

## 2 EXPERIMENTAL

### 2.1 Instruments and Chemicals

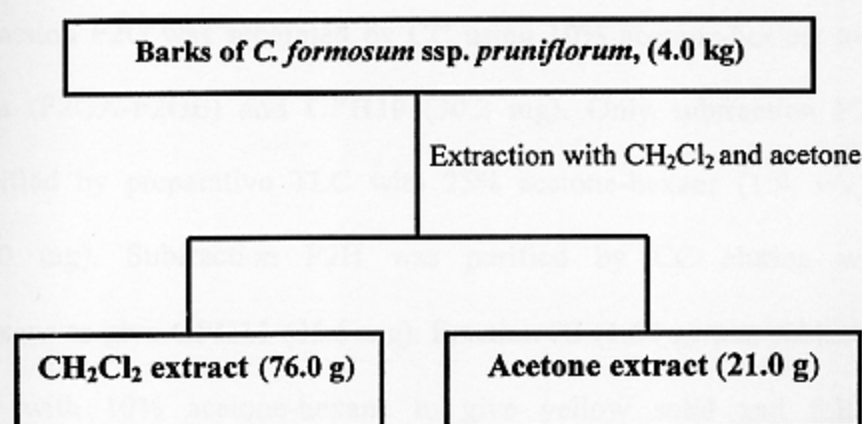
Melting points were determined on the Fisher-John melting point apparatus. Optical rotations were measured on a JASCO P-1020 digital polarimeter. UV-Vis and FT-IR spectra were recorded on SPECORD S 100 (Analytikjena) and Perkin-Elmer FTS FT-IR spectrophotometer, respectively. Single crystal X-ray diffraction measurements were collected using SMART 1-K CCD diffractometer with monochromated Mo-K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ) using  $\omega$ -scan mode and SHELXTL for structure solution and refinement. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded on a 500 MHz Varian UNITY INOVA and/or 300 MHz Bruker FT-NMR Ultra Shield<sup>TM</sup> spectrometers in  $\text{CDCl}_3$  or  $\text{CD}_3\text{OD}$  with TMS as the internal standard. Chemical shifts are reported in  $\delta$  (ppm) and coupling constants ( $J$ ) are expressed in hertz. EI and HREI mass spectra were measured on a Kratos MS 25 RFA spectrometer. Solvents for extraction and chromatography were distilled at their boiling point ranges prior to use except diethyl ether which was analytical grade reagent. Quick column chromatography (QCC) and column chromatography (CC) were carried out on silica gel 60 F<sub>254</sub> (Merck) and silica gel 100 (Merck), respectively.

## 2.2 Plant material

Barks and roots of *C. formosum* ssp. *pruniflorum* were collected in May 2004 from Nhong Khai Province in the northeastern part of Thailand. Identification was made by Prof. Puangpen Sirirugsa, Department of Biology, Faculty of Science, Prince of Songkla University and a specimen (No. 0012677) was deposited at Prince of Songkla University Herbarium.

## 2.3 Extraction

Ground-dried barks (4.0 kg) were extracted with  $\text{CH}_2\text{Cl}_2$  and acetone successively at room temperature (each  $2 \times 20$  L, for 5 days). The crude extracts were evaporated under reduced pressure to afford brownish crude  $\text{CH}_2\text{Cl}_2$  (76.0 g) and acetone (21.0 g) extracts. (see **Scheme 1**.)



**Scheme 1** Extraction of the barks of *C. formosum* ssp. *pruniflorum*

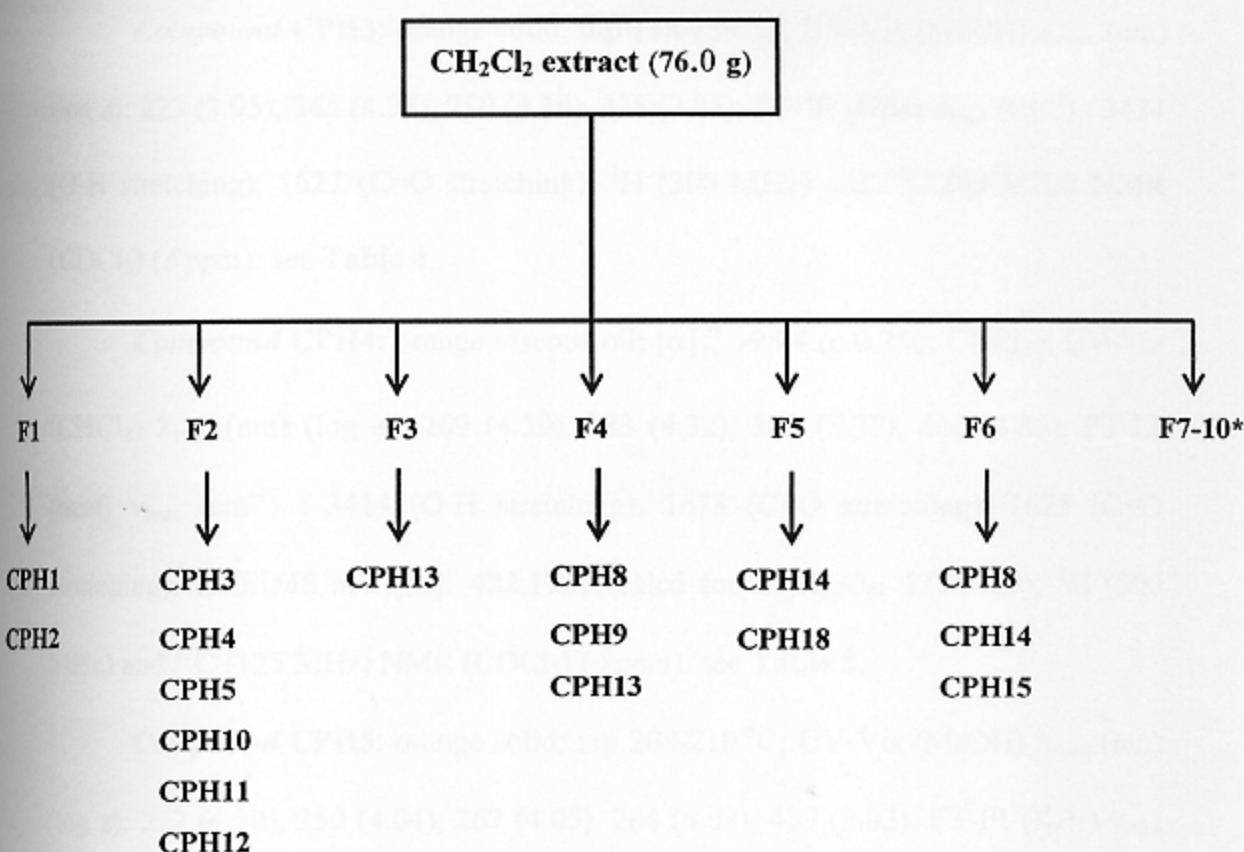
## 2.4 Isolation and Chemical Investigation

### 2.4.1. From the $\text{CH}_2\text{Cl}_2$ extract of the barks of *C. formosum* ssp.

#### *pruniflorum*

The crude  $\text{CH}_2\text{Cl}_2$  extract was subjected to QCC eluting with increasing polarities of EtOAc and acetone in hexane to afford 10 fractions (F1-F10). Fraction F1 (2.01 g) was separated by CC with 5% acetone-hexane to afford 3 subfractions (F1A-F1C). Subfraction F1B was further purified by CC with 10% EtOAc-hexane to give **CPH1** (3.0 mg) and **CPH2** (2.3 mg). Fraction F2 (58.06 g) was further separated by CC using a gradient of hexane with EtOAc to afford 10 subfractions (F2A-F2J). Subfraction F2C was separated by CC with 20% EtOAc-hexane to yield **CPH3** (68.2 mg) and **CPH4** (5.2 mg). Subfraction F2D was subjected to CC eluting with 60%  $\text{CH}_2\text{Cl}_2$ -hexane to give 3 fractions (F2DA-F2DC). Only subfraction F2DB was further purified by preparative TLC with 30%  $\text{CH}_2\text{Cl}_2$ -hexane to give **CPH12** (1.5 mg). Subfraction F2G was separated by CC using 10% acetone-hexane to afford 5 subfractions (F2GA-F2GE) and **CPH10** (30.2 mg). Only subfraction F2GB was further purified by preparative TLC with 25% acetone-hexane (1:9, v/v) to give **CPH5** (5.0 mg). Subfraction F2H was purified by CC eluting with 40%  $\text{CH}_2\text{Cl}_2$ -hexane to give **CPH11** (35.5 mg). Fraction F3 (1.54 g) was subjected to CC and eluted with 10% acetone-hexane to give yellow solid and followed by recrystallization from  $\text{CH}_3\text{OH}-\text{CHCl}_3$  (1:9, v/v) to yield **CPH13** (122.3 mg). Fraction F4 (2.02 g) was subjected to CC using 30% EtOAc-hexane to give 7 subfractions (F4A-F4G), **CPH13** (27.7 mg) and **CPH8** (2.4 mg). Subfraction F4F was further

purified by preparative TLC using 15% EtOAc-hexane to give **CPH9** (15.2 mg). Fraction F5 was subjected to CC and eluted with 10% acetone-hexane to afford 5 subfractions (F5A-F5E). Subfraction F5B was separated by CC using 20% acetone-hexane to give **CPH18** (4.5 mg). Subfraction F5D was separated by CC with 15% acetone-hexane to give **CPH14** (14.2 mg). Fraction F6 was subjected to CC with 15% acetone-hexane to afford 7 subfractions (F6A-F6G). Subfraction F6B was purified by CC with 30% EtOAc-hexane to yield **CPH15** (8.0 mg). Subfraction F6C was separated by CC with 15% acetone-hexane to afford 3 subfractions (F6CA-F6CC). Subfraction F6CB was further purified by preparative TLC eluting with 15% acetone-hexane to give **CPH8** (5.0 mg). Subfraction F6F was separated by CC 20% acetone-hexane to give **CPH14** (10.8 mg). (see **Scheme 2**.)



\* No further investigation

**Scheme 2** Isolation of compounds **CPH1**, **CPH2**, **CPH3**, **CPH4**, **CPH5**, **CPH8**,  
**CPH9**, **CPH10**, **CPH11**, **CPH12**, **CPH13**, **CPH14**, **CPH15** and **CPH18**

**Compound CPH1**: red-orange solid; mp 157-159 °C; UV-Vis (MeOH)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 221 (4.30), 253 (4.04), 265 (4.04), 286 (4.02), 480 (3.82); FT-IR (KBr)  $\nu_{\max}$  (cm<sup>-1</sup>): 3338 (O-H stretching), 1628 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$ ppm): see **Table 2**.

**Compound CPH2**: red-orange solid; mp 201-203 °C; UV-Vis (MeOH)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 220 (3.59), 278 (3.39), 425 (3.04); FT-IR (KBr)  $\nu_{\max}$  (cm<sup>-1</sup>): 3425 (O-H stretching), 1624 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$ ppm): see **Table 3**.



**Compound CPH3:** orange solid; mp 118-119 °C; UV-Vis (MeOH)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 223 (3.95), 265 (4.37), 280 (4.30), 438 (3.85); FT-IR (KBr)  $\nu_{\max}$  (cm<sup>-1</sup>): 3424 (O-H stretching), 1627 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 4**.

**Compound CPH4:** orange viscous oil;  $[\alpha]_D^{27}$  -98.4 (*c* 0.250, CHCl<sub>3</sub>); UV-Vis (CHCl<sub>3</sub>)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 269 (4.39), 283 (4.32), 366 (3.37), 440 (3.86); FT-IR (neat)  $\nu_{\max}$  (cm<sup>-1</sup>): 3414 (O-H stretching), 1673 (C=O stretching), 1625 (C=O stretching); HREIMS *m/z* [M]<sup>+</sup> 422.1737 (calcd for C<sub>25</sub>H<sub>26</sub>O<sub>6</sub>, 422.1729); <sup>1</sup>H (500 MHz) and <sup>13</sup>C (125 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 5**.

**Compound CPH5:** orange solid; mp 208-210 °C; UV-Vis (MeOH)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 222 (4.30), 250 (4.04), 267 (4.05), 284 (4.02), 437 (3.83); FT-IR (KBr)  $\nu_{\max}$  (cm<sup>-1</sup>): 3350 (O-H stretching), 1646 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 6**.

**Compound CPH8:** greenish brown viscous oil; UV-Vis (CHCl<sub>3</sub>)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 242 (3.48), 274 (3.65), 405 (2.97); FT-IR (KBr)  $\nu_{\max}$  (cm<sup>-1</sup>): 3414 (O-H stretching), 1632 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 9**.

**Compound CPH9:** greenish brown viscous oil; UV-Vis (CHCl<sub>3</sub>)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 241 (4.08), 278 (4.42), 318 (3.68), 335 (3.59), 405 (3.70); FT-IR (KBr)  $\nu_{\max}$  (cm<sup>-1</sup>): 3400 (O-H stretching), 1632 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 10**.

**Compound CPH10:** yellow-green solid; mp 114-116 °C;  $[\alpha]_D^{27}$  0 (*c* 2.1650, CHCl<sub>3</sub>); UV-Vis (MeOH)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 207 (4.66), 279 (4.17), 361 (4.29); FT-IR

(neat)  $\nu_{\max}$  ( $\text{cm}^{-1}$ ): 3446 (O-H stretching), 1613 (C=O stretching); HREIMS  $m/z$   $[\text{M}]^+$  782.3797 (calcd for  $\text{C}_{50}\text{H}_{54}\text{O}_8$ , 782.3819), EIMS  $m/z$  406 (35)  $[\text{M}]^+$ , 392 (2), 352 (35), 309 (100), 270 (79), 56 (69);  $^1\text{H}$  (300 MHz) and  $^{13}\text{C}$  (75 MHz) NMR ( $\text{CDCl}_3$ ) ( $\delta$  ppm): see **Table 11**.

**Compound CPH11**: yellow-green solid; mp 208-209 °C;  $[\alpha]_{\text{D}}^{27}$  0 ( $c$  0.6500,  $\text{CHCl}_3$ ); UV-Vis (MeOH)  $\lambda_{\max}$  (nm) ( $\log \epsilon$ ): 210 (3.90), 219 (3.91), 273 (3.91), 360 (3.59); FT-IR (KBr)  $\nu_{\max}$  ( $\text{cm}^{-1}$ ): 3446 (O-H stretching), 1631 (C=O stretching); HREIMS  $m/z$   $[\text{M}]^+$  674.2864 (calcd for  $\text{C}_{42}\text{H}_{42}\text{O}_8$ , 674.2880), EIMS  $m/z$  338 (50)  $[\text{M}]^+$ , 295 (100), 283 (55);  $^1\text{H}$  (300 MHz) and  $^{13}\text{C}$  (75 MHz) NMR ( $\text{CDCl}_3$ ) ( $\delta$  ppm): see **Table 12**.

**Compound CPH12**: yellow solid; mp 144-146 °C; UV-Vis (MeOH)  $\lambda_{\max}$  (nm) ( $\log \epsilon$ ): 205 (4.26), 223 (4.18), 253 (4.35), 327 (3.88), 369 (3.46); FT-IR (neat)  $\nu_{\max}$  ( $\text{cm}^{-1}$ ): 3424 (O-H stretching), 1642 (C=O stretching);  $^1\text{H}$  (300 MHz) and  $^{13}\text{C}$  (125 MHz) NMR ( $\text{CDCl}_3$ ) ( $\delta$  ppm): see **Table 13**.

**Compound CPH13**: brown-yellow solid; mp 183-184 °C; UV-Vis (MeOH)  $\lambda_{\max}$  (nm) ( $\log \epsilon$ ): 240 (4.28), 283 (4.62), 338 (4.25); FT-IR (KBr)  $\nu_{\max}$  ( $\text{cm}^{-1}$ ): 3446 (O-H stretching), 1649 (C=O stretching);  $^1\text{H}$  (300 MHz) and  $^{13}\text{C}$  (75 MHz) NMR ( $\text{CDCl}_3$ ) ( $\delta$  ppm): see **Table 14**.

**Compound CPH14**: yellow solid; mp 180-181 °C; UV-Vis (MeOH)  $\lambda_{\max}$  (nm) ( $\log \epsilon$ ): 203 (4.26), 253 (4.42), 287 (3.92), 328 (4.09); FT-IR (KBr)  $\nu_{\max}$  ( $\text{cm}^{-1}$ ): 3380 (O-H stretching), 1621 (C=O stretching);  $^1\text{H}$  (300 MHz) and  $^{13}\text{C}$  (75 MHz) NMR ( $\text{CDCl}_3$ ) ( $\delta$  ppm): see **Table 15**.

**Compound CPH15:** yellow solid; mp 218-219 °C; UV-Vis (CHCl<sub>3</sub>)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 245 (4.00), 297 (3.76), 337 (3.42); FT-IR (KBr)  $\nu_{\max}$  (cm<sup>-1</sup>): 3345 (O-H stretching), 1635 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 16**.

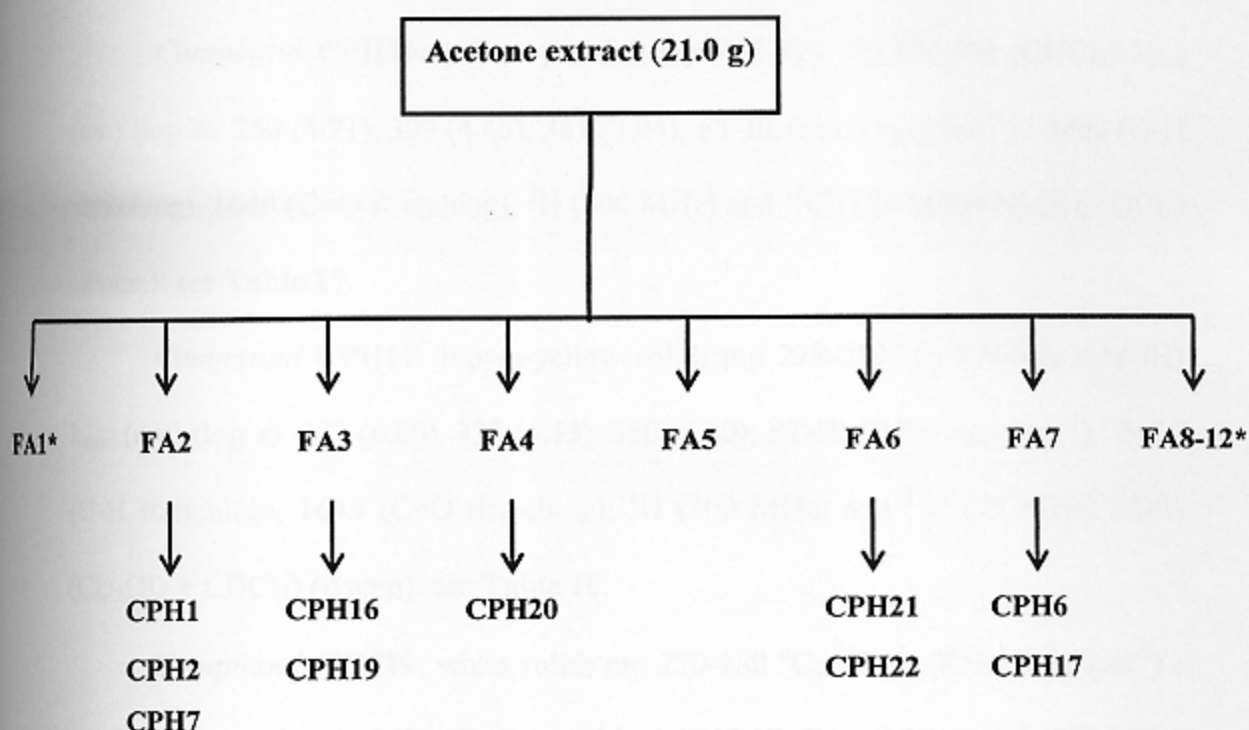
**Compound CPH18:** white solid; mp 193-194 °C; FT-IR (KBr)  $\nu_{\max}$  (cm<sup>-1</sup>): 3416 (O-H stretching), 1638 (C=C stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 19**.

#### 2.4.2. From the acetone extract of the barks of *C. formosum* ssp.

##### *pruniflorum*

The acetone extract was subjected to QCC eluting with a gradient of hexane-acetone to afford 12 fractions (FA1-FA12). Fraction FA2 (1.98 g) was subjected to CC with 3% acetone-hexane to afford 6 subfractions (FA2A-FA2F). Subfraction FA2B was further separated by CC with 5% EtOAc-hexane to give 4 subfractions (FA2B1-FA2B4) and **CPH7** (3.0 mg). Only subfraction FA2B1 was further purified by CC with 5% acetone-hexane to yield **CPH1** (2.6 mg) and **CPH2** (1.0 mg). Fraction FA3 was separated by CC and eluted with 10% EtOAc-hexane to give **CPH16** (4.0 mg) and **CPH19** (15.7 mg). Fraction FA4 was recrystallized from CH<sub>3</sub>OH-CHCl<sub>3</sub> (1:9, v/v) to yield **CPH20** (61.5 mg). Fraction FA6 was recrystallized from CH<sub>3</sub>OH-CHCl<sub>3</sub> (1:9, v/v) to give a mixture of **CHP21** and **CPH22** (34.2 mg). Fraction FA7 was subjected to CC eluting with 20% acetone-hexane to give **CPH6** (3.1 mg) and **CPH17** (5.0 mg). (see **Scheme 3**.)





\* No further investigation

**Scheme 3** Isolation of compounds **CPH1**, **CPH2**, **CPH6**, **CPH7**, **CPH16**, **CPH17**, **CPH19**, **CPH20**, **CPH21**, **CPH22**

**Compound CPH6**: orange solid; mp 254-256 °C; UV-Vis (MeOH)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 220 (4.29), 251 (4.04), 266 (4.05), 285 (4.02), 441 (3.85); FT-IR (neat)  $\nu_{\max}$  ( $\text{cm}^{-1}$ ): 3400 (O-H stretching), 1642 (C=O stretching);  $^1\text{H}$  (300 MHz) and  $^{13}\text{C}$  (75 MHz) NMR (DMSO) ( $\delta$  ppm): see **Table 7**.

**Compound CPH7**: orange solid; mp 224-226 °C; UV-Vis (MeOH)  $\lambda_{\max}$  (nm) (log  $\epsilon$ ): 208 (3.05), 224 (3.59), 265 (3.37), 285 (3.39), 424 (3.04); FT-IR (KBr)  $\nu_{\max}$  ( $\text{cm}^{-1}$ ): 3446 (O-H stretching), 1646 (C=O stretching);  $^1\text{H}$  (300 MHz) and  $^{13}\text{C}$  (125 MHz) NMR ( $\text{CDCl}_3$ ) ( $\delta$  ppm): see **Table 8**.

**Compound CPH16:** yellow powder; mp 212-214 °C; UV-Vis (CHCl<sub>3</sub>)  $\lambda_{\text{max}}$  (nm) (log  $\epsilon$ ): 260 (4.71), 309 (4.45), 381 (3.94); FT-IR (neat)  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3400 (O-H stretching), 1640 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (125 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 17**.

**Compound CPH17:** brown-yellow solid; mp 228-229 °C; UV-Vis (MeOH)  $\lambda_{\text{max}}$  (nm) (log  $\epsilon$ ): 282 (4.80), 338 (4.53), 380 (4.30); FT-IR (KBr)  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3415 (O-H stretching), 1646 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CD<sub>3</sub>OD + CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 18**.

**Compound CPH19:** white solid; mp 279-280 °C; FT-IR (KBr)  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3415 (O-H stretching), 1686 (C=O stretching), 1645 (C=C stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 20**.

**Compound CPH20:** colorless crystal; mp 245-247 °C; FT-IR (KBr)  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 1715 (C=O stretching); <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm): see **Table 21**.

**Compounds CPH21 and CPH22:** colorless crystal; <sup>1</sup>H (300 MHz) and <sup>13</sup>C (75 MHz) NMR (CDCl<sub>3</sub>) ( $\delta$  ppm).

## 2.5 Bioassays

### 2.5.1 Antibacterial assay

The isolated compounds from the barks of *C. formosum* ssp. *pruniflorum* were tested for antibacterial activities against, *Bacillus subtilis* (obtained from Department

of Industrial Biotechnology, Faculty of Agroindustrial, PSU), *Staphylococcus aureus* TISTR517 (obtained from Microbial Resources Center (MIRCEN), Bangkok, Thailand), *Streptococcus faecalis*, *Salmonella typhi*, *Shigella sonnei* and *Pseudomonas aeruginosa*. The last four microorganism were obtained from Department of Pharmacognosy and Botany, Faculty of Pharmacy, PSU. The antibacterial assay employed was the same as described in Boonsri *et al.* (Boonsri *et al.*, 2006). Vancomycin which was used as a standard showed antibacterial activity of 75 µg/mL.

### 2.5.2 Cytotoxic assay

The procedure for cytotoxic assay was performed by the sulphorhodamine B (SRB) assay as described by Skehan *et al.* (Skehan *et al.*, 1990). In this study, four cancer cell lines obtained from National Cancer Institute, Bangkok, Thailand, were used including MCF-7 (breast adenocarcinoma), KB (human oral cancer), HeLa (Human cervical cancer) and HT-29 (colon cancer). Camptothecin which was used as a standard showed cytotoxic activity in the range of 0.2-2.0 µg/mL.