3 RESULTS AND DISCUSSION

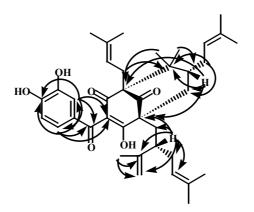
3.1 Structural determination of compounds isolated from the green fruits, ripe fruits, flowers and seeds of G. dulcis

The green fruits, ripe fruits and flowers of *Garcinia dulcis* Kurz. collected from Songkhla province in the southern part of Thailand. The extracts obtained from the green fruits, ripe fruits, flowers and seeds. Isolation and purification of acetone and methanolic extracts of the green fruits gave twenty compounds and six compounds, respectively. Twenty-five compounds were obtained from acetone extract of the ripe fruits. From the flowers, twenty-seven compounds were given from the acetone extract and three compounds were obtained from the methanolic extract. The dichloromethane and methanolic extracts of the seeds gave three and twelve compounds, respectively. The structures of all compounds were elucidated by analysis of spectroscopy.

GD1: (4R,6R,8R)-2-(3,4-dihydroxybenzoyl)-1-hydroxy-8-[(2S)-2-isopropenyl-5-methyl-4-hexenyl]-5,5-dimethyl-4,6-bis(3-methyl-2-butenyl)bicyclo[3.3.1] non-1-ene-3,9-dione (camboginol or garcinol)

GD1 is a yellow solid, m.p. 125-128 °C. Its molecular formula was $C_{38}H_{50}O_6$ as indicated by mass spectrum ([M]⁺ m/z 602). The optical rotation was [α]₀²⁹ -128.5 ° (c = 0.02 in CH₃OH). The ¹H NMR spectrum (**Table 24**) showed the resonances of aromatic protons H-15 (δ 6.61, d, J = 8.4 Hz), H-16 (δ 7.00, dd, J = 8.4, 2.0 Hz) and H-12 (δ 6.97, d, J = 2.0 Hz) which referred to the presence of 3,4-dihydroxybenzoyl group. The presence of three prenyl groups were observed. The signals at δ 5.10 (br t, H-18), 2.39 (br d, H-17), 1.68 (s, H-20), 1.57 (s, H-21) were due to a prenyl group at C-4. Whereas those at δ 4.95 (br t, H-25), 1.75 (s, H-28), 1.71 (s, H-27), 1.38 (br s, H-24) and δ 5.08 (br t, H-35), 1.89 (m, H-34), 1.82 (s, H-37), 1.56 (s, H-38) belonged to the prenyl group attached to CH-carbon. HMBC correlations showed the correlations of H-17 to C-5; H-24 to C-5, C-7 and H-34 to C-29, C-30, C-31 indicated that the prenyl groups were at C-4, C-6 and C-30, respectively. The spectrum further showed the signals of isopropenyl group at δ 4.43 (2H, br s, H-32) and 1.70 (3H, s, H-33) of which this group was located at C-30 according to the correlations of H-30 to C-31 and C-32. In addition, two methyl groups on a saturated carbon were observed at δ 1.05

(3H, s, H-22) and 1.61 (3H, s, H-23) and were placed at C-5 from the HMBC correlations of H-22 to C-4, C-5, C-6 and H-23 to C-22. Two groups of methylene protons were assigned from the resonances at δ 2.17 (1H, m, H-7), 1.90 (1H, m, H-7) and 2.74 (1H, m, H-29), 1.73 (1H, m, H-29) whereas two methine protons were indicated from the signals at δ 1.49 (1H, m, H-6) and 2.08 (1H, m, H-30). Four signals of carbonyl carbons were shown in the ¹³C NMR spectrum which corresponded to a non-enolisable carbonyl carbon (δ 209.17) and three enolisable carbonyl carbons (δ 194.01, 195.10 and 198.73). The remaining signals were assigned for eleven quaternary carbons, eight methine carbons, six methylene carbons and nine methyl carbons. The HMBC confirmed the above assignment, **GD1** was then identified as (4R,6R,8R)-2-(3,4-dihydroxybenzoyl)-1-hydroxy-8-[(2S)-2-isopropenyl-5-methyl-4-hexenyl]-5,5-dimethyl-4,6-bis(3-methyl-2-butenyl)bicyclo[3.3.1]non-1-ene-3,9-dione. The proposed structure, the spectral data, melting point and optical rotation were agreed with the structure of camboginol (Rama Rao, et al., 1980; Krishnamurthy, et al., 1981).



Major HMBC correlations of GD1

Table 24 The NMR spectral data of GD1

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-T	ype)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplica	$(ity, J_{\rm Hz})$	HMBC
Position	GD1	camboginol	GD1	camboginol	HWIDC
1	194.01 (C=O)	194.1	-	-	-
2	115.88 (C)	116.0	-	-	-
3	195.10 (C=O)	195.1	-	-	-
4	69.77 (C)	69.8	-	-	-
5	49.66 (C)	49.6	-	-	-
6	46.83 (CH)	47.0	1.49 (1H, m)	1.40-2.90 (m)	C-4, C-5, C-7, C-8,
					C-25
7	28.94 (CH ₂)	29.0	1.90 (1H, m)	1.40-2.90 (m)	C-24, C-29
			2.17 (1H, m)		C-24, C-29
8	57.96 (C)	58.0	-	-	-
9	209.17 (C=O)	209.1	-	-	-
10	198.73 (C=O)	198.9	-	-	-
11	127.70 (C)	127.8	-	-	-
12	116.44 (CH)	116.6	6.97 (1H, d, 2.0)	6.95 (d, 2.0)	C-2, C-10, C-14, C-16
13	143.70 (C)	143.8	-	-	-
14	149.83 (C)	150.0	-	-	-
15	114.37 (CH)	114.4	6.61 (1H, d, 8.4)	6.58 (d, 9.0)	C-11, C-13, C-14
16	124.03 (CH)	120.2	7.00 (1H, <i>dd</i> , 8.4, 2.0)	7.05 (q)	C-2, C-10, C-12, C-14
17	42.64 (CH ₂)	42.8 ^b	2.39 (2H, <i>br d</i>)	1.40-2.90 (m)	C-5, C-19
18	122.64 (CH)	122.7 ^a	5.10 (1H, <i>br t</i>)	4.90 (br t)	C-20, C-21
19	132.03 (C)	132.0	-	-	-
20	26.98 (CH ₃)	26.0°	1.68 (3H, s)	1.70	C-18, C-19, C-21
21	17.87 (CH ₃)	17.9 ^d	1.57 (3H, s)	1.56	C-18, C-19, C-20
22	27.01 (CH ₃)	26.5	1.05 (3H, s)	1.04 (s)	C-4, C-5, C-6
23	17.20 (CH ₃)	17.3	1.16 (3H, s)	1.16 (s)	C-6, C-22
24	32.61 (CH ₂)	32.7 ^b	1.38 (2H, <i>br s</i>)	1.40-2.90 (m)	C-5, C-7, C-26
25	123.76 (CH)	124.2ª	4.95 (1H, <i>br t</i>)	4.90 (br t)	C-27, C-28
26	132.98 (C)	132.9	-	-	-
27	26.03 (CH ₃)	25.8°	1.71 (3H, s)	1.70	C-25, C-26, C-28
28	18.18 (CH ₃)	18.2 ^d	1.75 (3H, s)	1.56	C-25, C-26, C-27

Table 24 (Continued)

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-T	Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)		IMDC	
Position	GD1	camboginol	GD1	camboginol	HMBC	
29	26.36 (CH ₂)	27.0 ^b	2.74 (1H, m)	1.40-2.90 (m)	C-1, C-8, C-9, C-30	
			1.73 (1H, m)		C-1, C-8, C-9, C-30	
30	43.57 (CH)	43.7	2.08 (1H, m)	1.40-2.90 (m)	C-8, C-29, C-31, C-32,	
					C-33, C-34, C-35	
31	148.06 (C)	148.1	-	-	-	
32	113.28 (CH ₂)	112.7	4.43 (2H, br s)	4.36 (br s)	C-31, C-33	
33	22.58 (CH ₃)	22.7	1.70 (3H, s)	1.70	C-30, C-31, C-32	
34	35.51 (CH ₂)	36.3 ^b	1.89 (2H, m)	1.70	C-29, C-30, C-31,	
					C-35, C-36	
35	120.02 (CH)	124.0°	5.08 (1H, br t)	4.9 (br t)	C-37, C-38	
36	135.34 (C)	135.3	-	-	-	
37	25.71 (CH ₃)	25.8°	1.82 (3H, s)	1.70	C-35, C-36, C-38	
38	17.64 (CH ₃)	17.7 ^d	1.56 (3H, s)	1.56	C-36, C-37	

^{*} Carbon type was deduced from DEPT experiments.

^a Olefinic methine carbons, ^b Methylene carbons, ^{c, d} Methyl carbons: Assignment maybe interchangeable.

GD2: Octadecanoic acid-2,3-dihydroxypropyl ester

$${^{3}_{CH_{2}-OH}}\atop{^{2}_{CH-OH}}\atop{^{1}_{CH_{2}-O-\overset{1'}{C}-CH_{2}-CH_{2}-(CH_{2})_{13}-CH_{2}-CH_{3}}}$$

GD2 is a white solid, m.p. 63-65 °C. The ¹H NMR spectrum (**Table 25**) exhibited two sharp triplet signals of H-2' at δ 2.35 (2H, J = 6.6 Hz) and H-18' at δ 0.88 (3H, J = 6.6 Hz), a quintet signal of H-3' at δ 1.63 (2H, J = 6.6 Hz) and a multiplet signal of H-4' - H-17' at δ 1.32-1.25 (28H). These assignments were supported by the HMBC correlations of H-2' to C-1', C-3', C-4'; H-3' to C-1', C-2', C-4'; H-17' to C-16', C-18' and H-18' to C-16', C-17'. The ¹H NMR spectra further showed two doublet of doublet signals of non equivalent methylene protons H-1 at δ 4.21 (1H, J = 11.2 and 4.2 Hz) and 4.15 (1H, J = 11.2 and 5.6 Hz), two doublet of doublet signals of non equivalent methylene protons H-3 at δ 3.70 (1H, J = 11.2 and 4.2 Hz) and 3.61 (1H, J = 11.2 and 5.6 Hz) and a multiplet signal of methine proton H-2 at δ 3.94. These signals were supported by the correlations of H-1 to C-2, C-3, C-1'; H-2 to C-1, C-3 and H-3 to C-1, C-2 in HMBC. The presence of hydroxy groups at C-2 and C-3 were confirmed by the appearance of two signals of acetyl protons (δ 2.05, 3H and 2.06, 3H) and the down field shifts of H-2 (δ 5.23, 1H, m) and H-3 (δ 4.27, 1H, dd, J = 8.4, 4.2 Hz and 4.29, 1H, dd, J = 8.4, 4.2 Hz) of its acetylated product (Table 26). Thus GD2 was assigned to be octadecanoic acid-2,3-dihydroxypropyl ester (Chupin, et al., 2001).

Major HMBC correlations of GD2

Structure of GD2(A)

Table 25 The NMR spectral data of GD2

Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	COSY	НМВС
1	65.16 (CH ₂)	4.21 (1H, dd, 11.2, 4.2)	H-2	C-2, C-3, C-1'
		4.15 (1H, dd, 11.2, 5.6)	H-2	C-2, C-3, C-1'
2	70.25 (CH)	3.94 (1H, m)	H-1, H-3	C-1, C-3
3	63.30 (CH ₂)	3.70 (1H, dd, 11.2, 4.2)	H-2	C-1, C-2
		3.61 (1H, dd, 11.2, 5.6)	H-2	C-1, C-2
1'	174.36 (C=O)	-	-	-
2'	34.14 (CH ₂)	2.35 (2H, t, 6.6)	H-3′	C-1', C-3', C-4'
3'	24.90 (CH ₂)	1.63 (2H, qn, 6.6)	H-2', H-4'	C-1', C-2', C-4'
4'	29.11 (CH ₂)		H-3'	C-16', C-(5'-15'), C-17'
	29.70 (CH ₂)			
	29.69 (CH ₂)			
	29.68 (CH ₂)			
	29.67 (CH ₂)			
	29.66 (CH ₂)			
5'-15'	29.64 (CH ₂)	\(\) 1.32-1.25 (28H, m)		
	29.63 (CH ₂)			
	29.58 (CH ₂)			
	29.43 (CH ₂)			
	29.35 (CH ₂)			
	29.23 (CH ₂)			
16'	31.91 (CH ₂)		H-17'	
17'	22.68 (CH ₂)		H-16',H-18'	C-16', C-(5'-15'), C-18'
18'	14.12 (CH ₃)	0.88 (3H, t, 6.6)	H-17′	C-16', C-17'
2-OH	-	2.46 (1H, d, 5.0)	-	-
3-ОН	-	2.02 (1H, br t, 5.8)		-

^{*} Carbon type was deduced from DEPT experiments.

Table 26 The ¹H NMR spectral data of GD2 and GD2(A)

Dogition	GD2	GD2(A)
Position	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)
1	4.21 (1H, dd, 11.2, 4.2)	4.15 (1H, dd, 7.8, 5.4)
	4.15 (1H, dd, 11.2, 5.6)	4.12 (1H, dd, 7.8, 5.4)
2	3.94 (1H, m)	5.23 (1H, m)
3	3.70 (1H, dd, 11.2, 4.2)	4.29 (1H, dd, 8.4, 4.2)
	3.61 (1H, dd, 11.2, 5.6)	4.27 (1H, dd, 8.4, 4.2)
2'	2.35 (2H, t, 6.6)	2.29 (2H, t, 7.0)
3'	1.63 (2H, qn, 6.6)	1.58 (2H, qn, 6.4)
4'-17'	1.32-1.25 (28H, m)	1.23 (28H, <i>m</i>)
18'	0.88 (3H, t, 6.6)	0.85 (3H, t, 6.4)
2-OCOCH ₃	-	2.05 (3H, s)*
3-OCOCH ₃	-	2.06 (3H, s)*

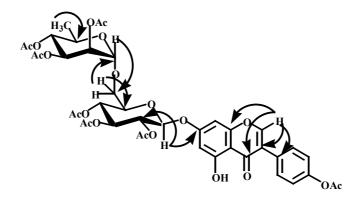
^{*} Assignment maybe interchangeable.

GD3 : 5-Hydroxy-4', 2'', 3'', 4'', 2''', 3''', 4'''-heptaacetateisoflavone 7-O-[α -rhamnopyranosyl-(1 \rightarrow 6)]- β -glucopyranoside

GD3 is yellow needles. EIMS showed the molecular ion of 872 which corresponded to $C_{41}H_{44}O_{21}$. The UV spectra (λ_{max} 323.0, 257.2 and 215.8 nm) indicated an isoflavone nucleus. IR absorption bands at 3410 cm⁻¹ and 1646 cm⁻¹ suggested the presence of hydroxyl group and carbonyl group, respectively.

The ¹H NMR spectrum (**Table 27**) revealed a *singlet* of a chelated hydroxyl group (5-OH) at δ 12.72 and a *singlet* of isoflavone type vinylic proton (H-2) at δ 7.97. The 4'-oxysubstituted pattern of B-ring was readily deduced from the proton signals forming an AA'BB' system at δ 7.58 (J = 8.3 Hz, H-2', H-6') and 7.19 (J = 8.3 Hz, H-3', H-5'). Two *doublets* at δ 6.42 and 6.55 with coupling constant of 2.5 Hz, represented the H-6 and H-8, respectively. The correlations of H-6 to C-4a, C-5, C-8 and H-8 to C-4, C-4a, C-6, C-7, C-8a confirmed the assignment of H-6 and H-8. A *doublet* at δ 5.18 (J = 6.8 Hz) was assigned for the anomeric proton H-1" of β -glucose unit. A *doublet* at δ 4.73 (J = 2.3 Hz) and a *doublet* at δ 1.18 (J = 5.0 Hz) were attributed to the anomeric proton H-1" and a methyl group of α -rhamnose unit, respectively. A (1 \rightarrow 6)-glycosidic linkage of rhamnose to glucose was proposed from the downfield shift of C-6". In addition, the linkage was confirmed by the HMBC correlation of the anomeric proton H-1" to C-6" (δ 66.27). The correlation of H-1"

to C-7 (δ 161.98) of the aglycone suggested that the glucose moiety was attached to the 7-hydroxyl of the aglycone. Moreover the sugar moieties were assigned to be acetyl derivatives according to the resonances of seven acetyl groups (δ 2.33, 2.09, 2.08, 2.07, 2.05 (6H) and 1.98). These data permitted the identification of **GD3** as 5-hydroxy-4', 2'', 3'', 4'', 2''', 3''', 4'''-heptaacetateisoflavone 7-O-[α -rhamnopyranosyl-(1 \rightarrow 6)]- β -glucopyranoside. This compound was a new naturally occurring but synthetically known isoflavone glycoside (Roesler, *et al.*, 1965).



Major HMBC correlations of GD3

 $Table\ 27\ \ The\ NMR\ spectral\ data\ of\ GD3$

Position	$\delta_{\!\scriptscriptstyle m C}^{}*$	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2	154.02 (CH)	7.97 (1H, s)	C-3, C-4, C-8a, C-1'
3	123.05 (C)	-	-
4	180.61 (C=O)	-	-
4a	107.53 (C)	-	-
5	162.58 (C)	-	-
6	100.51 (CH)	6.42 (1H, d, 2.5)	C-4a, C-5, C-8
7	161.98 (C)	-	-
8	94.67 (CH)	6.55 (1H, d, 2.5)	C-4, C-4a, C-6, C-7, C-8a
8a	157.59 (C)	-	-
1'	128.29 (C)	-	-
2', 6'	129.96 (CH)	7.58 (2H, d, 8.3)	C-3, C-2',6', C-3',5', C-4'
3', 5'	121.79 (CH)	7.19 (2H, d, 8.3)	C-1', C-3',5', C-4'
4'	150.75 (C)	-	-
Glc-1"	97.94 (CH)	5.18 (1H, d, 6.8)	C-7, C-2", C-3"
2''	73.57 (CH)	3.89 (1H, m)	C-3''
3''	69.18 (CH)	5.27 (1H, m)	C-1", C-4", 3"-OCOCH ₃
4''	70.81 (CH)	5.33 (1H, <i>t</i> , 8.0)	C-3", C-6"
5''	72.50 (CH)	5.31 (1H, <i>m</i>)	C-1", C-3", C-4", C-6"
6''	66.27 (CH ₂)	3.63 (2H, dd, 11.5, 6.9)	C-5", C-1""
Rha-1'''	98.20 (CH)	4.73 (1H, d, 2.3)	C-6", C-2"", C-3""
2'''	69.08 (CH)	5.27 (1H, m)	C-1''', C-4'''
3'''	68.70 (CH)	5.09 (1H, t, 8.0)	C-1''', C-4'''
4'''	70.74 (CH)	5.04 (1H, t, 8.0)	C-3''', C-6'''
5'''	66.61 (CH)	3.85 (1H, m)	C-3''', C-4'''
6'''	17.27 (CH ₃)	1.18 (3H, d, 5.0)	C-4''', C-5'''

Table 27 (Continued)

Position	$\delta_{_{\mathrm{C}}}*$	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
5-OH	-	12.72 (1H, s)	-
4'-OCOCH ₃	169.20 (C=O) ^a , 20.56 (CH ₃) ^b	1.98 (3H, s) ^c	C-4'
2"-OCOCH ₃	170.41 (C=O) ^a , 20.57 (CH ₃) ^b	$2.09 (3H, s)^{c}$	C-2''
3"- OCOCH ₃	169.90 (C=O) ^a , 20.73 (CH ₃) ^b	$2.05 (3H, s)^{c}$	C-3''
4"- OCOCH ₃	$169.88 \text{ (C=O)}^{a}, 20.74 \text{ (CH}_{3})^{b}$	2.07 (3H, s) ^c	C-4''
2""- OCOCH ₃	169.70 (C=O) ^a , 20.57 (CH ₃) ^b	$2.08 (3H, s)^{c}$	C-2'''
3'''- OCOCH ₃	$169.41 \text{ (C=O)}^{a}, 20.76 \text{ (CH}_{3})^{b}$	$2.05 (3H, s)^{c}$	C-3'''
4""- OCOCH ₃	169.38 (C=O) ^a , 21.11 (CH ₃) ^b	$2.33 (3H, s)^{c}$	C-4''

^{*} Carbon type was deduced from DEPT experiments.

^{a, b, c} Assignment maybe interchangeable.

GD4: 7-O-[α -Rhamnopyranosyl(1 \rightarrow 6)]- β -glucopyranoside-4'-methoxyisoflavone (derriscannoside A)

GD4 was isolated as brown needles. It is an isoflavone of which the characteristic resonance of the methine proton H-2 appeared as a singlet signal at δ 8.25. The AA'BB' resonances of aromatic protons H-2', H-6' and H-3', H-5' were displayed at δ 7.51 (d, J = 8.7 Hz) and 6.95 (d, J = 8.7 Hz), respectively. The signals of ABX type at δ 8.10 (d, J = 8.7 Hz), 7.11 (dd, J = 8.7 and 2.5 Hz) and 7.23 (d, J = 2.5 Hz) were proposed for the signals of H-5, H-6 and H-8, respectively. A singlet signal at δ 3.83 was assigned for 4'-OCH₂. The ¹H NMR spectrum (**Table 28**) further indicated the presence of two sugar moieties. A doublet signal at δ 4.96 with J = 7.0Hz was assigned for the anomeric proton of glucose (H-1"). A singlet resonance at δ 4.67 (H-1''') and a doublet resonance at δ 1.22 (J=5.6 Hz, H-6''') implied the presence of a rhamnose. The other proton signals of sugars appeared at δ 3.85-3.35. HMBC correlations of H-1 $^{\prime\prime}$ to C-7 (δ 161.30) and H-1 $^{\prime\prime\prime}$ to C-6 $^{\prime\prime}$ (δ 61.98) suggested that glucose was directly connected to isoflavone nucleus at C-7 and rhamnose was formed (1 \rightarrow 6) a glycosidic bond to glucose. The 13 C NMR and the DEPT spectra were shown in **Table 28**. The assignment suggested that **GD4** was 7-O-[α -rhamnopyranosyl $(1\rightarrow 6)$]- β -glucopyranoside-4'-methoxyisoflavone (derriscannoside A, Dianpeng, et al., 1999).

Major HMBC correlations of GD4

Table 28 The NMR spectral data of GD4

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2	153.21 (CH)	8.25 (1H, s)	C-4, C-8a, C-1'
3	123.65 (C)	-	-
4	175.14 (C=O)	-	-
4a	118.81 (C)	-	-
5	126.77 (CH)	8.10 (1H, d, 8.7)	C-4, C-7, C-8a
6	115.60 (CH)	7.11 (1H, dd, 8.7, 2.5)	C-4a, C-7
7	161.30 (C)	-	-
8	103.78 (CH)	7.23 (1H, d, 2.5)	C-4, C-4a, C-6, C-7, C-8a
8a	157.19 (C)	-	-
1'	124.09 (C)	-	-
2', 6'	129.84 (CH)	7.51 (2H, d, 8.7)	C-3, C-2',6', C-4'
3', 5'	113.45 (CH)	6.95 (2H, d, 8.7)	C-1', C-3',5', C-4'
4'	159.50 (C)	-	-
4'-OCH ₃	54.96 (CH ₃)	3.83 (3H, s)	C-4'

Table 28 (Continued)

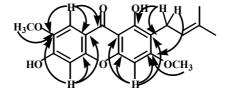
Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
Glc-1"	100.63 (CH)	4.96 (1H, d, 7.0)	C-7, C-3", C-5"
2''	83.24 (CH)	3.30-3.20 (1H, <i>m</i>)	C-3''
3''	76.97 (CH)	3.50-3.44 (1H, <i>m</i>)	C-2", C-4"
4''	73.02 (CH)	3.52-3.50 (1H, <i>m</i>)	C-3", C-5", C-6"
5''	75.82 (CH)	3.62-3.59 (1H, <i>m</i>)	C-3", C-4", C-5", C-6"
6''	61.98 (CH ₂)	3.54 (2H, <i>br s</i>)	C-4", C-5", C-1""
2′′-ОН	-	5.37 (1H, <i>br s</i>) ^a	-
3′′-ОН	-	5.07 (1H, <i>br s</i>) ^a	-
4''-OH	-	5.11 (1H, <i>br s</i>) ^a	-
Rha-1'''	100.53 (CH)	4.67 (1H, <i>br s</i>)	C-6", C-2"", C-3"", C-5""
2'''	70.29 (CH)	3.85 (1H, m)	C-2''', C-4'''
3'''	71.21 (CH)	3.80-3.65 (1H, m)	C-1''', C-4'''
4'''	72.56 (CH)	3.45-3.35 (1H, <i>m</i>)	C-3''', C-5'''
5'''	68.16 (CH)	3.65-3.60 (1H, <i>m</i>)	C-3''', C-4'''
6'''	17.61 (CH ₃)	1.22 (3H, d, 5.6)	C-4''', C-5'''
2′′′-ОН	-	4.45 (1H, <i>br s</i>) ^b	-
3′′′-ОН	-	4.36 (1H, <i>br s</i>) ^b	-
4′′′-OH	-	4.55 (1H, <i>br s</i>) ^b	-

^{*} Carbon type was deduced from DEPT experiments.

^{a, b} Assignment maybe interchangeable.

GD5: 1,6-Dihydroxy-3,7-dimethoxy-2-(3-methyl-2-butenyl)xanthone

GD5 is a pale yellow solid, m.p. 250-251 °C. The ¹H NMR spectrum (**Table 29**) showed two *singlet* signals of a chelated hydroxy proton (1-OH) at δ 13.10 and a non chealated hydroxy proton (6-OH) at δ 8.17. Three *singlet* signals were observed at δ 6.40, 6.92 and 7.56. These signals were assigned for aromatic protons H-4, H-5 and H-8, respectively. The deshielded aromatic proton at δ 7.56 was assigned to be at C-8 due to an anisotropic effect of the carbonyl group. Two singlet signals at δ 3.90 and 3.98 were assigned for two methoxy groups of 3-OCH₃ and 7-OCH₃, respectively. In addition, a set of signals of a prenyl side chain were shown in the spectrum, of which the methylene protons resonated at $\delta 3.34$ (d, H-1'), olefinic proton resonated at $\delta 5.21$ (t, H-2') and two methyl groups resonated at δ 1.66 (s, H-4') and 1.78 (s, H-5'). The ¹³C NMR and DEPT experiments indicated the presence of a carbonyl carbon, ten quaternary carbons, four methine carbons, a methylene carbon and four methyl carbons. The location of the prenyl side chain at C-2 was determined from HMBC correlations; $H-1'(\delta 3.34)$ correlated to C-1, C-2 and C-3. The location of aromatic protons H-4, H-5 and H-8 were indicated by correlations of H-4 to C-2, C-3, C-4a, C-9a; H-5 to C-7, C-8a, C-10a and H-8 to C-7, C-9, C-10a. Therefore **GD5** was proposed to be 1,6-dihydroxy-3,7-dimethoxy-2-(3-methyl-2-butenyl)xanthone or known as cowagarcinone B (Chairerk, 2001).



Major HMBC correlations of GD5

Table 29 The NMR spectral data of GD5

D :::	$\delta_{_{ m C}}$ (C-Ty	pe)*	$\delta_{\!\scriptscriptstyle m H}$ (multip	plicity, J_{Hz})	III MC
Position	GD5	**	GD5	**	HMBC
1	159.25 (C)	159.36	-	-	-
2	111.51 (C)	111.76	-	-	-
3	163.67 (C)	163.85	-	-	-
4	89.51 (CH)	89.58	6.40 (1H, s)	6.43 (1H, s)	C-2, C-3, C-4a, C-9a
4a	156.30 (C)	156.24	-	-	-
5	102.76(CH)	102.49	6.92 (1H, s)	6.94 (1H, s)	C-7, C-8a, C-10a
6	152.20 (C)	152.37	-	-	-
7	145.14 (C)	144.32	-	-	-
8	104.68(CH)	104.62	7.56 (1H, s)	7.61 (1H, s)	C-7, C-9, C-10a
8a	113.50 (C)	113.63	-	-	-
9	179.85(C=O)	179.86	-	-	-
9a	103.20 (C)	104.62	-	-	-
10a	153.24 (C)	152.54	-	-	-
1'	21.30 (CH ₂)	21.36	3.34 (2H,d,6.6)	3.37 (2H,d,6.5)	C-1, C-2, C-3, C-2', C-3'
2'	122.20(CH)	122.21	5.21 (1H,t, 6.9)	5.24 (1H, <i>t</i> ,7.0)	C-1', C-5'
3'	131.90 (C)	131.83	-	-	-
4'	25.79 (CH ₃)	25.80	1.66 (3H, s)	1.68 (3H, s)	C-2', C-3', C-5'

Table 29 (Continued)

D :::	$\delta_{_{ m C}}$ (C-Ty	pe)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)		HMDC	
Position	GD5	**	GD5	**	НМВС	
5′	17.78 (CH ₃)	17.80	1.78 (3H, s)	1.80 (3H, s)	C-2', C-3', C-4'	
1-OH	-	-	13.10 (1H, s)	13.00 (3H, s)	-	
6-OH	-	-	8.17 (1H, s)	6.34 (1H, s)	-	
3-OCH ₃	55.87 (CH ₃)	55.90	3.90 (3H, s)	3.92 (3H, s)	C-3	
7-OCH ₃	56.38 (CH ₃)	56.53	3.98 (3H, s)	4.01 (3H, s)	C-7	

^{*} Carbon type was deduced from DEPT experiments.

^{**} PGC8 (Chairerk, 2001)

GD6: 1,3,6-Trihydroxy-7-methoxy-2-(3-methyl-2-butenyl)-8-(3,7-dimethyl-2,6-octadienyl)xanthone (cowanin)

GD6 is a yellow solid, m.p. 137-138 °C. The ¹H NMR spectrum (**Table 30**) demonstrated the resonances of a chelated hydroxy group (1-OH) at δ 13.80 and two singlet signals of H-4 and H-5 at δ 6.29 and 6.81, respectively. A signal of methoxy group was shown at δ 3.81 and the methoxyl was located at C-7. The ^{1}H NMR spectrum also showed a characteristic signals of a prenyl unit at $\delta 5.31(1H, br t, H-2')$, 3.46 (2H, d, H-1'), 1.84 (3H, s, H-5') and 1.78 (3H, s, H-4'). The remaining signals were assigned for a geranyl side chain. A doublet at δ 4.10 and a broad triplet at δ 5.28 were assigned for the signals of methylene protons H-1" and an olefinic proton H-2", respectively. A broad triplet at δ 5.04 and a multiplet at δ 2.06 were the signals of an olefinic proton H-6" and two groups of methylene protons H-4" and H-5", respectively whereas three singlet signals at δ 1.85 (3H, s, H-9"), 1.61 (3H, s, H-10") and 1.56 (3H, s, H-8") were those of three methyl groups. Since the chemical shift of methylene protons H-1" of the geranyl side chain was appeared at lower field than that of H-1' of the prenyl side chain, the geranyl group thus was placed nearby the carbonyl group and the prenyl moiety was placed at C-2. Thus GD6 was assigned to be 1,3,6trihydroxy-7-methoxy-2-(3-methyl-2-butenyl)-8-(3,7-dimethyl-2,6-octadienyl) xanthone or known as cowanin (Na Pattalung, et al., 1994).

Table 30 The ¹H NMR spectral data of **GD6**

D :::	$\delta_{_{ m H}}$ (multip	plicity, $J_{\rm Hz}$)
Position	GD6*	cowanin**
4	6.29 (1H, s)	6.30 (1H, s)
5	6.81 (1H, s)	6.86 (1H, s)
1'	3.46 (2H, d, 7.0)	3.45 (2H, br d, 7.0)
2'	5.31 (1H, <i>br t</i> , 7.0)	5.28 (1H, br t, 7.0)
4'	1.78 (3H, s)	1.76 (3H, s)
5'	1.84 (3H, s)	1.82 (3H, s)
1''	4.10 (2H, d, 7.0)	4.09 (2H, br d, 7.0)
2''	5.28 (1H, br t, 7.0)	5.28 (1H, br t, 7.0)
4''	2.06 (1H, m)	2.03 (1H, m)
5''	2.06 (1H, m)	2.03 (1H, m)
6''	5.04 (1H, br t, 7.0)	5.03 (1H, <i>br t</i>)
8''	1.56 (3H, s)	1.59 (3H, s)
9''	1.85 (3H, s)	1.84 (3H, s)
10''	1.61 (3H, s)	1.61 (3H, s)
1-OH	13.80 (1H, s)	13.80 (1H, s)
7-OCH ₃	3.81 (3H, s)	3.80 (3H, s)

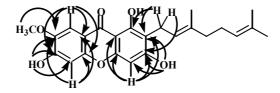
^{* 500} MHz, in CDCl_3

^{** 400} MHz, in CDCl₃

GD7: 1,3,6-Trihydroxy-7-methoxy-2-(3,7-dimethyl-2,6-octadienyl)xanthone (cowaxanthone)

GD7 is a yellow solid, m.p. 197-199 °C. EIMS showed the molecular ion of 410 which corresponded to C₂₄H₂₆O₆. A xanthone nucleus was elucidated from the UV spectra (λ_{max} 361.4, 320.5, 258.3 and 241.8 nm). IR absorption bands at 3369 cm $^{\text{--}1}$ and 1646 cm⁻¹ indicated the presence of hydroxyl stretching and conjugated carbonyl stretching, respectively. The ¹H NMR spectrum (**Table 31**) showed the resonances of a hydrogen-bonded phenolic hydroxy group at δ 13.35 (s, 1-OH), two free hydroxy protons at δ 6.39 (s, 3-OH) and 6.43 (s, 6-OH) and three singlet signals of three isolated aromatic protons at δ 7.58 (H-8), 6.92 (H-5) and 6.48 (H-4). The most deshielded aromatic proton signal, δ 7.58, was assigned for H-8 according to an anisotropic effect by the carbonyl group. The sharp singlet resonance at δ 4.00 belonged to the methoxy group which was located at C-7 according to HMBC correlation of OCH₃ protons to C-7. The remaining proton signals were assigned for the geranyl side chain which was located at C-2. Those signals were assigned as follow; three singlet signals at δ 1.84, 1.68 and 1.67 were of three vinylic methyl groups, a doublet signal at δ 3.48 was assigned for benzylic methylene protons H-1', a multiplet signal at $\delta 2.10$ (4H) was the signal of two groups of methylene protons H-4' and H-5' and two sets of broad triplet signals at δ 5.31 and 5.06 were the signals of two olefinic methine protons H-2' and H-6'. HMBC correlations between H-4 to C-2, C-3, C-4a, C-9a and 3-OH to C-2, C-3, C-4 confirmed the connection of isolated aromatic proton at C-4. The HMBC correlation of H-1' to C-1, C-2 and C-3 confirmed

the presence of the geranyl unit at C-2. The ¹³C NMR spectral data deduced from DEPT and HMQC experiments showed the signals of a carbonyl carbon, eleven quaternary carbons, five methine carbons, three methylene carbons and four methyl carbons. Accordingly, **GD7** was confirmed to be 1,3,6-trihydroxy-7-methoxy-2-(3,7-dimethyl-2,6-octadienyl)xanthone or cowaxanthone (Likhitwitayawuid, *et al.*, 1997).



Major HMBC correlations of GD7

Table 31 The NMR spectral data of GD7

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
1	160.11 (C)	-	-
2	108.72 (C)	-	-
3	162.24 (C)	-	-
4	94.18 (CH)	6.48 (1H, s)	C-2, C-3, C-4a, C-9a
4a	156.01 (C)	-	-
5	102.60 (CH)	6.92 (1H, s)	C-7, C-8a, C-9, C-10a
6	152.60 (C)	-	-
7	144.30 (C)	-	-
8	104.54 (CH)	7.58 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
8a	113.30 (C)	-	-
9	179.96 (C=O)	-	-
9a	103.04 (C)	-	-
10a	152.52 (C)	-	-

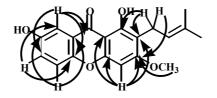
Table 31 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)*	δ_{H} (multiplicity, J_{Hz})	НМВС
1'	21.39 (CH ₂)	3.48 (2H, d, 7.2)	C-1, C-2, C-3, C-2', C-3'
2'	121.25 (CH)	5.31 (1H, br t, 7.2)	C-2, C-1', C-4', C-9'
3'	139.56 (C)	-	-
4'	39.73 (CH ₂)	2.10 (2H, m)	C-2', C-3', C-5', C-6', C-9'
5'	26.36 (CH ₂)	2.10 (2H, m)	C-3', C-4', C-6', C-7'
6'	123.73 (CH)	5.06 (1H, br t, 7.2)	C-4', C-10'
7'	132.12 (C)	-	-
8'	25.69 (CH ₃)	1.68 (3H, s)	C-6', C-7', C-10'
9′	16.27 (CH ₃)	1.84 (3H, s)	C-2', C-3', C-4'
10'	17.72 (CH ₃)	1.67 (3H, s)	C-6', C-7', C-8'
1-OH	-	13.35 (1H, s)	C-1, C-2, C-9a
3-OH	-	6.39 (1H, s)	C-2, C-3, C-4
6-OH	-	6.43 (1H, s)	C-5, C-6, C-7
7-OCH ₃	56.49 (CH ₃)	4.00 (3H, s)	C-7

^{*} Carbon type was deduced from DEPT experiments.

GD8: 1,7-Dihydroxy-3-methoxy-2-(3-methyl-2-butenyl)xanthone

GD8 is a yellow solid, m.p. 228-230 °C. The ¹H NMR spectrum (Table 32) showed two singlet signals, which disappeared upon addition of D₂O, of a chelated hydroxy group (1-OH) at δ 12.90 and a non-chelated hydroxy group (7-OH) at δ 9.25. The signals in aromatic region, δ 7.50 (d), 7.28 (d) and 7.16 (dd) appearing as ABX type were proposed for the signals of H-8, H-5 and H-6, respectively. These assignment were supported by ³J correlations of H-8 to C-6, C-10a, C-9; H-5 to C-7a, C-8a and H-6 to C-8, C-10a on HMBC experiment. A singlet signal of the aromatic proton H-4 was at δ 6.31 according to the correlation to C-2, C-3, C-9, C-4a and C-9a from the HMBC experiment. The spectrum further showed the typical signals of a prenyl side chain which appeared at δ 1.57 (3H, s, H-4') and 1.69 (3H, s, H-5'), 3.23 (2H, d, H-1') and 5.11 (1H, br t, H-2'). This prenyl unit was assigned to be at C-2 and was supported by the correlation of H-1' to C-1, C-2 and C-3. A signal of a methoxy group appearing as a *singlet* at δ 3.82 was indicated to be at C-3 by the 3J correlation of methoxy protons to C-3. The ¹³C NMR spectral data (**Table 32**) suggested that **GD8** contained two methyl carbons, a methylene carbon, five methine carbons, nine quaternary carbons, a methoxy carbon and a carbonyl carbon. The proposed structure, the spectral data and the melting point were found to be corresponded to 1,7dihydroxy-3-methoxy-2-(3-methyl-2-butenyl)xanthone (Mahabusarakam, et al., 1987).



Major HMBC correlations of GD8

Table 32 The NMR spectral data of GD8

Position	$\delta_{_{ m C}}$ (C-Type)*	δ_{H} (multiplicity, J_{Hz})	HMBC
1	159.00 (C)	-	-
2	111.00 (C)	-	-
3	164.00 (C)	-	-
4	89.47 (CH)	6.31 (1H, s)	C-2, C-3, C-4a, C-9, C-9a
4a	156.00 (C)	-	-
5	118.44 (CH)	7.28 (1H, d, 9.0)	C-7, C-8a, C-10a
6	124.24 (CH)	7.16 (1H, dd, 9.0, 2.5)	C-7, C-8, C-10a
7	149.50 (C)	-	-
8	108.85 (CH)	7.50 (1H, d, 2.5)	C-6, C-7, C-9, C-10a
8a	121.00 (C)	-	-
9	180.50 (C=O)	-	-
9a	103.00 (C)	-	-
10a	153.50 (C)	-	-
1'	21.20 (CH ₂)	3.23 (2H, d, 7.0)	C-1, C-2, C-3, C-2', C-3'
2'	122.10 (CH)	5.11 (1H, <i>br t</i> , 7.0)	C-1', C-4', C-5'
3'	131.50 (C)	-	-
4'	25.75 (CH ₃)	1.57 (3H, s)	C-2', C-3', C-5'
5'	17.50 (CH ₃)	1.69 (3H, s)	C-2', C-3', C-4'

Table 32 (Continued)

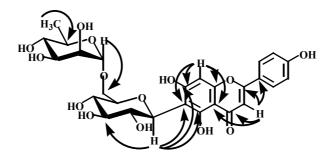
Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
1-OH	-	12.90 (1H, s)	-
3-OCH ₃	55.88 (CH ₃)	3.82 (3H, s)	C-3
7-OH	-	9.25 (1H, s)	C-6, C-7, C-8

^{*} Carbon type was deduced from DEPT experiments.

GD9: 5,7,4'-Trihydroxyflavone 6-C-[α -rhamnopyranosyl-(1 \rightarrow 6)]- β -glucopyranoside

GD9 is a yellow solid, m.p. 200-202 °C. The HRMS showed the [M+H]⁺ of 579.1742 corresponded to C₂₇H₃₁O₁₄. This compound exhibited UV absorption bands at 333.6, 271.7, 238.0 and 215.8 nm, a characteristic of a flavone nucleus and IR absorption bands at 3402 cm⁻¹ (a hydroxyl group), 1650 cm⁻¹ (a carbonyl group of an flavone). The ¹H NMR spectrum showed the characteristic resonances of a flavone proton at δ 6.63 (s, H-3), a hydrogen-bonded hydroxy proton at δ 13.31 (s, 5-OH) and an aromatic proton at δ 6.51 (s, H-8). Two doublet resonances (J = 9.3 Hz) at δ 7.83 (2H) and δ 6.90 (2H) were in agreement with the A_2B_2 type of aromatic protons H-2', H-6' and H-3', H-5'. The ¹H NMR (**Table 33**) further indicated the presence of two sugar moieties. The resonances of an anomeric proton H-1" and an anomeric carbon C-1" of a β -glucose were observed at δ 4.58 (d, J= 9.3 Hz) and δ 74.11 whereas the resonances of an anomeric proton H-1", an anomeric carbon C-1" and methyl protons (H-6''') of an α -rhamnose were shown at δ 4.51 (br s), 101.52 and 1.06 (d, J= 6.0 Hz). The proton signals of sugar moieties appeared at δ 3.42-3.40 (m, H-2", H-5"), 3.24 (t, H-3"), 3.16 (t, H-4"), 3.80 (d, H-6"), 4.51 (br s, H-1""), 4.10-4.02 (m, $\text{H-2}^{\prime\prime\prime}$), 3.34 (t, $\text{H-3}^{\prime\prime\prime}$), 3.14 (t, $\text{H-4}^{\prime\prime\prime}$), 3.92-3.90 (m, $\text{H-5}^{\prime\prime\prime}$) and 1.06 (d, $\text{H-6}^{\prime\prime\prime}$). The identification of 6-C-β-glucopyranoside was indicated by the HMBC correlation of H-1" to C-5, C-6 and C-7. The $(1\rightarrow 6)$ -C-glucosidic bond was assigned from the downfield shift of C-6" of glucose unit (δ 68.54) and the cross peak of H-1" to C-6".

The complete HMBC data confirmed the structure. 5,7,4'-Trihydroxyflavone 6-C-[α -rhamnopyranosyl-(1 \rightarrow 6)]- β -glucopyranoside was then assigned for **GD9**, a novel compound.



Major HMBC correlations of GD9

Table 33 The NMR spectral data of GD9

Position	$\delta_{\rm C}$ (C-Type)	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
2	165.19 (C)	-	-
3	103.77 (CH)	6.63 (1H, s)	C-2, C-4, C-4a, C-1'
4	183.24 (C=O)	-	-
4a	104.55 (C)	-	-
5	161.98 (C)	-	-
6	109.32 (C)	-	-
7	164.34 (C)	-	-
8	94.60 (CH)	6.51 (1H, s)	C-4, C-4a, C-6, C-7, C-8a
8a	157.65 (C)	-	-
1'	122.35 (C)	-	-
2', 6'	129.67 (CH)	7.83 (2H, d, 9.3)	C-2, C-2',6', C-4'
3', 5'	117.28 (CH)	6.90 (2H, d, 9.3)	C-1', C-3',5', C-4'
4'	161.55 (C)	-	-

Table 33 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
Glc-1"	74.11 (CH)	4.58 (1H, d, 9.3)	C-5, C-6, C-7, C-3", C-5"
2''	71.31 (CH)	3.42-3.40 (1H, <i>m</i>)	C-3''
3''	79.60 (CH)	3.24 (1H, t, 9.3)	C-2", C-4"
4''	71.47 (CH)	3.16 (1H, t, 9.3)	C-3", C-5", C-6"
5''	73.06 (CH)	3.42-3.40 (1H, <i>m</i>)	C-6"
6''	68.54 (CH ₂)	3.80 (1H, d, 9.3)	C-4", C-5"", C-1""
Rha-1'''	101.52 (CH)	4.51 (1H, <i>br s</i>)	C-6", C-2"", C-3"", C-5""
2'''	71.56 (CH)	4.10-4.02 (1H, m)	C-4'''
3'''	70.71 (CH)	3.34 (1H, <i>t</i> , 9.3)	C-1''', C-4'''
4'''	71.33 (CH)	3.14 (1H, t, 9.3)	C-3''', C-5'''
5'''	69.40 (CH)	3.92-3.90 (1H, m)	C-3''', C-4'''
6'''	18.61 (CH ₃)	1.06 (3H, d, 6.0)	C-4''', C-5'''
5-OH	-	13.31 (1H, s)	C-4a, C-5, C-6, C-1"

^{*} Carbon type was deduced from DEPT experiments.

GD10: 1,5,8-Trihydroxy-3-methoxy-2-(3-methyl-2-butenyl)xanthone

GD10 is a yellow solid, m.p. 195-197 °C. The ¹H NMR spectrum (**Table 34**) exhibited the resonances of H-4 (δ 6.49, s), H-6 (δ 7.25, d, J = 9.0 Hz), H-7 (δ 6.70, d, J = 9.0 Hz), 3-OCH₃ (δ 3.95) and two chelated hydroxy protons 1-OH (δ 12.12) and 8-OH and (δ 11.32). The characteristic signals of a prenyl side chain were observed at δ 5.22 (br t, H-2'), 3.36 (d, J = 7.0 Hz, H-1'), 1.69 (s, H-4') and 1.80 (s, H-5'). HMBC correlation of H-1' to C-2 indicated that the prenyl side chain was at C-2. The assignment of H-4 was confirmed by the correlation of H-4 to C-2, C-3, C-4a and C-9a. The NOE effect of H-4 to the methoxy proton resonances confirmed the location of 3-OCH₃. These assignment indicated that **GD10** was 1,5,8-trihydroxy-3-methoxy-2-(3-methyl-2-butenyl)xanthone which was first isolated from *Garcinia mangostana* (Parveen and Khan, 1988).

Major HMBC correlations of GD10

 $Table \ 34 \ \ The \ NMR \ spectral \ data \ of \ GD10$

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
1	159.26 (C)	-	-
2	112.83 (C)	-	-
3	165.05 (C)	-	-
4	89.96 (CH)	6.49 (1H, s)	C-2, C-3, C-4a, C-9a
4a	155.66 (C)	-	-
5	135.51 (C)	-	-
6	122.97 (CH)	7.25 (1H, d, 9.0)	C-5, C-8
7	109.95 (CH)	6.70 (1H, d, 9.0)	C-5, C-8
8	153.96 (C)	-	-
8a	107.57 (C)	-	-
9	184.40 (C=O)	-	-
9a	102.60 (C)	-	-
10a	142.78 (C)	-	-
1'	21.31 (CH ₂)	3.36 (2H, d, 7.0)	C-2, C-2'
2'	121.63 (CH)	5.22 (1H, br t, 7.5)	C-2, C-4', C-5'
3'	132.26 (C)	-	-
4'	25.82 (CH ₃)	1.69 (3H, s)	C-2', C-3', C-5'
5'	17.80 (CH ₃)	1.80 (3H, s)	C-2', C-3', C-4'
1-OH	-	12.12 (1H, s)	C-1, C-2, C-9a
8-OH	-	11.32 (1H, s)	C-7, C-8, C-8a
3-OCH ₃	56.13 (CH ₃)	3.95 (3H, s)	C-3

^{*} Carbon type was deduced from DEPT experiments.

GD11: 5,4'-Dihydroxy-3'-(3-methyl-2-butenyl)-2",2"-dimethylchromeno (5",6":6,7)isoflavone (chandalone)

GD11 is a pale yellow solid, m.p. 65-66 °C. The ¹H NMR spectrum (**Table 35**) demonstrated the *singlet* resonance of a chelated hydroxy group (5-OH) at δ 13.19 and two *singlet* signals of H-2 and H-8 at δ 7.82 and 6.34, respectively. The ABX pattern of aromatic protons H-2', H-5' and H-6' were shown at δ 7.25 (d, J = 2.5 Hz), 6.87 (d, J = 10.0 Hz) and 7.26 (d, J = 10.0 and 2.5 Hz), respectively. Two *doublet* signals of the vicinal olefinic protons at δ 5.64 and 6.74 and a *singlet* resonance of two methyl groups at δ 1.49 (6H) indicated the presence of a chromene ring. Correlations of H-4" to C-5, C-6, C-7 and H-8 to C-4a, C-6, C-7, C-8a, C-4 identified the position of the dimethylchromene ring to be at C-6 and C-7. In addition, a set of signal of a prenyl side chain was shown in the spectrum at δ 3.41 (d, H-1""), 5.36 (br t, H-2""), 1.79 (s, H-4"") and 1.80 (s, H-5""). The HMBC correlation of H-1"" to C-2', C-3' and C-4' suggested the position of the prenyl side chain at C-3'. 5,4'-Dihydroxy-3'-(3-methyl-2-butenyl)-2",2"-dimethylchromeno(5",6":6,7)isoflavone (Falshaw, *et al.*, 1969) then was proposed for **GD11**.

Major HMBC correlations of GD11

Table 35 The NMR spectral data of GD11

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
2	152.52 (CH)	7.82 (1H, s)	C-4, C-8a, C-1'
3	123.68 (C)	-	-
4	180.96 (C=O)	-	-
4a	106.09 (C)	-	-
5	159.45 (C)	-	-
6	156.89 (C)	-	-
7	105.53 (C)	-	-
8	94.83 (CH)	6.34 (1H, s)	C-4, C-4a, C-6, C-7, C-8a
8a	157.27 (C)	-	-
1'	122.93 (C)	-	-
2'	130.53 (CH)	7.25 (1H, d, 2.5)	C-3, C-4', C-6', C-1'''
3'	127.10 (C)	-	-
4'	154.74 (C)	-	-
5'	115.90 (CH)	6.87 (1H, d, 10.0)	C-1', C-3', C-4', C-6'
6'	128.18 (CH)	7.26 (1H, dd, 10.0, 2.5)	C-3, C-2', C-4'
2''	78.01 (C)	-	-
3''	128.13 (CH)	5.64 (1H, d, 10.0)	C-6, C-7, C-2", C-4", C-5"
4''	115.48 (CH)	6.74 (1H, d, 10.0)	C-5, C-6, C-7
1'''	29.82 (CH ₂)	3.41 (2H, d, 6.0)	C-2', C-3', C-4', C-2''', C-3'''
2'''	121.53 (CH)	5.36 (1H, <i>br t</i> , 6.0)	C-3', C-1''', C-4''', C-5'''
3'''	134.99 (C)	-	-
4'''	25.84 (CH ₃)	1.79 (3H, s)	C-2"', C-3"', C-5""
5'''	17.97 (CH ₃)	1.80 (3H, s)	C-2"', C-3"', C-4"'
5-OH	-	13.19 (1H, s)	C-4a, C-6
2"-(CH ₃) ₂	28.27 (CH ₃)	1.49 (6H, s)	2"-(CH ₃) ₂ , C-2", C-3"

^{*} Carbon type was deduced from DEPT experiments.

GD12: 5,7,4'-Trihydroxy-6,3'-bis(3-methyl-2-butenyl)isoflavone (lupalbigenin)

GD12 was isolated as a yellow solid, m.p. 122-125 °C. The ¹H NMR spectrum (**Table 36**) showed a sharp *singlet* signal of a chelated hydroxy group 5-OH at δ 13.17, a *singlet* signal of a vinylic proton H-2 at δ 7.80 and a *singlet* signal of an isolated aromatic proton H-8 at δ 6.40. The resonances of aromatic protons H-2', H-6' and H-5' were shown as ABX pattern at δ 7.23, 6.83 and 6.86, respectively. The signals of two prenyl side chains were observed. The first set resonated at δ 1.74 (s, H-4''), 1.83 (s, H-5''), 3.43 (d, H-1'') and 5.29 (s, H-2'') whereas the second appeared at δ 1.78 (6H, s, H-4''' and H-5'''), 3.39 (d, H-1''') and 5.35 (br t, H-2'''). HMBC correlations (**Table 36**) showed the correlation of H-1''' to C-5, C-6 and C-7. It's therefore confirmed that one prenyl side chain was at C-6. In addition, the correlation of H-1''' to C-2' and C-4' supported that another prenyl group was at C-3'. **GD12** was therefore assigned to be 5,7,4'-trihydroxy-6,3'-bis(3-methyl-2-butenyl)isoflavone which was known as lupalbigenin (Pistelli, et al, 1996).

Major HMBC correlations of GD12

 $Table \ 36 \ \ The \ NMR \ spectral \ data \ of \ GD12$

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2	152.40 (CH)	7.80 (1H, s)	C-3, C-4, C-8a, C-1', C-2', C-6'
3	123.60 (C)	-	-
4	180.92 (C=O)	-	-
4a	105.61 (C)	-	-
5	159.57 (C)	-	-
6	110.82 (C)	-	-
7	161.61 (C)	-	-
8	93.70 (CH)	6.40 (1H, s)	C-4a, C-6, C-7, C-8a, C-1''
8a	156.02 (C)	-	-
1'	122.89 (C)	-	-
2'	130.44 (CH)	7.23 (1H, d, 2.3)	C-3, C-4', C-6', C-1'''
3'	127.33 (C)	-	-
4'	154.75 (C)	-	-
5'	115.75 (CH)	6.86 (1H, d, 9.2)	C-1', C-3', C-4', C-6'
6'	128.04 (CH)	6.83 (1H, dd, 9.2, 2.3)	C-3, C-2', C-4', C-5'
1''	21.46 (CH ₂)	3.43 (2H, d, 8.0)	C-5, C-6, C-7, C-2", C-3", C-4", C-5"
2''	121.59 (CH)	5.29 (1H, br t, 8.0)	C-6, C-1", C-4", C-5"
3''	134.11 (C)	-	-
4''	25.78 (CH ₃)	1.74 (3H, s)	C-3", C-5"
5''	17.89 (CH ₃)	1.83 (3H, s)	C-3", C-4"
1'''	29.50 (CH ₂)	3.39 (2H, d, 8.0)	C-2', C-4', C-2''', C-3''', C-4''', C-5'''
2'''	121.76 (CH)	5.35 (1H, br t, 8.0)	C-3', C-1''', C-4''', C-5'''
3'''	134.41 (C)	-	-
4'''	25.80 (CH ₃)	1.78 (3H, s)	C-3''', C-5'''
5'''	17.87 (CH ₃)	1.78 (3H, s)	C-3''', C-4'''
5-OH	-	13.17 (1H, s)	C-4, C-4a, C-5, C-6

^{*} Carbon type was deduced from DEPT experiments.

GD13: 1,6-Dihydroxy-2',2'-dimethylchromano(5',6':2,3)-2",2"-dimethylchromano(5",6":8,7)xanthone (BR-xanthone A)

GD13 is a yellow solid, m.p. 182-183 °C. The ¹H NMR spectrum (**Table 37**) exhibited four *singlet* signals of 1-OH (δ 13.76), 6-OH (δ 6.39), H-4 (δ 6.25) and H-5 (δ 6.80). ¹H NMR spectrum revealed a characteristic signals of two dimethylchromane rings. The first set of resonances of a dimethylchromane ring with *gem*-dimethyl protons at δ 1.37 (6H, s) and two methylene protons at δ 1.84 (2H, t, H-3') and 2.72 (2H, t, H-4') was assigned to be at C-3 and C-2 by the correlation of H-4' to C-1 and C-2 in HMBC. The second set of signals of another dimethylchromane ring with *gem*-dimethyl protons and two methylene protons at δ 1.39 (s, 6H), 1.88 (t, H-3'') and 3.51 (t, H-4'') was suggested to be fused to the xanthone nucleus at C-7 and C-8 by the HMBC correlation of H-4'' to C-8, C-8a and C-10a. The ¹³C NMR and HMBC experiment corresponded to the assigned structure. Thus, **GD13** was 1,6-dihydroxy-2',2'-dimethylchromano(5',6':2,3)-2'',2''-dimethylchromano(5'',6'':8,7)xanthone or BR-xanthone A (Bulasubramanian, *et al.*, 1988).

Major HMBC correlations of GD13

 $Table\ 37\ \ The\ NMR\ spectral\ data\ of\ GD13$

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
1	159.98 (C)	-	-
2	102.94 (C)	-	-
3	154.31 (C)	-	-
4	93.36 (CH)	6.25 (1H, s)	C-2, C-3, C-9, C-9a
4a	159.95 (C)	-	-
5	100.65 (CH)	6.80 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	152.67 (C)	-	-
7	153.06 (C)	-	-
8	121.08 (C)	-	-
8a	110.00 (C)	-	-
9	182.06 (C=O)	-	-
9a	103.48 (C)	-	-
10a	138.98 (C)	-	-
2'	75.57 (C)	-	-
3'	31.45 (CH ₂)	1.84 (2H, t, 6.6)	C-2, C-2', C-4', 2'-(CH ₃) ₂
4'	15.70 (CH ₂)	2.72 (2H, t, 6.6)	C-1, C-2, C-2', C-3'
2''	73.81 (C)	-	-
3''	32.40 (CH ₂)	1.88 (2H, t, 6.6)	C-8, C-2", C-4", 2"-(CH ₃) ₂
4''	22.26 (CH ₂)	3.51 (2H, t, 6.6)	C-8, C-8a, C-10a, C-2", C-3"
1-OH	-	13.76 (1H, s)	-
6-OH	-	6.39 (1H, s)	-
2'-(CH ₃) ₂	26.08 (CH ₃)	1.37 (6H, s)	C-2', C-3', 2'-(CH ₃) ₂
2"-(CH ₃) ₂	26.42 (CH ₃)	1.39 (6H, s)	C-2", C-3", 2"-(CH ₃) ₂

^{*} Carbon type was deduced from DEPT experiments.

GD14: 1,3,6-Trihydroxy-7-methoxy-2,8-bis(3-methyl-2-butenyl)xanthone (mangostin)

GD14 was isolated as a yellow solid, m.p. 182-183 °C. The ¹H NMR (**Table 38**) revealed the presence of four sharp *singlet* signals at δ 13.80 (1-OH), 6.82 (H-5), 6.29 (H-4) and 3.81 (7-OCH₃). The signals of two prenyl side chains were observed. Those signals were two *broad triplet* signals of two olefinic protons (δ 5.29 and 5.27), two *doublet* signals of benzylic methylene protons (δ 4.06 and 3.46) and three *singlet* signals of four methyl groups (δ 1.85 (6H), 1.77 and 1.70). **GD14** was then identified to be 1,3,6-trihydroxy-7-methoxy-2,8-bis(3-methyl-2-butenyl)xanthone. The ¹H NMR spectral data and melting point were identical with mangostin, authentic sample, which was previously isolated from *Garcinia mangostana* (Mahabusarakam, *et al.*, 1987).

Table 38 The ¹H NMR spectral data of **GD14**

$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)			
GD14*	mangostin**		
13.80 (1H, s)	13.80 (s)		
6.82 (1H, s)	6.30 (s)		
6.29 (1H, s)	6.28 (s)		
5.29 (1H, <i>br t</i>)	5.28 (t, 7.0)		
5.27 (1H, <i>br t</i>)	5.28 (br s, 7.0)		
4.06 (2H, br d, 5.7)	4.11 (<i>d</i> , 7.0)		
3.81 (3H, s)	3.81 (s)		
3.46 (2H, <i>br d</i> , 6.9)	3.47-3.40 (<i>d</i> , 7.0)		
1.85 (6H, s)	1.84 (s), 1.85 (s)		
1.77 (3H, s)	1.78 (s)		
1.70 (3H, s)	1.70 (s)		

^{* 300} MHz in CDCl₃

^{** 400} MHz in CDCl₃

GD15: 5,7,4'-Trihydroxy-8,3'-bis(3-methyl-2-butenyl)isoflavone (isolupalbigenin)

GD15 is yellow needles, m.p. 164-166 °C. The ¹H NMR spectrum (**Table 39**) showed three *singlet* signals of 5-OH (δ 13.10), H-2 (δ 7.72) and H-6 (δ 6.37). The spectrum also showed the resonances of aromatic protons H-2', H-5' and H-6' which appeared as an ABX pattern at δ 7.13, 7.12 and 6.82, respectively. In addition, two sets of signals due to two prenyl side chains were present and were assigned to attach to the parent structure at C-8 and C-3'. The first showed the signals at δ 5.22 (H-2''), 3.30 (H-1'''), 1.74 (H-5'') and 1.63 (H-4'') whereas the second showed the signals at δ 5.30 (H-2'''), 3.30 (H-1''') and 1.67 (H-4''', H-5'''). The ¹³C NMR and the DEPT spectra suggested that **GD15** contained four methyl carbons, two methylene carbons, seven methine carbons, eleven quaternary carbons and a carbonyl carbon. The HMBC (**Table 39**) also showed the correlations of H-1''' to C-2', C-3', C-4' and H-1'' to C-7, C-8, C-8a suggested that the prenyl groups were at C-3' and C-8. **GD15** was therefore assigned to be 5,7,4'-trihydroxy-8,3'-bis(3-methyl-2-butenyl)isoflavone which was known as isolupalbigenin (Tahara, *et al.*, 1994).

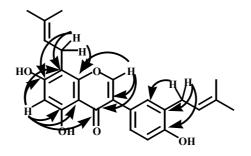


Table 39 The NMR spectral data of GD15

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
2	152.19 (CH)	7.72 (1H, s)	C-3, C-4, C-8a
3	123.62 (C)	-	-
4	180.87 (C=O)	-	-
4a	105.26 (C)	-	-
5	161.92 (C)	-	-
6	93.35 (CH)	6.37 (1H, s)	C-4, C-4a, C-5, C-7, C-8
7	159.49 (C)	-	-
8	111.72 (C)	-	-
8a	155.89 (C)	-	-
1'	122.16 (C)	-	-
2'	130.10 (CH)	7.13 (1H, <i>br d</i> , 1.2)	C-3, C-1', C-3', C-4', C-1'''
3'	128.10 (C)	-	-
4'	155.04 (C)	-	-
5'	127.59 (CH)	7.12 (1H, <i>br d</i> , 6.9)	C-4', C-6'
6'	115.27 (CH)	6.82 (1H, <i>br dd</i> , 6.9, 1.2)	C-1', C-2', C-4', C-5'
1''	21.43 (CH ₂)	3.30 (2H, <i>br d</i> , 6.6)	C-7, C-8, C-8a, C-2", C-3"
2''	122.28 (CH)	5.22 (1H, <i>br t</i> , 6.0)	C-8, C-1", C-4", C-5"

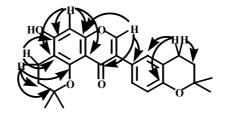
Table 39 (Continued)

Position	$\delta_{_{\! \mathrm{C}}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle ext{H}}$ (multiplicity, $J_{\scriptscriptstyle ext{Hz}}$)	НМВС
3''	132.64 (C)	-	-
4''	25.77 (CH ₃)	1.63 (3H, s)	C-3", C-5"
5''	17.85 (CH ₃)	1.74 (3H, s)	C-2", C-3", C-4"
1'''	29.60 (CH ₂)	3.30 (2H, br d, 6.6)	C-2', C-3', C-4', C-2''', C-3'''
2'''	122.40 (CH)	5.30 (1H, br t, 6.0)	C-2''', C-4''', C-5'''
3'''	131.73 (C)	-	-
4'''	25.77 (CH ₃)	1.67 (3H, s)	C-2''', C-3''', C-4''', C-5'''
5'''	17.82 (CH ₃)	1.67 (3H, s)	C-2''', C-3''', C-4''', C-5'''
5-OH	-	13.10 (1H, s)	C-4a, C-5

^{*} Carbon type was deduced from DEPT experiments.

GD16: 7-Hydroxy-2",2"-dimethylchromano(5",6":6,5)-2"",2""-dimethylchromano(5"',6":3',4')isoflavone

GD16 is a yellow solid, m.p. 178-180 $^{\circ}$ C. Its molecular formula of $C_{25}H_{26}O_5$ were established on the basis of HRMS ($[M]^+$ m/z 406.1733). The UV spectrum showed maxima absorptions of an isoflavone type at 328.4 and 263.4 nm. The IR spectrum showed the stretchings of a hydroxyl group at 3375 cm⁻¹ and a carbonyl group at 1652 cm⁻¹. The ¹H NMR spectrum (Table 40) showed signals of a characteristic isoflavone proton at δ 7.67 (s, H-2). An ABX pattern of aromatic protons H-2', H-5' and H-6' was present at δ 7.27 (d, J = 1.7 Hz), 6.73 (d, J = 8.5 Hz) and 7.15 (dd, J = 8.5 and 1.7 Hz), respectively whereas an isolated aromatic proton H-8 was at δ 6.42. The presence of two dimethylchroman rings were suggested from the resonances of four methyl groups at δ 1.33 (s, 2"'-Me₂), 1.39 (s, 2"-Me₂) and four methylene groups at δ 1.80 (t, J = 3.3 Hz, H-3'''), 1.82 (t, J = 3.3 Hz, H-3''), 2.66 (t, J = 6.6 Hz, H-4''), 2.80 (t, J = 6.6 Hz, H-4'''). The HMBC correlations of H-4'' to C-5, C-6, C-7, C-2", C-3" and H-4" to C-2', C-3', C-4', C-2", C-3" indicated that the dimethylchroman rings were fused to the isoflavone nucleus at C-5, C-6 and C-3', C-4', respectively. A carbonyl group was indicated from a carbon resonance at δ 177.50 and a stretching band in the IR spectrum at 1652 cm⁻¹. The complete HMBC data confirmed the structure of GD16 as 7-hydroxy-2",2"-dimethylchromano(5",6":6,5)- $2^{\prime\prime\prime},2^{\prime\prime\prime}$ -dimethylchromano $(5^{\prime\prime\prime},6^{\prime\prime\prime}:3^\prime,4^\prime)$ isoflavone. This compound appears to be novel.



 $Table\ 40\ \ \text{The NMR spectral data of }GD16$

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
2	149.63 (CH)	7.67 (1H, s)	C-4, C-8a, C-1'
3	117.42 (C)	-	-
4	177.50 (C=O)	-	-
4a	107.38 (C)	-	-
5	157.00 (C)	-	-
6	105.50 (C)	-	-
7	161.90 (C)	-	-
8	93.89 (CH)	6.42 (1H, s)	C-4, C-4a, C-6, C-7, C-8a
8a	159.88 (C)	-	-
1'	122.50 (C)	-	-
2'	130.80 (CH)	7.27 (1H, d, 1.7)	C-4', C-6', C-4'''
3'	120.50 (C)	-	-
4 ′	154.90 (C)	-	-
5'	116.87 (CH)	6.73 (1H, d, 8.5)	C-1', C-3', C-4'
6′	128.20 (CH)	7.15 (1H, dd, 8.5, 1.7)	C-2', C-4'
2''	75.25 (C)	-	-
3''	31.40 (CH ₂)	1.82 (2H, t, 3.3)	C-6, C-2", C-4", 2"-(CH ₃) ₂
4''	17.12 (CH ₂)	2.66 (2H, t, 6.6)	C-5, C-6, C-7, C-2", C-3"

Table 40 (Continued)

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2'''	74.27 (C)	-	-
3′′′	32.82 (CH ₂)	1.80 (2H, t, 3.3)	C-3', C-2''', C-4''', 2'''-(CH ₃) ₂
4'''	22.47 (CH ₂)	2.80 (2H, t, 6.6)	C-2', C-3', C-4', C-2''', C-3'''
2"-(CH ₃) ₂	26.56 (CH ₃)	1.39 (6H, s)	C-2", 2"-(CH ₃) ₂ , C-3"
2'''-(CH ₃) ₂	26.86 (CH ₃)	1.33 (6H, s)	C-2''', 2'''-(CH ₃) ₂ , C-3'''

^{*} Carbon type was deduced from DEPT experiments.

GD17: 1,3,6-Trihydroxy-2-(3-methyl-2-butenyl)-2"-methyl-2"-(4-methyl-3-pentenyl)chromeno(5",6":8,7)xanthone

GD17 is a yellow solid, m.p. 93-95 °C. The ¹H NMR spectrum (Table 41) exhibited three singlet signals of a hydrogen bonded hydroxy group 1-OH (δ 13.70) and two non chelated hydroxy groups (δ 6.41 and 6.26). The spectrum also showed two sharp singlet signals of two isolated aromatic protons H-4 and H-5 (δ 6.31 and 6.82). The placement of H-4 and H-5 were supported by the cross peaks of H-4 to C-2, C-3, C-4a, C-9a and H-5 to C-6, C-7, C-8a, C-9, C-10a. The signals of two methyl groups (δ 1.77, H-4' and 1.84, H-5'), methylene protons (δ 3.45, H-1') and an olefinic methine proton (δ 5.30, H-2') were the signals of a prenyl moiety. In HMBC the correlation of H-1' to C-1, C-2 and C-3 suggested that the prenyl unit was at C-2. Two doublet resonances at δ 8.09 and 5.79 were assigned for two vicinal protons H-4" and H-3" of a chromene ring. The chemical shift of H-4" suggested that the chromene ring was nearby the carbonyl group. Irradiation of an olefinic proton (H-3 $^{\prime\prime}$, δ 5.79) in the NOE experiment resulted in an enhancement of the signals of methylene protons (H-5", δ 1.82-1.78) and signal of methyl protons (2"-CH₃, δ 1.46). The evidences indicated that the chromene ring consisted of a methyl group and CH₂-R as shown in below.

The remaining resonances: two *singlet* signals of two methyl groups at δ 1.58 (H-9") and 1.67 (H-10"), a *triplet* signal of an olefinic proton (H-7") at δ 5.09 and a *multiplet* signal of methylene protons (H-6") at δ 2.13 were assigned for a prenyl unit. Irradiation at olefinic proton H-7" (δ 5.09) in a NOE experiment gave the enhancement of the signal of H-6" (δ 2.13). Irradiation at the methylene protons H-6" effected the signals of the methylene protons H-5" (δ 1.82-1.78). These results confirmed that the R group was a prenyl unit. The arrangement of the aromatic protons and substituent unit was confirmed by HMBC. The ¹³C NMR and DEPT experiments were shown in **Table 41**. **GD17** was then identified as 1,3,6-trihydroxy-2-(3-methyl-2-butenyl)-2"-methyl-2"-(4-methyl-3-pentenyl)chromeno(5",6":8,7)xanthone. The proposed structure, the spectral data and melting point was corresponded to cowagarcinone D (Chairerk, 2001).

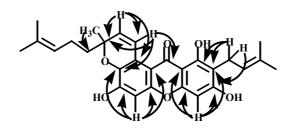


Table 41 The NMR spectral data of GD17

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
1	160.46 (C)	-	-
2	108.36 (C)	-	-
3	161.76 (C)	-	-
4	93.39 (CH)	6.31 (1H, s)	C-2, C-3, C-4a, C-9a
4a	155.31 (C)	-	-

Table 41 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)*	δ_{H} (multiplicity, J_{Hz})	HMBC
5	102.30 (CH)	6.82 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	150.74 (C)	-	-
7	136.79 (C)	-	-
8	119.56 (C)	-	-
8a	108.39 (C)	-	-
9	182.53 (C=O)	-	-
9a	103.78 (C)	-	-
10a	153.00 (C)	-	-
1'	21.43 (CH ₂)	3.45 (2H, d, 7.0)	C-1, C-2, C-3, C-2', C-3'
2'	121.37 (CH)	5.30 (1H, br t, 6.6)	C-1', C-4', C-5'
3'	135.74 (C)	-	-
4'	25.84 (CH ₃)	1.77 (3H, s)	C-2', C-3', C-5'
5′	17.92 (CH ₃)	1.84 (3H, s)	C-2', C-3', C-4'
2''	79.37 (C)	-	-
3''	131.43 (CH)	5.79 (1H, d, 9.9)	C-2", C-4"
4''	121.45 (CH)	8.09 (1H, d, 9.9)	C-7, C-2"
5''	40.36 (CH ₂)	1.82-1.78 (2H, m)	C-2", C-6"
6''	22.75 (CH ₂)	2.13 (2H, <i>m</i>)	C-7", C-8"
7''	123.64 (CH)	5.09 (1H, t, 6.6)	C-5", C-6", C-9"
8''	132.15 (C)	-	-
9''	17.66 (CH ₃)	1.58 (3H, s)	C-7", C-8", C-10"
10''	25.67 (CH ₃)	1.67 (3H, s)	C-7", C-8", C-9"
1-OH	-	13.70 (1H, s)	C-1, C-2, C-9a
3-ОН	-	6.26 (1H, s)	-
6-OH	-	6.41 (1H, s)	C-5, C-6, C-7, C-10a
2"-CH ₃	25.64 (CH ₃)	1.46 (3H, s)	C-2", C-3", C-5"

^{*} Carbon type was deduced from DEPT experiments.

GD18: 2-Hydroxy-1,2,3-propanetricarboxylic acid-1,3-dimethyl ester

$$\begin{array}{c|ccccc} H_{b} & O & & \\ H_{a} & C & & \| & \\ & & C & \\ & & O & \\ & & O & \\ & & \| & \\ HO & C & C & OH \\ & & & \\ & & & C & \\ & & & & \\ & & &$$

GD18 is a white solid, m.p. 110-112 °C. The ¹H NMR spectrum (**Table 42**) showed a sharp *singlet* of two methoxy groups 1-COOC \underline{H}_3 and 3-COOC \underline{H}_3 at δ 3.67 (6H) and two *doublets* of two *gem*-methylene protons H_a -1, H_a -3 and H_b -1, H_b -3 at δ 2.91 (2H, J = 15.5 Hz) and 2.83 (2H, J = 15.5 Hz). The correlations of 1-COOC \underline{H}_3 to 1-C=O; 3-COOC \underline{H}_3 to 3-C=O; H-1 to 1-C=O, C-2, C-3, 2-C=O and H-3 to C-1, C-2, 2-C=O, 3-C=O confirmed the positions of two methoxy groups and two methylene protons. The ¹³C NMR spectral data showed the signals of three carbonyl carbons (δ 175.25 and 170.48 x2), a quaternary carbon (δ 72.64), two methylene carbons (δ 42.80 x2) and two methyl carbons (δ 51.84 x2). Accordingly the structure of **GD18** was proposed to be 2-hydroxy-1,2,3-propanetricarboxylic acid-1,3-dimethyl ester. It was known as 1,5-dimethoxy citrate (Pyo, *et al.*, 2000).

Major HMBC correlations of GD18

Table 42 The NMR spectral data of GD18

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
1-C=O, 3-C=O	170.48 (C=O)	-	-
1	42.80 (CH ₂)	H_a : 2.91 (1H, d, 15.5) ^a	C-2, C-3, 1-C=O,
		H_b : 2.83 (1H, d, 15.5) ^b	2-C=O
2	72.64 (C)	-	-
3	42.80 (CH ₂)	H_a : 2.91 (1H, d, 15.5) ^a	C-1, C-2, 2-C=O,
		H_b : 2.83 (1H, d, 15.5) ^b	3-C=O
2-C=O	175.25 (C=O)	-	-
1-COO <u>CH</u> ₃ , 3-COO <u>CH</u> ₃	51.84 (CH ₃)	3.67 (6H, s)	1-C=O, 3-C=O

^{*} Carbon type was deduced from DEPT experiments.

^{a, b} Assignment maybe interchangeable.

GD19: 8-β-D-Glucopyranosyl-5,7-dihydroxy-2-(4-hydroxyphenyl)-4H-1benzopyran-4-one (vitexin or apigenin 8-*C*-glucoside)

GD19 is a yellow solid, m.p. 258-259 °C. The ¹H NMR spectrum (**Table 43**) indicated a flavone nucleus by the appearance of an α -methine proton at δ 6.78 (s, H-3). This proton was confirmed from the results of ³J cross peak of H-3 to C-4a, C-1' and ²J of H-3 to C-2, C-4. A *singlet* signal of a chelated hydroxy group was displayed at δ 13.15 (5-OH). A singlet signal of aromatic proton H-6 was shown at δ 6.27. Two doublets (J = 8.4 Hz) with integrating for two protons each at $\delta 8.02 \text{ (H-2')}$ and H-6') and 6.90 (H-3' and H-5') suggested the presence of a 1,4-disubstituted benzene nucleus. The resonances of a sugar moiety were observed at δ 4.67-3.25 whereas an anomeric proton resonated at δ 4.67 (d, H-1"). The coupling constant of 9.6 Hz suggested that it was a β -glucose moiety. According to HMBC correlation of H-1" (δ 4.67) to C-8 (δ 105.06) indicated the location of the glucose moiety at C-8. Therefore **GD19** 8- β -D-glucopyranosyl-5,7-dihydroxy-2-(4was characterized as hydroxyphenyl)-4H-1-benzopyran-4-one or known as vitexin or apigenin 8-Cglucoside (Evans, et al., 1957; Kartnig, et al., 1991).

 $Table \ 43 \ \ The \ NMR \ spectral \ data \ of \ GD19$

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
2	163.06 (C)	-	-
3	102.90 (CH)	6.78 (1H, s)	C-2, C-4, C-4a, C-1'
4	182.56 (C=O)	-	-
4a	104.50 (C)	-	-
5	161.60 (C)	-	-
6	98.60 (CH)	6.27 (1H, s)	C-4a, C-5, C-8
7	156.00 (C)	-	-
8	105.06 (C)	-	-
8a	160.87 (C)	-	-
1'	122.07 (C)	-	-
2', 6'	129.43 (CH)	8.02 (2H, d, 8.4)	C-2, C-2',6', C-4'
3', 5'	116.29 (CH)	6.90 (2H, d, 1.8)	C-1', C-3',5', C-4'
4'	164.50 (C)	-	-
Glc-1"	73.83 (CH)	4.67 (1H, d, 9.6)	C-7, C-8, C-8a, C-3''
2''	79.11 (CH)	3.52-3.40 (1H, <i>m</i>)	C-2"
3''	135.34 (CH)	3.52-3.40 (1H, <i>m</i>)	C-4", C-5"

Table 43 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
4''	82.29 (CH)	3.25 (1H, m)	C-4", C-5"
5''	25.79 (CH)	3.82 (1H, t, 8.7)	C-3''
6''	61.75 (CH ₂)	3.85-3.84 (2H, m)	C-4", C-5"
5-OH	-	13.15 (1H, s)	C-4a, C-5, C-6

^{*} Carbon type was deduced from DEPT experiments.

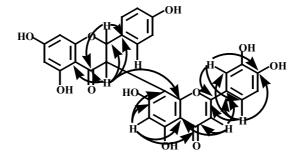
GD20: 5,7,4′,5″,7″,3″″,4″″-Heptahydroxy-[3,8″]-flavonylflavanone (morelloflavone)

GD20 is yellow crystals, m.p. 305-307 °C. EIMS showed the molecular ion of 556 which corresponded to $C_{30}H_{20}O_{11}$. Duplication of signals (in a ratio of 1: 0.38) in the ¹H NMR spectrum at 295 K suggested the existence of two conformers. The major reason led to the existence of such conformers being rotational restrictions about the interflavanyl C-3, C-8" bond. The minor signals given at 295 K sligthly disappeared at high temperature (340 K and 360 K) and only one set of signals was observed at 380 K. The ¹H NMR spectrum at 380 K (**Table 44**) showed two *singlet* signals of two chelated hydroxy protons 5-OH and 5"-OH at δ 12.80 and 12.00, two doublet resonances of H-2', H-6' and H-3', H-5' at δ 7.12 (2H, d, J = 8.4 Hz) and 6.54 (2H, d, J = 8.4 Hz), a broad singlet of H-6 and H-8 at δ 6.02 (2H), two doublets of H-2 and H-3 at δ 5.76 (J = 12.0 Hz) and 4.90 (J = 12.0 Hz), a singlet signal of a vinylic proton H-3" at δ 6.47 and a *singlet* signal of an isolated aromatic proton H-6" at δ 6.24. The resonances of aromatic protons H-2", H-6" and H-5" were shown as ABX pattern at δ 7.32 (d, J = 2.4 Hz), 7.21 (dd, J = 8.4 and 2.4 Hz) and 6.84 (d, J =8.4 Hz), respectively. The stereochemistry at C-2, C-3 positions is trans as judged from the *J*-value ($\delta_{\text{H-2}}$ 5.76, J = 12.0 Hz and $\delta_{\text{H-3}}$ 4.90, J = 12.0 Hz). ¹³C NMR spectrum at 380 K showed twenty eight signals (Table 44). The HMBC correlations at 380 K of H-2 to C-1', C-4; H-3 to C-2, C-4, C-1', C-7", C-8", C-8a" and H-6" to C-4", C-

4a", C-8" suggested that C-2 was connected to C-1' and C-3 was connected to C-8". In addition, the correlation of H-3" to C-4", C-4a", C-1" indicated that C-2" was connected to C-1".

The ¹³C NMR spectrum at 295 K also showed duplicate double signals of fifty-six carbons. The signals of major conformer (Table 45) showed two singlet signals of two chelated hydroxy protons 5-OH and 5"-OH at δ 13.10 and 12.30, two doublet resonances of H-2', H-6' and H-3', H-5' at δ 7.17 (2H, J = 8.4 Hz) and 6.41 (2H, J = 8.4 Hz), respectively. The spectrum further showed two singlets of H-6 and H-8 at δ 5.99 and 6.06, two *doublets* of H-2 and H-3 at δ 5.73 (J = 12.0 Hz) and 4.92 (J = 12.0 Hz), a singlet signal of a vinylic proton H-3" at δ 6.60 and a singlet signal of an isolated aromatic proton H-6" at δ 6.26. The resonances of aromatic protons H-2", H-5" and H-6" were shown as an ABX pattern at δ 7.45 (d, J = 3.4 Hz), 6.92 (d, J = 8.4 Hz) and 7.27 (dd, J = 8.4 and 3.4 Hz), respectively. The HMBC at 295 K showed the correlations of H-2 to C-1', C-4 and H-3 to C-2, C-4, C-4a, C-1', C-8a". It's therefore confirmed that C-2 was connected to the 4-substituted ring at C-1' and C-3 was connected to another flavone nucleus at C-8". In addition, the correlations of H- $6^{\prime\prime}$ to C-4 $^{\prime\prime}$, C-4 $a^{\prime\prime}$, C-7 $^{\prime\prime}$, C-8 $^{\prime\prime}$ and H-3 $^{\prime\prime}$ to C-4 $^{\prime\prime}$, C-4 $a^{\prime\prime}$, C-1 $^{\prime\prime\prime}$ supported the assignment. The signals of the minor conformer showed two singlet signals of two chelated hydroxy protons 5-OH and 5"-OH at δ 12.95 and 12.15, two doublet resonances of H-2', H-6' and H-3', H-5' at δ 7.12 (d, J = 8.4 Hz) and 6.41 (d, J = 8.4 Hz), two broad singlets of H-6 and H-8 at δ 5.99 and 6.60, two doublets of H-2 and H-3 at δ 5.60 (J = 12.0 Hz) and 5.01 (J = 12.0 Hz), a singlet signal of a vinylic proton H-3" at δ 6.60 and a *singlet* signal of an isolated aromatic proton H-6" at δ 6.26. The resonances of aromatic protons H-2", H-5" and H-6" were shown as an ABX pattern at δ 7.45 (d, J = 3.4 Hz), 6.99 (d, J = 8.4 Hz) and 7.27 (dd, J = 8.4 and 3.4 Hz), respectively. The HMBC of the minor conformer showed the same pattern of correlations as of the major conformer.

Acetylation of **GD20** with pyridine-acetic anhydride gave a pentaacetate **GD20(A)**, m.p. 188-190 °C. The ¹H NMR spectrum (**Table 46**) showed a *singlet* signal of a chelated hydroxy proton 5"-OH at δ 12.75, two *doublet* resonances of H-2, H-6 and H-3, H-5 at δ 7.41 (2H, d, J = 8.5 Hz) and 7.06 (2H, d, J = 8.5 Hz), respectively. The spectra showed two *doublets* of H-3' and H-5' at δ 6.62 (J = 2.0 Hz) and 6.60 (J = 2.0 Hz), respectively, a *singlet* signal of a vinylic proton H-3" at δ 6.99 and a *singlet* signal of an isolated aromatic proton H-6" at δ 6.98. The resonances of aromatic protons H-2", H-5" and H-6" were shown as an ABX pattern at δ 7.39 (d, J = 1.7 Hz), 7.29 (d, J = 8.5 Hz) and 7.21 (dd, J = 8.5 and 1.7 Hz), respectively. The interesting observation was the disappearence of two *doublet* signals at δ 5.76 (J = 12.0 Hz) and 4.90 (J = 12.0 Hz) due to H-2 and H-3 of **GD20**, respectively and instead a downfield *singlet* was seen at δ 6.63 (H- β). During acetylation, opening of the chromanone ring had occurred transforming the flavanone to the corresponding chalcone. The integration for five acetoxy groups was observed.



Major HMBC correlations of GD20

Structure of GD20(A) Major

HMBC correlations of GD20(A)

Table 44 The NMR spectral data of **GD20** at 380 K

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
2	82.01 (CH)	5.76 (1H, d, 12.0)	C-4, C-2',6', C-8
3	49.06 (CH)	4.90 (1H, d, 12.0)	C-2, C-4, C-1', C-7", C-8", C-8a"
4	196.47 (C=O)	-	-
4a	102.40 (C)	-	-
5	164.41 (C)	-	-
6	95.92 (CH)	6.02 (1H, s)	C-4, C-4a, C-5, C-7, C-8
7	167.16 (C)	-	-
8	97.03 (CH)	6.02 (1H, s)	C-4, C-7, C-8a
8a	163.51 (C)	-	-
1'	128.87 (C)	-	-
2', 6'	128.70 (CH)	7.12 (2H, d, 8.4)	C-2, C-2',6', C-3',5', C-4'
3', 5'	115.32 (CH)	6.54 (2H, d, 8.4)	C-2',6', C-3',5', C-4'
4'	158.01 (C)	-	-
2''	146.33 (C)	-	-
3''	103.51 (CH)	6.47 (1H, s)	C-4", C-4a", C-5", C-1"
4''	182.15 (C=O)	-	-
4a''	104.25 (C)	-	-

Table 44 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
5''	164.41 (C)	-	-
6''	99.28 (CH)	6.24 (1H, s)	C-4", C-4a", C-8"
7''	162.50 (C)	-	-
8''	101.20 (C)	-	-
8a''	155.90 (C)	-	-
1'''	122.49 (C)	-	-
2'''	114.32 (CH)	7.32 (1H, d, 2.4)	C-2", C-3"", C-4"", C-6""
3'''	161.14 (C)	-	-
4'''	150.14 (C)	-	-
5'''	116.91 (CH)	6.84 (1H, d, 8.4)	C-1''', C-4'''
6'''	119.38 (CH)	7.21 (1H, dd, 8.4, 2.4)	C-2''', C-3''', C-4'''
5-OH	-	12.80 (1H, s)	-
5′′-ОН	-	12.00 (1H, s)	-

^{*} Carbon type was deduced from DEPT experiments.

Table 45 The NMR spectral data of **GD20** at 295 K

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*		$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)		IIMDC
rosition	major	minor	major	minor	НМВС
2	81.38 (CH)	82.28	5.73 (1H, d, 12.0)	5.60	C-4, C-1'
3	49.06 (CH)	48.74	4.92 (1H, d, 12.0)	5.01	C-2, C-4, C-4a, C-1', C-8a''
4	196.68 (C=O)	196.91	-	-	-
4a	102.15 (C)	101.96	-	-	-
5	164.26 (C)	164.34	-	-	-
6	95.78 (CH)	95.78	5.99 (1H, s)	6.06	C-4, C-5, C-7, C-8
7	167.01 (C)	167.44	-	-	-

Table 45 (Continued)

D = = 141 =	$\delta_{\rm C}({ m C-Type})^*$		$\delta_{_{ m H}}$ (multiplicity, .	$J_{ m Hz}$)	IIMDC
Position	major	minor	major	minor	HMBC
8	96.70 (CH)	96.70	6.06 (1H, s)	5.99	C-4, C-7
8a	163.32 (C)	162.97	-	-	-
1'	128.62 (C)	128.10	-	-	-
2'	128.98 (CH)	129.32	7.17 (1H, d, 8.4)	7.12	C-2, C-4', C-5', C-6'
3'	114.89 (CH)	115.62	6.41 (1H, d, 8.4)	6.41	C-2', C-4', C-5'
4'	157.79 (C)	158.03	-	-	-
5'	114.89 (CH)	115.09	6.41 (1H, d, 8.4)	6.41	C-3', C-4', C-6'
6'	128.98 (CH)	128.10	7.17 (1H, d, 8.4)	7.12	C-2, C-2', C-4', C-5'
2''	146.10 (C)	146.44	-	-	-
3''	102.68 (CH)	103.50	6.60 (1H, s)	6.60	C-4", C-4a", C-1""
4''	182.15 (C=O)	182.12	-	-	-
4a''	103.62 (C)	104.12	-	-	-
5''	163.94 (C)	163.94	-	-	-
6''	99.08 (CH)	98.45	6.26 (1H, s)	6.26	C-4", C-4a", C-7", C-8"
7''	162.14 (C)	162.14	-	-	-
8''	101.01 (C)	100.52	-	-	-
8a''	155.71 (C)	155.01	-	-	-
1'''	121.50 (C)	121.99		-	-
2'''	113.72 (CH)	114.22	7.45 (1H, d, 3.4)	7.45	C-2", C-4"", C-6""
3'''	160.98 (C)	160.71	-	-	-
4'''	150.15 (C)	150.15	-	-	-
5'''	116.61 (CH)	115.62	6.92 (1H, d, 8.4)	6.99	C-1''', C-4'''
6'''	119.78 (CH)	118.49	7.27(1H,dd,8.4, 3.4)	7.27	C-2", C-1"", C-2"", C-4""
5-OH	-	-	13.10 (1H, s)	12.95	-
5′′-ОН	-	-	12.30 (1H, s)	12.15	C-4a", C-5", C-6"

^{*} Carbon type was deduced from DEPT experiments.

Table 46 The NMR spectral data of GD20(A)

Position	$\delta_{_{\mathrm{C}}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
1	135.76 (C)	-	-
2, 6	128.86 (CH)	7.41 (2H, d, 8.5)	C-2,6, C-4
3, 5	122.02 (CH)	7.06 (2H, d, 8.5)	C-1, C-3,5, C-4
4	151.49 (C)	-	-
α	108.10 (C)	-	-
β	108.90 (CH)	6.63 (1H, s)	C-2, C-α, C-7', C-8"
1'	107.30 (C)	-	-
2'	159.30 (C)	-	-
3'	109.20 (CH)	6.62 (1H, d, 2.0)	C-1', C-4', C-5'
4'	153.20 (C)	-	-
5'	110.06 (CH)	6.60 (1H, d, 2.0)	C-1', C-3', C-4', C-6'
6'	149.62 (C)	-	-
7'	182.62 (C=O)	-	-
2''	162.63 (C)	-	-
3''	77.25 (CH)	6.99 (1H, s)	C-4a", C-1""
4''	182.62 (C=O)	-	-
4a''	107.78 (C)	-	-
5''	151.35 (C)	-	-
6''	96.05 (CH)	6.98 (1H, s)	C-4a", C-5", C-7", C-8"
7''	145.48 (C)	-	-
8''	107.20 (C)	-	-
8a''	159.80 (C)	-	-
1'''	129.49 (C)	-	-
2'''	121.64 (CH)	7.39 (1H, d, 1.7)	C-2", C-3"", C-4"", C-6""
3'''	144.78 (C)	-	-
4'''	145.00 (C)	-	-
5'''	124.42 (CH)	7.29 (1H, d, 8.5)	C-1''', C-3''', C-4'''

Table 46 (Continued)

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
6'''	124.75 (CH)	7.21 (1H, dd, 8.5, 1.7)	C-2", C-2"", C-4""
5′′-ОН	-	12.75 (1H, s)	C-4a'', C-6''
5xOCOCH ₃	169.29 (C=O)	-	-
	168.94 (C=O)	-	-
	168.47 (C=O)	-	-
	167.82 (C=O)	-	-
	167.67 (C=O)	-	-
5xOCO <u>CH</u> ₃	21.13 (CH ₃)	2.25 (3H, s)	C=O
		2.24 (3H, s)	C=O
	21.00 (CH ₃)	2.49 (3H, s)	C=O
	20.70 (CH ₃)	2.34 (3H, s)	C=O
	20.63 (CH ₃)	2.37 (3H, s)	C=O

^{*} Carbon type was deduced from DEPT experiments.

GD21: 1,3-Dihydroxy-2,6-bis(3-methyl-2-butenyl)-2,2-dimethylchromeno (5"'',6"':8,7)xanthone

GD21 is a yellow solid, m.p. 119-120 °C. The HRMS showed the molecular ion of 446.1628 [M]⁺ corresponded to C₂₈H₂₀O₅. The UV spectrum showed maxima absorptions at 329.8, 300.5, 287.9, 265.0, 246.6 and 207.9 nm. The IR spectrum exhibited the absorption bands of O-H stretching at 3402 cm⁻¹ and C=O stretching at 1622 cm⁻¹. The ¹H NMR spectrum (**Table 47**) exhibited the resonances of a chelated hydroxy proton 1-OH (δ 13.70, s), H-4 (δ 6.26, s) and H-5 (δ 6.82, s). The assignment of H-4 was confirmed by the correlation of H-4 to C-2, C-3, C-4a and C-9a whereas the position of H-5 was proved by the correlation of H-5 to C-6, C-7, C-8a, C-9 and C-10a. The signals of two olefinic protons at δ 5.27 (br t, H-2') and 5.30 (br t, H-2''), two benzylic methylene protons at $\delta 3.57$ (d, H-1') and 3.46 (d, H-1'') and four methyl groups at δ 1.84 (H-4'), 1.50 (H-5'), 1.77 (H-4") and 1.87 (H-5") indicated the presence of two prenyl groups. The HMBC correlations of H-1' to C-1, C-2 and C-3 and H-1" to C-5, C-6 and C-7 supported that two prenyl groups were at C-2 and C-6. The signals of two methyl groups at δ 1.50 (6H, s, 2"'-Me₂) and vicinal olefinic protons at δ 5.77 (d, J = 10.2 Hz, H-3''') and 7.99 (d, J = 10.2 Hz, H-4''') associated with a chromene ring were present. The HMBC correlations of H-4" to C-2" and H-3" to C-8, C-2" and C-4" indicated that the chromene ring was fused to the xanthone nucleus at C-7 and C-8. These assignment indicated that GD21 was 1,3dihydroxy-2,6-bis(3-methyl-2-butenyl)-2,2-dimethylchromeno($5^{\prime\prime\prime}$, $6^{\prime\prime\prime}$:8,7)xanthone. This compound was a new xanthone derivative.

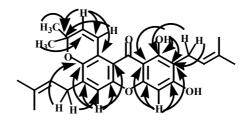


Table 47 The NMR spectral data of GD21

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	157.79 (C)	-	-
2	104.39 (C)	-	-
3	159.97 (C)	-	-
4	94.28 (CH)	6.26 (1H, s)	C-2, C-3, C-4a, C-9a
4a	156.54 (C)	-	-
5	102.42 (CH)	6.82 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	136.85 (C)	-	-
7	150.93 (C)	-	-
8	119.70 (C)	-	-
8a	108.56 (C)	-	-
9	182.47 (C=O)	-	-
9a	103.89 (C)	-	-
10a	153.07 (C)	-	-
1'	21.45 (CH ₂)	3.57 (2H, d, 7.2)	C-1, C-2, C-3
2'	121.07 (CH)	5.27 (1H, br t, 7.5)	C-2, C-4', C-5'

Table 47 (Continued)

Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
3'	131.33 (C)	-	-
4′	25.81 (CH ₃)	1.84 (3H, s)	C-2', C-3', C-5'
5'	17.98 (CH ₃)	1.68 (3H, s)	C-2', C-3', C-4'
1''	22.58 (CH ₂)	3.46 (2H, d, 7.2)	C-5, C-7, C-2", C-3"
2''	121.50 (CH)	5.30 (1H, br t, 7.5)	C-6, C-4", C-5"
3''	132.64 (C)	-	-
4''	25.86 (CH ₃)	1.77 (3H, s)	C-2", C-3", C-5"
5''	17.94 (CH ₃)	1.87 (3H, s)	C-2", C-3", C-4"
2'''	77.99 (C)	-	-
3'''	132.34 (CH)	5.77 (1H, d, 10.2)	C-8, C-2''', C-4'''
4'''	120.95 (CH)	7.99 (1H, d, 10.2)	C-8a, C-2'''
2'''-(CH ₃) ₂	27.35 (CH ₃)	1.50 (6H, s)	C-2''', C-3'''
1-ОН	-	13.70 (1H, s)	C-1, C-2, C-9a
3-ОН	-	6.35 (1H, s)	C-2

^{*} Carbon type was deduced from DEPT experiments.

GD22: 2,4,6-Trihydroxy-3,5-bis(3-methyl-2-butenyl)benzophenone (clusiaphenone B)

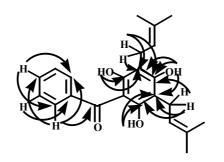


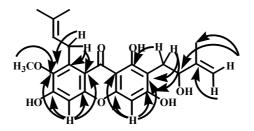
Table 48 The NMR spectral data of GD22

Position	$\delta_{_{ m C}}$ (C-Type)*		$\delta_{_{ m H}}$ (multi	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	
rosition	GD22	clusiaphenone B	GD22	clusiaphenone B	HMBC
1	104.58 (C)	104.56	-	-	-
2, 6	157.65 (C)	157.62	-	-	-
3, 5	106.33 (C)	106.33	-	-	-
4	161.07 (C)	161.03	-	-	-
1'	140.31 (C)	140.28	-	-	-
2', 6'	129.02 (CH)	128.98	7.66-7.62(2H, <i>m</i>)	7.64-7.46	C-3',5', C-4', C=O
3', 5'	127.95 (CH)	127.93	7.55-7.47(2H,m)	$\left.\right $ (5H, m)	C-1', C-2',6'
4'	132.04 (CH)	132.00	7.60-7.54(1H, <i>m</i>)	J	C-2',6'
1", 1""	21.84 (CH ₂)	21.82	3.34 (4H, <i>d</i> ,7.0)	3.33 (4H, d, 7.0)	C-2,6, C-3,5, C-4,
					C-2",2"", C-3",3""
2", 2""	121.85 (CH)	121.83	5.22(2H, brt,6.6)	5.22 (2H, br t)	C-1",1"", C-4",4"",
					C-5",5"
3", 3""	135.06 (C)	134.98	-	-	-
4", 4""	25.80 (CH ₃)	25.77	1.74 (6H, s)	1.73 (6H, s)	C-2",2"", C-5",5""
5", 5""	17.87 (CH ₃)	17.85	1.78 (6H, s)	1.78 (6H, s)	C-2",2"", C-4",4""
C=O	198.05	198.03	-	-	-
2-OH	-	-	8.92 (1H, s)	8.92 (1H, s)	C-2, C-3
4-OH	-	-	6.37 (1H, s)	6.37 (1H, s)	C-5, C-6
6-OH	-	-	8.92 (1H, s)	8.92 (1H, s)	C-3, C-4, C-5

^{*} Carbon type was deduced from DEPT experiments.

GD23: 1,3,6-Trihydroxy-2-(2-hydroxy-3-methyl-3-butenyl)-7-methoxy-8-(3-methyl-2-butenyl)xanthone (mangostenol)

GD23 was isolated as a yellow solid with m.p. 163-164 °C. The ¹H NMR spectrum (**Table 49**) showed four *singlet* signals of a deshielded hydroxy proton 1-OH (δ 13.90), methoxy protons (δ 3.79), H-4 (δ 6.37) and H-5 (δ 6.83). A prenyl side chain and a 2-hydroxy-3-methyl-3-butenyl moiety were detected. The resonances of the prenyl side chain appeared as follow; two singlets of two methyl groups at δ 1.88 (H-5'') and 1.70 (H-4''), a doublet of methylene protons (H-1'') at δ 4.09 and a broad triplet of an olefinic methine proton (H-2") at δ 5.27. The resonances of 2-hydroxy-3methyl-3-butenyl were appeared as follow; two singlets of methylene protons (H-4') at δ 5.00 and 4.88, a doublet of doublet of a hydroxy methine proton (H-2') at δ 4.41, two doublet of doublet signals of methylene protons (H-1') at δ 3.18 and 2.88, and a sharp singlet of methyl protons (H-5') at δ 1.83. The HMBC spectrum was used to support the structural assignment. In HMBC, H-1' showed long-range heteronuclear connectivities with C-1, C-2, C-3, C-2', C-3'; H-2' was coupled with C-2, C-1', C-3', C-4', C-5'; H-1" exhibited interactions with C-7, C-8, C-8a, C-2", C-3" and the methoxy signal (OCH₂) showed a cross-peak with C-7. Thus it was indicated that the methoxy, prenyl and 2-hydroxy-3-methyl-3-butenyl moieties were at C-7, C-8 and C-2, respectively. The ¹³C NMR spectrum and DEPT experiments showed the resonances of twelve quaternary carbons, four methine carbons, three methylene carbons, four methyl carbons and a carbonyl carbon. Therefore GD23 was assigned as 1,3,5trihydroxy-2-(2-hydroxy-3-methyl-3-butenyl)-7-methoxy-8-(3-methyl-2-butenyl) xanthone. It was the first reported xanthone in *G. dulcis*. Its physical and spectral data were in agreement with mangostenol which was previously isolated from *G. mangostana* (Suksamrarn, *et al.*, 2002).



 $Table\ 49\ \ The\ NMR\ spectral\ data\ of\ GD23$

Position	$\delta_{_{\! \mathrm{C}}}$ (C-Type)	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
1	160.74 (C)	-	-
2	107.83 (C)	-	-
3	162.94 (C)	-	-
4	93.78 (CH)	6.37 (1H, s)	C-2, C-3, C-4a, C-9, C-9a
4a	155.65 (C)	-	-
5	101.79 (CH)	6.83 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	155.56 (C)	-	-
7	143.08 (C)	-	-
8	137.29 (C)	-	-
8a	111.63 (C)	-	-
9	181.97 (C=O)	-	-
9a	103.12 (C)	-	-

Table 49 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
10a	155.40 (C)	-	-
1'	28.61 (CH ₂)	2.88 (1H, dd, 15.0, 8.1)	C-1, C-2, C-3, C-2', C-3'
		3.18 (1H, dd, 15.0, 2.1)	C-1, C-2, C-3, C-2', C-3'
2'	76.42 (CH)	4.41 (1H, d, 7.5)	C-2, C-1', C-3', C-4', C-5'
3'	147.08 (C)	-	-
4'	109.97 (CH ₂)	4.88 (1H, s)	C-2', C-3', C-5'
		5.00 (1H, s)	C-2', C-3', C-5'
5'	18.29 (CH ₃)	1.83 (3H, s)	C-2', C-3', C-4'
1''	26.32 (CH ₂)	4.09 (2H, d, 6.3)	C-7, C-8, C-8a, C-2", C-3"
2''	123.38 (CH)	5.27 (1H, br t, 6.3)	C-8, C-4", C-5"
3''	131.66 (C)	-	-
4''	25.73 (CH ₃)	1.70 (3H, s)	C-2", C-3", C-5"
5''	18.08 (CH ₃)	1.88 (3H, s)	C-2", C-3", C-4"
1-OH	-	13.90 (1H, s)	C-1, C-2, C-9a
7-OCH ₃	61.18 (CH ₃)	3.79 (3H, s)	C-7

 $[\]boldsymbol{*}$ Carbon type was deduced from DEPT experiments.

GD24: 1,3,6-Trihydroxy-2-(3-hydroxy-3-methylbutyl)-7-methoxy-8-(3-methyl-2-butenyl)xanthone (cratoxylone)

GD24 was isolated as yellow needles. The ¹H NMR spectrum (**Table 50**) exhibited the resonances of a chelated hydroxy proton 1-OH (δ 13.80, s), H-4 (δ 6.39, s), H-5 (δ 6.85, s), 7-OCH₃ (δ 3.80), a prenyl group and a 3-hydroxy-3-methylbutyl side chain. The resonances of the prenyl group were displayed at δ 5.28 (br t, H-2''), 4.14 (d, H-1''), 1.66 (s, H-4'') and 1.84 (s, H-5''). The HMBC correlation of H-1'' to C-7 and C-8 indicated that the prenyl group was at C-8. The resonances of 3-hydroxy-3-methylbutyl side chain were shown at δ 1.26 (δ H, s, H-4' and H-5'), 2.77 (δ H, sH, H-1') and 1.71 (δ H, sH, H-2'). The assignment of H-4 was confirmed by the correlation of H-4 to C-2, C-3, C-4a and C-9a whereas the position of H-5 was proved by the correlation of H-5 to C-6, C-7, C-8a and C-10a. These assignment indicated that **GD24** was 1,3,6-trihydroxy-2-(3-hydroxy-3-methylbutyl)-7-methoxy-8-(3-methyl-2-butenyl) xanthone. It was known as cratoxylone (Bennett, ϵ t ϵ L, 1993).

Major HMBC correlations of GD24

 $Table \ 50 \ \ The \ NMR \ spectral \ data \ of \ GD24$

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	163.00 (C)	-	-
2	112.37 (C)	-	-
3	161.70 (C)	-	-
4	93.40 (CH)	6.39 (1H, s)	C-2, C-3, C-4a, C-9a
4a	155.68 (C)	-	-
5	102.71 (CH)	6.85 (1H, s)	C-6, C-7, C-8a, C-10a
6	157.70 (C)	-	-
7	144.20 (C)	-	-
8	138.09 (C)	-	-
8a	111.93 (C)	-	-
9	182.89 (C=O)	-	-
9a	103.53 (C)	-	-
10a	156.23 (C)	-	-
1'	17.89 (CH ₂)	2.77 (2H, m)	C-2, C-2'
2'	43.23 (CH ₂)	1.71 (2H, m)	C-2, C-3', C-4',5'
3'	70.63 (C)	-	-
4', 5'	29.58 (CH ₃)	1.26 (6H, s)	C-2', C-3', C-4',5'
1''	26.87 (CH ₂)	4.14 (2H, d, 6.0)	C-7, C-8, C-2", C-3"
2''	124.78 (CH)	5.28 (1H, br t, 6.0)	C-8, C-1", C-4", C-5"
3''	131.43 (C)	-	-
4''	25.95 (CH ₃)	1.66 (3H, s)	C-2", C-3", C-5"
5''	18.29 (CH ₃)	1.84 (3H, s)	C-2", C-3", C-4"
1-OH	-	13.80 (1H, s)	-
7-OCH ₃	61.28 (CH ₃)	3.80 (3H, s)	C-7

^{*} Carbon type was deduced from DEPT experiments.

GD25: 1,3,6-Trihydroxy-8-(3-hydroxy-3-methylbutyl)-7-methoxy-2-(3-methyl-2-butenyl)xanthone (garcinone D)

GD25 is a yellow solid, m.p. 200-203 °C. It was an isomer of GD24. The difference was the position of side chains. The location of the prenyl group at C-2 was assigned to be at C-2 by the HMBC correlation of H-1' to C-1, C-2 and C-3 whereas the 3-hydroxy-3-methylbutyl side chain was located at C-8 by the HMBC correlation of H-1" to C-7, C-8 and C-8a. GD25 was then identified as 1,3,6-trihydroxy-8-(3-hydroxy-3-methylbutyl)-7-methoxy-2-(3-methyl-2-butenyl)xanthone. The proposed structure, the spectral data was agreed with the structure of garcinone D (Sen, *et al.*, 1986; Bennett, *et al.*, 1993).

Major HMBC correlations of GD25

 $Table \ 51 \ \ The \ NMR \ spectral \ data \ of \ GD25$

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	161.61 (C)	-	-
2	110.97 (C)	-	-
3	163.18 (C)	-	-
4	93.13 (CH)	6.40 (1H, s)	C-2, C-3, C-4a, C-9, C-9a
4a	157.60 (C)	-	-
5	102.52 (CH)	6.81 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	155.68 (C)	-	-
7	144.44 (C)	-	-
8	139.94 (C)	-	-
8a	111.83 (C)	-	-
9	182.79 (C=O)	-	-
9a	103.48 (C)	-	-
10a	156.24 (C)	-	-
1'	21.98 (CH ₂)	3.33 (2H, d, 7.2)	C-1, C-2, C-3, C-2', C-3'
2'	123.55 (CH)	5.28 (1H, <i>br t</i> , 7.0)	C-2, C-4', C-5'
3'	131.28 (C)	-	-
4'	25.90 (CH ₃)	1.65 (3H, s)	C-2', C-3', C-5'
5'	17.91 (CH ₃)	1.79 (3H, s)	C-2', C-3', C-4'
1''	23.22 (CH ₂)	3.46 (2H, <i>m</i>)	C-7, C-8, C-8a, C-2''
2''	45.65 (CH ₂)	1.77 (2H, m)	C-1", C-3"
3''	70.61 (C)	-	-
4", 5"	29.40 (CH ₃)	1.33 (6H, s)	C-2", C-3", C-4",5"
1-OH	-	13.82 (1H, s)	C-1, C-2, C-9a
7-OCH ₃	61.48 (CH ₃)	3.85 (3H, s)	C-7

^{*} Carbon type was deduced from DEPT experiments.

RD1: 1,6-Dihydroxy-7-methoxy-8-(3,7-dimethyl-2,6-octadienyl)-2',2'-dimethylchromeno(5',6':2,3)xanthone (fuscaxanthone A)

RD1 was isolated as yellow needles. The ¹H NMR spectrum (**Table 52**) showed a sharp singlet signal of a chelated hydroxyl proton (1-OH) at δ 13.72 and two singlet signals of aromatic protons H-4 and H-5 at δ 6.24 and 6.83. H-5 and H-4 were deduced from the HMBC correlations of H-5 to C-7, C-8a, C-9, C-10a and H-4 to C-2, C-4a, C-9, C-9a. A sharp singlet signal of methoxy protons was observed at δ 3.80 and the methoxy group was suggested to be at C-7, from the correlation of OCH₃ to C-7. The ¹H NMR spectrum also showed a characteristic signal of a dimethylchromene ring, of which the signal of gem-dimethyl protons was at δ 1.46 (s) and two olefinic protons (H-4' and H-3') were at δ 6.73 (d) and 5.56 (d). The correlation of H-4' to C-2 and C-1 precisely determined that the dimethylchromene ring was next to C-1. Three singlets of three methyl groups at δ 1.82 (H-9"), 1.60 (H-8") and 1.54 (H-10"), a doublet of methylene protons at $\delta 4.09$ (H-1"), two multiplets of methylene protons at δ 2.04-2.01 (H-5") and 2.01-1.98 (H-4") and two triplets of two olefinic methine protons at δ 5.25 (H-2") and 5.01 (H-6") implied the presence of a geranyl moiety. The deshielded chemical shift of the methylene protons H-1" (δ 4.09) suggested that the geranyl side chain was at peri position to the carbonyl group. The HMBC correlation of H-1" to C-7, C-8 and C-8a confirmed the location of the geranyl unit at C-8. According to the assignment, **RD1** was 1,6-dihydroxy-7-methoxy-8-(3,7-dimethyl-2,6-octadienyl)-2',2'-dimethylchromeno(5',6':2,3)xanthone. This compound was identical with fuscaxanthone A (Ito, *et al.*, 2003b) which was first isolated from *Garcinia fusca*.

Table 52 The NMR spectral data of RD1

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
1	159.82 (C)	-	-
2	104.46 (C)	-	-
3	157.93 (C)	-	-
4	94.12 (CH)	6.24 (1H, s)	C-2, C-4a, C-9, C-9a
4a	156.26 (C)	-	-
5	101.66 (CH)	6.83 (1H, s)	C-7, C-8a, C-9, C-10a
6	154.65 (C)	-	-
7	142.74 (C)	-	-
8	137.09 (C)	-	-
8a	112.17 (C)	-	-

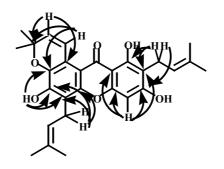
Table 52 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
9	181.94 (C=O)	-	-
9a	103.72 (C)	-	-
10a	155.71 (C)	-	-
2'	77.90 (C)	-	-
3'	127.12 (CH)	5.56 (1H, d, 9.9)	C-2, C-2', 2'-(CH ₃) ₂
4′	115.71 (CH)	6.73 (1H, d, 9.9)	C-1, C-2, C-3, C-2'
1''	26.55 (CH ₂)	4.09 (2H, d, 6.0)	C-7, C-8, C-8a, C-2", C-3", C-4"
2''	123.19 (CH)	5.25 (1H, br t, 6.0)	C-8, C-1", C-4", C-9"
3''	135.57 (C)	-	-
4''	39.69 (CH ₂)	2.01-1.98 (2H, m)	C-1", C-2", C-3", C-9"
5''	26.46 (CH ₂)	2.04-2.01 (2H, m)	C-3", C-4", C-6", C-7"
6''	124.27 (CH)	5.01 (1H, <i>br t</i> , 6.0)	C-4", C-5", C-10"
7''	131.26 (C)	-	-
8''	25.59 (CH ₃)	1.60 (3H, s)	C-6", C-7", C-10"
9''	16.47 (CH ₃)	1.82 (3H, s)	C-1", C-2", C-3", C-4", C-5"
10''	17.64 (CH ₃)	1.54 (3H, s)	C-6", C-7", C-8"
1-OH	-	13.72 (1H, s)	-
7-OCH ₃	61.97 (CH ₃)	3.80 (3H, s)	C-7
2'-(CH ₃) ₂	28.29 (CH ₃)	1.46 (6H, s)	C-2', C-3', C-4'

^{*} Carbon type was deduced from DEPT experiments.

RD2: 1,3,6-Trihydroxy-2,5-bis(3-methyl-2-butenyl)-2"',2"''-dimethylchromeno (5"'',6"'':8,7)xanthone (tovophyllin A)

RD2 is a yellow solid, m.p. 212-215 °C. The ¹H NMR spectral data (**Table** 53) demonstrated the resonances of a chelated hydroxy group 1-OH (δ 13.90, s), 6-OH $(\delta 6.32, s)$ and H-4 $(\delta 6.37, s)$. A singlet signal of two methyl groups 2"'-Me₂ $(\delta 1.52, s)$ 6H) and two doublet signals of two geminal olefinic protons H-3" (δ 5.79, J = 9.9 Hz) and H-4" (δ 8.02, J = 9.9 Hz) implied the presence of a 2,2-dimethylchromene ring. The deshielded effect on the resonance of H-4" suggested the location of the chromene ring nearby the carbonyl group. Two triplets at δ 5.32 (H-2') and 5.29 (H-2"), two doublets at δ 3.50 (H-1") and 3.60 (H-1") and four singlets at δ 1.70 (H-4"), 1.89 (H-5'), 1.80 (H-4") and 1.87 (H-5") were the typical resonances of two prenyl groups. HMBC correlations of H-1' to C-1, C-2, C-3 and H-1" to C-5, C-6, C-10a indicated that the prenyl side chains were at C-2 and C-5. The assignment of H-4 was confirmed by the correlation of H-4 to C-2, C-3, C-4a and C-9a. The position of 6-OH was proved by the correlation of 6-OH to C-5, C-6 and C-7. The ¹³C NMR spectral data showed the resonances of a carbonyl carbon, five methyl carbons, two methylene carbons, five methine carbons and fourteen quaternary carbons. RD2 was then identified as 1,3,6-trihydroxy-2,5-bis(3-methyl-2-butenyl)-2"",2""-dimethylchromeno (5"',6":8,7)xanthone. The proposed structure, the spectral data were in good agreement with the structure of tovophyllin A (Bennett, et al., 1993).



 $Table \ 53 \ \ The \ NMR \ spectral \ data \ of \ RD2$

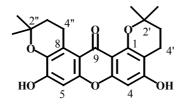
Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
1	160.80 (C)	-	-
2	106.65 (C)	-	-
3	161.80 (C)	-	-
4	94.26 (CH)	6.37 (1H, s)	C-2, C-3, C-4a, C-9a
4a	156.00 (C)	-	-
5	115.63 (C)	-	-
6	148.50 (C)	-	-
7	136.82 (C)	-	-
8	117.16 (C)	-	-
8a	119.90 (C)	-	-
9	181.50 (C=O)	-	-
9a	102.40 (C)	-	-
10a	150.90 (C)	-	-
1'	22.56 (CH ₂)	3.50 (2H, d, 5.4)	C-1, C-2, C-3, C-2', C-3'
2'	121.00 (CH)	5.32 (1H, br t, 6.9)	C-4', C-5'
3'	132.50 (C)	-	-
4'	27.43 (CH ₃)	1.70 (3H, s)	C-2', C-3', C-5'

Table 53 (Continued)

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
5′	17.97 (CH ₃)	1.89 (3H, s)	C-2', C-3', C-4'
1''	22.69 (CH ₂)	3.60 (2H, d, 7.2)	C-5, C-6, C-10a, C-2", C-3"
2''	120.50 (CH)	5.29 (1H, br t, 6.9)	C-5, C-4", C-5"
3''	132.32 (C)	-	-
4''	27.33 (CH ₃)	1.80 (3H, s)	C-2", C-3", C-5"
5''	18.17 (CH ₃)	1.87 (3H, s)	C-2", C-3", C-4"
2'''	78.00 (C)	-	-
3'''	131.55 (CH)	5.79 (1H, d, 9.9)	C-8, C-2'''
4'''	120.93 (CH)	8.02 (1H, d, 9.9)	C-7, C-2'''
1-OH	-	13.90 (1H, s)	C-1, C-2, C-9a
6-OH	-	6.32 (1H, s)	C-5, C-6, C-7
2""-(CH ₃) ₂	29.69 (CH ₃)	1.52 (6H, s)	C-2"', C-3"', 2"'-(CH ₃) ₂

^{*} Carbon type was deduced from DEPT experiments.

RD3: 3,6-Dihydroxy-2',2'-dimethylchromano(5',6':2,1)-2'',2''Dimethylchromano(5''',6''':8,7)xanthone (isonormangostin)



RD3 is a yellow solid, m.p. 210-212 $^{\circ}$ C. Its molecular formula of $C_{23}H_{24}O_6$ was established on the basis of mass spectrum ($[M]^{+}$ m/z 396.1571). The UV spectrum showed maximum absorptions bands at 355.3, 307.5, 254.5 and 212.2 nm. In the IR spectrum, absoption bands of O-H stretching and C=O stretching were shown at 3424 and 1609 cm⁻¹, respectively. The ¹H NMR spectrum (**Table 54**) contained two sharp singlet signals of isolated aromatic protons H-4 and H-5 at δ 6.27 and 6.65, respectively. The assignments were supported by the cross peaks of H-4 to C-2, C-3, C-4a, C-9, C-9a and H-5 to C-6, C-7, C-8a, C-9, C-10a. The resonances of gemdimethyl protons 2"-Me $_2$ at δ 1.36 (6H, s), and two methylene protons H-3" and H-4" at δ 1.84 (2H, t, J = 6.6 Hz) and 3.55 (2H, t, J = 6.6 Hz) implied the presence of a 2,2dimethylchromane ring. This ring was placed to be at C-8 and C-7 by the correlation of H-4" to C-6, C-7, C-8 and C-8a in HMBC. The spectrum further showed the signals of gem-dimethyl protons at δ 1.45 (s, 6H, 2'-Me₂) and two methylene protons at δ 1.82 (t, 6.9 Hz, H-3') and 2.65 (t, 6.9 Hz, H-4') corresponding to the second dimethylchromane ring. According to the HMBC correlation of H-4' to C-1, C-2 and C-3, this ring was then assigned to be fused at C-1 and C-2. The evidences then suggested that RD3 was 3,6-dihydroxy-2',2'-dimethylchromano(5',6':2,1)-2",2"dimethylchromano(5",6":8,7)xanthone or known as isonormangostin. It was a new naturally occurring but synthetically known compound (Lee, 1981).

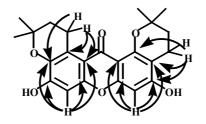


Table 54 The NMR spectral data of RD3

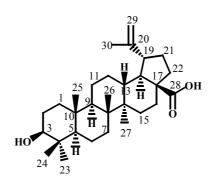
Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*	δ_{H} (multiplicity, J_{Hz})	НМВС
1	155.87 (C)	-	-
2	104.72 (C)	-	-
3	160.11 (C)	-	-
4	93.18 (CH)	6.27 (1H, s)	C-2, C-3, C-4a, C-9, C-9a
4a	156.80 (C)	-	-
5	99.80 (CH)	6.65 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	151.86 (C)	-	-
7	138.07 (C)	-	-
8	121.71 (C)	-	-
8a	113.36 (C)	-	-
9	177.89 (C=O)	-	-
9a	106.56 (C)	-	-
10a	150.73 (C)	-	-
2'	75.71 (C)	-	-
3'	31.46 (CH ₂)	1.82 (2H, t, 6.9)	C-2', C-4', 2'-(CH ₃) ₂
4'	16.93 (CH ₂)	2.65 (2H, t, 6.9)	C-1, C-2, C-3, C-2', C-3'
2"	74.76 (C)	-	-
3''	32.94 (CH ₂)	1.84 (2H, t, 6.6)	C-8, C-2", C-4", 2"-(CH ₃) ₂

Table 54 (Continued)

Position	$\delta_{_{\! \mathrm{C}}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
4''	22.51 (CH ₂)	3.55 (2H, t, 6.6)	C-6, C-7, C-8, C-8a, C-2", C-3"
2'-(CH ₃) ₂	26.26 (CH ₃)	1.45 (6H, s)	C-2', C-3', 2'-(CH ₃) ₂
2"-(CH ₃) ₂	26.34 (CH ₃)	1.36 (6H, s)	C-2", C-3", 2"-(CH ₃) ₂

^{*} Carbon type was deduced from DEPT experiments.

RD4: 3β -Hydroxylup-20(29)-en-oic acid (betulinic acid)



RD4 was isolated as a white solid, m.p. 282-285 °C. The ¹H NMR spectrum (**Table 55**) showed the resonances of an oxymethine proton (δ 3.18, dd, J = 11.0 and 5.0 Hz, H-3), five methyl groups (δ 0.97, H-27; 0.96, H-23; 0.94, H-26; 0.81, H-25 and 0.75, H-24) and isopropenyl side chain (δ 4.73, br d, J = 2.5 Hz, H_a-29; 4.77, br qd, J = 2.5 and 1.5 Hz, H_b-29; 1.69, s, H-30). ¹³C NMR spectral data (**Table 55**) showed the signal of a carbonyl carbon at δ 178.56. ¹H NMR and ¹³C NMR spectral data as well as melting point were in good agreement with those of 3 β -hydroxylup-20(29)-en-oic acid or betulinic acid (Macías, et al., 1994).

 $Table \ 55 \ \ The \ NMR \ spectral \ data \ of \ RD4$

n iii	RD4	betulinic acid	RD4	betulinic acid
Position	$\delta_{_{\mathrm{C}}}$ (C-Type)*	$\delta_{_{\! \mathrm{C}}}**$	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	$\delta_{_{ m H}}(J_{_{ m Hz}})**$
1	38.72 (CH ₂)	38.5 (t)	H _{ax} : 0.88 (1H, dd, 13.0, 4.5)	H _{ax} : 0.90
			H_{eq} : 1.65 (1H, m)	H _{eq} : 1.58
2	27.38 (CH ₂)	28.2 (t)	H _{ax} : 1.59 (1H, m)	H _{ax} : 1.82 (8.3)
			H_{eq} : 1.59 (1H, m)	H _{eq} : 1.82 (7.5)
3	78.76 (CH)	78.1 (<i>d</i>)	3.18 (1H, dd, 11.0, 5.0)	3.45 (8.3, 7.5)
4	38.86 (C)	39.4 (s)	-	-
5	55.34 (CH)	55.9 (<i>d</i>)	0.68 (1H, d, 10.0)	0.80
6	18.30 (CH ₂)	18.7 (t)	H _{ax} : 1.37 (1H, m)	H _{ax} : 1.40
			H_{eq} : 1.55 (1H, m)	H _{eq} : 1.40
7	34.34 (CH ₂)	34.7 (t)	1.37 (2H, m)	1.36
8	40.73 (C)	41.0 (s)	-	-
9	50.53 (CH)	50.9 (d)	1.27 (1H, m)	1.36
10	37.17 (C)	37.5 (s)	-	-
11	20.87 (CH ₂)	21.1 (t)	H_{ax} : 1.40 (1H, m)	H _{ax} : 1.36
			H_{eq} : 1.23 (1H, m)	H _{eq} : 1.15
12	25.54 (CH ₂)	26.0 (t)	H_{ax} : 1.15 (1H, m)	H _{ax} : 1.18
			H _{eq} : 1.71 (1H, m)	H _{eq} : 1.84
13	38.23 (CH)	39.2 (<i>d</i>)	2.28 (1H, m)	2.71
14	42.43 (C)	42.8 (s)	-	-
15	29.72 (CH ₂)	30.2 (t)	H_{ax} : 1.52 (1H, m)	H _{ax} : 1.73
			H_{eq} : 1.16 (1H, m)	H _{eq} : 1.19
16	32.33 (CH ₂)	32.8 (t)	H_{ax} : 1.50 (1H, m)	H _{ax} : 1.53
			H_{eq} : 2.25 (1H, m)	H _{eq} : 2.63
17	56.12 (C)	56.6 (s)	-	-
18	49.20 (CH)	49.7 (<i>d</i>)	1.57 (1H, m)	1.70 (8.5)
19	46.90 (CH)	47.7 (<i>d</i>)	3.04 (1H, dt, 11.0, 5.0)	3.51 (8.5, 4.1)

Table 55 (Continued)

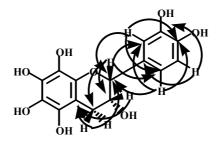
D = =:4: =	RD4	betulinic acid	RD4	betulinic acid
Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{_{\! \mathrm{C}}}$ **	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	$\delta_{_{ m H}}(J_{_{ m Hz}})**$
20	150.58 (C)	151.4 (s)	-	-
21	30.67 (CH ₂)	31.1 (t)	H _α : 1.94 (1H, m)	H _α : 2.20 (8.5)
			H_{β} : 1.41 (1H, m)	H _β : 1.52 (4.1)
22	37.09 (CH ₂)	37.4 (t)	H_{α} : 1.98 (1H, m)	H _α : 2.20
			H_{β} : 1.38 (1H, m)	Н _β : 1.54
23	28.01 (CH ₃)	38.5 (q)	0.96 (3H, s)	1.20
24	15.44 (CH ₃)	16.2 (q)	0.75 (3H, s)	0.98
25	16.13 (CH ₃)	16.3 (q)	0.81 (3H, s)	0.79
26	16.04 (CH ₃)	16.2 (q)	0.94 (3H, s)	1.03
27	14.67 (CH ₃)	14.8 (q)	0.97 (3H, s)	1.04
28	178.56 (C=O)	179.0 (s)	-	-
29	109.36 (CH ₂)	110.0 (t)	H _a : 4.73 (1H, <i>br d</i> , 2.5)	H _a : 4.92 (1.9)
			H _b : 4.77 (1H, br qd, 2.5, 1.5)	H _b : 4.74 (1.9)
30	19.38 (CH ₃)	19.4 (q)	1.69 (3H, s)	1.76

^{*} Carbon type was deduced from DEPT experiments.

^{**} Spectrum in pyridine- d_5 .

RD5: 3,5,6,7,8,3',4'-Heptahydroxyflavan

RD5 is a light brown crystals, m.p. 240-242 °C. The molecular weight of 290.0771 [M-O₂]⁺ corresponded to $C_{15}H_{14}O_8$. The optical rotation was $[\alpha]_D^{29}$: - 72.0 $^{\circ}$ (c = 0.012 in CH₃OH). The UV spectrum showed maximum absorptions at 280.6, 228.5 and 212.1 nm. The IR spectrum exhibited strong absorptions of O-H stretching at 3358 cm⁻¹ and C=C stretching at 1609 and 1517 cm⁻¹. The ¹H NMR spectrum of **RD5** (Table 56) exhibited the signals of aromatic protons H-5', H-6' at δ 6.82 (2H) and H-2' at δ 7.00. The resonances of methine protons H-2, H-3 were shown as singlets at δ 4.85 and 4.22 whereas the methylene protons H_a -4 and H_b -4 appeared as doublet of doublets at δ 2.87 (J =15.0 and 4.5 Hz) and 2.82 (J =15.0 and 3.0 Hz). The correlations of H-2 to H-3; H-3 to H-2, H-4 and H-4 to H-3 in the COSY experiment confirmed the assignment of protons. The enhancement of the resonances of H-3 and H_a-4 by irradiation of the H-2 signal in the NOE experiment indicated that these three protons were in the cis-configuration. The complete HMBC correlations confirmed the assigned structure. **RD5** was then assigned as 3,5,6,7,8,3',4'-heptahydroxyflavan. This compound is an C-3 epimer of elephantorrhizol (Moyo, et al., 1999). RD5 appears to be novel.



Major HMBC correlations of $RD5\,$

Table~56~~The NMR spectral data of RD5

Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	COSY	НМВС
2	78.47 (CH)	4.85 (1H, s)	H-3	C-3, C-4, C-1', C-2', C-6'
3	66.15 (CH)	4.22 (1H, <i>br s</i>)	H-2, H-4	C-4a, C-1'
4	27.89 (CH ₂)	H _a : 2.87 (1H, dd, 15.0, 4.5)	H-3	C-2, C-3, C-4a, C-5
		H _b : 2.82 (1H, dd, 15.0, 3.0)	H-3	C-2, C-3, C-4a, C-5
4a	98.76 (C)	-	-	-
5	155.90 (C)	-	-	-
6	130.66 (C)	-	-	-
7	155.78 (C)	-	-	-
8	156.43 (C)	-	-	-
8a	144.52 (C)	-	-	-
1'	130.66 (C)	-	-	-
2'	113.92 (CH)	7.00 (1H, s)	H-6'	C-2, C-1', C-3', C-4', C-6'
3'	144.39 (C)	-	-	-
4'	144.52 (C)	-	-	-
5'	114.84 (CH)	6.82 (1H, <i>br s</i>)	H-6′	C-1', C-3'
6'	118.20 (CH)	6.82 (1H, <i>br s</i>)	H-5'	C-2, C-1', C-2', C-4'

^{*} Carbon type was deduced from DEPT experiments.

RD6: Kaempferol 3-O- β -glucopyranosyl-7-O- α -rhamnopyranoside

The glycoside **RD6** was obtained as a yellow solid, m.p. 249-250 °C. The ¹H NMR spectrum (Table 57) exhibited a singlet signal of a chelated hydroxy proton 5-OH at δ 12.25 and the resonances of H-2', H-6' and H-3', H-5' as an AA'BB' pattern at δ 7.95 (J = 9.0 Hz, 2H) and 6.85 (J = 9.0 Hz, 2H), respectively. The resonances of meta protons H-6 and H-8 were observed at δ 6.37 (d, J = 2.1 Hz, 1H) and 6.67 (d, J = 2.1 Hz, 1H). It therefore indicated that RD6 was a derivative of kaemferol. The remaining resonances were the resonances of two sugar moieties which were assigned to be glucose and rhamnose. The resonance of the anomeric proton of glucose (H-1") was at δ 5.19 (d, J = 7.2 Hz) whereas the resonance of the anomeric proton of rhamnose (H-1''') was at δ 5.47 (br s). In 13 C NMR the signals of the anomeric carbons of glucose and rhamnose were at δ 102.54 and 99.22, respectively. HMBC correlations of H-1" to C-3 and H-1" to C-7 indicated the glucose moiety was linked to kaempferol at C-3 and the rhamnose unit was at C-7. The ¹³C NMR signals (**Table** 57) and DEPT spectra showed resonances of a carbonyl carbon, eight quaternary carbons, fourteen methine carbons, a methylene carbon and a methyl carbon. Thus glycoside **RD6** was kaempferol 3-O- β -glucopyranosyl-7-O- α -rhamnopyranoside (Mahabusarakam, 1992).

Major HMBC correlations of RD6

Table 57 The NMR spectral data of RD6

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
2	158.95 (C)	-	-
3	134.97 (C)	-	-
4	179.00 (C=O)	-	-
4a	107.12 (C)	-	-
5	161.58 (C)	-	-
6	101.04 (CH)	6.37 (1H, d, 2.1)	C-4a, C-5, C-8
7	162.83 (C)	-	-
8	96.00 (CH)	6.67 (1H, d, 2.1)	C-4a, C-7, C-8a
8a	157.37 (C)	-	-
1'	122.24 (C)	-	-
2', 6'	132.50 (CH)	7.95 (2H, d, 9.0)	C-2, C-2',6', C-4'
3', 5'	116.74 (CH)	6.85 (2H, d, 9.0)	C-1', C-3',5', C-4'
4'	160.95 (C)	-	-
Glc-1"	102.54 (CH)	5.19 (1H, d, 7.2)	C-3
2''	70.98 (CH)	3.13-3.12 (1H, m)	C-3", C-4"
3''	77.31 (CH)	3.25 (1H, t, 5.4)	C-1", C-4"
4''	78.04 (CH)	3.09 (1H, t, 5.4)	C-3"

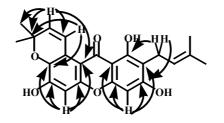
Table 57 (Continued)

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
5''	71.36 (CH)	3.70-3.69 (1H, m)	C-6"
6''	61.98 (CH ₂)	3.51 (2H, m)	C-4"
Rha-1'''	99.22 (CH)	5.47 (1H, br s)	C-7, C-5"
2'''	71.12 (CH)	3.68-3.67 (1H, m)	C-3", C-4"
3'''	75.28 (CH)	3.31 (1H, <i>m</i>)	C-4", C-5"
4'''	73.03 (CH)	3.36-3.35 (1H, m)	C-2", C-4", C-5", C-6"
5'''	70.84 (CH)	3.49 (1H, m)	C-3", C-4"
6'''	18.81 (CH ₃)	1.06 (3H, d, 9.0)	C-4", C-5"
5-OH	-	12.25 (1H, s)	C-4a, C-5, C-6

^{*} Carbon type was deduced from DEPT experiments.

RD7: 1,3,6-Trihydroxy-2-(3-methyl-2-butenyl)-2",2"-dimethylchromeno (5",6":8,7)xanthone (garcinone B)

RD7 is an orange solid, m.p. 186-187 °C. The ¹H NMR spectrum (**Table 58**) demonstrated a *singlet* resonance of a chelated hydroxy group 1-OH (δ 13.69) and two singlet signals of H-4 and H-5 (δ 6.31 and 6.81). Two singlet signals of two methyl groups 2"-Me₂ (δ 1.50 and 1.25) and two doublet signals of two geminal olefinic protons H-3" and H-4" (δ 5.82, J = 10.0 Hz and 8.02, J = 10.0 Hz) which were a part of a chromene ring were observed. The deshielded effect on the resonance of H-4 (δ 8.02) suggested that the chromene ring attached to the xanthone nucleus nearby the carbonyl group. The *singlet* of two methyl groups (δ 1.84, H-5' and 1.77, H-4'), a methylene protons (δ 3.45, H-1') and an olefinic methine proton (δ 5.29, H-2') were typical signals of a prenyl moiety. HMBC correlation of H-1' to C-1 and C-3 indicated that the prenyl side chain was at C-2. The assignment of H-4 was confirmed by the correlation of H-4 to C-2, C-3, C-4a and C-9a whereas the position of H-5 was proved by the correlation of H-5 to C-6, C-7, C-9 and C-10a. The ¹³C NMR spectral data (Table 58) indicated a carbonyl carbon, four methyl carbons, a methylene carbon, five methine carbons and twelve quaternary carbons. RD7 was then identified as 1,3,6trihydroxy-2-(3-methyl-2-butenyl)-2",2"-dimethylchromeno(5",6":8,7)xanthone. The proposed structure was identical to garcinone B (Sen, et al., 1982).



Major HMBC correlations of $\ensuremath{\mathbf{RD7}}$

Table 58 The NMR spectral data of RD7

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
1	160.41 (C)	-	-
2	102.29 (C)	-	-
3	161.80 (C)	-	-
4	93.36 (CH)	6.31 (1H, s)	C-2, C-3, C-4a, C-9a
4a	155.80 (C)	-	-
5	102.29 (CH)	6.81 (1H, s)	C-6, C-7, C-9, C-10a
6	151.00 (C)	-	-
7	136.50 (C)	-	-
8	125.50 (C)	-	-
8a	119.72 (C)	-	-
9	182.42 (C=O)	-	-
9a	108.50 (C)	-	-
10a	153.80 (C)	-	-
1'	21.41 (CH ₂)	3.45 (2H, d, 7.0)	C-1, C-3, C-2', C-3'
2'	121.54 (CH)	5.29 (1H, br t, 7.0)	C-1', C-4', C-5'
3'	134.00 (C)	-	-
4'	25.83 (CH ₃)	1.77 (3H, s)	C-2', C-5'
5'	17.91 (CH ₃)	1.84 (3H, s)	C-2', C-4'

Table 58 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
2''	77.20 (C)	-	-
3''	132.24 (CH)	5.82 (1H, d, 10.0)	C-8a, C-2", C-4", C-5"
4''	120.99 (CH)	8.02 (1H, d, 10.0)	C-7, C-2''
2"-(CH ₃) ₂	27.30 (CH ₃)	1.50 (3H, s)	C-2", C-3", 2"-(CH ₃) ₂
	31.92 (CH ₃)	1.25 (3H, s)	C-2", C-3", 2"-(CH ₃) ₂
1-OH	-	13.69 (1H, s)	-

^{*} Carbon type was deduced from DEPT experiments.

RD8: 1,3,6-Trihydroxy-7-methoxy-2,5-bis(3-methyl-2-butenyl)xanthone

RD8 is a bright yellow solid, m.p. 223-224 °C. The ¹H NMR spectrum (**Table 59**) exhibited the resonances of hydroxy protons 1-OH at δ 13.22 (s), 3-OH at δ 8.99 (s) and 6-OH at δ 7.11 (s), aromatic protons H-8 at δ 7.44 (s) and H-4 at δ 6.37 (s) and 7-OCH₃ at δ 3.96 (s). The NOE effect of H-8 (δ 7.44) to the methoxy resonance suggested the OCH₃ at C-7. The signals of two prenyl groups were displayed as followed: two *broad triplets* of olefinic methine protons H-2' (δ 5.26) and H-2'' (δ 5.25), two *doublets* of benzylic methylene protons H-1' (δ 3.38) and H-1'' (δ 3.56) and four *singlets* of methyl groups H-4' (δ 1.69), H-5' (δ 1.80), H-4'' (δ 1.65) and H-5'' (δ 1.85). The HMBC correlations of H-1' to C-1, C-2, C-3 and H-1'' to C-5, C-6, C-10a indicated that the prenyl side chains were at C-2 and C-5. The assignment of H-4 was confirmed by the correlation of H-4 to C-2, C-3, C-4a, C-9 and C-9a whereas the signal of H-8 was proved by the correlation of H-8 to C-6, C-7, C-8a, C-9 and C-10a. These assignment indicated that **RD8** was 1,3,6-trihydroxy-7-methoxy-2,5-bis(3-methyl-2-butenyl)xanthone. The structure of **RD8** and its melting point were identical to the previously isolated compound from *Garcinia cowa* (Na Pattalung, *et al.*, 1994).

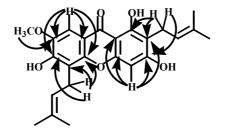


Table 59 The NMR spectral data of RD8

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	160.00 (C)	-	-
2	109.65 (C)	-	-
3	160.12 (C)	-	-
4	93.55 (CH)	6.37 (1H, s)	C-2, C-3, C-4a, C-9, C-9a
4a	155.86 (C)	-	-
5	115.30 (C)	-	-
6	150.05 (C)	-	-
7	144.05 (C)	-	-
8	101.99 (CH)	7.44 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
8a	112.94 (C)	-	-
9	180.21 (C=O)	-	-
9a	101.99 (C)	-	-
10a	149.50 (C)	-	-
1'	21.36 (CH ₂)	3.38 (2H, d, 6.9)	C-1, C-2, C-3, C-2', C-3'
2'	121.90 (CH)	5.26 (1H, <i>br t</i> , 7.2)	C-1', C-4', C-5'
3'	133.60 (C)	-	-
4'	25.70 (CH ₃)	1.69 (3H, s)	C-2', C-3', C-5'
5'	17.83 (CH ₃)	1.80 (3H, s)	C-2', C-3', C-4'

Table 59 (Continued)

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1''	22.34 (CH ₂)	3.56 (2H, d, 7.2)	C-5, C-6, C-10a, C-2", C-3"
2''	121.03 (CH)	5.25 (1H, <i>br t</i> , 7.2)	C-1", C-4", C-5"
3''	132.57 (C)	-	-
4''	25.75 (CH ₃)	1.65 (3H, s)	C-2", C-3", C-5"
5''	17.90 (CH ₃)	1.85 (3H, s)	C-2", C-3", C-4"
1-OH	-	13.22 (1H, s)	C-1, C-2, C-9a
3-ОН	-	$8.99 (1H, s)^{a}$	-
6-OH	-	$7.11 (1H, s)^{a}$	-
7-OCH ₃	56.33 (CH ₃)	3.96 (3H, s)	C-7

^{*} Carbon type was deduced from DEPT experiments.

^a Assignment maybe interchangeable.

RD9: 1,6-Dihydroxy-7-methoxy-8-(3-methyl-2-butenyl)-2',2'-dimethylchromeno (5',6':2,3)xanthone

RD9 was isolated as a yellow solid, m.p. 159-160 °C. The ¹H NMR spectrum (**Table 60**) showed four *singlet* signals of 1-OH (δ 13.60), H-5 (δ 6.75), H-4 (δ 6.15) and 7-OCH, (δ 3.71). The ¹H NMR spectrum showed characteristic signals of a dimethylchromene ring, of which the signal of gem-dimethyl protons resonated as a singlet at δ 1.37 and two doublet signals of two olefinic protons (H-4' and H-3') were at δ 6.64 and 5.48. The ³J correlations of H-4' to C-1 and H-3' to C-2 indicated that the dimethylchromene ring was fused to C-2 and C-3. In addition, the signals of two methyl groups at δ 1.60 (H-4") and 1.74 (H-5"), methylene protons at δ 3.99 (H-1") and an olefinic methine proton at δ 5.17 (H-2'') which were characteristic signals of a prenyl moiety were present in the spectrum. The chemical shift of the methylene protons H-1'' (δ 3.99) implied that H-1'' was deshielded by the carbonyl group, accordingly the prenyl side chain was proposed to be at C-8, a peri position to the carbonyl group. The HMBC correlation of H-1" to C-7, C-8 and C-8a confirmed the location of the prenyl unit at C-8. According to the assignment and HMBC correlations, 1,6-dihydroxy-7-methoxy-8-(3-methyl-2-butenyl)-2',2'dimethylchromeno(5',6':2,3)xanthone was the structure of **RD9** (Sen, et al., 1980).

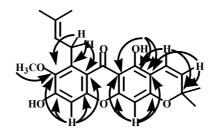


Table 60 The NMR spectral data of RD9

Position	$\delta_{_{\mathbb{C}}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
1	159.87 (C)	-	-
2	104.50 (C)	-	-
3	157.92 (C)	-	-
4	94.16 (CH)	6.15 (1H, s)	C-2, C-3, C-4a, C-9a
4a	156.27 (C)	-	-
5	101.69 (CH)	6.75 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	154.61 (C)	-	-
7	142.65 (C)	-	-
8	136.99 (C)	-	-
8a	112.17 (C)	-	-
9	182.00 (C=O)	-	-
9a	103.73 (C)	-	-
10a	155.74 (C)	-	-
2'	77.95 (C)	-	-
3'	127.16 (CH)	5.48 (1H, d, 10.2)	C-2, C-2'
4'	115.71 (CH)	6.64 (1H, d, 10.2)	C-1, C-2'
1''	26.55 (CH ₂)	3.99 (2H, d, 6.3)	C-7, C-8, C-8a, C-2", C-3"
2''	123.11 (CH)	5.17 (1H, <i>br t</i>)	C-1", C-4", C-5"

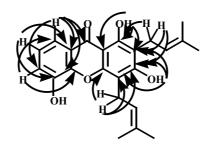
Table 60 (Continued)

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
3''	132.22 (C)	-	-
4''	25.84 (CH ₃)	1.60 (3H, s)	C-2", C-3", C-5"
5''	18.23 (CH ₃)	1.74 (3H, s)	C-2", C-3", C-4"
2'-(CH ₃) ₂	28.30 (CH ₃)	1.37 (6H, s)	C-2', C-3', C-4', 2'-(CH ₃) ₂
1-OH	-	13.60 (1H, s)	C-1, C-2, C-9a
7-OCH ₃	62.05 (CH ₃)	3.71 (3H, s)	C-7

^{*} Carbon type was deduced from DEPT experiments.

RD10: 1,3,5-Trihydroxy-2,4-bis(3-methyl-2-butenyl)xanthone (8-desoxygartanin)

RD10 is a yellow solid, m.p. 167-168 °C. The ¹H NMR spectral data (**Table 61**) demonstrated the resonances of a chelated hydroxy group 1-OH (δ 13.18, s), 3-OH (δ 6.53, s) and 5-OH (δ 5.74, br s). Two *doublet of doublet* signals at δ 7.76 (J = 7.8 and 1.8 Hz) and 7.27 (J = 7.8 and 1.8 Hz), and a *triplet* signal at δ 7.23 (J = 7.8 Hz) implied the presence of H-8, H-6 and H-7, respectively. The deshielded effect on resonance of H-8 (δ 7.76) suggested the location of H-8 nearby the carbonyl group. Two *broad triplet* signals at δ 5.29 (H-2') and 5.26 (H-2''), two *doublet* signals at δ 3.48 (H-1') and 3.54 (H-1'') and three *singlet* signals at δ 1.86 (H-5' and H-5''), 1.79 (H-4') and 1.75 (H-4'') were the resonances of two prenyl side chains. HMBC correlations of H-1' to C-1, C-2, C-3 and H-1'' to C-3, C-4, C-4a indicated that the prenyl side chains were at C-2 and C-4. The assignment of 3-OH was confirmed by the correlations corresponded to 1,3,5-trihydroxy-2,4-bis(3-methyl-2-butenyl)xanthone or 8-desoxygartanin (Govindachari, *et al.*, 1971).



 $Table\ 61\ \ The\ NMR\ spectral\ data\ of\ RD10$

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	158.68 (C)	-	-
2	109.13 (C)	-	-
3	160.91 (C)	-	-
4	105.44 (C)	-	-
4a	152.40 (C)	-	-
5	144.42 (C)	-	-
6	119.77 (CH)	7.27 (1H, dd, 7.8, 1.8)	C-8, C-10a
7	123.83 (CH)	7.23 (1H, <i>t</i> , 7.8)	C-5, C-8a
8	116.91 (CH)	7.76 (1H, dd, 7.8, 1.8)	C-6, C-8a, C-9, C-10a
8a	120.89 (C)	-	-
9	181.11 (C=O)	-	-
9a	103.32 (C)	-	-
10a	144.40 (C)	-	-
1'	22.09 (CH ₂)	3.48 (2H, d, 7.2)	C-1, C-2, C-3, C-2', C-3'
2'	121.19 (CH)	5.29 (1H, <i>br t</i> , 7.2)	C-1'
3'	136.18 (C)	-	-
4′	25.66 (CH ₃)	1.79 (3H, s)	C-2', C-3', C-5'

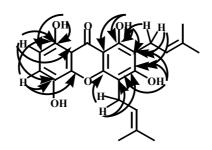
Table 61 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
5'	17.95 (CH ₃)	1.86 (3H, s)	C-2', C-3', C-4'
1''	21.65 (CH ₂)	3.54 (2H, d, 6.6)	C-3, C-4, C-4a, C-2", C-3"
2''	122.23 (CH)	5.26 (1H, <i>br t</i> , 7.2)	C-4, C-1"
3''	133.53 (C)	-	-
4''	25.87 (CH ₃)	1.75 (3H, s)	C-2", C-3", C-5"
5''	17.95 (CH ₃)	1.86 (3H, s)	C-2", C-3", C-4"
1-OH	-	13.18 (1H, s)	C-1, C-2, C-9a
3-ОН	-	6.53 (1H, s)	C-2, C-3, C-4
5-OH	-	5.74 (1H, <i>br s</i>)	-

^{*} Carbon type was deduced from DEPT experiments.

RD11: 1,3,5,8-Tetrahydroxy-2,4-bis(3-methyl-2-butenyl)xanthone (gartanin)

RD11 is a yellow solid, m.p. 165-167 °C. The ¹H NMR spectral data (**Table 62**) demonstrated the resonances of two chelated hydroxy groups 1-OH (δ 12.36, s), 8-OH (δ 11.28, s) and non chelated hydroxy group 3-OH (δ 6.62, s). Two *doublet* signals at δ 7.25 (J = 8.7 Hz) and 6.69 (J = 8.7 Hz) implied the presence of H-6 and H-7, respectively. Two *broad triplet* signals at δ 5.29 (H-2') and 5.26 (H-2''), two *doublet* signals at δ 3.49 (H-1') and 3.55 (H-1'') and three *singlet* signals at δ 1.88 (H-5' and H-5''), 1.81 (H-4') and 1.78 (H-4'') were the resonances of two prenyl side chains. HMBC correlations of H-1' to C-1, C-2, C-3 and H-1'' to C-3, C-4, C-4a indicated that the prenyl side chains were at C-2 and C-4. The assignment of 3-OH was confirmed by the correlation of 3-OH to C-2, C-3 and C-4. The ¹³C NMR spectral data and HMBC correlations corresponded to 1,3,5,8-tetrahydroxy-2,4-bis(3-methyl-2-butenyl)xanthone or gartanin (Govindachari, *et al.*, 1971).



Major HMBC correlations of RD11

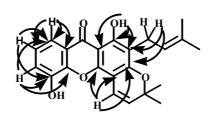
 $Table \ 62 \ \ The \ NMR \ spectral \ data \ of \ RD11$

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	158.16 (C)	-	-
2	109.51 (C)	-	-
3	161.65 (C)	-	-
4	105.81 (C)	-	-
4a	152.50 (C)	-	-
5	136.31 (C)	-	-
6	122.86 (CH)	7.25 (1H, d, 8.7)	C-5, C-8, C-10a
7	109.78 (CH)	6.69 (1H, d, 8.7)	C-5, C-8, C-8a
8	153.85 (C)	-	-
8a	107.17 (C)	-	-
9	184.73 (C=O)	-	-
9a	102.23 (C)	-	-
10a	142.84 (C)	-	-
1'	21.60 (CH ₂)	3.49 (2H, d, 6.9)	C-1, C-2, C-3, C-2', C-3'
2'	120.99 (CH)	5.29 (1H, <i>br t</i> , 6.9)	C-4', C-5'
3'	135.72 (C)	-	-
4'	25.89 (CH ₃)	1.81 (3H, s)	C-2', C-3', C-5'
5'	17.98 (CH ₃)	1.88 (3H, s)	C-2', C-3', C-4'
1''	22.01 (CH ₂)	3.55 (2H, d, 6.9)	C-3, C-4, C-4a, C-2", C-3"
2''	121.84 (CH)	5.26 (1H, br t, 6.9)	C-4", C-5"
3''	133.95 (C)	-	-
4''	25.70 (CH ₃)	1.78 (3H, s)	C-2", C-3", C-5"
5''	17.98 (CH ₃)	1.88 (3H, s)	C-2", C-3", C-4"
1-OH	-	12.36 (1H, s)	C-1, C-2, C-9a
3-ОН	-	6.62 (1H, s)	C-2, C-3, C-4
8-OH	-	11.28 (1H, s)	C-7, C-8

^{*} Carbon type was deduced from DEPT experiments.

RD12: 1,5,8-Trihydroxy-2-(3-methyl-2-butenyl)-2,2-dimethylchromeno (5",6":4,3)xanthone (morusignin J)

RD12 was isolated as a yellow solid with m.p. 183-185 °C. The ¹H NMR spectrum (**Table 63**) consisted of the signals of two chelated hydroxy protons 1-OH at δ 12.38, 8-OH at δ 11.35 and *ortho* protons H-6 and H-7 at δ 7.26 (d, J = 9.0 Hz) and 6.69 (d, J = 9.0 Hz). The resonances of prenylated unit were shown at δ 3.35 (2H, d, J = 7.5 Hz, H-1'), 5.24 (1H, br t, J = 8.0 Hz, H-2'), 1.70 (3H, s, H-4') and 1.82 (3H, s, H-5'). The resonances of methyl groups and methine proton associated to the characteristic signals of a chromene ring appeared at δ 1.50 (6H, s, 2"-Me₂), 5.66 (1H, d, J = 10.0 Hz, H-3") and 6.77 (1H, d, J = 10.0 Hz, H-4"). The ³J correlations of H-6 to C-5, C-8, C-10a and H-7 to C-5, C-8, C-8a as well as 8-OH to C-7, C-8, C-8a supported the position of *ortho* H-6 and H-7. The HMBC correlation of H-4" to C-3, C-4, C-4a confirmed the location of chromene ring. **RD12** is 1,5,8-trihydroxy-2-(3-methyl-2-butenyl)-2,2-dimethylchromeno(5",6":4,3)xanthone or known as morusignin J (Hano, *et al.*, 1993).



Major HMBC correlations of RD12

 $Table \ 63 \ \ The \ NMR \ spectral \ data \ of \ RD12$

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	159.29 (C)	-	-
2	112.55 (C)	-	-
3	159.97 (C)	-	-
4	102.06 (C)	-	-
4a	149.36 (C)	-	-
5	135.54 (C)	-	-
6	123.24 (CH)	7.26 (1H, d, 9.0)	C-5, C-8, C-10a
7	110.05 (CH)	6.69 (1H, d, 9.0)	C-5, C-8, C-8a
8	154.08 (C)	-	-
8a	107.34 (C)	-	-
9	184.43 (C=O)	-	-
9a	100.87 (C)	-	-
10a	142.70 (C)	-	-
1'	21.19 (CH ₂)	3.35 (2H, d, 7.5)	C-2, C-3, C-2', C-3'
2'	121.67 (CH)	5.24 (1H, br t, 8.0)	C-1', C-4', C-5'
3'	131.87 (C)	-	-
4'	25.82 (CH ₃)	1.70 (3H, s)	C-2', C-3', C-5'
5'	17.92 (CH ₃)	1.82 (3H, s)	C-2', C-3', C-4'
2''	78.39 (C)	-	-
3''	127.65 (CH)	5.66 (1H, d, 10.0)	C-4, C-2"
4''	114.78 (CH)	6.77 (1H, d, 10.0)	C-3, C-4, C-4a, C-2"
1-OH	-	12.38 (1H, s)	C-1, C-2, C-9a
8-OH	-	11.35 (1H, s)	C-7, C-8, C-8a
2"-(CH ₃) ₂	28.25 (CH ₃)	1.50 (6H, s)	C-2", C-3", 2"-(CH ₃) ₂

^{*} Carbon type was deduced from DEPT experiments.

RD13: 5,7,4'-Trihydroxyflavone (apigenin)

RD13 was a yellow solid, melting at 168-170 °C. The ¹H NMR spectrum (**Table 64**) indicated a flavone nucleus by the appearance of *singlet* signal of an α -methine proton (H-3) at δ 6.41. The signal for a chelated hydroxy group 5-OH (δ 12.75, s) and the signals of AA'BB' system of H-2', H-6' (δ 7.66, d, d = 8.5 Hz) and H-3', H-5' (δ 6.87, d, d = 8.5 Hz) were displayed in the spectrum. The resonances of *meta* coupling of H-6 and H-8 were detected at δ 6.17 (d, d = 2.1 Hz) and 6.35 (d, d = 2.1 Hz), respectively. The ¹³C NMR signals from DEPT spectra showed resonances of a carbonyl carbon, seven quaternary carbons and five methine carbons. The proof of vinylic proton H-3 was obtained from the results of ³d cross peaks of H-3 to C-4a and C-1' (δ 104.13 and 121.62) and ²d cross peaks of H-3 to C-2 and C-4 (δ 163.91 and 102.93). Therefore **RD13** was characterized as 5,7,4'-trihydroxyflavone which was known as apigenin (Berghöfer and Hölzl, 1987).

Major HMBC correlations of RD13

 $Table~64~\ \ The~NMR~spectral~data~of~RD13$

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
2	163.91 (C)	-	-
3	102.93 (CH)	6.41 (1H, s)	C-2, C-4, C-4a, C-1'
4	181.94 (C=O)	-	-
4a	104.13 (C)	-	-
5	161.64 (C)	-	-
6	99.05 (CH)	6.17 (1H, d, 2.1)	C-4a, C-5, C-8
7	157.45 (C)	-	-
8	93.87 (CH)	6.35 (1H, d, 2.1)	C-4, C-4a, C-6, C-7, C-8a
8a	163.93 (C)	-	-
1'	121.62 (C)	-	-
2', 6'	115.88 (CH)	7.66 (2H, d, 8.5)	C-2, C-1', C-3',5', C-4'
3', 5'	127.76 (CH)	6.87 (2H, d, 8.5)	C-2',6', C-4'
4'	160.86 (C)	-	-
5-OH	-	12.75 (1H, s)	-
7-OH	-	$10.14 (1H, s)^a$	-
4'-OH	-	$9.75 (1H, s)^{a}$	-

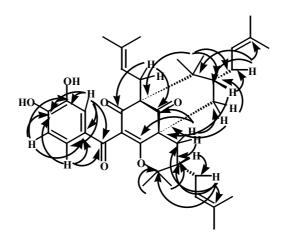
^{*} Carbon type was deduced from DEPT experiments.

^a Assignment maybe interchangeable.

RD14: (4R,6R,8S,30S)-2-(3,4-dihydroxybenzoyl)-5,5,31,31-tetramethyl-4,6,30-tris(3-methyl-2-butenyl)oxatricyclo[7.3.1.0]tridec-1-ene-3,9-dione (cambogin)

RD14 is a yellow solid, m.p. 238-240 °C. The optical rotation was $\left[\alpha\right]_{D}^{29}$ – 198.0 ° (c = 0.03 in CH₃OH). The ¹H NMR spectrum (**Table 65**) showed the resonances of aromatic protons H-15 (δ 6.72, d, J = 7.7 Hz), H-16 (δ 6.93, dd, J = 7.7, 2.2 Hz) and H-12 (δ 7.13, d, J = 2.2 Hz) which refered to the presence of 3,4-dihydroxybenzoyl group. The presence of three prenyl groups was observed. The signals of a prenyl group resonated at δ 4.77 (H-18), 2.49 and 2.31 (H-17), 1.51 (H-20), 1.49 (H-21) accordingly HMBC correlation of H-17 to C-3, C-9, was link to quaternary carbon C-4. The signals of two prenyl groups resonated at δ 5.16 (H-25), 2.56 and 1.98 (H-24), 1.72 (H-27), 1.57 (H-28) and δ 4.90 (H-35), 2.49 (H-34), 1.64 (H-37), 1.59 (H-38). These two groups were then connected to methine carbons C-6 and C-30, respectively, indicated by HMBC correlations of H-24 to C-5, C-7 and H-34 to C-29, C-30, C-31. The methine protons H-6 and H-30 resonated at δ 1.33 (m) and 1.43 (m). Two methyl groups on a saturated carbon C-5 were observed at δ 0.82 (s, H-22), 1.18 (s, H-23) and two methyl groups on a saturated carbon C-31 at δ 1.04 (s, H-32), 0.88 (s, H-33). Two *doublet of doublet* signals at δ 2.85 (1H, s) = 13.3 and 3.8 Hz)

and 0.99 (1H, J=13.3 and 1.5 Hz) were assigned for methylene protons H-7. The *multiplet* and *doublet* signals at δ 1.98 and 2.13 were suggested for resonated by methylene proton H-29. The fully HMBC data confirmed the assigned structure. **RD14** was then identified as $(4R,6R,8S,30S)-2-(3,4-\text{dihydroxybenzoyl})-5,5,31,31-\text{tetramethyl-4,6,30-tris(3-methyl-2-butenyl)oxatricyclo[7.3.1.0]tridec-1-ene-3,9-dione which agreed with the structure of cambogin (Rama Rao,$ *et al.*, 1980).



Major HMBC correlations of RD14

Table 65 The NMR spectral data of RD14

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
1	170.64 (C)	-	-
2	125.01 (C)	-	-
3	193.51 (C=O)	-	-
4	86.53 (C)	-	-
5	45.78 (C)	-	-
6	42.41 (CH)	1.33 (1H, <i>m</i>)	C-23
7	27.73 (CH ₂)	2.85 (1H, dd, 13.3, 3.8)	C-1, C-4, C-6, C-8, C-9, C-29
		0.99 (1H, dd, 13.3, 1.5)	C-4, C-6, C-8, C-9, C-29
8	50.83 (C)	-	-
9	206.68 (C=O)	-	-
10	191.61 (C=O)	-	-
11	129.04 (C)	-	-
12	115.40 (CH)	7.13 (1H, d, 2.2)	C-10, C-11, C-13, C-14, C-16
13	145.38 (C)	-	-
14	151.05 (C)	-	-
15	115.17 (CH)	6.72 (1H, d, 7.7)	C-11, C-13, C-14
16	122.12 (CH)	6.93 (1H, dd, 7.7, 2.2)	C-10, C-12, C-14
17	25.21 (CH ₂)	2.49 (1H, m)	C-9, C-18, C-19
		2.31 (1H, dd, 13.3, 3.8)	C-3, C-18, C-19
18	120.71 (CH)	4.77 (1H, br t, 6.6)	C-17, C-20, C-21
19	132.73 (C)	-	-
20	25.92 (CH ₃)	1.51 (3H, s)	C-18, C-19, C-21
21	18.12 (CH ₃)	1.49 (3H, s)	C-18, C-19, C-20
22	28.51 (CH ₃)	0.82 (3H, s)	C-4, C-6, C-23
23	21.17 (CH ₃)	1.18 (3H, s)	C-4, C-5, C-6, C-22, C-24
24	28.97 (CH ₂)	2.56 (1H, m)	C-26
		1.98 (1H, m)	C-5, C-7

Table 65 (Continued)

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
25	122.12 (CH)	5.16 (1H, br t, 6.6)	C-6, C-24, C-27, C-28
26	132.89 (C)	-	-
27	25.18 (CH ₃)	1.72 (3H, s)	C-25, C-26, C-28
28	17.98 (CH ₃)	1.57 (3H, s)	C-25, C-26, C-27
29	38.44 (CH ₂)	2.13 (1H, d, 13.3)	C-1, C-7, C-8, C-9, C-30
		1.98 (1H, m)	C-1, C-7, C-8, C-9, C-30
30	45.38 (CH)	1.43 (1H, m)	C-8, C-29, C-31, C-34
31	67.73 (C)	-	-
32	22.22 (CH ₃)	1.04 (3H, s)	C-30, C-31, C-33
33	26.35 (CH ₃)	0.88 (3H, s)	C-30, C-31, C-32
34	25.21 (CH ₂)	2.49 (2H, m)	C-29, C-30, C-31, C-36
35	125.34 (CH)	4.90 (1H, br t, 6.6)	C-34, C-37, C-38
36	132.17 (C)	-	-
37	25.89 (CH ₃)	1.64 (3H, s)	C-35, C-36, C-38
38	18.09 (CH ₃)	1.59 (3H, s)	C-35, C-36, C-37

^{*} Carbon type was deduced from DEPT experiments.

RD15: Kaempferol 3,7-di-*O*-α-rhamnopyranoside

RD15 was isolated as a yellow solid, m.p. 232-233 °C. The ¹H NMR spectrum (**Table 66**) indicated that **RD15** was a kaemferol glycoside. The aglycone moiety showed the pattern of *para* disubstitution of which H-2', H-6' and H-3', H-5' resonated as *doublets* at δ 7.77 and 6.95 (J = 8.2 Hz). The *meta* coupling of H-6 and H-8 were suggested from the two *broad singlets* resonances at δ 6.67 and 6.40. A *singlet* signal at δ 12.45 indicated that the hydroxy group (5-OH) was *peri* to a carbonyl group (C-4). Two *broad singlet* signals at δ 5.43 and 5.52 were the resonance of anomeric protons H-1" and H-1"', respectively. Two *doublets* with three protons each at δ 0.90 and 1.23 were the resonances of two methyl groups H-6" and H-6"', respectively. The resonances of sugar moieties corresponded to two molecules of rhamnose. The HMBC experiments revealed the correlations between the H-1" (δ 5.43) to C-3 (δ 134.93) and H-1"' (δ 5.52) to C-7 (δ 162.11), suggested that the C-1" and C-1"' of rhamnosyl units were linked to the 3-hydroxy and 7-hydroxy of the aglycone. Then it was assigned to be kaempferol 3,7-di-O- α -rhamnopyranoside. This compound has been isolated from *Measa ramentacea* (Mahabusarakam, 1992).

Major HMBC correlations of $RD15\,$

Table~66~~The NMR spectral data of RD15

Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	COSY	НМВС
2	158.12 (C)	-	-	-
3	134.93 (C)	-	-	-
4	178.36 (C=O)	-	-	-
4a	106.37 (C)	-	-	-
5	156.57 (C)	-	-	-
6	94.52 (CH)	6.67 (1H, br s)	H-8	C-4a, C-5, C-7, C-8
7	162.11 (C)	-	-	-
8	99.63 (CH)	6.40 (1H, br s)	Н-6	C-4a, C-6, C-7, C-8a
8a	161.60 (C)	-	-	-
1 '	120.83 (C)	-	-	-
2', 6'	130.77 (CH)	7.77 (2H, d, 8.2)	H-3',5'	C-2, C-2',6', C-3',5', C-4'
3', 5'	115.76 (CH)	6.95 (2H, d, 8.2)	H-2',6'	C-1', C-3',5', C-4'
4'	160.43 (C)	-	-	-
Rha-1''	101.93 (CH)	5.43 (1H, <i>br s</i>)	H-2"	C-3
2''	70.20 (CH)	4.14 (1H, <i>br s</i>)	H-1"	C-1", C-3"
3''	70.12 (CH)	3.90-3.20 (1H, m)	-	C-1", C-4", C-5"

Table 66 (Continued)

Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	COSY	НМВС
4''	71.84 (CH)	3.90-3.20 (1H, m)	-	C-2", C-5"
5''	70.77 (CH)	3.90-3.20 (1H, <i>m</i>)	-	C-4", C-6"
6''	17.58 (CH ₃)	0.90 (3H, d, 6.0)	-	C-5''
Rha-1'''	98.70 (CH)	5.52 (1H, br s)	Н-2′′′	C-7, C-3'''
2'''	70.10 (CH)	3.99 (1H, <i>br s</i>)	H-1'''	C-1''', C-4'''
3'''	70.99 (CH)	3.90-3.20 (1H, m)	-	C-2''', C-4'''
4'''	72.24 (CH)	3.90-3.20 (1H, <i>m</i>)	-	C-2''', C-6'''
5'''	70.39 (CH)	3.90-3.20 (1H, m)	-	C-4''', C-6'''
6'''	18.04 (CH ₃)	1.23 (3H, d, 5.4)	-	C-5'''
5-OH	-	12.45 (1H, s)	-	-

^{*} Carbon type was deduced from DEPT experiments.

RD16: 3,5,7,3',4'-Pentahydroxyflavan ((-) epicatechin)

RD16 was isolated as light brown crystals, m.p. 235-236 °C. The optical rotation was $\left[\alpha\right]_{D}^{29}$ – 72.0 ° (c = 0.038 in CH₃OH). The ¹H NMR spectrum (**Table 67**) suggested the presence of five aromatic protons, two methine protons, one methylene proton and five hydroxy groups. Two doublet resonances at δ 5.99 and 5.87 with the coupling constant of 1.8 Hz corresponded to the resonances of meta protons H-6 and H-8, respectively. A doublet at δ 6.99 (J = 1.8 Hz), a doublet of doublet at δ 6.74 (J = 7.2 and 1.8 Hz) and a *doublet* at δ 6.77 (J = 7.2 Hz) were assigned for the resonances of H-2', H-6' and H-5', respectively. The spectra further showed the resonances of H-2, H-3, H_a-4 and H_b-4 at δ 4.78 (s), 4.13 (s), 2.66 (d, J = 16.8 and 4.5 Hz) and 2.79 (d, J = 16.8 and 2.5 Hz). The chemical shift of H-2 and H-3 indicated that these two protons were next to oxygen-bearing carbons. In addition, the *broad singlet* resonances of H-2 and H-3 suggested that the configuration of H-2 and H-3 are cis (Sethi, et al., 1984). The evidence from the NOE experiments: irradiation at the resonances of H-2 and H-3 resulted in the enhancement of the resonances of H-3, H_a -4 and H-2, H_a -4, respectively, indicated that H-2, H-3 and H_a-4 are cis. The remaining resonances are the resonances of five hydroxy groups (δ 8.99, 8.84, 8.64, 8.50 and 4.26). The 13 C NMR spectrum (Table 67) showed the signals of fifteen carbons; five aromatic methine carbons, two oxymethine carbons, a methylene carbon and seven quaternary aromatic carbons. In HMBC (Table 67) ³J correlations between H-2 to C-4, C-2', C-6', C-9; H-2' to C-2, C-4', C-6' and H-5' to C-1', C-3' suggested that the aromatic ring

was linked to C-2. Consequently, **RD16** was proposed to be (-) epicatechin of which the (-) isomer was indicated from optical rotation ($[\alpha]_D^{29}$ –72.0 °). The assignment was in agreement with the previous data of (-) epicatechin (Sheehan, *et al.*, 1983).

Major HMBC correlations of **RD16** NOE difference experiments of **RD16**

Table 67 The ¹H NMR spectral data of **RD16**

D = = '4' = =	RD16			((-) epicatechin
Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}\left(mult.,J_{\scriptscriptstyle m Hz} ight)^{\!\scriptscriptstyle m a}$	HMBC	$\delta_{\!\scriptscriptstyle m C}$	$\delta_{_{ m H}}\left(mult.,J_{_{ m Hz}} ight)^{^{ m b}}$
2	78.33 (CH)	4.78 (1H, s)	C-3, C-4, C-9, C-1',	78.1	4.84 (1H, s)
			C-2', C-6'		
3	65.63 (CH)	4.13 (1H, <i>br s</i>)	C-10, C-1'	65.6	4.19 (1H, <i>br s</i>)
4	28.13 (CH ₂)	H _a : 2.66	C-2, C-3, C-5, C-10	27.9	2.84
		(1H, dd, 16.8, 4.5)			(1H, dd, 16.5, 4.5)
		H _b : 2.79	C-3, C-5, C-10		2.76
		(1H, dd, 16.8, 2.5)			(1H, dd, 16.5, 2.5)
5	156.58 (C)	-	-	156.4	-
6	95.71 (CH)	5.99 (1H, d, 1.8)	C-5, C-7, C-8, C-10	95.6	6.05 (1H, d, 2.0)
7	156.34 (C)	-	-	156.2	-
8	94.69 (CH)	5.87 (1H, d, 1.8)	C-6, C-7, C-9, C-10	94.5	5.96 (1H, d, 2.0)

Table 67 (Continued)

Dogition		RD16			(-) epicatechin
Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{_{ m H}}$ (mult., $J_{_{ m Hz}}$) $^{^{ m a}}$	HMBC	$\delta_{\!\scriptscriptstyle m C}$	$\delta_{_{ m H}}\left(mult.,J_{_{ m Hz}} ight)^{^{ m b}}$
9	155.73 (C)	-	-	155.5	-
10	98.69 (C)	-	-	98.3	-
1′	130.61 (C)	-	-	130.4	-
2'	114.50(CH)	6.99 (1H, d, 1.8)	C-2, C-1', C-3',	114.8	7.25 (1H, d, 2.0)
			C-4', C-6'		
3'	144.45 (C)	-	-	144.2	-
4′	144.56 (C)	-	-	144.4	-
5 ′	115.05(CH)	6.77 (1H, d, 7.2)	C-1', C-3', C-4'	117.8	6.83 (1H, d, 8.0)
6'	118.08(CH)	6.74	C-2, C-2', C-4',	114.1	6.80
		(1H, dd, 7.2, 1.8)	C-5'		(1H, d, 8.0, 2.0)
3-ОН	-	4.26 (1H, <i>br s</i>)	-	-	4.66 (1H, <i>br s</i>)
5-OH	-	8.99 (1H, <i>br s</i>)	C-5, C-6, C-10	-	9.14 (1H, <i>br s</i>)
7-OH	-	8.84 (1H, <i>br s</i>)	C-7, C-8	_	8.94 (1H, <i>br s</i>)
3′-OH	-	8.64 (1H, <i>br s</i>)	C-2', C-4'	_	8.82 (1H, br s)
4'-OH	-	8.50 (1H, <i>br s</i>)	C-3', C-5'	-	8.74 (1H, <i>br s</i>)

^{*} Carbon type was deduced from DEPT experiments.

^a Spectrum in $CDCl_3 + DMSO-d_6$

^b Spectrum in $CD_3COCD_3 + DMSO-d_6$

RD17: 1,6,7-Trihydroxy-3-methoxy-2,8-bis(3-methyl-2-butenyl)xanthone

RD17 is a yellow solid, m.p. 170-172 °C, HREIMS m/z 410.1731 [M]⁺ (calcd for $C_{24}H_{26}O_6$, 410.1729). The UV spectrum showed maxima absorptions at 367.5, 316.6, 260.5, 243.9 and 208.9 nm. The IR spectrum exhibited the absorption band of O-H stretching (3406 cm⁻¹) and C=O stretching (1642 cm⁻¹). The ¹H NMR spectrum (Table 68) exhibited the singlet resonances of a hydrogen bonded hydroxy proton 1-OH at δ 13.35, H-4 at δ 6.26, H-5 at δ 6.75 and 3-OCH, at δ 3.83. The methoxy group was thought to be at C-3 by the NOE effect of the methoxy resonance to H-4. The presence of two prenyl groups were characterized from two broad triplets at δ 5.16 (H-2') and 5.24 (H-2''), two doublets (J = 6.0 Hz) at δ 3.28 (H-1') and 4.27 (H-1'') and four singlets at δ 1.72 (H-5', H-4''), 1.60 (H-4') and 1.82 (H-5''). These groups were placed at C-2 and C-8 from the HMBC correlations of H-1' to C-1, C-3 and H-1" to C-7, C-8. The assignment of H-4 was confirmed by the correlation of H-4 to C-2, C-3, C-4a and C-9a whereas the position of H-5 was proved by the correlation of H-5 to C-6, C-7, C-8a and C-10a. The complete HMBC confirmed the structure of RD17 as 1,6,7-trihydroxy-3-methoxy-2,8-bis(3-methyl-2-butenyl)xanthone. It was a new xanthone derivative.

Major HMBC correlations of RD17

Table 68 The NMR spectral data of RD17

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	159.69 (C)	-	-
2	111.37 (C)	-	-
3	163.50 (C)	-	-
4	88.75 (CH)	6.26 (1H, s)	C-2, C-3, C-4a, C-9a
4a	155.29 (C)	-	-
5	101.13 (CH)	6.75 (1H, s)	C-6, C-7, C-8a, C-10a
6	150.67 (C)	-	-
7	139.60 (C)	-	-
8	127.41 (C)	-	-
8a	111.67 (C)	-	-
9	182.57 (C=O)	-	-
9a	101.13 (C)	-	-
10a	153.50 (C)	-	-
1'	21.35 (CH ₂)	3.28 (2H, d, 6.3)	C-1, C-3, C-2', C-3'
2'	122.35 (CH)	5.16 (1H, br t, 6.0)	C-2, C-1', C-4', C-5'
3'	132.00 (C)	-	-
4'	25.81 (CH ₃)	1.60 (3H, s)	C-2', C-3', C-5'
5'	17.78 (CH ₃)	1.72 (3H, s)	C-2', C-3', C-4'

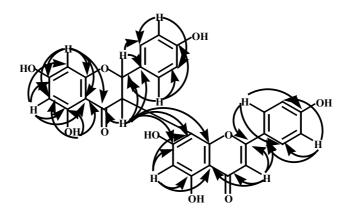
Table 68 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
1''	26.00 (CH ₂)	4.27 (2H, d, 6.3)	C-7, C-8, C-8a, C-2", C-3"
2''	121.52 (CH)	5.24 (1H, br t, 6.0)	C-8, C-1", C-4", C-5"
3''	136.00 (C)	-	-
4''	25.85 (CH ₃)	1.72 (3H, s)	C-2", C-3", C-5"
5''	18.08 (CH ₃)	1.82 (3H, s)	C-2", C-3", C-4"
1-OH	-	13.35 (1H, s)	C-1, C-2, C-9a
3-OCH ₃	55.82 (CH ₃)	3.83 (3H, s)	C-3

^{*} Carbon type was deduced from DEPT experiments.

FD1: (2S,3R)-2,3-dihydro-5,5",7,7"-tetrahydroxy-2,2"-bis(4-hydroxyphenyl)[3,8"-Bi-4H-1-benzopyran]-4,4"-dione (volkensiflavone)

FD1 was isolated as yellow needles, m.p. 247-250 °C. Duplication of signals (in a ratio of 1: 0.67) in the 1 H NMR spectrum of the optically active **FD1** in DMSO- d_{6} suggested the existence of two conformers at room temperature. The major reasons leading to the existence of such conformers is the rotational restriction about the interflavanyl C-3, C-8" bond. The sharp paired signals given at 295 K became the sharp signals at 390 K. The signals of major conformer (**Table 69**) showed the similar pattern to those of **GD20**. Except the resonances of aromatic protons H-2", H-6" and H-3", H-5" were shown as AA'BB' pattern at δ 7.70 (2H, d, J = 8.4 Hz) and 6.84 (2H, d, J = 8.4 Hz), respectively. The 13 C NMR data and HMBC correlations corresponded to the assigned structure. Thus **FD1** was assigned to be (2*S*,3*R*)-2,3-dihydro-5,5",7,7"-tetrahydroxy-2,2"-bis(4-hydroxyphenyl)-[3,8"-Bi-4H-1-benzopyran]-4,4"-dione which was known as volkensiflavone (Herbin, *et al.*, 1970).



Major HMBC correlations of ${\bf FD1}$

Table 69 The NMR spectral data of **FD1** at 390 K

Position	$\delta_{_{\! \mathrm{C}}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2	82.00 (CH)	5.65 (1H, d, 12.0)	C-1', C-2',6'
3	48.72 (CH)	4.90 (1H, d, 12.0)	C-2, C-4, C-2',6', C-7'', C-8'', C-8a''
4	196.64 (C=O)	-	-
4a	102.33 (C)	-	-
5	164.39 (C)	-	-
6	95.84 (CH)	5.90 (1H, <i>br s</i>)	C-4, C-4a, C-5, C-7, C-8
7	167.20 (C)	-	-
8	96.87 (CH)	6.00 (1H, br s)	C-4, C-4a, C-6, C-7, C-8a
8a	163.45 (C)	-	-
1'	129.81 (C)	-	-
2', 6'	128.79 (CH)	7.05 (2H, d, 8.4)	C-2, C-2',6', C-3',5'
3', 5'	115.18 (CH)	6.49 (2H, d, 8.4)	C-2',6', C-3',5'
4'	162.60 (C)	-	-
2''	164.27 (C)	-	-
3''	103.38 (CH)	6.51 (1H, s)	C-2", C-4", C-4a", C-1""

Table 69 (Continued)

Position	$\delta_{_{\mathbb{C}}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
4''	182.21 (C=O)	-	-
4a''	104.18 (C)	-	-
5''	161.04 (C)	-	-
6''	99.09 (CH)	6.15 (1H, s)	C-4a", C-5", C-7", C-8"
7''	161.60 (C)	-	-
8''	101.16 (C)	-	-
8a''	157.99 (C)	-	-
1'''	121.79 (C)	-	-
2''', 6'''	128.77 (CH)	7.70 (2H, d, 8.4)	C-2", C-2"',6"", C-4""
3''', 5'''	116.47 (CH)	6.84 (2H, d, 8.4)	C-1''', C-2''',6''', C-3''', 5'''
4'''	161.03 (C)	-	-
5-OH	-	12.10 (1H, s)	C-4a, C-5, C-6
5′′-ОН	-	12.85 (1H, s)	-

^{*} Carbon type was deduced from DEPT experiments.

FD2: 1-Hydroxy-2,3,4,6-tetramethoxyxanthone

FD2 is a yellow solid, m.p. 125-128 $^{\circ}$ C, HREIMS m/z 332.0892 [M] $^{+}$ (calcd for C₁₇H₁₆O₇, 332.0896). The UV spectrum showed maxima absorptions of xanthone type at 376.9, 309.3, 273.8 and 241.4 nm. The IR spectrum showed the stretchings of hydroxyl group at 3400 cm⁻¹ and carbonyl group at 1646 cm⁻¹. The ¹H NMR spectrum (**Table 70**) showed a *singlet* signal of a chelated hydroxy proton (1-OH) at δ 12.60. The signals in aromatic region, δ 7.84 (dd), 7.32 (d) and 7.26 (d) appearing as an ABX type were proposed for the signals of H-7, H-8 and H-5, respectively. These assignment were supported by the correlations of H-7 to C-5, C-6, C-9; H-8 to C-6, C-8a, C-10a and H-5 to C-6, C-7, C-10a on the HMBC experiment. Four singlet signals at δ 4.04, 4.16, 3.97 and 4.05 were assigned for four methoxy groups of 2-OCH₂, 3- OCH_3 , 4-OCH₃ and 6-OCH₃, respectively and were confirmed by the 3J coupling of methoxy protons to C-2, C-3, C-4 and C-6, respectively in HMBC. The ¹³C NMR and DEPT experiments indicated the presence of a carbonyl carbon, nine quaternary carbons, three methine carbons and four methyl carbons. The assignment suggested that **FD2** was 1-hydroxy-2,3,4,6-tetramethoxyxanthone. This compound appears to be novel.

Major HMBC correlations of FD2

 $Table \ 70 \ \ The \ NMR \ spectral \ data \ of \ FD2$

Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	150.60 (C)	-	-
2	132.75 (C)	-	-
3	154.19 (C)	-	-
4	135.48 (C)	-	-
4a	145.75 (C)	-	-
5	115.92 (CH)	7.26 (1H, d, 1.5)	C-6, C-7, C-10a
6	148.73 (C)	-	-
7	116.56 (CH)	7.84 (1H, dd, 7.8, 1.5)	C-5, C-6, C-9
8	123.71 (CH)	7.32 (1H, d, 7.8)	C-6, C-8a, C-10a
8a	120.91 (C)	-	-
9	181.71 (C=O)	-	-
9a	104.50 (C)	-	-
10a	146.40 (C)	-	-
1-OH	-	12.60 (1H, s)	-
2-OCH ₃	61.95 (CH ₃)	4.04 (3H, s)	C-2
3-OCH ₃	61.74 (CH ₃)	4.16 (3H, s)	C-3
4-OCH ₃	61.23 (CH ₃)	3.97 (3H, s)	C-4
6-OCH ₃	56.50 (CH ₃)	4.05 (3H, s)	C-6

^{*} Carbon type was deduced from DEPT experiments.

FD3: 1-Hydroxy-3,4,5-trimethoxyxanthone

FD3 is a yellow solid, m.p. 203-205 °C. EIMS showed the molecular ion of 302 which corresponded to $C_{16}H_{14}O_6$. The ¹H NMR spectral data (**Table 71**) exhibited a *singlet* signal of a chelated hydroxy group 1-OH at δ 12.76. A *singlet* signal of an isolated aromatic proton, δ 6.62, was observed and was assigned to be of H-2 according to the correlation to C-1, C-3, C-4 and C-9a on the HMBC experiment. Two *doublet of doublet* signals at δ 7.82 (J = 7.2 and 1.8 Hz) and 7.24 (J = 7.2 and 1.8 Hz), and a *triplet* signal at δ 7.30 (J = 7.2 Hz) implied the presence of H-8, H-6 and H-7, respectively. The deshielded effect on the resonance of H-8 (δ 7.82) suggested the location of H-8 nearby the carbonyl group. Three *singlet* signals of three methoxy groups appeared at δ 4.03 (5-OCH₃), 3.97 (3-OCH₃) and 3.93 (4-OCH₃). The assignment for three methoxy groups were supported by 3J coupling of 5-OCH₃ to C-5; 3-OCH₃ to C-3 and 4-OCH₃ to C-4 on the HMBC experiment. The 13 C NMR spectral data and HMBC correlations corresponded to 1-hydroxy-3,4,5-trimethoxyxanthone (Stout, *et al.*, 1973).

NOE difference experiments of FD3

Major HMBC correlations of FD3

 $Table \ 71 \ \ The \ NMR \ spectral \ data \ of \ FD3$

Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
1	154.09 (C)	-	-
2	90.92 (CH)	6.62 (1H, s)	C-1, C-3, C-4, C-9a
3	160.01 (C)	-	-
4	131.89 (C)	-	-
4a	153.17 (C)	-	-
5	148.23 (C)	-	-
6	115.47 (CH)	7.24 (1H, dd, 7.2, 1.8)	C-8, C-10a
7	123.62 (CH)	7.30 (1H, <i>t</i> , 7.2)	C-5, C-8a
8	116.66 (CH)	7.82 (1H, dd, 7.2, 1.8)	C-6, C-9, C-10a
8a	121.15 (C)	-	-
9	181.12 (C=O)	-	-
9a	104.28 (C)	-	-
10a	146.24 (C)	-	-
1-OH	-	12.76 (1H, s)	-
3-OCH ₃	56.41 (CH ₃)	3.97 (3H, s)	C-3
4-OCH ₃	60.90 (CH ₃)	3.93 (3H, s)	C-4
5-OCH ₃	56.43 (CH ₃)	4.03 (3H, s)	C-5

^{*} Carbon type was deduced from DEPT experiments.

FD4: 5,4'-Dihydroxy-7,3'-dimethoxyflavonol (rhamnazin)

FD4 was obtained as a yellow solid, m.p. 215-216 °C. EIMS showed the molecular ion of 330 which corresponded to $C_{17}H_{14}O_7$. ¹H NMR spectrum (**Table 72**) showed the resonances of a hydrogen-bonded hydroxy proton at δ 12.45 (s, 5-OH), two free hydroxy groups at δ 9.50 (s, 3-OH) and 9.76 (s, 4'-OH). Two *doublet* resonances (J = 2.3 Hz) at δ 6.34 (1H) and 6.76 (1H) were in agreement with the *meta*-coupling of aromatic protons H-6 and H-8. Two *doublet* signals at δ 7.78 (J = 2.3 Hz) and 6.94 (J = 8.7 Hz) and a *doublet of doublet* signal at δ 7.73 (J = 8.7 and 2.3 Hz) implied the presence of H-2, H-5 and H-6, respectively. Two methoxy signals at δ 3.86 and 3.85 were assigned for 7-OCH₃ and 3'-OCH₃. These assignment supported by the differential NOE technique: irradiation of 3'-OCH₃ effected the signal of H-2' and irradiation of 7-OCH₃ effected the signals of H-6 and H-8. The complete HMBC data confirmed the structure. 5,4'-Dihydroxy-7,3'-dimethoxyflavonol or known as rhamnazin was then assigned for **FD4** (Subhadhirasakul, *et al.*, 2003).

NOE difference experiments of **FD4** Ma

Major HMBC correlations of FD4

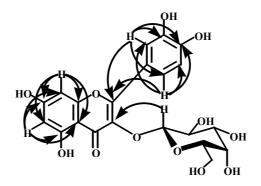
 $Table \ 72 \ \ The \ NMR \ spectral \ data \ of \ FD4$

Position	$\delta_{_{\! m C}}$ (C-Type)	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2	147.28 (C)	-	-
3	136.32 (C)	-	-
4	176.20 (C=O)	-	-
4a	104.23 (C)	-	-
5	160.54 (C)	-	-
6	97.69 (CH)	6.34 (1H, d, 2.3)	C-4a, C-5, C-7, C-8
7	165.15 (C)	-	-
8	92.30 (CH)	6.76 (1H, d, 2.3)	C-4, C-4a, C-6, C-7, C-8a
8a	156.30 (C)	-	-
1'	122.00 (C)	-	-
2'	112.02 (CH)	7.78 (1H, d, 2.3)	C-2, C-1', C-3', C-4', C-6'
3'	147.63 (C)	-	-
4'	149.18 (C)	-	-
5′	115.76 (CH)	6.94 (1H, d, 8.7)	C-3', C-4', C-6'
6'	122.10 (CH)	7.73 (1H, dd, 8.7, 2.3)	C-2, C-2', C-4'
3-ОН	-	9.50 (1H, s)	C-2, C-3, C-4
5-OH	-	12.45 (1H, s)	C-4a, C-5, C-6
7-OCH ₃	56.25 (CH ₃)	3.86 (3H, s)	C-7
3′-OCH ₃	56.08 (CH ₃)	3.85 (3H, s)	C-3'
4'-OH	-	9.76 (1H, s)	C-3', C-4', C-5'

^{*} Carbon type was deduced from DEPT experiments.

FD5: 2-(3,4-Dihydroxyphenyl)-3-(β-D-galactopyranosyloxy)-5,7-dihydroxy-4H-1benzopyran-4-one (hyperin or quercetin 3-*O*-β-galactopyranoside)

The glycoside **FD5** was isolated as a yellow solid, m.p. 235-236 °C. The optical rotation was $[\alpha]_D^{29}$: - 26.0 ° (c = 0.03 in CH₃OH). The ¹H NMR spectrum (**Table 73**) exhibited a *singlet* signal of a chelated hydroxy proton 5-OH at δ 12.60 and the resonances of H-2', H-5' and H-6' as an ABX pattern at δ 7.54 (d, J = 2.0 Hz), 6.81 (dd, J = 7.2 Hz) and 7.65 (d, J = 7.2, 2.0 Hz), respectively. The resonances of *meta* protons H-6 and H-8 were observed at δ 6.19 (d, J = 1.8 Hz) and 6.39 (d, J = 1.8 Hz). It therefore indicated that **FD5** was a derivative of quercetin. The remaining resonances were the resonances of a sugar moiety which was assigned to be galactose. The resonances of anomeric proton (H-1") was at δ 5.35 (d, J = 7.2 Hz). In the ¹³C NMR spectrum the anomeric carbon was at δ 102.12. HMBC correlation of H-1" to C-3 indicated that galactose was linked to quercetin at C-3. Thus glycoside **FD5** was 2-(3,4-dihydroxyphenyl)-3-(β -D-galactopyranosyloxy)-5,7-dihydroxy-4H-1-benzopyran-4-one or known as quercetin 3-O- β -galactopyranoside (Kartnig, *et al.*, 1985; Yasukawa and Takido, 1987).



Major HMBC correlations of FD5

 $Table \ 73 \ \ The \ NMR \ spectral \ data \ of \ FD5$

Position	$\delta_{_{\mathbb{C}}}$ (C-Type)	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
2	156.49 (C)	-	-
3	133.74 (C)	-	-
4	177.72 (C=O)	-	-
4a	104.11 (C)	-	-
5	161.46 (C)	-	-
6	98.98 (CH)	6.19 (1H, d, 1.8)	C-4a, C-5, C-7, C-8
7	164.57 (C)	-	-
8	93.81 (CH)	6.39 (1H, d, 1.8)	C-4, C-4a, C-6, C-7, C-8a
8a	156.58 (C)	-	-
1'	121.34 (C)	-	-
2'	116.23 (CH)	7.54 (1H, d, 2.0)	C-2, C-3', C-4', C-6'
3'	145.08 (C)	-	-
4′	148.74 (C)	-	-
5′	115.45 (CH)	6.81 (1H, d, 7.2)	C-1', C-3', C-4'
6'	122.19 (CH)	7.65 (1H, dd, 7.2, 2.0)	C-2, C-2', C-4', C-5'

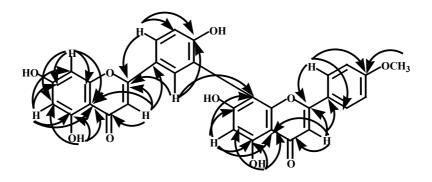
Table 73 (Continued)

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
Gal-1"	102.12 (CH)	5.35 (1H, d, 7.2)	C-3
2''	71.47 (CH)	3.57 (1H, t, 7.2)	C-1", C-5"
3''	76.07 (CH)	3.30 (1H, <i>m</i>)	C-4"
4''	68.18 (CH)	3.66 (1H, d, 2.7)	C-2", C-5"
5''	73.45 (CH)	3.37 (1H, m)	C-2"
6''	60.39 (CH ₂)	3.46 (1H, dd, 9.5, 5.0)	C-3", C-4"
		3.29 (1H, dd, 9.5, 2.5)	C-3", C-4"
5-OH	-	12.60 (1H, s)	-

^{*} Carbon type was deduced from DEPT experiments.

FD6: 5,7,5",7"-Tetrahydroxy-2-(4-hydroxyphenyl)-2"-(4-methoxyphenyl)-[3',8"-Bi-4H-1-benzopyran]-4,4"-dione (podocarpusflavone A)

FD6 was isolated as a yellow solid, m.p. 270-271 °C. EIMS showed the molecular ion of 552 which corresponded to $C_{31}H_{20}O_{10}$. The 1H NMR spectrum (**Table 74**) showed the presence of an ABX pattern (δ 7.89, H-2'; 7.20, H-5' and 7.82, H-6'), an aromatic protons with *meta* coupling (δ 6.27, H-6 and 6.39, H-8), an AA'BB' system (δ 7.55, H-2''', H-6''' and 6.84, H-3''', H-5'''), an isolated aromatic proton (δ 6.55, s, H-6''), two methine olefinic protons (δ 6.55, s, H-3 and 6.57, s, H-3'') and a methoxy group (δ 3.79, s, 4'''-OCH₃). The HMBC correlation of methoxy protons to C-4''' supported the location of 4'''-OCH₃. The HMBC correlations of H-3 to C2, C-4, C-4a, C-1' and H-3'' to C-2'', C-4'', C-4a'', C-1''' confirmed that C-2 was connected to the 4-hydroxysubstituted ring at C-1' and C-2'' connected to the methoxysubstituted ring at C-1'''. These data indicated a linkage of two flavone units between C-3' and C-8''. Thus **FD6** was assigned to be 5,7,5'',7''-tetrahydroxy-2-(4-dihydroxyphenyl)-2''-(4-methoxyphenyl)-[3',8''-Bi-4H-1-benzopyran]-4,4''-dione which was known as podocarpusflavone A (Pelter, *et al.*, 1971; Dasgupta, *et al.*, 1981; Markham, *et al.*, 1987).



Major HMBC correlations of ${\bf FD6}$

Table 74 The NMR spectral data of FD6

Position	$\delta_{_{ m C}}$ (C-Type)*	δ_{H} (multiplicity, J_{Hz})	HMBC
2	163.78 (C)	-	-
3	103.06 (CH)	6.55 (1H, s)	C-2, C-4, C-4a, C-1'
4	181.80 (C=O)	-	-
4a	103.32 (C)	-	-
5	161.54 (C)	-	-
6	98.97 (CH)	6.27 (1H, d, 2.3)	C-4a, C-5, C-7, C-8
7	163.74 (C)	-	-
8	93.75 (CH)	6.39 (1H, d, 2.3)	C-4a, C-6, C-7, C-8a
8a	157.35 (C)	-	-
1'	121.70 (C)	-	-
2'	130.96 (CH)	7.89 (1H, d, 2.4)	C-2, C-3, C-4', C-6', C-8''
3'	119.64 (C)	-	-
4'	158.80 (C)	-	-
5'	116.33 (CH)	7.20 (1H, d, 8.6)	C-1', C-3', C-4'
6'	127.15 (CH)	7.82 (1H, dd, 8.6, 2.4)	C-2, C-2', C-4'
2''	163.24 (C)	-	-

Table 74 (Continued)

Position	$\delta_{_{ m C}}$ (C-Type)*	δ_{H} (multiplicity, J_{Hz})	HMBC
3''	103.07 (CH)	6.57 (1H, s)	C-2", C-4", C-4a", C-1""
4''	182.24 (C=O)	-	-
4a''	104.06 (C)	-	-
5''	160.97 (C)	-	-
6''	99.06 (CH)	6.55 (1H, s)	C-4a", C-5", C-7", C-8"
7''	163.74 (C)	-	-
8''	104.33 (C)	-	-
8a''	161.35 (C)	-	-
1'''	123.01 (C)	-	-
2"", 6""	127.43 (CH)	7.55 (2H, d, 9.6)	C-2", C-2"", C-4"", C-6""
3''', 5'''	113.97 (CH)	6.84 (2H, d, 9.6)	C-1''', C-3''', C-4'''
4'''	161.93 (C)	-	-
5-OH	-	12.82 (1H, s)	C-4a, C-5, C-6
5′′-ОН	-	12.96 (1H, s)	C-4a", C-5", C-6"
4'''-OCH ₃	55.00 (CH ₃)	3.79 (3H, s)	C-4'''

^{*} Carbon type was deduced from DEPT experiments.

FD7: (2S,2"S,3R)-2"-(3,4-dihydroxyphenyl)-7"-(β-D-glucopyranosyloxy)-2,2",3,3"-tetrahydro-5,5",7-trihydroxy-2-(4-hydroxyphenyl)-[3,8"-Bi-4H-1-benzopyran]-4,4"-dione (xanthochymusside)

FD7 was isolated as a yellow solid, m.p. 222-224 °C. EIMS showed the molecular ion of 720 which corresponded to $C_{36}H_{32}O_{16}$. Duplication of signals (in a ratio of 1 : 0.6) in the ¹H NMR spectrum of the optically active **FD7** in CD₃OD suggested the existence of two conformers at room temperature, the major reasons leading to the existence of such conformers was rotational restrictions about the interflavanone C-3,C-8" bond. The ¹H and ¹³C NMR spectra at 300 K also showed the signals forming their respective pairs. The ¹H signals of major conformer (**Table 75**) showed the resonances of an AA'BB' pattern (δ 7.23, H-2', H-6' and 6.64, H-3', H-5'), an aromatic protons with *meta* coupling (δ 5.90, H-6 and 5.88, H-8), an ABX pattern (δ 6.81, H-2'''; 6.73, H-5''' and 6.70, H-6'''), an isolated aromatic proton (δ 6.38, H-6''), a methine proton (δ 5.08, H-2'') and a methylene protons (δ 2.95 and 2.56, H-3''). The HMBC correlations of H-2'' to C-4'', C-8a'', C-1''' and H-3'' to C-2'', C-4'', C-1''' supported the assignment. The HMBC correlations of H-2 to C-1', C-2'', C-6' and H-3 to C-7'', C-8'', C-8a'' confirmed that C-2 was connected to 4-substituted ring at C-1' and C-3 was connected to another flavanone nucleus at C-8''.

The resonances of β -glucose were observed of which an anomeric proton H-1'''' resonated at δ 4.99 (1H, d, J = 6.0 Hz), a methylene protons H-6'''' were at δ 3.92 (1H, d, J = 12.0 Hz) and 3.75 (1H, m) and the other protons were at δ 3.30-3.49. According to HMBC correlation of H-1'''' to C-7'' indicated the location of glucose moiety was at C-7''. On the basis of above considerations **FD7** was assigned to be $(2S,2''S,3R)-2''-(3,4-\text{dihydroxyphenyl})-7''-(\beta-D-\text{glucopyranosyloxy})-2,2'',3,3''-\text{tetrahydro-5,5''},7-\text{trihydroxy-2-(4-hydroxyphenyl})-[3,8''-Bi-4H-1-benzopyran]-4,4''-dione which was known as xanthochymusside (Konoshima, <math>et$ al., 1970b).

Major HMBC correlations of FD7

Table 75 The NMR spectral data (major conformer) of ${\bf FD7}$ at 300 K

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
2	83.75 (CH)	5.70 (1H, d, 12.0)	C-3, C-4, C-1', C-2',6'
3	49.94 (CH)	5.03 (1H, d, 12.0)	C-2, C-4, C-4a, C-1', C-7", C-8", C-8a"
4	198.41 (C=O)	-	-
4a	103.05 (C)	-	-
5	165.42 (C)	-	-
6	97.55 (CH)	5.90 (1H, <i>br s</i>)	C-4a, C-5, C-7, C-8
7	167.90 (C)	-	-
8	97.15 (CH)	5.88 (1H, <i>br s</i>)	C-4a, C-6, C-7
8a	163.93 (C)	-	-
1'	131.46 (C)	-	-
2', 6'	129.23 (CH)	7.23 (2H, d, 7.9)	C-2, C-2',6', C-4'
3', 5'	114.76 (CH)	6.64 (2H, br d, 7.9)	C-1', C-3',5', C-4'
4'	158.77 (C)	-	-
2''	80.88 (CH)	5.08 (1H, br t)	C-4", C-8a", C-1""
3''	44.81 (CH ₂)	2.95 (1H, br dd)	C-2", C-4", C-1""
		2.56 (1H, <i>br dd</i>)	C-2", C-4", C-1""
4''	198.41 (C=O)	-	-
4a''	104.78 (C)	-	-
5''	165.42 (C)	-	-
6''	96.14 (CH)	6.38 (1H, s)	C-4a", C-5", C-7", C-8"
7''	164.71 (C)	-	-
8''	104.29 (C)	-	-
8a''	161.93 (C)	-	-
1'''	131.46 (C)	-	-
2'''	130.70 (CH)	6.81 (1H, br d, 2.0)	C-2", C-1"", C-3"", C-4"", C-6""
3'''	146.34 (C)	-	-
4'''	146.65 (C)	-	-

Table 75 (Continued)

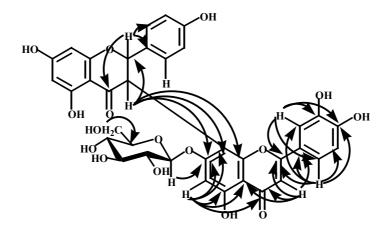
Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
5'''	116.42 (CH)	6.73 (1H, br d, 8.0)	C-1''', C-3'''
6'''	119.10 (CH)	6.70 (1H, br dd, 8.0, 2.0)	C-2", C-1"", C-2"", C-4"", C-5""
Glc-1''''	101.54 (CH)	4.99 (1H, d, 6.0)	C-7", C-3"", C-5""
2''''	75.01 (CH)	3.32-3.30 (1H, <i>m</i>)	C-3'''', C-5''''
3''''	78.25 (CH)	3.49-3.47 (1H, m)	C-2'''', C-3'''', C-4'''', C-5''''
4''''	70.95 (CH)	3.38 (1H, br t)	C-3'''', C-5''''', C-6'''''
5''''	78.11 (CH)	3.46-3.44 (1H, <i>m</i>)	C-3'''', C-5''''', C-6'''''
6''''	62.36 (CH ₂)	3.92 (1H, d, 12.0)	C-1'''', C-4''''
		3.75 (1H, m)	C-1'''', C-4''''

^{*} Carbon type was deduced from DEPT experiments.

FD8: (2*S*,3*R*)-2"-(3,4-Dihydroxyphenyl)-7"-(β-D-glucopyranosyloxy)-2,3-dihydro-5,7,5"-trihydroxy-2-(4-hydroxyphenyl)-[3,8"-Bi-4H-1-benzopyran]-4,4"-dione (fukugeside)

FD8 was isolated as a yellow solid, m.p. 245-247 °C. EIMS showed the molecular ion of 718 which corresponded to $C_{36}H_{30}O_{16}$. Duplication of signals (in a ratio of 1 : 0.27) in the ¹H NMR spectrum in CD₃OD suggested the existence of two conformers at room temperature, the major reason leading to the existence of such conformers is rotational restriction about the interflavanyl C-3,C-8" bond. The ¹H NMR spectrum at 300 K also showed the signals forming their respective pairs. The ¹³C NMR spectrum at 300 K showed fifty five signals with thirty two pairs. The ¹⁴H major conformer (**Table 76**) showed the AA'BB' resonances of H-2', H-6' at δ 7.09 (2H, d, J = 8.3 Hz) and H-3', H-5' at δ 6.34 (2H, d, J = 8.3 Hz), broad singlets of H-6 at δ 5.90 and H-8 at δ 5.97, doublets of H-2 at δ 5.70 (J = 11.9 Hz) and H-3 at δ 4.86 (J = 11.9 Hz), singlet of H-3" at δ 6.35 and singlet of H-6" at δ 6.58. The HMBC correlations of H-6" to C-4", C-4a", C-5", C-7", C-8" and H-3" to C-2", C-4", C-4a", C-1" supported the assignment. The resonances of H-2" (δ 7.29, δ 8.9 Hz) and H-5" (δ 6.84, δ 8.9 Hz) were shown as an ABX pattern. The HMBC correlation of H-2 to C-4, C-1' and C-2' confirmed that C-2 was

connected to the 4-substituted ring at C-1'. The correlation of H-3 to C-2, C-7'', C-8'' and C-8a'' suggested that C-3 was connected to another flavone nucleus at C-8''. The resonances of sugar moiety were observed at δ 5.14-3.30 of which anomeric proton H-1'''' resonating at δ 5.14 (1H, d). The coupling constant of 7.6 Hz suggested a β -glucose moiety. According to HMBC correlation of H-1'''' to C-7'' indicated the location of glucose moiety at C-7''. On the basis of above considerations **FD8** was assigned to be (2S,3R)-2''-(3,4-dihydroxyphenyl)-7''- $(\beta$ -D-glucopyranosyloxy)-2,3-dihydro-5,7,5''-trihydroxy-2-(4-hydroxyphenyl)-[3,8''-Bi-4H-1-benzopyran]-4,4''-dione which was known as fukugeside (Kanoshima, *et al.*, 1970a; 1970b).



Major HMBC correlations of FD8

Table 76 The NMR spectral data (major conformer) of ${\bf FD8}$ at 300 K

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
2	82.82 (CH)	5.70 (1H, d, 11.9)	C-4, C-1', C-2'
3	50.96 (CH)	4.86 (1H, d, 11.9)	C-2, C-4, C-7", C-8", C-8a"
4	197.76 (C=O)	-	-
4a	103.44 (C)	-	-
5	166.52 (C)	-	-
6	97.74 (CH)	5.90 (1H, <i>br s</i>)	C-4a, C-5, C-7, C-8
7	168.23 (C)	-	-
8	96.50 (CH)	5.97 (1H, br s)	C-4a, C-6, C-7, C-8a
8a	164.76 (C)	-	-
1'	130.36 (C)	-	-
2', 6'	129.35 (CH)	7.09 (2H, d, 8.3)	C-2, C-1', C-2',6', C-3',5', C-4'
3', 5'	115.40 (CH)	6.34 (2H, d, 8.3)	C-1', C-2',6', C-3',5', C-4'
4′	158.37 (C)	-	-
2''	165.65 (C)	-	-
3''	104.02 (CH)	6.35 (1H, s)	C-2", C-4", C-4a", C-1"
4''	183.79 (C=O)	-	-
4a''	106.39 (C)	-	-
5''	162.62 (C)	-	-
6''	99.43 (CH)	6.58 (1H, s)	C-4", C-4a", C-5", C-7", C-8"
7''	161.52 (C)	-	-
8''	104.46 (C)	-	-
8a''	156.57 (C)	-	-
1'''	130.25 (C)	-	-
2'''	114.40 (CH)	7.29 (1H, <i>br s</i>)	C-2", C-3"", C-4"", C-6""
3'''	146.52 (C)	-	-
4'''	151.02 (C)	-	-
5'''	116.79 (CH)	6.84 (1H, br d, 8.9)	C-3''', C-4'''

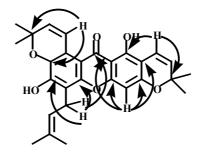
Table 76 (Continued)

Position	$\delta_{_{\mathbb{C}}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
6'''	120.88 (CH)	7.24 (1H, br d, 8.9)	C-2", C-2"", C-4"", C-5""
Glc-1''''	101.40 (CH)	5.14 (1H, d, 7.6)	C-7''
2''''	75.14 (CH)	3.51 (1H, m)	C-1'''', C-3'''', C-4''''
3''''	78.43 (CH)	3.49 (1H, m)	C-1'''', C-2'''', C-4''''
4''''	71.01 (CH)	3.30 (1H, <i>m</i>)	C-3'''', C-5'''', C-6''''
5''''	78.26 (CH)	3.34 (1H, m)	C-4'''', C-6''''
6''''	62.40 (CH ₂)	3.92 (1H, d, 10.9)	C-4'''', C-5''''
		3.73 (1H, dd, 5.6)	C-4'''', C-5''''

^{*} Carbon type was deduced from DEPT experiments.

FD9: 1,6-Dihydroxy-5-(3-methyl-2-butenyl)-2',2'-dimethylchromeno(5',6':2,3)-2''',2'''-dimethylchromeno(5''',6''':8,7)xanthone

FD9 was isolated as an orange solid, m.p. 218-220 °C. HRFABMS m/z 460.1929 $[\mathrm{M}]^{^{+}}$ (calcd for $\mathrm{C_{28}H_{28}O_6}$, 460.1886). The UV spectrum exhibited maximum absorption bands at 335.4, 301.0, 289.1, 221.0 and 205.6 nm. The absorption bands of O-H stretching and C=O stretching were shown in the IR spectrum at 3424 and 1615 cm⁻¹, respectively. The ¹H NMR spectrum (**Table 77**) showed the signals of two phenolic hydroxy groups at δ 13.70 (1-OH) and 6.34 (6-OH), an isolated aromatic proton at δ 6.30 (H-4). Characteristic signals of a prenyl group was present at δ 3.57 (H-1"), 5.28 (H-2"), 1.69 (H-4") and 1.88 (H-5") and the group was located at C-5 according to HMBC correlation of H-1" to C-6 and C-10a. The signals of four methyl groups at δ 1.49 (2"'-Me₂), 1.47 (2'-Me₂) and the vicinal olefinic protons at δ 5.77 (d, H-3'''), 7.98 (d, H-4'''), 5.57 (d, H-3') and 6.73 (d, H-4') associated with two chromene rings were present. The ³J correlations of H-4" to C-7 and H-4 to C-1 suggested that the chromene ring were connected to the parent structure at C-7, C-8 and C-2, C-3. The complete HMBC data confirmed the structure; 1,6-dihydroxy-5-(3 $methyl-2-butenyl)-2^{\prime},2^{\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime},2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime}-dimethylchromeno(5^{\prime},6^{\prime}:2,3)-2^{\prime\prime}-dimethylchr$ (5"',6"':8,7)xanthone was assinged for **FD9**. This compound appears to be novel.



Major HMBC correlations of FD9

Table 77 The NMR spectral data of FD9

Position	$\delta_{_{ m C}}$ (C-Type)*	δ_{H} (multiplicity, J_{Hz})	HMBC
1	159.81 (C)	-	-
2	103.83 (C)	-	-
3	157.78 (C)	-	-
4	94.20 (CH)	6.30 (1H, s)	C-3, C-4', C-2, C-9, C-4a
4a	150.96 (C)	-	-
5	108.36 (C)	-	-
6	148.63 (C)	-	-
7	136.58 (C)	-	-
8	117.14 (C)	-	-
8a	115.33 (C)	-	-
9	182.83 (C=O)	-	-
9a	104.29 (C)	-	-
10a	156.51 (C)	-	-
2'	77.95 (C)	-	-
3'	127.09 (CH)	5.57 (1H, d, 10.2)	C-2', C-2
4'	115.71 (CH)	6.73 (1H, d, 10.2)	C-2', C-1
1''	22.59 (CH ₂)	3.57 (2H, d, 7.5)	C-6, C-10a, C-8a

Table 77 (Continued)

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2''	120.96 (CH)	5.28 (1H, br t, 5.7)	C-5, C-4", C-5"
3''	132.67 (C)	-	-
4''	25.81 (CH ₃)	1.69 (3H, s)	C-2", C-3", C-5"
5''	18.00 (CH ₃)	1.88 (3H, s)	C-2", C-3", C-4"
2'''	76.87 (C)	-	-
3'''	131.39 (CH)	5.77 (1H, d, 10.2)	C-2''', C-8
4'''	120.99 (CH)	7.98 (1H, d, 10.2)	C-2''', C-7
1-OH	-	13.70 (1H, s)	C-2, C-1, C-9a
6-OH	-	6.34 (1H, s)	-
2'-(CH ₃) ₂	28.37 (CH ₃)	1.47 (6H, s)	C-2', C-3', 2'-(CH ₃) ₂
2'''-(CH ₃) ₂	27.43 (CH ₃)	1.49 (6H, s)	C-2''', C-3''', 2'''-(CH ₃) ₂

^{*} Carbon type was deduced from DEPT experiments.

FD10: (2S,2"S,3R)-2"-(3,4-dihydroxyphenyl)-2,3,2",3"-tetrahydro-5,7,5",7"-tetrahydroxy-2-(4-hydroxyphenyl)-[3,8"-Bi-4H-1-benzopyran]-4,4"-dione (GB-2a)

FD10 was isolated as a pale yellow solid, m.p. 215-216 °C. Duplication of signals (in a ratio of 1:0.4) in the ¹H NMR spectrum of the optically active FD10 in $\mathrm{DMSO}\text{-}d_{\scriptscriptstyle{6}}$ suggested the existence of two conformers at room temperature, the major reasons leading to the existence of such conformers being rotational restrictions about the interflavanone C-3, C-8" bond. The major conformer (Table 78) showed the resonances of aromatic protons H-2', H-6' and H-3', H-5' as an AA'BB' system at δ 7.13 and 6.79 (2H each, J = 8.5 Hz) and the resonances of H-2", H-5" and H-6" as an ABX pattern at δ 7.10 (d, J = 2.0 Hz), 6.64 (d, J = 8.5 Hz) and 6.73 (dd, J = 8.5 and2.0 Hz). It further showed the resonances of *meta* protons H-6 and H-8 at δ 5.90 and 5.79 (br s each) and an isolated aromatic proton H-6" at δ 5.86 (s). The resonances of trans-coupled protons H-2 and H-3 were observed at δ 5.73 and 4.56 (d, J = 12.3 Hz) whereas H-2" and H-3" resonated at δ 5.43 (br t) and 2.90 (m), 2.60 (m), respectively. The HMBC correlation of H-2 to C-1', C-2' and C-6' confirmed that the 4-substituted ring was connected to at C-2 and the correlation of H-2" to C-1" and C-2" confirmed that the 3,4-disubstituted ring was connected to C-2". The 3,8"-link of two flavanone was suggested from HMBC correlation of H-3 to C-7", C-8" and C-8a".

The complete HMBC confirmed the structure of **FD10** as (2S,3R,2''S)-2''-(3,4-dihydroxyphenyl)-2,3,2'',3''-tetrahydro-5,7,5'',7''-tetrahydroxy-2-(4-hydroxyphenyl)-[3,8''-Bi-4H-1-benzopyran]-4,4''-dione or known as GB-2a (Jackson,*et al.*1971; Gunatilaka,*et al.*, 1984).

Table 78 The NMR spectral data (major conformer) of **FD10** at 300 K

Position	$\delta_{_{\mathbb{C}}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2	82.20 (CH)	5.73 (1H, d, 12.3)	C-4, C-1', C-2',6'
3	47.76 (CH)	4.56 (1H, d, 12.3)	C-2, C-4, C-7", C-8", C-8a"
4	197.14 (C=O)	-	-
4a	101.81 (C)	-	-
5	165.34 (C)	-	-
6	96.51 (CH)	5.90 (1H, br s)	C-4, C-4a, C-5, C-7, C-8
7	166.86 (C)	-	-
8	95.93 (CH)	5.79 (1H, br s)	C-4, C-4a, C-7, C-8a
8a	163.27 (C)	-	-

Table 78 (Continued)

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1'	130.23 (C)	-	-
2', 6'	129.45 (CH)	7.13 (2H, d, 8.5)	C-2, C-1', C-2',6', C-4'
3', 5'	114.38 (CH)	6.79 (2H, d, 8.5)	C-2',6', C-3',5', C-4'
4'	158.13 (C)	-	-
2''	79.02 (CH)	5.43 (1H, <i>br t</i>)	C-4", C-1"", C-2""
3''	43.14 (CH ₂)	2.90 (1H, m)	C-2", C-4", C-1""
		2.60 (1H, <i>m</i>)	C-2", C-4", C-1""
4''	196.62 (C=O)	-	-
4a''	102.12 (C)	-	-
5''	164.09 (C)	-	-
6''	95.50 (CH)	5.86 (1H, s)	C-4, C-4a", C-5", C-7", C-8"
7''	161.19 (C)	-	-
8''	101.76 (C)	-	-
8a''	162.48 (C)	-	-
1'''	128.37 (C)	-	-
2'''	129.04 (CH)	7.10 (1H, d, 2.0)	C-2", C-3"", C-4"", C-6""
3'''	145.68 (C)	-	-
4'''	146.17 (C)	-	-
5'''	115.32 (CH)	6.64 (1H, d, 8.5)	C-1''', C-3''', C-4'''
6'''	118.25 (CH)	6.73 (1H, dd, 8.5, 2.0)	C-2", C-1"", C-2"", C-4""
5-OH	-	12.19 (1H, s)	C-4a, C-5, C-6
4'-OH	-	9.61 (1H, s)	C-4', C-3',5'
5′′-ОН	-	12.15 (1H, s)	C-4a", C-5", C-6"
3′′′-ОН	-	9.20 (1H, s)	C-3", C-4"
4′′′-OH	-	9.08 (1H, s)	C-3", C-4", C-5"

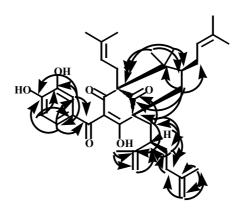
^{*} Carbon type was deduced from DEPT experiments.

FD11: (4R,6R,8S)-2-(13,14-dihydroxybenzoyl)-1-hydroxy-8-[(2S)-2-isopropenyl-5-methyl-5-hexenyl]-5,5-dimethyl-4,6-bis(3-methyl-2-butenyl)bicyclo [3.3.1]non-1-ene-3,9-dione (xanthochymol)

HO 15
$$\frac{20}{11}$$
 $\frac{21}{0}$ $\frac{22}{3}$ $\frac{23}{9}$ $\frac{24}{7}$ $\frac{28}{11}$ $\frac{3}{3}$ $\frac{9}{31}$ $\frac{29}{31}$ $\frac{1}{34}$ $\frac{29}{38}$ $\frac{1}{37}$

FD11 is a yellow solid, m.p. 139-140 °C. The optical rotation was $[\alpha]_D^{29}$ + 135.5 ° (c = 0.03 in CH₃OH). The resonances appearing as an ABX system at δ 6.66 (d, J = 8.5 Hz, H-15), 6.99 (dd, J = 8.5, 1.5 Hz, H-16) and 7.04 (d, J = 1.5 Hz, H-12) and the resonance of carbonyl carbon at δ 198.08 refered the presence of 3,4-dihydroxybenzoyl group. The resonances of protons of a prenyl group were observed at δ 5.10 (H-18), 2.77 (H-17) and 1.72 (H-20, H-21) and was found to be linked to quaternary carbon C-4 according to HMBC correlation of H-17 to C-3 and C-5. The resonances at δ 4.93 (H-25), 1.70 (H-27), 1.56 (H-28) and 1.28 (H-24) suggested the presence of the second prenyl group and it was indicated to be C-6 by 3J correlation of H-24 to C-5 and C-7. The methylene protons H-7 (δ 2.37), H-29 (δ 2.22), H-34 (δ 1.35) and H-35 (δ 2.04) were assigned from the HMBC experiments as followed: H-29 to C-9, C-34; H-30 to C-35; H-34 to C-29, C-30 and C-35. The resonances of H-32 (δ 4.43, 2H), H-33 (δ 1.83, 3H), H-37 (δ 4.65, 2H) and H-38 (δ 1.76, 3H) of isopropenyl groups were assigned to be linked to C-30 and C-35 by 3J correlations of H-32 to C-30; H-33 to C-31, C-32; H-37 to C-35, C-38 and H-38 to C-36, C-37. Methyl groups

on a saturated carbon resonated at δ 1.04 (H-22) and 1.19 (H-23) whereas methine protons H-6 and H-30 resonated at δ 1.90 and 2.62. The ¹³C NMR spectrum showed a non-enolisable carbonyl carbon at δ 209.41 and three enolisable carbonyl carbons at δ 198.08, 194.22 and 193.98. The HMBC correlations corresponded to the assigned structure. **FD11** was then identified as (4R,6R,8S)-2-(3,4-dihydroxybenzoyl)-1-hydroxy-8-[(2S)-2-isopropenyl-5-methyl-5-hexenyl]-5,5-dimethyl-4,6-bis(3-methyl-2-butenyl)bicyclo[3.3.1]non-1-ene-3,9-dione which agreed with the structure of xanthochymol (Blount and Williams, 1976; Rama Rao, *et al.*, 1980).



Major HMBC correlations of FD11

 $Table \ 79 \ \ The \ NMR \ spectral \ data \ of \ FD11$

D:4:	$\delta_{\rm C}$ (C	-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multipli	$(city, J_{\rm Hz})$	IMDC
Position	FD11	xanthochymol	FD11	xanthochymol	HMBC
1	193.98(C=O)	194.0 (C=O)	-	-	-
2	115.83 (C)	115.2 (C)	-	-	-
3	194.22(C=O)	194.9 (C=O)	-	-	-
4	69.68 (C)	70.0 (C)	-	-	-
5	49.55 (C)	49.8 (C)	-	-	-
6	46.74 (CH)	46.7 (CH)	1.90 (1H, m)	1.00-2.30 (m)	C-5, C-8, C-25
7	42.52 (CH ₂)	42.9 (CH ₂)	2.37 (2H, m)	1.00-2.30 (m)	C-5, C-8, C-9, C-24
8	57.78 (C)	58.1 (C)	-	-	-
9	209.41(C=O)	209.2 (C=O)	-	-	-
10	198.08(C=O)	198.6 (C=O)	-	-	-
11	127.89 (C)	127.9 (C)	-	-	-
12	116.45 (CH)	116.6 (CH)	7.04 (1H, d, 1.5)	7.00 (d, 2.0)	C-10, C-13, C-14, C-16
13	143.72 (C)	143.8 (C)	-	-	-
14	147.35 (C)	147.5 (C)	-	-	-
15	114.33 (CH)	114.2 (CH)	6.66 (1H, d, 8.5)	6.60 (d, 9.0)	C-11, C-13, C-14
16	120.04 (CH)	120.13 (CH)	6.99(1H,dd,8.5,1.5)	6.95 (d,9.0,2.0)	C-10, C-12, C-14, C-15
17	27.16 ^a (CH ₂)	27.2 ^{aa} (CH ₂)	2.77 (2H, m)	1.00-2.30 (m)	C-3, C-5
18	123.85 (CH)	123.9 (CH)	$5.10^{\circ} (1H, m)$	$5.00^{ee} (m)$	C-17, C-20, C-21
19	132.87 (C)	133.1 (C)	-	-	-
20	26.01 ^b (CH ₃)	26.3 ^{bb} (CH ₃)	1.72 ^f (3H, s)	1.71^{ff} (3H, s)	C-18, C-19, C-21
21	18.10°(CH ₃)	18.2 ^{cc} (CH ₃)	1.72^{f} (3H, s)	$1.56^{\text{ff}}(3\text{H}, s)$	C-18, C-19, C-20
22	26.91 (CH ₃)	26.6 (CH ₃)	1.04 (3H, s)	1.02 (3H, s)	C-4, C-5, C-6, C-23
23	17.14 (CH ₃)	17.4 (CH ₃)	1.19 (3H, s)	1.18 (3H, s)	C-4, C-5, C-6, C-22
24	29.26 ^a (CH ₂)	29.2 ^{aa} (CH ₂)	1.28 (2H, br s)	1.00-2.30	C-5, C-7, C-25, C-26
25	124.11(CH)	124.2 (CH)	4.93° (1H, m)	4.90 ^{ee} (1H, m)	C-24, C-27, C-28
26	135.21 (C)	135.4 (C)	-	-	-
27	25.78 ^b (CH ₃)	26.1 ^{bb} (CH ₃)	$1.70^{\rm f}$ (3H, s)	1.71^{ff} (3H, s)	C-25, C-26, C-28
28	17.84°(CH ₃)	18.5° (CH ₃)	1.56 ^f (3H, s)	1.56 ^{ff} (3H, s)	C-25, C-26, C-27

Table 79 (Continued)

Dogition	$\delta_{\rm c}$ (C-	-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)		IMDC
Position	FD11	xanthochymol	FD11	xanthochymol	HMBC
29	36.46 (CH ₂)	36.8 (CH ₂)	2.22 (2H, m)	1.00-2.30 (m)	C-9, C-34
30	43.27 (CH)	43.6 (CH)	2.62 (1H, m)	1.00-2.30 (m)	C-8, C-35
31	149.79 (C)	149.9 (C)	-	-	-
32	113.21(CH ₂)	113.5 (CH ₂)	4.43 (2H, br d, 4.0)	4.43 (2H, s)	C-30, C-33
33	22.53 ^d (CH ₃)	22.8 ^{dd} (CH ₃)	1.83 (3H, s)	1.81 (3H, s)	C-31, C-32
34	31.58 (CH ₂)	32.1 (CH ₂)	1.35 (2H, m)	1.00-2.30 (m)	C-29, C-30, C-35
35	35.42 (CH ₂)	35.7 (CH ₂)	2.04 (2H, m)	1.00-2.30 (m)	C-34, C-37, C-38
36	145.94 (C)	146.1 (C)	-	-	-
37	109.58(CH ₂)	109.8 (CH ₂)	4.65 (2H, <i>br d</i> ,12.5)	4.65 (2H, s)	C-35, C-38
38	22.64 ^d (CH ₃)	23.0 ^{dd} (CH ₃)	1.76 (3H, s)	1.75 (3H, s)	C-36, C-37

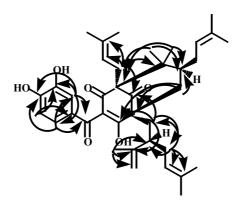
^{*} Carbon type was deduced from DEPT experiments.

^{a-f, aa-ff} Assignment maybe interchangeable.

FD12: (4S,6R,8S)-2-(3,4-dihydroxybenzoyl)-1-hydroxy-8-[(2S)-2-isopropenyl-5-methyl-4-hexenyl]-5,5-dimethyl-4,6-bis(3-methyl-2-butenylbicyclo[3.3.1] non-1-ene-3,9-dione (guttiferone E)

FD12 is a yellow solid, m.p. 128-130 °C. The optical rotation was $[\alpha]_D^{29}$ + 96.0 ° (c = 0.03 in CH₃OH). The ¹H NMR spectrum (**Table 80**) showed the ABX resonances of H-15 at δ 6.65 (d, J = 8.1 Hz), H-16 at δ 6.95 (dd, J = 8.1, 2.0 Hz) and H-12 at δ 7.01 (d, J = 2.0 Hz). The presence of three prenyl groups were observed. The signals at δ 5.09 (H-18), 2.75 and 2.60 (H-17), 1.81 (H-20), 1.71 (H-21) associated to a prenyl group at C-4. Whereas those at δ 5.05 (H-25), 1.74 (H-28), 1.71 (H-27), 1.32 (H-24) and δ 4.94 (H-35), 2.20 (H-34), 1.68 (H-37), 1.61 (H-38) belonged to the prenyl groups attached to CH-carbons. HMBC correlations of H-17 to C-4, C-5; H-24 to C-7 and H-34 to C-29, C-30, C-35, C-36 indicated that the prenyl groups were at C-4, C-6 and C-30, respectively. The signals of isopropenyl group at δ 4.44, 4.39 (H-32) and 1.55 (H-33) of which this group was located at C-30 according to the HMBC correlations of H-32 and H-33 to C-30. Two methyl groups on a saturated carbon were resonated at δ 1.17 (H-22) and 1.02 (H-23) and were located at C-5 from the correlations of H-22 to C-4, C-5, C-6 and H-23 to C-4, C-6 in HMBC. Two groups of methylene protons were assigned from the resonances at δ 2.12, 2.05 (H-7) and 2.15,

1.91 (H-29) whereas two methine protons were indicated from the resonances at δ 1.45 (H-6) and 2.36 (H-30). Four signals of carbonyl carbons were shown in the ¹³C NMR spectrum which corresponded to a non-enolisable carbonyl carbon (δ 209.25) and three enolisable carbonyl carbons (δ 198.59, 194.56 and 193.90). **FD12** is virtually identical to **GD1** (camboginol or garcinol). However, the optical rotation of **FD12** was opposite in sign compared to that of **GD1**, $[\alpha]_D^{29}$ –128.5°. Thus, **FD12** was then identified as (4S,6R,8S)-2-(3,4-dihydroxybenzoyl)-1-hydroxy-8-[(2S)-2-isopropenyl-5-methyl-4-hexenyl]-5,5-dimethyl-4,6-bis(3-methyl-2-butenyl)bicyclo[3.3.1]non-1-ene-3,9-dione or known as guttiferone E (Gustafson, *et al.*, 1992).



Major HMBC correlations of FD12

 $Table~80~{\rm The~NMR~spectral~data~of~FD12}$

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-	Type)*	$\delta_{_{ m H}}$ (multipl	$(icity, J_{ m Hz})$	IMDC
Position	FD12	guttiferoneE	FD12	guttiferone E	HMBC
1	194.56 (C)	194.8	-	-	-
2	115.88 (C)	118.6	-	-	-
3	193.90(C=O)	194.0	-	-	-
4	69.74 (C)	69.4	-	-	-
5	49.59 (C)	49.9	-	-	-
6	46.83 (CH)	48.0	1.45 (1H, m)	1.47 (m)	C-4, C-8, C-22, C-23,
					C-25
7	42.63 (CH ₂)	43.3	2.05 (1H, m)	2.06 (<i>dd</i> ,14.1,7.2)	C-1, C-6, C-8
			2.12 (1H, m)	2.21 (<i>dd</i> ,14.0,1.0)	C-1, C-6, C-8
8	57.88 (C)	59.8	-	-	-
9	209.25(C=O)	208.9	-	-	-
10	198.59(C=O)	195.8	-	-	-
11	127.86 (C)	129.7	-	-	-
12	116.51(CH)	117.2	7.01 (1H, d, 2.0)	7.22 (d, 2.2)	C-10, C-13, C-14, C-16
13	143.73 (C)	146.2	-	-	-
14	148.03 (C)	152.5	-	-	-
15	114.34 (CH)	115.0	6.65 (1H, d, 8.1)	6.69 (d, 8.3)	C-11, C-13, C-14
16	124.10 (CH)	125.2	6.95(1H,dd,8.1,2.0)	7.00 (dd, 8.3, 2.2)	C-10, C-12, C-14
17	26.38 (CH ₂)	27.0	2.60 (1H, br d)	2.48 (<i>dd</i> ,13.3,4.6)	C-4, C-5, C-18, C-19
			2.75 (1H, <i>br d</i>)	2.67 (dd,13.3,9.4)	C-4, C-5, C-18, C-19
18	120.13(CH)	121.5	5.09 (1H, br t)	4.98 (m)	C-20, C-21
19	135.21 (C)	135.3	-	-	-
20	26.05 (CH ₃)	26.4	1.81 (3H, s)	1.71 (d, 1.0)	C-18, C-19, C-21
21	17.94 (CH ₃)	18.5 ^a	1.71 (3H, s)	1.65 (s)	C-18, C-19, C-20
22	22.71 (CH ₃)	23.2	1.17 (3H, s)	1.12 (s)	C-4, C-5, C-6, C-23
23	27.02 (CH ₃)	27.3	1.02 (3H, s)	0.95 (s)	C-4, C-6, C-22
24	29.68 (CH ₂)	30.5	1.32 (2H, <i>br s</i>)	2.06 (2H, m)	C-7, C-26
25	122.65(CH)	125.6	5.05 (1H, br t)	4.88 (m)	C-6, C-27, C-28
26	132.90 (C)	133.5	-	-	-

Table 80 (Continued)

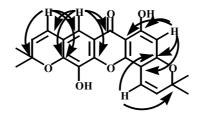
Position	$\delta_{\rm C}$ (C-'	Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)		e)* $\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)		НМВС
Position	FD12	guttiferoneE	FD12	guttiferone E	HMBC		
27	25.82 (CH ₃)	25.9 ^b	1.71 (3H, s)	1.65 (s)	C-25, C-26, C-28		
28	18.16 (CH ₃)	18.3°	1.74 (3H, s)	1.52 (s)	C-25, C-26, C-27		
29	39.19 (CH ₂)	36.6	1.91 (1H, m)	1.71 (dd,13.8,3.7)	C-9, C-31, C-34		
			2.15 (1H, m)	2.23(dd,13.8,10.6)	C-1, C-30		
30	43.60 (CH)	45.9	2.36 (1H, m)	2.34 (m)	C-8, C-33, C-34		
31	149.81 (C)	150.0	-	-	-		
32	112.68(CH ₂)	112.5	4.39 (1H, br s)	4.35 (d, 2.4)	C-30, C-33		
			4.44 (1H, br s)	4.54 (dd, 2.4, 1.2)	C-30, C-33		
33	17.64 (CH ₃)	18.2	1.55 (3H, s)	1.73 (s)	C-30, C-31, C-32		
34	28.93 (CH ₂)	33.9	2.20 (2H, m)	2.01 (m)	C-29, C-30, C-35, C-36		
35	123.90(CH)	123.9	4.94 (1H, <i>br t</i>)	4.98 (m)	C-37, C-38		
36	131.95 (C)	132.8	-	-	-		
37	25.73 (CH ₃)	26.0 ^b	1.68 (3H, s)	1.63 (d, 1.2)	C-35, C-36, C-38		
38	17.89 (CH ₃)	18.1	1.61 (3H, s)	1.57 (d, 1.0)	C-35, C-36, C-37		

^{*} Carbon type was deduced from DEPT experiments.

^{a, b} Assignment maybe interchangeable.

FD13: 1,5-Dihydroxy-2',2'-dimethylchromeno(5',6':4,3)-2'',2''-dimethylchromeno(5'',6'':7,6)xanthone (rheediaxanthone A)

FD13 was obtained as a yellow solid with m.p. 256-257 °C. The ¹H NMR spectrum (Table 81) consisted of the characteristic signal of a chelated hydroxy proton (1-OH) at δ 13.18, a sharp *singlet* signal of isolated aromatic proton H-8 at δ 7.54 and a doublet signal of aromatic proton H-2 at δ 6.31 (J = 0.6 Hz). Aromatic protons H-8 and H-2 were deduced from the HMBC correlations of H-8 to C-6, C-9, C-10a, C-4" and H-2 to C-1, C-3, C-4, C-9a. The ¹H NMR spectrum further showed the characteristic signals of two dimethylchromene rings. Gem-dimethyl protons resonated as a singlet at δ 1.59 (6H) and two olefinic protons (H-3" and H-4") appeared as two doublet signals at δ 5.79 and 6.51 (1H each, J = 10.5 Hz). Irradiation of the lowestfield aromatic H-8 caused a NOE enhancement of the olefinic proton H-4", suggesting the fusion of the dimethylchromene ring to C-6 and C-7 with an ether linkage at C-6. In addition, HMBC correlations of H-4" to C-6 and C-8 were shown. The gemdimethyl protons and two olefinic protons (H-3' and H-4') of another dimethylchromene ring resonated at δ 1.54 (s, 6H), 5.67 (d, J = 9.6 Hz, H-3') and 6.95 (dd, J = 9.6, 0.6 Hz, H-4'), respectively. The HMBC correlations of H-4' to C-3 and H-3' to C-4 precisely determined that it was at C-3 and C-4. The ¹³C NMR and DEPT experiments indicated the existence of a carbonyl carbon, four methyl carbons, six methine carbons and twelve quaternary carbons. FD13 then was assigned for 1,5dihydroxy-2,2-dimethylchromeno(5',6':4,3)-2,2-dimethylchromeno(5'',6'':7,6) xanthone which corresponded to rheediaxanthone A (Delle Monache, et al., 1981).



 $Table~81~\ The~NMR~spectral~data~of~FD13$

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	163.09 (C)	-	-
2	99.38 (CH)	6.31 (1H, d, 0.6)	C-1, C-3, C-4, C-9a
3	160.53 (C)	-	-
4	101.36 (C)	-	-
4a	151.57 (C)	-	-
5	132.37 (C)	-	-
6	145.01 (C)	-	-
7	114.73 (C)	-	-
8	113.48 (CH)	7.54 (1H, s)	C-6, C-9, C-10a, C-4"
8a	117.79 (C)	-	-
9	180.31 (C=O)	-	-
9a	102.35 (C)	-	-
10a	144.72 (C)	-	-
2'	78.19 (C)	-	-
3'	127.14 (CH)	5.67 (1H, d, 9.6)	C-4, C-2'
4'	115.16 (CH)	6.95 (1H, dd, 9.6, 0.6)	C-3, C-2'
2''	78.98 (C)	-	-
3''	131.01 (CH)	5.79 (1H, d, 10.5)	C-2", C-4"

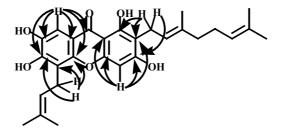
Table 81 (Continued)

Position	$\delta_{_{\mathrm{C}}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
4''	121.43 (CH)	6.51 (1H, d, 10.5)	C-6, C-8, C-2"
1-OH	-	13.18 (1H, s)	-
2'-(CH ₃) ₂	28.27 (CH ₃)	1.54 (6H, s)	C-2', C-3', 2'-(CH ₃) ₂
2"-(CH ₃) ₂	28.45 (CH ₃)	1.59 (6H, s)	C-2", C-3", 2"-(CH ₃) ₂

^{*} Carbon type was deduced from DEPT experiments.

FD14: 1,3,6,7-Tetrahydroxy-2-(3,7-dimethyl-2,6-octadienyl)-5-(3-methyl-2-butenyl)xanthone

FD14 is a yellow solid, m.p. 185-187 $^{\circ}$ C. Its molecular formula of $C_{28}H_{32}O_6$ was established on the basis of mass spectrum ($[M]^{+}m/z$ 464.2199). The UV spectrum showed the maxima absorption bands at 369.1, 322.7, 258.4, 242.9 and 207.4 nm. The IR spectrum showed the absorption bands of hydroxy group at 3450 cm⁻¹ and conjugated carbonyl group at 1711 cm⁻¹. The ¹H NMR spectrum (**Table 82**) exhibited the resonances of a chelated hydroxy proton 1-OH at δ 13.19, aromatic protons H-4 at δ 6.42 and H-8 at δ 7.33. The characteristic signals of a prenyl unit were displayed at δ 3.60 (d, H-1"), 5.29 (br t, H-2"), 1.69 (s, H-4"), 1.88 (s, H-5"). The prenyl unit was located at C-5 by HMBC correlation of H-1" to C-5, C-6 and C-10a. It also exhibited the typical signal of a geranyl group: three singlets of three methyl groups at δ 1.57 (H-8'), 1.80 (H-9') and 1.52 (H-10'), a doublet of methylene protons at δ 3.37 (H-1'). two multiplets of methylene protons at δ 1.99 (H-4') and 2.03 (H-5') and two broad triplets of two olefinic methine protons at δ 5.29 (H-2') and 5.02 (H-6'). The correlation of H-1' to C-1, C-2 and C-3 in the HMBC indicated that the geranyl side chain was at C-2. The ¹³C NMR spectral data (Table 89) showed 28 signals for 28 carbon atoms. These evidences indicated that **FD14** was 1,3,6,7-tetrahydroxy-2-(3,7dimethyl-2,6-octadienyl)-5-(3-methyl-2-butenyl)xanthone. This compound was a new xanthone derivative.



Major HMBC correlations of FD14

Table 82 The NMR spectral data of FD14

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
1	159.82 (C)	-	-
2	110.43 (C)	-	-
3	162.59 (C)	-	-
4	93.44 (CH)	6.42 (1H, s)	C-2, C-3, C-4a, C-9a
4a	156.12 (C)	-	-
5	112.77 (C)	-	-
6	150.79 (C)	-	-
7	141.91 (C)	-	-
8	105.14 (CH)	7.33 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
8a	115.57 (C)	-	-
9	180.64 (C=O)	-	-
9a	102.57 (C)	-	-
10a	150.26 (C)	-	-
1'	21.46 (CH ₂)	3.37 (2H, d, 6.0)	C-1, C-2, C-3, C-2', C-3'
2'	122.45 (CH)	5.29 (1H, <i>br t</i>)	C-1', C-4'
3'	132.24 (C)	-	-
4'	39.84 (CH ₂)	1.99 (2H, m)	C-3', C-5', C-6'
5'	21.71 (CH ₂)	2.03 (2H, m)	C-3', C-4', C-6'

Table 82 (Continued)

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
6'	124.27 (CH)	5.02 (1H, <i>br t</i>)	C-5', C-8', C-10'
7'	131.54 (C)	-	-
8'	25.58 (CH ₃)	1.57 (3H, s)	C-6', C-7', C-10'
9′	17.85 (CH ₃)	1.80 (3H, s)	C-2', C-3', C-4', C-5'
10'	17.62 (CH ₃)	1.52 (3H, s)	C-6', C-7', C-8'
1''	22.45 (CH ₂)	3.60 (2H, d, 6.0)	C-5, C-6, C-10a, C-2", C-3"
2''	121.38 (CH)	5.29 (1H, <i>br t</i>)	C-1", C-4", C-5"
3''	135.90 (C)	-	-
4''	25.81 (CH ₃)	1.69 (3H, s)	C-2", C-3", C-5"
5''	16.20 (CH ₃)	1.88 (3H, s)	C-2", C-3", C-4"
1-OH	-	13.19 (1H, s)	-

^{*} Carbon type was deduced from DEPT experiments.

FD15: 1,6-Dihydroxy-7-methoxy-8-(3-methyl-2-butenyl)-2',2'-dimethylchromano (5',6':2,3)xanthone (3-isomangostin)

FD15 was isolated as a yellow solid, m.p. 154-156 °C. The ¹H NMR spectrum (**Table 83**) showed four *singlet* signals of 1-OH at δ 13.72, H-5 at δ 6.83, H-4 at δ 6.23 and 7-OCH₃ at δ 3.80. The signals of *gem*-dimethyl protons and two groups of methylene protons (H-4' and H-3') associated with a chroman ring resonated at δ 1.37 (s), 2.71 (t) and 1.83 (t), respectively. The correlation of H-4' to C-1, C-2 and C-3 precisely determined that the dimethylchroman ring was next to C-1 and C-4. In addition, the signals of two methyl groups, methylene protons and an olefinic methine proton which were the characteristic signals of a prenyl moiety were present at δ 1.69 (H-4"), 1.83 (H-5"), 4.09 (H-1") and 5.27 (H-2"), respectively. The chemical shift of the methylene protons H-1'' (δ 4.09) implied that H-1'' was deshielded by a carbonyl group, accordingly the prenyl side chain was proposed to be at C-8, a peri position to the carbonyl group. The HMBC correlation of H-1" to C-7, C-8 and C-8a confirmed the location of the prenyl unit at C-8. According to the assignment and HMBC correlations (Table 83), 1,6-dihydroxy-7-methoxy-8-(3-methyl-2-butenyl)-2',2'dimethylchromano(5',6':2,3)xanthone or 3-isomangostin was proposed for FD15 (Mahabusarakam, et al., 1987).

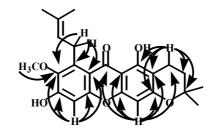


Table 83 The NMR spectral data of FD15

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
1	160.57 (C)	-	-
2	103.75 (C)	-	-
3	154.68 (C)	-	-
4	93.99 (CH)	6.23 (1H, s)	C-2, C-3, C-4a, C-9, C-9a, C-4'
4a	160.69 (C)	-	-
5	101.61 (CH)	6.83 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	155.83 (C)	-	-
7	142.37 (C)	-	-
8	136.89 (C)	-	-
8a	112.07 (C)	-	-
9	182.00 (C=O)	-	-
9a	102.83 (C)	-	-
10a	154.47 (C)	-	-
2'	76.02 (C)	-	-
3'	31.86 (CH ₂)	1.83 (2H, t, 5.8)	C-2, C-2', C-4', 2'-(CH ₃) ₂
4'	16.08 (CH ₂)	2.71 (2H, t, 5.8)	C-1, C-2, C-3, C-2', C-3'
1''	26.51 (CH ₂)	4.09 (2H, d, 5.6)	C-7, C-8, C-8a, C-2", C-3"
2''	123.23 (CH)	5.27 (1H, br t, 5.6)	C-8, C-4", C-5"

Table 83 (Continued)

Position	$\delta_{_{\! \mathrm{C}}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
3''	132.06 (C)	-	-
4''	25.82 (CH ₃)	1.69 (3H, s)	C-2", C-3", C-5"
5''	18.20 (CH ₃)	1.83 (3H, s)	C-2", C-3", C-4"
1-OH	-	13.72 (1H, s)	C-1, C-9a
7-OCH ₃	-	3.80 (3H, s)	C-7
2'-(CH ₃) ₂	26.72 (CH ₃)	1.37 (6H, s)	C-2', C-3', 2'-(CH ₃) ₂

^{*} Carbon type was deduced from DEPT experiments.

FD16: 1,6-Dihydroxy-2-(3-methyl-2-butenyl)-3-methoxy-2'',2''-dimethylchromeno(5'',6'':8,7)xanthone

FD16 is a yellow solid, m.p. 213-215 $^{\circ}$ C. Its molecular formula of $C_{24}H_{24}O_6$ were established on the basis of mass spectrum ($[M]^{+}$ m/z 408.1573). The UV spectrum showed maxima absorptions at 331.2, 322.6, 265.8, 244.8 and 208.8 nm. The IR spectrum exhibited the absorption bands of O-H stretching at 3479 cm⁻¹ and C=O stretching at 1587 cm⁻¹. The ¹H NMR spectrum of FD16 (Table 84) indicated that it was a methoxy derivative of **RD7**. The ¹H NMR spectrum showed the resonances of a chelated hydroxy group 1-OH at δ 13.35 (br s), aromatic protons H-4 at δ 6.36 (s), H-5 at δ 6.83 (s), methoxy protons 3-OCH₃ at δ 3.91 (s) and proton resonances corresponded to a prenyl group at δ 1.80 (H-5', s), 1.68 (H-4', s), 3.36 (H-1', d) and 5.23 (H-2', br t). The charecteristic signals of two methyl groups (2"-Me₃) and vicinal olefinic protons (H-3", H-4") associated with a chromene ring were shown at δ 1.50, 5.83 and 8.04, respectively. The deshielded effect on resonance of H-4" suggested that the chromene ring was attached to the xanthone nucleus nearby the carbonyl group. The enhancement of the resonance of H-4 by irradiation of methoxy protons in the NOE experiment suggested that the methoxy group was at C-3. HMBC correlation of H-1' to C-1, C-2 and C-3 indicated that the prenyl unit was at C-2. The assigned structure was further confirmed by HMBC correlations. Thus, FD16 was proposed to be 1,6-dihydroxy-2-(3-methyl-2-butenyl)-3-methoxy-2",2"-dimethylchromeno (5".6":8,7)xanthone. This compound appears to be novel.

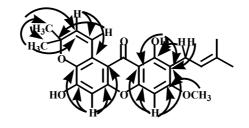


Table 84 The NMR spectral data of FD16

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
1	159.60 (C)	-	-
2	111.46 (C)	-	-
3	163.61 (C)	-	-
4	88.93 (CH)	6.36 (1H, s)	C-2, C-3, C-4a, C-9a
4a	155.52 (C)	-	-
5	102.20 (CH)	6.83 (1H, s)	C-6, C-7, C-8a, C-10a
6	150.75 (C)	-	-
7	136.77 (C)	-	-
8	119.73 (C)	-	-
8a	108.78 (C)	-	-
9	182.41 (C=O)	-	-
9a	102.20 (C)	-	-
10a	153.01 (C)	-	-
1'	21.33 (CH ₂)	3.36 (2H, d, 6.7)	C-1, C-2, C-3, C-2', C-3'
2'	122.27 (CH)	5.23 (1H, <i>br t</i> , 7.0)	C-1', C-4', C-5'
3'	131.74 (C)	-	-
4'	25.81 (CH ₃)	1.68 (3H, s)	C-2', C-3', C-5'
5'	17.78 (CH ₃)	1.80 (3H, s)	C-2', C-3', C-4'

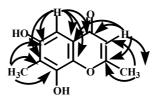
Table 84 (Continued)

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2''	71.00 (C)	-	-
3''	132.21 (CH)	5.83 (1H, d, 9.7)	C-8, C-4"
4''	121.02 (CH)	8.04 (1H, d, 9.7)	C-2"
2"-(CH ₃) ₂	27.34 (CH ₃)	1.50 (3H, s)	C-3", 2"-(CH ₃) ₂
	29.69 (CH ₃)	1.50 (3H, s)	C-3", 2"-(CH ₃) ₂
1-OH	-	13.35 (1H, <i>br s</i>)	C-1, C-2, C-9a
3-OCH ₃	55.84 (CH ₃)	3.91 (3H, s)	C-3

^{*} Carbon type was deduced from DEPT experiments.

FD17: 6,8-Dihydroxy-2,7-dimethyl-4H-chromen-4-one

FD17 was isolated as a yellow solid, m.p. 227-229 °C. Its molecular formula of $C_{11}H_{10}O_4$ were established on the basis of mass spectrum ([M]⁺ m/z 206.0570). The UV spectrum showed maximum absorption bands at 327.2, 300.5, 258.6, 252.7, 226.6, 219.6 and 207.9 nm. The IR spectra showed absorption bands of O-H stretching (3343 cm⁻¹) and C=O stretching (1657 cm⁻¹). The ¹³C NMR spectrum showed a carbonyl carbon at δ 181.58. The ¹H NMR spectrum showed a *singlet* signal of olefinic proton H-3 at δ 5.99, a *singlet* signal of aromatic proton H-5 at δ 6.33 and two *singlet* signals of methyl protons 2-CH₃ at δ 2.39 and 7-CH₃ at δ 2.13. These assignment were supported by the HMBC correlations of H-3 to C-2, C-4, C-4a, 2-CH₃; H-5 to C-4, C-4a, C-6, C-7, C-8a; 2-CH₃ to C-2, C-3 and 7-CH₃ to C-6, C-7, C-8. **FD17** then was assigned for 6,8-dihydroxy-2,7-dimethyl-4H-chromen-4-one which was a new natural occurring compound.



Major HMBC correlations of FD17

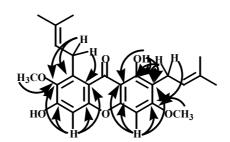
Table~85 ~~ The ~NMR ~spectral ~data ~of ~FD17

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
2	165.63 (C)	-	-
3	106.88 (CH)	5.99 (1H, s)	C-2, C-4, C-4a, 2-CH ₃
4	181.58 (C=O)	-	-
4a	102.99 (C)	-	-
5	97.57 (CH)	6.33 (1H, s)	C-4, C-4a, C-6, C-7, C-8a
6	160.98 (C)	-	-
7	101.22 (C)	-	-
8	154.60 (C)	-	-
8a	158.20 (C)	-	-
2-CH ₃	19.38 (CH ₃)	2.39 (3H, s)	C-2, C-3
7-CH ₃	6.43 (CH ₃)	2.13 (3H, s)	C-6, C-7, C-8

^{*} Carbon type was deduced from DEPT experiments.

FD18: 1,6-Dihydroxy-3,7-dimethoxy-2,8-bis(3-methyl-2-butenyl)xanthone (β-mangostin)

FD18 is a yellow solid, m.p. 179-180 °C. The ¹H NMR spectrum (**Table 86**) indicated the presence of a chelated hydroxy group (1-OH) at δ 13.44, a non chelated hydroxy group (6-OH) at δ 6.40, two isolated aromatic protons (H-4 and H-5) at δ 6.35 and 6.85, two methoxy groups (3-OCH₃ and 7-OCH₃) at δ 3.91 and 3.82, and two prenylated units. The resonances of two prenylated units were shown as followed: four methyl groups at δ 1.69 (s, H-4'), 1.80 (s, H-5'), 1.70 (s, H-4") and 1.83 (s, H-5"), two olefinic methine protons at δ 5.24 (br t, H-2") and 5.27 (br t, H-2") and two benzylic methylene protons at δ 3.36 (d, H-1") and 4.10 (d, H-1"). The HMBC correlations confirmed that **FD18** is 1,6-dihydroxy-3,7-dimethoxy-2,8-bis(3-methyl-2-butenyl)xanthone or known as β -mangostin (Yates and Bhat, 1968).



Major HMBC correlations of FD18

 $Table~86~{\rm The~NMR~spectral~data~of~FD18}$

Table 80 The twink spectral data of PD18			
Position	$\delta_{_{\! m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
1	159.76 (C)	-	-
2	111.51 (C)	-	-
3	163.52 (C)	-	-
4	88.81 (CH)	6.35 (1H, s)	C-2, C-3, C-4a, C-9, C-9a
4a	155.71 (C)	-	-
5	101.48 (CH)	6.85 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	154.44 (C)	-	-
7	142.57 (C)	-	-
8	137.03 (C)	-	-
8a	112.38 (C)	-	-
9	181.98 (C=O)	-	-
9a	101.13 (C)	-	-
10a	155.24 (C)	-	-
1'	21.35 (CH ₂)	3.36 (2H, d, 7.0)	C-1, C-2, C-3, C-2', C-3'
2'	122.31 (CH)	5.24 (1H, <i>br t</i> , 7.0)	C-1', C-4', C-5'
3'	131.72 (C)	-	-
4'	25.83 (CH ₃)	1.69 (3H, s)	C-2', C-3', C-5'
5'	17.79 (CH ₃)	1.80 (3H, s)	C-2', C-3', C-4'
1''	26.55 (CH ₂)	4.10 (2H, d, 6.5)	C-7, C-8, C-8a, C-2", C-3"
2''	123.20 (CH)	5.27 (1H, br t, 6.5)	C-1", C-4", C-5"
3''	132.10 (C)	-	-
4''	25.84 (CH ₃)	1.70 (3H, s)	C-2", C-3", C-5"
5''	18.23 (CH ₃)	1.83 (3H, s)	C-2", C-3", C-4"
1-OH	-	13.44 (1H, s)	C-1, C-2, C-9a
6-OH	-	6.40 (1H, s)	-
3-OCH ₃	55.84 (CH ₃)	3.91 (3H, s)	C-3
7-OCH ₃	56.05 (CH ₃)	3.82 (3H, s)	C-7

^{*} Carbon type was deduced from DEPT experiments.

FD19: 6-{4-[(E)-2-Carboxyethenyl]phenoxy}-3,4,5-trihydroxytetrahydro-2H-pyran-2-carboxylic acid

FD19 was isolated as a white solid, m.p. 281-283 °C. It had molecular formula $C_{15}H_{16}O_9$ ([M+K]⁺ m/z 379.0428). The IR spectrum exibited absorption bands for O-H stretching stretching (3439 cm⁻¹) and C=O stretching (1642 cm⁻¹). An AA'BB' pattern at δ 7.61 (d, H-5, H-9) and 7.06 (d, H-6, H-8) in the ¹H NMR spectrum (**Table 87**) indicated the *para*-substituted nucleus of which the substitutents were β -glucouronic acid and vinylic carbonyl groups. The resonance of anomeric proton of sugar moiety was observed as a *doublet* at δ 4.92 (H-1') with coupling constant of 7.1 Hz whereas the anomeric carbon resonated at δ 100.43 and the carboxyl carbon (C-6') was at δ 172.28. The signals of *trans* vinylic protons were present at δ 7.50 (d, H-3) and 6.39 (d, H-2) with the coupling constant of 15.9 Hz. The vinylic carboxyl group was suggested from the carbonyl resonance at δ 168.57 and HMBC correlation of H-3 to C-1, C-2, C-5 and C-9. **FD19** was then assigned to be 6-{4-[(E)-2-carboxyethenyl] phenoxy}-3,4,5-trihydroxytetrahydro-2H-pyran-2-carboxylic acid.

Major HMBC correlations of FD19

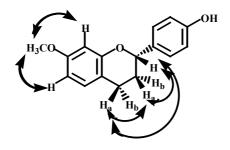
Table 87 The NMR spectral data of FD19

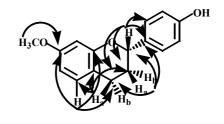
Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
1	168.57 (C=O)	-	-
2	118.38 (CH)	6.39 (1H, d, 15.9)	C-1, C-4
3	143.37 (CH)	7.50 (1H, d, 15.9)	C-1, C-2, C-5,9
4	128.53 (C)	-	-
5, 9	130.01 (CH)	7.61 (2H, d, 8.6)	C-3, C-5,9, C-6,8, C-7
6, 8	117.02 (CH)	7.06 (2H, d, 8.6)	C-4, C-6,8, C-7
7	159.33 (C)	-	-
1'	100.43 (CH)	4.92 (1H, d, 7.1)	C-7
2'	73.51 (CH)	3.50-3.10 (1H, m)	-
3'	72.39 (CH)	3.50-3.10 (1H, m)	-
4'	74.41 (CH)	3.50-3.10 (1H, m)	-
5'	76.90 (CH)	3.50-3.10 (1H, m)	-
6′	172.28 (C=O)	-	-

^{*} Carbon type was deduced from DEPT experiments.

SD1: (2S)-4'-Hydroxy-7-methoxyflavan

SD1 was isolated as a colourless solid, m.p. 140-142 °C. The optical rotation was $\left[\alpha\right]_{D}^{29} - 10.6^{\circ}$ (c = 0.018 in CH₃OH). The ¹H NMR spectrum of **SD1** indicated that it was a methoxy derivative of SD11. The resonance of a methoxy group was displayed at δ 3.73. The group was suggested to be at C-7 by the 3J coupling of the methoxy protons to C-7. The resonances of AA'BB' system for H-2', H-6' (δ 7.28, 2H, J = 9.2 Hz) and H-3', H-5' (δ 6.87, 2H, J = 9.2 Hz) and the resonances of an ABX system for H-5 (δ 6.96, d, J = 8.5 Hz), H-6 (δ 6.44, dd, J = 8.5, 2.6 Hz) and H-8 (δ 6.38, d, J = 2.6 Hz) were observed. The spectra further showed a doublet of doublet resonance of H-2 at δ 4.96 (1H, J = 13.1, 2.9 Hz), a quartet of doublet of H_a-3 at δ 2.12 (J = 12.0, 2.4 Hz) and a multiplet resonance of H_b-3 at δ 1.98, a doublet of doublet of doublet of doublet of H_a-4 at δ 2.88 (1H, J = 5.0, 1.0 Hz) and a doublet of doublet of doublet resonance of H_b -4 at δ 2.68 (J = 12.0, 5.0 and 3.0 Hz). The negative $[\alpha]_D$ value indicated that C-2 possesses an S configuration (Achenbach, et al., 1988). The enhancement of the signal of H-2 and H_a-4 upon irradiation at the resonance of H_a-3 indicated that proton H-2, H_a-3 and H_a-4 are cis. In HMBC (**Table 88**) ³J correlations of H-2 to C-4, C-2',6' and H-2',6' to C-2, C-6',2', C-4' suggested that the aromatic ring was linked to C-2. Thus SD1 was proposed to be (2S)-4'-hydroxy-7methoxyflavan (Achenbach, et al., 1988).





NOE difference experiments of SD1

Table 88 The NMR spectral data of SD1

Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
2	78.28 (CH)	4.96 (1H, dd, 13.1, 2.9)	C-3, C-4, C-1', C-2',6'
3	30.64 (CH ₂)	H _a : 2.12 (1H, qd, 12.0, 2.4)	C-2, C-4, C-4a, C-1'
		H _b : 1.98 (1H, m)	C-2, C-4, C-4a, C-1'
4	25.01 (CH ₂)	H _a : 2.88 (1H, <i>dddd</i> , 5.0, 1.0)	C-2, C-3, C-4a, C-5, C-8a
		H _b : 2.68 (1H, <i>ddd</i> , 12.0, 5.0, 3.0)	C-2, C-3, C-4a, C-5, C-8a
4a	115.56 (C)	-	-
5	130.71 (CH)	6.96 (1H, d, 8.5)	C-4, C-6, C-7, C-8a
6	107.68 (CH)	6.44 (1H, dd, 8.5, 2.6)	C-4a, C-7, C-8
7	159.98 (C)	-	-
8	102.16 (CH)	6.38 (1H, d, 2.6)	C-4a, C-6, C-7, C-8a
8a	156.89 (C)	-	-
1'	133.72 (C)	-	-
2', 6'	128.23 (CH)	7.28 (2H, d, 9.2)	C-2, C-2',6', C-3',5', C-4'
3', 5'	115.84 (CH)	6.87 (2H, d, 9.2)	C-1', C-2',6', C-3',5', C-4'
4'	157.80 (C)	-	-
7-OCH ₃	55.37 (CH ₃)	3.73 (3H, s)	C-7
4'-OH	_	8.40 (1H, <i>br s</i>)	-

^{*} Carbon type was deduced from DEPT experiments.

SD2: 4,4'-Dihydroxy-2'-methoxychalcone (isoliquiritigenin 2'-methyl ether)

HO
$$\beta$$
 α
 β
 α
 β
 α
 β
 α
 β
 α
 β
 α

SD2 was isolated as a yellow solid, m.p. 218-220 °C. The ¹H NMR spectrum (**Table 89**) showed an AA'BB' pattern, of which two *doublet* signals at δ 7.54 (J = 9.2 Hz) and 6.81 (J = 9.2 Hz) belonged to H-2, H-6 and H-3, H-5, respectively. It was thus indicated that **SD2** contained a *para*-substituted nucleus. A *doublet* at δ 6.50 (J = 2.3 Hz), a *doublet of doublet* at δ 6.44 (J = 8.1, 2.3 Hz) and a *doublet* at δ 7.50 (J = 8.1 Hz) were assigned for the resonances of H-3', H-5' and H-6', respectively. Two *doublet* signals at δ 7.35 and 7.45 with a coupling constant of 16.1 Hz corresponded to *trans* vinylic protons, H- α and H- β . A methoxy group resonated at δ 3.83 and was located at C-2' by the 3J coupling of methoxy protons to C-2'. The correlations of H- α to C=O, C- β , C-1 and H- β to C=O, C- α , C-2, C-6 in HMBC suggested the positions of H- α and H- β . The 13 C NMR spectrum showed: a carbonyl carbon, five quaternary carbons, seven methine carbons and a methyl carbons corresponded with the assigned structure. **SD2** was therefore assigned to be 4,4'-dihydroxy-2'-methoxychalcone (Achenbach, *et al.*, 1988).



Major HMBC correlations of SD2

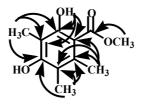
 $Table~89~{\rm The~NMR~spectral~data~of~SD2}$

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
1	126.20 (C)	-	-
2, 6	130.40 (CH)	7.54 (2H, d, 9.2)	C-2,6, C-4, C-β
3, 5	116.05 (CH)	6.81 (2H, d, 9.2)	C-1, C-3,5, C-4
4	159.79 (C)	-	-
1'	120.45 (C)	-	-
2'	162.72 (C)	-	-
3'	99.46 (CH)	6.50 (1H, d, 2.3)	C-1', C-2', C-4', C-5'
4'	160.57 (C)	-	-
5'	108.04 (CH)	6.44 (1H, dd, 8.1, 2.3)	C-1', C-3'
6'	132.37 (CH)	7.50 (1H, d, 8.1)	C-2', C-4', C=O
α	124.13 (CH)	7.35 (1H, d, 16.1)	C-1, C-β, C=O
β	141.56 (CH)	7.45 (1H, <i>d</i> , 16.1)	C-2,6, C-α, C=O
2'-OCH ₃	55.83 (CH ₃)	3.83 (3H, s)	C-2'
C=O	189.12 (C=O)	-	-

^{*} Carbon type was deduced from DEPT experiments.

SD3: 2,4-dihydroxy-3,5,6-trimethylbenzoic acid methyl ester (methyl 2,4-dihydroxy-3,5,6-trimethylbenzoate)

SD3 was isolated as a yellow solid, m.p. 78-80 °C. The ¹H NMR spectrum (**Table 90**) showed six *singlet* signals. They were assigned for a chelated hydroxy proton 2-OH (δ 11.48), a non chelated hydroxy proton 4-OH (δ 5.25), methoxy protons (δ 3.95) and methyl protons 3-CH₃, 5-CH₃ (δ 2.15, 6H, s) and 6-CH₃ (δ 2.45, 3H, s). The presence of a carbonyl group was suggested from carbon resonance at δ 172.74. HMBC correlation of methoxy protons to C=O suggested that the methoxy group was linked to carbonyl carbon. Six quaternary carbons were indicated from the ¹³C NMR spectrum. **SD3** then was assigned for 2,4-dihydroxy-3,5,6-trimethylbenzoic acid methyl ester which was known as methyl 2,4-dihydroxy-3,5,6-trimethylbenzoate (Soman, *et al.*, 1999). These assignment were supported by HMBC correlations.



Major HMBC correlations of SD3

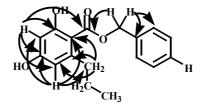
 $Table \ 90 \ \ The \ NMR \ spectral \ data \ of \ SD3$

Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
1	106.15 (C)	-	-
2	159.57 (C)	-	-
3	107.32 (C)	-	-
4	156.74 (C)	-	-
5	114.86 (C)	-	-
6	137.53 (C)	-	-
7	101.22 (C)	-	-
2-ОН	-	11.48 (1H, s)	C-1, C-2, C-3
3-CH ₃	8.04 (CH ₃)	2.15 (3H, s)	C-2, C-3, C-4
4-OH	-	5.25 (1H, s)	-
5-CH ₃	11.88 (CH ₃)	2.15 (3H, s)	C-4, C-5, C-6
6-CH ₃	18.82 (CH ₃)	2.45 (3H, s)	C-1, C-5, C-6
C=O	172.74 (C=O)	-	-
OCH ₃	51.88 (CH ₃)	3.95 (3H, s)	C=O

^{*} Carbon type was deduced from DEPT experiments.

SD4: 2,4-Dihydroxy-6-propyl benzoic acid phenylmethyl ester

SD4 was isolated as a colourless solid, m.p. 89-90 °C. The ¹H NMR spectrum (**Table 91**) showed a *singlet* signal of a chelated hydroxy proton 2-OH at δ 12.00 and two *doublet* signals of *meta* protons H-3 (δ 6.31, J = 2.7 Hz) and H-5 (δ 6.21, J = 2.7 Hz). The resonances of a propyl side chain was displayed at δ 2.69 (2H, br t, H-1'), 1.40 (2H, *sextet*, H-2') and 0.66 (3H, t, H-3'). The propyl group was suggested to be at C-6 by the correlation of H-1' to C-1, C-5, C-6. The ¹H NMR showed the signals of oxybenzyl group of which the resonances of an oxymethylene protons were at δ 5.31 and the phenyl protons were at δ 7.35 (2H, m, H-2" and H-6"), 7.39 (2H, m, H-3" and H-5") and 7.37 (1H, m, H-4"). Irradiation at oxymethylene protons (δ 5.31) in NOE experiment gave the enhancement of the signal of H-2" and H-6" (δ 7.35). The compound was then proposed to be ester derivative. These assignment were supported by HMBC correlations as shown in **Table 91**. **SD4** was then assigned for 2,4-dihydroxy-6-propyl benzoic acid phenylmethyl ester which was known as benzyl 2,4- dihydroxy-6-propyl benzoate (Elix, et al., 1998).



Major HMBC correlations of SD4

Table 91 The NMR spectral data of SD4

Position	$\delta_{\!\scriptscriptstyle m C}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
1	104.69 (C)	-	-
2	165.17 (C)	-	-
3	101.43 (CH)	6.31 (1H, d, 2.7)	C-2, C-4, C-5, C=O
4	160.83 (C)	-	-
5	111.22 (CH)	6.21 (1H, d, 2.7)	C-1, C-3, C-4, C=O, C-1'
6	148.85 (C)	-	-
1'	38.87 (CH ₂)	2.69 (2H, br t)	C-1, C-5, C-6, C-2', C-3'
2'	25.02 (CH ₂)	1.40 (2H, sextet)	C-6, C-1', C-3'
3'	13.90 (CH ₃)	0.66 (3H, <i>t</i> , 7.2)	C-1', C-2'
1''	134.95 (C)	-	-
2", 6"	128.68 (CH) ^a	$7.35 (2H, m)^{b}$	C-1", C-2",6", C-3",5", C-4"
3", 5"	129.00 (CH) ^a	$7.39 (2H, m)^{b}$	C-1", C-2",6", C-3",5", C-4"
4''	128.69 (CH) ^a	7.37 (1H, <i>m</i>)	C-2",6", C-3",5"
2-ОН	-	12.00 (1H, s)	-
C=O	171.44 (C=O)	-	-
OCH ₂	67.52 (CH ₂)	5.31 (2H, s)	C-1", C-2", C=O

^{*} Carbon type was deduced from DEPT experiments.

^{a, b} Assignment maybe interchangeable.

SD5: 1,7-Dihydroxyxanthone (euxanthone)

SD5 was isolated as a yellow solid, m.p. 228-230 °C. The ¹H NMR (**Table 92**) revealed the presence of a sharp *singlet* signal of a chelated hydroxy proton (1-OH) at δ 12.62. Two *doublet of doublet* signals at δ 6.79 (J = 8.0, 0.5 Hz), 6.93 (J = 8.0, 0.5 Hz) and a *triplet* signal at δ 7.59 (J = 8.0 Hz) were assigned as an ABM pattern of H-2, H-4 and H-3, respectively. A *doublet* signal at δ 7.42 (J = 9.0 Hz), a *doublet* signal at δ 7.62 (J = 3.0 Hz) and a *doublet of doublet* signal at δ 7.32 (J = 9.0 and 3.0 Hz) revealed the presence of an ABX pattern which was assigned for H-5, H-8 and H-6. **SD5** was then identified to be 1,7-dihydroxyxanthone. Its ¹H NMR spectral data and melting point were in agreement with euxanthone (Fujita, *et al.*, 1992).

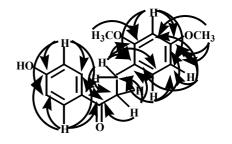
Table 92 The ¹H NMR spectral data of **SD5**

D = = 141 =	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)			
Position	SD5	euxanthone		
2	6.79 (1H, dd, 8.0, 0.5)	6.77 (1H, dd, 8.4, 0.5)		
3	7.59 (1H, <i>t</i> , 8.0)	7.59 (1H, <i>t</i> , 8.8)		
4	6.93 (1H, dd, 8.0, 0.5)	6.95 (1H, dd, 8.4, 0.5)		
5	7.42 (1H, d, 9.0)	7.41 (1H, <i>d</i> , 9.3)		
6	7.32 (1H, dd, 9.0, 3.0)	7.33 (1H, <i>dd</i> , 9.3, 2.9)		
8	7.62 (1H, d, 3.0)	7.63 (1H, d, 2.9)		
1-OH	12.62 (1H, s)	12.62 (1H, s)		
7-OH	-	5.01 (1H, s)		

SD6: 4'-Hydroxy-2,4-dimethoxydihydrochalcone (loureirin A)

HO 4 A
$$1$$
 B 1 OCH₃

SD6 was isolated as a colourless solid, m.p. 128-130 °C. The ¹H NMR spectrum (**Table 93**) demonstrated the resonances of an ABX pattern of H-3 (δ 6.44, d, J = 2.4 Hz), H-5 (δ 6.41, dd, J = 7.8 and 2.4 Hz) and H-6 (δ 7.07, d, J = 7.8 Hz), and an AA'BB' pattern of H-2', H-6' (δ 7.92, 2H, d, J = 8.4 Hz) and H-3', H-5' (δ 6.90, 2H, d, J = 8.4 Hz). The presence of two methoxy groups were shown by the resonance at δ 3.79 (6H) and it was indicated at C-2 and C-4 by the ³J coupling of methoxy groups to C-2 and C-4. The enhancement of H-3 by irradiation at the resonance of the methoxy protons suggested that H-3 was nearby the methoxy groups. The spectrum further showed resonances of H- α at δ 3.18 (2H, t, J = 7.2 Hz) and H- β at δ 2.87 (2H, t, J = 7.2 Hz). The HMBC correlations of H- α to C=O, H- β , C-1 and H- β to C=O, H- α , C-1, C-2, C-6 suggested an aromatic ring B was attached to C- β . The ¹³C NMR spectrum and HMBC were in agreement with assigned structure. Therefore, **SD6** was characterized as 4'-hydroxy-2,4-dimethoxydihydrochalcone or known as loureirin A (Meksuriyen and Cordell, 1988).



Major NOE difference experiments of SD6

Major HMBC correlations of SD6

Table 93 The NMR spectral data of SD6

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	121.64 (C)	-	-
2	158.25 (C)	-	-
3	98.49 (CH)	6.44 (1H, d, 2.4)	C-1, C-2, C-4, C-5
4	159.36 (C)	-	-
5	103.89 (CH)	6.41 (1H, dd, 7.8, 2.4)	C-1, C-3, C-4
6	130.21 (CH)	7.07 (1H, d, 7.8)	C-2, C-4, C-5, C-β
1'	129.19 (C)	-	-
2', 6'	130.93 (CH)	7.92 (2H, d, 8.4)	C-2',6', C-3',5', C-4', C=O
3', 5'	115.47 (CH)	6.90 (2H, d, 8.4)	C-1', C-2',6', C-3',5', C-4'
4'	161.21 (C)	-	-
α	38.88 (CH ₂)	3.18 (2H, t, 7.2)	C-1, C- <i>β</i> , C=O
β	25.62 (CH ₂)	2.87 (2H, t, 7.2)	C-1, C-2, C-6, C-α, C=O
2-OCH ₃	55.33 (CH ₃) ^a	3.79 (3H, s)	C-2
4-OCH ₃	55.15 (CH ₃) ^a	3.79 (3H, s)	C-4
C=O	200.72 (C=O)	-	-

^{*} Carbon type was deduced from DEPT experiments.

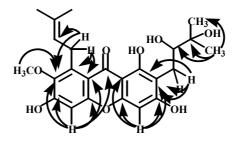
^a Assignment maybe interchangeable.

SD7: 1,3,6-Trihydroxy-2-(2,3-dihydroxy-3-methylbutyl)-7-methoxy-8-(3-methyl-2-butenyl)xanthone

SD7 was isolated as a yellow solid with m.p. 188-190 °C. A xanthone nucleus was elucidated from the UV spectra (λ_{max} 318.4, 244.9 and 213.2 nm). The IR spectra showed absorption bands of hydroxy (3397 cm⁻¹) and conjugated carbonyl (1612 cm⁻¹) functionalities. It had molecular formula $C_{24}H_{28}O_8$ ([M]⁺ m/z 444.1788).

The ¹H NMR spectrum (**Table 94**) showed a *singlet* signal of a deshielded hydroxy proton 1-OH (δ 13.82), a *singlet* resonance of methoxy protons 7-OCH₃ (δ 3.79), two *singlet* signals of two isolated aromatic protons H-4 and H-5 (δ 6.26 and 6.74). The typical resonances of a prenyl side chain appeared as follow: two *singlets* of two methyl groups at δ 1.84 (H-5'') and 1.69 (H-4''), a *doublet* at δ 4.00 (H-1'') and a *broad triplet* at δ 5.26 (H-2''). This side chain was assigned at C-8 according to the low field chemical shift of H-1'' (δ 4.00) and confirmed by HMBC correlation of H-1'' to C-7, C-8, C-8a, C-2'' and C-3''. A *doublet of doublet* of methylene protons at δ 3.14 (2H, J = 12.3 and 9.3 Hz, H-1'), a *triplet* of hydroxy methine proton at δ 4.77 (1H, J = 8.4 Hz, H-2') and two sharp *singlets* of two methyl groups at δ 1.83 (H-5') and 1.32 (H-4') corresponded to the resonances of a 2,3-dihydroxy-3-methylbutyl side chain. It was assigned at C-2 according to the HMBC correlation of H-1' to C-1, C-2, C-3, C-2' and C-3'. The resonance of methoxy protons (OCH₃) showed a cross-peak with a quaternary aromatic carbon C-7 signal, this indicated the position of OCH₃ at C-7. The ¹³C NMR spectrum and DEPT experiments corresponded to the assigned

structure. Therefore, **SD7** was assigned as 1,3,6-trihydroxy-2-(2,3-dihydroxy-3-methylbutyl)-7-methoxy-8-(3-methyl-2-butenyl)xanthone. It was a new naturally occurring xanthone.



Major HMBC correlations of SD7

Table 94 The NMR spectral data of SD7

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	HMBC
1	157.19 (C)	-	-
2	107.51 (C)	-	-
3	166.33 (C)	-	-
4	88.30 (CH)	6.26 (1H, s)	C-2, C-3, C-4a, C-9a
4a	157.19 (C)	-	-
5	101.97 (CH)	6.74 (1H, s)	C-6, C-7, C-8a, C-9, C-10a
6	155.67 (C)	-	-
7	143.40 (C)	-	-
8	137.36 (C)	-	-
8a	111.51 (C)	-	-
9	182.24 (C=O)	-	-
9a	104.08 (C)	-	-
10a	156.01 (C)	-	-
1'	26.86 (CH ₂)	3.14 (2H, dd, 12.3, 9.3)	C-1, C-2, C-3, C-2', C-3'
2'	91.77 (CH)	4.77 (1H, t, 8.4)	C-1'
3'	71.88 (C)	-	-
4'	25.37 (CH ₃)	1.32 (3H, s)	C-2', C-3', C-5'
5'	24.07 (CH ₃)	1.83 (3H, s)	C-2', C-3', C-4'
1''	26.41 (CH ₂)	4.00 (2H, d, 6.0)	C-7, C-8, C-8a, C-2", C-3"
2''	123.45 (CH)	5.26 (1H, br t, 5.1)	C-1", C-4", C-5"
3''	131.91 (C)	-	-
4''	25.87 (CH ₃)	1.69 (3H, s)	C-2", C-3", C-5"
5''	18.20 (CH ₃)	1.84 (3H, s)	C-2", C-3", C-4"
1-OH	-	13.82 (1H, s)	C-1, C-2, C-9a
7-OCH ₃	61.25 (CH ₃)	3.79 (3H, s)	C-7

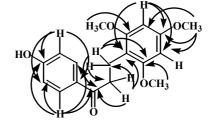
^{*} Carbon type was deduced from DEPT experiments.

SD8: 4'-Hydroxy-2,4,6-trimethoxydihydrochalcone (loureirin B)

$$H_3CO$$
 4
 OCH_3
 OCH_3

SD8 was isolated as a colourless solid, m.p. 132-133 °C. The ¹H NMR spectrum (**Table 95**) showed the resonances of an AA'BB' system of H-2', H-6' (δ 7.93) and H-3', H-5' (δ 6.93) and the resonances of H- α (δ 2.95, 2H) and H- β (δ 2.87, 2H) as in **SD6**. The *singlet* resonance of three methoxy groups (δ 3.80) and two *doublet* resonances of *meta*-protons (δ 6.24) implied the presence of a tetrasubstituted ring B. This assignment was confirmed by irradiation at the resonances of methoxy groups, it gave positive enhancement of the signals of H-3 and H-5. The ¹³C NMR spectrum (**Table 95**) showed 14 signals for eighteen carbons. Thus **SD8** was elucidated as 4'-hydroxy-2,4,6-trimethoxydihydrochalcone. It was known as loureirin B (Meksuriyen and Cordell, 1988).

$$HO$$
 H
 OCH_3
 OCH_3



Major NOE difference experiments of SD8

Major HMBC correlations of SD8

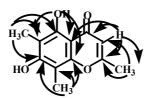
 $Table \ 95 \ \ The \ NMR \ spectral \ data \ of \ SD8$

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	110.08 (C)	-	-
2, 6	159.61 (C)	-	-
3, 5	91.38 (CH)	6.24 (2H, s)	C-1, C-4, C-3,5
4	160.76 (C)	-	-
1'	130.25 (C)	-	-
2', 6'	131.30 (CH)	7.93 (2H, d, 7.8)	C-2',6', C-3',5', C-4', C=O
3', 5'	115.87 (CH)	6.93 (2H, d, 7.8)	C-1', C-2',6', C-3',5', C-4'
4'	162.38 (C)	-	-
α	38.89 (CH ₂)	2.95 (2H, dt, 9.6, 4.0)	C-1, C- <i>β</i> , C=O
β	19.28 (CH ₂)	2.87 (2H, dt, 9.6, 4.0)	C-1, C-2, C-6, C-α, C=O
2-OCH ₃	55.93 (CH ₃)	3.80 (3H, s)	C-2, C-3
4-OCH ₃	55.51 (CH ₃)	3.80 (3H, s)	C-3, C-4, C-5
6-OCH ₃	55.93 (CH ₃)	3.80 (3H, s)	C-5, C-6
C=O	198.71 (C=O)	-	-

^{*} Carbon type was deduced from DEPT experiments.

SD9: 5,7-Dihydroxy-2,6,8-trimethylchromone

SD9 was isolated as a light brown solid, m.p. 276-277 °C. The ¹H NMR spectrum (**Table 96**) showed the signals of a chelated hydroxy proton 5-OH at δ 13.05 (s), an olefinic proton H-3 at δ 6.20 (s) and methyl protons 2-CH₃ (δ 2.43, s), 6-CH₃ (δ 2.08, s) and 8-CH₃ (δ 2.19, s). The ¹³C NMR spectrum and DEPT experiments showed the resonances of a carbonyl carbon, seven quaternary carbons, a methine carbon and three methyl carbons which agreed with the assigned structure. The enhancement of H-3 by irradiation at the resonance of the methyl protons 2-CH₃ confirmed the positions of H-3. HMBC correlations also confirmed the assignment. **SD9** then was assigned for 5,7-dihydroxy-2,6,8-trimethylchromone which was known as 8-methyleugenitol (Ito and Lai, 1978).



Major HMBC correlations of SD9

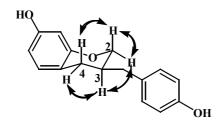
 $Table \ 96 \ \ The \ NMR \ spectral \ data \ of \ SD9$

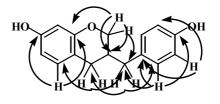
Position	$\delta_{_{\rm C}}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	НМВС
2	167.32 (C)	-	-
3	107.39 (CH)	6.20 (1H, s)	C-2, C-4, C-4a, 2-CH ₃
4	182.08 (C=O)	-	-
4a	103.11 (C)	-	-
5	155.92 (C)	-	-
6	106.75 (C)	-	-
7	159.66 (C)	-	-
8	101.53 (C)	-	-
8a	152.89 (C)	-	-
2-CH ₃	19.85 (CH ₃)	2.43 (3H, s)	C-2, C-3
5-OH	-	13.05 (1H, s)	C-4a, C-5, C-6
6-CH ₃	7.99 (CH ₃)	2.08 (3H, s)	C-5, C-6, C-7
8-CH ₃	7.87 (CH ₃)	2.19 (3H, s)	C-7, C-8, C-8a

^{*} Carbon type was deduced from DEPT experiments.

SD10: 7-Hydroxy-3-(4-hydroxybenzyl)chroman

SD10 was isolated as a pink solid, m.p. 150-152 °C. The optical rotation was $[\alpha]_D^{26}$ + 63.0 ° (c = 0.018 in CH₃OH). The ¹H NMR spectrum (**Table 97**) exhibited the resonances of seven aromatic protons, one methine proton, three methylene protons and two hydroxy groups. The AA'BB' resonances at δ 7.07 (2H, J = 7.2 Hz) and 6.79 (2H, J = 7.2 Hz) corresponded to protons H-2', H-6' and H-3', H-5', respectively. The ABX resonances at δ 6.24 (d, J = 2.4 Hz), 6.33 (dd, J = 8.4 and 2.4 Hz) and 6.80 (d, J= 8.4 Hz) belonged to H-8, H-6 and H-5, respectively. A doublet of doublet of doublet resonance of H_{eq} -2 (δ 4.10, 1H, J = 9.6, 2.4 and 1.3 Hz), a doublet of doublet resonance of H_{ax} -2 (δ 3.72, 1H, J = 10.8 and 8.4 Hz), a *multiplet* resonance of H_{ax} -3 (δ 2.16), two doublet of doublet signals of H_{eq} -4 and H_{ax} -4 (δ 2.65, 1H, J = 15.4 and 5.5 Hz and δ 2.39, 1H, J = 15.4 and 8.8 Hz) and two doublet of doublet signals of H-9 (δ 2.59 and 2.59, 1H each, J = 13.2 and 7.7 Hz) were observed. The remaining signals are the resonances of two hydroxy groups (δ 8.16 and 8.12). The NOE experiment: irradiation at the resonances of H_{ax} -2 (δ 3.72) and H_{ax} -3 (δ 2.16) resulted in the enhancement of the resonance of H_{eq} -2 (δ 4.10). The 13 C NMR spectrum (**Table 97**) showed the signals of five quaternary carbons, six methine carbons and three methylene carbons. Consequently, SD10 was proposed to be 7-hydroxy-3-(4-hydroxybenzyl)chroman. The assignment was in agreement with the previous data of 7-hydroxy-3-(4hydroxybenzyl)chroman (Masaoud, et al., 1995; Meksuriyen, et al., 1987).





Major NOE difference experiments of SD10

Major HMBC correlations of SD10

Table 97 The NMR spectral data of SD10

Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{\!\scriptscriptstyle m H}$ (multiplicity, $J_{\scriptscriptstyle m Hz}$)	¹ H- ¹ H COSY	НМВС
2	70.41 (CH ₂)	H _{ax} : 3.72	H _{eq} -2, H-3	C-3, C-4, C-8a, C-9
		(1H, dd, 10.8, 8.4)		
		H _{eq} : 4.10	H _{ax} -2, H-3	C-3, C-4, C-8a, C-9
		(1H, ddd, 9.6, 2.4, 1.3)		
3	35.35 (CH)	H _{ax} : 2.16 (1H, <i>m</i>)	H-2, H-4,	C-2, C-4, C-9
			H-9	
4	30.95 (CH ₂)	H _{ax} : 2.39	H-3, H _{eq} -4	C-2, C-3, C-4a, C-5,
		(1H, dd, 15.4, 8.8)		C-8a, C-9
		H _{eq} : 2.65	H-3, H _{ax} -4	C-2, C-3, C-4a, C-5,
		(1H, dd, 15.4, 5.5)		C-8a, C-9
4a	113.27 (C)	-	-	-
5	131.14 (CH)	6.80 (1H, d, 8.4)	Н-6	C-4, C-7, C-8a
6	108.76 (CH)	6.33 (1H, dd, 8.4, 2.4)	H-5, H-8	C-4a, C-7, C-8
7	156.18 (C)	-	-	-
8	103.54 (CH)	6.24 (1H, d, 2.4)	Н-6	C-4a, C-6, C-7, C-8a
8a	157.40 (C)	-	-	-

Table 97 (Continued)

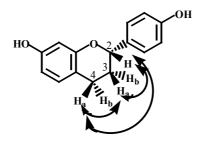
Position	$\delta_{\rm C}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	¹ H- ¹ H COSY	НМВС
9	37.62 (CH ₂)	2.59 (1H, dd, 13.2, 7.7)	H-3, H-9	C-2, C-3, C-4, C-1',
				C-2',6'
		2.52 (1H, dd, 13.2, 7.7)	H-3, H-9	C-2, C-3, C-4, C-1',
				C-2',6'
1'	130.72 (C)	-	-	-
2', 6'	131.08 (CH)	7.07 (2H, d, 7.2)	H-3', H-5'	C-9, C-1', C-3',5', C-4'
3', 5'	115.95 (CH)	6.79 (2H, d, 7.2)	H-2', H-6'	C-1', C-3',5', C-4'
4'	156.58 (C)	-	-	-
7-OH	-	8.16 (1H, <i>br s</i>) ^a	-	-
4'-OH	-	8.12 (1H, <i>br s</i>) ^a	-	-

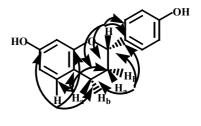
^{*} Carbon type was deduced from DEPT experiments.

^a Assignment maybe interchangeable.

SD11: (2S)-7,4'-Dihydroxyflavan

SD11 was isolated as brown needles. The optical rotation was $\left[\alpha\right]_{D}^{29}$ -129.0 $^{\circ}$ (c = 0.023 in CH₃OH). The ¹H NMR spectrum (**Table 98**) showed the resonances of an AA'BB' system for protons H-2', H-6' and H-3', H-5' at δ 7.27 (2H, J = 8.0 Hz) and 6.86 (2H, J = 8.0 Hz) and the resonances of an ABX system for protons H-5, H-6 and H-8 at δ 6.88 (d, J = 8.0 Hz), 6.38 (dd, J = 8.0 and 2.4 Hz) and 6.32 (d, J = 2.4 Hz). The spectra further showed a doublet of doublet resonance of H-2 at δ 4.93 (1H, J = 10.5 and 1.8 Hz), a quartet of doublet of H_a-3 at δ 2.09 (J = 12.6 and 2.4 Hz) and a multiplet resonance of H_b -3 at δ 1.96, a multiplet of H_a -4 at δ 2.85 and a doublet of doublet of doublet resonance of H_b -4 at δ 2.64 (J = 15.6, 4.8 and 3.6 Hz). The negative $[\alpha]_D$ value indicated that C-2 possesses an S configuration (Achenbach, et al., 1988). The enhancement of the signals of H-2 and H_a-4 upon irradiation at the resonance of H_a -3 indicated that protons H-2, H_a -3 and H_a -4 are cis. In HMBC (**Table 98**), 3J correlations of H-2 to C-4, C-2',6' and H-2',6' to C-2, C-6',2', C-4' suggested that the aromatic ring was linked to C-2. The ¹³C NMR spectrum (Table 98) showed 13 signals for fifteen carbons. Consequently, SD11 was proposed to be (2S)-7,4'-dihydroxyflavan (Achenbach, et al., 1988).





NOE difference experiments of SD11

Major HMBC correlations of SD11

Table 98 The NMR spectral data of SD11

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	НМВС
2	78.12 (CH)	4.93 (1H, dd, 10.5, 1.8)	C-3, C-4, C-1', C-2',6'
3	30.70 (CH ₂)	H _a : 2.09 (1H, qd, 12.6, 2.4)	C-4, C-4a, C-1'
		H _b : 1.96 (1H, <i>m</i>)	C-4, C-4a, C-1'
4	24.98 (CH ₂)	H _a : 2.85 (1H, <i>m</i>)	C-2, C-3, C-4a, C-5, C-8a
		H _b : 2.64 (1H, <i>ddd</i> , 15.6, 4.8, 3.6)	C-2, C-3, C-4a, C-5, C-8a
4a	113.63 (C)	-	-
5	130.70 (CH)	6.88 (1H, d, 8.0)	C-4, C-7, C-8a
6	108.69 (CH)	6.38 (1H, dd, 8.0, 2.4)	C-4a, C-7, C-8
7	157.67 (C)	-	-
8	103.80 (CH)	6.32 (1H, d, 2.4)	C-4a, C-6, C-7, C-8a
8a	156.78 (C)	-	-
1'	133.77 (C)	-	-
2', 6'	128.18 (CH)	7.27 (2H, d, 8.0)	C-2, C-2',6', C-3',5', C-4'
3', 5'	115.80 (CH)	6.86 (2H, d, 8.0)	C-1', C-3',5', C-4'
4'	157.29 (C)	-	-

^{*} Carbon type was deduced from DEPT experiments.

SD12: 4,4'-Dihydroxy-2-methoxydihydrochalcone (loureirin C)

SD12 was isolated as a pink solid, m.p. 178-179 °C. The ¹H NMR spectrum (**Table 99**) showed the resonances of an ABX pattern of H-3 (δ 6.46, d), H-5 (δ 6.35, dd) and H-6 (δ 6.99, d), an AA'BB' pattern of H-2', H-6' (δ 7.92, 2H, d) and H-3', H-5' (δ 6.92, 2H, d) and the resonances of H- α (δ 3.11, 2H) and H- β (δ 2.85, 2H) as in **SD8**. Only one methoxy group was shown by the resonance at δ 3.80 and it was located at C-2 by the ³J correlation of methoxy group to C-2. The enhancement of H-3 by irradiation at the resonance of the methoxy protons confirmed the position of OCH₃ at C-2. ¹³C NMR spectrum (**Table 99**) and HMBC confirmed the assigned structure. Therefore, **SD12** was characterized as 4,4'-dihydroxy-2-methoxydihydrochalcone which was known as loureirin C (Meksuriyen and Cordell, 1988).

$$H_3$$
CO H_3 CO H_4 OH H_3 CO H_4 OH

Major NOE difference experiments of SD12 Major HMBC correlations of SD12

Table 99 The NMR spectral data of SD12

Position	$\delta_{_{ m C}}$ (C-Type)*	$\delta_{_{ m H}}$ (multiplicity, $J_{_{ m Hz}}$)	HMBC
1	121.07 (C)	-	-
2	159.28 (C)	-	-
3	99.60 (CH)	6.46 (1H, d, 3.0)	C-1, C-2, C-4, C-5
4	158.00 (C)	-	-
5	107.35 (CH)	6.35 (1H, dd, 8.0, 3.0)	C-1, C-3, C-4
6	130.98 (CH)	6.99 (1H, d, 8.0)	C-2, C-4, C-β
1'	130.10 (C)	-	-
2', 6'	131.24 (CH)	7.92 (2H, d, 9.6)	C-2',6', C-3',5', C-4', C=O
3', 5'	115.90 (CH)	6.92 (2H, d, 9.6)	C-1', C-3',5', C-4'
4'	162.44 (C)	-	-
α	39.31 (CH ₂)	3.11 (2H, <i>t</i> , 9.0)	C-1, C- <i>β</i> , C=O
β	25.80 (CH ₂)	2.85 (2H, t, 9.0)	C-1, C-2, C-6, C-α, C=O
2-OCH ₃	55.46 (CH ₃)	3.80 (3H, s)	C-2
C=O	198.71 (C=O)	-	-

^{*} Carbon type was deduced from DEPT experiments.

3.2 Relationship of compounds from G. dulcis

Investigation of *G. dulcis* has revealed that polyisoprenyl benzophenones, biflavonoids and xanthones are three main components in this plant. The flavans, isoflavans, flavones, isoflavones, flavonols, chromones, chalcones, glycerol, triterpene and dimethoxy citrate were also present. The presence of both xanthones and benzophenones *in G. dulcis* is in agreement with the biosynthesis path way that xanthones were formed on an arylpolyketide pathway involving a benzophenone key intermediate (Bennett and Lee, 1989). The relationships among xanthone derivatives can be discussed.

GD14 is a precursor of GD13, GD23, GD24, GD25, RD3, RD7, RD9, FD15 and SD7. GD13, RD3 and FD15 were obtained from cyclization of prenyl side chains. RD7 and RD9 were resulted from cyclization and reduction of the prenyl unit. GD23, GD24, GD25 and SD7 were obtained from oxidation of the prenyl side chain. The oxidation of the prenyl side chain to a 2-hydroxy-3-methyl-3-butenyl side chain was reported by Tahara and co-workers (1989).

FD9, FD16 and RD1 were obtained from cyclization and reduction of the prenyl side chain of RD2, RD17 and GD6, respectively. GD17 was the product from cyclization and reduction of the geranyl side chain of GD6.

$$\begin{array}{c} \text{cyclization \&} \\ \text{H}_3\text{CO} \\ \text{H}_0 \\ \text{GD6} \\ \end{array}$$

Benzophenones which were found in *G. dulcis* were **GD1**, **GD22**, **RD14**, **FD11** and **FD12**. Camboginol (**GD1**) is a major component in the green fruits but not in the ripe fruits. Whereas cambogin (**RD14**) which was a cyclization product of **GD1** was isolated from the ripe fruits. It could be assumed that **GD1** was produced in the green fruits then become **RD14** in the ripe fruits. **GD1**, **RD14**, **FD11** and **FD12** might be derived biogenetically from maclurin by the attachment of three or more isoprenyl side chains and successive cyclodehydrogenation and/or cyclization. **GD1** and **RD14** were obtained from reaction of intermediate **I** and **II**, respectively (Ramo Rao, *et al.*, 1980; Karanjgoakar, *et al.*, 1973; Locksley, *et al.*, 1967). Reaction of three dimethylally pyrophosphate (DMAPP) units on the phloroglucinol ring of maclurin can give rise to an intermadiate **I** which can further react with two additional DMAPP to give camboginol (**GD1**) (Rama Rao, *et al.*, 1980). Alternatively, reaction of intermediate **II** with one DMAPP and a isopentenylpyrophosphate can lead to xanthochymol (**FD11**) (Rama Rao, *et al.*, 1980).

GD20, FD1, FD6, FD7, FD8 and FD10 are biflavonoids. FD6 is C-3',C-8" interflavonoid linkage whereas the others are C-3,C-8" interflavonoid linkage. GD20, FD1 and FD8 are two related flavanone-flavone dimers. FD7 and FD10 contained two flavanone units whereas FD6 contained two flavone units. FD7 and FD8 are biflavonoid glycosides of FD10 and GD20, respectively. Apigenin (RD13) was a precursor of these biflavonoids and flavanonols, RD6 and RD15.

Among seventy-four isolated compounds, only five of them were previously reported to be isolated from this plant. Morelloflavone (GD20), volkensiflavone (FD1)

and GB-2a (FD10) were previously isolated from the leaves whereas podocarpusflavone A (FD6) and euxanthone (SD5) were obtained from the branches and the stem bark, respectively.

3.3 Evaluation of antioxidation activity

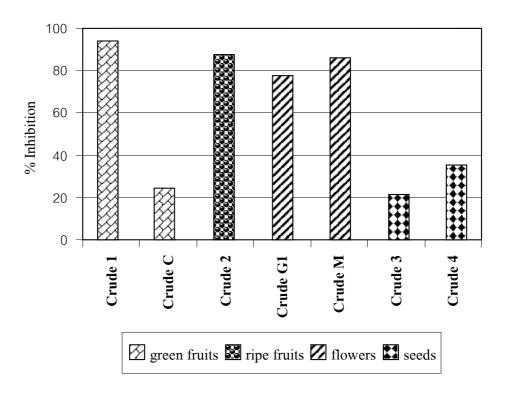
Phenolic compounds were known to be the antioxidant with an excellent hydrogen or electron donors (Shahidi, *et al.*, 1992). The isolated compounds from *G. dulcis* were therefore considerable interest in the studies of antioxidative activity.

Estimation of antioxidative effects has been carried out by various methods. The DPPH assay is one of the methods used for antioxidant testing (Tamura, *et al.*, 1990). DPPH is a stable free radical which shows a purple colour and a strong absorption at 517 nm. When DPPH radical accepts an electron or hydrogen radical, a more stable compound will be formed and consequently its characteristic absorption at 517 nm vanishes. The capacity of the substances to donate electrons can be estimated from the degree of loss of color (Blois, 1958). Coexistence of an antioxidant compound (AH) and free radical (DPPH) leads to the disappearance of DPPH free radical and the appearance of the free radical (A).

AH +
$$O_2N$$
 NO_2 O_2N NO_2 O_2N NO_2

3.3.1 Free radical scavenging activity of the crude extracts

To determine the scavenging activity, the crude extracts of *G. dulcis* were tested at the final concentration of 100 µg/mL. The activity was monitored by following the decrease of the absorbance of the solution at 517 nm for 30 min. The results were expressed as % inhibition (**Figure 26**). The activity was exhibited by the acetone extracts from the fruits (**Crude 1** and **Crude 2**) and flowers (**Crude G1**) as well as by the methanolic extract from the flowers (**Crude M**).



Crude 1, 2, G1 = acetone extract

Crude C, M, 4 = methanolic extract

Crude 3 = dichloromethane extract

Figure 26 Radical scavenging activity of the crude extracts

3.3.2 Free radical scavenging activity of the pure compounds

To determine the active constituents in the extracts and the pure consituents were examined. The samples were tested at the final concentration of 10 μM. The absorption of the solution were measured at 517 nm at 30 min. Ascorbic acid and BHT were used as reference compounds. The activity was expressed in the % inhibition (Table 22). The results showed that GD1, GD20, RD5, RD14, RD16, FD5, FD8, FD11 and FD12 exhibited higher activity than that of BHT (Table 100 and Figure 27). The other compounds showed moderate to weak activity.

Table 100 % inhibition of tested compounds and standard antioxidant (10 μM)

Sample	% Inhibition \pm S.D.
	(10 µM, 30 min)
Control	-
RD5	86.89 ± 0.03
RD16	81.97 ± 0.06
Ascorbic acid	77.05 ± 0.01
GD1	73.77 ± 0.03
RD14	68.85 ± 0.04
FD11	60.66 ± 0.05
FD12	59.02 ± 0.04
FD5	57.38 ± 0.05
FD8	55.74 ± 0.03
GD20	50.81 ± 0.03
ВНТ	42.62 ± 0.02
FD7	36.07 ± 0.04
FD10	32.79 ± 0.01

Sample	% Inhibition ± S.D.
	(10 µM, 30 min)
GD10	24.59 ± 0.02
GD9	22.13 ± 0.01
GD11	19.67 ± 0.02
GD12	19.67 ± 0.01
GD3	18.03 ± 0.03
GD4	18.03 ± 0.01
GD14	18.03 ± 0.01
RD8	18.03 ± 0.02
RD13	18.03 ± 0.02
RD17	18.03 ± 0.02
GD19	17.21 ± 0.02
GD7	16.39 ± 0.01
FD9	16.39 ± 0.01
GD15	15.57 ± 0.03

Table 100 (Continued)

	T
Sample	% Inhibition \pm S.D.
	(10 μM, 30 min)
GD5	14.75 ± 0.01
GD6	14.75 ± 0.02
GD8	14.75 ± 0.03
GD13	14.75 ± 0.01
GD16	14.75 ± 0.02
RD6	14.75 ± 0.05
RD7	14.75 ± 0.04
FD14	14.75 ± 0.03
GD17	13.11 ± 0.03
GD22	11.48 ± 0.04
RD12	11.48 ± 0.01
RD15	11.48 ± 0.02
RD3	9.84 ± 0.01
SD10	9.83 ± 0.03
GD23	8.20 ± 0.03
RD10	8.20 ± 0.03
FD4	8.20 ± 0.02
FD13	8.20 ± 0.01
SD1	8.20 ± 0.03
SD5	6.56 ± 0.01
SD7	6.56 ± 0.02

Sample	% Inhibition \pm S.D.
	(10 µM, 30 min)
FD1	4.92 ± 0.01
FD6	4.92 ± 0.05
SD11	4.92 ± 0.01
GD25	3.28 ± 0.05
FD15	3.28 ± 0.01
FD17	3.28 ± 0.01
SD2	3.28 ± 0.01
SD12	3.28 ± 0.01
GD21	1.64 ± 0.02
RD1	1.64 ± 0.02
RD9	1.64 ± 0.02
RD11	1.64 ± 0.02
FD3	1.64 ± 0.01
FD16	1.64 ± 0.01
FD18	1.64 ± 0.01
FD19	1.64 ± 0.01
SD3	1.64 ± 0.01
SD4	1.64 ± 0.01
SD6	1.64 ± 0.01
SD8	1.64 ± 0.01
SD9	1.64 ± 0.01

control = 0.05 mM DPPH

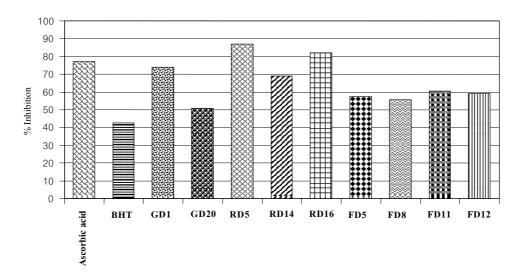


Figure 27 Radical scavenging activity of pure compounds

3.3.3 Evaluation of 50 % inhibitory concentration (IC₅₀)

The assessment of the activity was extended for GD1, GD20, RD5, RD14, RD16, FD5, FD8, FD11 and FD12. The oxidation effect was evaluated as the concentration required to scavenge 50% DPPH free radical (IC₅₀). Their IC₅₀ were exhibited at 6.90, 13.00, 5.90, 7.20, 6.10, 10.50, 11.40, 8.50 and 10.00 μ M, respectively whereas IC₅₀ of ascorbic acid and BHT were shown at 6.50 and 19.00 μ M (Table 101 and Figure 28). Moreover the results indicated that RD5 and RD16 acted as a radical scavenger more effective than ascorbic acid.

Table 101 IC_{50} values of tested compounds and standard antioxidant

Sample	$IC_{50} \pm S.D. \ (\mu M, 30 \ min)$
RD5	5.90 ± 0.02
RD16	6.10 ± 0.01
Ascorbic acid	6.50 ± 0.01
GD1	6.90 ± 0.02
RD14	7.20 ± 0.01
FD11	8.50 ± 0.01
FD12	10.00 ± 0.02
FD5	10.50 ± 0.01
FD8	11.40 ± 0.01
GD20	13.00 ± 0.02
ВНТ	19.00 ± 0.01

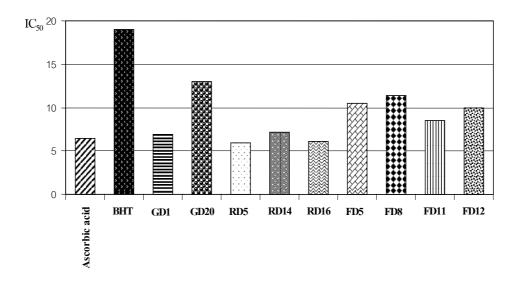


Figure 28 IC_{50} of pure compounds

The mechanism of trapping radical by **GD1**, **GD20**, **RD5**, **RD14**, **RD16**, **FD5**, **FD8**, **FD12** and **FD13** exhibited by a phenolic characteristic which was proposed to donate hydrogen radical to DPPH. The phenoxy radical which formed was stabilized through an intramolecular hydrogen bonding. A subsequent interaction with a second DPPH radical afforded the dehydro form of these compounds as a final product (Shahidi, *et al.*, 1992; Keawpradub, *et al.*, 2001).

The structure-activity study by Lin and co-worker in 1997 indicated that hydroxyl groups and at least on flavone unit in the biflavonoids are required for activity and C-3, C-8" linkages is necessary for biflavanones to exhibit activity. Early work by Clemetson and Andersen showed that many flavonoids and other plant phenolics protected againt ascorbic acid destruction. They note that flavonoids with either 3',4' B-ring substitution, were active inhibitors (Larson, 1988). **RD5**, **RD16** and **FD8**, are flavonoids with 3',4' B-ring substitution. The effect on trapping of the radical of those compounds could be noticed.

Polyhydroxy chalcones are biosynthetic intermediates between cinnamic acids and flavonoids, also show considerable antioxidant activity for tard. Chalcones with two adjacent hydroxy groups were almost fully effective, introduction of

additional hydroxyl groups leading to only slight increases in inhibitory activity whereas hydrogenation of the chalcone double bond increased their antioxidant activity to some extent (Larson, 1988). The unsaturated chalcone **SD2** and saturated chalcones **SD6**, **SD8** and **SD12** have no adjacent hydroxy groups.

3.4 Biological activities of known compounds

This research have led to the isolation of seventy-five compounds from G. dulcis. Furthermore we have reviewed the research which involving some of known compounds isolated from G. dulcis and found that some of known compounds have been evaluated for some of the biological activities by many research groups. Cowanin (GD6), mangostin (GD14), 1,3,6-trihydroxy-7-methoxy-2,5-bis(3-methyl-2-butenyl) xanthone (RD8) and cowaxanthone (GD7) were reported to showed antibacterial activity againts S. aureus ATCC 25923 and S. aureus penicillin-resistant (Sundaram, et al., 1983; Mahabusarakam, et al., 1986; Na Pattalung, et al., 1994). Cowanin (GD6), cowaxanthone (GD7), β -mangostin (FD18) and 1,7-dihydroxyxanthone (SD5) exhibited moderate activity againt Plasmodium falciparum comparable to that of pyrimethamine (Likhitwitayawuid, et al., 1998a; 1998b). β-Mangostin (FD18), garcinone B (RD7), garcinone D (GD25), α -mangostin (GD14), mangostenol (GD23) and 1,7-dihydroxy-3-methoxy-2-(3-methyl-2-butenyl)xanthone (GD8) have been tested for antituberculosis potential, α -mangostin (GD14), β -mangostin (FD18) and garcinone B (RD7) exhibited strong inhibitory effect against Mycobacterium tuberculosis (Suksamrarn, et al., 2003). α -Mangostin (GD14) and β -mangostin (FD18) showed significant inhibitory cancer chemopreventive activity. The inhibition activity of cowanin (GD6) and cowaxanthone (GD7) were weaker than that of α - (GD14) and β-mangostin (FD18) (Ito, et al., 2003). Camboginol (GD1) and xanthochymol (FD11)

have shown to prossess antibacterial (Iinuma, et al., 1996) and anti-topoisomerase I and II activities (Tosa, et al., 1997). Anti-HIV activity has been described for guttiferone E (FD12) (Gustafson, et al., 1992). Camboginol (GD1) has been reported to possess antioxidative, antibiotic and anti-HIV activities and suppressed colonic aberrant crypt foci (ACF) formation (Ito, et al., 2003) and antimicrobial properties (Bakana, et al., 1987). Camboginol (GD1) and cambogin (RD14) has been reported to antioxidative activity (Ito, et al., 2003). Camboginol (GD1) is known to be an antioxidant, and its radical scavenging ability against DPPH is greater than that of DL-α-tocophenol. Free radical scavenging effects of camboginol and cambogin were stronger than that of vitamin E. Camboginol (GD1) and cambogin (RD14) caused rapid decolorization of the DPPH solution, indicating marked radical scavenging potencies (Ito, et al., 2003). Morelloflavone (GD20) and apigenin (RD13) showed the moderate activity in the HIV-1 RT enzyme assay, were test with human PBM cells infected with HIV-1 (strain LAV-1) whereas volkensiflavone (FD1) showed weakly active (Lin, et al., 1997).