discussion	42
3.1 Structure Elucidation	41
PK1	44
PK2	45

CONTENTS (continued)

	Page
PK3	47
PK4	50
PK5	51
PK6	54
PK7	56
PK8	58
PK9	61
PK10	64
PK11	67
PK12	69
PK13	72
PK14	75
PK15	77
PK16	79
PK17	81
PK18	83
3.2 Relationships of flavonoids in this study	85
Conclusion	87
References	91
Appendix	95
¹ H-NMR and ¹³ C-NMR spectrum of compounds PK1-PK18	96
¹ H-NMR and ¹³ C-NMR spectral data of known compounds from literatures	114
Vitae	124

LIST OF TABLES

Table		Page
1	Compounds isolated from the plants of <i>Artocarpus</i> genus	2
2	Physical characteristic and weights of fractions obtained from QCC of the dichloromethane extract (root bark)	30
3	Physical characteristic and weights of fractions obtained from QCC of the dichloromethane extract (leaves)	34
4	¹ H, ¹³ C and HMBC spectral data of PK2	47
5	¹ H- ¹ H COSY spectral data of PK2	47
6	¹ H, ¹³ C and HMBC spectral data of PK3	49
7	¹ H- ¹ H COSY spectral data of PK3	50
8	¹ H, ¹³ C and HMBC spectral data of PK4	51
9	¹ H, ¹³ C and HMBC spectral data of PK5	53
10	¹ H, ¹³ C and HMBC spectral data of PK6	56
11	¹ H- ¹ H COSY spectral data of PK6	56
12	¹ H, ¹³ C and HMBC spectral data of PK7	58
13	¹ H, ¹³ C and HMBC spectral data of PK8	60
14	¹ H- ¹ H COSY spectral data of PK8	61
15	¹ H, ¹³ C and HMBC spectral data of PK9	63
16	¹ H, ¹³ C and HMBC spectral data of PK10	66
17	¹ H, ¹³ C and HMBC spectral data of PK11	69
18	¹ H, ¹³ C and HMBC spectral data of PK12	71
19	¹ H, ¹³ C and HMBC spectral data of PK13	74
20	¹ H- ¹ H COSY spectral data of PK13	75
21	¹ H, ¹³ C and HMBC spectral data of PK14	77
22	¹ H, ¹³ C and HMBC spectral data of PK15	79
23	¹ H, ¹³ C and HMBC spectral data of PK16	81
24	¹ H, ¹³ C and HMBC spectral data of PK17	83
25	¹ H, ¹³ C and HMBC spectral data of PK18	85

LIST OF TABLES (continued)

Table		Page
A-1	¹ H and ¹³ C NMR spectral data of (<i>E</i>)-4-(3',4'-dimethoxyphenyl)-3-butenyl acetate	114
A-2	¹ H and ¹³ C NMR spectral data of 4-hydroxybenzaldehyde	114
A-3	¹ H and ¹³ C NMR spectral data of cycloheterophyllin	115
A-4	¹ H and ¹³ C NMR spectral data of (<i>E</i>)-4-(3',4'-dimethoxyphenyl)but-3-en-1-ol	116
A-5	¹ H and ¹³ C NMR spectral data of cudraflavone B	117
A-6	¹ H and ¹³ C NMR spectral data of cycloartobiloxanthone	118
A-7	¹ H and ¹³ C NMR spectral data of artelastoxanthone	119
A-8	¹³ C NMR spectral data of <i>trans</i> -feluric acid	120
A-9	¹ H and ¹³ C NMR spectral data of artobiloxanthone	121
A-10	¹ H and ¹³ C NMR spectral data of artonin E	122
A-11	¹ H and ¹³ C NMR spectral data of artocarpanone	123

LIST OF ILLUSTRATIONS

Scheme		Page
1	Biogenetic relationship of the flavonoids compound	20
2	Prenylation of a flavone at C-3 position	21
3	Radical cyclization of 3-prenylflavone into an oxepinoflavone	22
4	Cyclization of 3-isoprenylflavone in the formation of pyranoflavone	22
5	Oxidative coupling reaction in the formation of dihydrobenzoxanthone skeleton	23
6	Proposed biogenetic route of the cyclopentenoxanthone and xanthonolide types of compounds	24
7	Biogenetic route to dihydroxanthone derivative	25
8	Extraction of the crude extracts from the root bark of <i>A. elasticus</i>	29
9	Isolation of compounds PK1-PK12 from dichloromethane extract of the root bark of <i>A. elasticus</i>	30
10	Extraction of the crude extracts from the leaves of <i>A. elasticus</i>	33
11	Isolation of compounds PK13-PK18 from dichloromethane extract of the leaves of <i>A. elasticus</i>	35
12	Relationship of prenylated flavones from <i>A. elasticus</i>	86
13	Relationship of prenylated dihydrochalcones from <i>A. elasticus</i>	87
Figure		
1	<i>Artocarpus elasticus</i>	2
A-1	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK1	96
A-2	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK2	97
A-3	¹³ C-NMR (75 MHz) (CDCl ₃) spectrum of PK2	97
A-4	¹ H-NMR (300 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK3	98
A-5	¹³ C-NMR (75 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK3	98
A-6	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK4	99
A-7	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK5	100
A-8	¹³ C-NMR (75 MHz) (CDCl ₃) spectrum of PK5	100

LIST OF ILLUSTRATIONS (continued)

Figure		Page
A-9	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK6	101
A-10	¹³ C-NMR (75 MHz) (CDCl ₃) spectrum of PK6	101
A-11	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK7	102
A-12	¹³ C-NMR (75 MHz) (CDCl ₃) spectrum of PK7	102
A-13	¹ H-NMR (300 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK8	103
A-14	¹³ C-NMR (75 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK8	103
A-15	¹ H-NMR (300 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK9	104
A-16	¹³ C-NMR (75 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK9	104
A-17	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK10	105
A-18	¹³ C-NMR (75 MHz) (CDCl ₃) spectrum of PK10	105
A-19	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK11	106
A-20	¹³ C-NMR (75 MHz) (CDCl ₃) spectrum of PK11	106
A-21	¹ H-NMR (300 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK12	107
A-22	¹³ C-NMR (75 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK12	107
A-23	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK13	108
A-24	¹³ C-NMR (75 MHz) (CDCl ₃) spectrum of PK13	108
A-25	¹ H-NMR (300 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK14	109
A-26	¹³ C-NMR (75 MHz) (CDCl ₃ +DMSO- <i>d</i> ₆) spectrum of PK14	109
A-27	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK15	110
A-28	¹³ C-NMR (75 MHz) (CDCl ₃) spectrum of PK15	110
A-29	¹ H-NMR (300 MHz) (acetone- <i>d</i> ₆) spectrum of PK16	111
A-30	¹³ C-NMR (75 MHz) (acetone- <i>d</i> ₆) spectrum of PK16	111
A-31	¹ H-NMR (300 MHz) (acetone- <i>d</i> ₆) spectrum of PK17	112
A-32	¹³ C-NMR (75 MHz) (acetone- <i>d</i> ₆) spectrum of PK17	112
A-33	¹ H-NMR (300 MHz) (CDCl ₃) spectrum of PK18	113
A-34	¹³ C-NMR (75 MHz) (CDCl ₃) spectrum of PK18	113

LIST OF ABBREVIATIONS AND SYMBOLS

<i>s</i>	=	singlet
<i>d</i>	=	doublet
<i>t</i>	=	triplet
<i>m</i>	=	multiplet
<i>dd</i>	=	doublet of doublet
<i>dt</i>	=	doublet of triplet
<i>mt</i>	=	multiplet of triplet
<i>qd</i>	=	quartet of doublet
<i>td</i>	=	triplet of doublet
<i>ddd</i>	=	doublet of doublet of doublet
<i>br</i>	=	broad
<i>br s</i>	=	broad singlet
<i>br d</i>	=	broad doublet
<i>g</i>	=	gram
<i>kg</i>	=	kilogram
<i>mg</i>	=	milligram
%	=	percent
<i>nm</i>	=	nanometer
<i>m.p.</i>	=	melting point
cm^{-1}	=	reciprocal centimeter (wave number)
δ	=	chemical shift relative to TMS
<i>J</i>	=	coupling constant
λ_{max}	=	maximum wavelength
ν	=	absorption frequencies
ϵ	=	molar extinction coefficient
$^{\circ}\text{C}$	=	degree of celcius
<i>MHz</i>	=	Megahertz
<i>ppm</i>	=	part per million
<i>IR</i>	=	Infrared

LIST OF ABBREVIATIONS AND SYMBOLS (continued)

UV	=	Ultraviolet-Visible
NMR	=	Nuclear Magnetic Resonance
2D NMR	=	Two Dimensional Nuclear Magnetic Resonance
COSY	=	Correlated Spectroscopy
DEPT	=	Distortionless Enhancement by Polarization Transfer
HMBC	=	Heteronuclear Multiple Bond Correlation
HMQC	=	Heteronuclear Multiple Quantum Coherence
CC	=	column chromatography
TMS	=	tetramethylsilane
Acetone- d_6	=	deuteroacetone
DMSO- d_6	=	deuterodimethylsulphoxide
$CDCl_3$	=	deuteriochloroform
MeOH	=	methanol
CH_2Cl_2	=	dichloromethane
TLC	=	thin layer chromatography
MIC	=	Minimum Inhibition Concentration

CHAPTER 1

INTRODUCTION

1.1 Introduction

A natural product is a chemical compound produced by a living organism found in nature. It usually has a pharmacological or biological activity for use in pharmaceutical drug discovery and drug design. All plants produce chemical compounds as part of their normal metabolic activities. Herbalism is a traditional medicinal or folk medicine practice based on the use of plants and plant extracts. Herbal plant is abundantly found in Thailand, so the research of the chemical constituents in Thai herbal plant is necessary.

The *Artocarpus* genus belongs to the mulberry family, Moraceae. It is a large evergreen tree consisting of 60 species approximately. Most species of *Artocarpus* are widespread in Southeast Asia; a few cultivated species are more widely distributed, especially *A. altilis* and *A. heterophyllus*. Economically the genus is of appreciable importance as a source of edible fruits, such as *A. heterophyllus* (Jack-fruit), *A. champeden* (Chempedak), and *A. altilis* (Breadfruit). Recently, there have been increasing reports of prenylated flavonoids. In spite of the structural diversity of them, they have been isolated from a rather limited number of plant families especially the Leguminosae, Moraceae and Asteraceae. They were most frequently found in roots and bark, but also occur in the aerial parts, buds and seeds. Prenylflavonoids isolated from *A. communis* and *A. elasticus* revealed significant cytotoxic effect against human cancer cell lines (Cidade *et al.*, 2001). The root bark and the heartwood have been described as containing chemical compounds with antioxidant properties. Wei and co-workers (Wei *et al.*, 2005) indicated that artocarpanone from the roots of *A. heterophyllus* significantly inhibits the LPS-induced NO production and iNOS protein expression in RAW 264.7 cells which the large amount of NO produced in response to lipopolysaccharide (LPS) plays an important role in inflammatory conditions (Stoclet *et al.*, 1998). The antifungal and antimicrobial effect of flavonoids is mainly attributed to presence of phenolic

compounds which have high affinity for proteins and act as inhibitors of microbial enzymes. Many of the isoprenylated flavonoids also showed potent cytotoxic activity against various cell lines, including murine leukemia P388, KB, mouse L-1210 and colon 38, inhibition of arachidonate 5-lipoxygenase, antiplatelet activity, and antibacterial activity against cariogenic bacteria (Nomura *et al.*, 1998). There are several activities in *Artocarpus* genus. In Thailand, fourteen species of *Artocarpus*: *A. altilis* สาเก, *A. alitssimus* ไสน, *A. chaplasha* หาดสี้าน, *A. dadah* หาดรุ่ม, *A. elasticus* กะออก, *A. gomezianus* หาดหนูน, *A. heterophyllus* ขนุน, *A. integer* จำปาตะ, *A. kemando* ขนุนป่า, *A. lacucha* มะหาด, *A. lanceifolius* ขนุนป่า, *A. nitidus* มะหาดข่อย, *A. rigidus* spp. *asperulus* ขนุนป่าน, *A. rigidus* spp. *rigidus* ขนุนป่า are widely distributed (Smitinand, T. 2544).

1.2 Review of Literatures

1.2.1 The Chemical Constituents of *Artocarpus* genus

The chemical constituents which were isolated from *Artocarpus* genus before 2008 were summarized in the thesis of Aeesoh Yanya (2009). The additional constituents of this genus from 2008 to 2010 were summarized in **Table 1** (Based on SciFinder Scholar database). Several of compounds have been reported in the *Artocarpus* genus, such as 2-arylbenzofuran, flavonoids, stibenoids, triterpenes *etc.*

Table 1 Compounds isolated from the plants of *Artocarpus* genus

Compounds	Structure	Bibliography
1. <i>A. altilis</i>		
fruits		
artocarpesin	1	Amarasinghe
artoindonesianin F	2	<i>et al.</i> , 2008
3 β -acetoxyolean-12-en-11-one	3	
cycloartenyl acetate	4	
isoartocarpesin	5	

Table 1 (continued)

Compounds	Structure	Bibliography
(3-methyl-2-butenyl)-(<i>E</i>)-2,3',4,5'-stilbenetetrol	6	
moracin M	7	
norartocarpanone	8	
norartocarpetin	9	
oxyresveratrol	10	
sitosterol	11	
sitosterol β -D-glucopyranoside	12	
2. <i>A. communis</i>		
leaves		
5'-geranyl-2',4',4'-trihydroxychalcone	13	Fang <i>et al.</i> ,
isolespeol	14	2008
lespeol	15	
3,4,2',4'-tetrahydroxy-3'-geranyldihydro- chalcone	16	
xanthoangelol	17	
cortex of roots		
artochamin B	18	Lin <i>et al.</i> ,
artochamin D	19	2009
artocommunol CC	20	
artoflavone A	21	
artomunoisoxanthone	22	
cyclogeracommunin	23	
dihydroartomunoxanthone	24	
3. <i>A. elasticus</i>		
non specified		
artelastoheterol	25	Lin <i>et al.</i> ,
artanol A	26	2009
cycloartelastoxanthone	27	
cycloartobiloxanthone	28	

Table 1 (continued)

Compounds	Structure	Bibliography
wood		
artocarpin	29	Musthapa <i>et al.</i> , 2009
artoindonesianin E1	30	
cycloartocarpin	31	
cudraflavones A	32	
cudraflavones C	33	
4. A. heterophyllus		
twigs		
artocarpesin	1	Zheng <i>et al.</i> , 2009
artocarpin	29	
artoheterophyllin A	34	
artoheterophyllin B	35	
artoheterophyllin C	36	
artoheterophyllin D	37	
artoinin A	38	
artoinin J	39	
<i>p</i> -counmaric acid	40	
cudraflavones B	41	
cycloheterophyllin	42	
dihydrophaseic acid 4'- <i>O</i> - β -D-glucopyranoside	43	
2-(2,4-dihydroxy-6-methoxyphenyl)-5-hydroxy-7-methoxy-6-(3-methyl-1-buten-1-yl)-3-(3-methyl-2-buten-1-yl)-4 <i>H</i> -1-benzopyran-4-one	44	
(<i>E</i>)-5-(6-hydroxybenzofuran-2-yl)-4-(3-methylbut-1-enyl)benzene-1,3-diol	45	
6-prenyl- 4',5,7-trihydroxyflavone	46	
4-hydroxybenzoic acid	47	
licoflavone C	48	

Table 1 (continued)

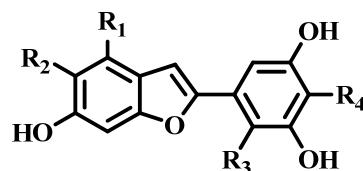
Compounds	Structure	Bibliography
moracin M	7	
norartocarpetin	9	
vanillic acid	49	
pulps		
<i>cis</i> -antheraxanthin	50	de Faria
<i>all-trans-α</i> -carotene	51	<i>et al.</i> , 2009
<i>9-cis-β</i> -carotene	52	
<i>13-cis-β</i> -carotene	53	
<i>15-cis-β</i> -carotene	54	
<i>all-trans-β</i> -carotene	55	
<i>all-trans-α</i> -cryptoxanthin	56	
<i>all-trans-β</i> -cryptoxanthin	57	
<i>all-trans</i> -lutein	58	
<i>cis</i> -luteoxanthin	59	
<i>all-trans</i> -luteoxanthin	60	
<i>all-trans</i> -neochrome	61	
<i>9-cis</i> -neoxanthin	62	
<i>all-trans</i> -neoxanthin	63	
<i>9-cis</i> -violaxanthin	64	
<i>all-trans</i> -zeaxanthin	65	
<i>cis</i> -zeinoxanthin	66	
<i>all-trans</i> -zeinoxanthin	67	
fruits		
artocarpesin	1	Fang <i>et al.</i> ,
norartocarpetin	9	2008
oxyresveratrol	10	

Table 1 (continued)

Compounds	Structure	Bibliography
5. <i>A. lowii</i>		
leaves		
2',4-dihydroxy-3',4'-(2,2-dimethylchromene) chalcone	68	Jamil <i>et al.</i> , 2008
2',4'-dihydroxy-4-methoxy-3'-prenyldihydro chalcone	69	
2',4',4-trihydroxy-3'-prenylchalcone	70	
6. <i>A. nobilis</i>		
root bark		
artobiloxanthone	71	Jayasinghe
artoinin E	72	<i>et al.</i> , 2008
artoinin E 2'-methylether	73	
artoinin V 2'-methylether	74	
cycloartobiloxanthone	28	
dihydroisoartoinin E 2'-methylether	75	
isoartoinin E 2'-methylether	76	
7. <i>A. tonkinensis</i>		
leaves		
alphitonin-4-O- β -D-glucopyranoside	77	Dang <i>et al.</i> ,
artokin-4'-O- β -D-glucopyranoside	78	2009
kaempherol-3-O- β -D-glucopyranoside	79	
maesopsin-4-O- β -D-glucopyranoside	80	

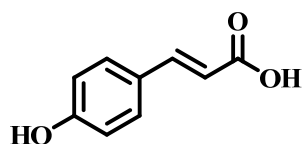
Structures of compounds from *Artocarpus* genus

a. 2-arylbenzofurans

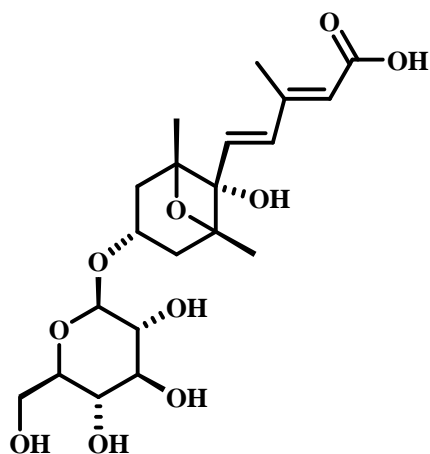


R ₁	R ₂	R ₃	R ₄	
H	H	H	H	7 : moracin M
	OMe		H	34 : artoheterophyllin A
H	H	H		45 : (<i>E</i>)-5-(6-hydroxybenzofuran-2-yl)-4-(3-methylbut-1-enyl)benzene-1,3-diol

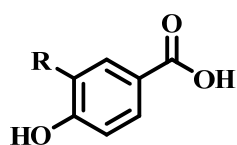
b. carboxylic acid



40 : *p*-coumaric acid



43 : dihydrophaseic acid 4'-*O*- β -D-glucopyranoside

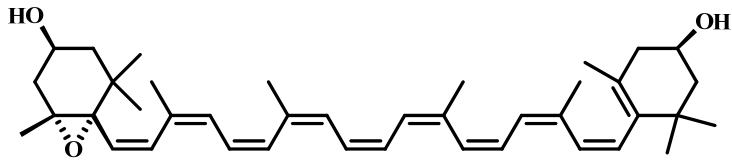
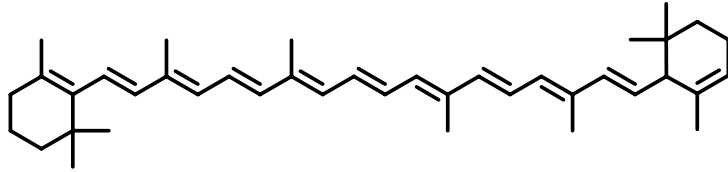
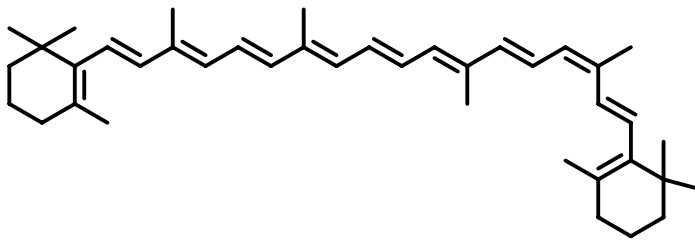
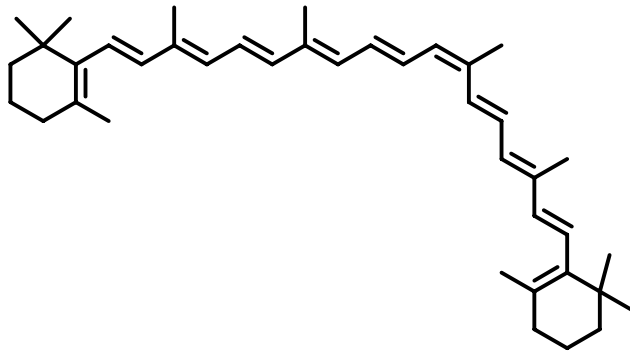
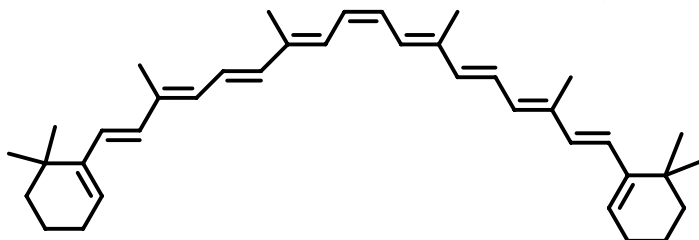
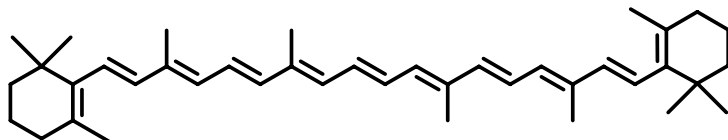


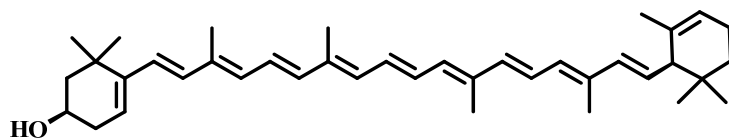
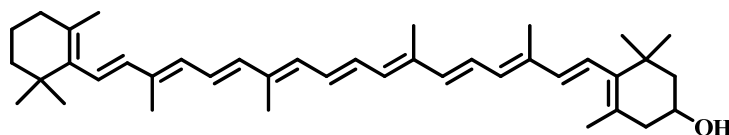
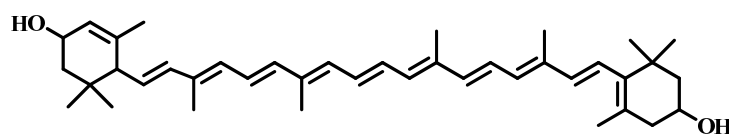
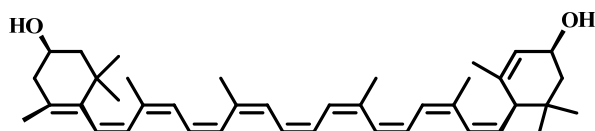
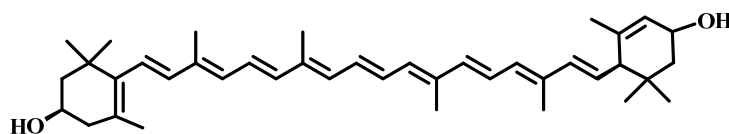
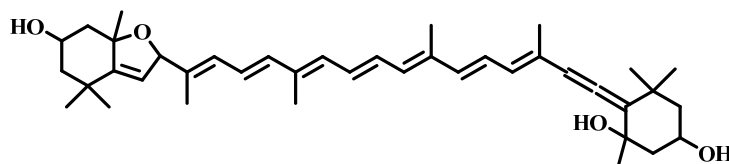
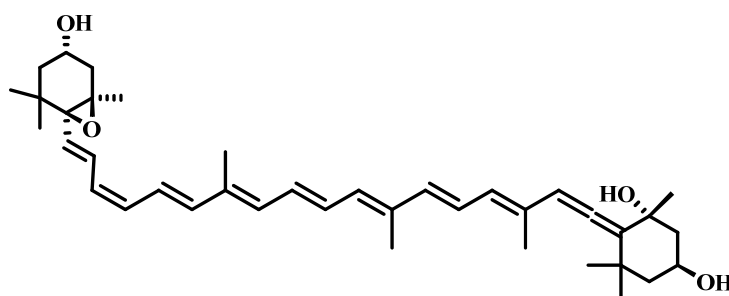
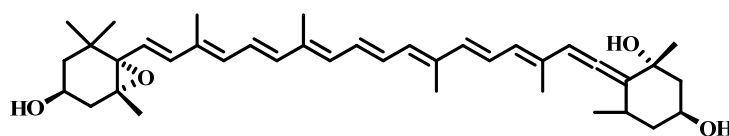
R

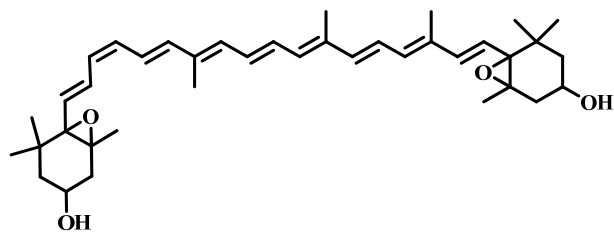
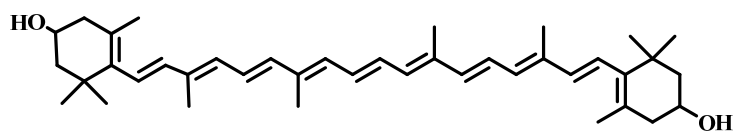
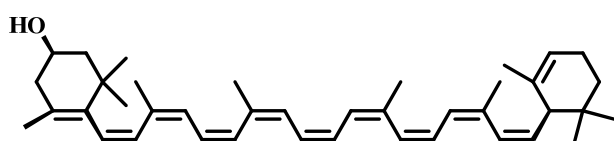
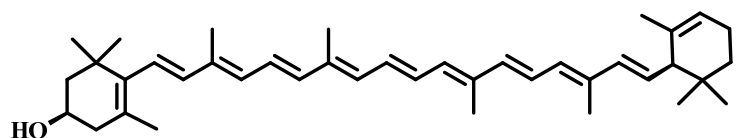
47 : H 4-hydroxybenzoic acid

49 : OMe vanillic acid

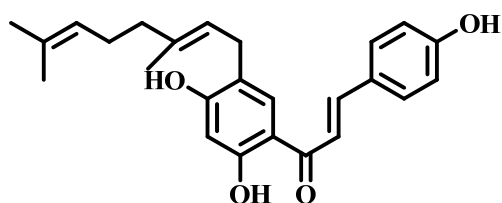
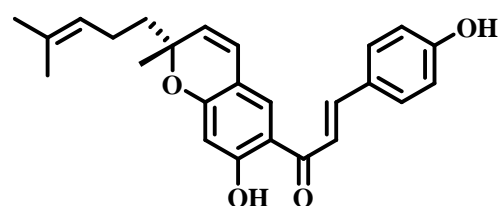
c. carotenoid

50 : *cis*-antheraxanthin51 : all-*trans*- α -carotene52 : 9-*cis*- β -carotene53 : 13-*cis*- β -carotene54 : 15-*cis*- β -carotene55 : all-*trans*- β -carotene

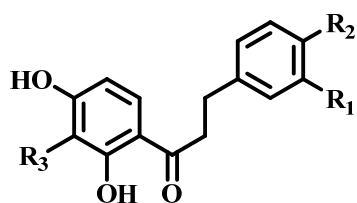
56 : all-*trans*- α -cryptoxanthin57 : all-*trans*- β -cryptoxanthin58 : all-*trans*-lutein59 : *cis*-luteoxanthin60 : all-*trans*-luteoxanthin61 : all-*trans*-neochrome62 : 9-*cis*-neoxanthin63 : all-*trans*-neoxanthin

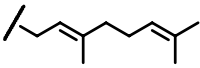
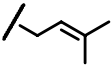
64 : 9-*cis*-violaxanthin65 : all-*trans*-zeaxanthin66 : *cis*-zeinoxanthin67 : all-*trans*-zeinoxanthin

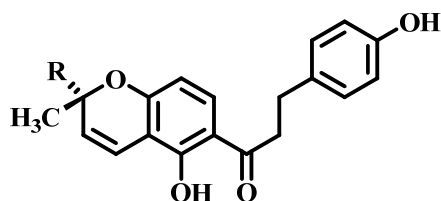
d. chalcones

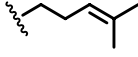
13 : 5'-geranyl-2',4',4'-
trihydroxychalcone

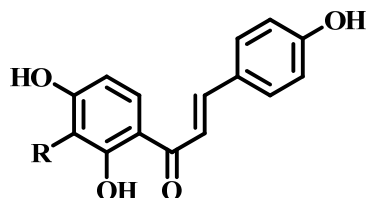
14 : isolespeol

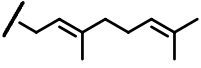
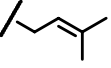


R_1	R_2	R_3	
16 : OH	OH		3,4,2',4'-tetrahydroxy-3'-geranyldihydrochalcone
69 : H	OMe		2',4'-dihydroxy-4-methoxy-3'-prenyldihydrochalcone

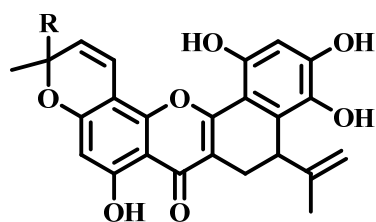


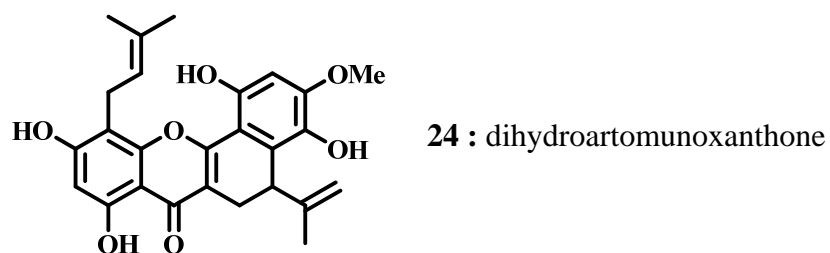
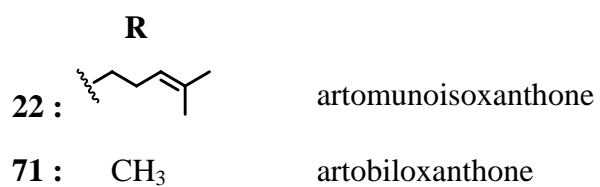
	R	
15 :		lespeol
68 :	CH ₃	2',4'-dihydroxy-3',4'-(2,2-dimethylchromene)chalcone



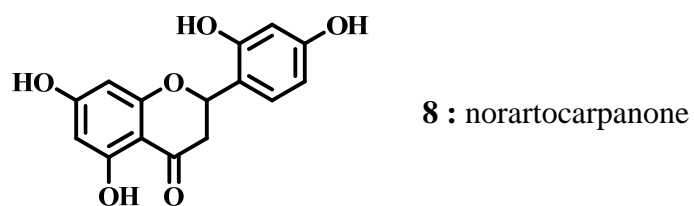
	R	
17 :		xanthoangelol
70 :		2',4',4'-trihydroxy-3'-prenylchalcone

e. dihydrobenzoxanthones

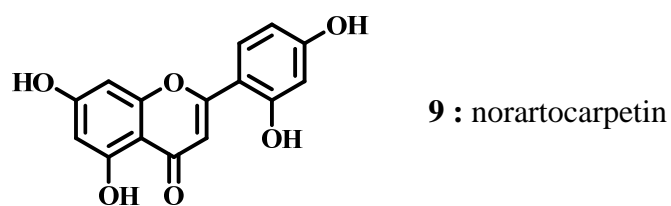




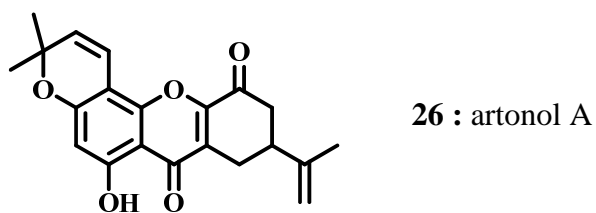
f. flavanones



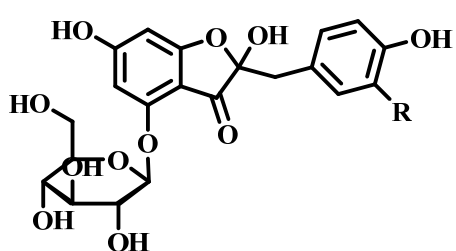
g. flavones



h. flavonoid-derived xanthenes



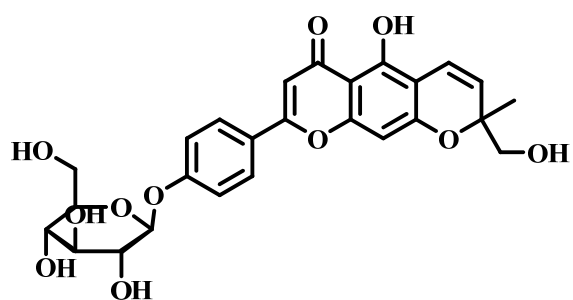
i. flavonoid glycosides



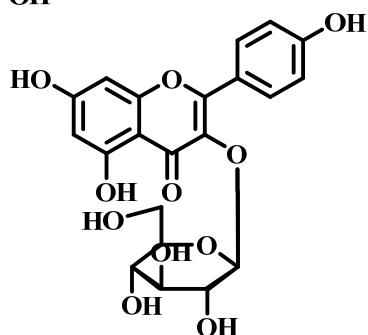
R

77 : OH alphonin-4-O- β -D-glucopyranoside

80 : H maesopsin-4-O- β -D-glucopyranoside

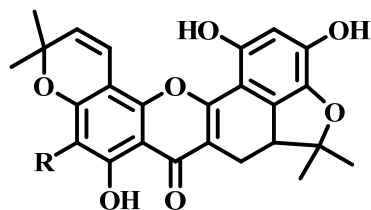


78 : artonin-4'-O- β -D-glucopyranoside

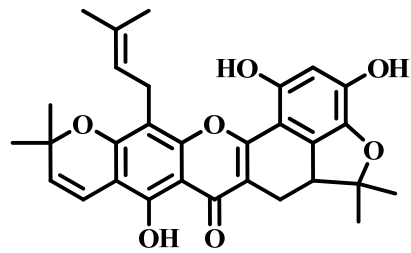
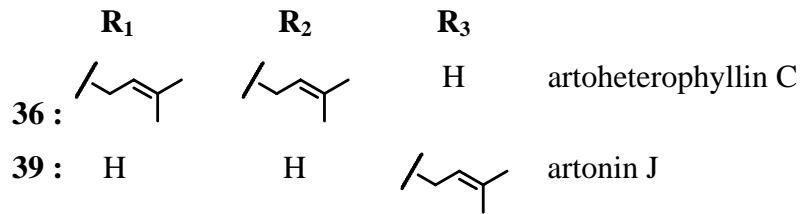
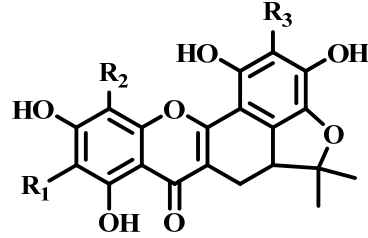
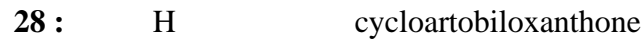
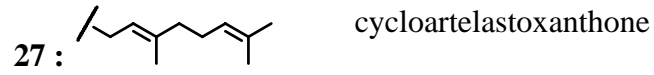


79 : kaempferol-3-O- β -D-glucopyranoside

j. furanodihydrobenzoxanthenes

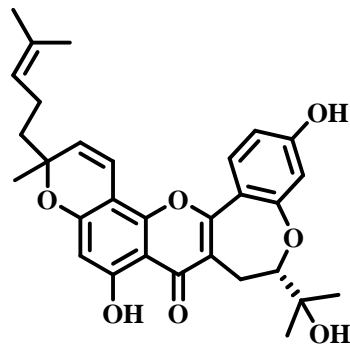


R

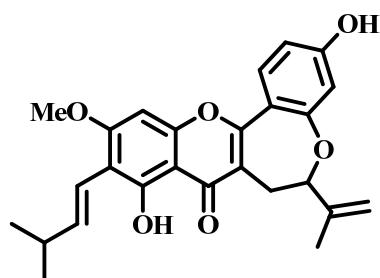


38 : artoinin A

k. oxepinoflavones

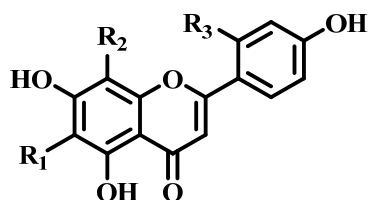


20 : artocommunol CC

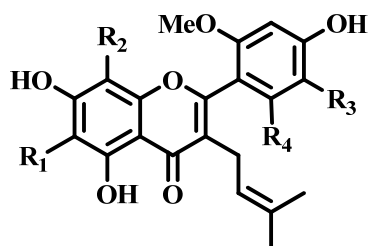


30 : artoindonesianin E1

1. prenylated flavones



	R ₁	R ₂	R ₃	
1 :		H	OH	artocarpesin
5 :		H	OH	isoartocarpesin
46 :		H	H	6-prenyl-4',5,7-trihydroxyflavone
48 : H		H	H	licoflavone C

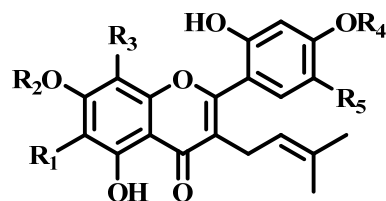


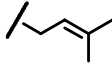
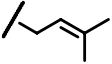
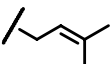
	R ₁	R ₂	R ₃	R ₄	
42 :		H	H	OH	cycloheterophyllin
44 :	OMe	H	H	OH	2-(2,4-dihydroxy-6-methoxyphenyl)-5-hydroxy-7-methoxy-6-(3-methyl-1-buten-1-yl)-3-(3-

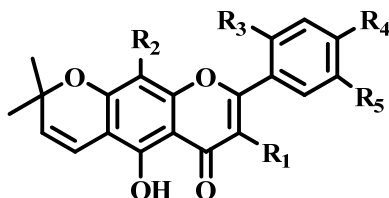
methyl-2-buten-1-yl)-4*H*-1-benzopyran-4-one

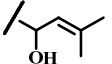
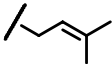
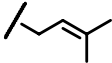
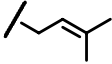
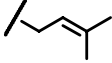
74 : H  OH H artonin V 2'-methylether

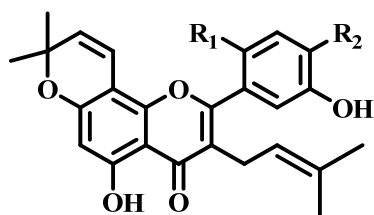
75 :  H OH H dihydroisoartonin E 2'-methylether



	R ₁	R ₂	R ₃	R ₄	R ₅	
19 :	H	H		Me	OH	artochamin D
29 :		Me	H	H	H	artocarpin
33 :		H	H	H	H	cudraflavones C

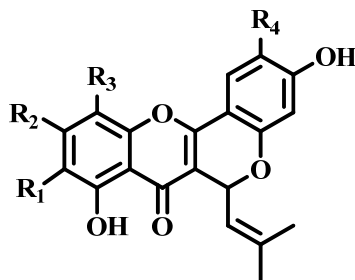


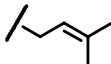
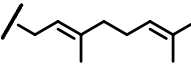
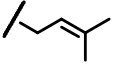
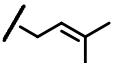
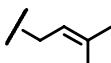
	R ₁	R ₂	R ₃	R ₄	R ₅	
25 :			OH	OH	OH	artelastoheterol
37 :			OH	OH	H	artoheterophyllin D
76 :		H	OMe	OH	OH	isoartonin E 2'-methylether

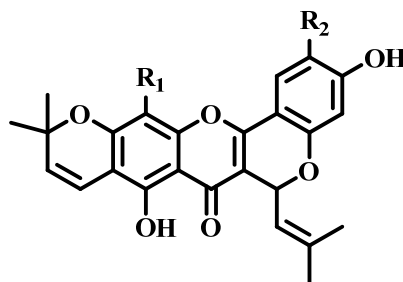


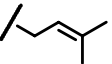
	R₁	R₂	
21 :	OH	OMe	artoflavone A
72 :	OH	OH	artonin E
73 :	OMe	OH	artonin E 2'-methylether

m. pyranoflavones

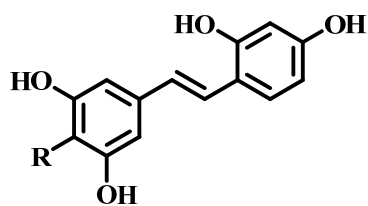


	R₁	R₂	R₃	R₄	
18 :	H	OH		OH	artocharin B
23 :	H	OH		H	cyclogeracommunin
31 :		OMe	H	H	cycloartocarpin
35 :		OH		OH	artoheterophyllin B

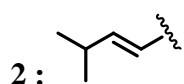


	R₁	R₂	
32 :	H	H	cudraflavones A
41 :		OH	cudraflavones B

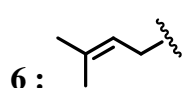
n. stilbenoids



R



artoindonesianin F



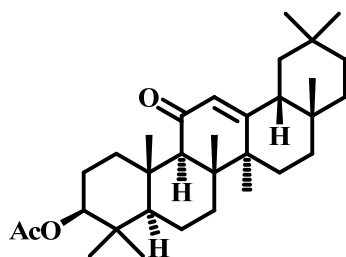
(3-methyl-2-butenyl)-(E)-

2,3,4,5'-stilbenetetrrol

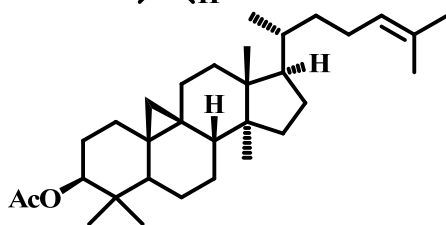


oxyresveratrol

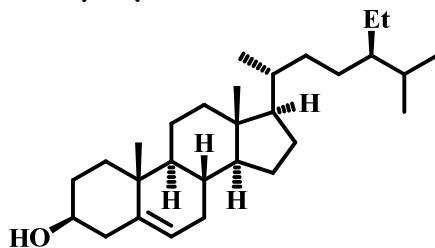
o. terpenoids



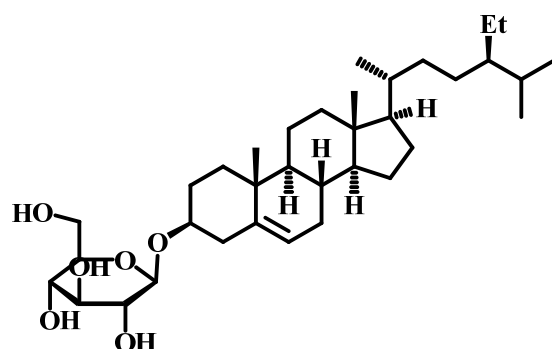
3 : 3 β -acetoxyolean-12-en-11-one



4 : cycloartenyl acetate



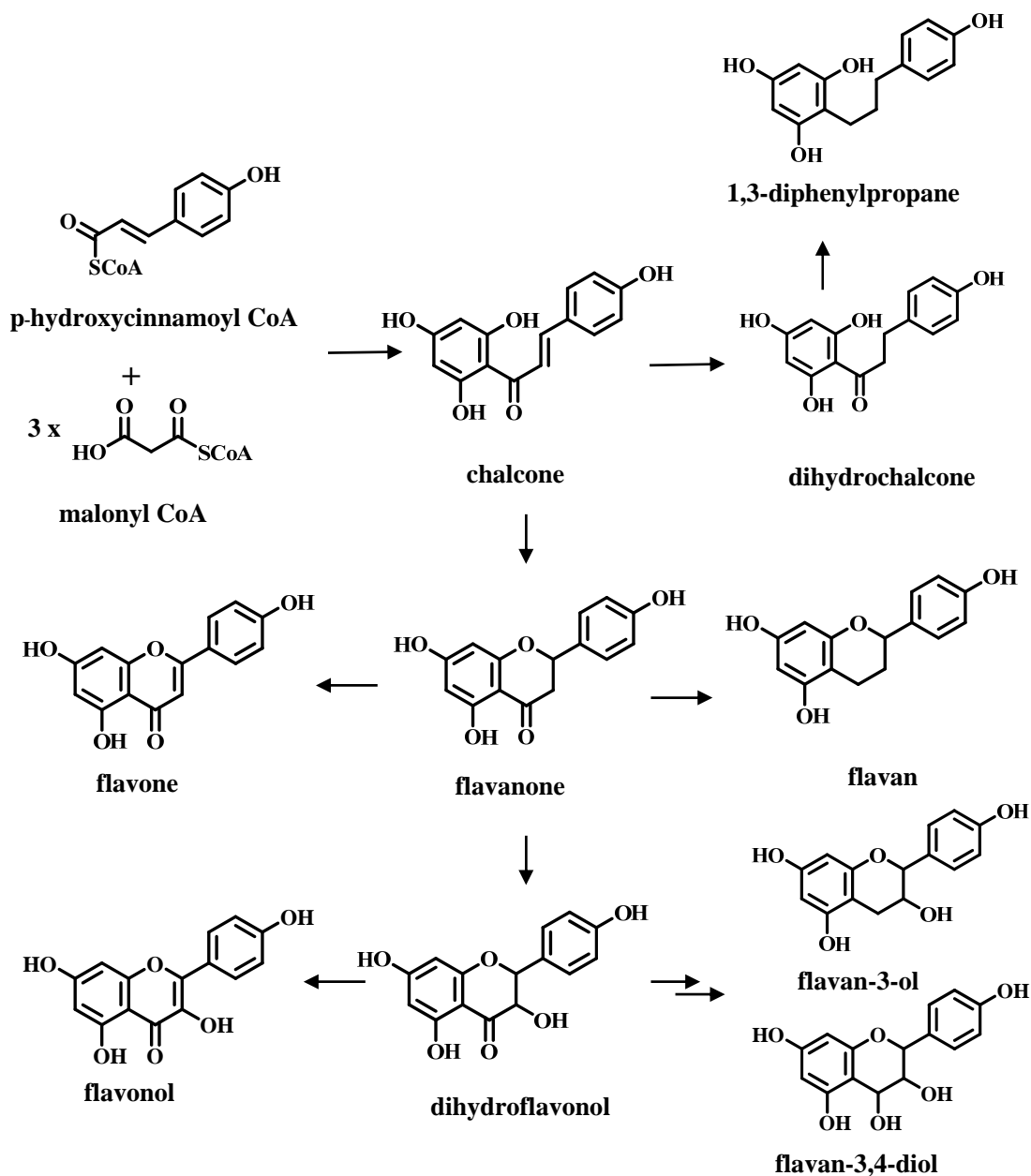
11 : sitosterol



12 : sitosterol- β -D-glucopyranoside

1.2.2 Biogenetic relationship of some flavonoid compounds

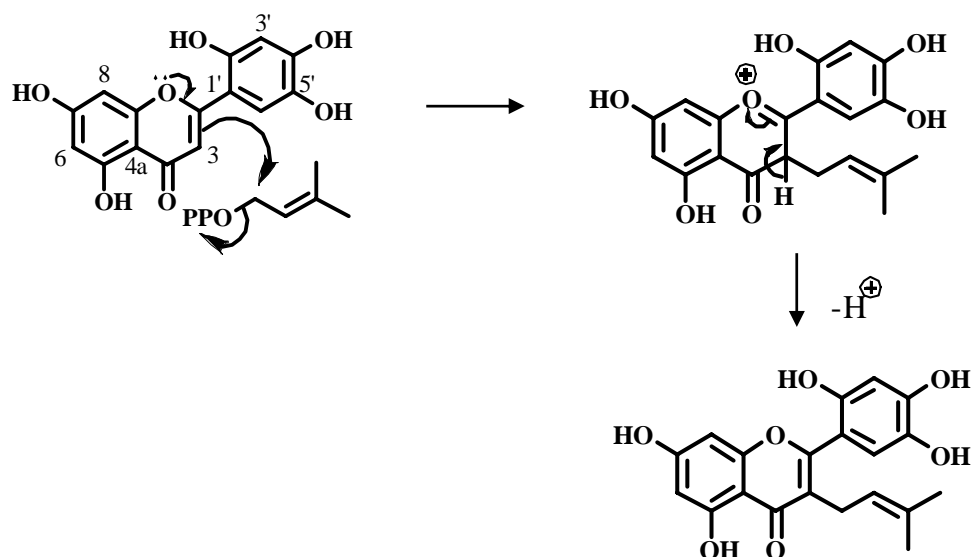
Many phenolic compounds, primarily flavonoids, apart from stilbenoids and 2-arylbenzofuran, have been isolated from *Artocarpus* species. The flavonoid constituents may be further classified according to their skeletons, namely chalcones, flavanones, flavones, flavan-3-ol, and 3-isoprenylflavones. Flavanones represent branch-point intermediates in the biosynthesis of other classes of flavonoids. Biogenetic relationship of the flavonoids compound was shown in **Scheme 1**.



Scheme 1 Biogenetic relationship of the flavonoids compound

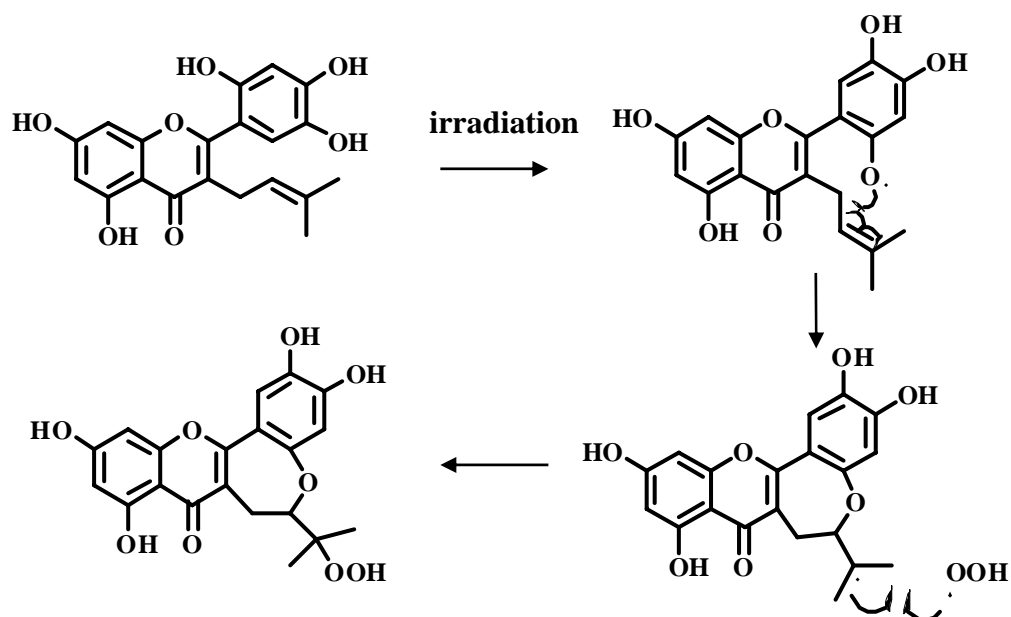
The major chemical compositions of *Artocarpus* plants were prenylated flavonoids. Generally, most flavonoids were *C*-prenylated, whereas *O*-prenylation is quite rare. *C*-prenylation takes place more frequently on ring A at position 6/8, as well as positions 3'/5' especially in flavanones and flavones (Barron and Ibrahim, 1996). Notable among the prenylated flavones is the frequent substitution at position 3. The formation of 3-prenylflavone derivatives from simple

flavones has been suggested to involve formally selective isoprenylation of the flavones as indicated in **Scheme 2** (Sultanbawa *et al.*, 1989).



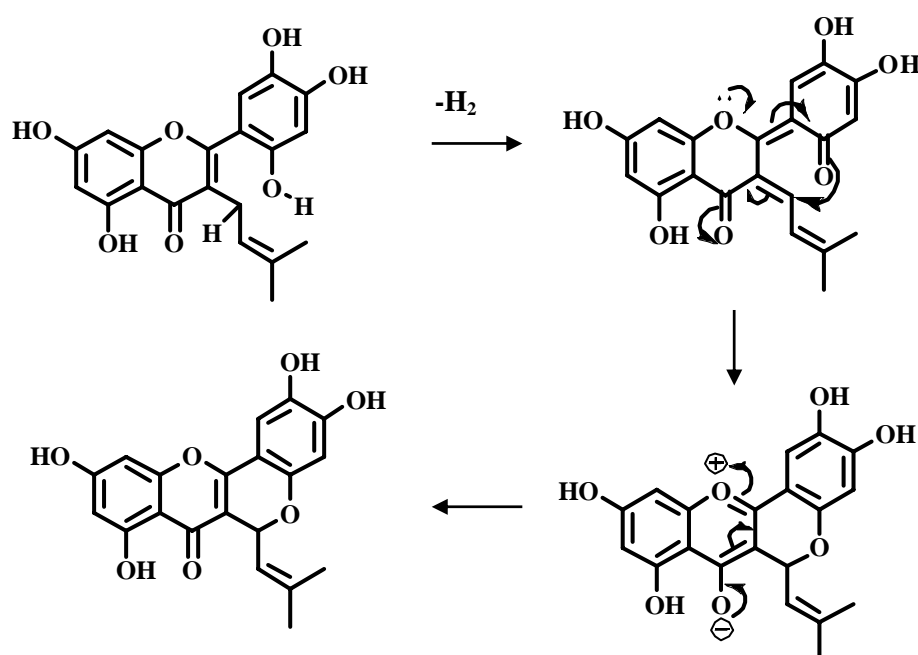
Scheme 2 Prenylation of a flavone at C-3 position

Modification of the isoprenyl side chains may occur by further oxidation, reduction, dehydration and/or cyclization. In common, cyclization of the isoprenyl side chains with adjacent phenolic groups gave the pyrano or furano derivatives. The isoprenyl substituent at C-3 position is always found in the form of a carbocyclic ring or an oxygen-bearing ring fused with rings B and C. Some hypotheses about the biogenesis of the *Artocarpus* flavonoids and related compounds have been reported in the literature (Hakim *et al.*, 2006). The 3-prenylflavones serve as precursors for several different structural types of flavonoids. The oxepinoflavone skeleton was provided by photo-oxidative cyclization derived from 3-isoprenyl-2',4',5'-trioxygenated flavones as presented by **Scheme 3** (Aida *et al.*, 1996).



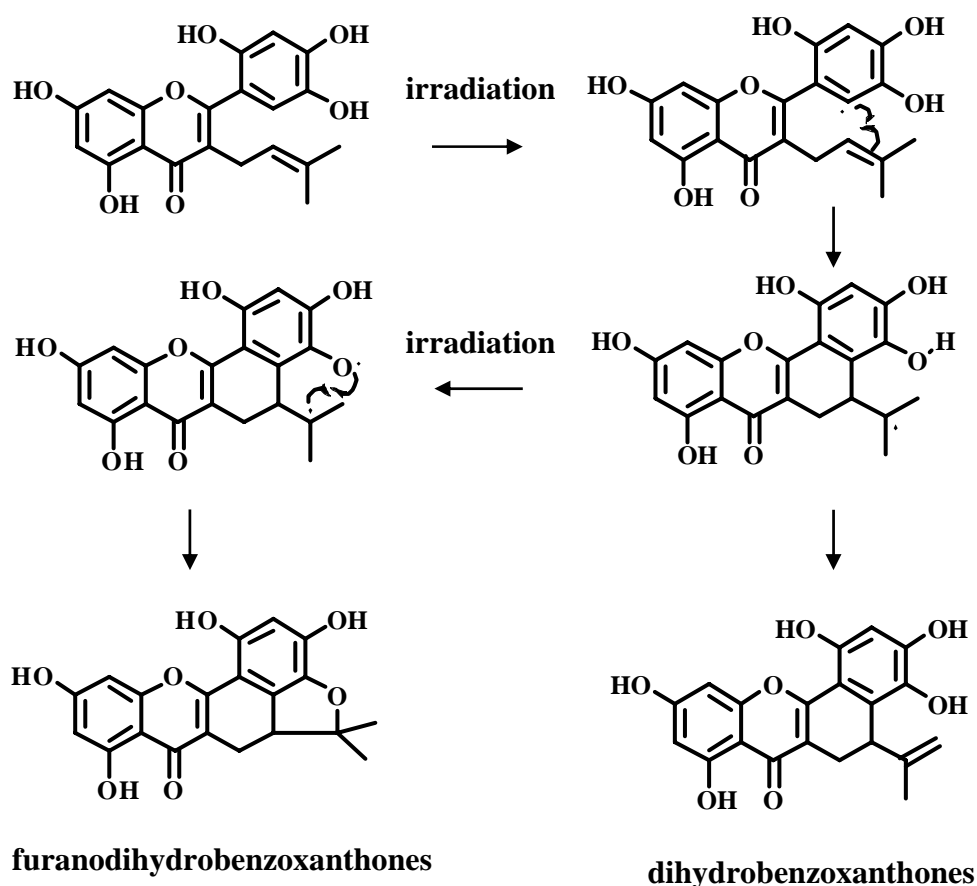
Scheme 3 Radical cyclization of 3-prenylflavone into an oxepinoflavone

In the same way, the pyranoflavone skeleton may be proposed that it derived from 3-isoprenyl-2',4',5'-trioxygenated flavones through cyclization as indicated in **Scheme 4** (Hakim *et al.*, 2006).



Scheme 4 Cyclization of 3-isoprenylflavone in the formation of pyranoflavone

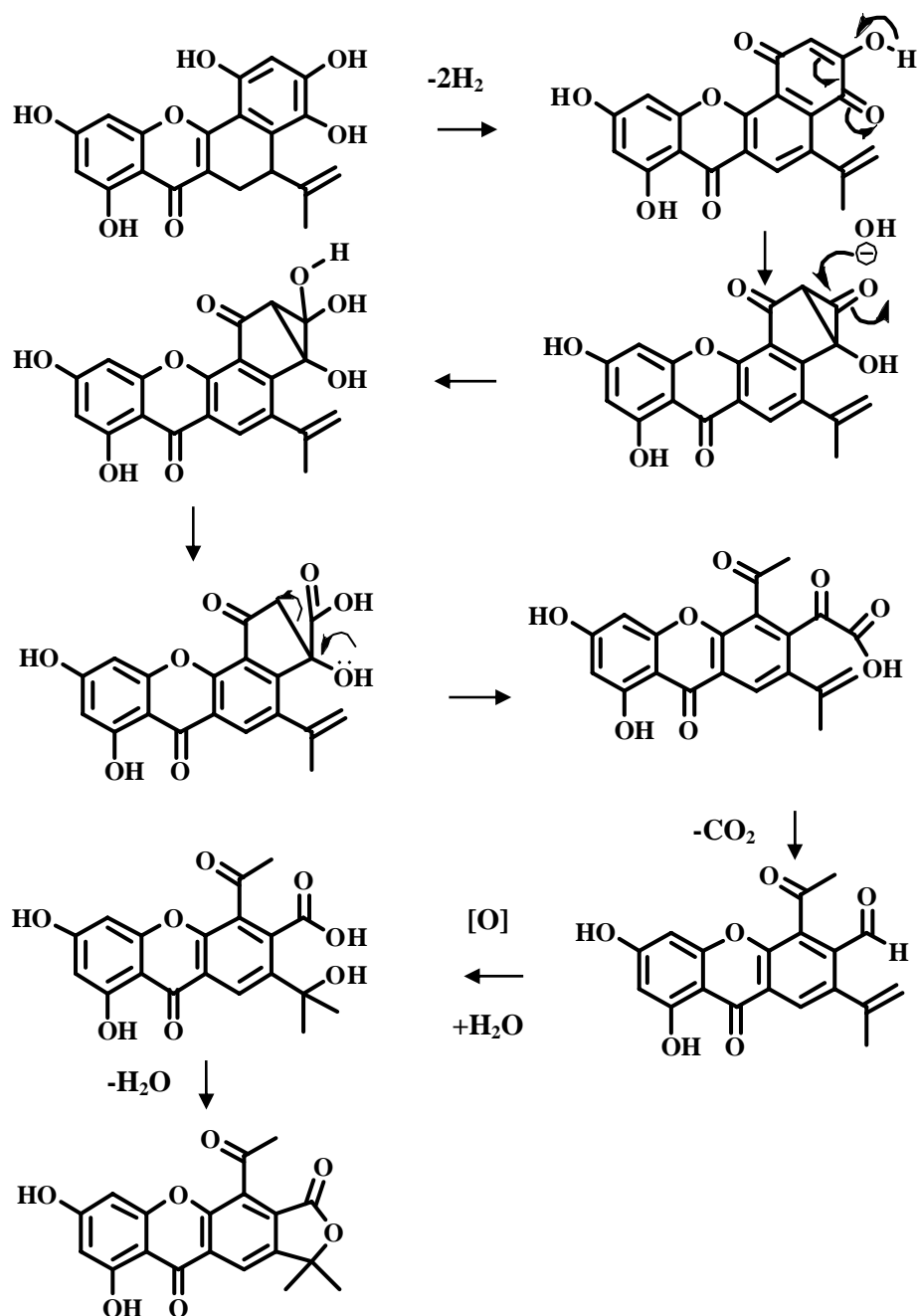
Similarly, it can assume that the dihydrobenzoxanthone skeleton is biologically derived from 3-isoprenyl-2',4',5'-trioxygenated flavones through oxidative cyclization as shown in **Scheme 5**. The hypothesis was confirmed by treatment of artonin E with the radical reagent diphenyl picryl hydrazyl (DPPH) to produce artobiloxanthone and cycloartobiloxanthone (Hano *et al.*, 1989; Aida *et al.*, 1996).



Scheme 5 Oxidative coupling reaction in the formation of dihydrobenzoxanthone skeleton

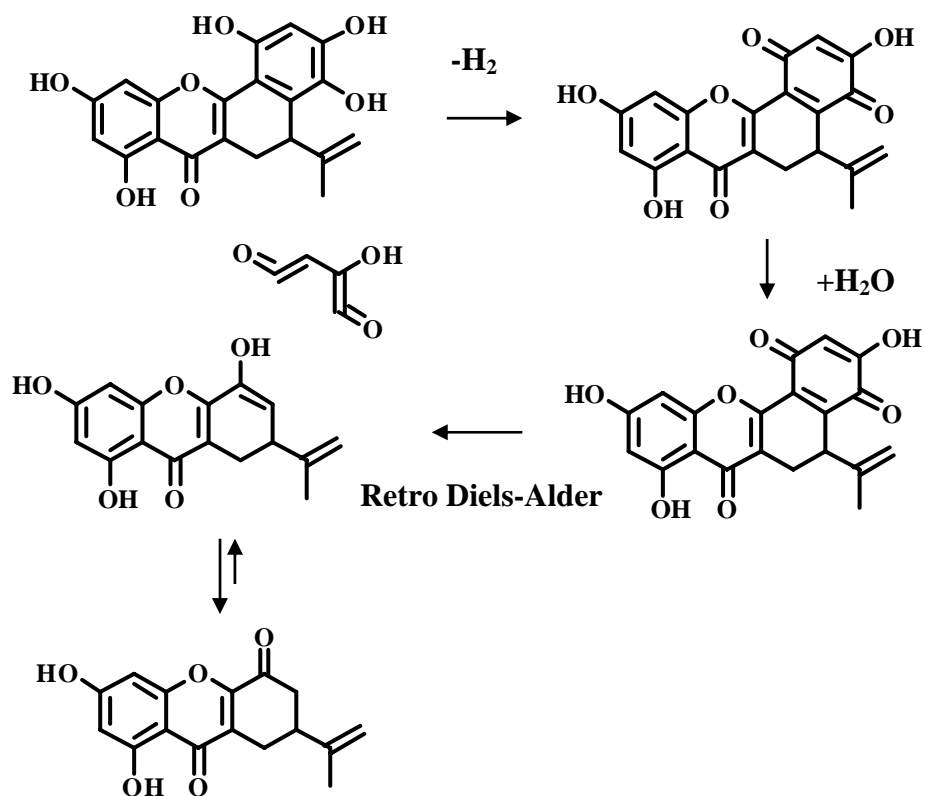
The dihydrobenzoxanthone skeleton may be further rearranged to other xanthone skeletons of quinonobenzoxanthone, cyclopentenoxanthone, xanthonolide, and dihydroxanthone. The quinonobenzoxanthones are biologically assumed to be derived from dihydrobenzoxanthone through oxidation reactions. Favorskii rearrangement of dihydrobenzoxanthone intermediates led to the cyclopente-

noxanthone skeleton, and may be further oxidation into xanthonolide derivatives, as described in **Scheme 6** (Hakim *et al.*, 2006).



Scheme 6 Proposed biogenetic route of the cyclopentenoxanthone and xanthonolide types of compounds

Moreover, the dihydroxanthone skeleton may be derived from a dihydrobenzoxanthone hydrate through a retro Diels-Alder reaction, as described in **Scheme 7**. The explanation of this hypothesis was confirmed by Aida and co-worker for the production of artonol A and B (Aida *et al.*, 1997).



Scheme 7 Biogenetic route to dihydroxanthone derivative

1.2.3 The Biological Activity of *Artocarpus* genus

Artocarpus plants have been used as traditional medicine in Indonesia against inflammation and malarial fever (Nomura *et al.*, 1998). Prenylflavonoids isolated from *Artocarpus elasticus* revealed significant cytotoxic effects against human cancer cell lines (Cidade *et al.*, 2001; Ko *et al.*, 2005). In the west part of Java, *A. elasticus* has been used to treat inflammation, female contraception (bark), dysentery (latex), and tuberculosis (young leaves) (Musthapa *et al.*, 2009). Many members of the *Artocarpus* genus have also been used as traditional folk medicine in Southeast Asia for the treatment of inflammation, malarial fever, and to treat ulcers, abscess, and diarrhea (Nomura *et al.*, 1998).

The biological activities of compounds from *A. elasticus* such as norartocarpetin, artocarpesin, isolespeol, artobiloxanthone and cycloartobiloxanthone have been reported to show tyrosinase inhibitory activity (Zheng *et al.*, 2009), anti-inflammatory activity (Dang *et al.*, 2009; Cerqueira *et al.*, 2008; Fang *et al.*, 2008), cytotoxic activity (Fang *et al.*, 2008; Musthapa *et al.*, 2009), anti-oxidant activity (Jamil *et al.*, 2008; Jayasinghe *et al.*, 2008; Lin *et al.*, 2009).

1.2.4 *Artocarpus elasticus*

Artocarpus elasticus (**Figure 1**) that is a perennial plant, is widely found in the southern of Thailand. It can grow as high as 40 meters. Its branches are spreading and its outer bark is smooth dark brown while inner bark is light brown. The leaves are large 12-30 c.m., wide 20-55 c.m. and bright green. Fruits are cylindrical-shaped; at first the rind is green and turning yellow-brown when ripe. It will be mature during July-October. *A. elasticus* is locally known as “Ka-ok”. It is not only found in the southern part of Thailand but also been found in Myanmar, Malaysia, and Indonesia. In the previously report, there are only one report on the chemical constituents from root bark and no report from leaves, so we are motivated to investigate its compositions in detail.



Figure 1 *Artocarpus elasticus*

1.3 Objectives

The objective of this work was to investigate the chemical constituents from the root bark and leaves of *A. elasticus*.

CHAPTER 2

EXPERIMENTAL

2.1 Instruments and Chemicals

Melting points were determined on a digital Electrothermal 9100 Melting Point Apparatus. The UV spectra were measured with a SPECORD S 100 (Analytikjena) and principle bands (λ_{\max}) were recorded as wavelengths (nm) and $\log \varepsilon$ in MeOH solution. The optical rotation $[\alpha]_D$ was measured in chloroform and methanol solution with Sodium D line (590 nm) on a JASCO P-1020 digital polarimeter. The IR spectra were measured with a Perkin-Elmer FTS FT-IR spectrophotometer. The NMR spectral data were recorded using 300 MHz Bruker FTNMR Ultra ShieldTM spectrometers in CDCl_3 , acetone- d_6 and DMSO- d_6 with TMS as the internal standard. Chemical shifts are reported in δ (ppm) and coupling constants (J) are expressed in hertz. EI and HREI mass spectra were measured on MAT 95 XL Mass spectrometer. Solvents for extraction and chromatography were distilled at their boiling point ranges prior to use except chloroform was analytical grade reagent. Quick column chromatography (QCC) and column chromatography (CC) were carried out on silica gel 60 H (Merck) and silica gel 100 (Merck), respectively.

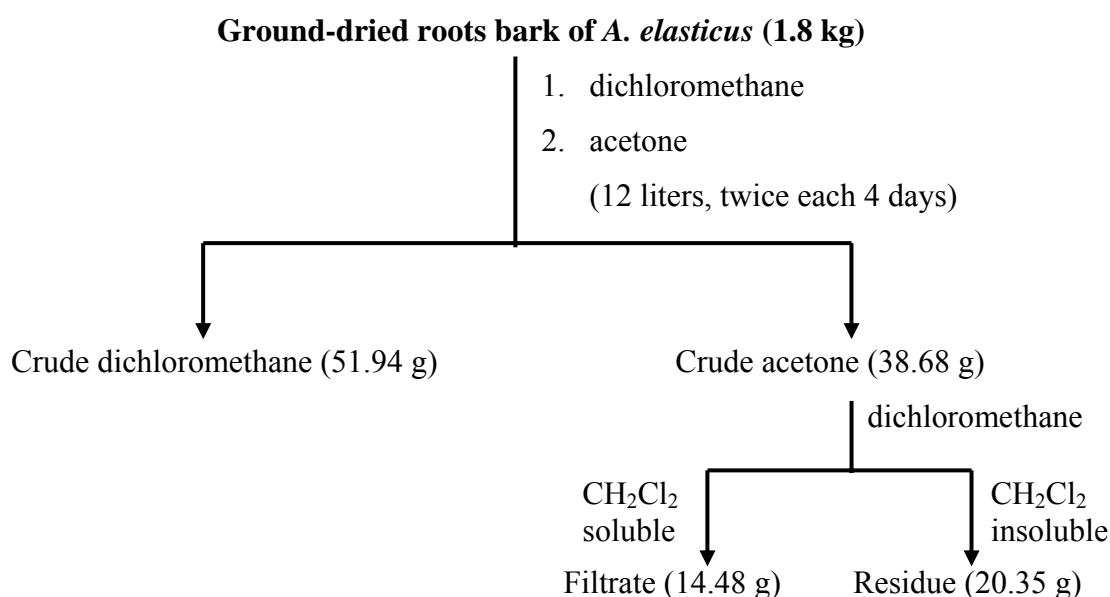
2.2 Plant Material

The root bark of *A. elasticus* was collected from Amphur Kuraburi, Phang Nga province in the southern part of Thailand in May 2008. Identification was made by Mr. Charernsak Saewai, Department of Biology, Faculty of Science, Prince of Songkla University. The specimen (A. Yanya 1Phang-nga: Kuraburi 2/4/2009) have been deposited in the Herbarium of the Department of Biology, Faculty of Science, Prince of Songkla University, Hat Yai, Songkhla, Thailand.

2.3 Extraction and Isolation

A. Root bark

Ground-dried root bark of *Artocarpus elasticus* (1.8 kg) were successively immersed in dichloromethane and acetone at room temperature (each extract 4 days). After removal of solvents, the dark brown viscous dichloromethane extract (51.94 g) and acetone extract (38.68 g) were obtained, respectively. The process of extraction was shown in **Scheme 8**.



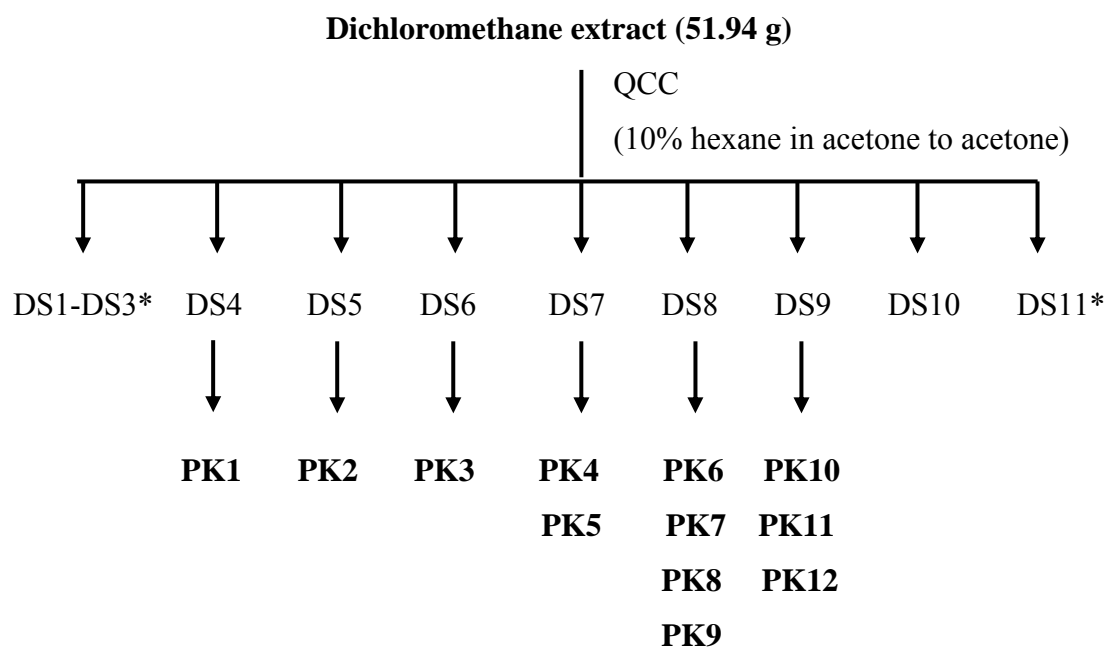
Scheme 8 Extraction of the crude extracts from the root bark of *A. elasticus*

2.3.1 Purification of dichloromethane extract

The dichloromethane extract and the dichloromethane soluble of acetone extract were combined (66.42 g). The extract was chromatographed on quick column chromatography over silica gel 60 using mixed hexane-acetone and acetone as eluent. Fractions with the similar characteristic on TLC were combined to afford 11 fractions (DS1-DS11) (**Table 2**). Further purification of each fraction gave twelve pure compounds (**Scheme 9**).

Table 2 Physical characteristic and weights of fractions obtained from QCC of the dichloromethane extract

Fraction	Weight (g)	Physical characteristic
DS1	3.5635	yellow gel
DS2	16.2667	orange gel
DS3	1.8281	yellow-brown viscous liquid
DS4	0.4929	yellow solid
DS5	1.2326	dark-brown viscous liquid
DS6	1.1245	dark-brown viscous liquid
DS7	2.8728	dark-brown viscous liquid
DS8	4.6366	dark-brown viscous liquid
DS9	5.2972	dark-brown viscous liquid
DS10	0.6546	dark-brown viscous liquid
DS11	5.2632	brown solid



* No further investigation

Scheme 9 Isolation of compounds **PK1-PK12** from dichloromethane extract of the root bark of *A. elasticus*

Fraction DS4 (0.4929 g) was purified by column chromatography over silica gel and eluted with 10% dichloromethane in hexane to give fractions DS4A-DS4D. Subfraction DS4B (41.2 mg) was recrystallized from methanol to yield a mixture of β -sitosterol and stigmasterol (**PK1**; 32.6 g) as colorless needles.

Fraction DS5 (1.2326 g) was further purified by column chromatography over silica gel and eluted with a gradient of acetone-hexane (5% to 20% acetone in hexane) solvent system to give fractions DS5A-DS5F. Fraction DS5D (463.2 mg) was rechromatographed on column chromatography and eluted with 15% acetone in hexane solvent system to give a yellow gum of **PK2** (24.2 mg).

Fraction DS6 (1.1245 g) was further purified by column chromatography over silica gel and eluted with a gradient of acetone-hexane (10% to 20% acetone in hexane) solvent system to give fractions DS6A-DS6K. Fraction DS6F (178.7 mg) was rechromatographed by using 15% acetone in hexane as eluent to give a red-brown gum of **PK3** (13.2 mg).

Fraction DS7 (2.8728 g) was further purified by column chromatography over silica gel and eluted with a gradient of acetone-hexane (20% to 30% acetone in hexane) solvent system to give fractions DS7A-DS7G. Fraction DS7C (555.0 mg) was further purified by column chromatography over silica gel and eluted with a mixed solvent of hexane-dichloromethane-acetone (8:1:1) to give fractions DS7C1-DS7C7. Fraction DS7C4 (90.2 mg) was rechromatographed on column chromatography using 20% acetone in hexane as eluent give colorless gum of **PK4** (2.5 mg). Fraction DS7F (134.4 mg) was further purified by column chromatography over silica gel and eluted with 15% acetone in hexane give fractions DS7F1-DS7F7. Fraction DS7F4 (51.2 mg) was further purified by column chromatography over silica gel and eluted with a mixed solvent of hexane-dichloromethane-acetone (3:1:1) to afford an orange solid of **PK5** (7.7 mg) in subfraction DS7F4C.

Fraction DS8 (4.6366 g) was further purified by column chromatography over silica gel and eluted with a gradient of acetone-hexane (20% to 30 % acetone in hexane) solvent system to give fractions DS8A- DS8I. Fraction DS8C (334.4 mg) was further purified by column chromatography over silica gel and eluted with a mixed solvent of hexane-dichloromethane-acetone (3:1:1) to give fractions DS8C1-DS8C5. Fraction DS8C4 (169.2 mg) was rechromatographed on

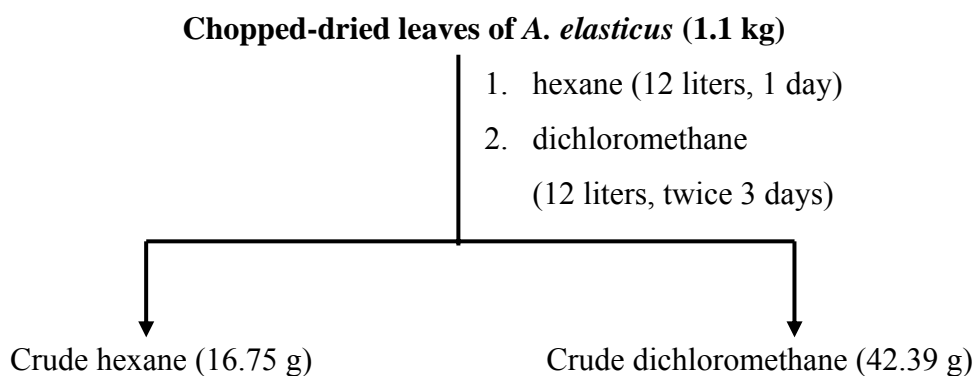
column chromatography and eluted with a mixed solvent of hexane-dichloromethane-acetone (8:1:1) to give a yellow gum of **PK6** (3.4 mg). Fraction DS8D (340.6 mg) was further purified by column chromatography over silica gel and eluted with a mixed solvent of hexane-dichloromethane-acetone (3:1:1) to give fractions DS8D1-DS8D8. Fraction DS8D4 (104.9 mg) was rechromatographed on column chromatography and eluted with a mixed solvent of hexane-dichloromethane-acetone (3:1:1) to give fractions DS8D4A-DS8D4K. Subfraction DS8D4H was further purified by preparative TLC with 25% acetone in hexane to give a yellow solid of **PK7** (4.8 mg). Fraction DS8D6 was filtered and washed with dichloromethane to give **PK8** (7.2 mg). Fraction DS8F (377.1 mg) was further separated by column chromatography over silica gel and eluted with a mixed solvent of hexane-dichloromethane-acetone (3:1:1) to give a yellow solid of **PK9** (29.9 mg).

Fraction DS9 (5.2972 g) was further purified by column chromatography over silica gel and eluted with a gradient of hexane-acetone (10% to 30% acetone in hexane) solvent system to give fractions DS9A-DS9G. Fraction DS9B (197.9 mg) which appears in a yellow solid mixed with yellow viscous liquid, was dissolved in hexane to give a yellow solid of **PK10** (12.6 mg). Fraction DS9C (297.9 mg) was further purified by column chromatography over silica gel and eluted with a mixed solvent of hexane-dichloromethane-acetone (3:1:1) to give fractions DS9C1-DS9C4. Fraction DS9C3 (123.4 mg) was rechromatographed on column chromatography and eluted with a mixed solvent of hexane-dichloromethane-acetone (3:1:1) to give a brown solid of **PK11** (16.9 mg).

Fraction DS9D (694.1 mg) was further purified by column chromatography over silica gel and eluted with 15% acetone in hexane to give fractions DS9D1-DS9D6. Fraction DS9D3 (26.4 mg) was further purified by column chromatography over silica gel and eluted with 25% acetone in hexane to give a yellow solid of **PK12** (4.1 mg).

B. Leaves

Chopped-dried leaves of *A. elasticus* (1.1 kg) was immersed at room temperature in hexane (extract 1 day) for get rid of chlorophyll compound which have been a major component in this extract. After the solution and the residue were each other isolated, the residue was further immersed in dichloromethane at room temperature (extract 3 days). The solvent was evaporated under reduced pressure to give dichloromethane extract as green-brown viscous gum (42.39 g). The process of extraction was shown in **Scheme 10**.



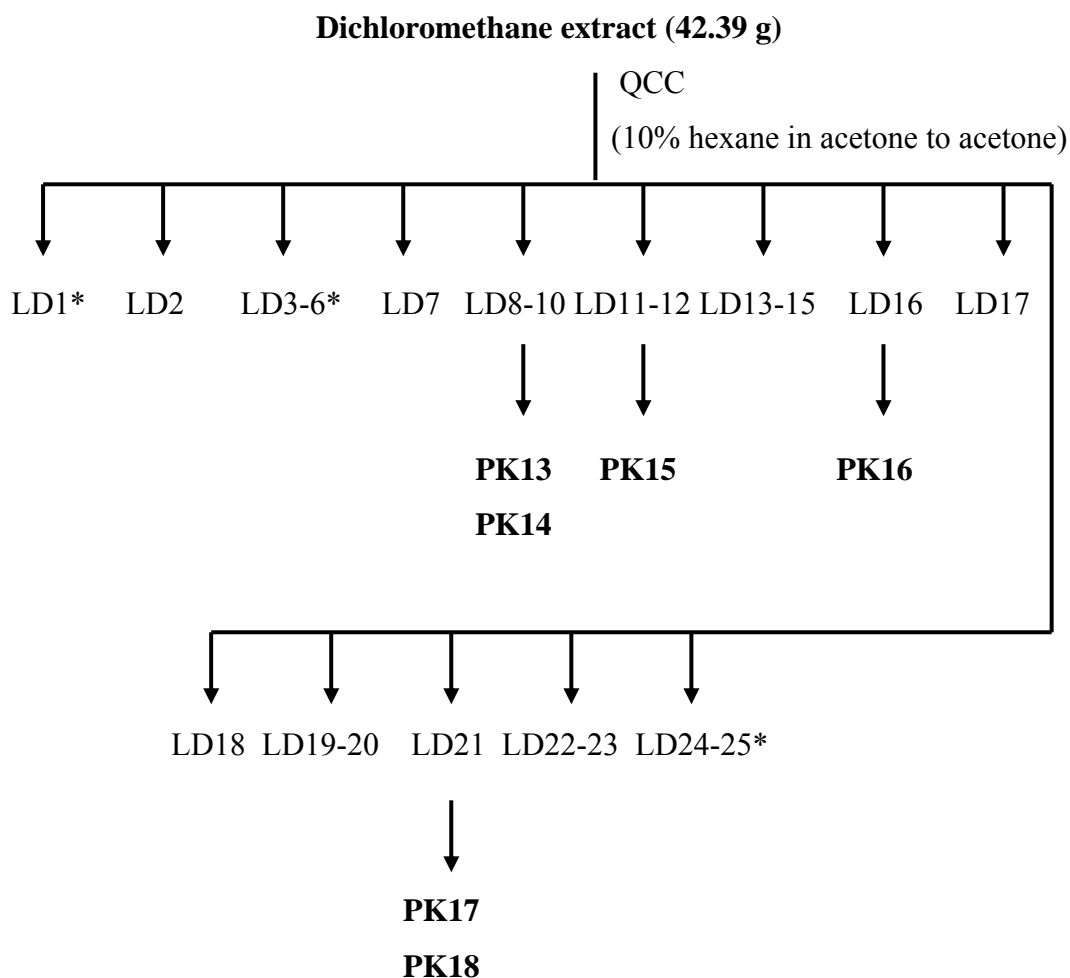
Scheme 10 Extraction of the crude extracts from the leaves of *A. elasticus*

2.3.2 Purification of dichloromethane extract

The dichloromethane extract (42.39 g) was chromatographed on quick column chromatography over silica gel 60 using solvent of increasing polarity from hexane through acetone. Fractions with the similar characteristic on TLC were combined to afford 25 fractions (LD1-LD25) (**Table 3**). Further purification of each fraction gave six pure compounds (**Scheme 11**).

Table 3 Physical characteristic and weights of fractions obtained from QCC of the dichloromethane extract

Fraction	Weight (g)	Physical characteristic
LD1	2.8928	yellow gel
LD2	1.8750	orange gum
LD3	1.6742	green viscous liquid
LD4	3.3039	green viscous liquid
LD5	1.2316	green viscous liquid
LD6	0.8712	green viscous liquid
LD7	1.2311	green viscous liquid
LD8	3.3490	brown solid
LD9	1.2562	green-yellow viscous liquid
LD10	0.7602	green-yellow viscous liquid
LD11	1.1365	green-yellow viscous liquid
LD12	0.7605	yellow viscous liquid
LD13	1.3433	yellow viscous liquid
LD14	2.2632	yellow viscous liquid
LD15	1.2109	brown viscous liquid
LD16	1.4309	brown viscous liquid
LD17	0.6549	brown viscous liquid
LD18	1.4390	brown viscous liquid
LD19	1.3412	brown viscous liquid
LD20	0.4256	brown viscous liquid
LD21	2.1172	dark-brown viscous liquid
LD22	1.3411	dark-brown viscous liquid
LD23	1.9134	dark-brown viscous liquid
LD24	1.2122	dark-brown viscous liquid
LD25	2.6129	dark-brown viscous liquid



* No further investigation

Scheme 11 Isolation of compounds **PK13-PK18** from dichloromethane extract of the leaves of *A. elasticus*

Fraction LD8 (3.3490 g) as a solid was filtered and washed with dichloromethane to give **PK13** (1.9213 g) as a major component of dichloromethane extract. Similarly, fraction LD9 (1.2562 g), LD10 (0.7602 g), LD11 (1.1365 g) and LD12 (0.7605 g) was filtered and washed with dichloromethane to give **PK13** (0.9871 g). On the basis of TLC characteristics, the filtrate of LD8, LD9 and LD10 were combined (2.3710 g). In the same, the filtrated of LD11 and LD12 were combined (0.5382 g) before the fraction was further purified.

The filtrate (2.3710 g) was further purified by column chromatography over silica gel and eluted with a gradient of acetone-hexane (15% to 25% acetone in hexane) to give fractions LD8.10A-LD8.10H. Fraction LD8.10E (300.7 mg) was further separated by column chromatography over silica gel and eluted with a mixed solvent of hexane-dichloromethane-acetone (8:1:1) to give a green gum of **PK14** (8.4 mg) in fraction LD8.10E4.

The filtrate (538.2 mg) was further purified by column chromatography over silica gel and eluted with a gradient of acetone-hexane (15% to 25% acetone in hexane) to give fractions LD11.12A-LD11.12F. Fraction LD11.12D (191.8 mg) was further purified by column chromatography over silica gel and eluted with a mixed solvent of hexane-dichloromethane-acetone (7:2:1) to give fractions LD11.12D1-LD11.12D4. Subfraction LD11.12D2 (50.1 mg) was separated by column chromatography over silica gel using a mixed solvent of hexane-dichloromethane-acetone (7:2:1) as eluent to afford a yellow solid of **PK15** (5.9 mg).

Fraction LD16 (1.4309 g) was purified by column chromatography over silica gel and eluted with 15% acetone in hexane to give fractions LD16A-LD16G. Fraction LD16E (212.3 mg) was further purified by column chromatography over silica gel and eluted with a gradient of acetone-hexane (15% to 25% acetone in hexane) to afford a pale yellow solid of **PK16** (10.2 mg) in subfraction DS16E3.

Fraction LD21 (2.1172 g) was further purified by column chromatography over SephadexTM LH-20 and eluted with dichloromethane-methanol (4:1) to give fractions LD21A-LD21N. Fraction LD21K (101.2 mg) as a solid was filtered and washed with dichloromethane to give a yellow solid of **PK17** (15.9 mg). The filtrate was collected and combined with LD21J and LD21L based on TLC characteristics to give fraction LD21J.L. This fraction was further purified by column chromatography over SephadexTM LH-20 and eluted with dichloromethane-methanol (4:1) to give a yellow solid of **PK18** (17.2 mg).

PK1:

a mixture of β -sitosterol and stigmasterol, colorless needles

IR (neat) ν_{\max} (cm^{-1}): 3425 (O-H stretching) and 1642 (C=C stretching)

^1H NMR spectral data; δ_{H} 3.57-3.47 (*m*, H-3), δ_{H} 5.36-5.34 (*br d*, $J = 5.1$ Hz, H-6),
5.16 (*dd*, $J = 8.4, 15.1$ Hz, H-22) and 5.01 (*dd*, $J = 8.4, 15.1$ Hz, H-23)

PK2:

(*E*)-4-(3',4'-dimethoxy-phenyl)-3-butenyl acetate, colorless viscous liquid

UV λ_{\max} (CHCl_3) ($\log \epsilon$): 244 (2.45), 273 (2.40) and 305 (2.42) nm

IR (neat) ν_{\max} (cm^{-1}): 1738 (C=O stretching) and 1515 (C=C stretching)

^1H NMR and ^{13}C NMR spectral data, see **Table 4**.

PK3:

5a,6-dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-9-(3-methyl-2-buten-1-yl)-
5*H*,7*H*,11*H*-benzofuro[3,4-*bc*]pyrano[3,2-*h*]xanthen-7-one, yellow solid

$[\alpha]_{\text{D}}^{28} = -31^\circ$ (*c* 0.1, acetone)

m.p. 249-251 °C

UV λ_{\max} (CH_3OH) ($\log \epsilon$): 236 (4.46), 256 (4.42), 265 (4.41), 278 (4.50), 335 (4.09)
and 391 (4.27) nm

IR (neat) ν_{\max} (cm^{-1}): 3371 (O-H stretching) and 1629 (C=O stretching)

^1H NMR and ^{13}C NMR spectral data, see **Table 6**.

PK4:

4-hydroxybenzaldehyde, colorless gum

UV λ_{\max} (CH_3OH) ($\log \epsilon$): 257 (3.68) and 275 (3.04) nm

IR (neat) ν_{\max} (cm^{-1}): 3367 (O-H stretching) and 1684 (C=O stretching)

^1H NMR and ^{13}C NMR spectral data, see **Table 8**.

PK5:

2,3,8-trihydroxy-11,11-dimethyl-13-(3-methyl-2-butenyl)-6-(2-methyl-1-propenyl)-
6*H*,7*H*,11*H*-bis[1]benzopyrano[4,3-*b*:6',7'-*e*]pyran-7-one, orange solid

$[\alpha]_{\text{D}}^{28} = -11^\circ$ (*c* 0.1, acetone)

m.p. 221-222 °C

UV λ_{\max} (CH_3OH) ($\log \epsilon$): 242 (5.00), 244 (4.98), 300 (5.04) and 384 (4.80) nm

IR (neat) ν_{\max} (cm^{-1}): 3345 (O-H stretching) and 1625 (C=O stretching)

^1H NMR and ^{13}C NMR spectral data, see **Table 9**.

PK6:

(*E*)-4-(3',4'-dimethoxyphenyl)but-3-en-1-ol, colorless viscous liquid

UV λ_{\max} (CHCl_3) ($\log \epsilon$): 240 (2.27) and 271 (2.14) nm

IR (neat) ν_{\max} (cm^{-1}): 3419 (O-H stretching) and 1515 (C=C stretching)

^1H NMR and ^{13}C NMR spectral data, see **Table 10**.

PK7:

8-(2,4-dihydroxyphenyl)-5-hydroxy-2,2-dimethyl-7-(3-methyl-2-butenyl)-2*H*,6*H*-benzo[1,2-*b*:5,4-*b'*]dipyran-6-one, yellow solid

m.p. 125-126 °C

UV λ_{\max} (CH_3OH) ($\log \epsilon$): 237 (4.23), 283, (3.59) and 344 (3.34) nm

IR (neat) ν_{\max} (cm^{-1}): 3230 (O-H stretching) and 1599 (C=O stretching)

^1H NMR and ^{13}C NMR spectral data, see **Table 12**.

PK8:

5*a*,6-dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-5*H*,7*H*,11*H*-benzofuro[3,4-*bc*]pyrano[3,2-*h*]xanthen-7-one, yellow solid

$[\alpha]_{\text{D}}^{28} = +5^\circ$ (*c* 0.2, CH_2Cl_2)

m.p. 284-285 °C

UV λ_{\max} (CH_3OH) ($\log \epsilon$): 228 (4.56), 257 (4.42), 272 (4.58), 312 (4.19), 330 (4.25) and 391 (4.33) nm

IR (neat) ν_{\max} (cm^{-1}): 3405 (O-H stretching) and 1642 (C=O stretching)

^1H NMR and ^{13}C NMR spectral data, see **Table 13**.

PK9:

6,7-dihydro-5,9,11,14-tetrahydroxy-3,3-dimethyl-6-(1-methylethenyl)-(-)-3*H*,8*H*-pyrano[3',2':4,5]benzo[1,2-*c*]xanthen-8-one, red-brown gum

$[\alpha]_{\text{D}}^{27} = -82^\circ$ (*c* 0.2, acetone)

UV λ_{\max} (CH_3OH) ($\log \epsilon$): 263 (4.39), 269 (4.36), 307 (3.71) and 379 (4.11) nm

IR (neat) ν_{\max} (cm^{-1}): 3402 (O-H stretching) and 1655 (C=O stretching)

^1H NMR and ^{13}C NMR spectral data, see **Table 15**.

PK10:

8,9-dihydro-6,10,11,13-tetrahydroxy-3,3-dimethyl-9-(1-methylethenyl)-3*H*,7*H*-benzo[*c*]pyrano[3,2-*h*]xanthen-7-one, yellow solid

$[\alpha]_D^{26} = +43^\circ$ (*c* 0.2, CH₂Cl₂)

m.p. 163-164 °C

UV λ_{\max} (CH₃OH) (log ϵ): 227 (3.85), 272 (3.93) and 384 (3.56) nm

IR (neat) ν_{\max} (cm⁻¹): 3349 (O-H stretching) and 1652 (C=O stretching)

¹H NMR and ¹³C NMR spectral data, see **Table 16**.

PK11:

(*E*)-3-(4'-hydroxy-3'-methoxyphenyl)-2-propenoic acid, brown-yellow gum

UV λ_{\max} (CH₃OH) (log ϵ): 244 (3.39), 273 (3.36), 299 (3.28) and 385 (3.02) nm

IR (neat) ν_{\max} (cm⁻¹): 3350 (O-H stretching) 1712 (C=O stretching) and 1513 (C=C stretching)

¹H NMR and ¹³C NMR spectral data, see **Table 17**.

PK12:

5-hydroxy-8,8-dimethyl-3-(3-methyl-2-butenyl)-2-(2,4,5-trihydroxyphenyl)-4*H*,8*H*-benzo[1,2-*b*:3,4-*b'*]dipyran-4-one, brown-yellow solid

m.p. 217-219 °C

UV λ_{\max} (CH₃OH) (log ϵ): 224 (4.34), 258 (4.41), 266 (4.47), 271 (4.47), 302 (3.90) and 352 (3.89) nm

IR (neat) ν_{\max} (cm⁻¹): 3402 (O-H stretching) and 1655 (C=O stretching)

¹H NMR and ¹³C NMR spectral data, see **Table 18**.

PK13:

1-(2,4-dihydroxyphenyl)-3-(8-hydroxy-2,2-dimethyl-7-(3-methylbut-2-enyl)-2*H*-chromen-6-yl)propan-1-one, brown-yellow gum

UV λ_{\max} (CH₃OH) (log ϵ): 231 (3.19), 273 (3.21) and 313 (2.89) nm

IR (neat) ν_{\max} (cm⁻¹): 3383 (O-H stretching) and 1634 (C=O stretching)

¹H NMR and ¹³C NMR spectral data, see **Table 19**.

PK14:

1-(2,4-dihydroxyphenyl)-3-(3,4-dihydroxy-2,5-bis(3-methylbut-2-enyl)phenyl)propan-1-one, yellow solid

m.p. 170 °C

UV λ_{\max} (CH₃OH) (log ϵ): 237 (3.24), 274 (3.26) and 313 (3.28) nm

IR (neat) ν_{\max} (cm⁻¹): 3422 (O-H stretching) and 1629 (C=O stretching)

¹H NMR and ¹³C NMR spectral data, see **Table 21**.

PK15:

1-(2,4-dihydroxyphenyl)-3-(7-((3,3-dimethyloxiran-2-yl)methyl)-8-hydroxy-2,2-dimethyl-2*H*-chromen-6-yl)propan-1-one, yellow gum

$[\alpha]_{\text{D}}^{26} = -14^{\circ}$ (*c* 0.1, acetone)

UV λ_{\max} (CH₃OH) (log ϵ): 245 (3.28), 274 (3.27) and 311 (3.33) nm

IR (neat) ν_{\max} (cm⁻¹): 3390 (O-H stretching) and 1631 (C=O stretching)

¹H NMR and ¹³C NMR spectral data, see **Table 22**.

PK16:

(*S*)-2-(2,4-dihydroxyphenyl)-5-hydroxy-7-methoxychroman-4-one, pale yellow solid

$[\alpha]_{\text{D}}^{27} = -3^{\circ}$ (*c* 0.2, acetone)

m.p. 210-211 °C

UV λ_{\max} (CH₃OH) (log ϵ): 244 (3.30), 273 (3.28) and 305 (3.32) nm

IR (neat) ν_{\max} (cm⁻¹): 3343 (O-H stretching) and 1597 (C=O stretching)

¹H NMR and ¹³C NMR spectral data, see **Table 23**.

PK17:

1-(2,4-dihydroxyphenyl)-3-(4-hydroxy-2,2-dimethyl-5-(3-methylbut-2-enyl)-2,7*b*-dihydro-1*aH*-oxireno[2,3-*c*]chromen-6-yl)propan-1-one, a yellow solid

$[\alpha]_{\text{D}}^{26} = +7^{\circ}$ (*c* 0.2, acetone)

m.p. 179-180 °C

UV λ_{\max} (CH₃OH) (log ϵ): 243 (3.27), 275 (3.26) and 310 (3.32) nm

IR (neat) ν_{\max} (cm⁻¹): 3223 (O-H stretching) and 1624 (C=O stretching)

¹H NMR and ¹³C NMR spectral data, see **Table 24**.

PK18:

1-(2,4-dihydroxyphenyl)-3-(7-hydroxy-6-(3-methylbut-2-enyl)benzofuran-5-yl)
propan-1-one, yellow gum

UV λ_{max} (CH₃OH) (log ϵ): 228 (3.18), 276 (3.19) and 314 (2.99) nm

IR (neat) ν_{max} (cm⁻¹): 3352 (O-H stretching) and 1629 (C=O stretching)

¹H NMR and ¹³C NMR spectral data, see **Table 25**.

CHAPTER 3

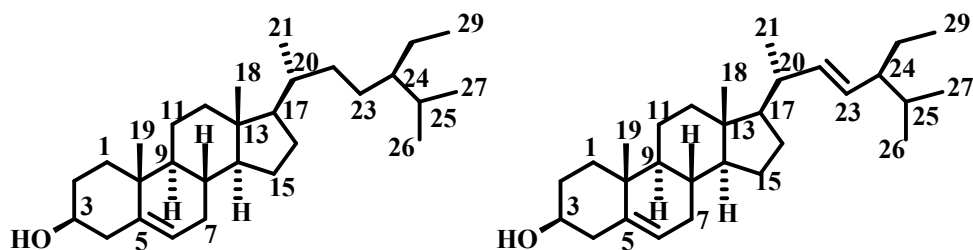
RESULTS AND DISCUSSION

3.1 Structure elucidation of compounds from the root bark and leaves of *A. elasticus*

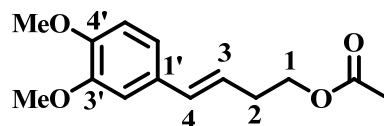
The crude dichloromethane and acetone extracts from the root bark of *A. elasticus* were subjected to repeated quick column and column chromatography over silica gel to furnish twelve known compounds. They were identified as a triterpenoids; a mixture of β -sitosterol and stigmasterol (**PK1**), (*E*)-4-(3',4'-dimethoxy-phenyl)-3-butenyl acetate (**PK2**), 5a,6-dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-9-(3-methyl-2-buten-1-yl)-5*H*,7*H*,11*H*-benzofuro[3,4-*bc*]pyrano[3,2-*h*]xanthen-7-one (**PK3**), 4-hydroxybenzaldehyde (**PK4**), 2,3,8-trihydroxy-11,11-dimethyl-13-(3-methyl-2-butenyl)-6-(2-methyl-1-propenyl)-6*H*,7*H*,11*H*-bis[1]benzopyrano[4,3-*b*:6',7'-*e*]pyran-7-one (**PK5**), (*E*)-4-(3',4'-dimethoxyphenyl)but-3-en-1-ol (**PK6**), 2-(2,4-dihydroxyphenyl)-5-hydroxy-8,8-dimethyl-3-(3-methylbut-2-enyl)pyrano[3,2-*g*]chromen-4(8*H*)-one (**PK7**), 5a,6-dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-5*H*,7*H*,11*H*-benzofuro[3,4-*bc*]pyrano[3,2-*h*]xanthen-7-one (**PK8**), 6,7-dihydro-5,9,11,14-tetrahydroxy-3,3-dimethyl-6-(1-methylethenyl)-(-)-3*H*,8*H*-pyrano[3',2':4,5]benzo[1,2-*c*]xanthen-8-one (**PK9**), 8,9-dihydro-6,10,11,13-tetrahydroxy-3,3-dimethyl-9-(1-methylethenyl)-3*H*,7*H*-benzo[*c*]pyrano[3,2-*h*]xanthen-7-one (**PK10**), (*E*)-3-(4'-hydroxy-3'-methoxyphenyl)-2-propenoic acid (**PK11**) and 5-hydroxy-8,8-dimethyl-3-(3-methyl-2-butenyl)-2-(2,4,5-trihydroxyphenyl)-4*H*,8*H*-benzo[1,2-*b*:3,4-*b*]dipyran-4-one (**PK12**). Purification of the crude dichloromethane from the leaves furnish five new prenylated dihydrochalcones; 1-(2,4-dihydroxyphenyl)-3-(8-hydroxy 2,2-dimethyl-7-(3-methylbut-2-enyl)-2*H*-chromen-6-yl)propan-1-one (**PK13**), 1-(2,4-dihydroxyphenyl)-3-(3,4-dihydroxy-2,5-bis(3-methylbut-2-enyl)phenyl)propan-1-one (**PK14**), 1-(2,4-dihydroxyphenyl)-3-(7-((3,3-dimethyloxiran-2-yl)methyl)-8-hydroxy-2,2-dimethyl-2*H*-chromen-6-yl)propan-1-one (**PK15**), 1-(2,4-dihydroxyphenyl)-3-(4-hydroxy-2,2-dimethyl-5-(3-methylbut-2-enyl)-2,7*b*-dihydro-1*aH*-oxireno[2,3-*c*]chromen-6-yl)propan-1-one (**PK17**) and 1-(2,4-dihydroxyphenyl)-3-(7-hydroxy-6-(3-

methylbut-2-enyl)benzofuran-5-yl)propan-1-one (**PK18**), and one known compound; (*S*)-2-(2,4-dihydroxyphenyl)-5-hydroxy-7-methoxychroman-4-one (**PK16**).

Their structures were elucidated mainly by 1D and 2D NMR spectroscopic data: ^1H , ^{13}C NMR, DEPT 135°, DEPT 90°, HMQC, HMBC, COSY and NOESY. The physical data of the known compounds were also compared with the reported values.

PK1**Mixture of β -sitosterol and stigmasterol**

The mixture of **PK1** was obtained as colorless needles. The ^1H NMR spectrum exhibited the characteristic resonances of an oxymethine protons at δ_{H} 3.57-3.47 (*m*, H-3), three olefinic protons at δ_{H} 5.36-5.34 (*br d*, $J = 5.1$ Hz, H-6), 5.16 (*dd*, $J = 8.4, 15.1$ Hz, H-22) and 5.01 (*dd*, $J = 8.4, 15.1$ Hz, H-23). These spectral data were in agreement with that of the mixture of β -sitosterol and stigmasterol (**PK1**) (Boonnak, 2006).

PK2**(E)-4-(3',4'-Dimethoxyphenyl)-3-butenyl acetate**

PK2 was obtained as colorless viscous liquid. The UV spectrum showed maximum absorption bands at 244, 273 and 305 nm. The IR spectrum showed the stretching of C=O of ester group at 1738 cm^{-1} and C=C at 1515 cm^{-1} . The ^1H NMR spectral data (**Table 4**) displayed an ABX signal of aromatic protons at δ_{H} 6.81 (*d*, $J = 8.1$ Hz, H-5'), 6.87 (*dd*, $J = 8.1, 1.8$ Hz, H-6') and 6.90 (*d*, $J = 1.8$ Hz, H-2') indicating a trisubstituted benzene ring. A substituent group was assigned for 3-butenyl acetate side chain which the resonances of *trans*-vinylic protons H-3 and H-4 were at δ_{H} 6.02 and 6.41 ($J = 15.9$ Hz), methylene protons H-1 and H-2 were at δ_{H} 4.18 (*t*, $J = 6.9$ Hz) and 2.53 (*qd*, $J = 6.9, 1.8$ Hz), and acetyl proton was at δ_{H} 2.06 (*s*). The HMBC correlations of H-1 to C=O and vinylic carbon C-3 confirmed the structure of that side chain. The side chain was positioned at C-1' according to the HMBC correlations of H-4 to C-2' (δ_{C} 108.8) and C-6' (δ_{C} 119.1) and of H-3 to C-1' (δ_{C} 130.5). The *ortho*-methoxyl groups which resonated at δ_{H} 3.90 (*s*, 3H) and δ_{H} 3.88 (*s*, 3H) were proposed for 3'-OCH₃ and 4'-OCH₃, respectively. Therefore, **PK2** was identified as (*E*)-4-(3',4'-dimethoxyphenyl)-3-butenyl acetate (Han *et al.*, 2003).

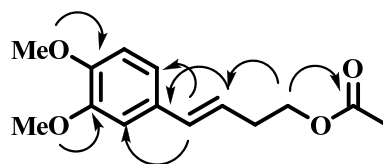
Major HMBC of **PK2**

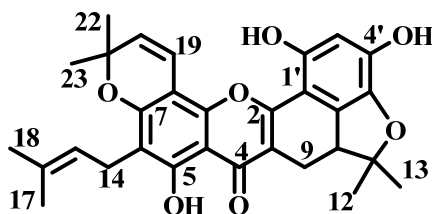
Table 4 ^1H , ^{13}C and HMBC spectral data of **PK2**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1	4.18 (2H, <i>t</i> , 6.9)	63.8 (CH ₂)	1-OC(O)CH ₃ , C-3, C-2
2	2.53 (2H, <i>qd</i> , 6.9, 1.8)	32.3 (CH ₂)	C-1, C-3, C-4
3	6.02 (1H, <i>dt</i> , 15.9, 6.9)	128.3 (CH)	C-1, C-2, C-1'
4	6.41 (1H, <i>d</i> , 15.9)	132.0 (CH)	C-2, C-1', C-2', C-6'
1'	-	130.5 (C)	-
2'	6.90 (1H, <i>d</i> , 1.8)	108.8 (CH)	C-4, C-4', C-6'
3'	-	149.1 (C)	-
4'	-	148.6 (C)	-
5'	6.81 (1H, <i>d</i> , 8.1)	111.2 (CH)	C-1', C-3', C-4', C-6'
6'	6.87 (1H, <i>dd</i> , 8.1, 1.8)	119.1 (CH)	C-4, C-2', C-4'
3'-OCH ₃	3.90 (3H, <i>s</i>)	56.0 (OCH ₃)	C-3'
4'-OCH ₃	3.88 (3H, <i>s</i>)	55.8 (OCH ₃)	C-4'
1-OC(O)CH ₃	-	171.1 (C=O)	-
1-OC(O)CH ₃	2.06 (3H, <i>s</i>)	21.0 (CH ₃)	1-OC(O)CH ₃

recorded in CDCl₃

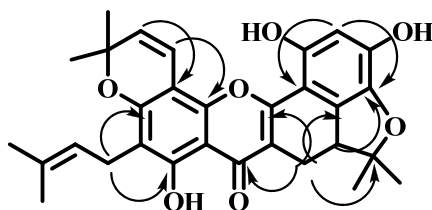
Table 5 ^1H - ^1H COSY spectral data of **PK2**

Proton (δ_{ppm})	Correlated proton (δ_{ppm})
H-4 (6.41)	H-3 (6.02)
H-3 (6.02)	H-4 (6.41), H-2 (2.53)
H-2 (2.53)	H-3 (6.02), H-1 (4.18)

PK3**5a,6-Dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-9-(3-methyl-2-buten-1-yl)-5H,7H,11H-benzofuro[3,4-bc]pyrano[3,2-h]xanthen-7-one : artonin F**

PK3 is a yellow solid, m.p. 249-251 °C, $[\alpha]_{\text{D}}^{28} = -31^{\circ}$ (c 0.1, acetone). The UV spectrum showed maximum absorption bands at 236, 256, 265, 278, 335 and 391 nm. Its IR spectrum showed absorption bands for hydroxyl (3371 cm^{-1}) and conjugated carbonyl (1629 cm^{-1}). The ^1H NMR spectrum (**Table 6**) indicated the presence of three hydroxyl groups (5-OH, δ_{H} 13.42, s ; 2'-OH, δ_{H} 7.78, s ; and 4'-OH, δ_{H} 9.23, s) and one aromatic proton (H-3', δ_{H} 6.38, s). The ^1H -NMR spectrum also indicated the resonances of two methyl groups at δ_{H} 1.35 and 1.66 (each 3H, s) and an ABX spin system at δ_{H} 2.40 (H_{α} -9, t , $J = 15.0\text{ Hz}$), 3.23 (H_{β} -9, dd , $J = 15.0, 7.2\text{ Hz}$), and 3.41 (H-10, dd , $J = 15.0, 7.2\text{ Hz}$), assignable to a furanodihydrobenzoxanthone skeleton (Hakim *et al.*, 2006). In the HMBC experiments, the protons resonated at δ_{H} 3.41 had correlations with carbons resonated at δ_{C} 131.8 (C-6'), 137.2 (C-5'), 93.5 (C-11), 22.7 (C-12) and 28.1 (C-13), confirming the cyclic was formed between C-3 and C-6' whereas the furan moiety was formed at C-5' and C-6' of the aromatic ring. Moreover, a 2,2-dimethylchromene ring was detected from the characteristic signals at δ 6.76 (d , $J = 9.9\text{ Hz}$, H-19), 5.57 (d , $J = 9.9\text{ Hz}$, H-20) and 1.46 (s , 22-CH₃ and 23-CH₃). The correlations of H-19 to C-7 and C-8a and of H-20 to C-8 confirmed the orientation of a chromene ring at C-7 and C-8 position. The spectrum further showed the resonances of methylene protons (H-14, δ_{H} 3.33, d , $J = 7.2\text{ Hz}$), an olefinic proton (H-15, δ_{H} 5.24, mt , $J = 7.2\text{ Hz}$) and methyl protons (CH₃-17, δ_{H} 1.81, s and CH₃-18, δ_{H} 1.68, s), indicating the presence of a prenyl group. The HMBC correlations of H-14 to C-5 (δ_{C} 158.7) and C-7 (δ_{C} 156.4) confirmed the location of the isoprenyl group

at C-6. The HMBC experiments (**Table 6**) supported the assignment. **PK3** was then identified to be artonin F (Hano *et al.*, 1990).



Major HMBC of **PK3**

Table 6 ^1H , ^{13}C and HMBC spectral data of **PK3**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
2	-	160.3 (C)	-
3	-	111.5 (C)	-
4	-	180.7 (C=O)	-
4a	-	104.3 (C)	-
5	-	158.7 (C)	-
6	-	112.5 (C)	-
7	-	156.4 (C)	-
8	-	100.5 (C)	-
8a	-	149.2 (C)	-
9	3.23 (1H $_{\beta}$, <i>dd</i> , 15.0, 7.2) 2.40 (1H $_{\alpha}$, <i>t</i> , 15.0)	19.9 (CH $_2$)	C-2, C-3, C-4, C-10, C-6' C-2, C-3, C-4, C-10, C-11, C-6'
10	3.41 (1H, <i>dd</i> , 15.0, 7.2)	46.6 (CH)	C-9, C-11, C-12, C-13, C-1', C-5', C-6'
11	-	93.5 (C)	-
12	1.35 (3H, <i>s</i>)	22.7 (CH $_3$)	C-10, C-11, C-13
13	1.66 (3H, <i>s</i>)	28.1 (CH $_3$)	C-10, C-12
14	3.33 (2H, <i>d</i> , 7.2)	21.3 (CH $_2$)	C-5, C-6, C-7, C-15, C-16
15	5.24 (1H, <i>mt</i> , 7.2)	122.1 (CH)	C-6, C-14, C-17, C-18
16	-	131.3 (C)	-
17	1.81 (3H, <i>s</i>)	17.9 (CH $_3$)	C-15, C-16, C-18

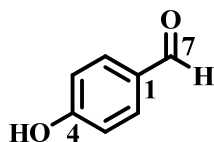
Table 6 (continued)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
18	1.68 (3H, <i>s</i>)	25.8 (CH ₃)	C-15, C-16, C-17
19	6.76 (1H, <i>d</i> , 9.9)	115.3 (CH)	C-7, C-8, C-8a, C-20, C-21
20	5.57 (1H, <i>d</i> , 9.9)	127.1 (CH)	C-8, C-21, C-22, C-23
21	-	77.3 (C)	-
22	1.46 (3H, <i>s</i>)	28.0 (CH ₃)	C-20, C-21, C-23
23	1.46 (3H, <i>s</i>)	28.0 (CH ₃)	C-20, C-21, C-22
1'	-	103.4 (C)	-
2'	-	150.1 (C)	-
3'	6.38 (1H, <i>s</i>)	104.6 (CH)	C-2, C-1', C-2', C-4', C-5'
4'	-	146.4 (C)	-
5'	-	137.2 (C)	-
6'	-	131.8 (C)	-
5-OH	13.42 (1H, <i>s</i>)	-	C-4, C-4a, C-5, C-6
2'-OH	7.78 (1H, <i>s</i>)	-	C-1', C-2', C-3'
4'-OH	9.23 (1H, <i>s</i>)	-	C-5'

recorded in CDCl₃+DMSO-*d*₆

Table 7 ¹H-¹H COSY spectral data of **PK3**

Proton (δ_{ppm})	Correlated proton (δ_{ppm})
H _{α} -9 (2.40) \longleftrightarrow	H _{β} -9 (3.23), H-10 (3.41)
H _{β} -9 (3.23) \longleftrightarrow	H _{α} -9 (2.40), H-10 (3.41)
H-10 (3.41) \longleftrightarrow	H _{α} -9 (2.40), H _{β} -9 (3.23)
H-14 (3.33) \longleftrightarrow	H-15 (5.24), H-17 (1.81), H-18 (1.68)
H-15 (5.24) \longleftrightarrow	H-14 (3.33), H-17 (1.81), H-18 (1.68)
H-17 (1.81) \longleftrightarrow	H-14 (3.33), H-15 (5.24), H-18 (1.68)
H-18 (1.68) \longleftrightarrow	H-14 (3.33), H-15 (5.24), H-17 (1.81)
H-19 (6.76) \longleftrightarrow	H-20 (5.57)

PK4**4-hydroxybenzaldehyde**

PK4 was obtained as a colorless gum. The UV spectrum showed absorption bands at λ_{\max} 257 and 275 nm, indicating the presence of a benzene chromophore. Its IR spectrum showed absorption bands for hydroxyl and carbonyl groups at 3367 and 1684 cm^{-1} , respectively. The ^1H NMR spectrum displayed characteristic signals of a *para*-disubstituted benzene at δ_{H} 7.81 (*d*, $J = 8.4$ Hz, 2H) and 6.96 (*d*, $J = 8.4$ Hz, 2H) and a singlet of an aldehyde proton at δ_{H} 9.88 (1H, *s*, CHO). The ^{13}C NMR spectrum exhibited the signal of a carbonyl carbon of aldehyde group at δ_{C} 190.6. The HMBC experiments were summarized in **Table 8**. Accordingly, the structure of **PK4** was assigned as 4-hydroxybenzaldehyde (Jang *et al.*, 2004).

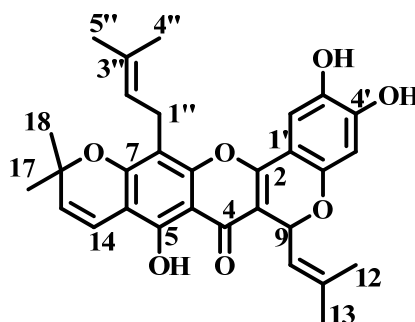
Table 8 ^1H , ^{13}C and HMBC spectral data of **PK4**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1	-	130.0 (C)	-
2/6	7.81 (2H, <i>d</i> , 8.4)	132.5 (CH)	C-3, C-4, C-7
3/5	6.96 (2H, <i>d</i> , 8.4)	116.1 (CH)	C-1, C-2
4	-	161.0 (C)	-
7	9.88 (1H, <i>s</i>)	190.6 (C)	-

recorded in CDCl_3

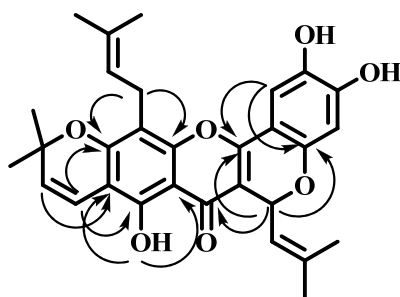
PK5

2,3,8-Trihydroxy-11,11-dimethyl-13-(3-methyl-2-butenyl)-6-(2-methyl-1-propenyl)-6*H*,7*H*,11*H*-bis[1]benzopyrano[4,3-*b*:6',7'-*e*]pyran-7-one
: cycloheterophyllin



PK5 is an orange solid, m.p. 221-222 °C (220 °C; Wei *et al.*, 2005), $[\alpha]_D^{28} = -11^\circ$ (*c* 0.1, acetone) ($[\alpha]_D^{23} = -2^\circ$ (*c* 0.1, acetone); Wei *et al.*, 2005). The UV spectrum showed maximum absorption bands at 242, 244, 300 and 384 nm. The IR spectrum showed the O-H and C=O stretching at 3345 and 1625 cm^{-1} , respectively. The ^{13}C NMR revealed the presence of 30 carbons, including a carbonyl group (δ_{C} 178.8) and six methyl groups, corresponding to a triprenylated flavonoid. The ^1H NMR spectral data (**Table 9**) displayed a hydrogen-bonded hydroxyl group at δ_{H} 12.96 (5-OH) and two isolated aromatic protons at δ_{H} 6.50 (H-3') and 7.26 (H-6'). The spectrum further showed the characteristic signals of 2,2-dimethylchromene ring (δ_{H} 6.72, *d*, $J = 10.2$ Hz, H-14; δ_{H} 5.62, *d*, $J = 10.2$ Hz, H-15; δ_{H} 1.46, *s*, CH₃-17 and δ_{H} 1.47, *s*, CH₃-18) and a prenyl side chain (δ_{H} 5.24, *mt*, $J = 7.2$ Hz, H-2'', 1H; δ_{H} 3.49, *d*, $J = 7.2$ Hz, H-1'', 2H; δ_{H} 1.86, *s*, H-4'', 3H; δ_{H} 1.69, *s*, H-5'', 3H). In the HMBC experiment, proton H-14 (δ_{H} 6.72) correlated to quaternary carbons C-5 (δ_{C} 154.4), C-6 (δ_{C} 105.4), C-7 (δ_{C} 156.5), indicating that a 2,2-dimethylchromene ring was fused to C-6 and C-7 in the A ring. The HMBC correlations of H-1'' to C-7 (δ_{C} 156.5), C-8 (δ_{C} 107.6) and C-8a (δ_{C} 153.6) confirmed the location of the isoprenyl group at C-6. The oxidative cyclization between the allylic methylene of a C-3 prenyl side chain with the C-2' hydroxyl group of the B ring led to 2*H* benzopyran ring system. It exhibited signals of two vinyl methyl groups at δ_{H} 1.95 (*s*, H-12) and δ_{H} 1.69 (*s*, H-13), an olefinic proton at δ_{H} 5.46 (*d*, $J = 9.0$ Hz, H-10) and an oxy-methine proton at δ_{H} 6.20

(*d*, $J = 9.0$ Hz, H-9). The oxidative cyclization was confirmed by the long-range cross peaks of H-9 (δ_{H} 6.20) to C-2 (δ_{C} 155.3), C-3 (δ_{C} 109.9), C-4 (δ_{C} 178.8), C-2' (δ_{C} 151.7). Thus, this compound was 2,3,8-trihydroxy-11,11-dimethyl-13-(3-methyl-2-butenyl)-6-(2-methyl-1-propenyl)-6*H*,7*H*,11*H*-bis[1]benzopyrano[4,3-*b*:6',7'-*e*]pyran-7-one which was corresponded to the previously isolated, cycloheterophyllin (Wei *et al.*, 2005).



Major HMBC of **PK5**

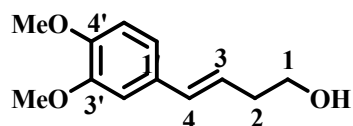
Table 9 ^1H , ^{13}C and HMBC spectral data of **PK5**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
2	-	155.3 (C)	-
3	-	109.9 (C)	-
4	-	178.8 (C=O)	-
4a	-	105.4 (C)	-
5	-	154.4 (C)	-
6	-	105.4 (C)	-
7	-	156.5 (C)	-
8	-	107.6 (C)	-
8a	-	153.6 (C)	-
9	6.20 (1H, <i>d</i> , 9.0)	69.4 (CH)	C-2, C-3, C-4, C-11, C-2'
10	5.46 (1H, <i>d</i> , 9.0)	121.0(CH)	C-12, C-13
11	-	139.4 (C)	-
12	1.95 (3H, <i>s</i>)	18.6 (CH ₃)	C-10, C-11, C-13
13	1.69 (3H, <i>s</i>)	25.9 (CH ₃)	C-10, C-11, C-12

Table 9 (continued)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
14	6.72 (1H, <i>d</i> , 10.2)	115.9 (CH)	C-5, C-7, C-16
15	5.62 (1H, <i>d</i> , 10.2)	127.9 (CH)	C-6, C-16, C-17, C-18
16	-	77.7 (C)	-
17	1.46 (3H, <i>s</i>)	28.1 (CH ₃)	C-15, C-16
18	1.47 (3H, <i>s</i>)	28.2 (CH ₃)	C-15, C-16
1'	-	108.1 (C)	-
2'	-	151.7 (C)	-
3'	6.50 (1H, <i>s</i>)	104.8 (CH)	C-9, C-1', C-2', C-4', C-5'
4'	-	149.5 (C)	-
5'	-	138.6 (C)	-
6'	7.26 (1H, <i>s</i>)	109.4 (CH)	C-2, C-1', C-2', C-4', C-5'
1''	3.49 (2H, <i>d</i> , 7.2)	21.5 (CH ₂)	C-7, C-8, C-8a, C-2'', C-3''
2''	5.24 (1H, <i>mt</i> , 7.2)	122.1 (CH)	C-1'', C-4'', C-5''
3''	-	131.7 (C)	-
4''	1.86 (3H, <i>s</i>)	18.1 (CH ₃)	C-3'', C-5''
5''	1.69 (3H, <i>s</i>)	25.8 (CH ₃)	C-3'', C-4''
5-OH	12.96 (1H, <i>s</i>)	-	C-4a, C-5, C-6

recorded in CDCl₃

PK6**(E)-4-(3',4'-Dimethoxyphenyl)but-3-en-1-ol**

PK6 was obtained as colorless viscous liquid. The UV spectrum showed maximum absorption bands at 240 and 271 nm. The IR spectrum showed the O-H and C=C stretching at 3419 cm^{-1} and 1515 cm^{-1} , respectively. Its ^1H NMR spectral data showed the resonances of H-5' (δ_{H} 6.82, *d*), H-6' (δ_{H} 6.90, *dd*), H-2' (δ_{H} 6.93, *d*), 3'-OCH₃ (δ_{H} 3.92, *s*), 4'-OCH₃ (δ_{H} 3.89, *s*), H-4 (δ_{H} 6.46, *d*, $J = 15.6\text{ Hz}$), H-3 (δ_{H} 6.09, *dt*, $J = 15.6, 6.6\text{ Hz}$), H-2 (δ_{H} 2.50, *qd*, $J = 6.6, 1.2\text{ Hz}$) and H-1 (δ_{H} 3.78, *t*, $J = 6.6\text{ Hz}$). Its ^1H NMR, ^{13}C NMR and HMBC spectral data were similar to those of **PK2**, except the absence of a C=O carbon signal. Therefore, **PK6** was identified as (*E*)-4-(3',4'-dimethoxyphenyl)but-3-en-1-ol (Han *et al.*, 2003). The HMBC correlations (**Table 10**) confirmed the assigned structure.

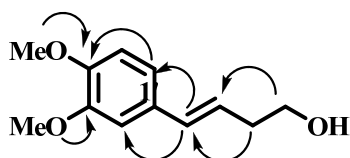
Major HMBC of **PK6**

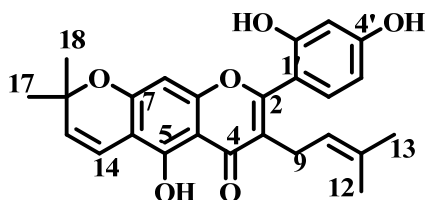
Table 10 ^1H , ^{13}C and HMBC spectral data of **PK6**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1	3.78 (2H, <i>t</i> , 6.6)	62.1 (CH ₂)	C-2, C-3
2	2.50 (2H, <i>qd</i> , 6.6, 1.2)	36.4 (CH ₂)	C-1, C-3, C-4
3	6.09 (1H, <i>dt</i> , 15.6, 6.6)	128.4 (CH)	C-1, C-2, C-1'
4	6.46 (1H, <i>d</i> , 15.6)	132.5 (CH)	C-2, C-1', C-2', C-6'
1'	-	130.4 (C)	-
2'	6.93 (1H, <i>d</i> , 1.8)	108.7 (CH)	C-4, C-4', C-6'
3'	-	148.6 (C)	-
4'	-	149.1 (C)	-
5'	6.82 (1H, <i>d</i> , 8.1)	111.2 (CH)	C-1', C-3', C-4'
6'	6.90 (1H, <i>dd</i> , 8.1, 1.8)	119.1 (CH)	C-4, C-2', C-4'
3'-OCH ₃	3.92 (3H, <i>s</i>)	55.8 (OCH ₃)	C-3'
4'-OCH ₃	3.89 (3H, <i>s</i>)	55.9 (OCH ₃)	C-4'

recorded in CDCl₃

Table 11 ^1H - ^1H COSY spectral data of **PK6**

Proton (δ_{ppm})		Correlated proton (δ_{ppm})
H-4 (6.46)	↔	H-3 (6.09)
H-3 (6.09)	↔	H-4 (6.46), H-2 (2.50)
H-2 (2.50)	↔	H-3 (6.09), H-1 (3.78)

PK7**2-(2,4-Dihydroxyphenyl)-5-hydroxy-8,8-dimethyl-3-(3-methylbut-2-enyl)pyrano [3,2-g]chromen-4(8H)-one : cudraflavone B**

PK7 is a yellow solid, m.p. 125-126 °C (126 °C; Ryu *et al.*, 2009). The UV spectrum showed maximum absorption bands at 237, 283, and 344 nm. The IR spectrum exhibited the absorption bands of hydroxyl group at 3230 cm^{-1} . The ^1H NMR spectral data (**Table 12**) revealed the presence of a chelated hydroxyl group (δ_{H} 13.18, *s*), an isolated aromatic proton (δ_{H} 6.27, *s*), and a 1,2,4-trisubstituted benzene [δ_{H} 7.21 (*d*, $J = 9.0$ Hz, H-6'), 6.52 (*d*, $J = 9.0$ Hz, H-5') and 6.51 (*s*, H-3')]. The spectrum further showed signals corresponded to a prenyl group [δ_{H} 3.13 (*d*, $J = 6.6$ Hz, 2H, H-9), 5.17 (*mt*, $J = 6.6$ Hz, 1H, H-10), 1.66 (*s*, 3H, H-13) and 1.48 (*s*, 3H, H-12)] and a 2,2-dimethylchromene ring [δ_{H} 6.73 (*d*, $J = 10.2$ Hz, 1H, H-14), 5.48 (*d*, $J = 10.2$ Hz, 1H, H-15), 1.46 (*s*, 6H, H-17 and H-18)]. The HMBC correlations of H-9 to carbonyl carbon (δ_{C} 182.2), C-2 (δ_{C} 159.3) indicated that a prenyl side chain was at C-3 position. While the correlations of H-14 to C-5 (δ_{C} 156.5), C-7 (δ_{C} 159.2) and of H-15 to C-6 (δ_{C} 105.4) indicated that a chromene ring was fused at C-6 and C-7 position. Thus **PK7** was assigned as 2-(2,4-dihydroxyphenyl)-5-hydroxy-8,8-dimethyl-3-(3-methylbut-2-enyl)pyrano[3,2-g]chromen-4(8H)-one which corresponded to cudraflavone B (Ryu *et al.*, 2009).

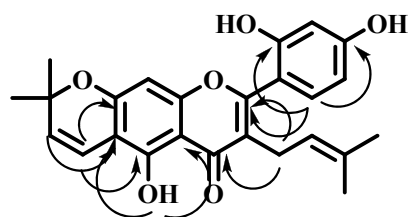
Major HMBC of **PK7**

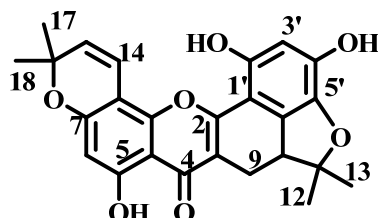
Table 12 ^1H , ^{13}C and HMBC spectral data of **PK7**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
2	-	159.3 (C)	-
3	-	121.4 (C)	-
4	-	182.2 (C=O)	-
4a	-	103.6 (C)	-
5	-	156.5 (C)	-
6	-	105.4 (C)	-
7	-	159.2 (C)	-
8	6.27 (1H, <i>s</i>)	94.6 (CH)	C-4a, C-6, C-7, C-8a
8a	-	157.0 (C)	-
9	3.13 (2H, <i>d</i> , 6.6)	24.4 (CH ₂)	C-2, C-3, C-4, C-10, C-11
10	5.17 (1H, <i>mt</i> , 6.6)	120.9 (CH)	-
11	-	133.3 (C)	-
12	1.48 (3H, <i>s</i>)	17.7 (CH ₃)	C-10, C-11, C-13
13	1.66 (3H, <i>s</i>)	25.6 (CH ₃)	C-10, C-11, C-12
14	6.73 (1H, <i>d</i> , 10.2)	115.6 (CH)	C-5, C-7, C-16
15	5.48 (1H, <i>d</i> , 10.2)	128.0 (CH)	C-6
16	-	77.9 (C)	-
17	1.46 (3H, <i>s</i>)	28.2 (CH ₃)	C-15, C-18
18	1.46 (3H, <i>s</i>)	28.2 (CH ₃)	C-15, C-17
1'	-	112.6 (C)	-
2'	-	155.2 (C)	-
3'	6.51 (1H, <i>s</i>)	103.8 (CH)	C-1', C-2', C-5'
4'	-	159.0 (C)	-
5'	6.52 (1H, <i>d</i> , 9.0)	108.4 (CH)	C-1', C-3', C-4'
6'	7.21 (1H, <i>d</i> , 9.0)	131.6 (CH)	C-2, C-2', C-4'
5-OH	13.18 (1H, <i>s</i>)	-	C-4a, C-5, C-6

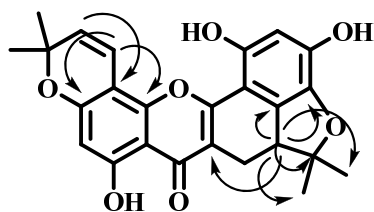
recorded in CDCl₃

PK8

5a,6-Dihydro-1,3,8-trihydroxy-5,5,11,11-tetramethyl-5*H*,7*H*,11*H*-benzofuro[3,4-*bc*]pyrano[3,2-*h*]xanthen-7-one : cycloartobiloxanthone



PK8 is a yellow solid, m.p. 284-285 °C (285-287 °C; Sultanbawa *et al.*, 1989), $[\alpha]_D^{28} = +5^\circ$ (*c* 0.2, CH₂Cl₂) ($[\alpha]_D^{20} = +80^\circ$ (*c* 0.2, CH₂Cl₂); Ren *et al.*, 2010). The UV spectrum showed maximum absorption bands at 228, 257, 272, 312, 330 and 391 nm. Its IR spectrum showed the stretching of hydroxyl (3405 cm⁻¹) and conjugated carbonyl group (1642 cm⁻¹). The ¹H NMR spectrum (**Table 13**) exhibited the signals of a chelated hydroxyl proton (5-OH) at δ_H 13.22 and non-chelated hydroxyl protons (4'-OH and 2'-OH) at δ_H 9.17 and δ_H 7.86. Two singlet signals at δ_H 6.25 and δ_H 6.38 were assigned for isolated aromatic protons H-6 and H-3'. The ¹H-NMR spectrum also indicated the resonances of two methyl groups at δ_H 1.35 and 1.67 (each 3H, *s*) and an ABX spin system at δ_H 2.41, 3.22, and 3.39, assignable to a furanodihydrobenzoxanthone skeleton (Hakim *et al.*, 2006) as those of **PK3**. The difference was the disappearance of a prenyl group signal but instead the resonances of aromatic protons (H-6, δ_H 6.25, *s*). Moreover, a 2,2-dimethylchromene ring was detected from the characteristic signals at δ_H 6.75 (*d*, H-14), 5.57 (*d*, H-15) and 1.42 (*s*, 17-CH₃ and 18-CH₃). The correlations of H-14 to C-7 and C-8a and of H-15 to C-8 confirmed the orientation of a chromene ring at C-7 and C-8 position. The assigned structure of **PK8** was in agreement with cycloartobiloxanthone (Sultanbawa *et al.*, 1989).

Major HMBC of **PK8****Table 13** ^1H , ^{13}C and HMBC spectral data of **PK8**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
2	-	160.9 (C)	-
3	-	112.0 (C)	-
4	-	181.0 (C=O)	-
4a	-	105.1 (C)	-
5	-	162.0 (C)	-
6	6.25 (1H, <i>s</i>)	100.4 (CH)	C-4a, C-5, C-7, C-8
7	-	159.1 (C)	-
8	-	101.3 (C)	-
8a	-	151.2 (C)	-
9	3.22 (1H $_{\beta}$, <i>dd</i> , 15.0, 7.2) 2.41 (1H $_{\alpha}$, <i>t</i> , 15.0)	20.2 (CH $_2$)	C-2, C-3, C-4, C-10, C-6' C-2, C-3, C-10, C-11, C-6'
10	3.39 (1H, <i>dd</i> , 15.0, 7.2)	46.9 (CH)	C-3, C-9, C-11, C-12, C-13, C-5', C-6'
11	-	93.9 (C)	-
12	1.35 (3H, <i>s</i>)	23.0 (CH $_3$)	C-10, C-11, C-13
13	1.67 (3H, <i>s</i>)	28.4 (CH $_3$)	C-10, C-11, C-12
14	6.75 (1H, <i>d</i> , 9.9)	115.3 (CH)	C-7, C-8, C-8a, C-16
15	5.57 (1H, <i>d</i> , 9.9)	127.7 (CH)	C-8, C-16, C-17, C-18
16	-	78.2 (C)	-
17	1.42 (3H, <i>s</i>)	28.5 (CH $_3$)	C-14, C-15, C-16, C-18
18	1.42 (3H, <i>s</i>)	28.5 (CH $_3$)	C-14, C-15, C-16, C-17
1'	-	103.7 (C)	-
2'	-	150.5 (C)	-
3'	6.38 (1H, <i>s</i>)	105.1 (CH)	C-1', C-2', C-4', C-5'

Table 13 (continued)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
4'	-	146.8 (C)	-
5'	-	137.6 (C)	-
6'	-	132.2 (C)	-
5-OH	13.22 (1H, <i>s</i>)	-	C-4, C-4a, C-5, C-6
2'-OH	7.86 (1H, <i>s</i>)	-	C-1', C-2', C-3'
4'-OH	9.17 (1H, <i>s</i>)	-	C-3', C-4', C-5'

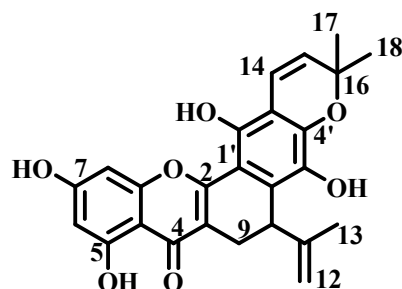
recorded in $\text{CDCl}_3 + \text{DMSO}-d_6$

Table 14 $^1\text{H}-^1\text{H}$ COSY spectral data of **PK8**

Proton (δ_{ppm})		Correlated proton (δ_{ppm})
$\text{H}_{\alpha}\text{-9}$ (2.41)	\longleftrightarrow	$\text{H}_{\beta}\text{-9}$ (3.22), H-10 (3.39)
$\text{H}_{\beta}\text{-9}$ (3.22)	\longleftrightarrow	$\text{H}_{\alpha}\text{-9}$ (2.41), H-10 (3.39)
H-10 (3.39)	\longleftrightarrow	$\text{H}_{\alpha}\text{-9}$ (2.41), $\text{H}_{\beta}\text{-9}$ (3.22)
H-14 (6.75)	\longleftrightarrow	H-15 (5.57)

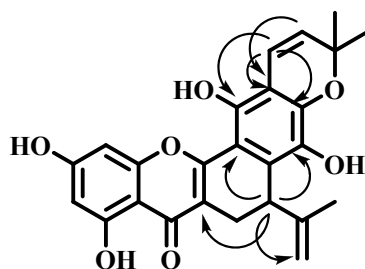
PK9

6,7-Dihydro-5,9,11,14-tetrahydroxy-3,3-dimethyl-6-(1-methylethenyl)-(-)-3*H*,8*H*-pyrano[3',2':4,5]benzo[1,2-*c*]xanthen-8-one : artelastoxanthone



PK9 is a red-brown gum, $[\alpha]_D^{27} = -82^\circ$ (c 0.2, acetone) ($[\alpha]_D^{28} = -67^\circ$ (c 0.2, acetone); Ko *et al.*, 2005). The UV spectrum showed maximum absorption bands at 263, 269, 307 and 379 nm. The IR spectrum showed the stretching of hydroxyl group at 3402 cm^{-1} and carbonyl group at 1655 cm^{-1} . The ^1H NMR spectral data (**Table 15**) showed the signals of a chelated hydroxyl proton 5-OH at δ_{H} 12.98, phenolic hydroxyl groups 2'-OH, 5'-OH at δ_{H} 7.78, 5.46 and *meta*-aromatic protons H-6, H-8 at δ_{H} 6.35, 6.40 with $J = 1.8$ Hz. The characteristic signals of 2, 2-dimethylchromene ring were shown at δ_{H} 5.64 (*d*, H-15), δ_{H} 6.74 (*d*, H-14), δ_{H} 1.49 (*s*, CH₃-18) and δ_{H} 1.52 (*s*, CH₃-17). It was placed at C-3' and C-4' position due to the HMBC correlation of H-14 to C-2', C-3', C-4' and of H-15 to C-3'. The ^1H NMR spectrum further showed an ABX system signal of non-equivalent methylene protons H _{α} -9 (δ_{H} 2.58, *dd*, $J = 16.2, 6.9$ Hz), H _{β} -9 (δ_{H} 3.39, *dd*, $J = 16.2, 1.5$ Hz) and a methine proton H-10 (δ_{H} 3.96, *d*, $J = 6.9$ Hz). The signal of non-equivalent vinylic protons (δ_{H} 4.34, *s*, H _{α} -12 and δ_{H} 4.71, *s*, H _{β} -12) and methyl proton (δ_{H} 1.81, *s*, H-13), corresponding to an isopropenyl group, were shown in the spectrum. The 3J HMBC correlations of H-10 to C-3, C-12, C-1', C-5' and C-6' suggested the point of attachment of C-10 to isopropenyl group and to C-6' of the aromatic ring. This evidence indicated that the cyclic was formed between C-3 and C-6' position, whereas the isoprenyl group was linked at C-10. The coupling constant value of 6.9 Hz suggested the *trans*-axial position of protons H _{α} -9 and H-10, consequently. These signals are the characteristic signals of a dihydrobenzoxanthone skeleton (Hakim *et*

al., 2006). The spectral data and assignments corresponded to the previously isolated, artelastoxanthone (Ko *et al.*, 2005).



Major HMBC of **PK9**

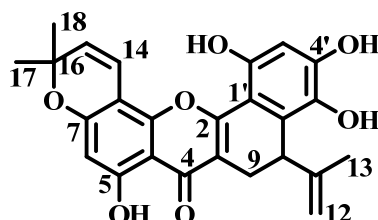
Table 15 ^1H , ^{13}C and HMBC spectral data of **PK9**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
2	-	159.7 (C)	-
3	-	111.5 (C)	-
4	-	180.0 (C=O)	-
4a	-	104.5 (C)	-
5	-	162.4 (C)	-
6	6.35 (1H, <i>d</i> , 1.8)	99.8 (CH)	C-4a, C-5, C-7, C-8
7	-	163.0 (C)	-
8	6.40 (1H, <i>d</i> , 1.8)	93.6 (CH)	C-4a, C-6, C-7, C-8a
8a	-	155.9 (C)	-
9	3.39 (1H $_{\beta}$, <i>dd</i> , 16.2, 1.5) 2.58 (1H $_{\alpha}$, <i>dd</i> , 16.2, 6.9)	21.5 (CH $_2$)	C-2, C-3, C-4, C-10, C-11, C-6' C-2, C-3, C-10, C-11
10	3.96 (1H, <i>d</i> , 6.9)	36.5 (CH)	C-3, C-9, C-11, C-12, C-13, C-1', C-5', C-6'
11	-	144.3 (C)	-
12	4.71 (1H $_{\beta}$, <i>s</i>) 4.34 (1H $_{\alpha}$, <i>s</i>)	111.7 (CH $_2$)	C-10, C-13 C-10, C-13
13	1.81 (3H, <i>s</i>)	21.6 (CH $_3$)	C-10, C-11, C-12
14	6.74 (1H, <i>d</i> , 10.0)	116.3 (CH)	C-2', C-3', C-4', C-16
15	5.64 (1H, <i>d</i> , 10.0)	128.5 (CH)	C-3', C-16, C-17, C-18
16	-	78.3 (C)	-

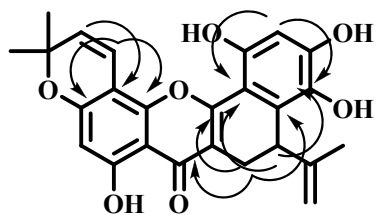
Table 15 (continued)

Position	δ_{H} (<i>mult, J_{Hz}</i>)	δ_{C} (C-Type)	HMBC
17	1.52 (3H, <i>s</i>)	28.2 (CH ₃)	C-15, C-16, C-18
18	1.49 (3H, <i>s</i>)	28.1 (CH ₃)	C-15, C-16, C-17
1'	-	105.2 (C)	-
2'	-	144.9 (C)	-
3'	-	108.8 (C)	-
4'	-	143.8 (C)	-
5'	-	135.6 (C)	-
6'	-	126.7 (C)	-
5-OH	12.98 (1H, <i>s</i>)	-	C-4, C-4a, C-5, C-6
2'-OH	7.78 (1H, <i>s</i>)	-	C-1', C-2', C-3'
5'-OH	5.46 (1H, <i>s</i>)	-	C-4', C-5', C-6'

recorded in CDCl₃+DMSO *d*₆

PK10**8,9-Dihydro-6,10,11,13-tetrahydroxy-3,3-dimethyl-9-(1-methylethenyl)-3H,7H-benzo[*c*]pyrano[3,2-*h*]xanthen-7-one : artobiloxanthone**

PK10 is a yellow solid, m.p. 163-164 °C (162-164 °C; Sultanbawa *et al.*, 1989), $[\alpha]_D^{26} = +43^\circ$ (*c* 0.2, CH₂Cl₂) ($[\alpha]_D^{20} = +60^\circ$ (*c* 0.2, CH₂Cl₂); Ren *et al.*, 2010). The UV spectrum showed maximum absorption bands at 227, 272 and 384 nm. The IR spectrum showed O-H stretching and C=O stretching at 3349 and 1652 cm⁻¹, respectively. The ¹H NMR spectrum had resonances associated with a chelated hydroxyl group at δ_H 13.01 (5-OH) and an ABX spin system at δ_H 2.59 (1H, *dd*, *J* = 16.2, 6.9 Hz, H_α-9), δ_H 3.36 (1H, *dd*, *J* = 16.2, 1.5 Hz, H_β-9) and δ_H 3.88 (1H, *dd*, *J* = 6.9, 1.5 Hz, H-10), attributed to the isoprenyl moiety located at the C-3 position, similar to the arrangement found for related compound, **PK9**. The ¹H NMR spectrum also indicated the presence of a 2,2-dimethylchromene ring by the resonance of two vinylic methine protons H-14 (δ_H 6.54, *d*, *J* = 10.2 Hz), H-15 (δ_H 5.64, *d*, *J* = 10.2 Hz), and two methyl groups 17-CH₃ (δ_H 1.46, *s*), 18-CH₃ (δ_H 1.48, *s*). This moiety was placed at C-7 and C-8 of the flavone skeleton due to the HMBC correlations of H-14 to C-7 (δ_C 159.2), C-8 (δ_C 104.8), and C-8a (δ_C 149.8) and of H-15 to C-8 (δ_C 104.8). Two isolated aromatic protons were indicated from the resonances at δ_H 6.29 (H-6, *s*) and δ_H 6.51 (H-3', *s*). The HMBC experiments (**Table 16**) showed long-range correlations between the singlet at δ_H 6.29 (H-6) and the quaternary carbon signals at δ_C 112.5 (C-4a) and δ_C 104.8 (C-8), locating of H-6 on the A-ring. Another aromatic proton (δ_H 6.51) was assigned in according with 1,2,4,5,6-pentasubstituted B-ring based on the biogenetic pattern of constituents in *Artocarpus* genus (Hakim *et al.*, 2006). Consequently, artobiloxanthone was assigned the structure **PK10** (Jayasinghe *et al.*, 2008).



Major HMBC of PK10

Table 16 ^1H , ^{13}C and HMBC spectral data of PK10

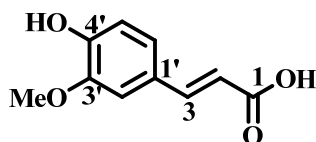
Position	δ_{C} (C-Type)	δ_{H} (mult, J _{HZ})	HMBC
2	159.7 (C)	-	-
3	110.9 (C)	-	-
4	180.1 (C=O)	-	-
4a	112.5 (C)	-	-
5	161.7 (C)	-	-
6	100.7 (CH)	6.29 (1H, s)	C-4a, C-5, C-7, C-8
7	159.2 (C)	-	-
8	104.8 (C)	-	-
8a	149.8 (C)	-	-
9	21.8 (CH ₂)	3.36 (1H _β , dd, 16.2, 1.5) 2.59 (1H _α , dd, 16.2, 6.9)	C-2, C-3, C-4, C-10, C-11, C-6' C-2, C-3, C-4, C-10, C-11, C-6'
10	37.9 (CH)	3.88 (1H, dd, 6.9, 1.5)	C-9, C-11, C-12, C-13, C-1', C-5', C-6'
11	144.7 (C)	-	-
12	112.6 (CH ₂)	4.78 (1H _β , br s) 4.46 (1H _α , br s)	C-10, C-13 C-10, C-11, C-13
13	21.0 (CH ₃)	1.79 (3H, s)	C-10, C-11, C-12
14	114.0 (CH)	6.54 (1H, d, 10.2)	C-7, C-8a, C-15, C-16
15	128.6 (CH)	5.64 (1H, d, 10.2)	C-8, C-16, C-17, C-18
16	78.3 (C)	-	-
17	27.9 (CH ₃)	1.46 (3H, s)	C-15, C-16, C-18
18	28.1 (CH ₃)	1.48 (3H, s)	C-15, C-16, C-17
1'	105.2 (C)	-	-

Table 16 (continued)

Position	δ_C (C-Type)	δ_H (<i>mult</i> , J_{Hz})	HMBC
2'	150.8 (C)	-	-
3'	103.0 (CH)	6.51 (1H, <i>s</i>)	C-1', C-2', C-4', C-5'
4'	150.4 (C)	-	-
5'	135.0 (C)	-	-
6'	127.7 (C)	-	-
5-OH	-	13.01 (1H, <i>br</i>)	-
*OH	-	7.50 (1H, <i>br</i>)	-

recorded in CDCl₃

*The position not identified

PK11**(E)-3-(4'-Hydroxy-3'-methoxyphenyl)-2-propenoic acid**

PK11 was obtained as a brown-yellow gum. The UV spectrum showed maximum absorption bands at 244, 273, 299 and 385 nm. The IR spectrum showed the O-H, C=O and C=C stretching at 3350, 1712 and 1513 cm^{-1} , respectively. The ^1H NMR spectral data (**Table 17**) displayed an ABX signal of aromatic protons at δ_{H} 6.94 (*d*, $J = 8.4$ Hz, H-5'), δ_{H} 7.05 (*d*, $J = 1.8$ Hz, H-2') and δ_{H} 7.13 (*dd*, $J = 8.4, 1.8$ Hz, H-6'). The spectrum further showed the resonance of vinylic proton H-3 at δ_{H} 7.59 and H-2 at δ_{H} 6.48. Their large coupling constant ($J = 15.9$ Hz) indicated *trans* configuration. In addition, the spectrum also showed the signal of a methoxyl group at δ_{H} 3.95. The high field signal of *ortho*-oxygenated aromatic carbons had resonated at δ_{C} 146.8 and 147.9 due to a mesomeric effect. The HMBC correlations of H-3 to C-1' (δ_{C} 127.8), C-2' (δ_{C} 109.7) and C-6' (δ_{C} 122.8) and of H-2 to C-1' (δ_{C} 127.8) correctly determined that the side chain was connected at C-1'. The correlations of H-6' to an oxygenated aromatic carbons (δ_{C} 147.9) whereas of a methoxyl group to another one, confirming the methoxyl group and the hydroxyl group were at C-3' and C-4', respectively. The structure of **PK11** was identified as (*E*)-3-(4'-hydroxy-3'-methoxyphenyl)-2-propenoic acid. It was corresponded to *trans*-feluric acid (Kelley *et al.*, 1976).

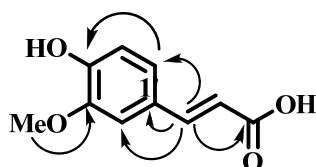
Major HMBC of **PK11**

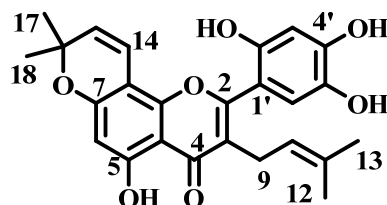
Table 17 ^1H , ^{13}C and HMBC spectral data of **PK11**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1	-	183.4 (C=O)	-
2	6.48 (1H, <i>d</i> , 15.9)	121.8 (CH)	C-3, C-1'
3	7.59 (1H, <i>d</i> , 15.9)	140.5 (CH)	C-1, C-2, C-1', C-2', C-6'
1'	-	127.8 (C)	-
2'	7.05 (1H, <i>d</i> , 1.8)	109.7 (CH)	C-3, C-3', C-4', C-6'
3'	-	146.8 (C)	-
4'	-	147.9 (C)	-
5'	6.94 (1H, <i>d</i> , 8.4)	114.8 (CH)	C-1', C-3', C-4'
6'	7.13 (1H, <i>dd</i> , 8.4, 1.8)	122.8 (CH)	C-3, C-2', C-5'
3'-OCH ₃	3.95 (3H, <i>s</i>)	56.0 (OCH ₃)	C-3'
4-OH	8.09 (1H, <i>s</i>)	-	-

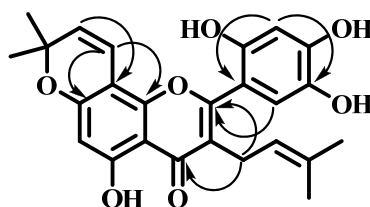
recorded in CDCl₃

PK12

**5-Hydroxy-8,8-dimethyl-3-(3-methyl-2-butenyl)-2-(2,4,5-trihydroxyphenyl)-4*H*,
8*H*-benzo[1,2-*b*:3,4-*b'*]dipyran-4-one : artonin E**



PK12 is a brown-yellow solid, m.p. 217-219 °C. The UV spectrum showed maximum absorption bands at 224, 258, 266, 271, 302 and 352 nm. The IR spectrum exhibited the absorption bands of hydroxyl group (3402 cm^{-1}) and carbonyl group (1655 cm^{-1}). The ^1H NMR spectral data (**Table 18**) revealed the presence of a hydrogen bonded hydroxyl group ($\delta_{\text{H}} 13.21, s$), three non-bonded hydroxyl groups ($\delta_{\text{H}} 8.56, 8.38, \text{ and } 7.54$), and three isolated aromatic protons ($\delta_{\text{H}} 6.19, s, \text{H-6}$; $\delta_{\text{H}} 6.58, s, \text{H-3'}$ and $\delta_{\text{H}} 6.79, s, \text{H-6'}$). The presence of a prenyl group was observed from characteristic signals of methylene protons (H-9, $\delta_{\text{H}} 3.14, d$), an olefinic methine proton (H-10, $\delta_{\text{H}} 5.12, mt$) and methyl protons ($\text{CH}_3\text{-12}, \delta_{\text{H}} 1.47, s$ and $\text{CH}_3\text{-13}, \delta_{\text{H}} 1.61, s$). The correlation of H-9 to carbonyl group ($\delta_{\text{C}} 182.5$) indicated that a prenyl side chain connected to C-3 position. The characteristic signals of a 2,2-dimethylchromene ring were shown at $\delta_{\text{H}} 5.48 (d, \text{H-15})$, $\delta_{\text{H}} 6.62 (d, \text{H-14})$, and $\delta_{\text{H}} 1.44 (s, \text{CH}_3\text{-17 and } s, \text{CH}_3\text{-18})$. It was placed at C-7 and C-8 position due to the HMBC correlation of H-14 to C-7, C-8, C-8a and of H-15 to C-8. Thus **PK12** was assigned as artonin E (Jayasinghe *et al.*, 2008).



Major HMBC of **PK12**

Table 18 ^1H , ^{13}C and HMBC spectral data of **PK12**

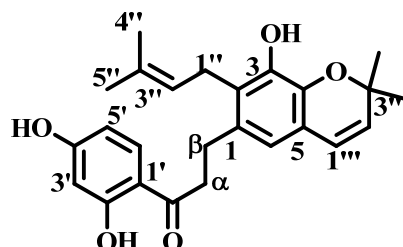
Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
2	-	161.2 (C)	-
3	-	120.8 (C)	-
4	-	182.5 (C=O)	-
4a	-	105.0 (C)	-
5	-	161.5 (C)	-
6	6.19 (1H, <i>s</i>)	99.2 (CH)	C-4a, C-5, C-7, C-8
7	-	158.8 (C)	-
8	-	100.8 (C)	-
8a	-	152.4 (C)	-
9	3.14 (2H, <i>d</i> , 6.6)	24.2 (CH ₂)	C-2, C-3, C-4, C-10, C-11
10	5.12 (1H, <i>mt</i> , 6.6)	121.5 (CH)	-
11	-	132.0 (C)	-
12	1.47 (3H, <i>s</i>)	17.5 (CH ₃)	C-10, C-11, C-13
13	1.61 (3H, <i>s</i>)	25.7 (CH ₃)	C-10, C-11, C-12
14	6.62 (1H, <i>d</i> , 9.9)	115.2 (CH)	C-7, C-8, C-8a, C-16
15	5.48 (1H, <i>d</i> , 9.9)	126.5 (CH)	C-8, C-16, C-17, C-18
16	-	77.7 (C)	-
17	1.44 (3H, <i>s</i>)	28.0 (CH ₃)	C-15, C-16, C-18
18	1.44 (3H, <i>s</i>)	28.0 (CH ₃)	C-15, C-16, C-17
1'	-	110.7 (C)	-
2'	-	148.8 (C)	-
3'	6.58 (1H, <i>s</i>)	104.0 (CH)	C-1', C-2', C-5'
4'	-	147.9 (C)	-
5'	-	137.6 (C)	-
6'	6.79 (1H, <i>s</i>)	116.2 (CH)	C-2, C-4', C-5'

Table 18 (continued)

Position	δ_{H} (mult, J_{Hz})	δ_{C} (C-Type)	HMBC
5-OH	13.21 (1H, <i>s</i>)	-	C-4a, C-5, C-6
*OH	8.56 (1H, <i>s</i>)	-	-
*OH	8.38 (1H, <i>s</i>)	-	-
*OH	7.54 (1H, <i>s</i>)	-	-

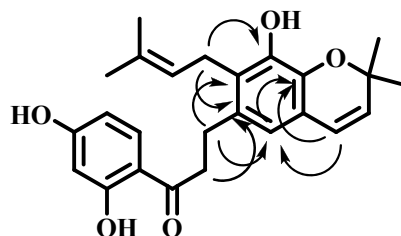
recorded in $\text{CDCl}_3 + \text{DMSO-}d_6$

* The position not identified

PK13**1-(2,4-Dihydroxyphenyl)-3-(8-hydroxy-2,2-dimethyl-7-(3-methylbut-2-enyl)-2H-chromen-6-yl)propan-1-one**

PK13 was obtained as a green-yellow gum. The UV spectrum showed maximum absorption bands at 231, 273 and 313 nm. The IR spectrum exhibited absorption bands at 3383 and 1634 cm^{-1} for a hydroxyl group and a carbonyl group. The ^1H NMR spectral data (**Table 19**) showed singlet resonance of a hydroxyl proton (2'-OH) at δ_{H} 12.83, isolated aromatic proton (H-6) at δ_{H} 6.42 and aromatic protons attributed to 1,2,4-trisubstituted benzene at δ_{H} 7.58 (*d*, $J = 8.7$ Hz, H-6'), 6.35 (*dd*, $J = 8.7, 2.1$ Hz, H-5') and 6.38 (*d*, $J = 2.1$ Hz, H-3'). The aromatic proton (H-6') resonated at low field because of mesomeric effect of carbonyl group in *ortho* position. The spectrum further showed signals of two methylene groups which were coupled to each other ($J = 9.6, 6.9$ Hz) at δ_{H} 3.11 (α -CH₂) and 2.93 (β -CH₂). The HMBC correlations of α -CH₂ to C-1 (δ_{C} 131.5) while β -CH₂ to C-2 (δ_{C} 126.4), C-6 (δ_{C} 117.7) suggested that they were the α - and β - methylene proton of dihydrochalcone skeleton (Wang, *et al.*, 2007). The characteristic signals of a prenyl side chain [δ_{H} 3.37 (*d*, $J = 6.6$ Hz, H-1''), 5.14 (*mt*, $J = 6.6$ Hz, H-2''), 1.72 (*s*, H-4''), 1.66 (*s*, H-5'')] and of a 2,2-dimethylchromene ring [δ_{H} 6.25 (*d*, $J = 8.7$ Hz, H-1'''), 5.56 (*d*, $J = 8.7$ Hz, H-2'''), 1.44 (*s*, H-4''' and H-5''')] were displayed in the spectrum. The prenyl group was placed at C-2 according to the correlation of H-2'' to C-2 (δ_{C} 126.4) and of H-1'' to C-1 (δ_{C} 131.5), C-3 (δ_{C} 142.4). The correlations of H-1''' to C-4 (δ_{C} 137.5), C-6 (δ_{C} 117.7) and of H-6 to C-4 (δ_{C} 137.5), C-1''' (δ_{C} 121.9) correctly determined that the chromene ring was at C-4 and C-5 position. The ^{13}C NMR spectrum showed 25 carbon signals separated by DEPT experiment into 11 quaternary, 7 methine, 3 methylene and 4 methyl carbons. The proposed structure of **PK13** was in agreement with molecular ion of m/z 408.1932 (C₂₅H₂₈O₅). Consequently, a new

dihydrochalcone derivative, 1-(2,4-dihydroxyphenyl)-3-(8-hydroxy-2,2-dimethyl-7-(3-methylbut-2-enyl)-2H-chromen-6-yl)propan-1-one, was assigned for **PK13**.



Major HMBC of **PK13**

Table 19 ^1H , ^{13}C and HMBC spectral data of **PK13**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1	-	131.5 (C)	-
2	-	126.4 (C)	-
3	-	142.4 (C)	-
4	-	137.5 (C)	-
5	-	118.7 (C)	-
6	6.42 (1H, <i>s</i>)	117.7 (CH)	C_{β} , C-2, C-4, C-5, C-1'''
C=O	-	204.0 (C=O)	-
α	3.11 (2H, <i>dd</i> , 9.6, 6.9)	39.5 (CH ₂)	C_{β} , C-1
β	2.93 (2H, <i>dd</i> , 9.6, 6.9)	27.2 (CH ₂)	C_{α} , C-1, C-2, C-6
1'	-	113.5 (C)	-
2'	-	163.1 (C)	-
3'	6.38 (1H, <i>d</i> , 2.1)	103.4 (CH)	C-1', C-2', C-4', C-5'
4'	-	165.1 (C)	-
5'	6.35 (1H, <i>dd</i> , 8.7, 2.1)	107.8 (CH)	C-1', C-3'
6'	7.58 (1H, <i>d</i> , 8.7)	132.1 (CH)	C-1', C-2', C-4'
1''	3.37 (2H, <i>d</i> , 6.6)	25.3 (CH ₂)	C-1, C-2, C-3, C-2'', C-3''
2''	5.14 (1H, <i>mt</i> , 6.6)	122.8 (CH)	C-2, C-1'', C-4'', C-5''
3''	-	131.8 (C)	-
4''	1.72 (3H, <i>s</i>)	17.8 (CH ₃)	C-2'', C-3'', C-5''
5''	1.66 (3H, <i>s</i>)	25.6 (CH ₃)	C-2'', C-3''

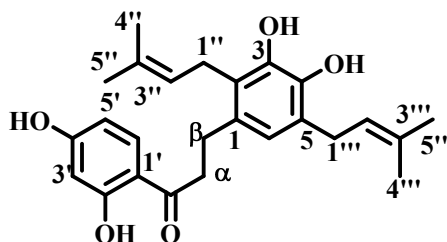
Table 19 (continued)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1'''	6.25 (1H, <i>d</i> , 8.7)	121.9 (CH)	C-4, C-5, C-6, C-3'''
2'''	5.56 (1H, <i>d</i> , 8.7)	129.9 (CH)	C-5, C-4''', C-5'''
3'''	-	77.1 (C)	-
4'''	1.44 (3H, <i>s</i>)	28.0 (CH ₃)	C-2''', C-3'''
5'''	1.44 (3H, <i>s</i>)	28.0 (CH ₃)	C-2''', C-3'''
2'-OH	12.83 (1H, <i>s</i>)	-	-

recorded in CDCl₃

Table 20 ¹H-¹H COSY spectral data of **PK13**

Proton (δ_{ppm})		Correlated proton (δ_{ppm})
H- α (3.11)	\longleftrightarrow	H- β (2.93)
H-3' (6.38)	\longleftrightarrow	H-5' (6.35)
H-5' (6.35)	\longleftrightarrow	H-3' (6.38), H-6' (7.58)
H-6' (7.58)	\longleftrightarrow	H-5' (6.35)
H-1'' (3.37)	\longleftrightarrow	H-2'' (5.14)
H-1''' (6.25)	\longleftrightarrow	H-2''' (5.56)

PK14**1-(2,4-dihydroxyphenyl)-3-(3,4-dihydroxy-2,5-bis(3-methylbut-2-enyl)phenyl)propan-1-one**

PK14 was obtained as a yellow solid, m.p. 170 C°. The UV spectrum showed maximum absorption bands at 237, 274 and 313 nm. The IR spectrum showed absorption band of a hydroxyl group at 3422 cm⁻¹ and a carbonyl group at 1629 cm⁻¹. Its ¹H NMR spectral data (**Table 21**) showed signals corresponded to the α - and β -methylene protons (δ_{H} 3.29, *dd*, $J = 8.1, 7.2$ Hz; δ_{H} 2.91 (*dd*, $J = 8.1, 7.2$ Hz), a 1,2,4-trisubstituted benzene (δ_{H} 7.54, *d*, $J = 9.3$ Hz, H-6'; δ_{H} 6.36, *d*, $J = 9.3$ Hz, H-5'; δ_{H} 6.38, *s*, H-3'), an aromatic proton (δ_{H} 6.49, H-6) on ring B, a hydrogen bonded hydroxyl group (δ_{H} 12.84), two prenyl groups (δ_{H} 3.36, *d*, $J = 7.2$ Hz, H-1''; δ_{H} 5.14, *mt*, $J = 7.2$ Hz, H-2''; δ_{H} 1.65, *s*, H-4''; δ_{H} 1.71, *s*, H-5'' and δ_{H} 3.28, *d*, $J = 7.2$ Hz, H-1'''; δ_{H} 5.28, *mt*, $J = 7.2$ Hz, H-2'''; δ_{H} 1.70, *s*, H-4'''; δ_{H} 1.71, *s*, H-5'''). The ¹³C NMR spectral data and HMBC correlations suggested that it was a dihydrochalcone with a prenylated side chain at C-2 as for **PK13**. The second prenyl group was placed at C-5 position according to the HMBC correlations of H-1''' to C-4 (δ_{C} 140.2), C-6 (δ_{C} 120.6) and of olefinic proton (H-2''') to C-5 (δ_{C} 125.9) indicated that this side chain was at C-5 position. The molecular ion of m/z 410.2087 (C₂₅H₃₀O₅) was in agreement with the proposed structure. Therefore a new dihydrochalcone structure, 1-(2,4-dihydroxyphenyl)-3-(3,4-dihydroxy-2,5-bis(3-methylbut-2-enyl)phenyl)propan-1-one, was assigned for **PK14**.

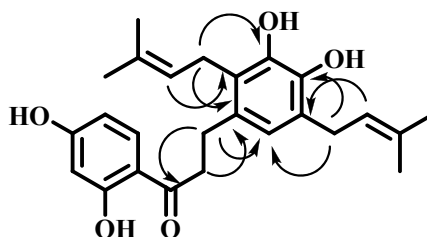
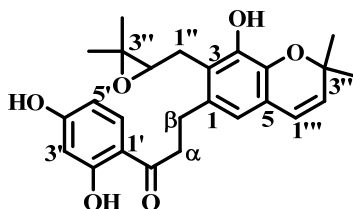
Major HMBC of **PK14**

Table 21 ^1H , ^{13}C and HMBC spectral data of **PK14**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1	-	130.4 (C)	-
2	-	124.2 (C)	-
3	-	142.6 (C)	-
4	-	140.2 (C)	-
5	-	125.9 (C)	-
6	6.49 (1H, <i>s</i>)	120.6 (CH)	C-2, C-4
C=O	-	203.6 (C=O)	-
α	3.29 (2H, <i>dd</i> , 8.1, 7.2)	38.8 (CH ₂)	C=O, C $_{\beta}$, C-1
β	2.91 (2H, <i>dd</i> , 8.1, 7.2)	27.1 (CH ₂)	C=O, C $_{\alpha}$, C-2, C-6
1'	-	112.4 (C)	-
2'	-	164.2 (C)	-
3'	6.38 (1H, <i>s</i>)	102.8 (CH)	C-1', C-4', C-5'
4'	-	164.7 (C)	-
5'	6.36 (1H, <i>d</i> , 9.3)	107.8 (CH)	C-1', C-3'
6'	7.54 (1H, <i>d</i> , 9.3)	131.6 (CH)	C-1', C-4', C=O
1''	3.36 (2H, <i>d</i> , 7.2)	25.0 (CH ₂)	C-1, C-2, C-3, C-2'', C-3''
2''	5.14 (1H, <i>mt</i> , 7.2)	123.0 (CH)	C-2, C-1'', C-4'', C-5'
3''	-	131.0 (C)	-
4''	1.65 (3H, <i>s</i>)	17.3 (CH ₃)	C-2'', C-3''
5''	1.71 (3H, <i>s</i>)	25.2 (CH ₃)	C-2'', C-3''
1'''	3.28 (2H, <i>d</i> , 7.2)	28.1 (CH ₂)	C-4, C-5, C-6, C-2''', C-3'''
2'''	5.28 (1H, <i>mt</i> , 7.2)	122.1 (CH)	C-5, C-1''', C-4''', C-5'''
3'''	-	132.0 (C)	-
4'''	1.71 (3H, <i>s</i>)	17.4 (CH ₃)	C-2''', C-3'''
5'''	1.70 (3H, <i>s</i>)	25.3 (CH ₃)	C-2''', C-3'''
3-OH	6.86 (1H, <i>s</i>)	-	C-2, C-3, C-4
4-OH	6.98 (1H, <i>s</i>)	-	C-3, C-4, C-5
2'-OH	12.84 (1H, <i>s</i>)	-	C-1', C-2', C-3'
4'-OH	9.86 (1H, <i>s</i>)	-	C-3', C-4', C-5'

recorded in CDCl₃+DMSO-*d*₆

PK15**1-(2,4-Dihydroxyphenyl)-3-(7-((3,3-dimethyloxiran-2-yl)methyl)-8-hydroxy-2,2-dimethyl-2*H*-chromen-6-yl)propan-1-one**

PK15 was obtained as a yellow gum, $[\alpha]_D^{26} = -14^\circ$ (c 0.1, acetone). The UV spectrum showed maximum absorption bands at 245, 274 and 311 nm. The IR spectrum showed absorption band of a hydroxyl group at 3390 cm^{-1} and a carbonyl group at 1631 cm^{-1} . The ^1H NMR spectral data (**Table 22**) disclosed the signals of a 1,2,4-trisubstituted benzene (δ_{H} 7.58, d , $J = 8.4$ Hz, H-6'; 6.35, dd , $J = 8.4, 1.8$ Hz, H-5'; 6.37, d , $J = 1.8$ Hz, H-3'), an aromatic proton (δ_{H} 6.45, H-6) on ring B, α - and β -methylene protons (δ_{H} 3.13, dd , $J = 8.4, 6.6$ Hz; 2.87, dd , $J = 8.4, 6.6$ Hz), hydrogen bonded hydroxyl group (δ_{H} 12.83, s , 2'-OH) and a chromene ring at C-4/C-5 position (δ_{H} 6.27; d , $J = 9.6$ Hz, H-1'''; 5.57, d , $J = 9.6$ Hz, H-2'''; 1.43, s , H-4'''; 1.44, s , H-5''') as for **PK13**. The replacement of the prenyl side chain at C-3 by 3,3-dimethyloxiran-2-yl-methyl group was indicated by the resonances of non-equivalent methylene protons (δ_{H} 2.96, dd , $J = 11.7, 5.4$ Hz; 2.70, dd , $J = 11.7, 5.4$ Hz), an oxy-methine proton (δ_{H} 3.82, t , $J = 5.4$ Hz) and two methyl groups (δ_{H} 1.37, s ; δ_{H} 1.32, s). The appearing of non-equivalent methylene proton H-1'' suggested that it connected to a chiral carbon C-2''. In ^{13}C NMR spectrum, oxy-methine carbon (C-2'') and oxy-quarternary carbon (C-3'') resonated at δ_{C} 69.8 at δ_{C} 76.4, respectively. A molecular ion in the HREI-MS at m/z 424.1880 which corresponded to a molecular formula of $\text{C}_{25}\text{H}_{28}\text{O}_6$ confirmed that **PK15** was 1-(2,4-dihydroxyphenyl)-3-(7-((3,3-dimethyloxiran-2-yl)methyl)-8-hydroxy-2,2-dimethyl-2*H*-chromen-6-yl)propan-1-one, a new dihydrochalcone.

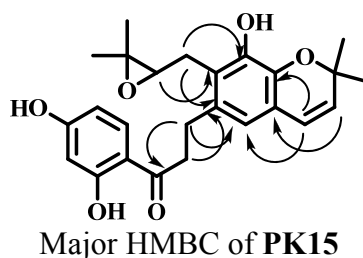
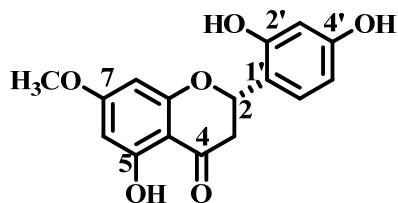


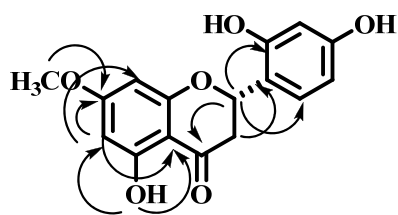
Table 22 ^1H , ^{13}C and HMBC spectral data of **PK15**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1	-	131.9 (C)	-
2	-	118.7 (C)	-
3	-	141.7 (C)	-
4	-	140.0 (C)	-
5	-	120.4 (C)	-
6	6.45 (1H, <i>s</i>)	118.1 (CH)	C $_{\beta}$, C-2, C-4, C-1'''
C=O	-	203.8 (C=O)	-
α	3.13 (2H, <i>dd</i> , 8.4, 6.6)	38.4 (CH ₂)	C=O, C $_{\beta}$, C-1
β	2.87 (2H, <i>dd</i> , 8.4, 6.6)	26.6 (CH ₂)	C=O, C $_{\alpha}$, C-6
1'	-	113.7 (C)	-
2'	-	165.2 (C)	-
3'	6.37 (1H, <i>d</i> , 1.8)	103.6 (CH)	C-1', C-2', C-5'
4'	-	163.0 (C)	-
5'	6.35 (1H, <i>dd</i> , 8.4, 1.8)	107.9 (CH)	C-1', C-3', C-4'
6'	7.58 (1H, <i>d</i> , 8.4)	132.2 (CH)	C=O, C-2', C-4'
1''	2.96 (1H, <i>dd</i> , 11.7, 5.4)	29.6 (CH ₂)	C-1, C-2, C-3, C-2'', C-3''
	2.70 (1H, <i>dd</i> , 11.7, 5.4)		C-1, C-2, C-3, C-2'', C-3''
2''	3.82 (1H, <i>t</i> , 5.4)	69.8 (CH)	C-2, C-4'', C-5''
3''	-	76.4 (C)	-
4''	1.37 (3H, <i>s</i>)	21.8 (CH ₃)	C-2'', C-3'', C-5''
5''	1.32 (3H, <i>s</i>)	24.4 (CH ₃)	C-2'', C-3'', C-4''
1'''	6.27 (1H, <i>d</i> , 9.6)	122.2 (CH)	C-4, C-5, C-6, C-3'''
2'''	5.57 (1H, <i>d</i> , 9.6)	130.9 (CH)	C-5, C-1''', C-3''', C-4''', C-5'''
3'''	-	76.3 (C)	-
4'''	1.43 (3H, <i>s</i>)	27.7 (CH ₃)	C-2''', C-3''', C-5'''
5'''	1.44 (3H, <i>s</i>)	27.5 (CH ₃)	C-2''', C-3''', C-4'''
2'-OH	12.83 (1H, <i>s</i>)	-	C-1', C-2', C-3'

recorded in CDCl₃

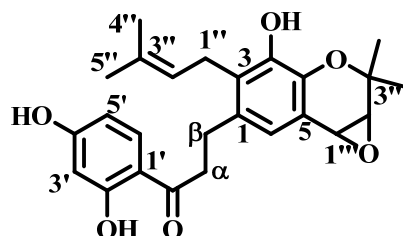
PK16**(S)-2-(2,4-Dihydroxyphenyl)-5-hydroxy-7-methoxychroman-4-one**

PK16 is a pale yellow solid, m.p. 210-211 °C (210-212 °C); Wei *et al.*, 2005), $[\alpha]_D^{27} = -3^\circ$ (*c* 0.2, acetone) ($[\alpha]_D^{24} = -2^\circ$ (*c* 0.2, acetone); Wei *et al.*, 2005). The UV spectrum showed maximum absorption bands at 244, 273 and 305 nm. Its IR spectrum showed the stretching of hydroxyl (3343 cm^{-1}) and carbonyl group (1697 cm^{-1}). The ^{13}C NMR revealed the presence of 16 carbons separated by DEPT experiment into 8 quaternary, 6 methine, 1 methylene and 1 methyl carbons. The ^1H NMR spectral data (**Table 23**) displayed an ABX signal of aromatic protons at δ_{H} 7.32 (*d*, $J = 8.1$ Hz, H-6'), δ_{H} 6.48 (*d*, $J = 2.4$ Hz, H-3') and δ_{H} 6.44 (*dd*, $J = 8.1, 2.4$ Hz, H-5'), and *meta*-aromatic protons at δ_{H} 6.03 (H-6) and δ_{H} 6.05 (H-8) with $J = 2.4$ Hz. The spectrum further showed the doublet of doublet resonance of an oxy-methine proton at δ_{H} 5.73 ($J = 13.2, 3.0$, H-2) and the doublet of doublet resonance of the adjacent non-equivalent methylene protons (H-3) at δ_{H} 3.21 ($J = 17.1, 13.2$ Hz; H_α) and δ_{H} 2.74 ($J = 17.1, 3.0$ Hz, H_β). The signals at δ 3.21 and 2.74 were ascribed to the *trans*- and *cis*- orientation with H-2, respectively. The spectrum also exhibited the resonances of methoxyl group at δ_{H} 3.84. It was placed at C-7 by HMBC correlations of the methoxyl group (δ_{H} 3.84), H-6 (δ_{H} 6.03) and H-8 (δ_{H} 6.05) to C-7 (δ_{C} 167.8). The correlations of oxy-methine proton (H-2) to C-2' (δ_{C} 155.5) and C-6' (δ_{C} 128.1) confirmed the linkage of C-2 and C-1'. This compound was flavanones, named (*S*)-2-(2,4-dihydroxyphenyl)-5-hydroxy-7-methoxychroman-4-one. These assignment indicated that **PK16** was artocarpanone (Wei *et al.*, 2005).

Major HMBC of **PK16****Table 23** ^1H , ^{13}C and HMBC spectral data of **PK16**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
2	5.73 (1H, <i>dd</i> , 13.2, 3.0)	74.6 (CH)	C-4, C-1', C-2', C-6'
3	3.21 (1H $_{\alpha}$, <i>dd</i> , 17.1, 13.2) 2.74 (1H $_{\beta}$, <i>dd</i> , 17.1, 3.0)	41.7 (CH ₂)	C-2, C-4, C-1' C-4, C-4a, C-1'
4	-	197.2 (C=O)	-
4a	-	102.8 (C)	-
5	-	164.1 (C)	-
6	6.03 (1H, <i>d</i> , 2.4)	94.5 (CH)	C-4a, C-5, C-7, C-8
7	-	167.8 (C)	-
8	6.05 (1H, <i>d</i> , 2.4)	93.6 (CH)	C-4a, C-6, C-7, C-8a
8a	-	163.8 (C)	-
1'	-	116.4 (C)	-
2'	-	155.5 (C)	-
3'	6.48 (1H, <i>d</i> , 2.4)	102.6 (CH)	C-1', C-2', C-4', C-5'
4'	-	158.7 (C)	-
5'	6.44 (1H, <i>dd</i> , 8.1, 2.4)	107.2 (CH)	C-1', C-3', C-4'
6'	7.32 (1H, <i>d</i> , 8.1)	128.1 (CH)	C-2, C-2', C-4'
5-OH	12.17 (1H, <i>s</i>)	-	C-4a, C-5, C-6
7-OMe	3.84 (3H, <i>s</i>)	55.3 (CH ₃)	C-7

recorded in acetone- d_6

PK17**1-(2,4-Dihydroxyphenyl)-3-(4-hydroxy-2,2-dimethyl-5-(3-methylbut-2-enyl)-2,7b-dihydro-1aH-oxireno[2,3-c]chromen-6-yl)propan-1-one**

PK17 was obtained as a yellow solid, m.p. 179-180 C°, $[\alpha]_D^{26} = +7^\circ$ (*c* 0.2, acetone). The UV spectrum showed maximum absorption bands at 243, 275 and 310 nm. The IR spectrum exhibited the presence of hydroxyl (3223 cm^{-1}) and carbonyl (1624 cm^{-1}) groups. The ^1H NMR spectral data and HMBC correlation (**Table 24**) were much closely to those of **PK13** with the replacement of olefinic proton signals at δ_{H} 6.25 (*d*, $J = 8.7$ Hz, H-1''') and δ_{H} 5.56 (*d*, $J = 8.7$ Hz, H-2''') by signals of oxy-methine proton at δ_{H} 4.47 and 3.54 (2 x *br d*, $J = 7.8$ Hz, H-1''' and H-2'''). The signals of dihydrochalcone skeleton disclosed the signals of α - and β -methylene proton (δ_{H} 3.15-3.21, *m*; 2.92-2.95, *m*), a 1,2,4-trisubstituted benzene (δ_{H} 7.77 (*d*, $J = 8.7$ Hz, H-6'); 6.42 (*d*, $J = 8.7, 2.4$ Hz, H-5'); 6.33 (*d*, $J = 2.4$ Hz, H-3')), an aromatic proton (δ_{H} 6.86, H-6) on ring B, a hydrogen bonded hydroxyl group (δ_{H} 12.83), a prenyl group (δ_{H} 3.39 (*d*, $J = 6.6$ Hz, H-1''); 5.14 (*mt*, $J = 6.6$ Hz, H-2''); 1.71 (*s*, H-4''); 1.63 (*s*, H-5'')). The HMBC correlation of H-1''' to C-3''' (δ_{C} 78.9) and of H-2''' to C-1''' (δ_{C} 69.0) confirm that the epoxy group was at C-1''' and C-2'''. A molecular ion in the HREI-MS at m/z 424.1865, corresponding to a molecular formula of $\text{C}_{25}\text{H}_{28}\text{O}_6$, so this compound have been epoxide structure. Thus a new dihydrochalcone structure, 1-(2,4-dihydroxyphenyl)-3-(4-hydroxy-2,2-dimethyl-5-(3-methylbut-2-enyl)-2,7b-dihydro-1aH-oxireno[2,3-c] chromen-6-yl) propan-1-one, was assigned for **PK17**.

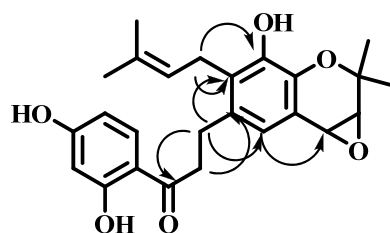
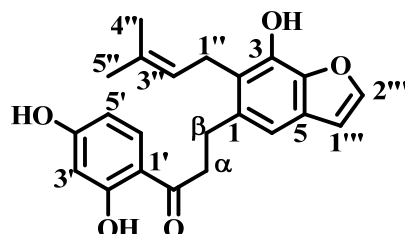
Major HMBC of **PK17**

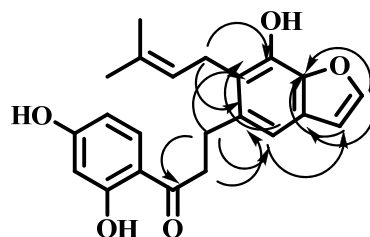
Table 24 ^1H , ^{13}C and HMBC spectral data of **PK17**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1	-	131.4 (C)	-
2	-	124.8 (C)	-
3	-	142.9 (C)	-
4	-	138.1 (C)	-
5	-	123.8 (C)	-
6	6.86 (1H, <i>s</i>)	118.1 (CH)	C_{β} , C-2, C-4, C-5, C-1'''
C=O	-	204.3 (C=O)	-
α	3.15-3.21 (2H, <i>m</i>)	39.3 (CH_2)	C=O, C_{β} , C-1
β	2.92-2.95 (2H, <i>m</i>)	27.3 (CH_2)	C=O, C_{α} , C-1, C-2, C-6
1'	-	113.0 (C)	-
2'	-	165.4 (C)	-
3'	6.33 (1H, <i>d</i> , 2.4)	102.7 (CH)	C-1', C-2', C-5'
4'	-	164.6 (C)	-
5'	6.42 (1H, <i>dd</i> , 8.7, 2.4)	107.8 (CH)	C-1', C-3', C-4'
6'	7.77 (1H, <i>d</i> , 8.7)	132.7 (CH)	C=O, C-4'
1''	3.39 (2H, <i>d</i> , 6.6)	24.9 (CH_2)	C-2, C-3, C-3''
2''	5.14 (1H, <i>mt</i> , 6.6)	122.4 (CH)	C-1'', C-4'', C-5''
3''	-	130.2 (C)	-
4''	1.71 (3H, <i>s</i>)	17.1 (CH_3)	C-2'', C-3'', C-5''
5''	1.63 (3H, <i>s</i>)	24.9 (CH_3)	C-2'', C-3'', C-4''
1'''	4.47 (1H, <i>br d</i> , 7.8)	69.0 (CH)	C-3'''
2'''	3.54 (1H, <i>br d</i> , 7.8)	76.2 (CH)	C-1'''
3'''	-	78.9 (C)	-
4'''	1.44 (3H, <i>s</i>)	26.1 (CH_3)	C-3''', C-5'''
5'''	1.19 (3H, <i>s</i>)	18.7 (CH_3)	C-3''', C-4'''
3-OH	7.18 (1H, <i>s</i>)	-	C-2, C-3
2'-OH	12.83 (1H, <i>s</i>)	-	C-1', C-2', C-3'

recorded in acetone- d_6

PK18**1-(2,4-Dihydroxyphenyl)-3-(7-hydroxy-6-(3-methylbut-2-enyl)benzofuran-5-yl)propan-1-one**

PK18 was obtained as a yellow gum. The UV spectrum showed maximum absorption bands at 228, 276 and 314 nm. The IR spectrum exhibited the presence of hydroxyl (3352 cm^{-1}) and carbonyl (1629 cm^{-1}) groups. Its ^1H NMR spectral data and HMBC correlations (**Table 25**) suggested that it had the same dihydrochalcone core as for **PK13**, showing signals of α - and β -methylene protons [δ_{H} 3.17-3.22 (*m*, 2H) and 3.08-3.14 (*m*, 2H)], aromatic protons of a 1,2,4-trisubstituted benzene (δ_{H} 7.60 (*d*, $J = 8.7$ Hz, H-6'); 6.36 (*dd*, $J = 8.7, 2.4$ Hz, H-5'); 6.40 (*d*, $J = 2.4$ Hz, H-3')), an aromatic proton (δ_{H} 7.02, H-6) on ring B, a hydrogen bonded hydroxyl group (δ_{H} 12.78), a prenyl group (δ_{H} 3.52 (*d*, $J = 6.6$ Hz, H-1''); 5.19 (*mt*, $J = 6.6$ Hz, H-2''); 1.79 (*s*, H-4''); 1.70, *s*, H-5'')), without the signals of a 2,2-dimethylchromene ring. The assignment of a furan ring was indicated from the resonances of olefinic protons at δ_{H} 6.68 (*d*, $J = 2.1$ Hz, H-1''') and 7.55 (*d*, $J = 2.1$ Hz, H-2'''). The later one was shown at the lower field due to deshielding effect by oxygen atom. The correlations of olefinic protons H-1''' and H-2''' to C-4 (δ_{C} 142.7), C-5 (δ_{C} 126.6) correctly determined that the furan ring was at C-4 and C-5 position. In ^{13}C NMR spectrum, oxy- sp^2 carbon (C-2''') and sp^2 carbon (C-1''') resonated at δ_{C} 144.6 at δ_{C} 106.9, respectively. A molecular ion in the HREI-MS at m/z 366.1461 which corresponded to a molecular formula of $\text{C}_{22}\text{H}_{22}\text{O}_5$ confirmed that **PK18** was 1-(2,4-dihydroxyphenyl)-3-(7-hydroxy-6-(3-methylbut-2-enyl)benzofuran-5-yl)propan-1-one, a new dihydrochalcone.

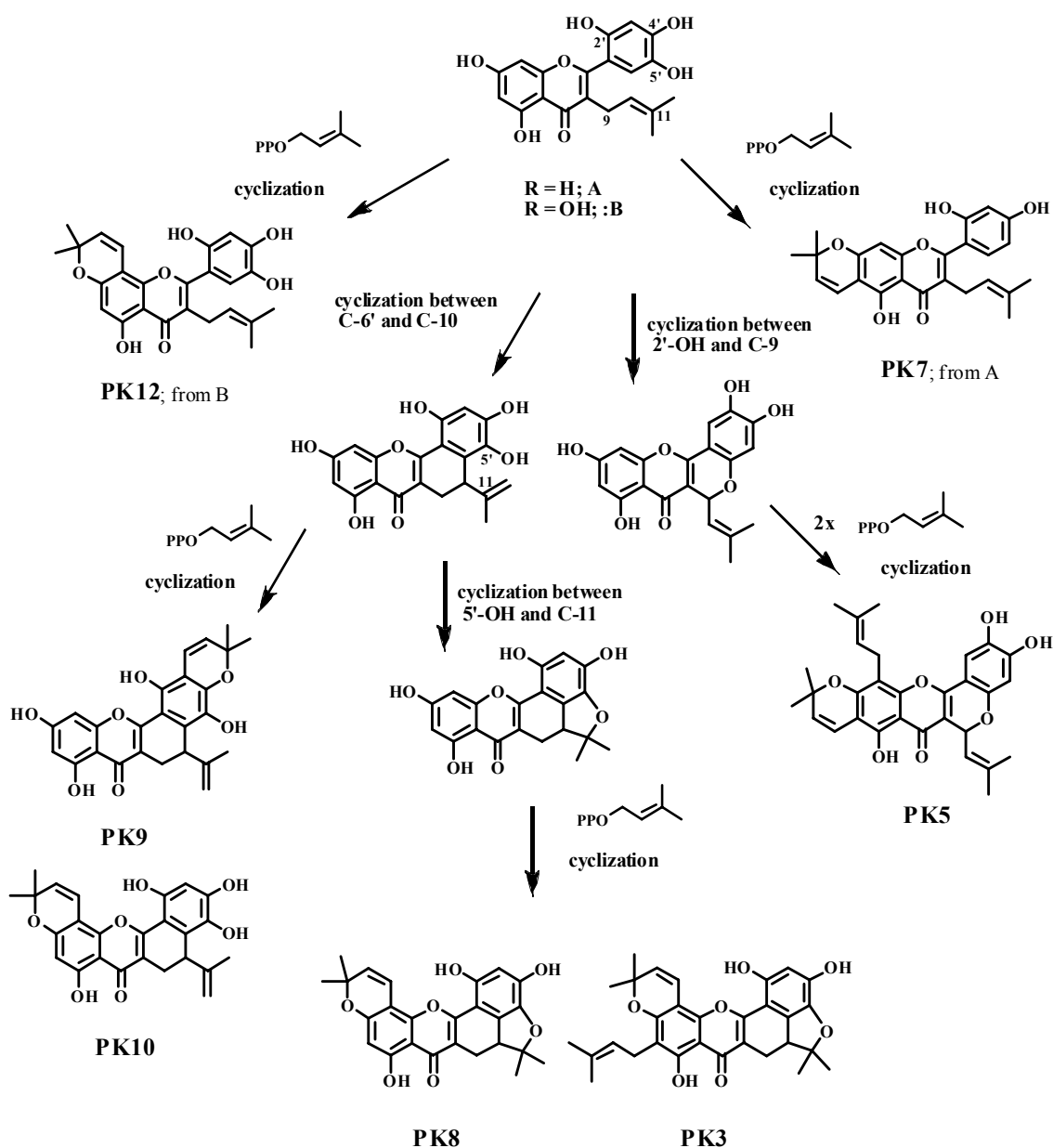
Major HMBC of **PK18****Table 25** ^1H , ^{13}C and HMBC spectral data of **PK18**

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C} (C-Type)	HMBC
1	-	135.1 (C)	-
2	-	121.6 (C)	-
3	-	139.4 (C)	-
4	-	142.7 (C)	-
5	-	126.6 (C)	-
6	7.02 (1H, <i>s</i>)	112.9 (CH)	C_{β} , C-2, C-4, C-1'''
C=O	-	203.7 (C=O)	-
α	3.17-3.22 (2H, <i>m</i>)	39.8 (CH_2)	C=O, C_{β} , C-1
β	3.08-3.14 (2H, <i>m</i>)	28.2 (CH_2)	C=O, C_{α} , C-2, C-6
1'	-	113.6 (C)	-
2'	-	165.3 (C)	-
3'	6.40 (1H, <i>d</i> , 2.4)	103.6 (CH)	C-2', C-4', C-5'
4'	-	163.1 (C)	-
5'	6.36 (1H, <i>dd</i> , 8.7, 2.4)	107.9 (CH)	C-1', C-3'
6'	7.60 (1H, <i>d</i> , 8.7)	132.1 (CH)	C=O, C-1', C-4'
1''	3.52 (2H, <i>d</i> , 6.6)	25.2 (CH_2)	C-1, C-2, C-3, C-2'', C-3''
2''	5.19 (1H, <i>t</i> , 6.6)	122.8 (CH)	-
3''	-	133.1 (C)	-
4''	1.79 (3H, <i>s</i>)	17.9 (CH_3)	C-2'', C-3'', C-5''
5''	1.70 (3H, <i>s</i>)	25.7 (CH_3)	C-2'', C-3'', C-4''
1'''	6.68 (1H, <i>d</i> , 2.1)	106.9 (CH)	C-4, C-5, C-2'''
2'''	7.55 (1H, <i>d</i> , 2.1)	144.6 (CH)	C-4, C-5, C-1''
2'-OH	12.78 (1H, <i>s</i>)	-	C-1', C-2', C-3'

recorded in CDCl_3

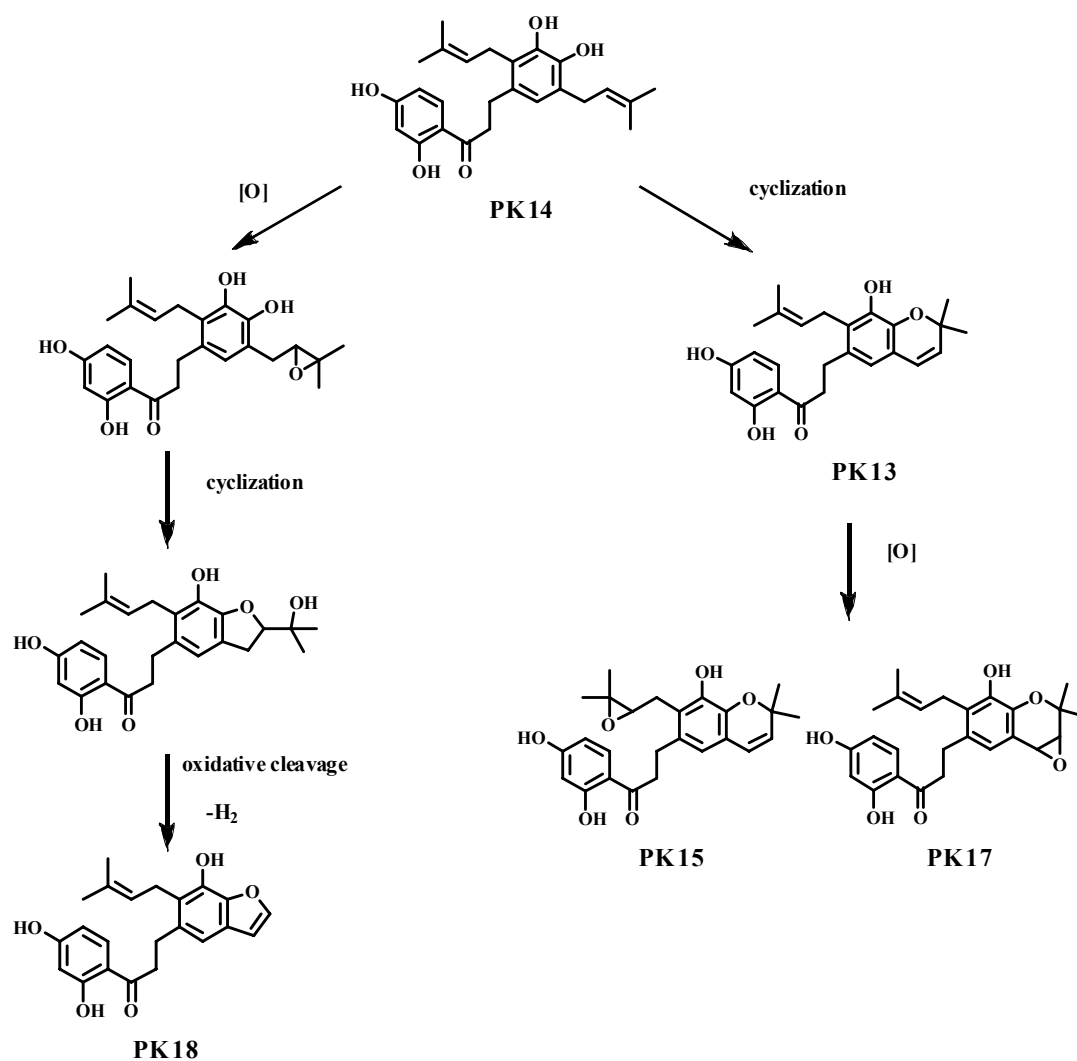
3.2 Relationship of flavonoids in this study

Investigation of *A. elasticus* has revealed that flavonoids is main components in this plant. The various classes of flavonoids is in agreement with the biogenetic relationship involving a 3-prenylatedflavone as a key intermediate. The relationship of flavones in this study can be discussed in **Scheme 12**.



Scheme 12 Relationship of prenylated flavones from *A. elasticus*

The major component isolated from the leaves of *A. elasticus* is **PK14** which has the relationship with other compounds as shown in **Scheme 13**.

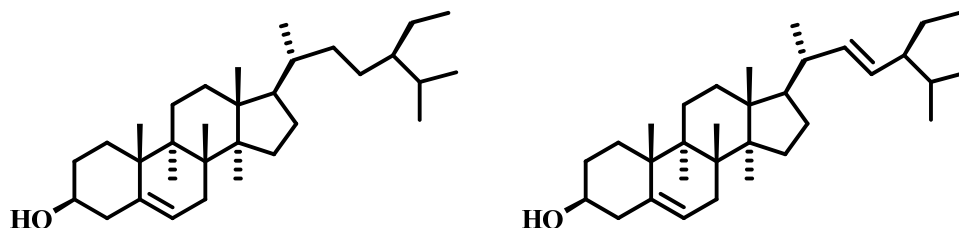


Scheme 13 Relationship of prenylated dihydrochalcones from *A. elasticus*

Conclusion

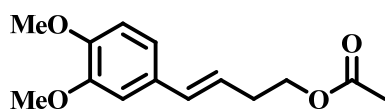
Investigation of the chemical constituents from the root bark and leaves of *A. elasticus* led to the isolation of various types of compounds. A triterpenoids (**PK1**), two phenylbutenoids (**PK2** and **PK6**), two furanodihydrobenzoxanthenes (**PK3** and **PK8**), a benzaldehyde derivatives (**PK4**), a pyranoflavones (**PK5**), two 3-prenylated flavones (**PK7** and **PK12**), two dihydrobenzoxanthenes (**PK9** and **PK10**), a phenylpropanoids (**PK11**) were obtained from the root. A flavanone (**PK16**) and five new prenylated dihydrochalcones (**PK13**, **PK14**, **PK15**, **PK17** and **PK18**) were isolated from the leaves. **PK13**, **PK14**, **PK15**, **PK17** and **PK18** are new compounds. **PK2**, **PK4**, **PK5**, **PK6**, **PK7**, **PK9**, **PK10** and **PK16** were obtained for the first time from this plant. Since this plant has been reported to have anti-inflammatory activity and cytotoxicity, further study on the antibacterial and antiprotozoa activity of the isolated compound should be performed.

Triterpenoids

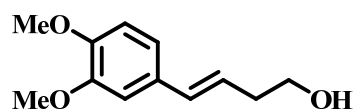


PK1 : a mixture of β -sitosterol and stigmasterol

Phenylbutenoids

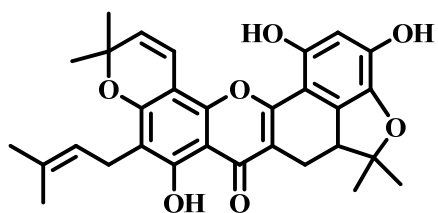


PK2 : (*E*)-4-(3',4'-dimethoxyphenyl)-
butenyl acetate

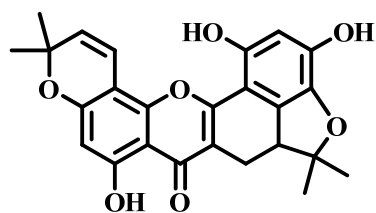


PK6 : (*E*)-4-(3',4'-dimethoxyphenyl)
but-3-en-1-ol

Furanodihydrobenzoxanthenes

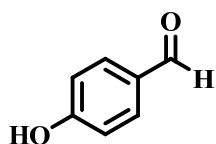


PK3 : artonin F



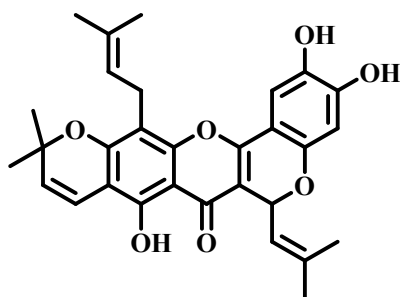
PK8 : cycloartobiloxanthone

Benzaldehyde derivatives



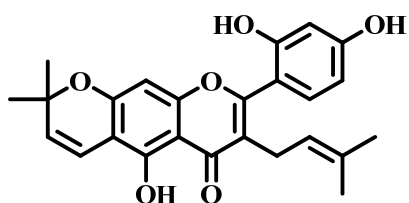
PK4 : 4-hydroxybenzaldehyde

Pyranoflavones

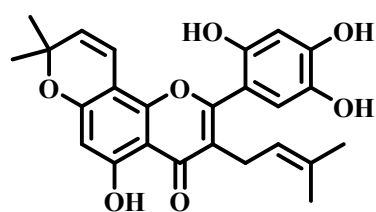


PK5 : cycloheterophyllin

3-Prenylated flavones

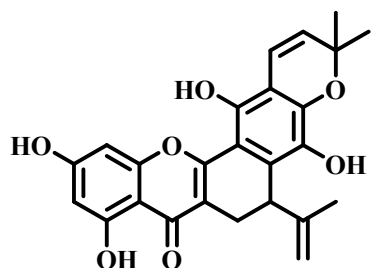


PK7 : cudraflavone B

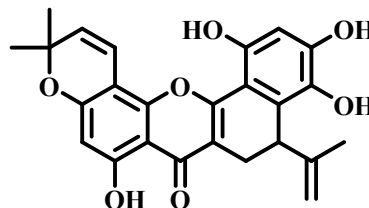


PK12 : artonin E

Dihydrobenzoxanthones

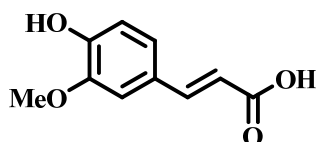


PK9 : artelastoxanthone



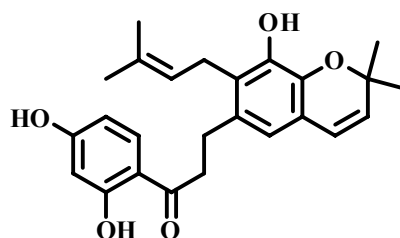
PK10 : artobiloxanthone

Phenylpropanoids

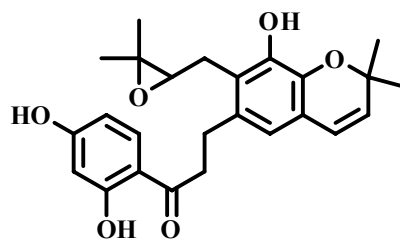


PK11 : *trans*-feluric acid

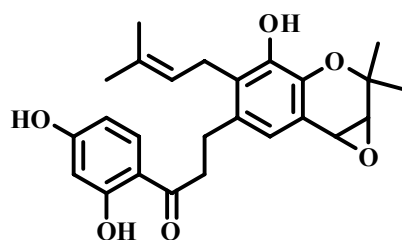
Prenylated dihydrochalcones



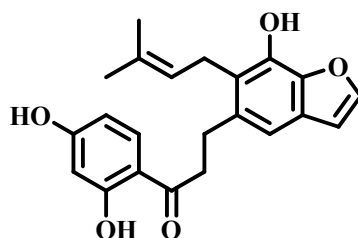
PK14 : 1-(2,4-dihydroxyphenyl)-3-(3,4- dihydroxy-2,5-
bis(3-methylbut-2- enyl) phenyl)propan-1-one



PK15 : 1-(2,4-dihydroxyphenyl)-3-(7- ((3,3-dimethyloxiran-2-yl) methyl)-
8-hydroxy-2,2-dimethyl-2*H*-chromen-6-yl) propan-1-one

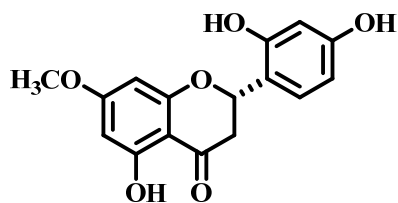


PK17 : 1-(2,4-dihydroxyphenyl)-3-(4-hydroxy-2,2-dimethyl-5-(3-methylbut-2-enyl)-2,7*b*-dihydro-1*aH*-oxireno[2,3-*c*] chromen-6-yl)propan-1-one



PK18 : 1-(2,4-dihydroxyphenyl)-3-(7-hydroxy-6-(3-methylbut-2-enyl)benzofuran-5-yl)propan-1-one

Flavonones



PK16 : artocarpanone

REFERENCES

- Aida, M.; Yamagami, Y.; Hano, Y.; Nomura, T. 1996. "Formation of dihydrobenzoxanthone skeleton from 3-isoprenylated 2',4',5'-trioxygenated flavone", *Heterocycles*, 43, 2561-2565.
- Aida, M.; Yamagami, Y.; Hano, Y.; Nomura, T. 1997. "Artonols A, B, C, D and E, five new isoprenylated phenols from the bark of *Artocarpus communis* Forst.", *Heterocycles*, 45, 163-175.
- Amarasinghe, N. R.; Jayasinghe, L.; Hara, N.; Fujimoto, Y. 2008. "Chemical constituents of the fruits of *Artocarpus altilis*", *Biochem. Syst. Ecol.*, 36, 323-325.
- Arung, E. T.; Yoshikawa, K.; Shimizu, K.; Kondo, R. 2010. "Isoprenoid-substituted flavonoids from wood of *Artocarpus heterophyllus* on B16 melanoma cells: Cytotoxicity and structural criteria" *Fitoterapia*, 81, 120-123.
- Barron, D.; Ibrahim, R. K. 1996. "Isoprenylated flavonoids-A survey", *Phytochemistry*, 43(5), 921-982.
- Boonnak, N. 2006. "Chemical constituents from the barks of *Cratoxylum formosum* spp. *pruniflorum*", M.Sc., Prince of Songkla University.
- Cerqueira, F.; Cidade, H., van Ufford, L.; Beukelman, C.; Kijjoa, A.; Nascimento, M. S. J. 2008, "The natural prenylated flavone artelastin is an inhibitor of ROS and NO production", *Intl. Immunopharmacol.*, 8, 597-602.
- Cidade, H. M.; Nascimento, M. S. Pinto, M. M. M.; Kijjoa, A.; Silva, A. M.; Herz, W. 2001. "Artelastocarpin and carpelastofuran, two new flavones, and cytotoxicities of prenyl flavonoids from *Artocarpus elasticus* against three cancer cell lines.", *Planta Med.*, 67, 867-870.

- Dang, D. T. N.; Eriste, E.; Liepinsh, E.; Trinh, T. T.; Harris, H. E.; Sillard, R.; Larsson, P. 2009. "A novel anti-inflammatory compound, artonkin-4'-O-glucoside, from the leaves of *Artocarpus tokinensis* suppresses experimentally induced arthritis", *Scand. J. Immunol.*, 69, 110-118.
- de Faria, A. F.; de Rosso V. V.; Mercadante A. Z. 2009. "Carotenoid composition of jackfruit (*Artocarpus heterophyllus*), determined by HPLC-PDA-MS/MS", *Plant Foods Hum Nutr*, 64, 108–115.
- Fang, S. C.; Hsu, C. L.; Yu, Y. S.; Yen, G. C. 2008. "Cytotoxic effects of new geranyl chalcone derivatives isolated from the leaves of *Artocarpus communis* in SW 872 Human liposarcoma cells", *J. Agric. Food Chem.*, 56, 8859-8868.
- Jamil, S.; Sirat, H. M.; Jantan, I.; Aimi, N.; Kitajima, M. 2008. "A new prenylated dihydrochalcone from the leaves of *Artocarpus lowii*", *J. Nat. Med.*, 62, 321-324.
- Jang, D. S.; Han, A. R.; Park, G.; Jhon, G. J.; Seo, E. K. 2004. "Flavonoids and aromatic compounds from the rhizomes of *Zingiber zerumbet*" *Arch. Pharm. Res.*, 27(4), 386-389.
- Jayasinghe, U. L. B.; Samarakoon, T. B.; Kumarihamy, B. M. M.; Hara, N.; Fujimoto, Y. 2008. "Four new prenylated flavonoids and xanthenes from the root bark of *Artocarpus nobilis*", *Fitoterapia*, 79, 37-41.
- Hakim, E. H.; Yurnawilis, A.; Aimi, N.; Kitajima, M.; Takayama, H. 2002. "Artoindonesianin P, a new prenylated flavone with cytotoxic activity from *Artocarpus lanceifolius*", *Fitoterapia*, 73, 668–673.
- Hakim, E. H.; Juliawaty, L. D.; Syah, Y. M.; Achmak, S. A. 2005. "Molecular diversity of *Artocarpus champeden* (Moraceae): A species endemic to Indonesia", *Molecular Diversity*, 9, 149-158.
- Hakim, E. H.; Achmad, S. A.; Juliawaty, L. D.; Makmur, L.; Syah, Y. M.; Aimi, N.; Kitajima, M.; Takayama, H.; Ghisalberti, E. L. 2006. "Prenylated flavonoids and related compounds of the Indonesian *Artocarpus* (Moraceae)", *J. Nat. Med.*, 60, 161-184.

- Han, A. R.; Lee, E. J.; Min, H. Y.; Kim, H. R.; Lee, S. K.; Seo, E. K. 2003. "A potential cytotoxic principle of *Zingiber cassumunar*", *Natural Product Sciences*, 9(2), 109-111.
- Hano, Y.; Aida, M.; Shiina, M.; Nomura, T.; Kawai, T.; Ohe, H.; Kagei, K. 1989. "Artonins A and B, two new prenylflavones from the root bark of *Artocarpus heterophyllus* Lamk.", *Heterocycles*, 29, 147-1453.
- Hano, Y.; Yamagami, Y.; Kobayashi, M.; Isohata, R.; Nomura, T. 1990. "Artonins E and F, two new prenylflavones from the bark of *Artocarpus communis* Forst", *Heterocycles*, 31(5), 877-882.
- Hwang, J. H.; Hong, S. S.; Han, X. H.; Hwang, J. S.; Lee, D.; Lee, H.; Yun, Y. P.; Kim, Y.; Ro, J. S.; Hwang, B. Y. 2007. "Prenylated xanthenes from the root bark of *Cudrania tricuspidata*", *J. Nat. Prod.*, 70, 1207-1209.
- Kelly, C. J.; Harruff, R. C.; Carmack, M. 1976. "The polyphenolic acids of *Lithospermum ruderale* II. Carbon-13 nuclear magnetic resonance of lithospermic and rosmarinic acids", *J. Org. Chem.*, 41(3), 449-454.
- Ko, H. H.; Lu, Y. H.; Yang, S.Z.; Won, S. J.; Lin, C. N. 2005. "Cytotoxic prenylflavonoids from *Artocarpus elasticus*", *J. Nat. Prod.*, 68, 1692-1695.
- Lin, K. W.; Liu, C. H.; Tu, H. Y.; Ko, H. H.; Wei, B. L. 2009. "Antioxidant prenylflavonoids from *Artocarpus communis* and *Artocarpus elasticus*", *Food Chemistry*, 115, 558-562.
- Musthapa, I.; Juliawaty, L. D.; Syah, Y. M.; Hakim, E. H.; Latip, J.; Ghisalberti, E. L. 2009. "An oxepinoflavone from *Artocarpus elasticus* with cytotoxic activity against P-388 Cells", *Arch. Pharm. Res.*, 32(2), 191-194.
- Nomura, T.; Hano, Y.; Aida, M. 1998. "Isoprenoid-substituted flavonoids from *Artocarpus* Plants (Moraceae)", *Heterocycles*, 47, 1179-1205.
- Ren, Y.; Kardono, L. B. S.; Riswan, S.; Chai, H.; Farnsworth, N. R.; Soejarto, D. D.; Blanco, E. J. C.; Kinghorn, A. D. 2010. "Cytotoxic and NF- κ B inhibitory constituents of *Artocarpus rigida*", *J. Nat. Prod.*, 73, 949-955.

- Ryu, Y. B.; Long, M. J. C.; Lee, J. W.; Ryu, H. W.; Kim, J. Y.; Lee, W. S.; Park, K. H. 2009. "Structural characteristics of flavanones and flavones from *Cudrania tricuspidata* for neuraminidase inhibition" *Bioorg. Med. Chem. Lett.*, 19, 4912–4915.
- Sritularak, B.; Tantituvanont, A.; Chanvorachote¹, P.; Meksawan, K.; T. Miyamoto, T.; Kohno, Y.; Likhitwitayawuid, K. 2010. "Flavonoids with free radical scavenging activity and nitric oxide inhibitory effect from the stem bark of *Artocarpus gomezianus*", *J. Med. Plant. Res.*, 4(5), 387-392.
- Stoclet, J. C.; Muller, B.; Andriantsitohaina, R.; Kleschyov, A. 1998. "Overproduction of nitric oxide in pathophysiology of blood vessels", *Biochemistry*, 37, 826-832.
- Sultanbawa, M. U. S.; Surendrakumar, S. 1989. Two pyranodihydrobenzoxanthenes from *Artocarpus nobilis*", *Phytochemistry*, 28, 599-605.
- Wang, Y.; Xu, K.; Lin, L.; Pan, Y. ; Zheng, X. 2007. "Geranyl flavonoids from the leaves of *Artocarpus altilis*", *Phytochemistry*, 68, 1300-1306.
- Wei, B. L.; Weng, J. R.; Chiu, P. H.; Hung, C. F.; Wang, J. P.; Lin, C. N. 2005. "Antiinflammatory flavonoids from *Artocarpus heterophyllus* and *Artocarpus communis*", *J. Agric. Food Chem.*, 53, 3867-3871.
- Zheng, Z. P.; Chen, S.; Wang, S.; Wang, X. C.; Cheng, K. W.; Wu, J. J.; Yang, D.; Wang, M. 2009. "Chemical components and tyrosinase inhibitors from the twigs of *Artocarpus heterophyllus*", *J. Agric. Food Chem.*, 57, 6649-6655.

APPENDIX

1. ^1H -NMR and ^{13}C -NMR spectrum of compounds PK1-PK18

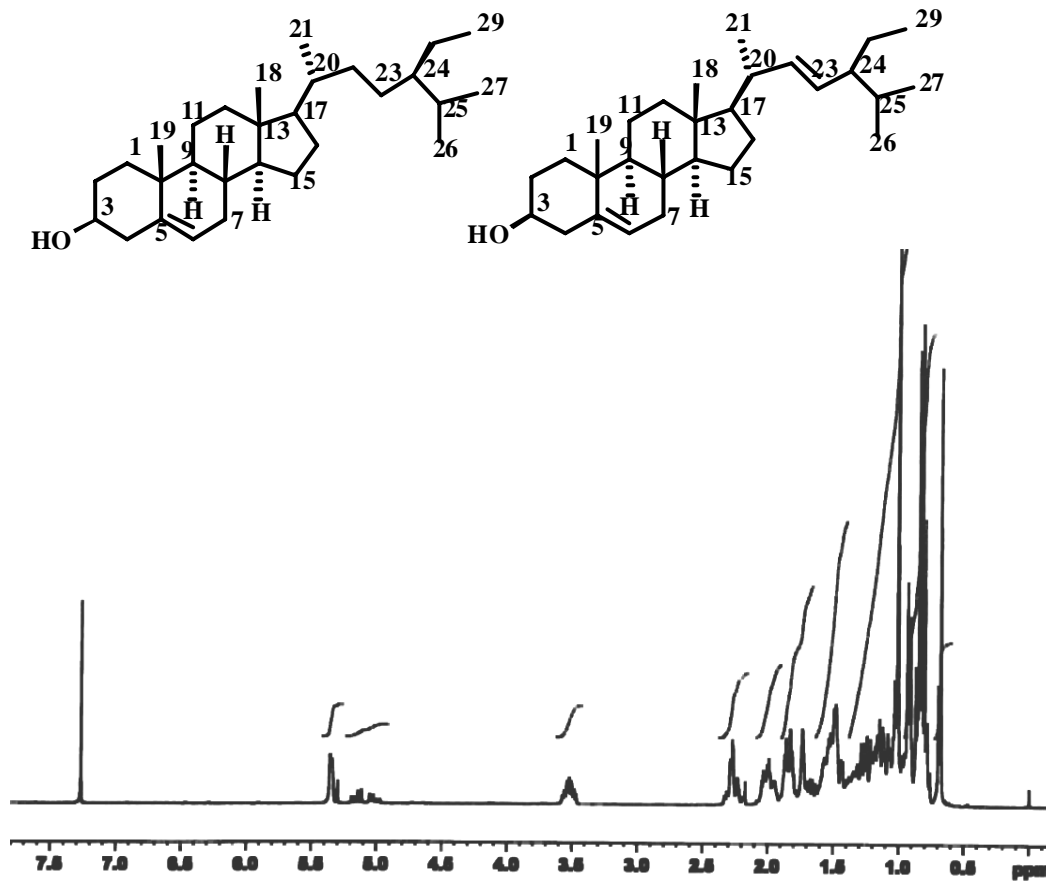


Figure A-1 ^1H -NMR (300 MHz) (CDCl_3) spectrum of PK1

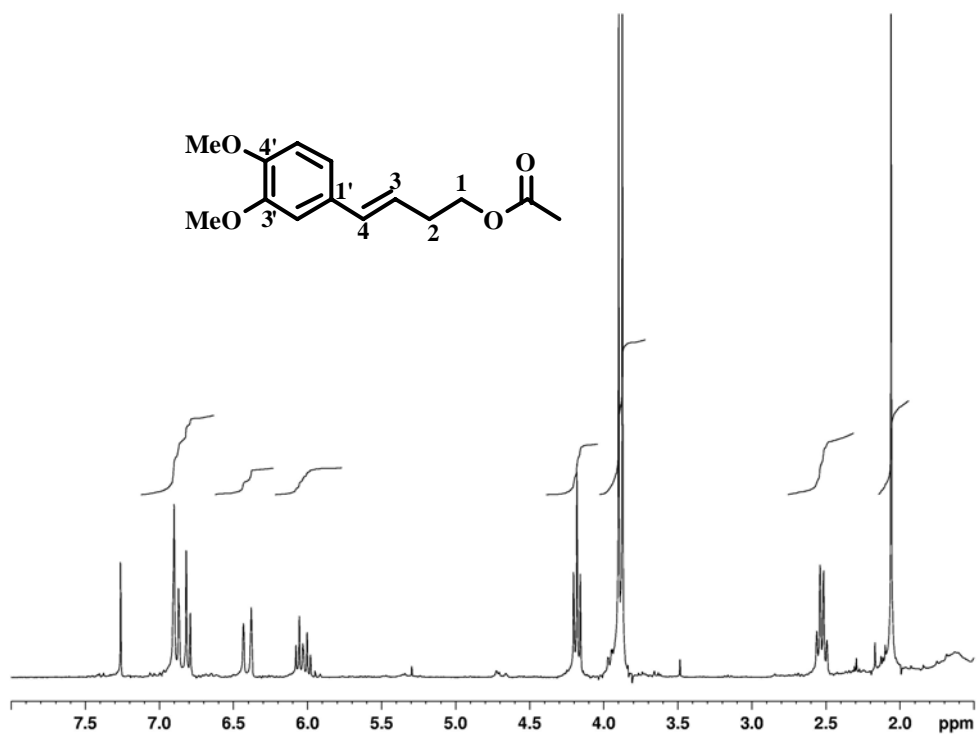


Figure A-2 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of **PK2**

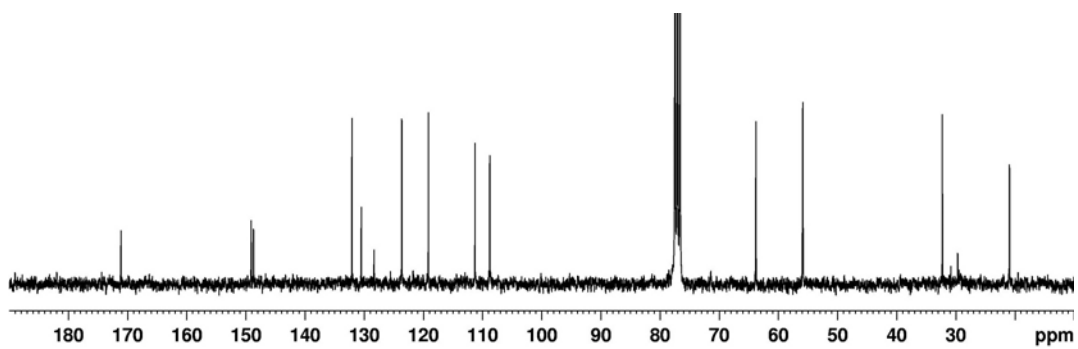


Figure A-3 $^{13}\text{C-NMR}$ (75 MHz) (CDCl_3) spectrum of **PK2**

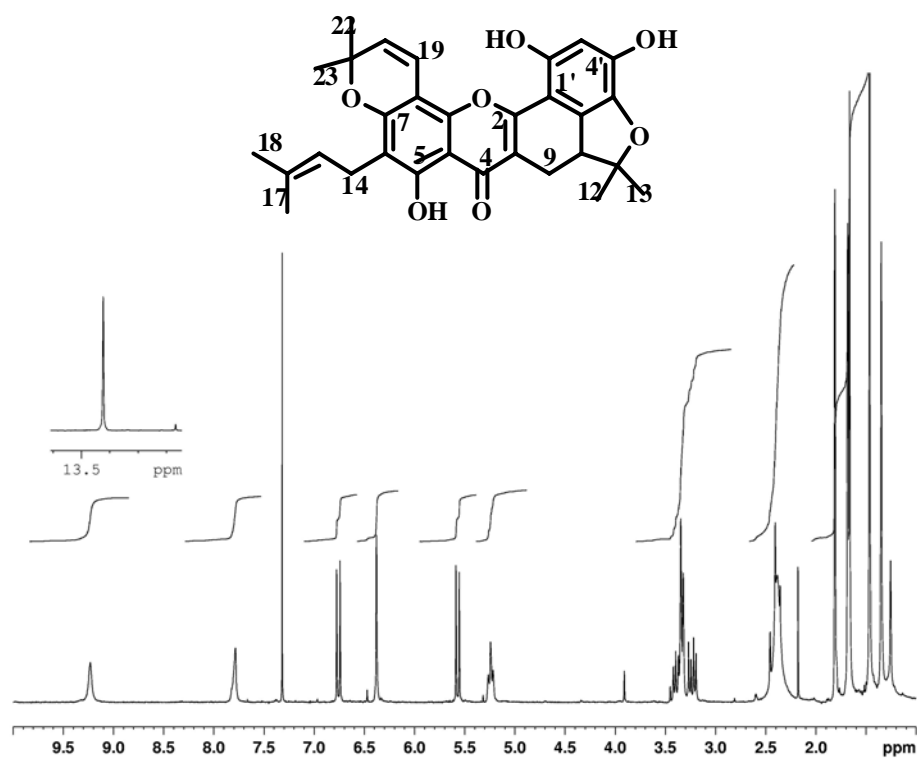


Figure A-4 $^1\text{H-NMR}$ (300 MHz) ($\text{CDCl}_3 + \text{DMSO-}d_6$) spectrum of **PK3**

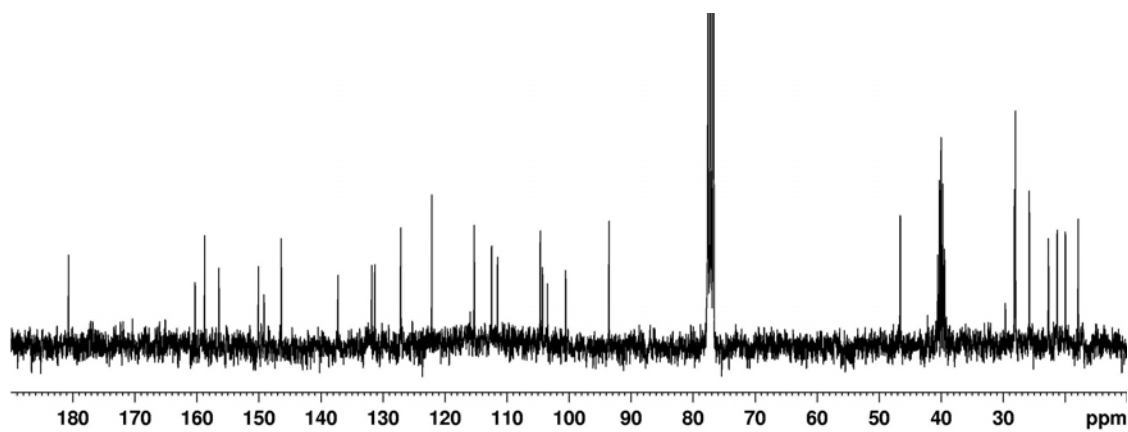


Figure A-5 $^{13}\text{C-NMR}$ (75 MHz) ($\text{CDCl}_3 + \text{DMSO-}d_6$) spectrum of **PK3**

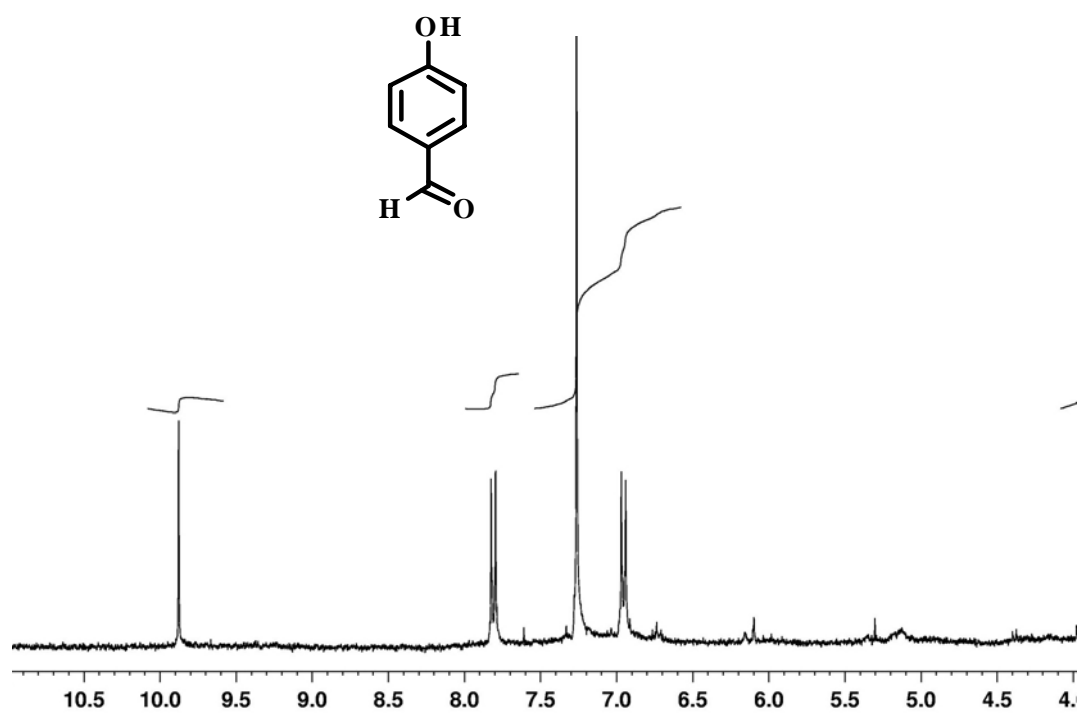


Figure A-6 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of **PK4**

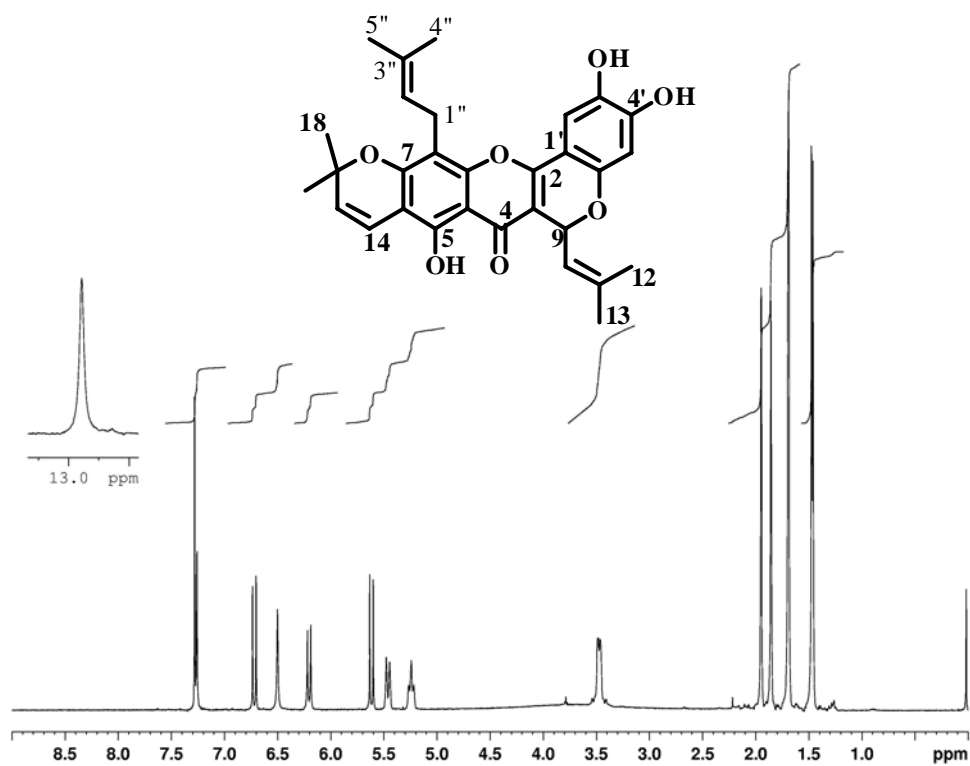


Figure A-7 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of PK5

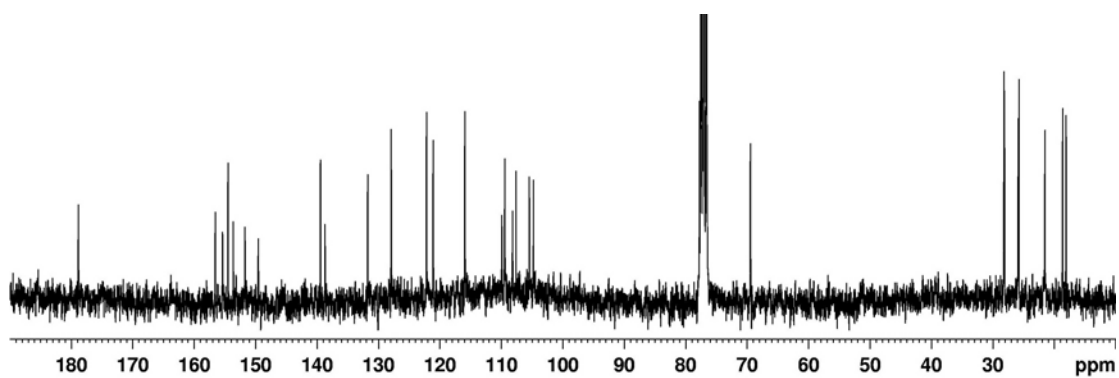


Figure A-8 $^{13}\text{C-NMR}$ (75 MHz) (CDCl_3) spectrum of PK5

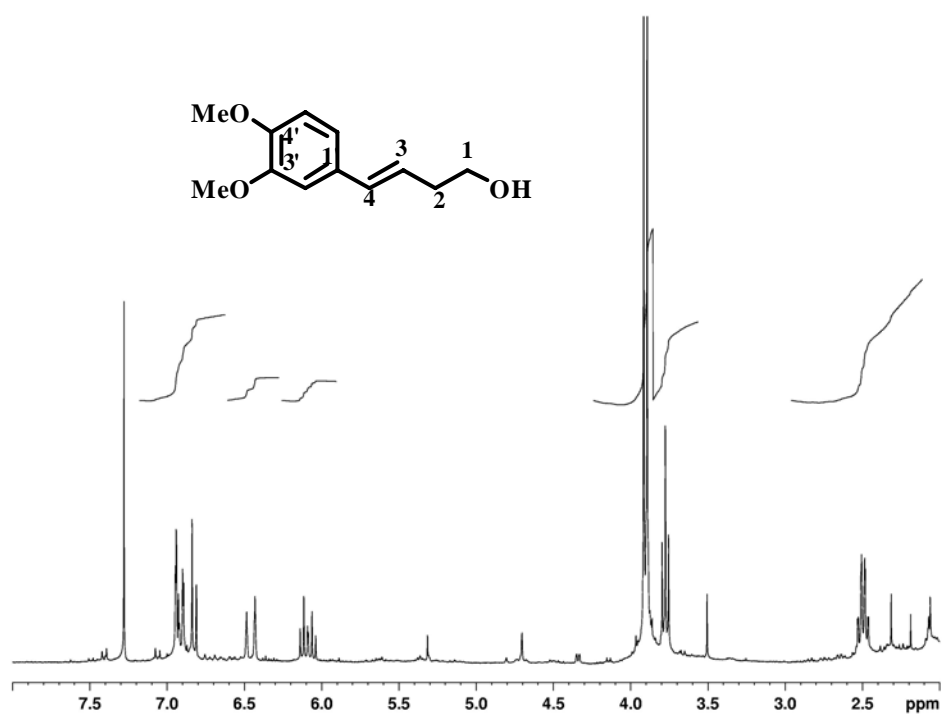


Figure A-9 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of **PK6**

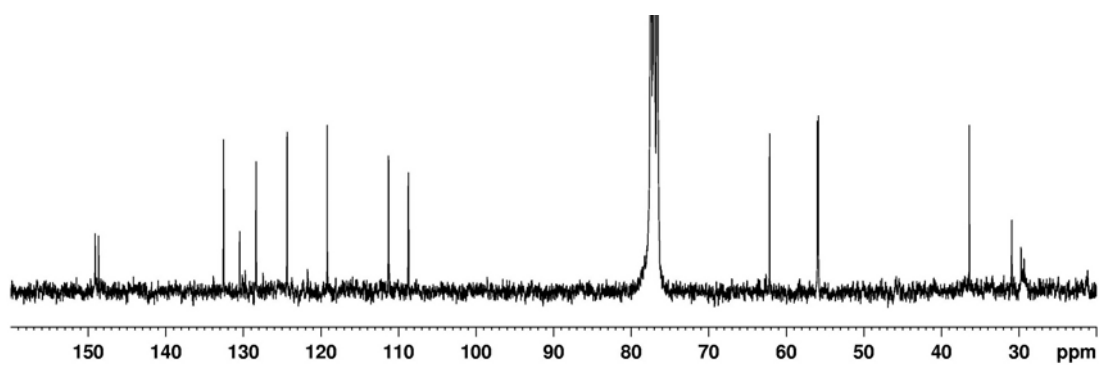


Figure A-10 $^{13}\text{C-NMR}$ (75 MHz) (CDCl_3) spectrum of **PK6**

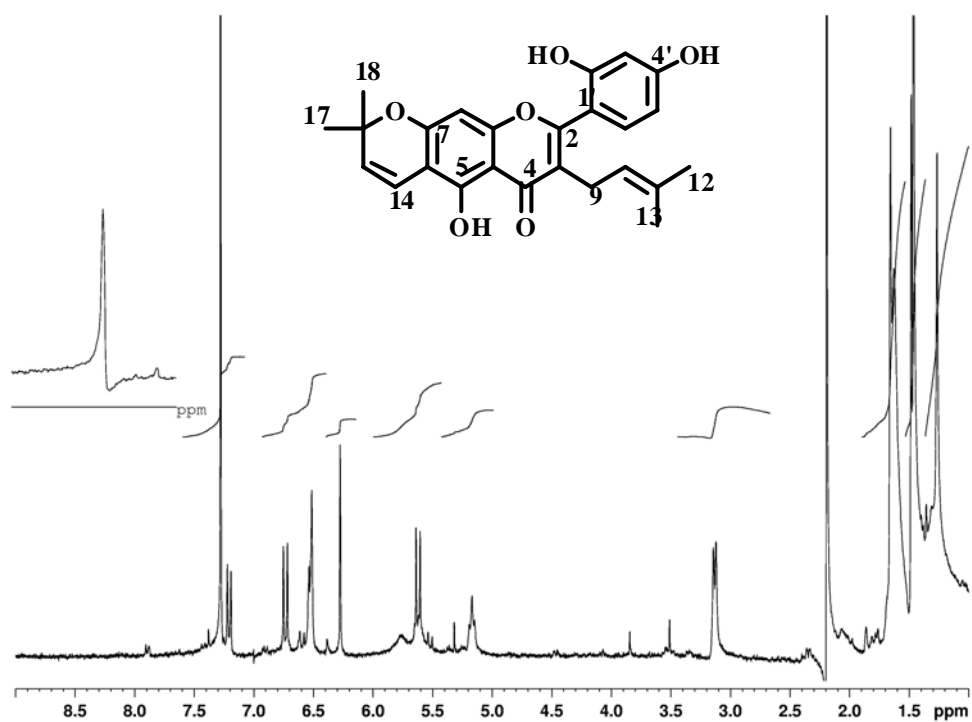


Figure A-11 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of **PK7**

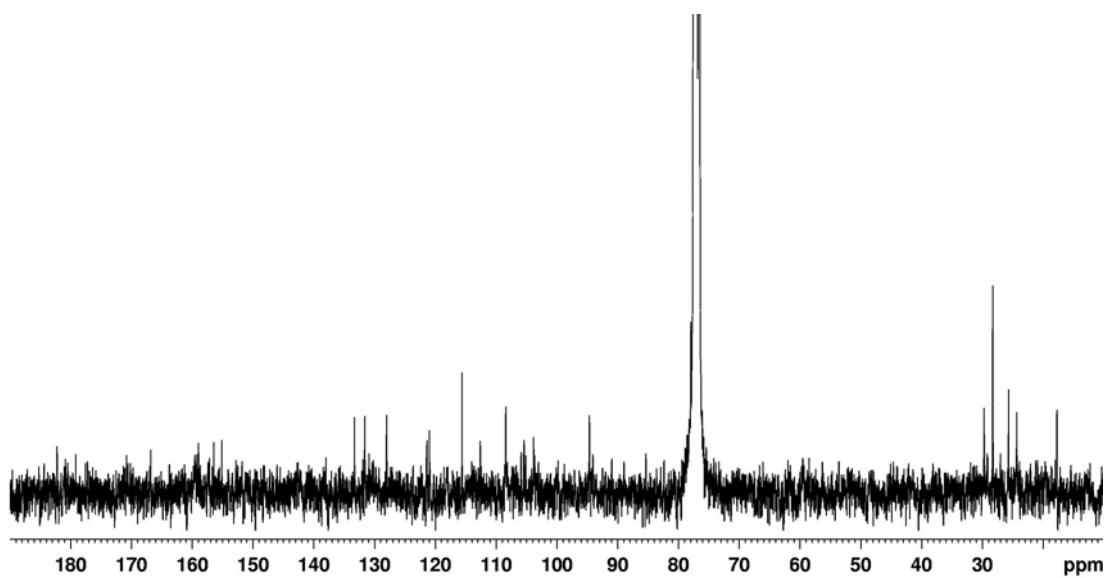


Figure A-12 $^{13}\text{C-NMR}$ (75 MHz) (CDCl_3) spectrum of **PK7**

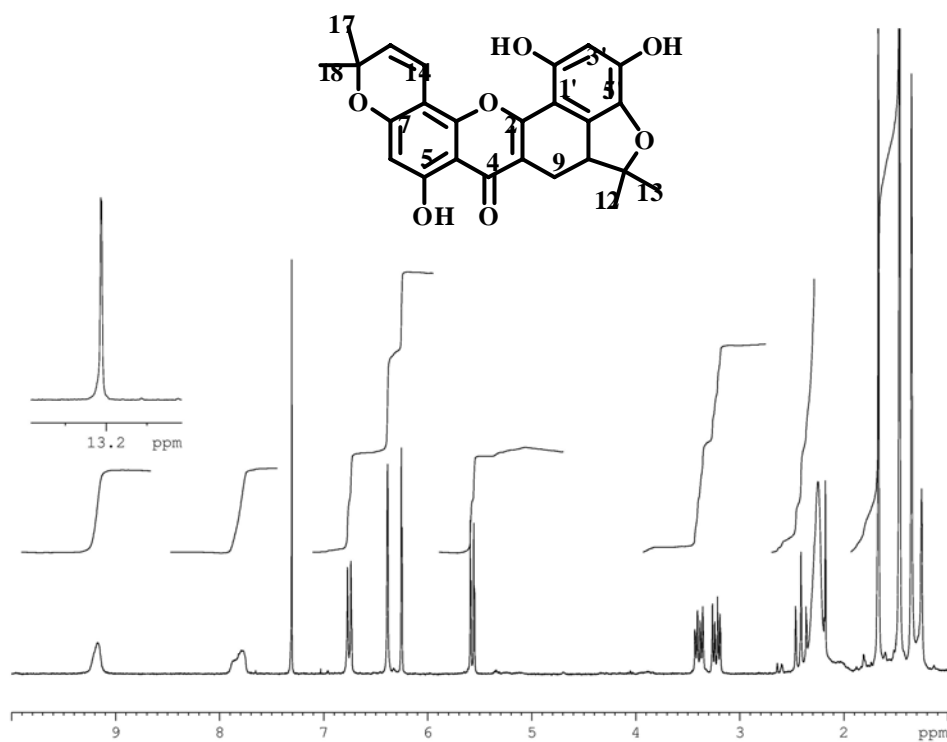


Figure A-13 $^1\text{H-NMR}$ (300 MHz) ($\text{CDCl}_3+\text{DMSO-}d_6$) spectrum of **PK8**

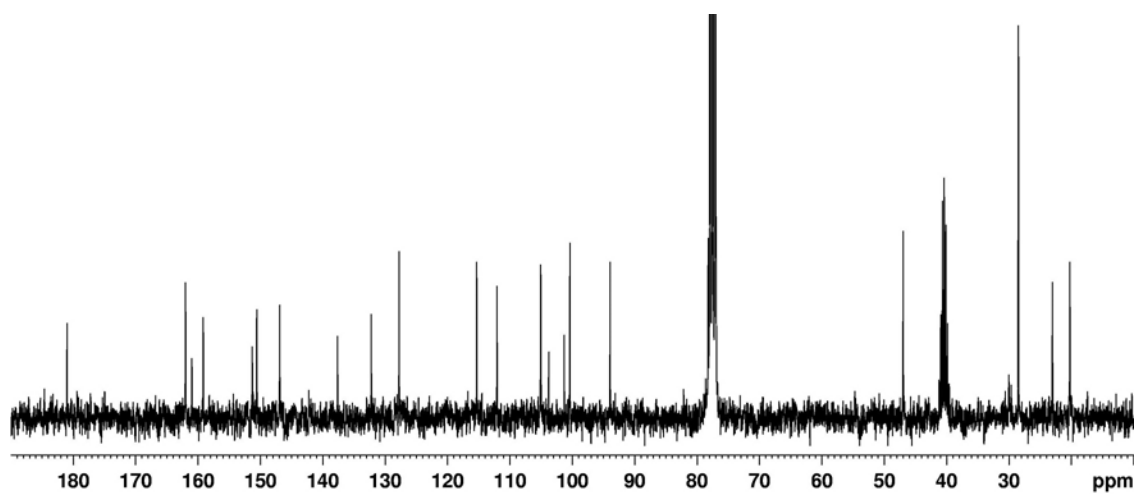


Figure A-14 $^{13}\text{C-NMR}$ (75 MHz) ($\text{CDCl}_3+\text{DMSO-}d_6$) spectrum of **PK8**

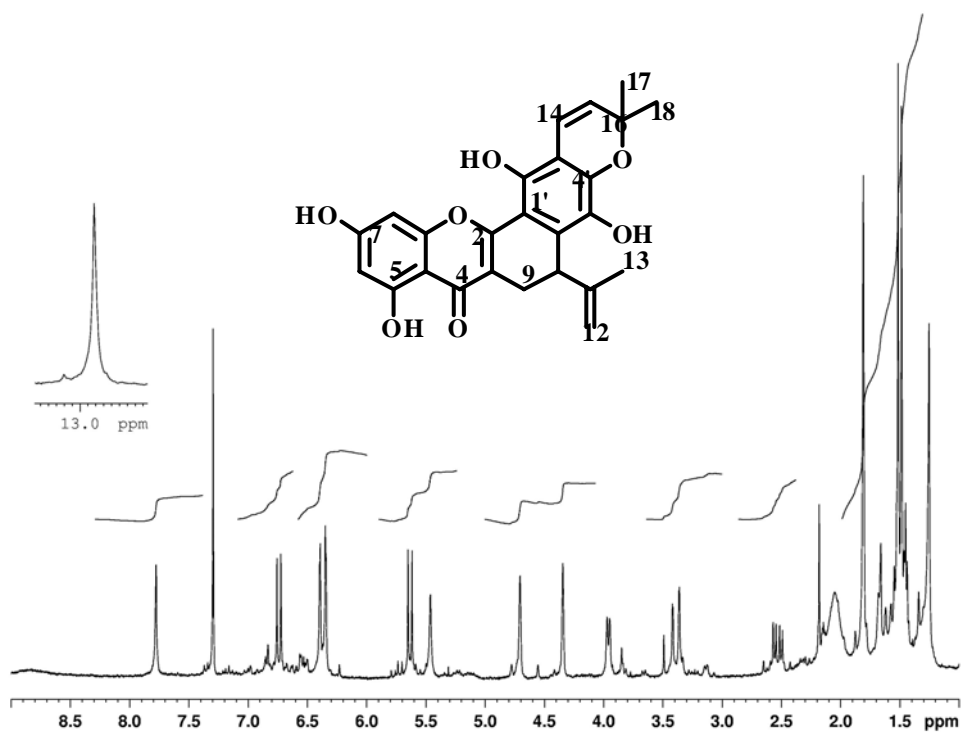


Figure A-15 $^1\text{H-NMR}$ (300 MHz) ($\text{CDCl}_3 + \text{DMSO-}d_6$) spectrum of **PK9**

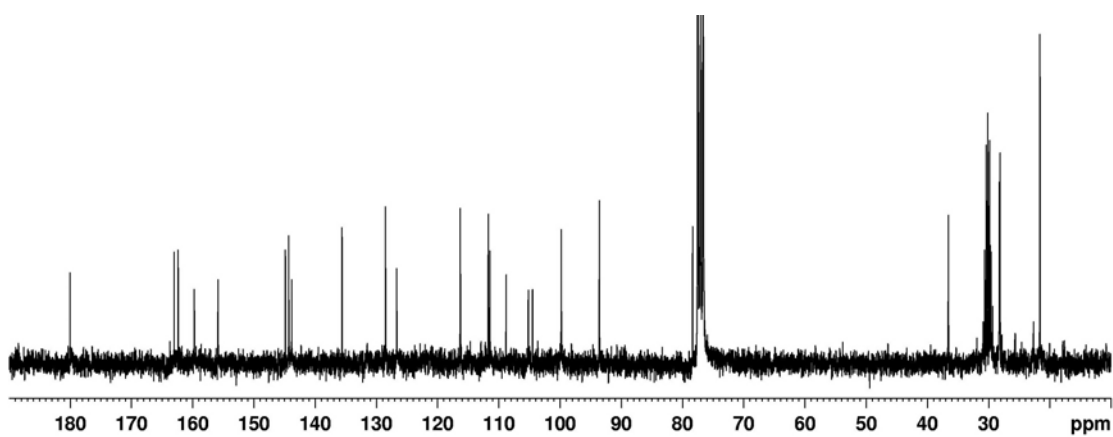


Figure A-16 $^{13}\text{C-NMR}$ (75 MHz) ($\text{CDCl}_3 + \text{DMSO-}d_6$) spectrum of **PK9**

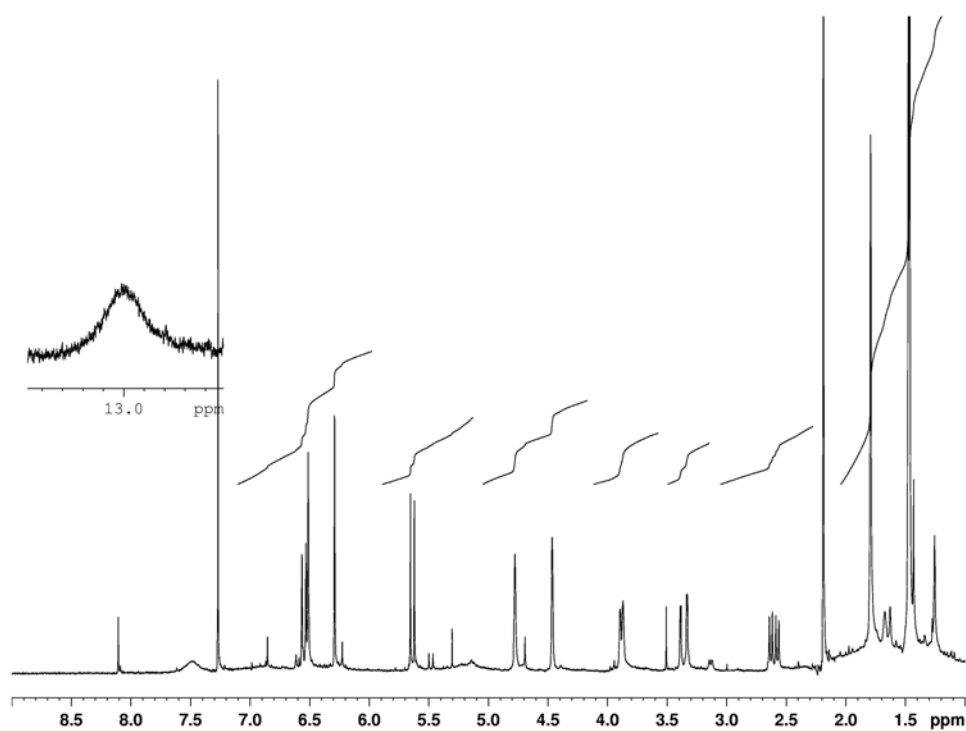


Figure A-17 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of **PK10**

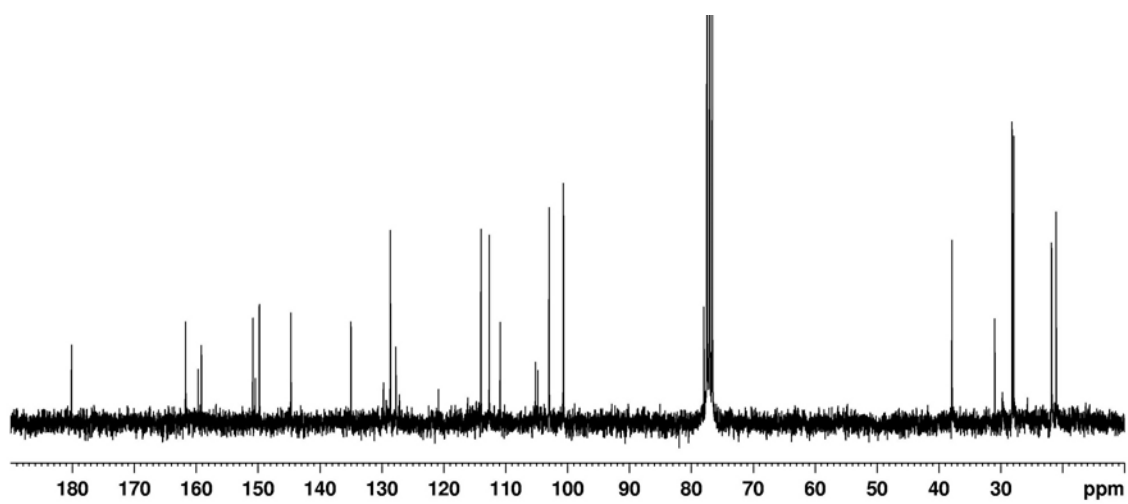


Figure A-18 $^{13}\text{C-NMR}$ (75 MHz) (CDCl_3) spectrum of **PK10**

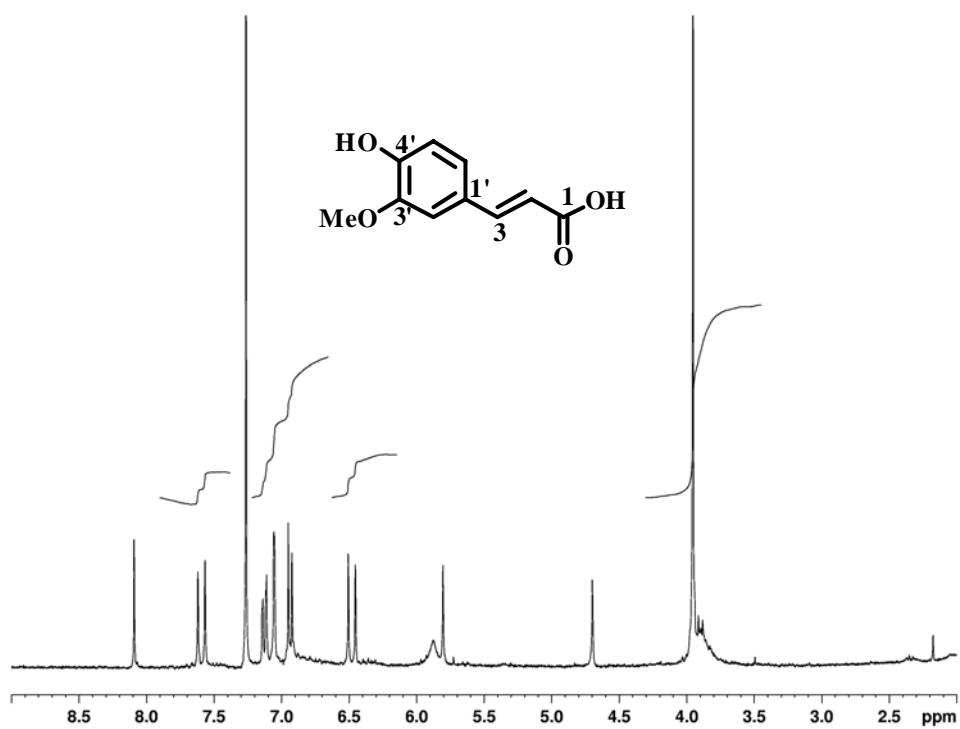


Figure A-19 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of **PK11**

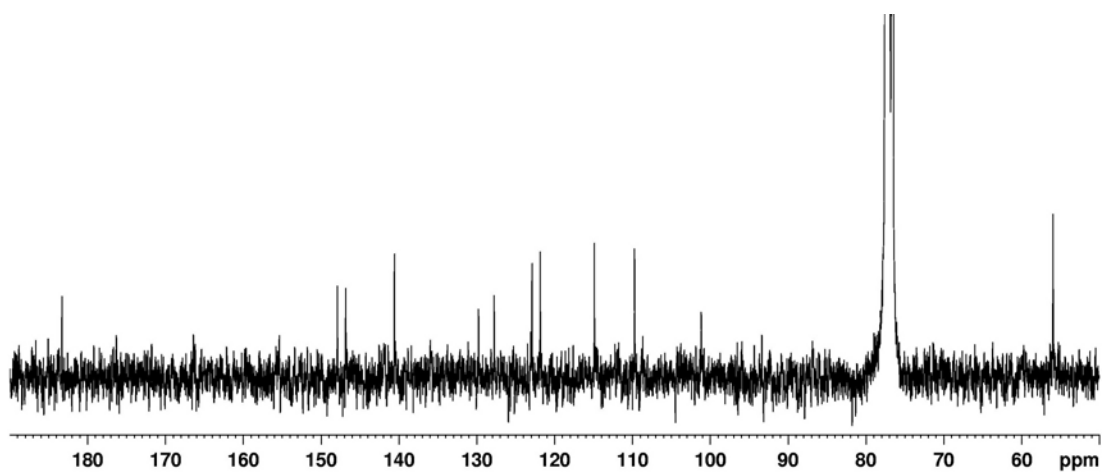


Figure A-20 $^{13}\text{C-NMR}$ (75 MHz) (CDCl_3) spectrum of **PK11**

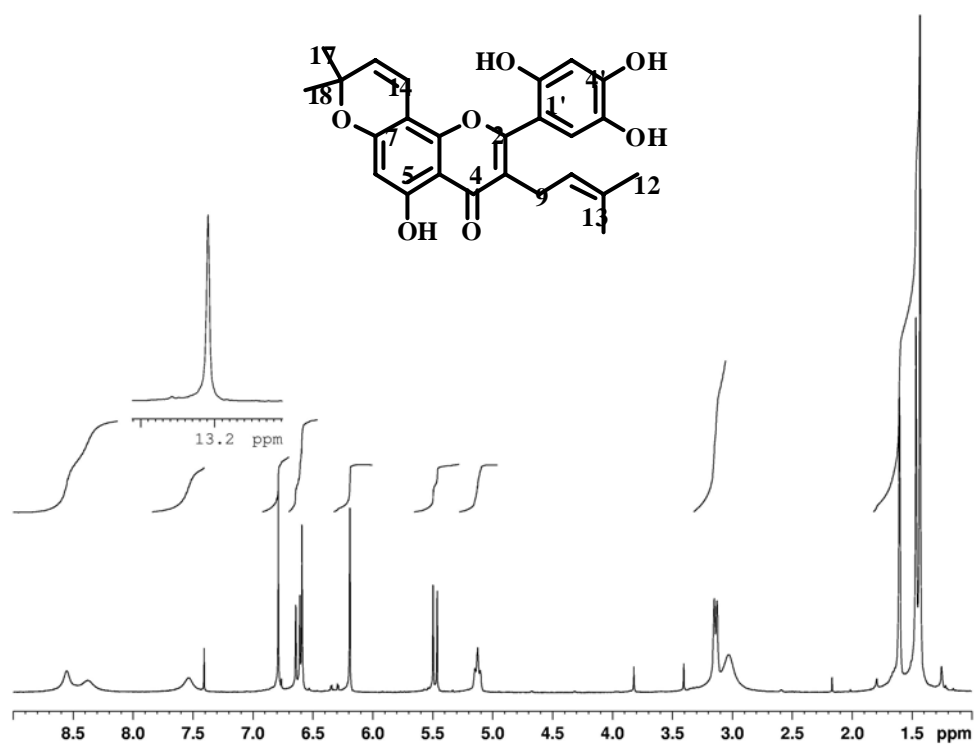


Figure A-21 ¹H-NMR (300 MHz) (CDCl₃+DMSO-*d*₆) spectrum of **PK12**

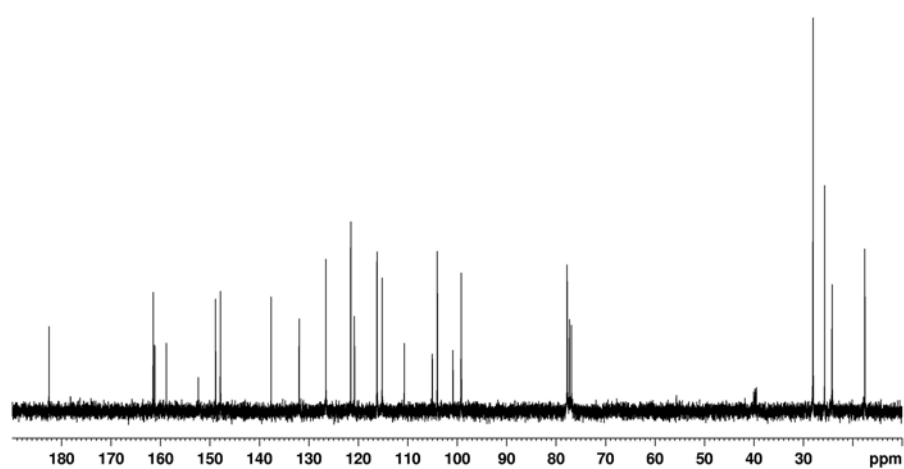


Figure A-22 ¹³C-NMR (75 MHz) (CDCl₃+DMSO-*d*₆) spectrum of **PK12**

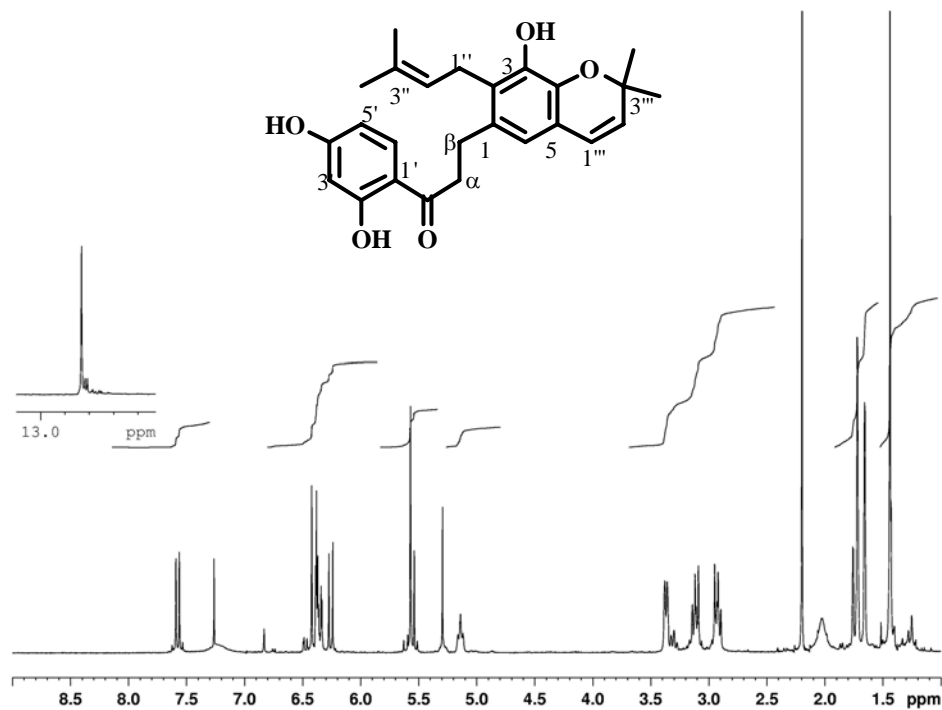


Figure A-23 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of **PK13**

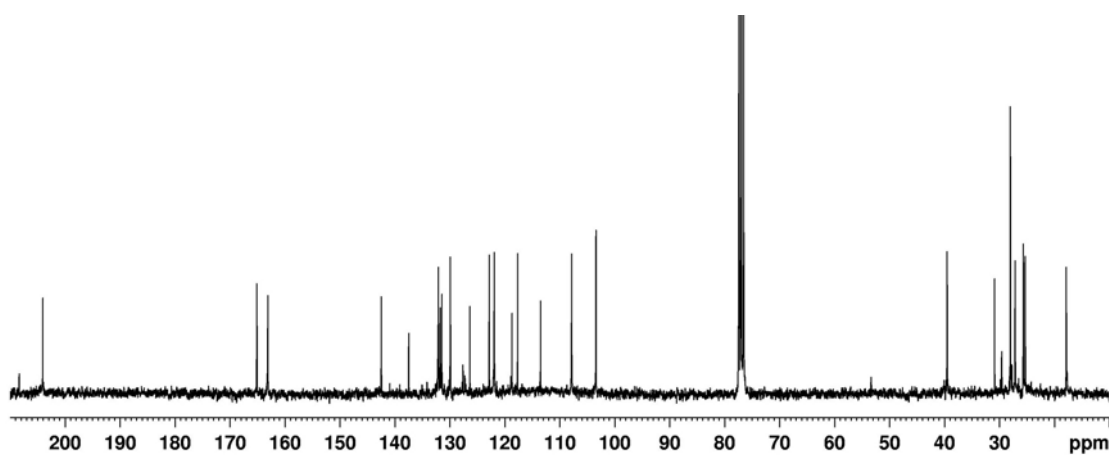


Figure A-24 $^{13}\text{C-NMR}$ (75 MHz) (CDCl_3) spectrum of **PK13**

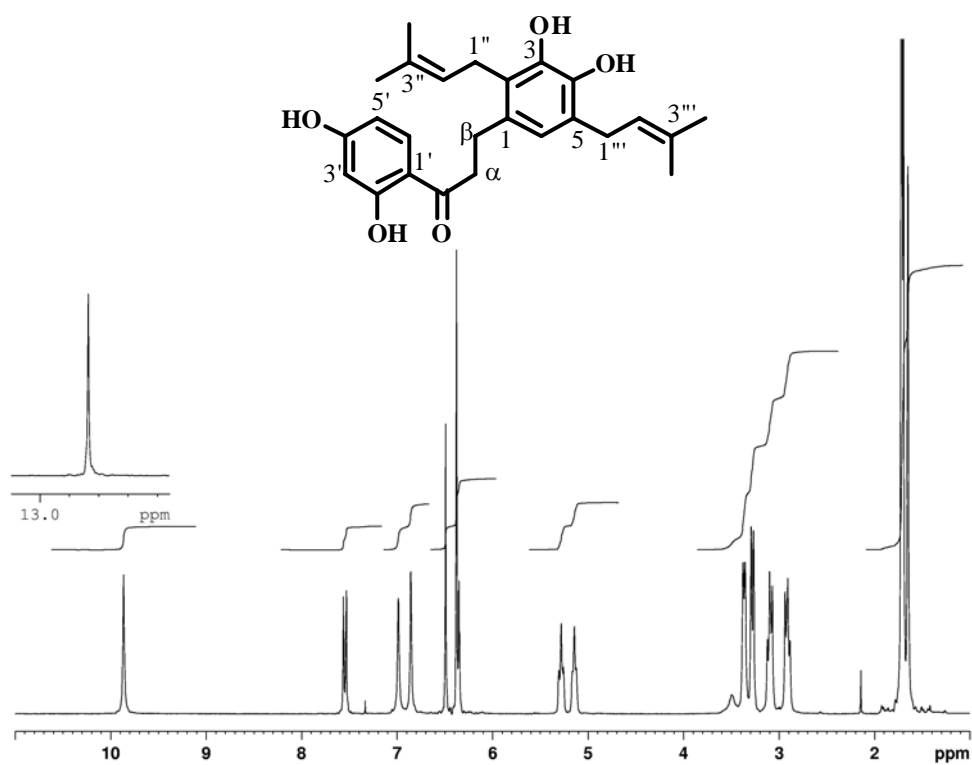


Figure A-25 $^1\text{H-NMR}$ (300 MHz) ($\text{CDCl}_3+\text{DMSO-}d_6$) spectrum of **PK14**

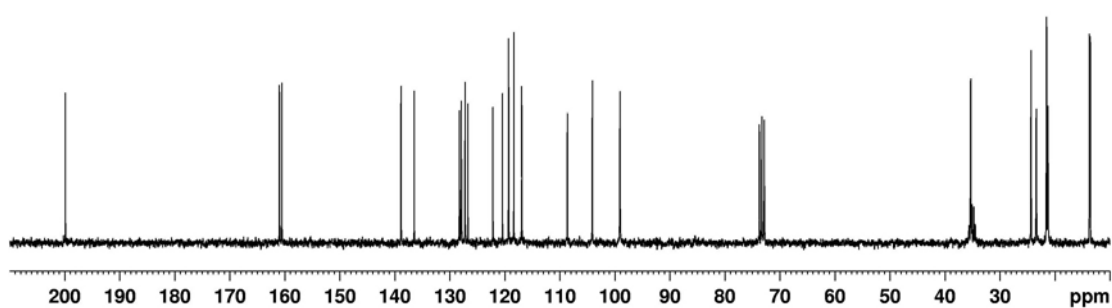


Figure A-26 $^{13}\text{C-NMR}$ (75 MHz) ($\text{CDCl}_3+\text{DMSO-}d_6$) spectrum of **PK14**

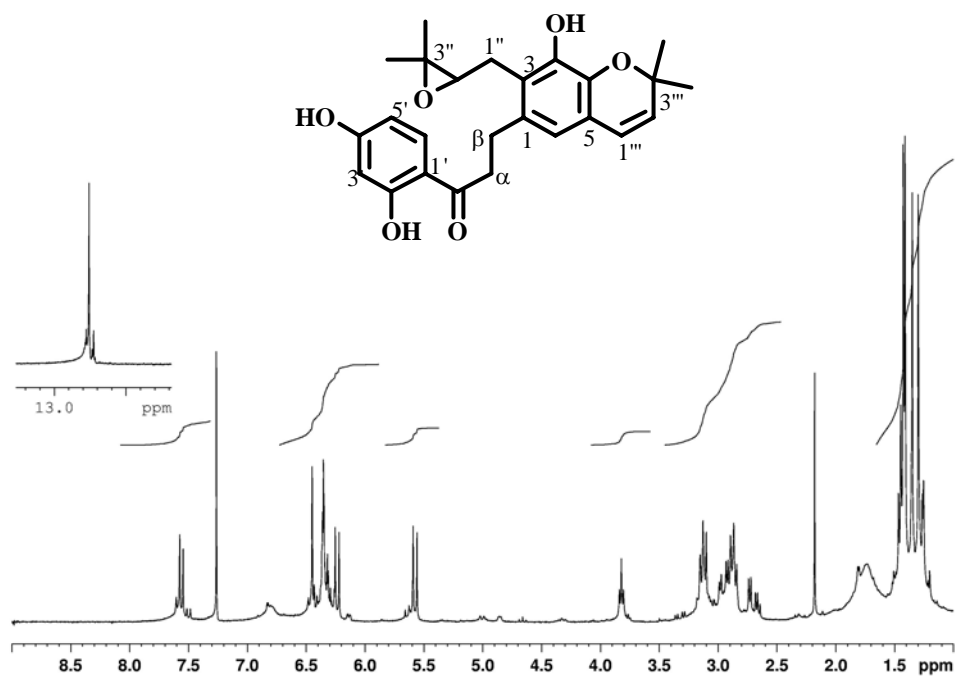


Figure A-27 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of **PK15**

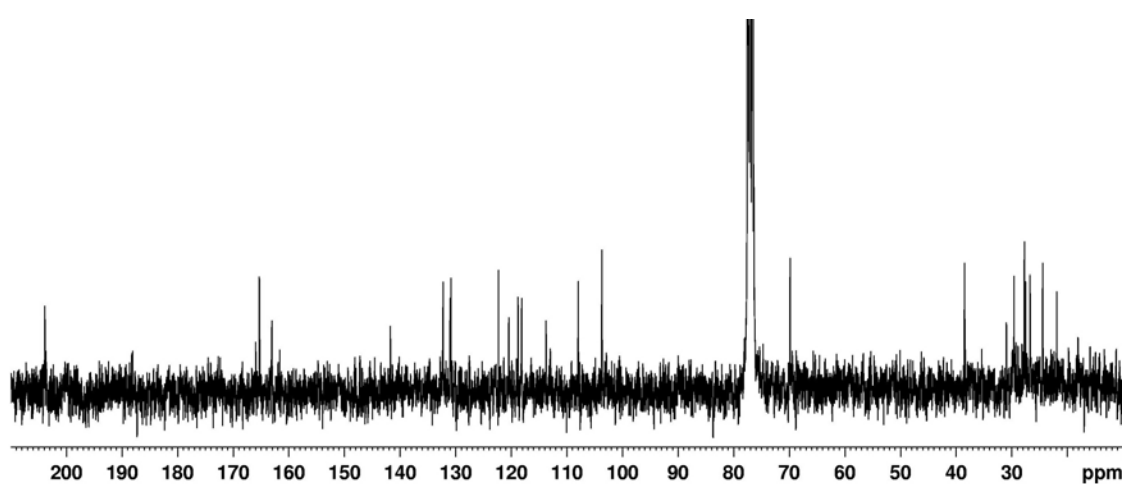


Figure A-28 $^{13}\text{C-NMR}$ (75 MHz) (CDCl_3) spectrum of **PK15**

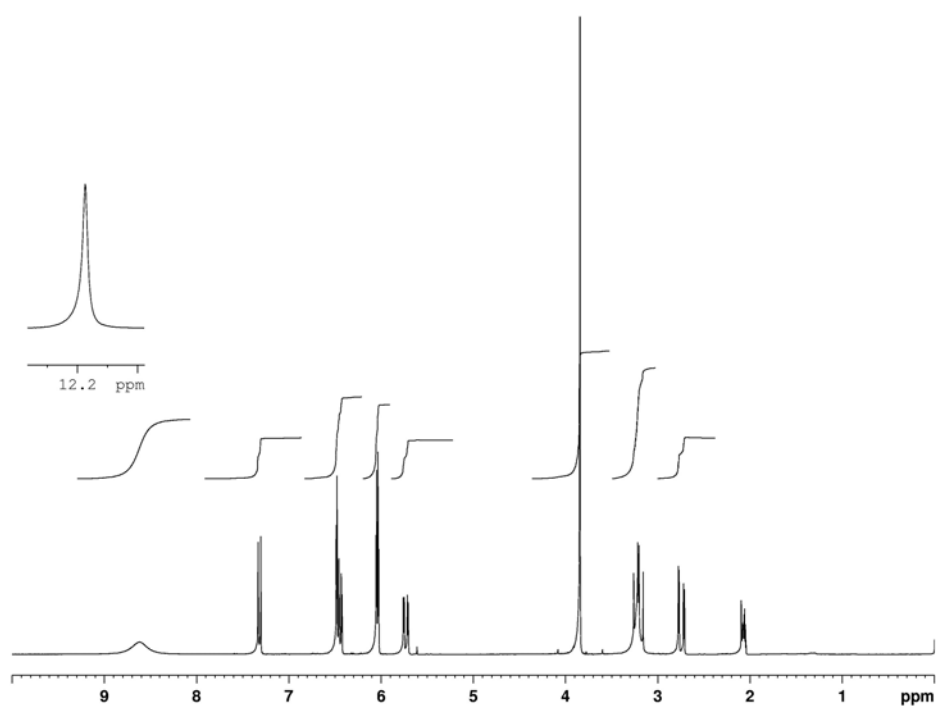


Figure A-29 $^1\text{H-NMR}$ (300 MHz) ($\text{acetone-}d_6$) spectrum of **PK16**

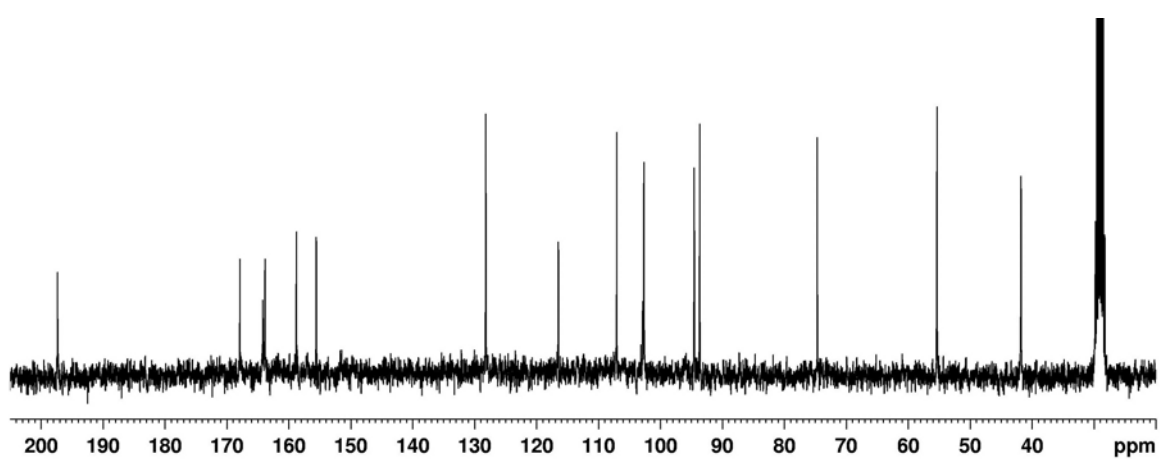


Figure A-30 $^{13}\text{C-NMR}$ (75 MHz) ($\text{acetone-}d_6$) spectrum of **PK16**

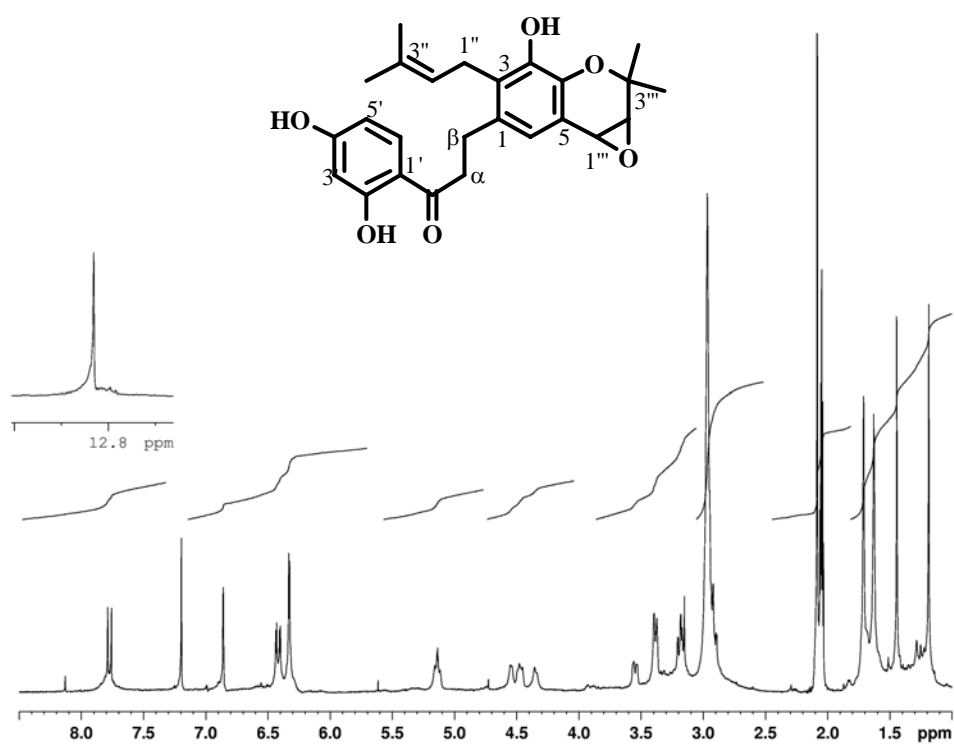


Figure A-31 $^1\text{H-NMR}$ (300 MHz) (acetone- d_6) spectrum of **PK17**

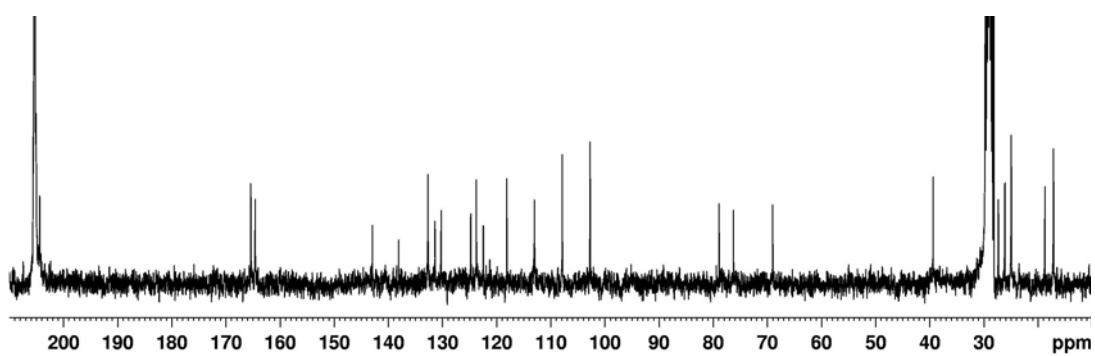


Figure A-32 $^{13}\text{C-NMR}$ (75 MHz) (acetone- d_6) spectrum of **PK17**

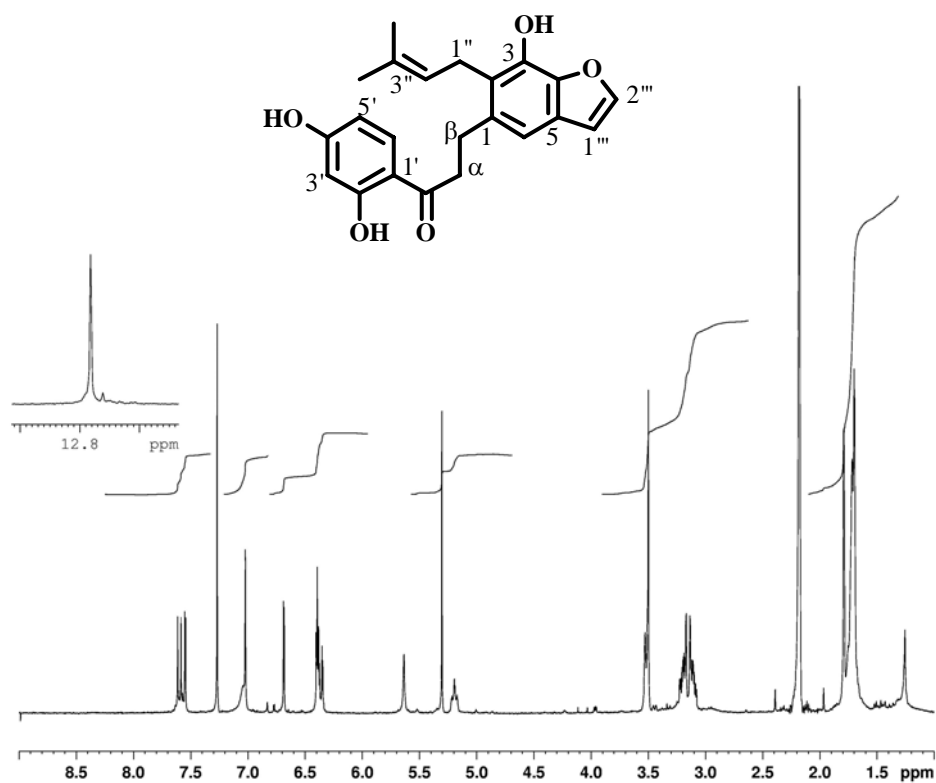


Figure A-33 $^1\text{H-NMR}$ (300 MHz) (CDCl_3) spectrum of **PK18**

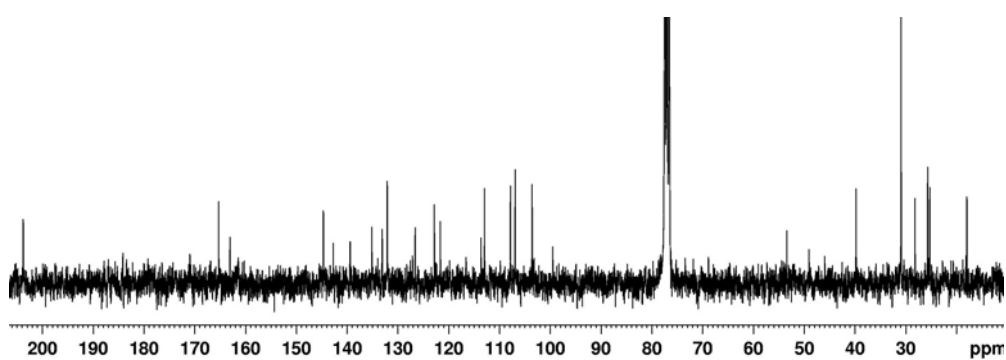


Figure A-34 $^{13}\text{C-NMR}$ (75 MHz) (CDCl_3) spectrum of **PK18**

2. ^{13}C -NMR and ^1H -NMR spectral data of known compounds from literatures

Table A-1 ^1H and ^{13}C NMR spectral data of (*E*)-4-(3',4'-dimethoxyphenyl)-3-butenyl acetate (Han *et al.*, 2003)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
1	4.18 (2H, <i>t</i> , 6.8)	64.1
2	2.53 (2H, <i>qd</i> , 6.8, 1.2)	32.5
3	6.03 (1H, <i>dt</i> , 16.0, 6.8)	123.8
4	6.41 (1H, <i>d</i> , 16.0)	132.3
1'	-	130.6
2'	6.91 (1H, <i>d</i> , 1.8)	108.8
3'	-	149.3
4'	-	148.8
5'	6.81 (1H, <i>d</i> , 8.0)	111.4
6'	6.88 (1H, <i>dd</i> , 8.0, 1.8)	119.3
3'-OCH ₃	3.90 (3H, <i>s</i>)	56.2
4'-OCH ₃	3.88 (3H, <i>s</i>)	56.0
1-OC(O)CH ₃	-	171.3
1-OC(O)CH ₃	2.06 (3H, <i>s</i>)	21.2

recorded in CDCl₃

Table A-2 ^1H and ^{13}C NMR spectral data of 4-hydroxybenzaldehyde (Jang *et al.*, 1990)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
1	-	129.9
2/6	7.82 (2H, <i>d</i> , 8.6)	132.5
3/5	6.98 (2H, <i>d</i> , 8.6)	116.0
4	-	161.6
7	9.86 (1H, <i>s</i>)	191.2

recorded in CDCl₃

Table A-3 ^1H and ^{13}C NMR spectral data of cycloheterophyllin (Wei *et al.*, 2005)

Position	δ_{H} (mult, J_{Hz})	δ_{C}	Position	δ_{H} (mult, J_{Hz})	δ_{C}
2	-	157.4	16	-	79.3
3	-	110.7	17	1.46 (3H, <i>s</i>)	28.9
4	-	180.1	18	1.48 (3H, <i>s</i>)	29.0
4a	-	106.5	1'	-	108.5
5	-	156.1	2'	-	153.1
6	-	106.7	3'	6.47 (1H, <i>s</i>)	106.1
7	-	158.7	4'	-	152.8
8	-	109.1	5'	-	142.1
8a	-	155.1	6'	7.29 (1H, <i>s</i>)	110.6
9	6.14 (1H, <i>d</i> , 9.4)	70.6	1''	3.49 (2H, <i>dd</i> , 7.2, 3.2)	22.8
10	5.51 (1H, <i>d</i> , 9.4)	123.8	2''	5.28 (1H, <i>m</i>)	122.8
11	-	139.2	3''	-	132.8
12	1.93 (3H, <i>s</i>)	19.3	4''	1.68 (3H, <i>s</i>)	26.2
13	1.68 (3H, <i>s</i>)	26.2	5''	1.87 (3H, <i>s</i>)	19.0
14	6.65 (1H, <i>d</i> , 10.0)	116.9	5-OH	13.25 (1H, <i>s</i>)	-
15	5.74 (1H, <i>d</i> , 10.0)	129.8			

recorded in CDCl_3

Table A-4 ^1H and ^{13}C NMR spectral data of (*E*)-4-(3',4'-dimethoxyphenyl)but-3-en-1-ol (Han *et al.*, 2003)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
1	3.75 (2H, <i>t</i> , 6.4)	62.1
2	2.47 (2H, <i>qd</i> , 6.4, 7.2)	36.4
3	6.07 (1H, <i>dt</i> , 15.6, 7.2)	124.4
4	6.43 (1H, <i>d</i> , 15.6)	132.5
1'	-	130.4
2'	6.92 (1H, <i>d</i> , 1.8)	108.6
3'	-	149.0
4'	-	148.6
5'	6.81 (1H, <i>d</i> , 8.2)	111.2
6'	6.89 (1H, <i>dd</i> , 8.2, 1.8)	119.1
3'-OCH ₃	3.90 (3H, <i>s</i>)	55.8
4'-OCH ₃	3.84 (3H, <i>s</i>)	55.9

recorded in CDCl₃

Table A-5 ^1H and ^{13}C NMR spectral data of cudraflavone B (Ryu *et al.*, 2009)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
2	-	159.7
3	-	121.4
4	-	182.3
4a	-	105.4
5	-	156.4
6	-	105.1
7	-	159.2
8	6.25 (1H, s)	94.7
8a	-	157.2
9	3.11 (2H, <i>brd</i> , 6.7)	24.3
10	5.11 (1H, <i>m</i>)	121.0
11	-	133.2
12	1.44 (3H, <i>s</i>)	17.7
13	1.52 (3H, <i>s</i>)	25.7
14	6.70 (1H, <i>d</i> , 10.0)	115.6
15	5.59 (1H, <i>d</i> , 10.0)	128.0
16	-	78.0
17	1.44 (3H, <i>s</i>)	28.3
18	1.44 (3H, <i>s</i>)	28.3
1'	-	112.5
2'	-	155.2
3'	6.49 (1H, <i>d</i> , 0.3)	103.8
4'	-	159.5
5'	6.50 (1H, <i>dd</i> , 8.9, 0.3)	108.4
6'	7.17 (1H, <i>d</i> , 8.9)	131.6
5-OH	13.10 (1H, <i>s</i>)	-

recorded in CDCl_3

Table A-6 ^1H and ^{13}C NMR spectral data of cycloartobiloxanthone (Sultanbawa *et al.*, 1989)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
2	-	162.8/163.1
3	-	102.1
4	-	181.9
4a	-	105.2
5	-	152.4
6	6.14 (1H, <i>s</i>)	102.2
7	-	162.8/163.1
8	-	105.2
8a	-	158.6
9	3.21 (1H $_{\beta}$, <i>dd</i> , 14.5, 7.0) 2.36 (1H $_{\alpha}$, <i>t</i> , 14.5)	20.5
10	3.43 (1H, <i>dd</i> , 14.0, 7.0)	47.7
11	-	93.9
12	1.34 (3H, <i>s</i>)	22.9
13	1.67 (3H, <i>s</i>)	29.1
14	6.92 (1H, <i>d</i> , 10.0)	116.3
15	5.64 (1H, <i>d</i> , 10.0)	128.2
16	-	78.9
17	1.47 (3H, <i>s</i>)	28.3
18	1.47 (3H, <i>s</i>)	28.3
1'	-	113.0
2'	-	151.9
3'	6.43 (1H, <i>s</i>)	106.8
4'	-	147.8
5'	-	138.6
6'	-	134.0
5-OH	13.33 (1H, <i>s</i>)	-
2'-OH	8.70 (1H, <i>s</i>)	-
4'-OH	8.85 (1H, <i>s</i>)	-

recorded in acetone d_6

Table A-7 ^1H and ^{13}C NMR spectral data of artelastoxanthone (Ko *et al.*, 2005)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
2	-	161.9
3	-	112.7
4	-	181.6
4a	-	105.6
5	-	163.8
6	6.28 (1H, <i>d</i> , 2.4)	100.4
7	-	165.1
8	6.58 (1H, <i>d</i> , 2.4)	95.6
8a	-	158.1
9	3.38 (1H $_{\beta}$, <i>dd</i> , 16.0, 2.0) 2.45 (1H $_{\alpha}$, <i>dd</i> , 16.0, 6.4)	22.8
10	3.98 (1H, <i>d</i> , 6.4)	38.4
11	-	145.9
12	4.64 (1H $_{\beta}$, <i>s</i>) 4.31 (1H $_{\alpha}$, <i>s</i>)	112.5
13	1.77 (3H, <i>s</i>)	22.6
14	6.76 (1H, <i>d</i> , 10.0)	117.9
15	5.75 (1H, <i>d</i> , 10.0)	130.4
16	-	78.9
17	1.47 (3H, <i>s</i>)	28.7
18	1.45 (3H, <i>s</i>)	28.7
1'	-	107.8
2'	-	146.1
3'	-	111.1
4'	-	146.0
5'	-	137.9
6'	-	129.2

Table A-7 (continued)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
5-OH	13.17 (1H, <i>s</i>)	-
7-OH	9.63 (1H, <i>s</i>)	-
2'-OH	8.02 (1H, <i>s</i>)	-
5'-OH	7.58 (1H, <i>s</i>)	-

recorded in CDCl_3

Table A-8 ^{13}C NMR spectral data of *trans*-feluric acid (Kelley *et al.*, 1976)

Position	δ_{C}
1	175.8
2	121.1
3	141.3
1'	127.7
2'	110.5
3'	147.1
4'	146.4
5'	115.3
6'	121.9
3'-OCH ₃	55.6
4-OH	-

recorded in acetone d_6 -D₂O (9:1)

Table A-9 ^1H and ^{13}C NMR spectral data of artobiloxanthone (Jayasinghe *et al.*, 2008)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
2	-	161.7
3	-	110.8
4	-	180.1
4a	-	104.8
5	-	159.5
6	6.9 (1H, <i>s</i>)	100.6
7	-	159.2
8	-	100.4
8a	-	151.0
9	3.36 (1H $_{\beta}$, <i>dd</i> , 16.6, 1.7) 2.63 (1H $_{\alpha}$, <i>dd</i> , 16.6, 7.8)	21.7
10	3.86 (1H, <i>br d</i> , 7.0)	38.1
11	-	149.8
12	4.80 (1H $_{\alpha}$, <i>br s</i>) 4.51 (1H $_{\alpha}$, <i>br s</i>)	112.8
13	1.79 (3H, <i>s</i>)	20.9
14	6.54 (1H, <i>d</i> , 10.2)	113.9
15	5.64 (1H, <i>d</i> , 10.2)	128.7
16	-	77.9
17	1.46 (3H, <i>s</i>)	27.9
18	1.48 (3H, <i>s</i>)	28.1
1'	-	105.2
2'	-	150.4
3'	6.51 (1H, <i>s</i>)	103.0
4'	-	144.8
5'	-	134.7
6'	-	127.7
5-OH	13.01 (1H, <i>s</i>)	-

recorded in CDCl_3

Table A-10 ^1H and ^{13}C NMR spectral data of artonin E (Jayasinghe *et al.*, 2008)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
2	-	163.2
3	-	122.0
4	-	183.9
4a	-	105.9
5	-	162.7
6	6.14 (1H, <i>s</i>)	100.1
7	-	160.5
8	-	102.2
8a	-	153.8
9	3.11 (2H, <i>br d</i> , 7.0)	24.9
10	5.10 (1H, <i>m</i>)	122.6
11	-	133.0
12	1.59 (3H, <i>br s</i>)	17.6
13	1.41(3H, <i>br s</i>)	25.9
14	6.61 (1H, <i>d</i> , 10.0)	115.8
15	5.59 (1H, <i>d</i> , 10.0)	128.2
16	-	79.1
17	1.43 (3H, <i>s</i>)	28.4
18	1.43 (3H, <i>s</i>)	28.4
1'	-	111.7
2'	-	150.1
3'	6.45 (1H, <i>s</i>)	104.7
4'	-	150.0
5'	-	-
6'	6.69 (1H, <i>s</i>)	-

recorded in CD_3OD

Table A-11 ^1H and ^{13}C NMR spectral data of artocarpanone (Wei *et al.*, 2005)

Position	δ_{H} (<i>mult</i> , J_{Hz})	δ_{C}
2	5.73 (1H, <i>dd</i> , 14.0, 3.0)	75.9
3	3.21 (1H, <i>dd</i> , 17.0, 14.0) 2.74 (1H, <i>dd</i> , 17.0, 3.0)	43.0
4	-	198.5
4a	-	104.1
5	-	160.0
6	6.02 (1H, <i>d</i> , 2.2)	95.7
7	-	169.1
8	6.05 (1H, <i>d</i> , 2.2)	94.8
8a	-	165.1
1'	-	117.7
2'	-	156.7
3'	6.47 (1H, <i>d</i> , 2.0)	103.9
4'	-	165.4
5'	6.43 (1H, <i>dd</i> , 8.0, 2.0)	108.3
6'	7.32 (1H, <i>d</i> , 8.0)	129.4
5-OH	12.17 (1H, <i>s</i>)	-
7-OMe	3.85 (3H, <i>s</i>)	56.6

recorded in acetone- d_6

VITAE

Name Mr. Prakit Chaithada

Student ID 5110220115

Educational Attainment

Degree	Name of Institution	Year of Graduation
B.SC. (Chemistry)	Chulalongkorn University	2004

Scholarship Awards during Enrolment

Center for Innovation in Chemistry (PERCH-CIC), Commission on Higher Education, Ministry of Education

Publication and Proceedings

Prakit Chaithada and Wilawan Mahabusarakam. "Prenylated flavones, phenylbutenoids and phenylpropanoids from the root bark of *Artocarpus elasticus*" The 1st Current Drug Development International Conference, Woraburi Phuket Resort & Spa, Phuket, Thailand, May 6-8, 2010. (Poster Presentation)