



**Anti-tyrosinase and anti-microbial activities of Thai medicinal plants**

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Master of Pharmacy in Pharmaceutical Sciences**

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ชื่อวิทยานิพนธ์	ฤทธิ์ยับยั้งเอนไซม์ไทโรซิเนสและฤทธิ์ต้านเชื้อจุลินทรีย์จากสมุนไพรไทย
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### บทคัดย่อ

ทดสอบฤทธิ์ยับยั้งเอนไซม์ไทโรซิเนสจากสารสกัดหยาบชั้นเอทานอลของสมุนไพรไทยจำนวน 77 ตัวอย่าง พบว่ามี 11 ตัวอย่างที่มีฤทธิ์ยับยั้งเอนไซม์ไทโรซิเนสที่ดี (ยับยั้งมากกว่า 50%) และเมื่อนำตัวอย่างดังกล่าวทั้ง 11 ตัวอย่างมาทดสอบฤทธิ์ต้านเชื้อจุลินทรีย์ พบว่าสารสกัดหยาบของรากจำปาจะมีฤทธิ์ยับยั้งเอนไซม์ไทโรซิเนสและฤทธิ์ต้านเชื้อจุลินทรีย์ที่ดี นำสารสกัดจากรากจำปามาสกัดแยกองค์ประกอบทางเคมี โดยใช้เทคนิคทางโครมาโตกราฟี พบสารบริสุทธิ์ 5 ชนิด และสารผสม 1 ชนิด ได้แก่ Artocarpanone (1), Artocarpin (2), Cudraflavone C (3), Alkene (4), Lanosterol (5) และสารผสมของ  $\beta$ -Sitosterol และ Stigmasterol (6) และพบว่า (1) มีฤทธิ์ยับยั้งเอนไซม์ไทโรซิเนสที่ดี สำหรับ (2) และ (3) มีฤทธิ์ต้านเชื้อจุลินทรีย์ที่ดี ดังนั้น (1), (2) และ (3) จึงน่าสนใจที่จะศึกษาเพิ่มเติมเพื่อพัฒนาไปเป็นผลิตภัณฑ์เพื่อผิวขาวและยารักษาสิวต่อไป

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### ABSTRACT

Seventy seven samples of ethanol crude extract from Thai medicinal plants were examined of anti-tyrosinase activity. Eleven samples showed the potential of anti-tyrosinase activity (more than 50% of tyrosinase inhibition). Anti-microbial activity of these samples also examined. The ethanol root extract of *Artocarpus integer* showed the potential of anti-tyrosinase and anti-microbial activities. The chemical constituents of *A. integer* root extracts were isolated by chromatographic techniques. Five pure compounds and one mixture compound were isolated as Artocarpanone (**1**), Artocarpin (**2**), Cudraflavone C (**3**), Alkene (**4**), Lanosterol (**5**) and mixture of  $\beta$ -Sitosterol and Stigmasterol (**6**). And (**1**) exhibited anti-tyrosinase effect, (**2**) and (**3**) also showed the potential of anti-microbial activity. (**1**), (**2**) and (**3**) are interesting for further study in order to provide possibilities for the development of new whitening and anti-acne agents from *A. integer*.

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## LIST OF ABBREVIATIONS AND SYMBOLS

$\alpha$	=	Alpha
$\beta$	=	Beta
cm	=	Centimeter
$\delta$	=	Chemical shift
CFU	=	Colony forming unit
COSY	=	Correlation spectroscopy
$J$	=	Coupling constant
cAMP	=	cyclic adenosine monophosphate
$^{\circ}\text{C}$	=	Degree Celsius
Chloroform- $d$	=	Deuterated chloroform
DMSO- $d_6$	=	Deuterated dimethylsulfoxide
Methanol- $d_4$	=	Deuterated methanol
$\text{Na}_2\text{HPO}_4$	=	Disodium hydrogen phosphate
DEPT	=	Distortionless enhancement by polarization transfer
d	=	Doublet (for NMR spectra)
dd	=	Doublet of doublets (for NMR spectra)
$\approx$	=	Estimate
g	=	Gram
$\gamma$	=	Gamma
Hz	=	Hertz
hr	=	Hour
HMBC	=	$^1\text{H}$ -detected heteronuclear multiple bond coherence
HMQC	=	$^1\text{H}$ -detected heteronuclear multiple quantum coherence
kg	=	Kilogram
$m/z$	=	Mass to charge ratio
$\text{IC}_{50}$	=	Median inhibitory concentration
MHz	=	Megahertz

## LIST OF ABBREVIATIONS (CONTINUED)

MSH	=	Melanocyte stimulating hormone
mg	=	Milligram
mL	=	Milliliter
mm	=	Millimeter
μg	=	Microgram
μL	=	Microliter
μM	=	Micromolar
m	=	Multiplet (for NMR spectra)
nm	=	Nanometer
s	=	Singlet (for NMR spectra)
NaCl	=	Sodium chloride
NaH <sub>2</sub> PO <sub>4</sub> ·2H <sub>2</sub> O	=	Sodium dihydrogen phosphate anhydrate
spp.	=	In the plural in place of the specific epithet
ppm	=	Part per million
%	=	Percentage
t	=	Triplet (for NMR spectra)

## CHAPTER 1

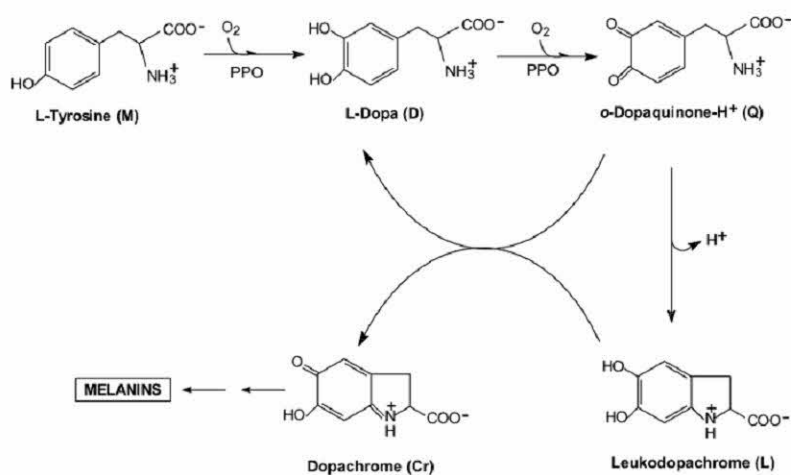
### INTRODUCTION

#### 1.1 Background

The people living in the tropical zone have the dark-colored skin and amount of melanin more than the people living in the other area (Jablonski and Chaplin, 2002). Asia is one in the tropical zone and the women in this area hopeful to have a white skin tone. The major to a white skin is against the ultraviolet (UV) radiation by covering or filtering agent, because the UV radiation is effect to skin dark-colored by indirection (Department of medical Sciences, 2001; Tengamnuay *et al.*, 2006). Some of cosmetic products for whitening and depigmentation, addition substances for accelerate to skin white such as hydroquinone, corticosteroids and mercury, but they are harm to health of skin, such as irreversible cutaneous damage, ochronosis, accumulation of mercury in the body and poisoning (Karioti *et al.*, 2007). These agents effect to melanogenesis, such as hydroquinone, it inhibits tyrosinase enzyme activity, which effects to the transformation of tyrosine to L-Dopa. Thus the melanogenesis and skin will be inhibition and whiteness, respectively. But effect of hydroquinone for a while, because it irritates to skin when using for a long time or more than 6 months, the skin is redness and burning (Department of medical Sciences, 2001).

Melanins are polyphenolic pigments and cause to dark-coloured (López-Serrano *et al.*, 2004). This pigment shows brown and black colour (National Science Museum, 2007). Melanin is distributed in the living things of the natural and has many different properties. In mammals, it is found in the eye, hair and skin (Hoogduijn *et al.*, 2004). The function of melanin is to defend the skin from UV radiation damage and removing reactive oxygen species (ROS). Various dermatological disorders result in the accumulation of an excessive level of melanin, such as melasma, age spots and sites of actinic damage (Kim and Uyama, 2005).

Tyrosinase, also known as polyphenoloxidase (Pênalver *et al.*, 2005) is an enzyme copper-containing (Gąsowska *et al.*, 2006), which is ubiquitously distributed in nature. Tyrosinase of mammals can be found in retina and skin. Tyrosinase is important to catalyst in the melanogenesis and change the other procession, such as in the vertebrates and fruits, this enzyme is important to the browning of fruits and vegetables (Clausa and Decker, 2006; García-Molina *et al.*, 2005). In the melanogenesis, this enzyme catalyses two distinct reactions as hydroxylation and oxidation. The enzyme in this reaction may call monophenolase and diphenolase, respectively. Monophenolase will change L-Tyrosine (M) to L-Dopa (D) and diphenolase will change L-Dopa (D) to *o*-Dopaquinone-H<sup>+</sup> (Q) and pass the intermediate finally to melanins (scheme 1.1) (García-Molina *et al.*, 2005).



**Scheme 1.1 Mechanism of melanogenesis**

(García-Molina *et al.*, 2005)

Acne is one dermal disease, which causes hurt social and psychological effects on sufferers. *Propionibacterium acnes* is anaerobic pathogen which cause to acne vulgaris or acne inflammation. However, *Staphylococcus epidermidis* and *S. aureus* are aerobic organism which cause abscess, usually involve to acne inflammation (Kumar *et al.*, 2007; Niyomkam, 2007 and Athikomkulchai *et al.*, 2008). Some mediators of inflammation, such as prostaglandin E2 (PGE<sub>2</sub>), it can be stimulate the melanogenesis (Petit and Piérard, 2003).

Screening of tyrosinase inhibition that is important to decrease melanin and may be developing into new drug to treatment for hyperpigmentation (Okunji *et al.*, 2007) and useful

in cosmetology (Kiken and Cohen, 2002). Tyrosinase inhibitors from natural products might help to solve the problem that concerns with the addition of the agent which is harmful to skin's health in cosmetic products. Therefore, studies of anti-tyrosinase and anti-microbial activities will be useful for development the product, which for whitening and treatment of skin infection, especially acne that usually find in the patient with hyperpigmentation (Kongcharoensuntorn *et al.*, 2005 and Chaisawadi *et al.*, 2008).

In this study, started by screening of anti-tyrosinase activity of 77 crude extracts from Thai medicinal plants by used kojic acid and water extract of *Artocarpus lakoocha* as positive controls, these agents are accepted to be the standard because they have high potential of tyrosinase inhibition activity. The crude extracts which showed more than 50% of tyrosinase inhibition would be selected for anti-microbial test. *Artocarpus integer* root was selected for phytochemical investigation, because it showed high potential of anti-tyrosinase and anti-microbial activities. Then this study was concerned with the isolation, purification and structure determination of chemical compounds from *A. integer*.

## 1.2 Objects

The main objectives in this investigation are as follows:

1. Preliminary screening of anti-tyrosinase activity from Thai medicinal plant extracts.
2. Preliminary screening of anti-microbial activity from Thai medicinal plant extracts which show high potential of tyrosinase inhibitory activity.
3. Isolation of the chemical constituents from the selected Thai medicinal plant extract.
4. Determination of anti-tyrosinase and anti-microbial activities of isolated pure compounds.



## CHAPTER 2

### HISTORICAL

#### 2.1. Melanin in mammalian

Melanins are extensively distributed pigments and are found in bacteria, fungi, plants and animals. Melanins of mammalian are divided into two types as eumelanin is brown or black and pheomelanin is red or yellow in color (Kim and Uyama, 2005). They are a heterogeneous polyphenol-like biopolymer with a complex structure where are synthesized by melanocyte cell in the basal layer of dermis, which contain specific enzyme controlling the produce of the melanins known as tyrosinase. Melanins synthesis takes place in specific organelles known as melanosomes. Melanosomes are transferred from the melanocyte cells into surrounding keratinocyte cells to protect the skin from UV radiation damage and removing reactive oxygen species (ROS) (Hoogduijn, 2004; Kim and Uyama, 2005 and Petit and Piérard, 2003).

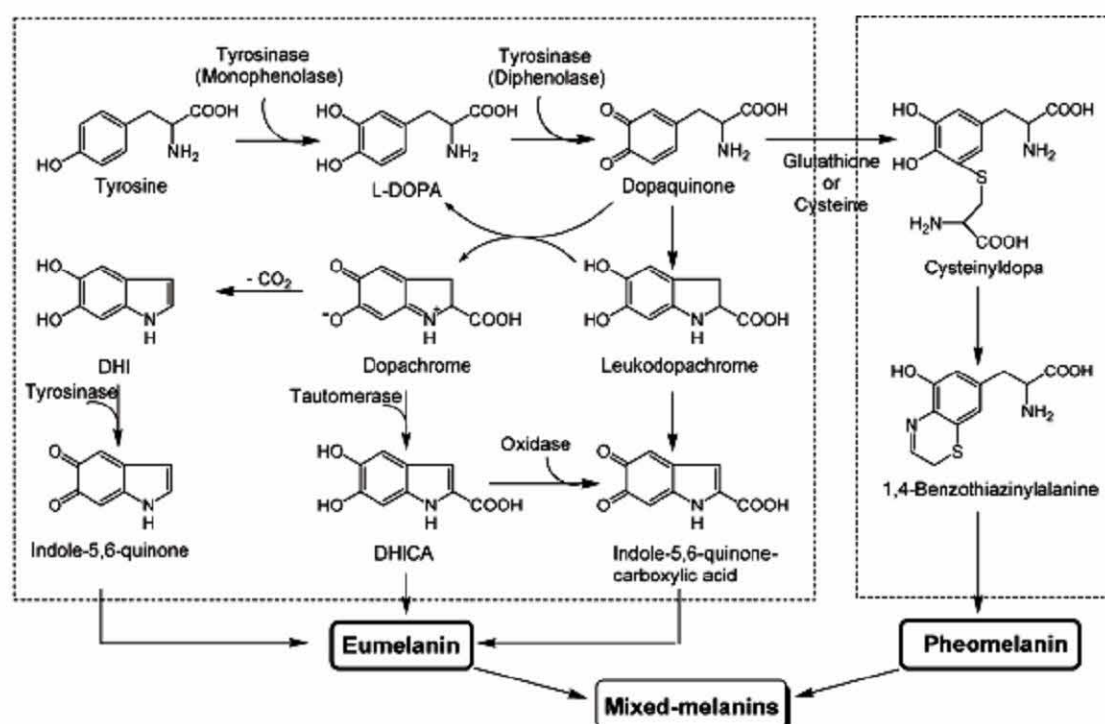
##### 2.1.1 Melanogenesis

Melanogenesis is started with the first step of L-tyrosine hydroxylation into L-3, 4-dihydroxyphenylalanine (L-Doap) and L-Dopa oxidation into *o*-dopaquinone catalyzed by tyrosinase. This first step is the rate-limiting step in melanogenesis. Dopaquinone can undergo two different type reactions (see in scheme 2.1) (Kim and Uyama, 2005; Petit and Piérard, 2003).

The first reaction is *o*-dopaquinone cyclization, started by amino group undergoes an intramolecular 1, 4-addition to the benzene ring into leukodopachrome and quickly oxidize to dopachrome. The dopachrome formation is lead to synthesis eumelanin which can undergo two difference type reactions as (1) decarboxylation of dopachrome into 5, 6-dihydroxyindole (DHI), its following oxidation into indole-5, 6-quinone and (2) enzymatically transformed into 5,6-dihydroxyindole-2-carboxylic acid (DHICA) by dopachrome tautomerase and DHICA can be oxidation into indole-5, 6-quinone carboxylic acid. Indole-5, 6-

quinone, indole-5, 6-quinone carboxylic acid and DHICA are subsequent polymerization into eumelanin (Kim and Uyama, 2005; Petit and Piérard, 2003).

The second reaction is the group of sulfhydryl compounds such as glutathione and cysteine nucleophilically attacks *o*-dopaquinone to create cysteinyl-dopa or glutathionyl-dopa. The cysteinyl-dopa or glutathionyl-dopa are subsequent cyclization and polymerization into pheomelanin (Kim and Uyama, 2005; Petit and Piérard, 2003). The interaction between the eumelanin and pheomelanin compounds gives increase to a heterogeneous merge of mixed-type melanins (Kim and Uyama, 2005).



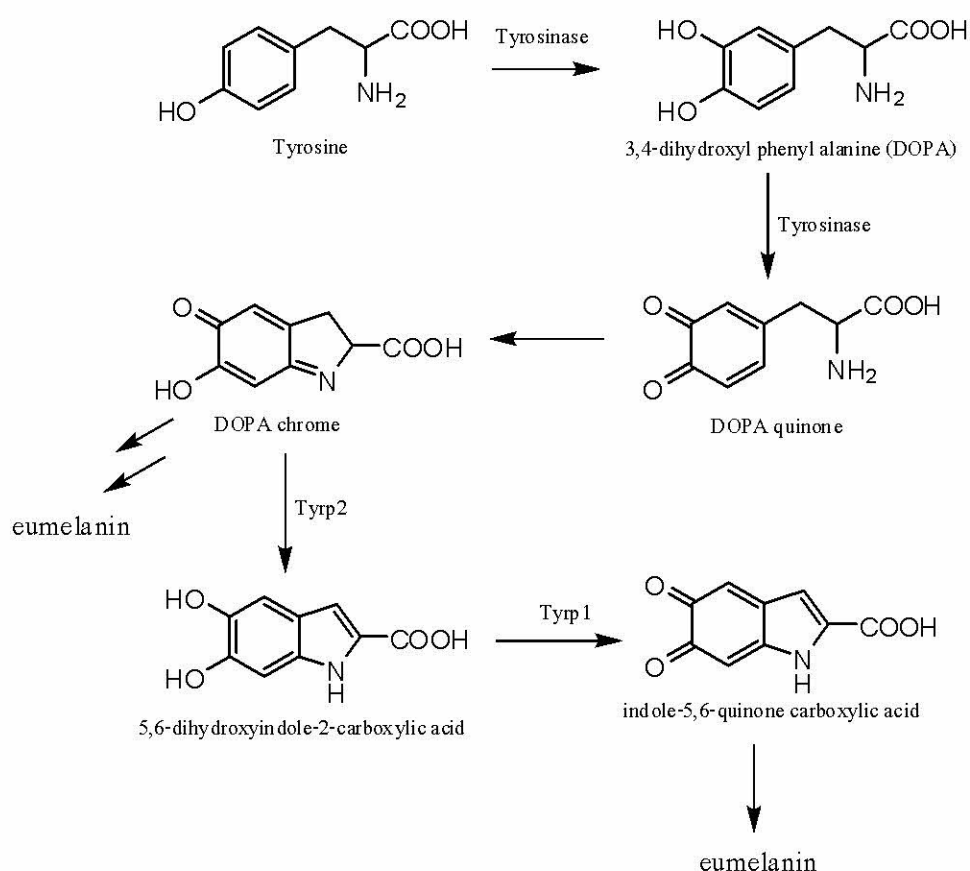
**Scheme 2.1 Step of melanogenesis in human's skin**

(Kim and Uyama, 2005)

The initiation of the melanogenesis procedure may be involved with ultraviolet radiation, free radical, reactive oxygen species, inflammatory mediators and hormone (Petit and Piérard, 2003). These factors may act in combination which response to make the skin disorder, such as age spot, melasma, freckles and other forms of melanin hyperpigmentation (Kim and Uyama, 2005; Petit and Piérard, 2003).

### 2.1.2 Anti-tyrosinase and tyrosinase inhibitor

Melanogenesis is a highly cooperative process carried out by tyrosinase family proteins, which include tyrosinase, tyrosinase-related protein 1 (tyrp1) and tyrosinase-related protein 2 (tyrp2) (see in scheme 2.2) (Wang and Hebert, 2006). On the contrary, if one step of the melanogenesis fails, melanins disappear on the skin. However, tyrosinases are the first and rate-limiting step of melanogenesis (see in scheme 2.2) (Clausa and Decker, 2006). Therefore, tyrosinase inhibitors have become increasingly important for depigmentation and whitening products that may be used to preclude or treat hyperpigmentation (Kim and Uyama, 2005).



**Scheme 2.2 Melanogenesis pathway regulated by tyrosinase, tyrosinase family proteins Tyrp1 and Tyrp2 (Wang and Hebert, 2006)**

### 2.1.2.1 Enzyme tyrosinase

Tyrosinase (1.14.18.1) also known as polyphenoloxidase (PPO) is the copper-containing enzyme. It uses its molecular oxygen of binuclear copper center to catalyze two different enzymatic reactions (Claus and Decker, 2006; [Wang and Hebert, 2006](#)), the hydroxylation of L-tyrosine to L-3, 4-dihydroxyphenylalanine (L-Dopa) and the subsequent oxidation of L-Dopa to DOPAquinone (see scheme 2.2). Tyrosinase of human is a type I membrane glycoprotein that contains 529 amino acid ([Wang and Hebert, 2006](#)).

### 2.1.2.2 Tyrosinase inhibitors

The commercials of tyrosinase inhibitors are contain in cosmetic and whitening products, such as kojic acid, arbutin and azelaic acid (Kim and Uyama, 2005; Petit and Piérard, 2003). However, many previously reports are studied for tyrosinase inhibition from medicinal plants, which development for whitening and anti-browning agents. Some of tyrosinase inhibitors from previously reports are showed in table 2.1

### 2.1.3 Anti-tyrosinase assay by enzymetic and cell culture methods

The *in vitro* mushroom tyrosinase inhibition assay in basic step evaluate the direct consequence of a given skin whitener on tyrosinase activity. The substrate of this enzyme is L-tyrosine and the reaction involve the occurrence of the co-substrate is L-Dopa (see in scheme 2.1). The activity of tyrosinase and tyrosinase inhibitor are quantified following the detection of dopachrome at 475 (Petit and Piérard, 2003) or 492 nm by spectrometer, this method as known dopachrome method (Sritularak, 1998a; Sritularak, 1998b). Other cell-free enzymatic analyze can be performed, for order to investigation, Tyrp-1 and tyrp-2 activities (Petit and Piérard, 2003).

Table 2.1 Anti-tyrosinase activity from medicinal plants

Medicinal plants	Compound/ Crude extract	IC <sub>50</sub>	Reference
<i>Camellia sinensis</i>	Epigallocatechin-3- <i>O</i> - gallate (EGCG)	34.58 $\mu$ M	No <i>et al.</i> ,1999
<i>Camellia sinensis</i>	Gallocatechin-3- <i>O</i> - gallate (GCG)	17.34 $\mu$ M	No <i>et al.</i> ,1999
<i>Camellia sinensis</i>	Epicatechin-3- <i>O</i> - gallate (ECG)	34.10 $\mu$ M	No <i>et al.</i> ,1999
<i>Morus alba</i> (leaves)	Mulberroside F	0.49 $\mu$ M	Lee <i>et al.</i> , 2002
<i>Citrus</i> sp. (peel)	Nobiletin	42.6 $\mu$ M	Sasaki and Yoshizaki, 2002
<i>Pharbitis nil</i> (seed)	Ethanol extract	24.9 $\mu$ g/mL	Wang <i>et al.</i> , 2006
<i>Sophora japonica</i> (flower)	Ethanol extract	95.6 $\mu$ g/mL	Wang <i>et al.</i> , 2006
<i>Spatholobus suberectus</i> (stem)	Ethanol extract	83.9 $\mu$ g/mL	Wang <i>et al.</i> , 2006
<i>Morus alba</i> (leaves)	Ethanol extract	78.3 $\mu$ g/mL	Wang <i>et al.</i> , 2006

## 2.2 Acne

Acne expands in follicular pilosebaceous unit. These units are largest on the face, neck and back. The developments of acne cause from the four major pathogenic factors are increased sebum production, disorders of the microflora, cornification of the pilosebaceous duct and inflammation. The microorganisms are commensally of normal skin as *Propionibacterium acnes*, *Staphylococcus aureus* and *S. epidermidis*, proliferate rapidly during teenager and often involved in the development of acnes (Niyomkan, 2008).

### 2.2.1 Acne vulgaris

Acne vulgaris is the majority general skin disease, a chronic inflammatory disorder in teenager consist pilosebaceous follicles. *P. acnes* is anaerobic microorganism, it is cause to acne vulgaris by metabolize sebaceous triglycerides into fatty acids, which chemotactically attract neutrophils. *S. aureus* is aerobic microorganism, usually involves in superficial infections within the sebaceous unit ([Traidej-Chomnawang, 2005](#)).

### 2.2.2 Acne treatments and anti-bacterial activity from medicinal plants

Variety of the most suitable therapy should be based on an inclusive assessment of the patient and pharmacological profile of the agents. The options of treatment for acne are as follows (Niyomkan, 2008).

(1) Physical therapy: There are many treatments including acne surgery, laser and phototherapy and using intralesional corticosteroid.

(2) Systemic therapy: The two major systemic modalities used in acne are antibiotics and estrogenic hormone. Both represents major step forward in the therapy of this disease.

(3) Local therapy: The general therapy including topical agents' tretinoin, azelaic acid, salicylic acid, benzoyl peroxide and anti-bacterials (clindamycin, erythromycin).

In previously reports, many medicinal plants were examined for antimicrobial activity against microorganisms frequently involved in acne inflammation such as *Propionibacterium acnes*, *Staphylococcus epidermidis* and *S. aureus*. Anti-bacterial activity from some medicinal plants is against *P. acne*, *S. aureus* and *S. epidermidis* shown in table 2.2.



**Table 2.2 Anti-bacterial activity from some medicinal plants**

Medicinal plants	Microorganism	Compound/ Crude extract	MIC (mg/mL)	MBC (mg/mL)	Reference
<i>Alpinia galanga</i>	<i>P. acnes</i>	1'- acetoxychavicol acetate	0.062	0.250	Niyomkamn <i>et al.</i> , 2007
<i>Coscinium fenestratum</i>	<i>P. acnes</i>	Ethanol extract	0.049	0.049	Kumar <i>et al.</i> , 2007
	<i>S. epidermidis</i>	Ethanol extract	0.049	0.165	
<i>Curcuma aromatica</i>	<i>P. acnes</i>	Ethanol extract	5	-	Giwanon
	<i>S. epidermidis</i>	Ethanol extract	5-10	-	<i>et al.</i> , 2006
	<i>S. aureus</i>	Ethanol extract	5-10	-	

- = not detect

## 2.3 Phytochemical investigation techniques

The generally phytochemical investigation techniques are chromatographic and spectroscopic techniques. These techniques are useful for isolation and identification of compound from the mixture compounds.

### 2.3.1 Chromatographic techniques

Chromatography is a method of separation in which the components to be separated are scattered in the middle of two phases, one of which is stationary phase while the mobile phase moves in an accurate direction. The purpose of chromatography is to separate the components of a mixture for purification. The chromatographic techniques for isolation of compound are as follow (Ettre, 1993; <http://en.wikipedia.org/wiki/Chromatography>):

(1) Column chromatography is a method used to purify compounds from mixtures of compounds. It is often used for preparative applications on scales from micrograms up to kilograms. The classical preparative chromatography column is a glass tube with a diameter from 5 mm to 50 mm and a height of 50 cm to 1 m with a tap at the bottom. Two methods are normally used to prepare a column; the dry method and the wet method. The stationary phase in column chromatography is a solid. The most common stationary phase for column chromatography is silica gel. The mobile phase is either a pure solvent or a mixture of dissimilar solvents and has also been selected so that the different compounds can be separated successfully ([http://en.wikipedia.org/wiki/Column\\_chromatography](http://en.wikipedia.org/wiki/Column_chromatography)).

(2) Thin layer chromatography (TLC) is a chromatography technique used to separate mixtures. It is performed on a sheet of glass, plastic, or aluminum foil, which is coated with a thin layer of adsorbent material, usually silica gel and aluminium oxide. Different compounds in the sample mixture move at dissimilar rates suitable to the differences in their attraction to the stationary phase, and because of differences in solubility in the solvent (Stoddard, 2007; [http://en.wikipedia.org/wiki/Thin\\_layer\\_chromatography](http://en.wikipedia.org/wiki/Thin_layer_chromatography)).

### 2.3.2 Spectroscopic technique

The majority important methods for identify compound are the nuclear magnetic resonant spectroscopy ( $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopy), the mass spectrometry, the infrared and the UV/Visible spectroscopy. These methods are the main tool of modern chemistry for the identification of molecular structures and used to determine and confirm molecular structures. ([www.oc-praktikum.de/en/articles/pdf/Spectroscopy\\_en.pdf](http://www.oc-praktikum.de/en/articles/pdf/Spectroscopy_en.pdf)). These techniques were used in this study.

### 2.3.3 Chemical constituents of *Artocarpus* spp.

In this study, we have been selecting *Artocarpus integer* for phytochemical investigation. *A. integer* is the tree in genus *Artocarpus*. *Artocarpus* belongs to the Moraceae family. The chemical constituents of plants in the genus *Artocarpus* can be classified into five



groups, namely flavonoids, triterpenoids, steroids, stilbenes and miscellaneous substances. The chemical constituents which were isolated from *Artocarpus* spp. from previously reports are showed in table 2.3 and the chemical structure of isolated compounds are showed in figure 2.1 (except the structure numbers 42, 43, 44, 119, 136, 137 and 138)

**Table 2.3 Chemical constituents of *Artocarpus* spp.**

Compound	Plant part	Scientific name	Reference
Acetylcholine [1]	Seed	<i>A. heterophyllus</i>	Pereira <i>et al.</i> , 1962
Afzelechin-(4 $\beta$ →8)-chatechin [2]	Leaf	<i>A. heterophyllus</i>	An <i>et al.</i> , 1992
Albanin A [3]	Root	<i>A. gomezianus</i>	Likhiwitayawuid
	Stem	<i>A. heterophyllus</i>	<i>et al.</i> , 2000 Arung <i>et al.</i> , 2006
$\gamma$ -Amimobutyric acid [4]	Leaf	<i>A. altilis</i>	Durand <i>et al.</i> , 1962
$\alpha$ -Amyrin [5]	Latex	<i>A. altilis</i>	Ultee, 1949
$\alpha$ -Amyrin acetate [6]	Fruit	<i>A. altilis</i>	Altman and Zito, 1976
Andalasin A [7]	Root	<i>A. gomezianus</i>	Likhitwitayawuid and Sritularak, 2001
Apigenin [8]	Heartwood	<i>A. altilis</i>	Shimizu <i>et al.</i> , 1998
Arbutin [9]	Leaf	<i>A. gomezianus</i>	Kingroungpet, 1994
Artelasticin [10]	Heartwood	<i>A. elasticus</i>	Kijjoa <i>et al.</i> , 1996
		<i>A. lanceifolius</i>	Syah <i>et al.</i> , 2001
Artelasticinol [11]	Root bark	<i>A. elasticus</i>	Ko <i>et al.</i> , 2005
Artelastin [12]	Heartwood	<i>A. elasticus</i>	Kijjoa <i>et al.</i> , 1996
Artelastinin [13]	Heartwood	<i>A. elasticus</i>	Kijjoa <i>et al.</i> , 1998

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Artelastofuran [14]	Heartwood	<i>A. elasticus</i>	Kijjoa <i>et al.</i> , 1998
	Heartwood	<i>A. lanceifolius</i>	Syah <i>et al.</i> , 2001
Artelastoheterol [15]	Root bark	<i>A. elasticus</i>	Ko <i>et al.</i> , 2005
Artelastoxanthone [16]	Root bark	<i>A. elasticus</i>	Ko <i>et al.</i> , 2005
Artocarbene [19]	Root bark	<i>A. heterophyllus</i>	Lu and Lin, 1994
	Heartwood	<i>A. incisus</i>	Shimizu <i>et al.</i> , 1997
Artocarpanone [20]	Heartwood	<i>A. heterophyllus</i>	Radhakrishnan <i>et al.</i> , 1965
	Heartwood	<i>A. hirsuta</i>	Venkataraman, 1972
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
Artocarpanone A [21]	Root bark	<i>A. heterophyllus</i>	Lin <i>et al.</i> , 1995
Artocarpesin [22]	Heartwood	<i>A. altilis</i>	Shimizu <i>et al.</i> , 1998
	Heartwood	<i>A. chaplacha</i>	Rao <i>et al.</i> , 1972 Kijjoa <i>et al.</i> , 1996
	Heartwood	<i>A. elasticus</i>	Venkataraman, 1972
	Heartwood	<i>A. gomezianus</i>	
	Heartwood	<i>A. heterophyllus</i>	Radhakrishnan <i>et al.</i> , 1965
	Heartwood	<i>A. hirsuta</i>	Venkataraman, 1972
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
Artocarpetin [23]	Heartwood	<i>A. heterophyllus</i>	Venkataraman, 1972
	Heartwood	<i>A. hirsuta</i>	Venkataraman, 1972
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Artocarpetin A [24]	Root	<i>A. heterophyllus</i>	Lin <i>et al.</i> , 1995
Artocarpetin B [25]	Root	<i>A. heterophyllus</i>	Chung <i>et al.</i> , 1995
Artocarpin [26]	Heartwood	<i>A. altilis</i>	Venkataraman, 1972
	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
	Heartwood	<i>A. communis</i>	Han <i>et al.</i> , 2006
	Heartwood	<i>A. champeden</i>	Euis <i>et al.</i> , 2005
	Heartwood	<i>A. chaplacha</i>	Rao <i>et al.</i> , 1972
	Heartwood	<i>A. elasticus</i>	Kijjoa <i>et al.</i> , 1996
	Heartwood	<i>A. gomezianus</i>	Venkataraman, 1972
	Heartwood	<i>A. heterophyllus</i>	Radhakrishnan <i>et al.</i> , 1965
	Heartwood	<i>A. hirsuta</i>	Venkataraman, 1972
	Heartwood	<i>A. lakoocha</i>	Venkataraman, 1972
Artocarpol A [27]	Root bark	<i>A. rigida</i>	Chung <i>et al.</i> , 2000
Artocarpol B [28]	Root bark	<i>A. rigida</i>	Ko <i>et al.</i> , 2000
Artocarpol C [29]	Root bark	<i>A. rigida</i>	Ko <i>et al.</i> , 2000
Artocarpol D [30]	Root bark	<i>A. rigida</i>	Ko <i>et al.</i> , 2000
Artocarpol E [31]	Root bark	<i>A. rigida</i>	Ko <i>et al.</i> , 2000
Artocarpol F [32]	Root bark	<i>A. rigida</i>	Ko <i>et al.</i> , 2000
Artocarpol G [33]	Root bark	<i>A. rigida</i>	Lu <i>et al.</i> , 2002
Artocarpol H [34]	Root bark	<i>A. rigida</i>	Lu <i>et al.</i> , 2002
Artocarpol I [35]	Root	<i>A. rigida</i>	Lu <i>et al.</i> , 2003
Artocarpone A [36]	Bark	<i>A. champeden</i>	Widyawaruyanti <i>et al.</i> , 2007

**Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)**

Compound	Plant part	Scientific name	Reference
Artocarpone B [37]	Bark	<i>A. champeden</i>	Widyawaruyanti <i>et al.</i> , 2007
Artocarpus chalcone AC-3-1 [38]	Leaf	<i>A. altilis</i>	Wang <i>et al.</i> , 2007b
Artocarpus chalcone AC-5-1 [39]	Leaf	<i>A. altilis</i>	Wang <i>et al.</i> , 2007b
Artocarpus flavone KB-2 [40]	Bark	<i>A. communis</i>	Fujimoto <i>et al.</i> , 1990
Artocarpus flavone KB-3 (Artonin E) [41]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
	Bark	<i>A. communis</i>	Fujimoto <i>et al.</i> , 1990
	Bark	<i>A. kemando</i>	Seo <i>et al.</i> , 2003
	Trunk	<i>A. lanceifolius</i>	Cao <i>et al.</i> , 2003
Artocarpus integrifolia $\alpha$ -D-Galactose specific lectin [42]	Bark	<i>A. rigida</i>	Hano <i>et al.</i> , 1990b
	Seed	<i>A. heterophyllus</i>	Suresh <i>et al.</i> , 1982
Artocarpus lactin C [43]	Seed	<i>A. integer</i>	Hashim <i>et al.</i> , 1992
Artocarpus lectin CE-A-I [44]	Seed	<i>A. integrifolia</i>	Ferreia <i>et al.</i> , 1992
Artochamin A (Artoindonesianin D) [45]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
	Root	<i>A. kemando</i>	Hakim <i>et al.</i> , 2006
Artochamin B [46]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
Artochamin C [47]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
Artochamin D [48]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
Artochamin E [49]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
Artocommunol CA [50]	Root	<i>A. communis</i>	Chan <i>et al.</i> , 2003
Artocommunol CB [51]	Root	<i>A. communis</i>	Chan <i>et al.</i> , 2003
Artocommunol CC [52]	Root	<i>A. communis</i>	Chan <i>et al.</i> , 2003

**Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)**

Compound	Plant part	Scientific name	Reference
Artocommunol CD [53]	Root	<i>A. communis</i>	Chan <i>et al.</i> , 2003
Artocommunol CE [54]	Root	<i>A. communis</i>	Chan <i>et al.</i> , 2003
Artoflavanone [55]	Root	<i>A. heterophyllus</i>	Dayal and Seshadri, 1974
Artogomezianol [56]	Root	<i>A. gomezianus</i>	Likhitwitayawuid and Sritularak., 2001
Artogomezianone [57]	Heartwood	<i>A. gomezianus</i>	Likhitwitayawuid <i>et al.</i> , 2006
Artoindonesianin A [58]	Root	<i>A. champeden</i>	Hakim <i>et al.</i> , 1999
Artoindonesianin A-2 [59]	Heartwood	<i>A. champeden</i>	Syah <i>et al.</i> , 2006b
Artoindonesianin A-3 [60]	Heartwood	<i>A. champeden</i>	Syah <i>et al.</i> , 2006b
Artoindonesianin B [61]	Root	<i>A. champeden</i>	Hakim <i>et al.</i> , 1999
	Root	<i>A. kemandu</i>	Hakim <i>et al.</i> , 2006
Artoindonesianin E [62]	Bark	<i>A. champeden</i>	Widyawaruyanti <i>et al.</i> , 2007
Artoindonesianin G [63]	Heartwood	<i>A. lanceifolius</i>	Syah <i>et al.</i> , 2001
Artoindonesianin H [64]	Heartwood	<i>A. lanceifolius</i>	Syah <i>et al.</i> , 2001
Artoindonesianin I [65]	Heartwood	<i>A. lanceifolius</i>	Syah <i>et al.</i> , 2001
Artoindonesianin J [66]	Root and bark	<i>A. bracteata</i>	Ersam <i>et al.</i> , 2002
Artoindonesianin L [67]	Root bark	<i>A. rotunda</i>	Suhartati <i>et al.</i> , 2001
Artoindonesianin M [68]	Heartwood	<i>A. champeden</i>	Euis <i>et al.</i> , 2005
Artoindonesianin N [69]	Bark	<i>A. gomezianus</i>	Hakim <i>et al.</i> , 2002b
Artoindonesianin O [70]	Bark	<i>A. gomezianus</i>	Hakim <i>et al.</i> , 2002b
Artoindonesianin P [71]	Bark	<i>A. lanceifolius</i>	Hakim <i>et al.</i> , 2002a
Artoindonesianin Q [72]	Bark	<i>A. champeden</i>	Syah <i>et al.</i> , 2002b

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Artoindonesianin R [73]	Bark	<i>A. champeden</i>	Syah <i>et al.</i> , 2002b; Widyawaruyanti <i>et al.</i> , 2007
Artoindonesianin S [74]	Heartwood	<i>A. champeden</i>	Syah <i>et al.</i> , 2002b
Artoindonesianin T [75]	Heartwood	<i>A. champeden</i>	Syah <i>et al.</i> , 2002b
Artoindonesianin U [76]	Heartwood	<i>A. champeden</i>	Syah <i>et al.</i> , 2004
Artoindonesianin V [77]	Heartwood	<i>A. champeden</i>	Syah <i>et al.</i> , 2004
Artoindonesianin X [78]	Root bark	<i>A. fretessi</i>	Soekamto <i>et al.</i> , 2003
Artoindonesianin Y [79]	Root bark	<i>A. fretessi</i>	Soekamto <i>et al.</i> , 2003
Artoindonesianin Z-1 [80]	Bark	<i>A. lanceifolius</i>	Syah <i>et al.</i> , 2006a
Artoindonesianin Z-2 [81]	Bark	<i>A. lanceifolius</i>	Syah <i>et al.</i> , 2006a
Artoindonesianin Z-3 [82]	Bark	<i>A. lanceifolius</i>	Hakim <i>et al.</i> , 2006
Artolastochromene [83]	Heartwood	<i>A. elasticus</i>	Kijjoa <i>et al.</i> , 1996
Artolastofuran [84]	Heartwood	<i>A. elasticus</i>	Kijjoa <i>et al.</i> , 1998
Artomunoflavanone [85]	Root	<i>A. communis</i>	Weng <i>et al.</i> , 2006
Artomunoxanthentrione [86]	Root bark	<i>A. communis</i>	Shieh and Lin, 1992
Artomunoxanthone [87]	Root bark	<i>A. communis</i>	Shieh and Lin, 1992
Artomunoxanthotrione epoxide [88]	Root bark	<i>A. communis</i>	Lin <i>et al.</i> , 1992
Artonin A [89]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
	Root	<i>A. champeden</i>	Hakim <i>et al.</i> , 1999
	Bark		Widyawaruyanti <i>et al.</i> , 2007
	Root bark	<i>A. heterophyllus</i>	Hano <i>et al.</i> , 1989
	Bark	<i>A. fretessi</i>	Soekamto <i>et al.</i> , 2003
Artonin B [90]	Root bark	<i>A. heterophyllus</i>	Hano <i>et al.</i> , 1990a
Artonin C [91]	Root bark	<i>A. heterophyllus</i>	Hano <i>et al.</i> , 1990a
Artonin D [92]	Root bark	<i>A. heterophyllus</i>	Hano <i>et al.</i> , 1990a

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Artonin F [93]	Bark	<i>A. communis</i>	Hano <i>et al.</i> , 1990c
	Root bark	<i>A. elasticus</i>	Ko <i>et al.</i> , 2005
Artonin G [94]	Bark	<i>A. rigida</i>	Hano <i>et al.</i> , 1990b
Artonin H [95]	Bark	<i>A. rigida</i>	Hano <i>et al.</i> , 1990b
Artonin I [96]	Root bark	<i>A. heterophyllus</i>	Hano <i>et al.</i> , 1990a
Artonin J [97]	Root bark	<i>A. heterophyllus</i>	Aida <i>et al.</i> , 1993
	Root bark	<i>A. teysmanii</i>	Makmur <i>et al.</i> , 2000
Artonin K [98]	Bark	<i>A. altilis</i>	Aida <i>et al.</i> , 1997
	Root bark	<i>A. heterophyllus</i>	Aida <i>et al.</i> , 1993
Artonin L [99]	Root bark	<i>A. heterophyllus</i>	Aida <i>et al.</i> , 1993
Artonin M [100]	Bark	<i>A. rigida</i>	Hano <i>et al.</i> , 1993
Artonin N [101]	Bark	<i>A. rigida</i>	Hano <i>et al.</i> , 1993
Artonin O [102]	Bark	<i>A. rigida</i>	Hano <i>et al.</i> , 1993
Artonin P [103]	Bark	<i>A. rigida</i>	Hano <i>et al.</i> , 1993
Artonin Q [104]	Bark	<i>A. heterophyllus</i>	Aida <i>et al.</i> , 1994
Artonin R [105]	Bark	<i>A. heterophyllus</i>	Aida <i>et al.</i> , 1994
Artonin S [106]	Heartwood	<i>A. elasticus</i>	Cidade <i>et al.</i> , 2001
	Bark	<i>A. heterophyllus</i>	Aida <i>et al.</i> , 1994
Artonin T [107]	Bark	<i>A. heterophyllus</i>	Aida <i>et al.</i> , 1994
Artonin U [108]	Bark	<i>A. heterophyllus</i>	Aida <i>et al.</i> , 1994
Artonin V [109]	Root bark	<i>A. altilis</i>	Hano <i>et al.</i> , 1994
Artonin X [110]	Bark	<i>A. heterophyllus</i>	Shinomiya <i>et al.</i> , 1995
Artonol A [111]	Bark	<i>A. communis</i>	Aida <i>et al.</i> , 1997
Artonol B [112]	Bark	<i>A. communis</i>	Aida <i>et al.</i> , 1997

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Artonol C [113]	Bark	<i>A. communis</i>	Aida <i>et al.</i> , 1997
Artonol D [114]	Bark	<i>A. communis</i>	Aida <i>et al.</i> , 1997
Artonol E [115]	Bark	<i>A. communis</i>	Aida <i>et al.</i> , 1997
Artostilbene A [116]	Wood	<i>A. chama</i>	Wang <i>et al.</i> , 2007a
Artostilbene B [117]	Wood	<i>A. chama</i>	Wang <i>et al.</i> , 2007a
Aurantiamide acetate [118]	Seed	<i>A. heterophyllus</i>	Chakraborty and Mandal, 1981
$\beta$ -Amyrin acetate [119]	Latex	<i>A. altilis</i>	Ultee, 1949
	Latex	<i>A. elasticus</i>	Ultee, 1949
	Bark	<i>A. lakoocha</i>	Kapel and Joshi, 1960
Betulin [120]	Root bark	<i>A. heterophyllus</i>	Lu and Lin, 1994
Betulinic acid [121]	Root	<i>A. heterophyllus</i>	Dayal and Seshadri, 1974
	Root bark		Lu and Lin, 1994
Butyosoermol [122]	Fruit	<i>A. heterophyllus</i>	Barton, 1951
Carpachromene [123]	Root	<i>A. bracteata</i>	Ersam <i>et al.</i> , 2002
	Bark		
Carpelastofuran [124]	Heartwood	<i>A. elasticus</i>	Cidade <i>et al.</i> , 2001
Chaplashin [125]	Heartwood	<i>A. chaplacha</i>	Rao <i>et al.</i> , 1972
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
Chromanoartobilochromen A [126]	Trunk bark	<i>A. nobilis</i>	Kumar <i>et al.</i> , 1977
Chromanoartobilochromen B [127]	Trunk bark	<i>A. nobilis</i>	Pavanasasivum <i>et al.</i> , 1974; Kumar <i>et al.</i> , 1977



Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Cudraflavone A	Stem	<i>A. altilis</i>	Chen <i>et al.</i> , 1993
(Isocyclomorusin) [128]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
	Root bark	<i>A. communis</i>	Shieh and Lin, 1992
	Heartwood		
	Root	<i>A. gomezianus</i>	Likhitwitayawuid <i>et al.</i> , 2000
	Heartwood		Likhitwitayawuid <i>et al.</i> , 2000
	Root bark	<i>A. heterophyllus</i>	Lin <i>et al.</i> , 1995
Cudraflavone C [129]	Heartwood	<i>A. champeden</i>	Euis <i>et al.</i> , 2005
	Heartwood	<i>A. communis</i>	Han <i>et al.</i> , 2006
	Root bark	<i>A. glaucus</i>	Hakim <i>et al.</i> , 2006
	Root	<i>A. gomezianus</i>	Likhitwitayawuid <i>et al.</i> , 2000
	Stem	<i>A. heterophyllus</i>	Arung <i>et al.</i> , 2006
Cyanomaclurin [130]	Heartwood	<i>A. heterophyllus</i>	Radhakrishnan <i>et al.</i> , 1965
	Heartwood	<i>A. hirsuta</i>	Venkataraman, 1972
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
Cycloaltilisinsin [131]	Stem	<i>A. altilis</i>	Chen <i>et al.</i> , 1993
Cycloart-23-ene-3 $\beta$ -25-diol [132]	Fruit	<i>A. altilis</i>	Altman and Zito, 1976
	Fruit	<i>A. heterophyllus</i>	Kielland and Malterud, 1994

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Cycloart-24-ene-3 $\beta$ -ol (Cycloartenol) [133]	Fruit	<i>A. altilis</i>	Altman and Zito, 1976; Pavanasasivam and Sultanbawa, 1973 Barton, 1951
	Fruit	<i>A. heterophyllus</i>	Nogueira and Correia, 1958
	Wood		Pavanasasivam and Sultanbawa, 1973; Barik <i>et al.</i> , 1994
	Bark	<i>A. lakoocha</i>	Pavanasasivam and Sultanbawa, 1973
	Latex		Pavanasasivam and Sultanbawa, 1973
	Bark		Pavanasasivam and Sultanbawa, 1973
	Heartwood	<i>A. nobilis</i>	Pavanasasivam and Sultanbawa, 1973
Cycloart-25-ene-3 $\beta$ -24-diol [134]	Fruit	<i>A. altilis</i>	Altman and Zito, 1976
	Fruit	<i>A. heterophyllus</i>	Kielland and Malterud , 1994
Cycloartelastoxanthendiol [135]	Root bark	<i>A. elasticus</i>	Ko <i>et al.</i> , 2005
Cycloartelastoxanthone [136]	Root bark	<i>A. elasticus</i>	Ko <i>et al.</i> , 2005

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Cycloartenone [137]	Bark	<i>A. altilis</i>	Pavanasasivam and Sultsnbswa, 1973
	Fruit	<i>A. heterophyllus</i>	Barton, 1951
	Bark		Pavanasasivam and Sultanbawa, 1973
	Root		Dayal and Seshadri, 1974
	Latex		Pant and Chaturvedi, 1989
	Bark	<i>A. lakoocha</i>	Barik <i>et al.</i> , 1994
	Bark	<i>A. nobilis</i>	Pavanasasivam and Sultsnbswa, 1973
	Heartwood		Pavanasasivam and Sultsnbswa, 1973
Cycloartenyl actate [138]	Bark	<i>A. altilis</i>	Pavanasasivam and Sultsnbswa, 1973
	Stem bark	<i>A. chaplacha</i>	Mahato, Banerjee and Chakravarti, 1971
	Bark	<i>A. heterophyllus</i>	Pavanasasivam and Sultsnbswa, 1973
	Bark	<i>A. nobilis</i>	Pavanasasivam and Sultsnbswa, 1973
	Heartwood		Pavanasasivam and Sultsnbswa, 1973

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Cycloartenyl acetate [139]	Bark	<i>A. altilis</i>	Pavanasasivam and Sultsnbswa, 1973
	Stem bark	<i>A. chaplacha</i>	Mahato, Banerjee and Chakravarti, 1971
	Bark	<i>A. heterophyllus</i>	Pavanasasivam and Sultsnbswa, 1973
	Bark	<i>A. nobilis</i>	Pavanasasivam and Sultsnbswa, 1973
	Heartwood		Pavanasasivam and Sultsnbswa, 1973
Cycloartobiloxanthone [140]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
	Bark	<i>A. communis</i>	Hano <i>et al.</i> , 1990c
	Root bark	<i>A. elasticus</i>	Ko <i>et al.</i> , 2005
	Bark	<i>A. nobilis</i>	Sultanbawa and Surendrakumar, 1989
	Bark	<i>A. rigida</i>	Hano <i>et al.</i> , 1990b
Cycloartocarpesin [141]	Heartwood	<i>A. chaplacha</i>	Rao <i>et al.</i> , 1972
	Heartwood	<i>A. heterophyllus</i>	Parthasarathy <i>et al.</i> , 1969
	Heartwood	<i>A. hirsutus</i>	Venkataraman, 1972
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compounds	Plant part	Scientific name	Reference
Cycloartocarpin [142]	Heartwood	<i>A. altilis</i>	Venkataraman, 1972
	Heartwood	<i>A. communis</i>	Han <i>et al.</i> , 2006
	Heartwood	<i>A. chaplacha</i>	Rao <i>et al.</i> , 1972
	Heartwood	<i>A. elasticus</i>	Pense <i>et al.</i> , 1976
	Heartwood	<i>A. gomezianus</i>	Venkataraman, 1972
	Heartwood	<i>A. heterophyllus</i>	Venkataraman, 1972
	Heartwood	<i>A. hirsuta</i>	Venkataraman, 1972
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
	Root	<i>A. kemando</i>	Hakim <i>et al.</i> , 2006
	Heartwood	<i>A. lakoocha</i>	Venkataraman, 1972
Cycloartocarpin A [143]	Root	<i>A. chama</i>	Wang <i>et al.</i> , 2004
	Root bark	<i>A. heterophyllus</i>	Lu and Lin, 1994
	Bark	<i>A. maingayii</i>	Hakim <i>et al.</i> , 2006
Cycloartomunin [144]	Root bark	<i>A. communis</i>	Lin and Shieh, 1991
Cycloartomunoxanthone [145]	Root bark	<i>A. communis</i>	Lin and Shieh, 1991
Cyclochampedol [146]	Bark	<i>A. champeden</i>	Achmad <i>et al.</i> , 1996
Cyclocommunin	Stem	<i>A. altilis</i>	Chen <i>et al.</i> , 1993
(Isocyclomulberrin) [147]	Root bark	<i>A. communis</i>	Lin and Shieh, 1991
Cyclocommunol [148]	Root bark	<i>A. communis</i>	Lin and Shieh, 1991
Cyclocommunomethanol [149]	Root	<i>A. communis</i>	Weng <i>et al.</i> , 2006
Cycloheterophyllin [150]	Bark	<i>A. heterophyllus</i>	Rao <i>et al.</i> , 1971
	Root bark		Hano <i>et al.</i> , 1989
	Bark	<i>A. champeden</i>	Widyawaruyanti <i>et al.</i> , 2007
Cyclointegrin [151]	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
9,19-cyclolanost-25-ene-3 $\beta$ ,24-diol (24R, 24S) [152]	Latex	<i>A. heterophyllus</i>	Barik <i>et al.</i> , 1997

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
9,19-cyclolanost-3-one-24,25-diol (24R, 24S) [153]	Latex	<i>A. heterophyllus</i>	Barik <i>et al.</i> , 1994
Cyclomorusin [154]	Stem	<i>A. altilis</i>	Chen <i>et al.</i> , 1993
	Root bark	<i>A. communis</i>	Lin and Shieh, 1991
Cyclomulberrin [155]	Stem	<i>A. altilis</i>	Chen <i>et al.</i> , 1993
	Root bark	<i>A. communis</i>	Lin and Shieh, 1991
Dihydroartomunoxanthone [156]	Root	<i>A. communis</i>	Weng <i>et al.</i> , 2006
Dihydrocycloartomunin [157]	Root bark	<i>A. communis</i>	Lin and Shieh, 1991
(-)-Dihydrofuranoartobilochromen A [158]	Trunk bark	<i>A. nobilis</i>	Kumar <i>et al.</i> , 1977
(-)-Dihydrofuranoartobilochromen B-1 [159]	Trunk bark	<i>A. nobilis</i>	Kumar <i>et al.</i> , 1977
(-)-Dihydrofuranoartobilochromen B-2 [160]	Trunk bark	<i>A. nobilis</i>	Kumar <i>et al.</i> , 1977
Dihydroisocycloartomunin [161]	Root bark	<i>A. altilis</i>	Lin and Shieh, 1992
(+)Dihydromorin [162]	Heartwood	<i>A. altilis</i>	Shimizu <i>et al.</i> , 1998
	Root bark	<i>A. communis</i>	Su <i>et al.</i> , 2002
	Heartwood	<i>A. heterophyllus</i>	Venkataraman, 1972
	Heartwood	<i>A. hirsuta</i>	Venkataraman, 1972
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
5,7-Dihydroxyflavone-3-O- $\alpha$ -L- rhamnoside [163]	Root bark	<i>A. lakoocha</i>	Chauhan and Kumari, 1979
1-Dotriacontanol [164]	Leaf	<i>A. gomezianus</i>	Kingroungpet, 1994
Engeletin [165]	Stem	<i>A. altilis</i>	Chen <i>et al.</i> , 1993
Furanoartobilochromen A [166]	Bark	<i>A. nobilis</i>	Pavanasasivum <i>et al.</i> , 1974

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Furanoartobilochromen B-1 [167]	Bark	<i>A. nobilis</i>	Pavanasasivum <i>et al.</i> , 1974
Furanoartobilochromen B-2 [168]	Bark	<i>A. nobilis</i>	Pavanasasivum <i>et al.</i> , 1974
Galangin-3-O- $\alpha$ -L-(-)- rhamnopyranoside [169]	Root bark	<i>A. lakoocha</i>	Chauhan and Kumari, 1979
Galangin-3-O- $\beta$ -D-galactopyranosyl- (1 $\rightarrow$ 4)- $\alpha$ -L-rhamnopyranoside [170]	Root bark	<i>A. lakoocha</i>	Chauhan and Kumari, 1979
3'-Geranyl-2',3,4,4'- tetrahydrochalcone [171]	Leaf	<i>A. incisus</i>	Shimizu <i>et al.</i> , 2000
	Leaf	<i>A. nobilis</i>	Jayasinghe <i>et al.</i> , 2004
	Fruit		Jayasinghe <i>et al.</i> , 2006
8-Geranyl-3',4',7-trihydroxyflavanone [172]	Fruit	<i>A. nobilis</i>	Jayasinghe <i>et al.</i> , 2006
8-Geranyl-4',5,7-trihydroxyflavanone [173]	Fruit	<i>A. nobilis</i>	Jayasinghe <i>et al.</i> , 2006
8-Geranyl-4',7-dihydroxyflavanone [174]	Fruit	<i>A. nobilis</i>	Jayasinghe <i>et al.</i> , 2006
Gemichalcone B [175]	Heartwood	<i>A. communis</i>	Han <i>et al.</i> , 2006
Gemichalcone C [176]	Heartwood	<i>A. communis</i>	Han <i>et al.</i> , 2006
Heteroartonin A [177]	Root	<i>A. heterophyllus</i>	Chung <i>et al.</i> , 1995
Heteroflavanone A [178]	Root bark	<i>A. heterophyllus</i>	Lu and Lin, 1993
Heteroflavanone B [179]	Root bark	<i>A. heterophyllus</i>	Lu and Lin, 1993
Heteroflavanone C [180]	Root bark	<i>A. heterophyllus</i>	Lu and Lin, 1994
		<i>A. champeden</i>	Widyawaruyanti <i>et al.</i> , 2007

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Heterophyllin [181]	Root bark	<i>A. heterophyllus</i>	Hano <i>et al.</i> , 1989
	Bark	<i>A. champeden</i>	Widyawarutanti <i>et al.</i> , 2007
Heterophyllol [182]	Root bark	<i>A. heterophyllus</i>	Lu and Lin, 1993
5'-Hydroxycudraflavone A [183]	Heartwood	<i>A. champeden</i>	Syah <i>et al.</i> , 2004
	Bark	<i>A. scortechinii</i>	Hakim <i>et al.</i> , 2006
5-Hydroxy-7-2'-4'-trimethoxyflavone [184]	Stem	<i>A. lakoocha</i>	Pavaro and Reutrakul, 1976
9-Hydroxytridecyldocasanoate [185]	Root bark	<i>A. heterophyllus</i>	Lu and Lin, 1994
4-Hydroxytridecyldocasanoate [186]	Latex	<i>A. heterophyllus</i>	Pant and Chaturvedi, 1989
Integrin [187]	Heartwood	<i>A. elasticus</i>	Pense <i>et al.</i> , 1976
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
Isoartocarpetin [188]	Heartwood	<i>A. incisus</i>	Shimizu <i>et al.</i> , 1998
Isobacachalcone [189]	Heartwood	<i>A. communis</i>	Han <i>et al.</i> , 2006
Isocycloartenyl acetate [190]	Bark	<i>A. chaplacha</i>	Mahato <i>et al.</i> , 1971
Isocycloheterophyllin [191]	Bark	<i>A. heterophyllus</i>	Rao <i>et al.</i> , 1973
Isonymphaeol-B [192]	Fruit	<i>A. nobilis</i>	Jayasinghe <i>et al.</i> , 2006
Jacalin [193]	Seed	<i>A. heterophyllus</i>	Hagieara <i>et al.</i> , 1988; Ferreia <i>et al.</i> , 1992
Kaempferol-3-O- $\beta$ -D-xylanopyranoside [194]	Root bark	<i>A. lakoocha</i>	Chauhan <i>et al.</i> , 1982
Kozonol C [195]	Root and bark	<i>A. bracteata</i>	Ersam <i>et al.</i> , 2002
Kuwanon C [196]	Stem	<i>A. heterophyllus</i>	Arung <i>et al.</i> , 2006
Kuwanon R [197]	Root bark	<i>A. heterophyllus</i>	Shinomiya <i>et al.</i> , 1995
Kuwanon T [198]	Root bark	<i>A. heterophyllus</i>	Shinomiya <i>et al.</i> , 1995



Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Lakoochin A [199]	Root	<i>A. lakoocha</i>	Puntumchai <i>et al.</i> , 2004
Lakoochin B [200]	Root	<i>A. lakoocha</i>	Puntumchai <i>et al.</i> , 2004
Lespeol [201]	Fruit	<i>A. nobilis</i>	Jayasinghe <i>et al.</i> , 2006
Lupeol [202]	Root bark	<i>A. lakoocha</i>	Chauhan and Kumari, 1979
Lupeol acetate [203]	Root bark	<i>A. altilis</i>	Shieh and Lin, 1992
	Stem bark	<i>A. chaplacha</i>	Mahato <i>et al.</i> , 1971
	Latex	<i>A. elasticus</i>	Utee, 1949
	Leaf	<i>A. gomezianus</i>	Kingroungpet, 1994
	Bark	<i>A. heterophyllus</i>	Kapil and Joshi, 1960
	Bark	<i>A. lakoocha</i>	Kapil and Joshi, 1960
Lymphoagglutinin [204]	Seed	<i>A. heterophyllus</i>	Arora <i>et al.</i> , 1987
	Seed	<i>A. hirsuta</i>	Arora <i>et al.</i> , 1987
	Seed	<i>A. lakoocha</i>	Arora <i>et al.</i> , 1987
Mesoerythritol [205]	Leaf	<i>A. gomezianus</i>	Venkataraman, 1972
Morachalcone A [206]	Heartwood	<i>A. communis</i>	Han <i>et al.</i> , 2006
Morin [207]	Heartwood	<i>A. altilis</i>	Venkataraman, 1972
	Heartwood	<i>A. gomezianus</i>	Venkataraman, 1972
	Heartwood	<i>A. heterophyllus</i>	Radhakrishnan <i>et al.</i> , 1965;
	Heartwood	<i>A. hirsuta</i>	Parthasarathy <i>et al.</i> ,
	Heartwood	<i>A. integer</i>	1969; Venkataraman, 1972 Pendse <i>et al.</i> , 1976

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Morusin [208]	Bark	<i>A. communis</i>	Fujimoto <i>et al.</i> , 1990
Mulberrin [209]	Bark	<i>A. fretessi</i>	Soekamto <i>et al.</i> , 2003
Mulberrochromene [210]	Bark	<i>A. fretessi</i>	Soekamto <i>et al.</i> , 2003
(+)-Norartocarpanone [211]	Heartwood	<i>A. incisus</i>	Shimizu <i>et al.</i> , 1998
Norartocarpetin [212]	Heartwood	<i>A. altilis</i>	Venkataraman, 1972
	Heartwood	<i>A. dadah</i>	Su <i>et al.</i> , 2002
	Root bark	<i>A. fretessi</i>	Soekamto <i>et al.</i> , 2003
	Heartwood	<i>A. gomezianus</i>	Venkataraman, 1972
	Heartwood	<i>A. heterophyllus</i>	Radhakrishnan <i>et al.</i> , 1965
	Heartwood	<i>A. hirsuta</i>	Venkataraman, 1972
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
	Root bark	<i>A. kemando</i>	Hakim <i>et al.</i> , 2006
Norartocarpin [213]	Heartwood	<i>A. elasticus</i>	Pense <i>et al.</i> , 1976
	Heartwood	<i>A. heterophyllus</i>	Venkataraman, 1972
	Heartwood	<i>A. lakoocha</i>	Venkataraman, 1972
Norcycloartocarpin [214]	Heartwood	<i>A. lakoocha</i>	Venkataraman, 1972
Oxydihydroartocarpesin [215]	Heartwood	<i>A. heterophyllus</i>	Pathasarathy <i>et al.</i> , 1969
	Heartwood	<i>A. hirsuta</i>	1969
	Heartwood	<i>A. integer</i>	Venkataraman, 1972
Oxydihydromorusin [216]	Bark	<i>A. rigida</i>	Hano, Inami and
			Nomura, 1990b
Oxyisocyclointegrin [217]	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
Oxyresveratrol [218]	Heartwood	<i>A. chaplacha</i>	Rao <i>et al.</i> , 1972
	Bark	<i>A. gomezianus</i>	Hakim <i>et al.</i> , 2002b
	Heartwood		Likhitwitayawuid <i>et al.</i> , 2000
	Heartwood	<i>A. lakoocha</i>	Venkataraman, 1972; Mongolsuk, Robertson and Towers, 1957
4-Prenyloxyresceratrol [219]	Heartwood	<i>A. incisus</i>	Shimizu <i>et al.</i> , 1997
Procyanidin B-3 [220]	Leaf	<i>A. heterophyllus</i>	An <i>et al.</i> , 1992
Procyanidin C-1 [221]	Leaf	<i>A. heterophyllus</i>	An <i>et al.</i> , 1992
Querectin-3-O- $\alpha$ -L-rhamnopyranoside [222]	Root bark	<i>A. lakoocha</i>	Chauhan <i>et al.</i> , 1982
Recinoleic acid [223]	Seed oil	<i>A. heterophyllus</i>	Daulatabad and Mirajkar, 1989
Resorcinol [224]	Heartwood	<i>A. chaplacha</i>	Rao <i>et al.</i> , 1972
	Root	<i>A. gomezianus</i>	Sritularrak, 1998
	Heartwood	<i>A. lakoocha</i>	Venkataraman, 1972
$\beta$ -Resorcyaldehyde [225]	Heartwood	<i>A. chaplacha</i>	Rao <i>et al.</i> , 1972
Resveratrol [226]	Heartwood	<i>A. chaplacha</i>	Rao <i>et al.</i> , 1972
	Root	<i>A. gomezianus</i>	Likhitwitayawuid <i>et al.</i> , 2000
	Heartwood	<i>A. lakoocha</i>	Venkataraman, 1972
Rubraflavone C [227]	Root bark	<i>A. rigida</i>	Lu <i>et al.</i> , 2002
Simiarenol [228]	Leaf	<i>A. gomezianus</i>	Kingroungpet, 1994
Stigmasterol [229]	Root	<i>A. gomezianus</i>	Sritularak, 1998
	Heartwood	<i>A. lakoocha</i>	Wetwitayaklung, 1994

Table 2.3 Chemical constituents of *Artocarpus* spp. (continued)

Compound	Plant part	Scientific name	Reference
$\beta$ -sitosterol [230]	Root bark	<i>A. altilis</i>	Shieh and Lin, 1992
	Stem bark	<i>A. chaplacha</i>	Mahato <i>et al.</i> , 1971
	Heartwood	<i>A. elasticus</i>	Pendse <i>et al.</i> , 1976
	Leaf	<i>A. gomezianus</i>	Kingroungpet, 1994
	Heartwood	<i>A. heterophyllus</i>	Pathasarathy <i>et al.</i> , 1969
	Root		Dayal and Seshadri, 1974
	Root bark		Lu and Lin, 1994
	Heartwood	<i>A. integer</i>	Pendse <i>et al.</i> , 1976
	Root bark	<i>A. lakoocha</i>	Chauhan and Kumari, 1979
Ursolic acid [231]	Root	<i>A. heterophyllus</i>	Dayal and Seshadri, 1974
	Root bark		Lu and Lin, 1994
Xanthoangelol [232]	Leaf	<i>A. nobilis</i>	Jayasinghe <i>et al.</i> , 2004
	Fruit		Jayasinghe <i>et al.</i> , 2006
Xanthoangelol B [233]	Leaf	<i>A. nobilis</i>	Jayasinghe <i>et al.</i> , 2004
	Fruit		Jayasinghe <i>et al.</i> , 2006

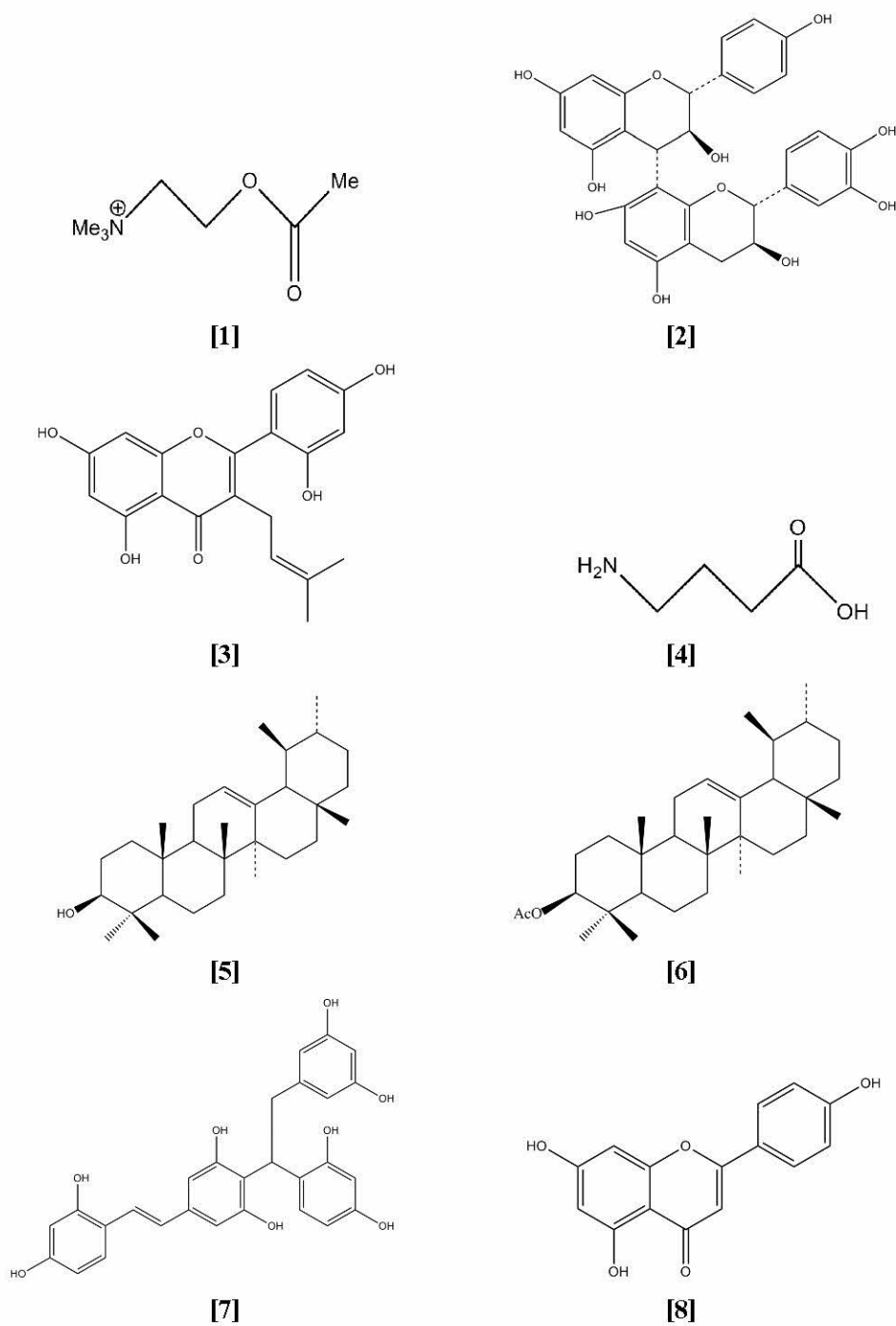


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3

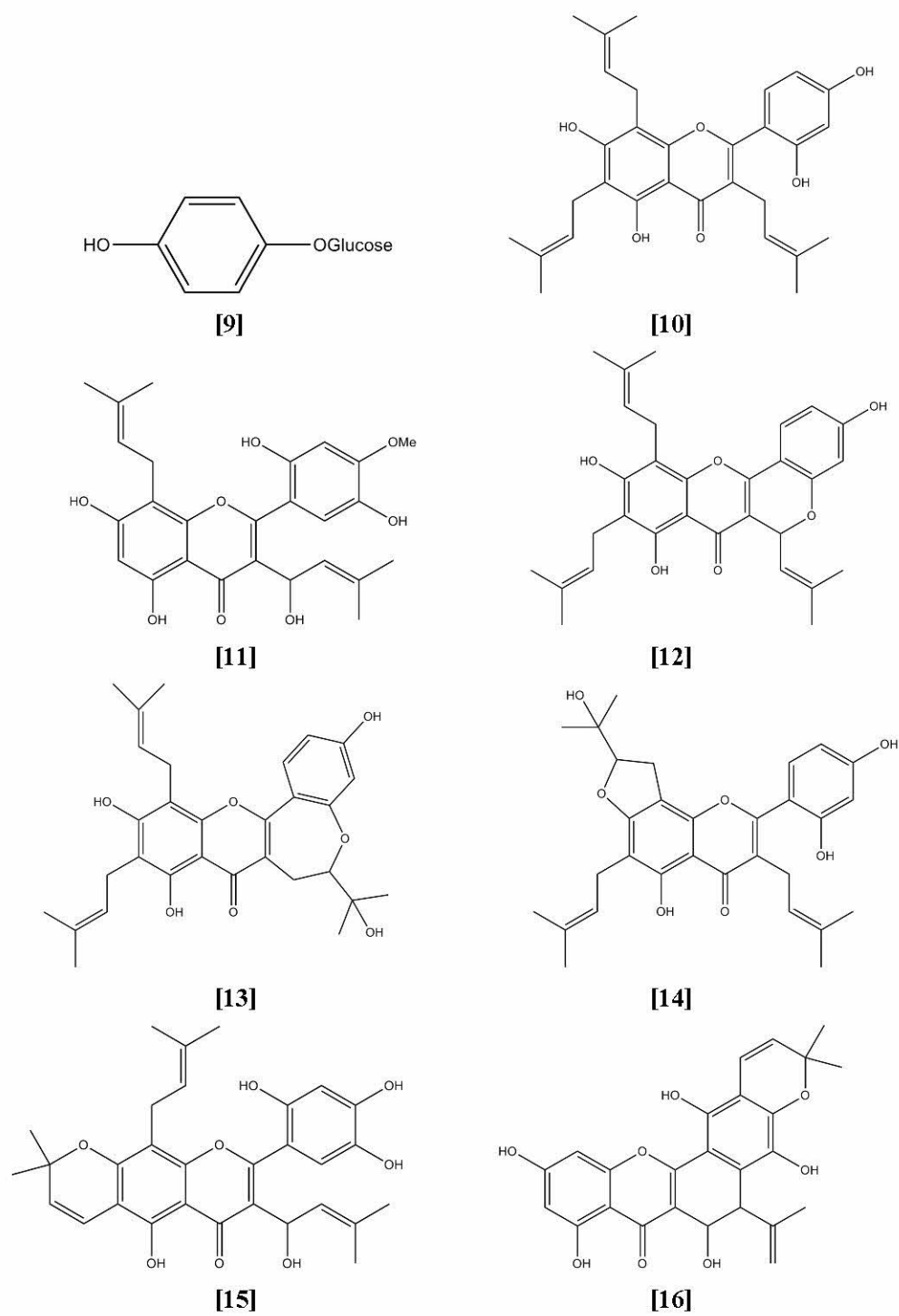
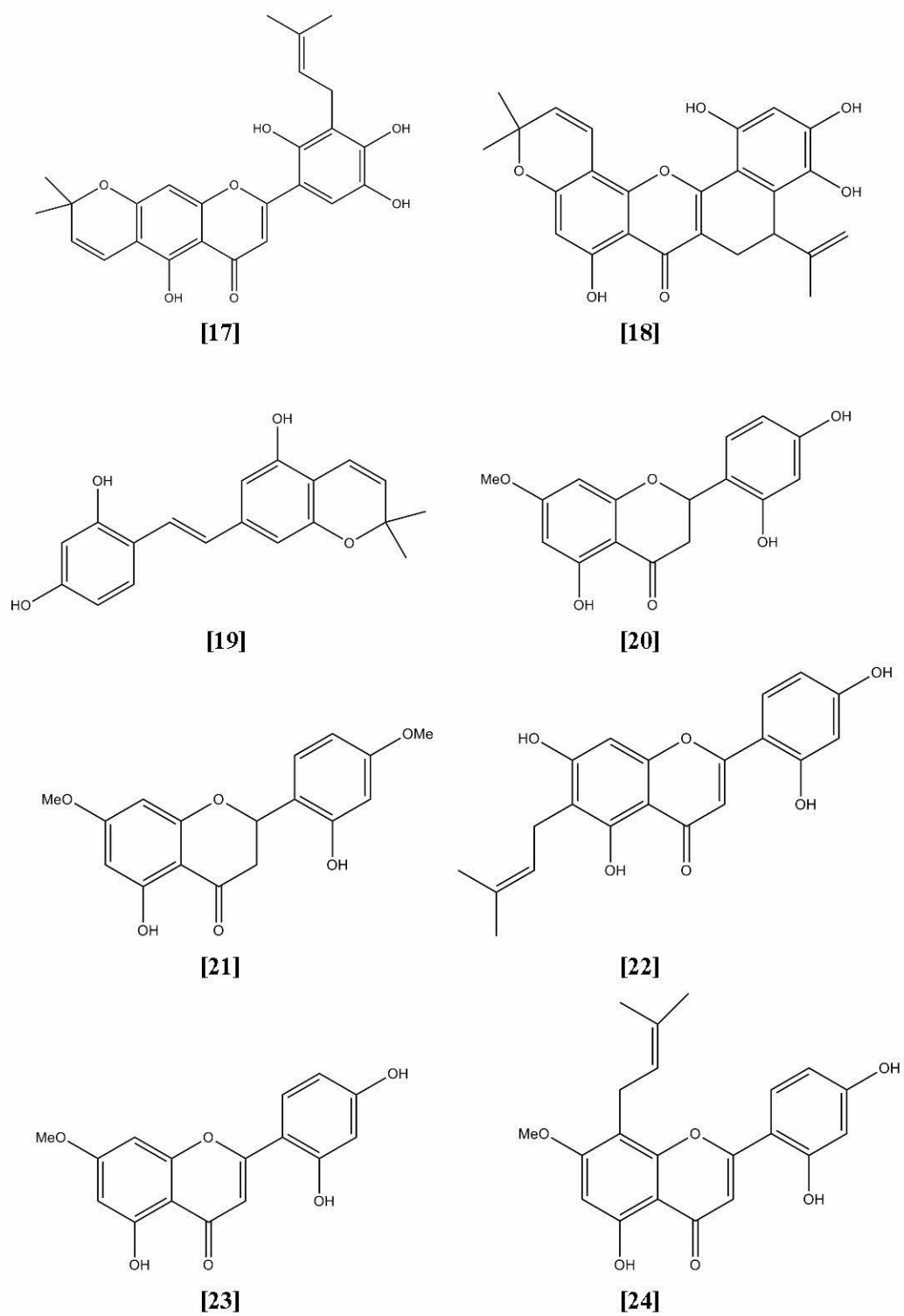


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)



**Figure 2.1** Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

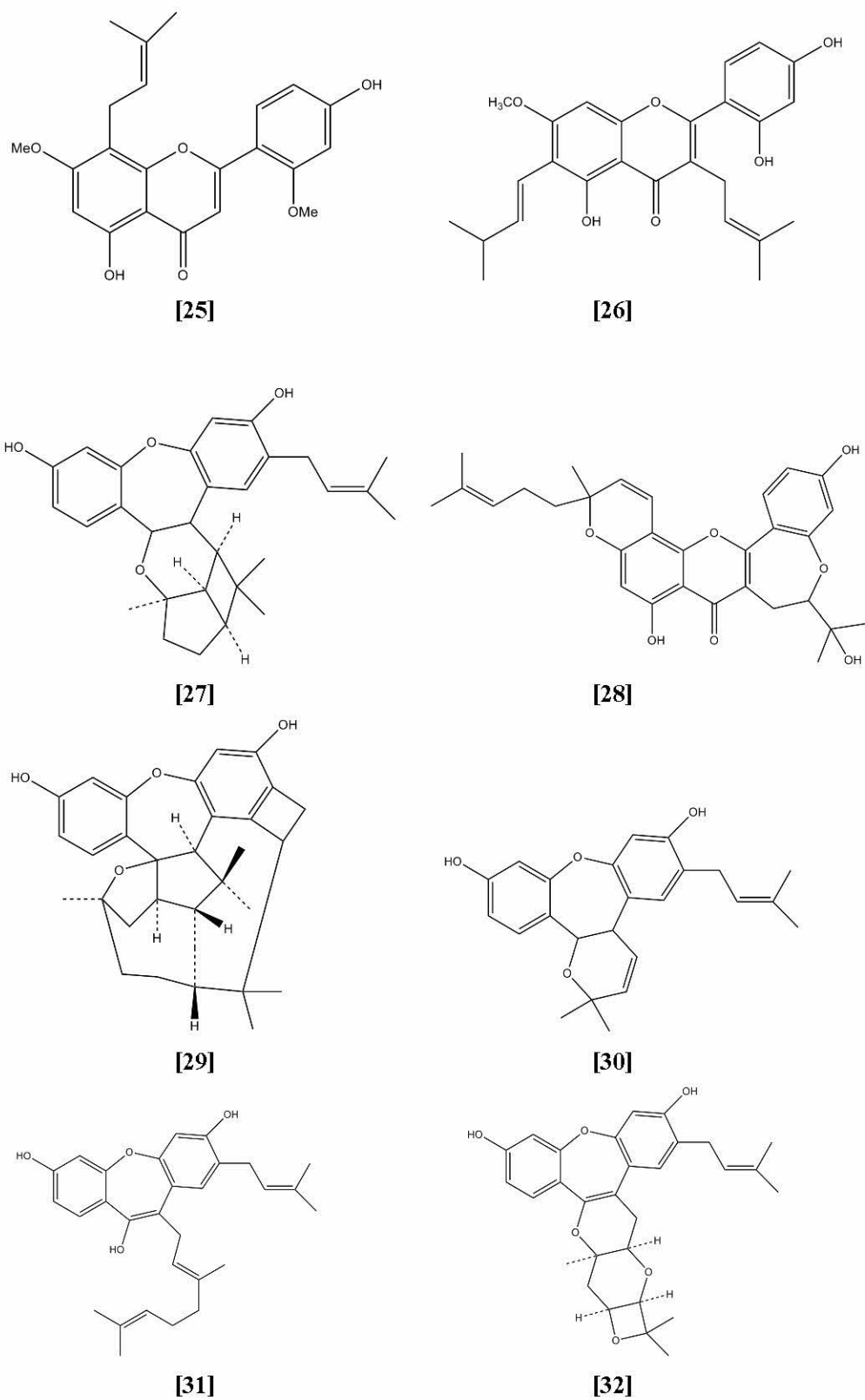


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)



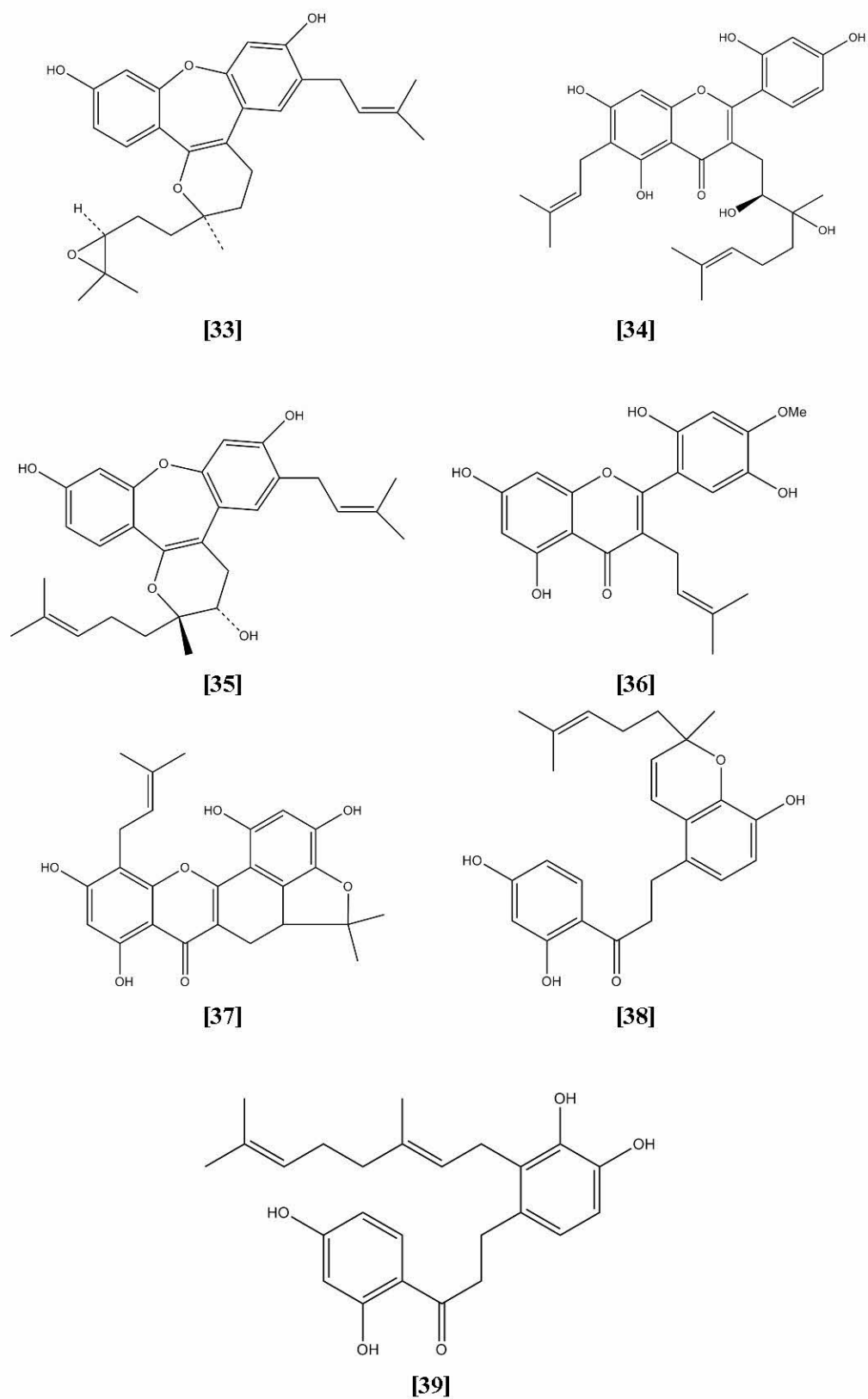


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

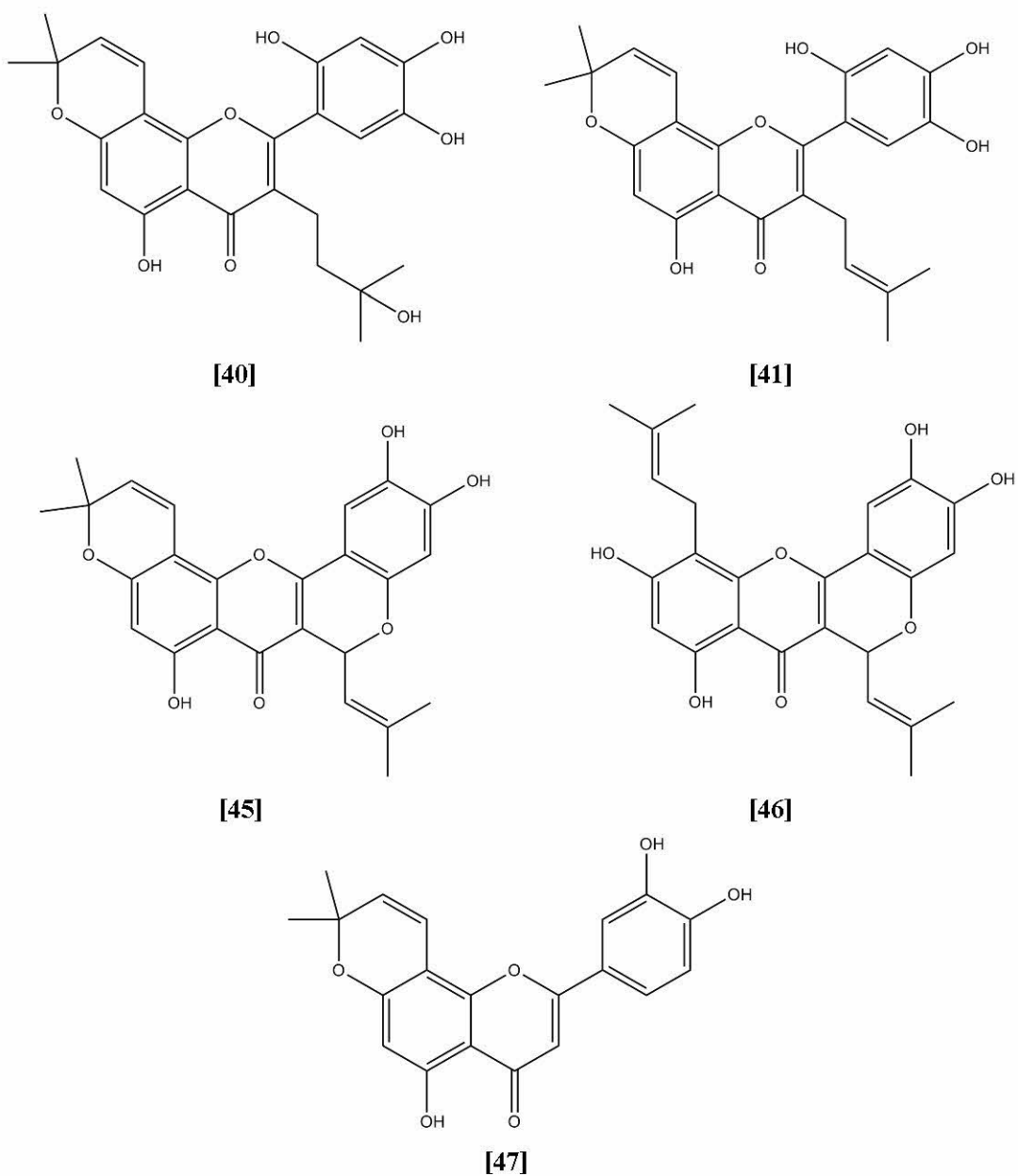


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

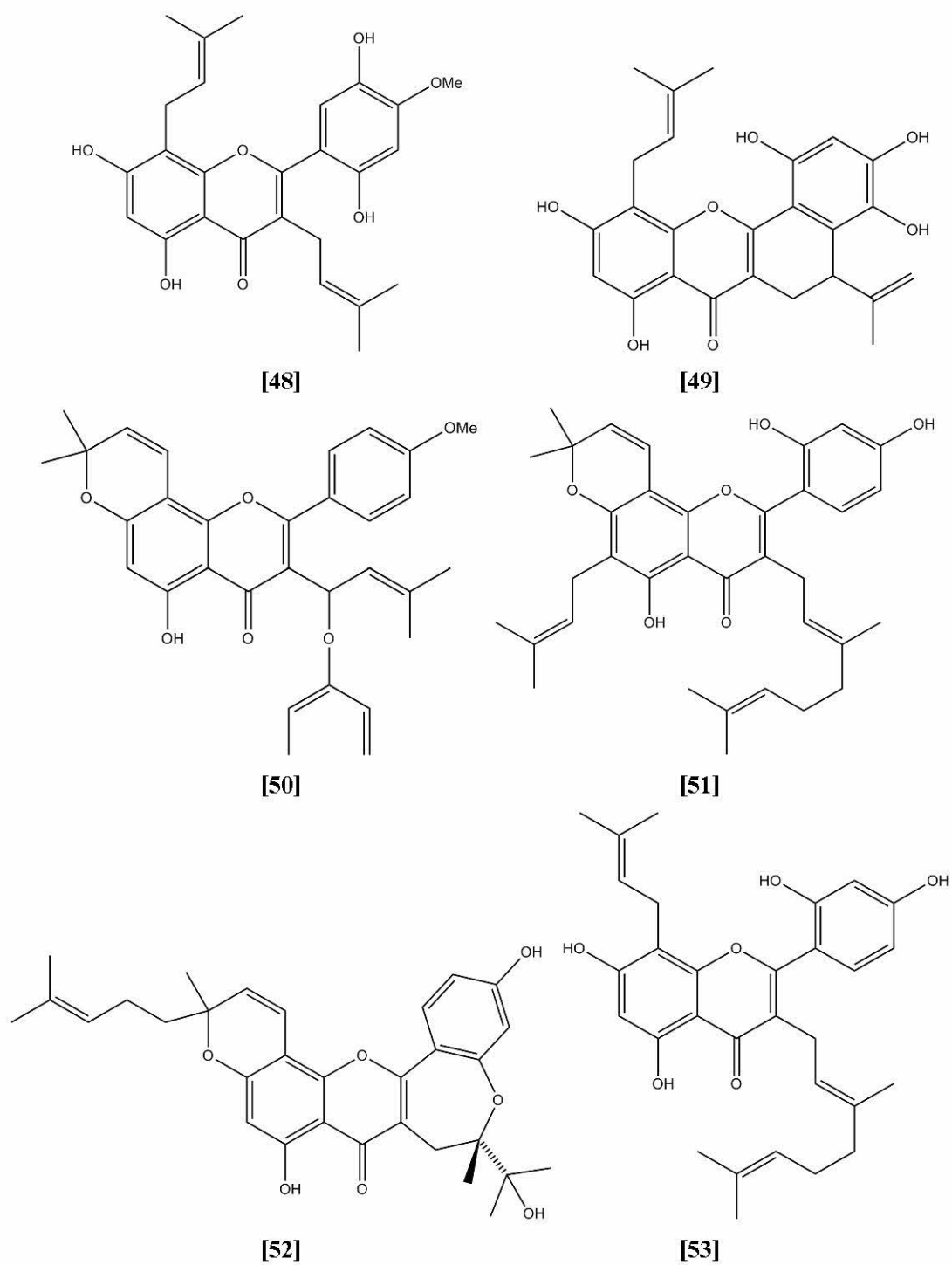


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

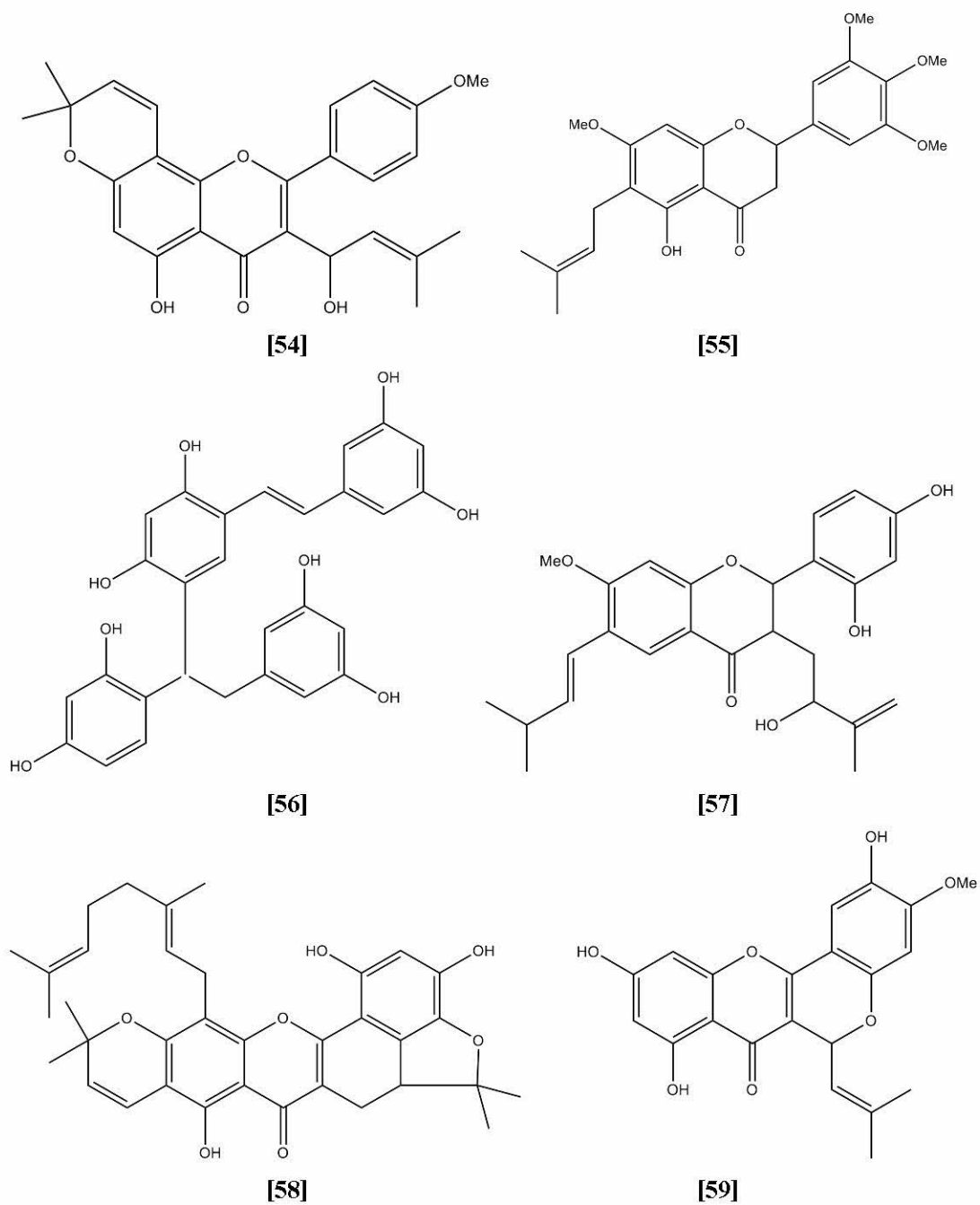
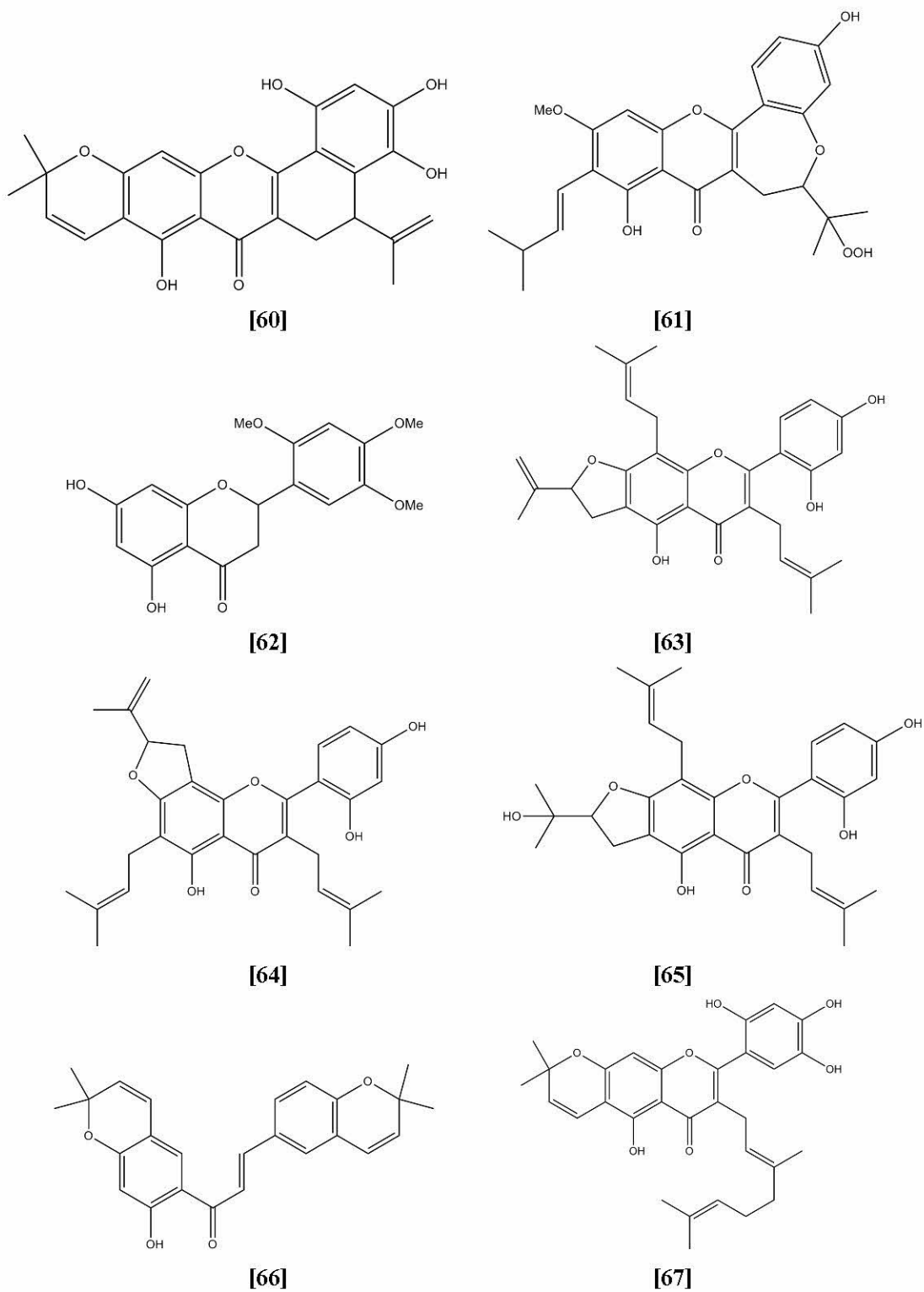


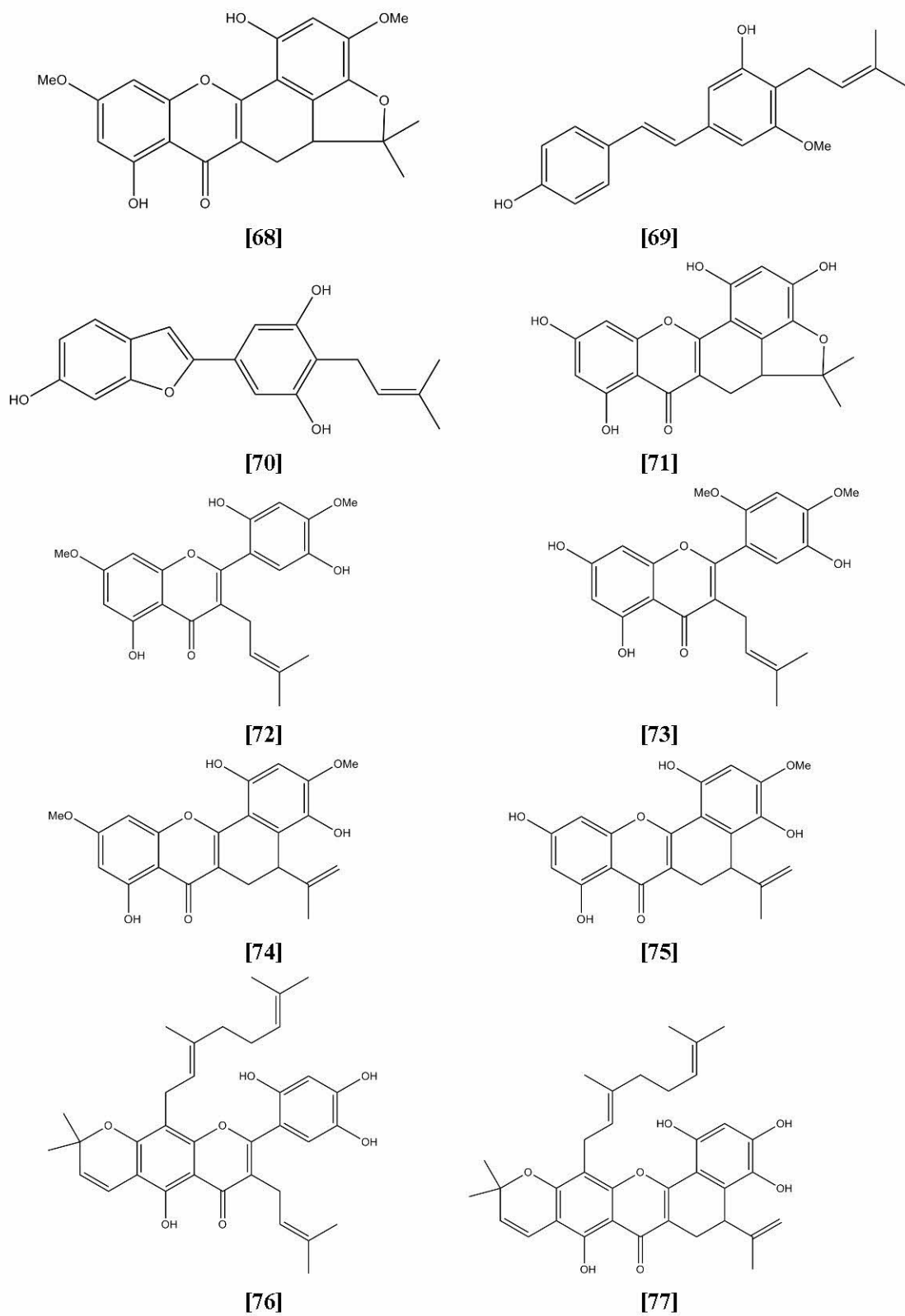
Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)



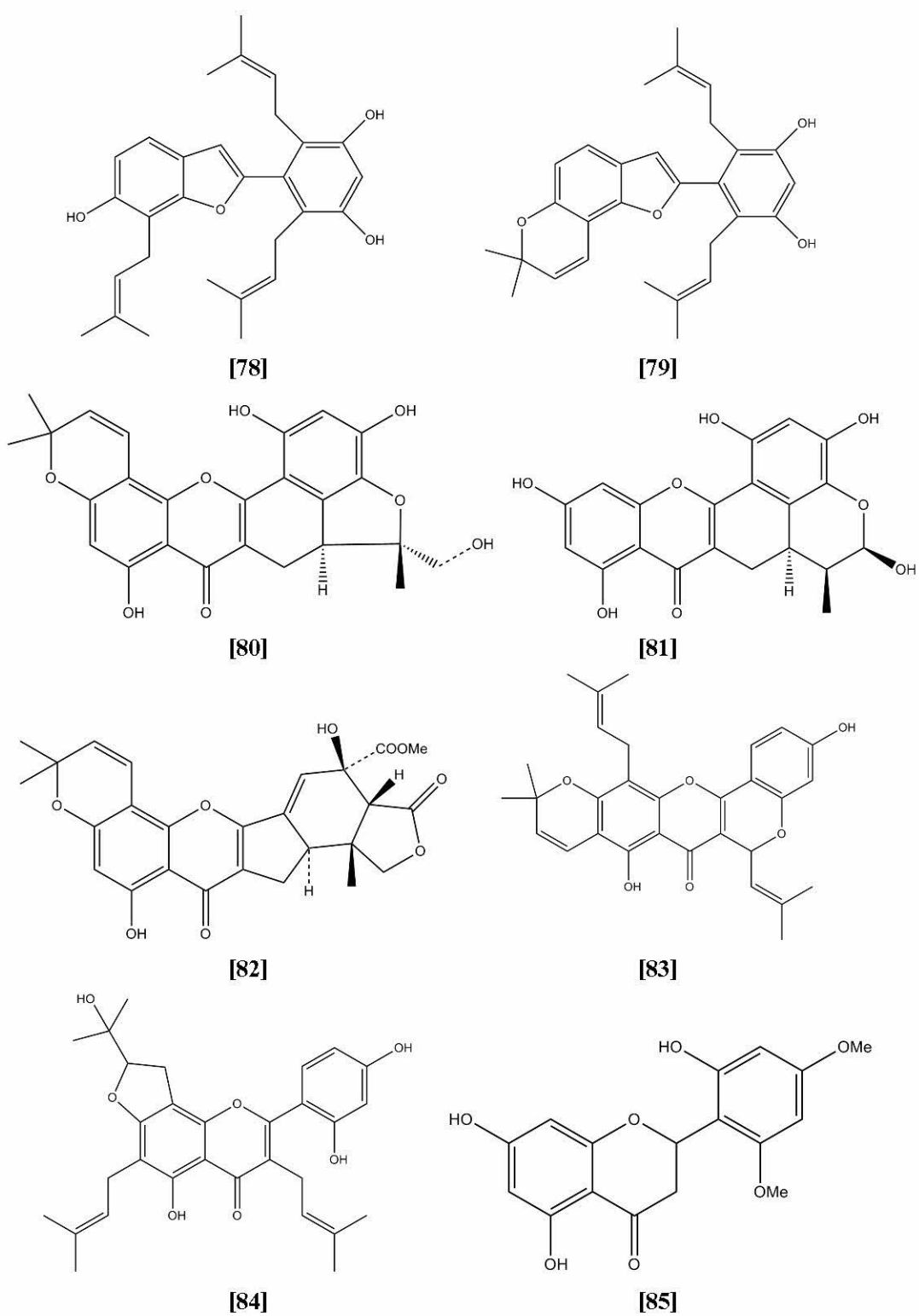
**Figure 2.1** Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)



**Figure 2.1** Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)



**Figure 2.1** Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

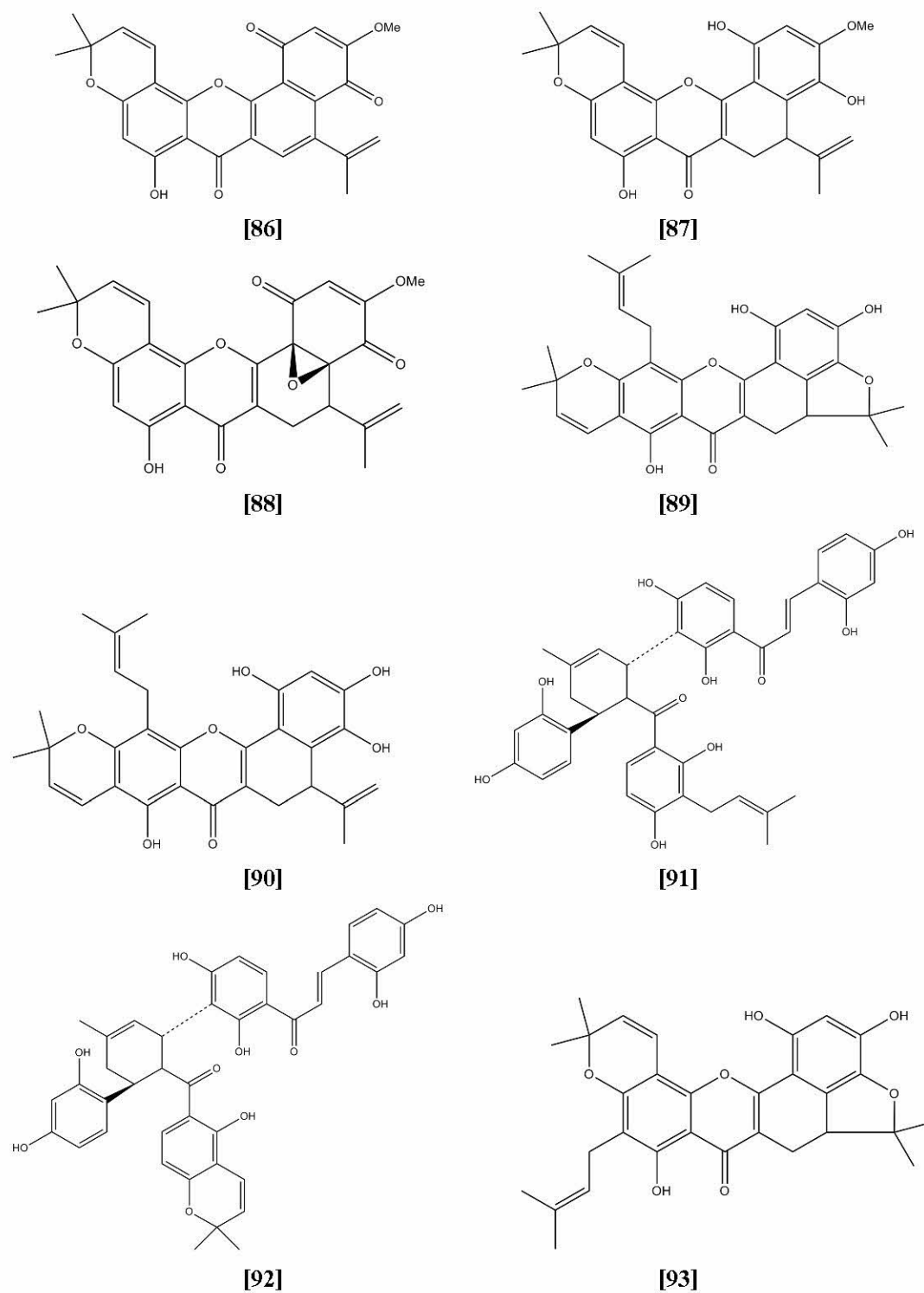


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)



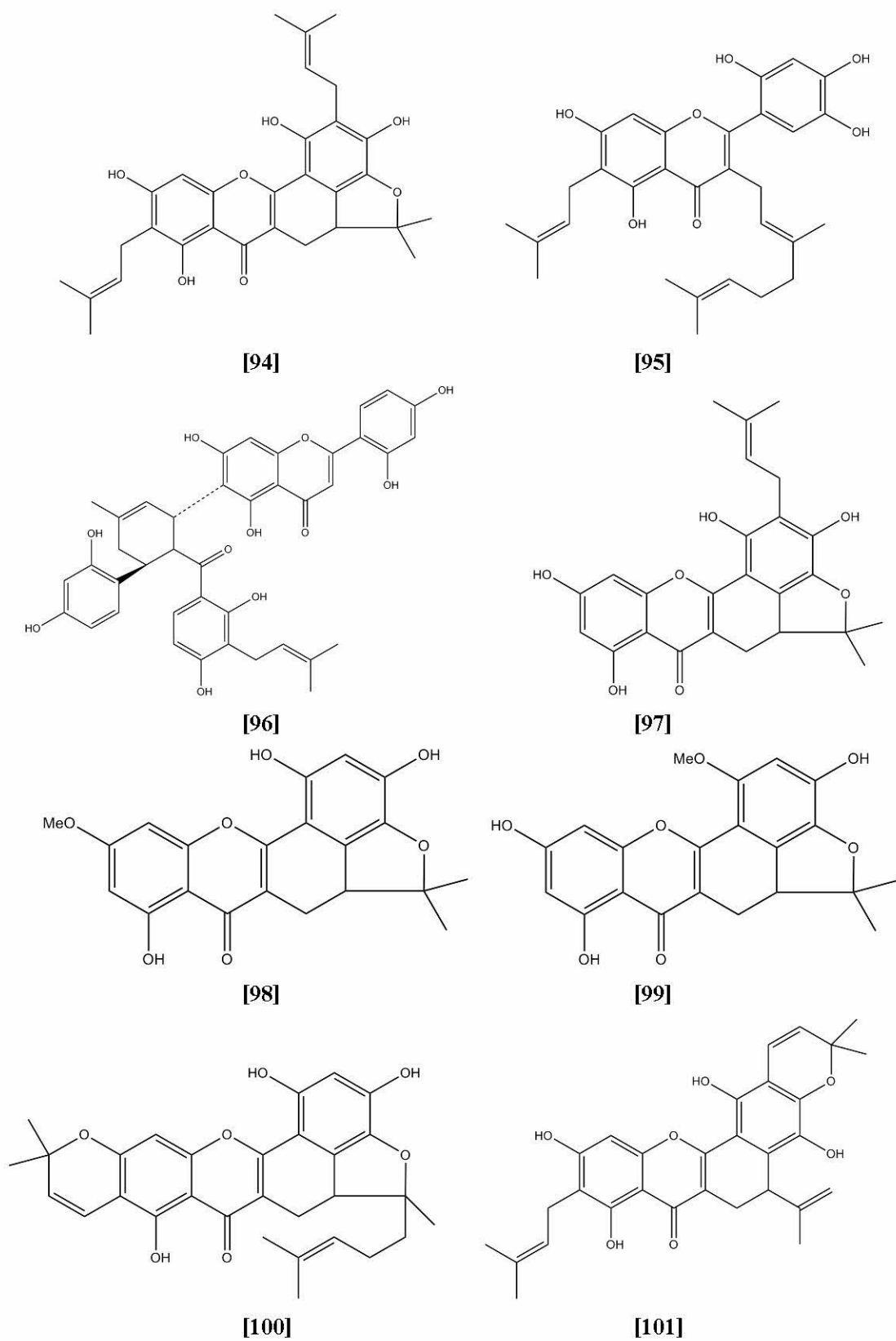


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

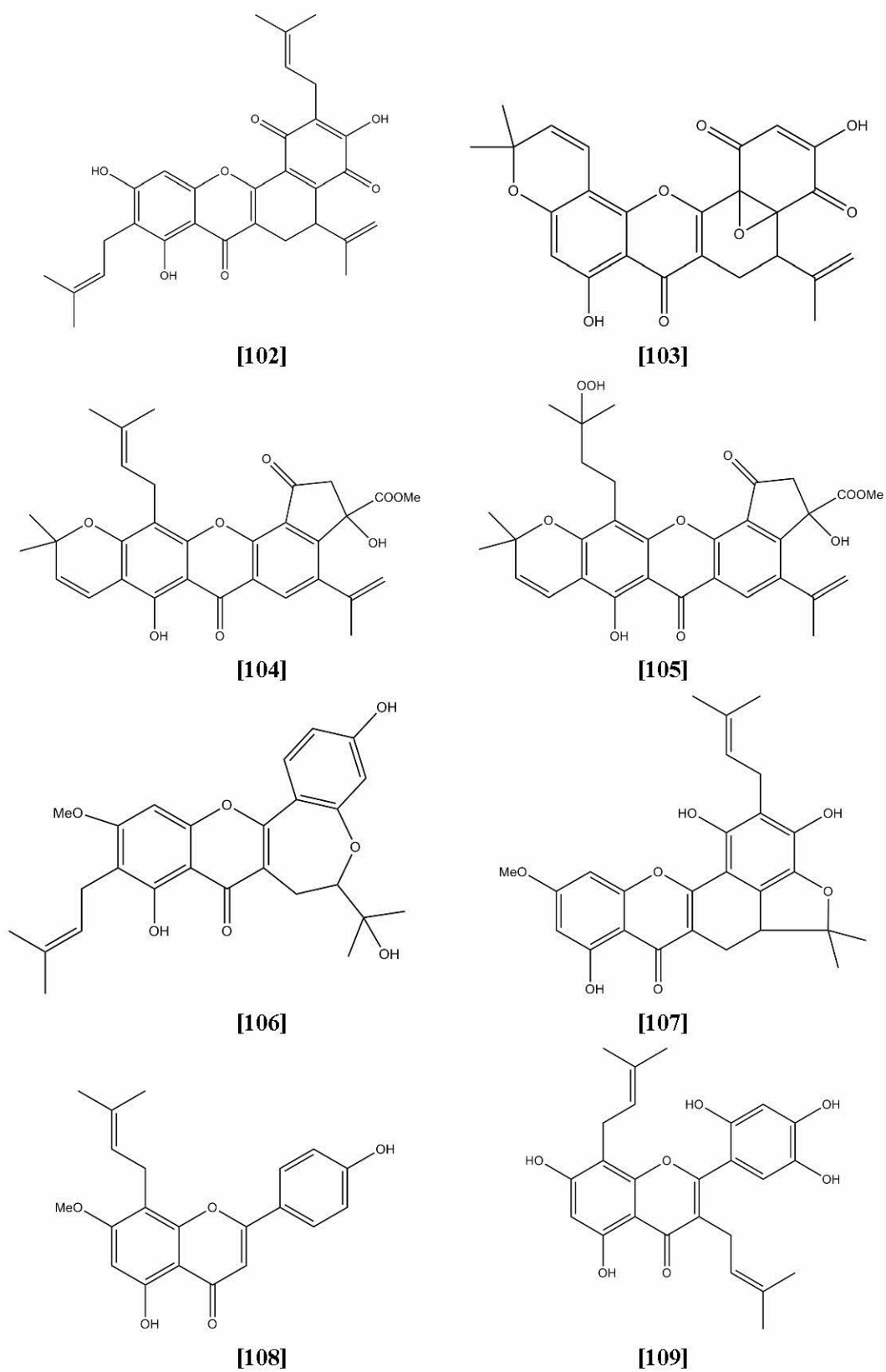
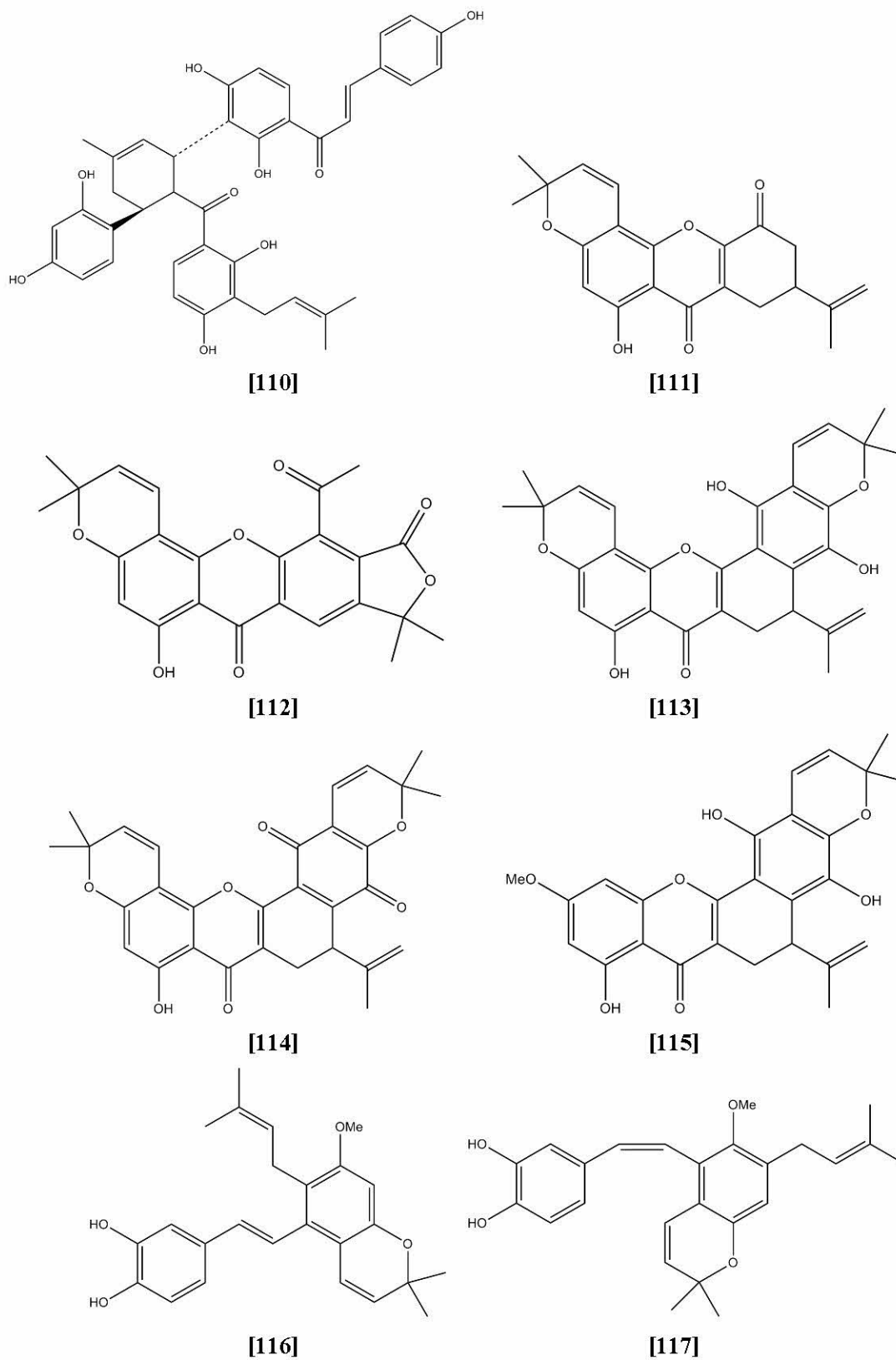


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)



**Figure 2.1** Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

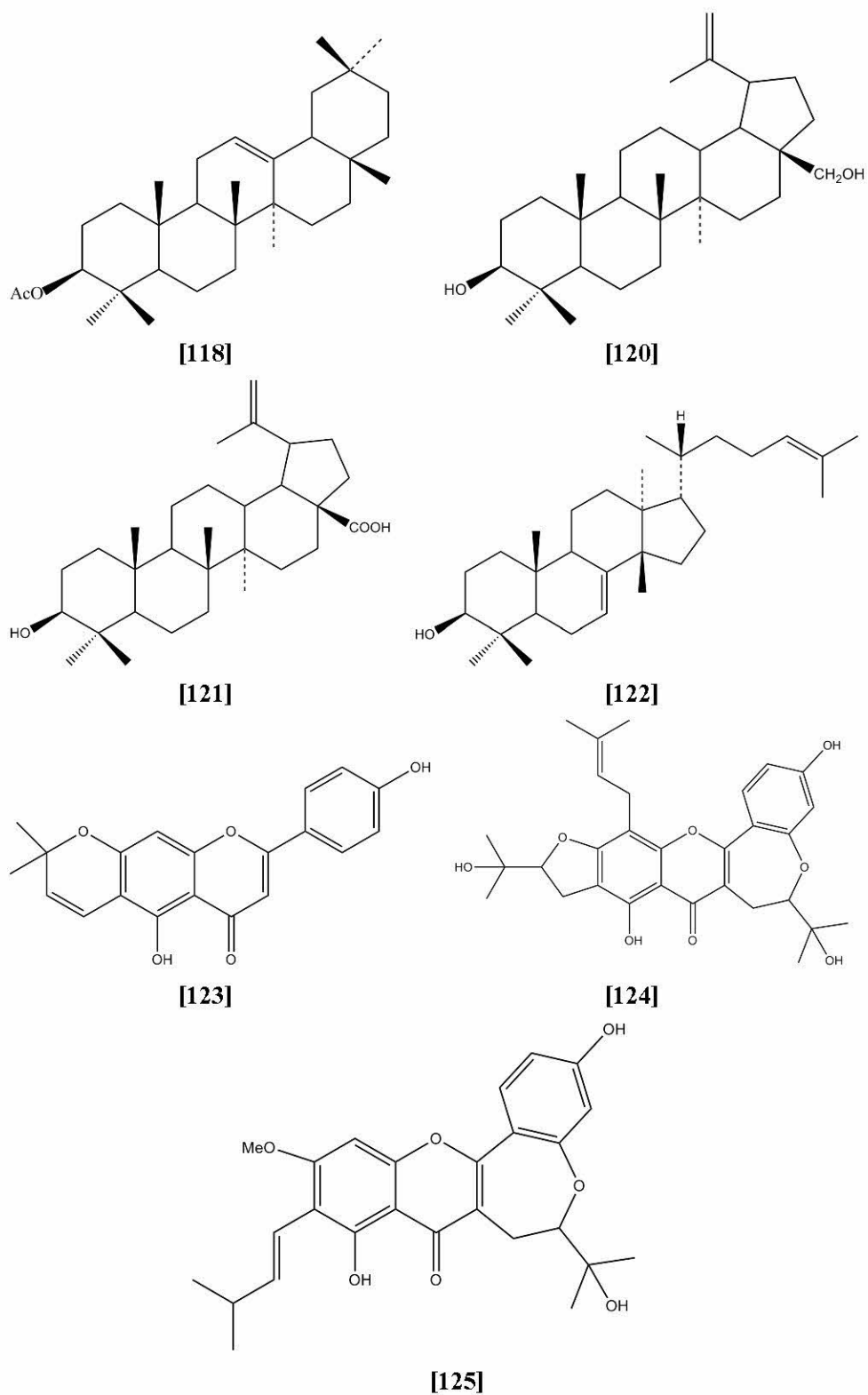


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

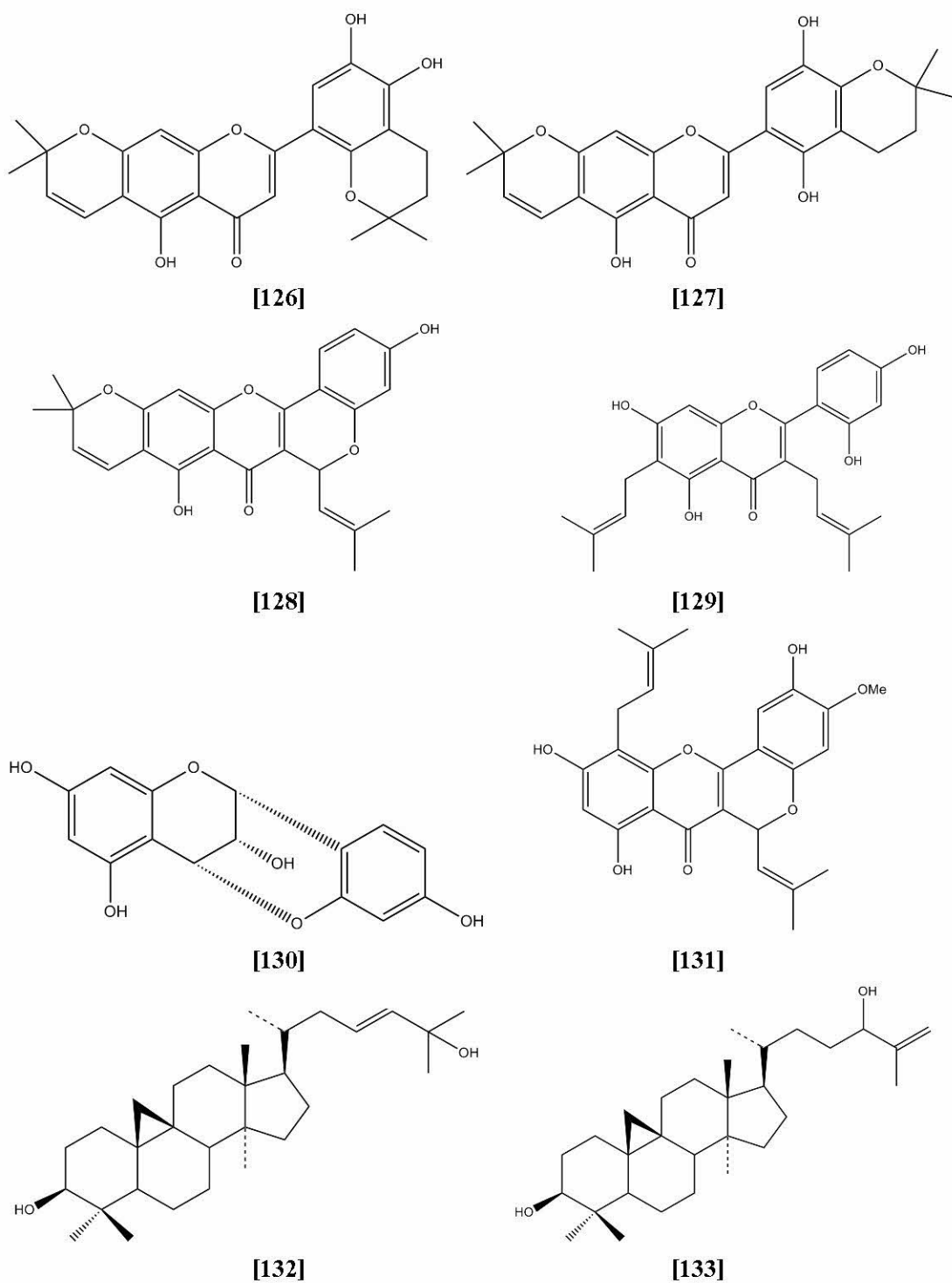


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

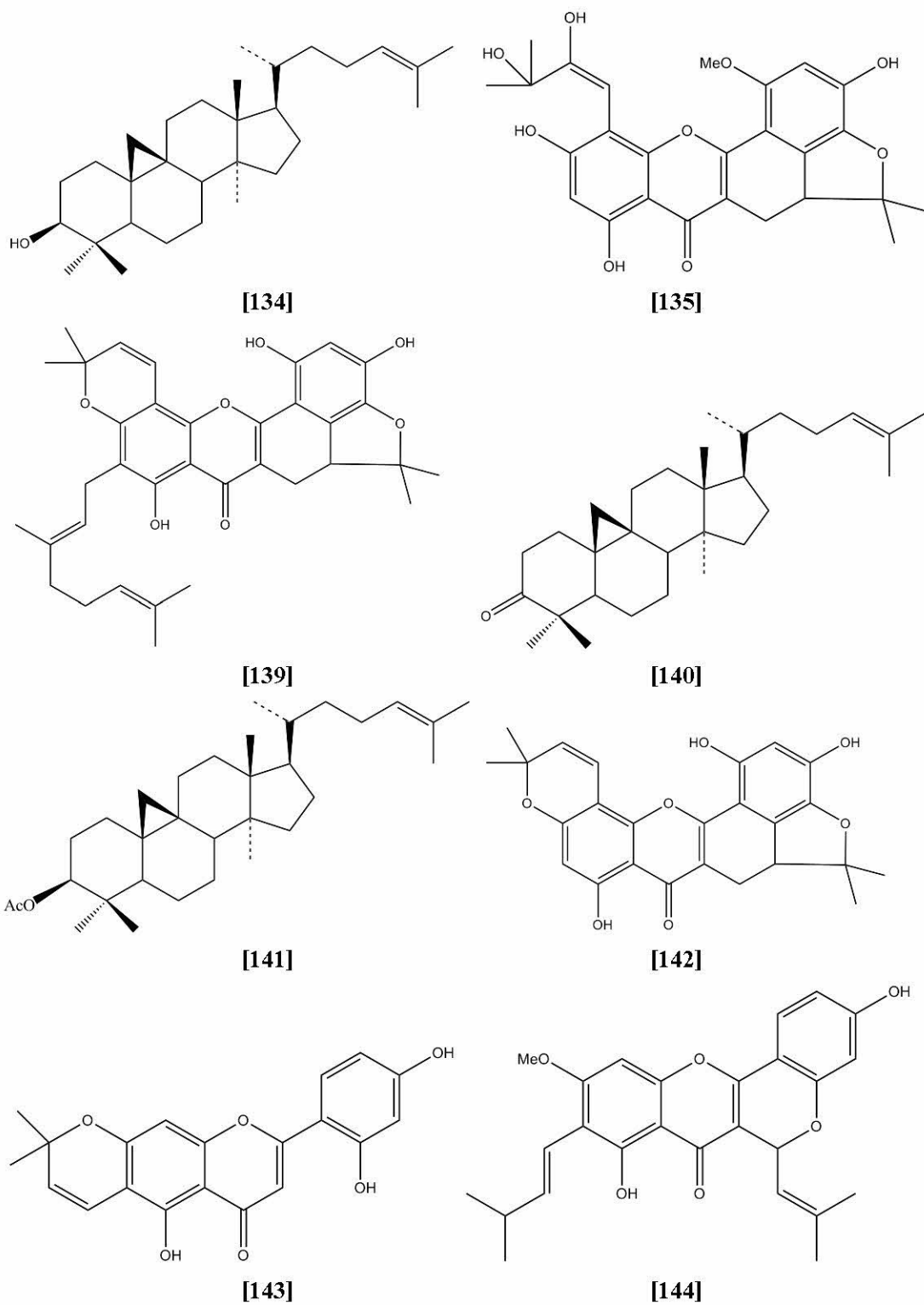


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

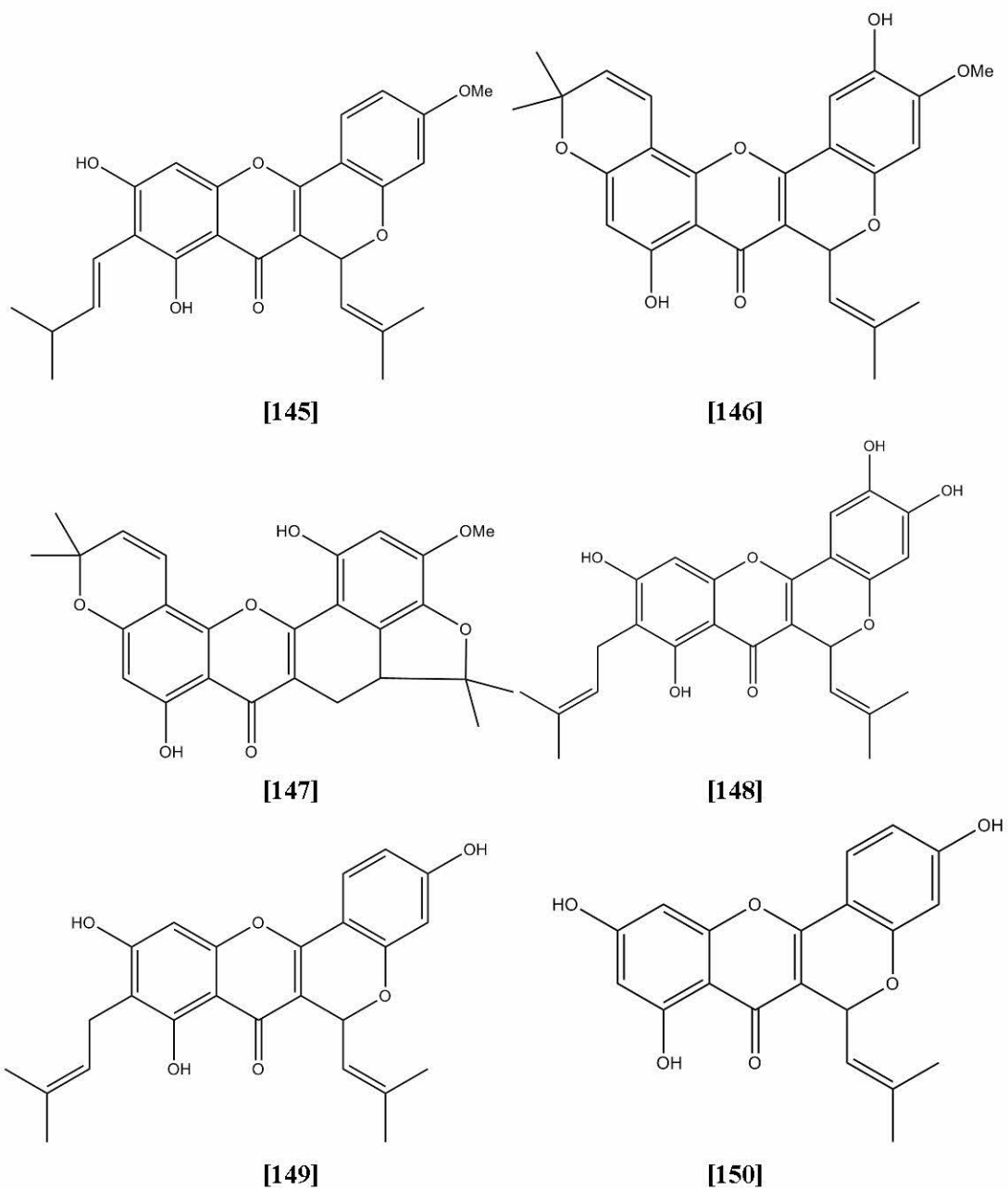
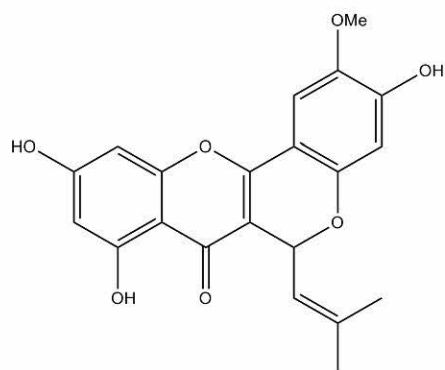
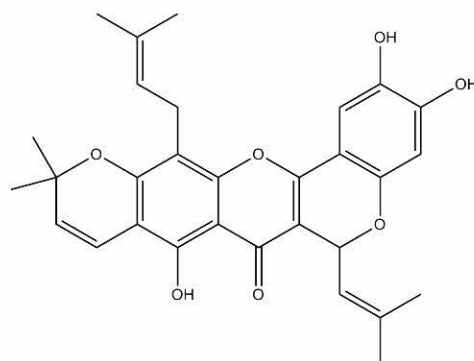


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

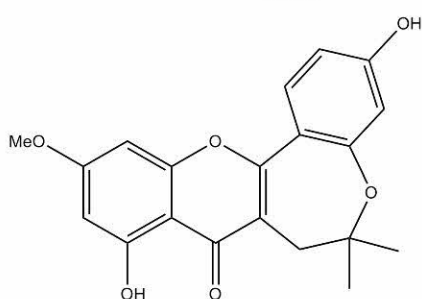
Numbers of these structures are related to table 2.3 (continued)



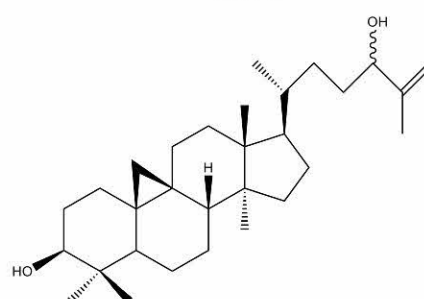
[151]



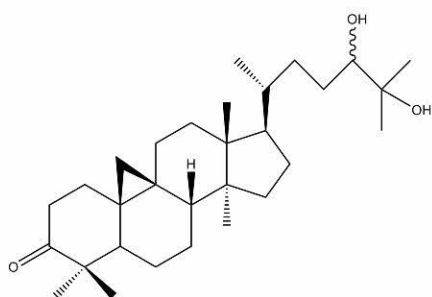
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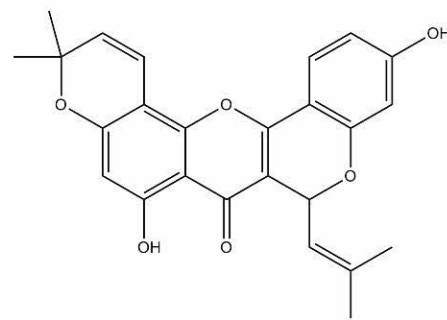
[153]



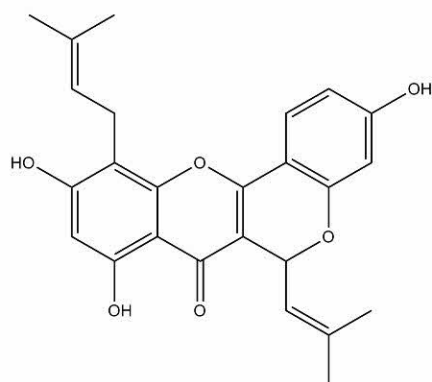
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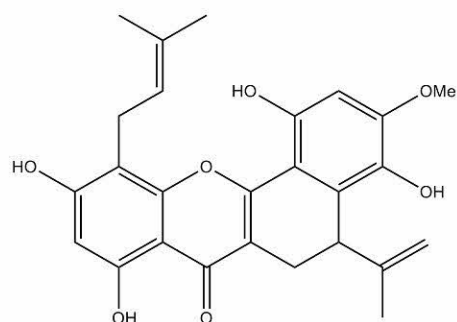
[155]



[156]



[157]



[158]

Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)



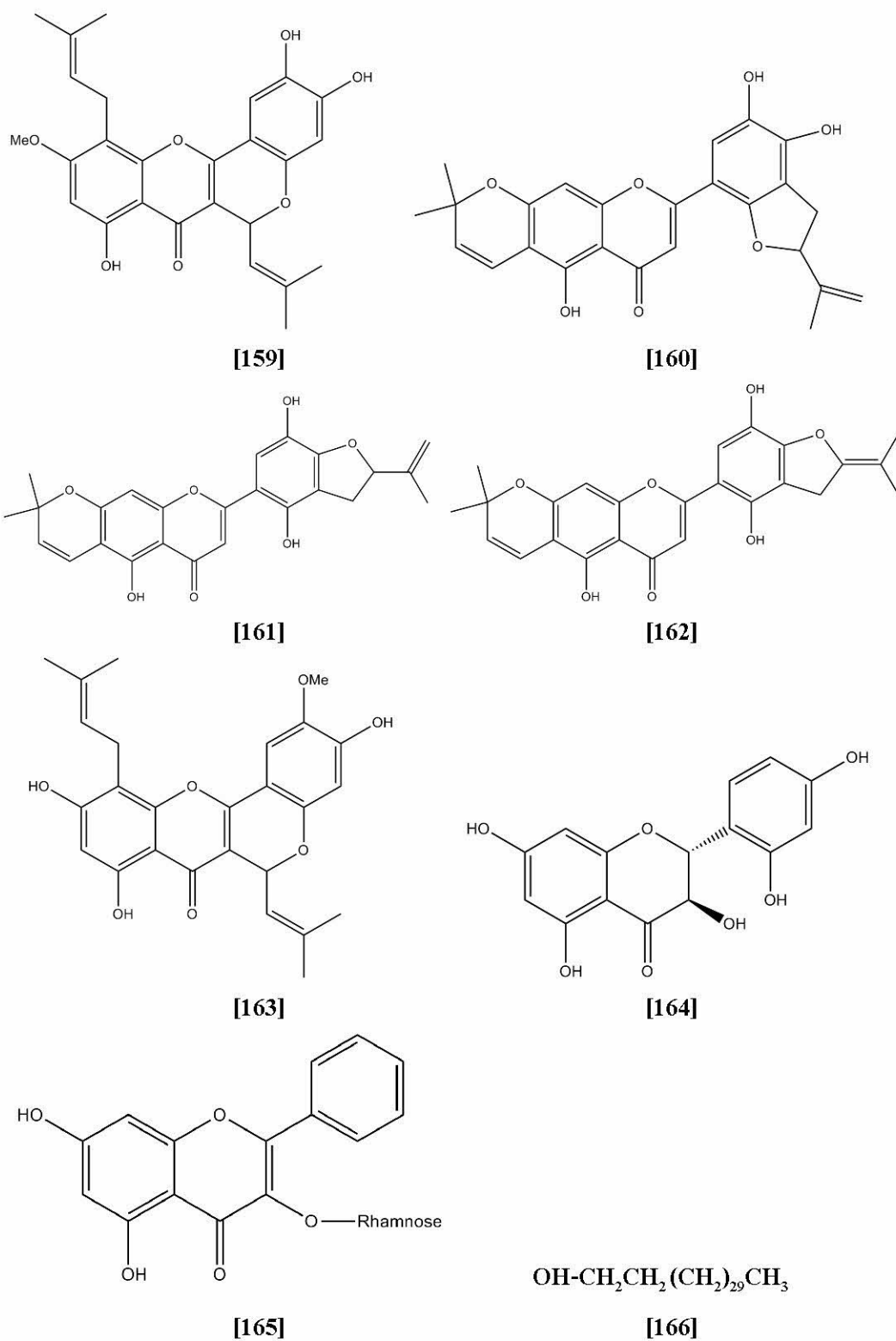


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

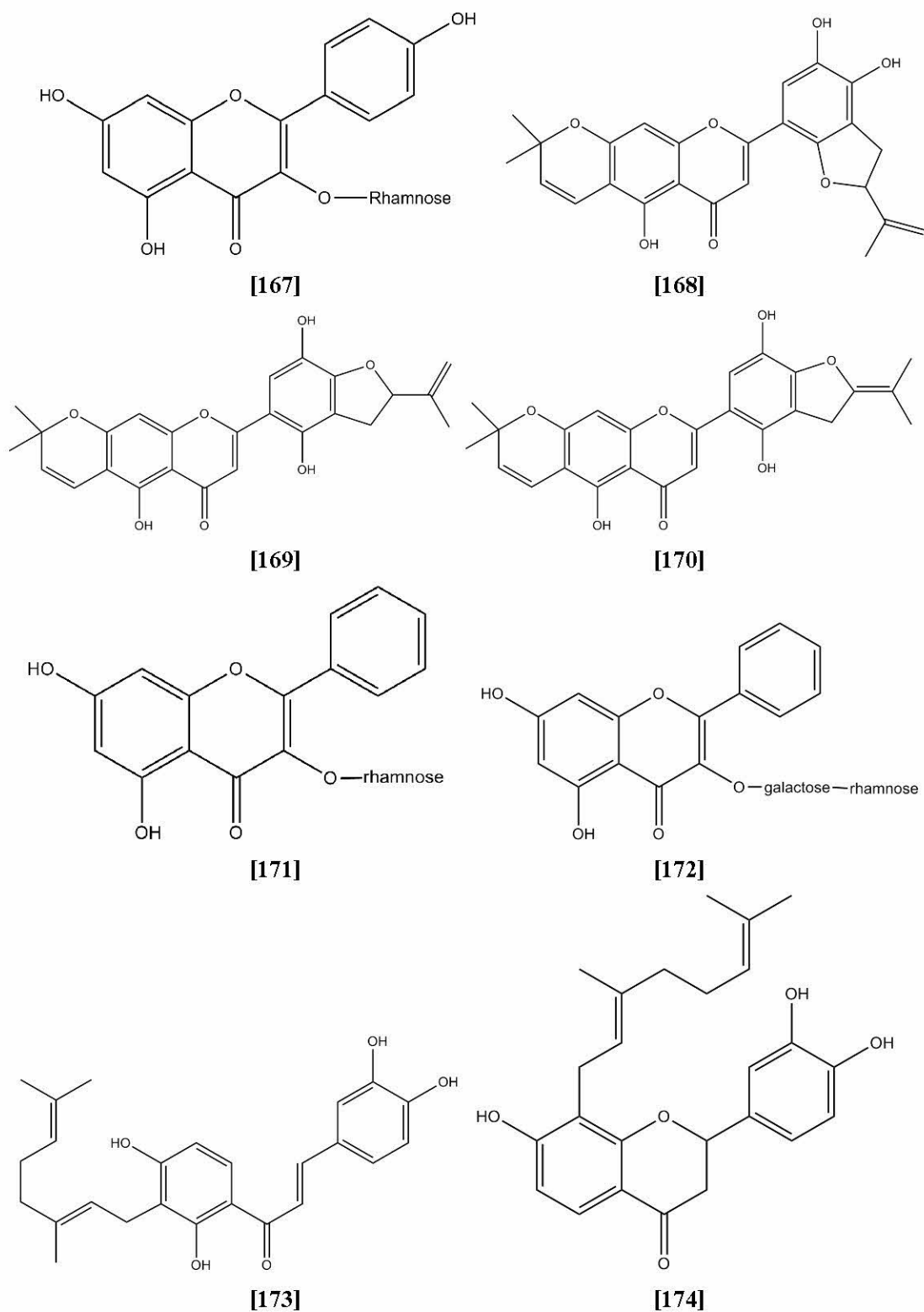


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

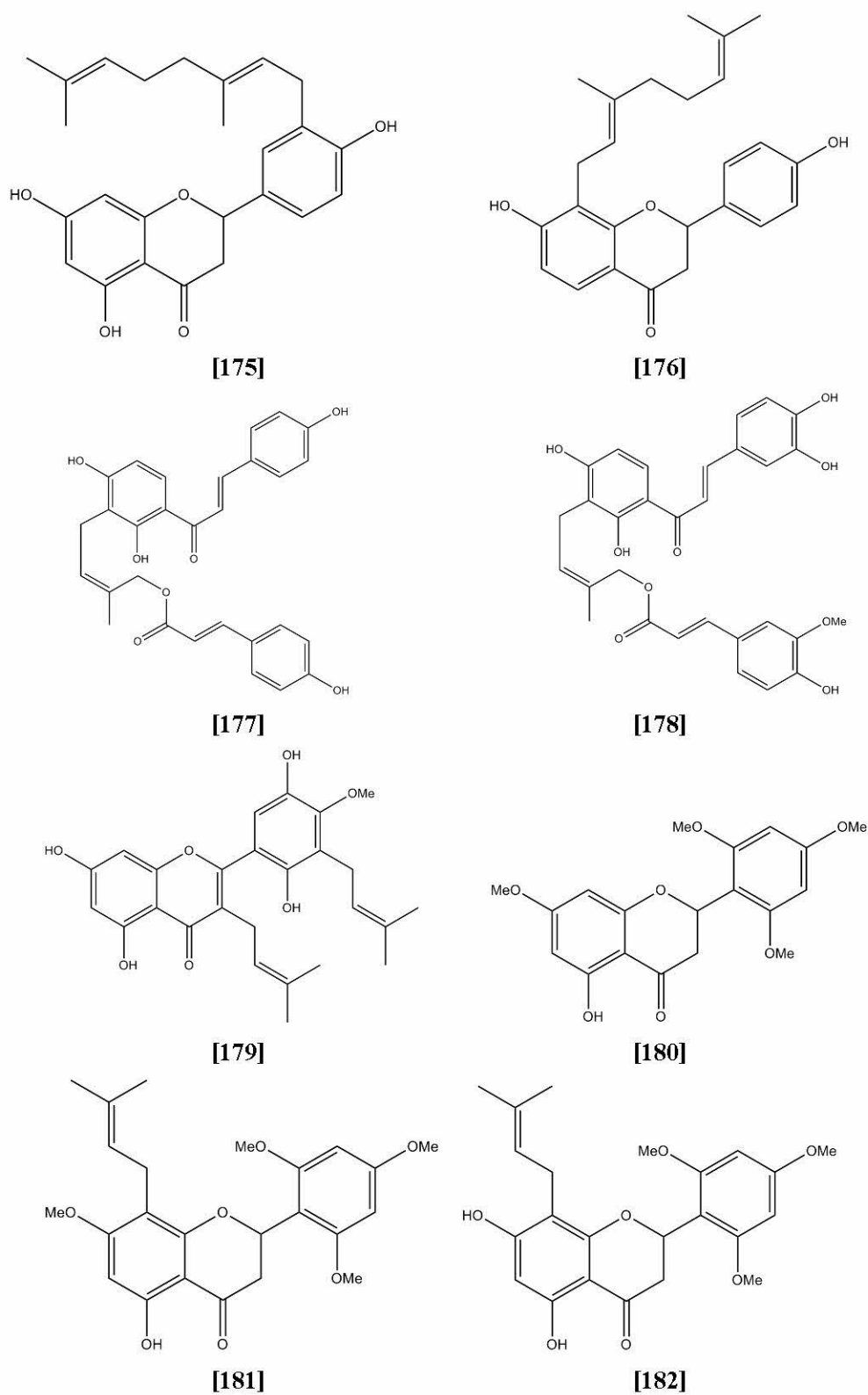


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

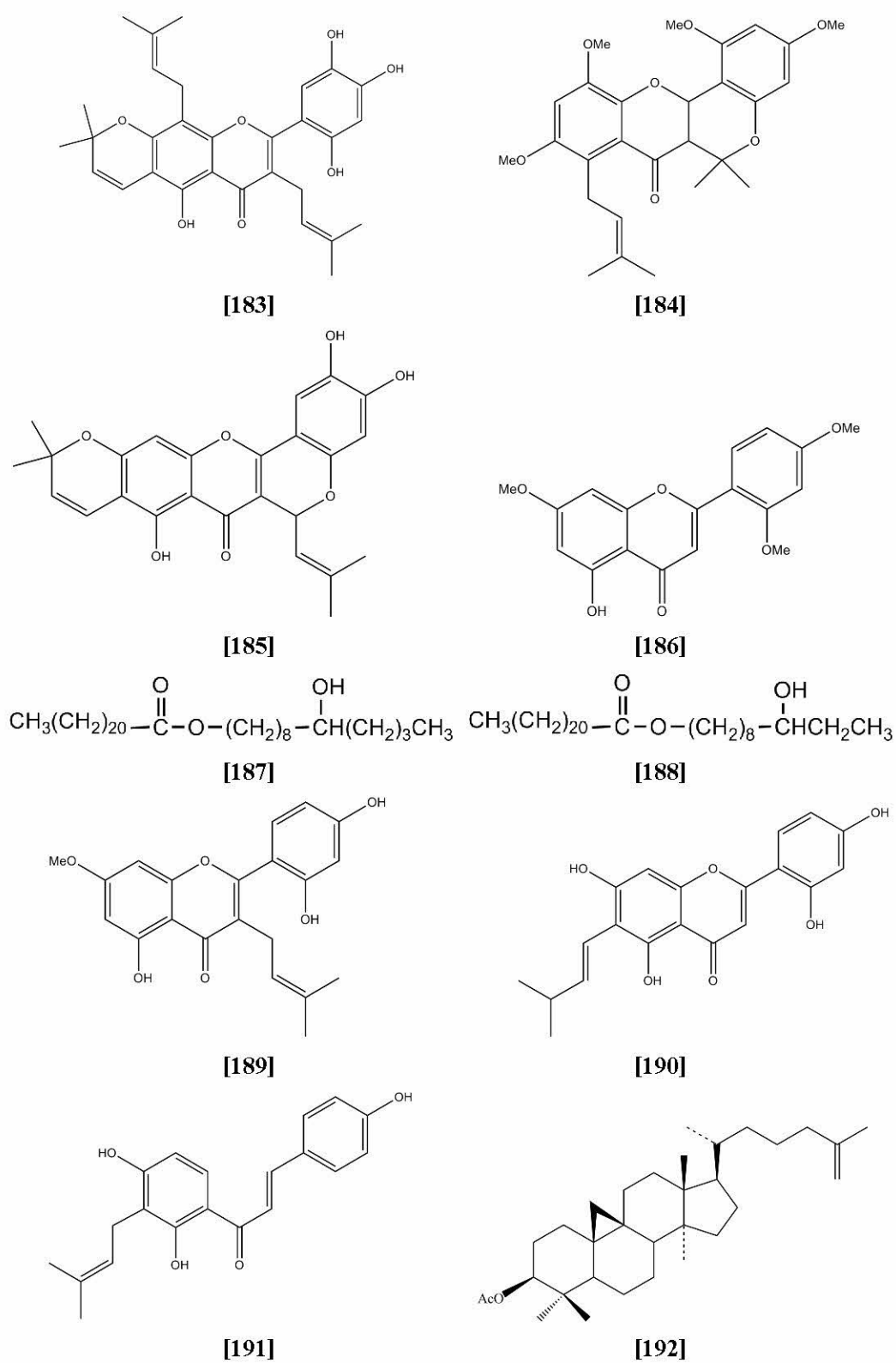


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

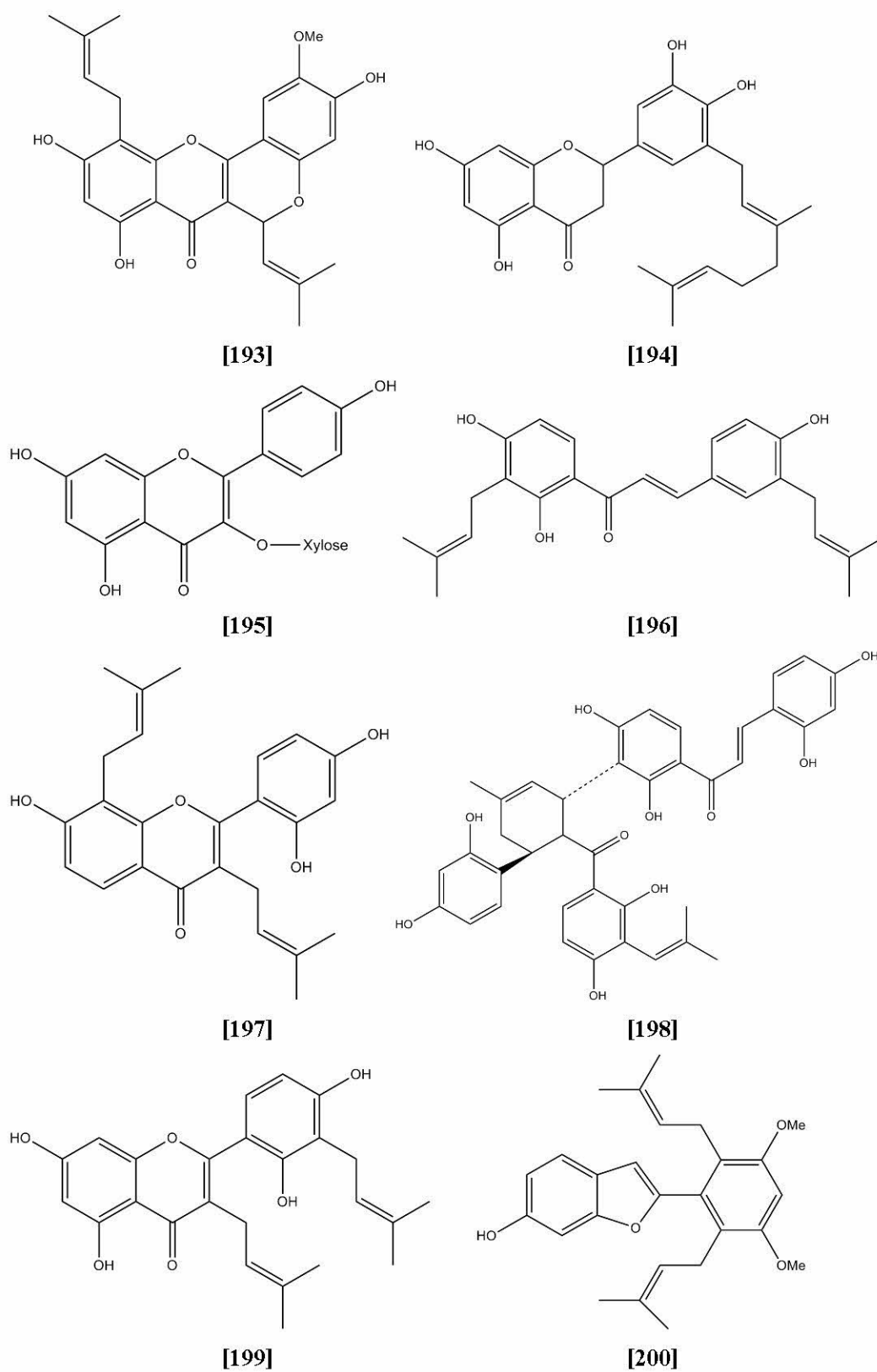


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

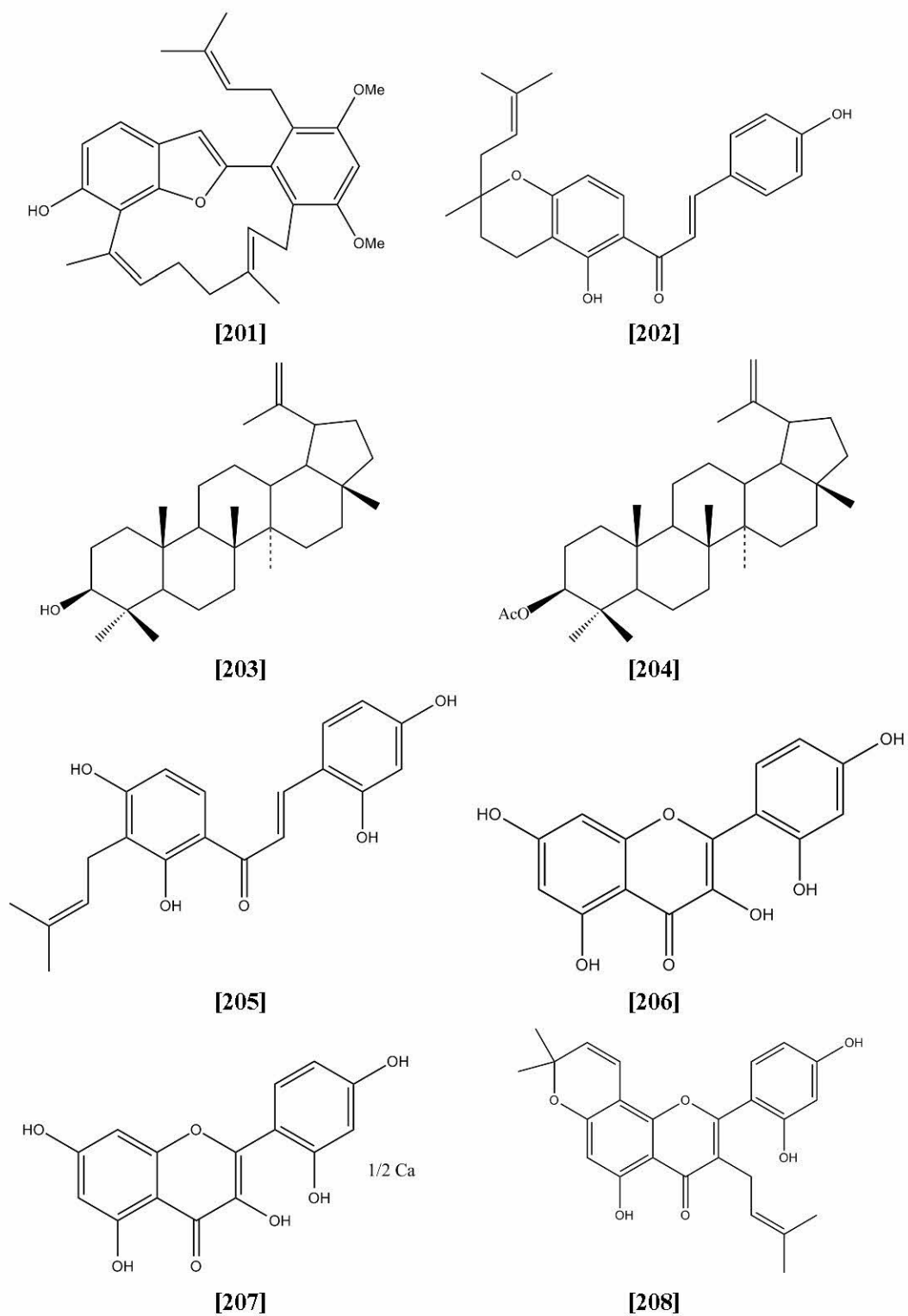


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

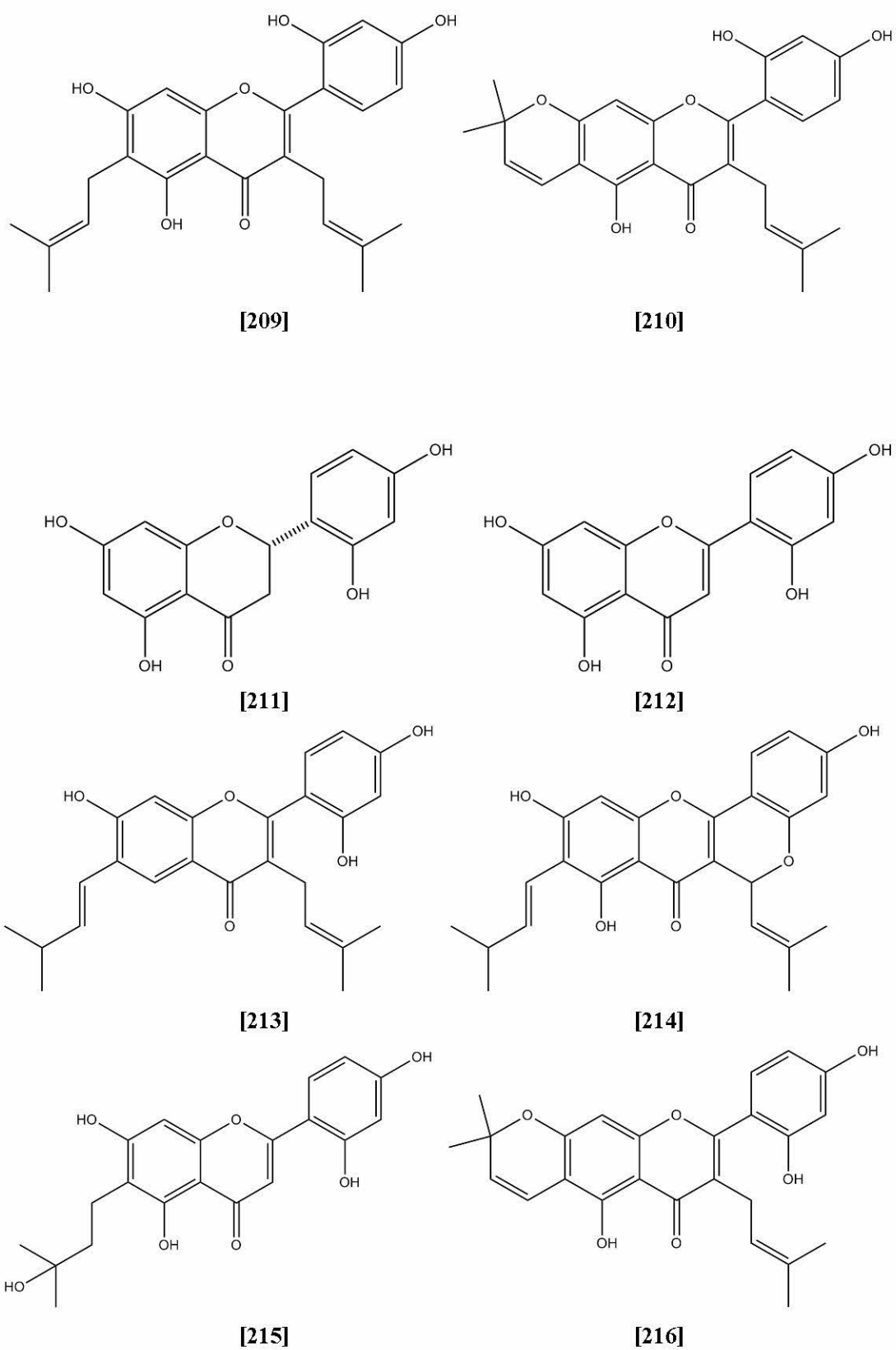


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

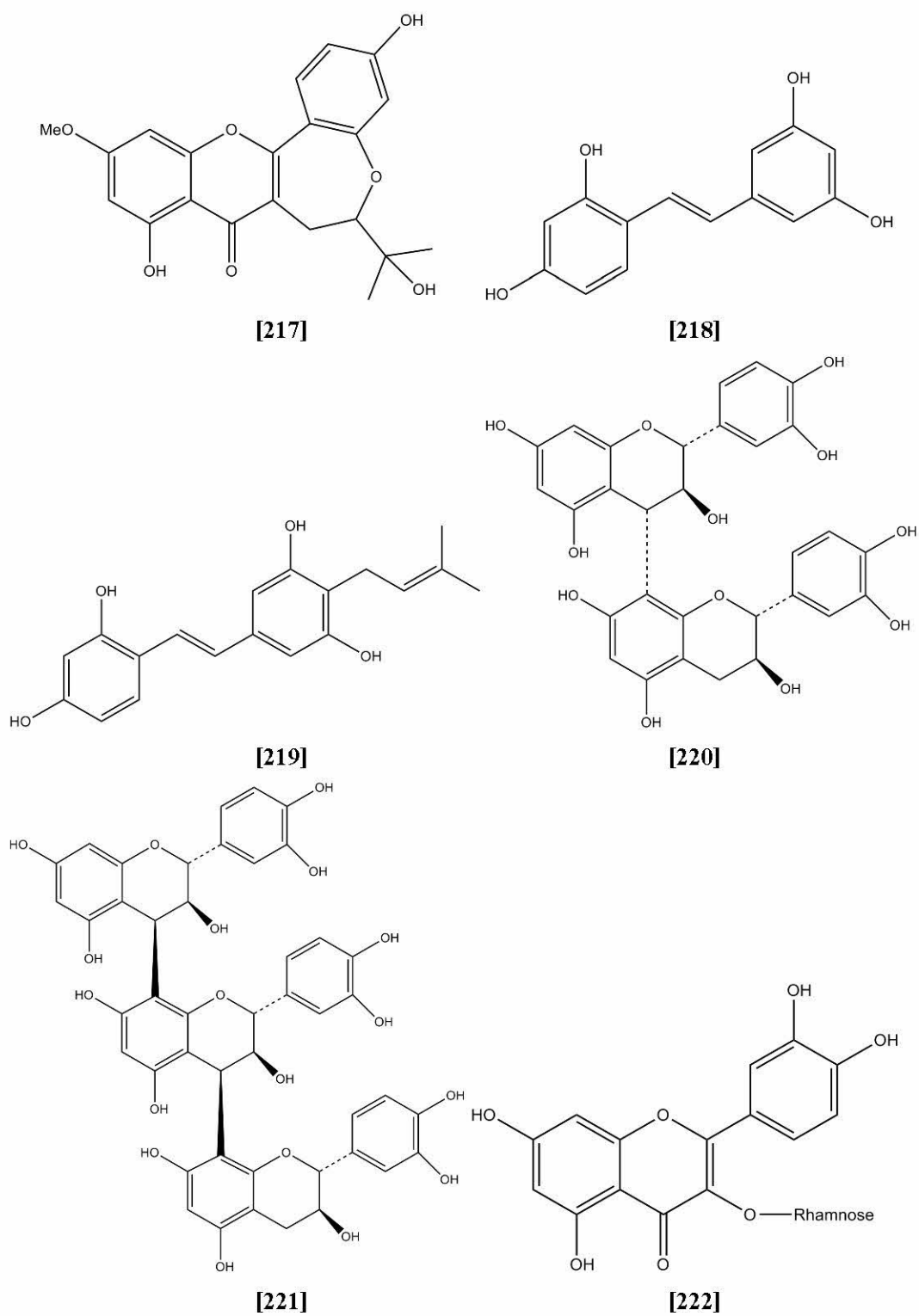


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)



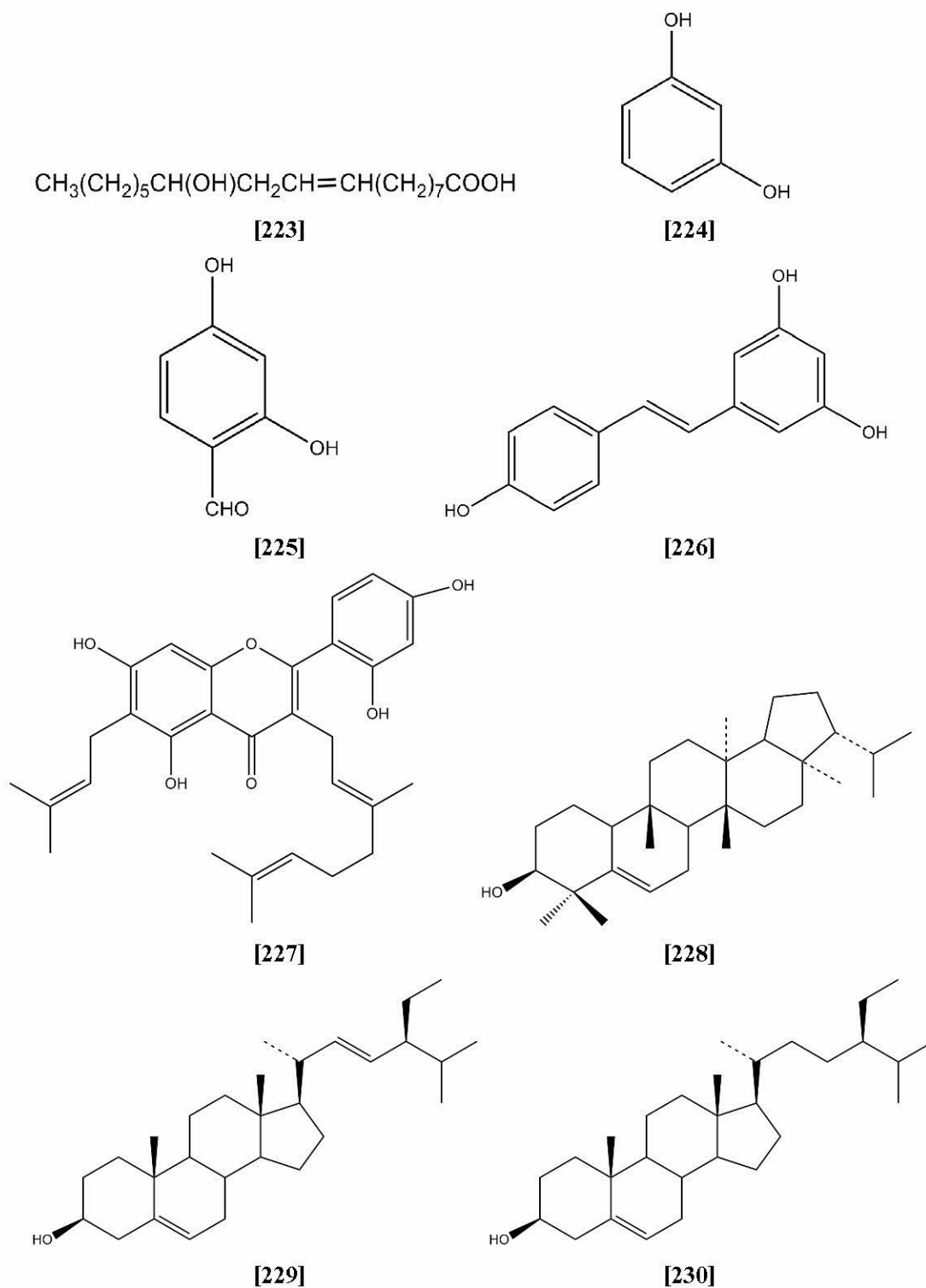


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

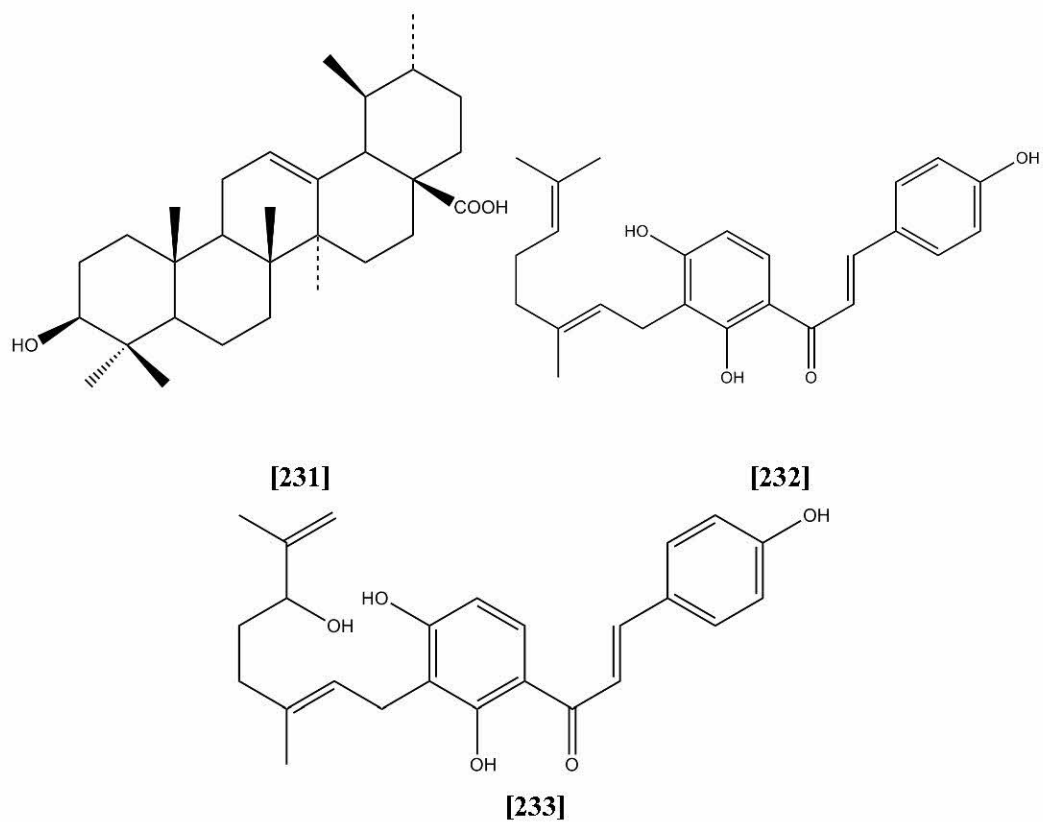


Figure 2.1 Structure of compounds previously isolated from *Artocarpus* spp.

Numbers of these structures are related to table 2.3 (continued)

## CHAPTER 3

### EXPERIMENTAL

#### 3.1 Material

##### 3.1.1 Plant material

Seventy seven samples of Thai medicinal plants were selected in order to prepare the crude extract for screening of anti-tyrosinase activity as shown in table 3.1. 21 samples are from Moraceae family because many previously reports of the plants in this family have shown high potential anti-tyrosinase activity (Sritularak *et al.*, 1998a, Likhitwitayawuid and Sritularak, 2001, Lee *et al.*, 2002, Arung *et al.*, 2006).

**Table 3.1 Seventy seven samples from Thai medicinal plants tested for anti-tyrosinase and anti-microbial activities determination**

No.	Family	Scientific name	Thai name	Plant part
1	Acanthaceae	<i>Barleria lupulina</i>	เสลดพังพอน	Leaf
2	Acanthaceae	<i>Andrographis paniculata</i>	ฟ้าทะลายโจร	Leaf
3	Acanthaceae	<i>Clinacanthus nutans</i>	พญาปล้องทอง	Leaf
4	Acanthaceae	<i>Thunbergia laurifolia</i>	รางจืด	Leaf
5	Anacardiaceae	<i>Spondias cytherea</i>	มะกอกฝรั่ง	Leaf
6	Annonaceae	<i>Desmos chinensis</i>	สายหยุด	Stem bark
7	Annonaceae	<i>Desmos chinensis</i>	สายหยุด	Leaf
8	Averrhoaceae	<i>Averrhoa bilimbi</i>	ตะลิงปลิง	Fruit
9	Averrhoaceae	<i>Averrhoa bilimbi</i>	ตะลิงปลิง	Juice
10	Capparidaceae	<i>Crateva magna</i>	กุ่มน้ำ	Branch
11	Capparidaceae	<i>Cleoma viscosa</i>	ผักเสี้ยนผี	Leaf

**Table 3.1 Seventy seven samples from Thai medicinal plants tested for anti-tyroisnase and anti-microbial activities determination (continued)**

No.	Family	Scientific name	Thai name	Plant part
12	Caesalpiniaceae	<i>Caesalpinia sappan</i>	ฝาง	Heartwood
13	Compositae	<i>Chromolaena odoratum</i>	สาบเสือ	Leaf
14	Compositae	<i>Saussurea lappa</i>	โกฐกระดูก	Rhizome
15	Combretaceae	<i>Combretum quadrangulare</i>	สะแกนา	Leaf
16	Euphorbiaceae	<i>Antidesma sp.</i>	มะเฒ่า	Leaf
17	Euphorbiaceae	<i>Bridelia ovata</i>	มะกา	Leaf
18	Euphorbiaceae	<i>Baccaurea macrophylla</i>	ลั้งแข	Fruit
19	Euphorbiaceae	<i>Euphorbia tirucalli</i>	พญาไร้ใบ	Aerial
20	Euphorbiaceae	<i>Phyllanthus acidus</i>	มะยม	Fruit
21	Euphorbiaceae	<i>Phyllanthus acidus</i>	มะยม	Root
22	Euphorbiaceae	<i>Sauropus changiana</i>	ลั้งมังกร	Leaf
23	Guttiferae	<i>Garcinia cowa</i>	ชะมวง	Leaf
24	Guttiferae	<i>Garcinia atroviridis</i>	ส้มแขก	Leaf
25	Lauraceae	<i>Cinnamomum iners</i>	อบเชย	Branch
26	Labiatae	<i>Ocimum tenuiflorum</i>	กระเพรา	Leaf
27	Labiatae	<i>Ocimum americanum</i>	แมงลัก	Leaf
28	Labiatae	<i>Ocimum basilicum</i>	โหระพา	Leaf
29	Leguminosae	<i>Acacia concinna</i>	ส้มป่อย	Leaf
30	Lythraceae	<i>Punica granatum</i>	ทับทิม	Peel (young)
31	Lythraceae	<i>Punica granatum</i>	ทับทิม	Peel (old)
32	Meliaceae	<i>Melia toosenden</i>	เถียน	Stem bark
33	Meliaceae	<i>Aglaia andamanica</i>	สังกะโต้ง	Leaf
34	Melastomataceae	<i>Melastoma malabathricum</i>	โคลงเคลง	Leaf
35	Moraceae	<i>Artocarpus heterophyllus</i>	ขนุน	Leaf
36	Moraceae	<i>Artocarpus altitis</i>	สาเก	Stem bark
37	Moraceae	<i>Artocarpus altitis</i>	สาเก	Leaf

**Table 3.1 Seventy seven samples from Thai medicinal plants tested for anti-tyroisnase and anti-microbial activities determination (continued)**

No.	Family	Scientific name	Thai name	Plant part
38	Moraceae	<i>Artocarpus altitis</i>	สาเก	Heartwood
39	Moraceae	<i>Artocarpus integer</i>	จำปาตะ	Root
40	Moraceae	<i>Artocarpus integer</i>	จำปาตะ	Wood
41	Moraceae	<i>Artocarpus integer</i>	จำปาตะ	Root bark
42	Moraceae	<i>Artocarpus integer</i>	จำปาตะ	Bark
43	Moraceae	<i>Streblus asper</i>	ข่อย	Leaf
44	Moraceae	<i>Streblus asper</i>	ข่อย	Branch
45	Moraceae	<i>Streblus asper</i>	ข่อย	Heartwood
46	Moraceae	<i>Streblus asper</i>	ข่อย	Stem bark
47	Moraceae	<i>Ficus benjamina</i>	ไทร	Leaf
48	Moraceae	<i>Ficus benjamina</i>	ไทร	Branch
49	Moraceae	<i>Ficus benjamina</i>	ไทร	Wood
50	Moraceae	<i>Ficus racemosa</i>	มะเดื่ออุทุมพร	Wood
51	Moraceae	<i>Ficus religiosa</i>	โพธิ์	Leaf
52	Moraceae	<i>Ficus religiosa</i>	โพธิ์	Branch
53	Moraceae	<i>Ficus religiosa</i>	โพธิ์	Stem bark
54	Moraceae	<i>Ficus religiosa</i>	โพธิ์	Wood
55	Moraceae	<i>Cudrania javanensis</i>	แกแล	Wood
56	Malvaceae	<i>Gossypium arboretum</i>	ฝ้ายแดง	Aerial
57	Mimosaceae	<i>Albizia procera</i>	ทึงถ่อน	Stem bark
58	Mimosaceae	<i>Mimosa</i> sp.	ไมยราบ (ลำต้นแดง)	Aerial
59	Mimosaceae	<i>Mimosa</i> sp.	ไมยราบ (ลำต้นแดง)	Flower
60	Piperaceae	<i>Piper nigrum</i>	พริกไทย	Leaf
61	Piperaceae	<i>Piper chaba</i>	คึบลิ	Leaf
62	Rubiaceae	<i>Hydnophytum formicarum</i>	หัวร้อยรู	Rhizome
63	Rubiaceae	<i>Hydnophytum formicarum</i>	หัวร้อยรู	Rhizome

**Table 3.1 Seventy seven samples from Thai medicinal plants tested for anti-tyroisnase and anti-microbial activities determination (continued)**

No.	Family	Scientific name	Thai name	Plant part
64	Rutaceae	<i>Citrus ichangensis</i>	มะขาม	Peel
65	Saururaceae	<i>Houttuynia cordata</i>	พลูคาว	Aerial
66	Sapindaceae	<i>Nephelium lappaceum</i>	เงาะ (โรงเรียน)	Peel
67	Sapindaceae	<i>Cardiospermum halicacabum</i>	โศกกะออม	Stem
68	Salvadoraceae	<i>Azima sarmentosa</i>	หนามพุงคอก	Root
69	Solanaceae	<i>Solanum ferox</i>	มะอึ๊ก	Branch
70	Solanaceae	<i>Capsicum frutescens</i>	พริก	Fruit
71	Simaroubaceae	<i>Eurycoma longifolia</i>	ปลาไหลเผือก	Root
72	Taccaceae	<i>Tacca leontopetaloides</i>	เท้าชายม่อม	Stem
73	Verbenaceae	<i>Vitex trifolia</i>	กนทีสอ	Fruit
74	Verbenaceae	<i>Vitex trifolia</i>	กนทีสอ	Leaf
75	Verbenaceae	<i>Vitex</i> sp.	กนทีสอ (แดง)	Leaf
76	Umbelliferae	<i>Conioselinum univittatum</i>	โกลฐหัวบัว	Rhizome
77	Zingiberaceae	<i>Languas galangal</i>	ข่า	Rhizome

### 3.1.2 Microbial, Media, Antibiotic, Enzyme and Chemical

#### (1) Microbial

##### Microbial

- *Staphylococcus aureus* ATCC 25923

##### Source

- Department of Pathology, Faculty of Medicine, Prince of Songkla University

- *Staphylococcus epidermidis* TISTR 517

- Thailand Institute of Scientific and Technology Research

- *Candida albicans* TISTR 5779

- Thailand Institute of Scientific and Technology Research

- *Propionibacterium acnes* DMST 14916

- Department of Medical Science, Ministry of Public Health, Thailand

- *Trichophyton mentagophytes*

- Department of Medical Science, Ministry of Public Health, Thailand

- *Trichophyton rubrum*

- Department of Medical Science, Ministry of Public Health, Thailand

### (2) Media

#### Media

- Mueller Hinton Agar (MHA)

#### Source

- Difco, Bacto Dickinson and Company, Spark USA.

- Mueller Hinton Broth (MHB)

- Difco, Bacto Dickinson and Company, Spark USA.

- Brain Heart Infusion Broth (BHIB)

- Difco, Bacto Dickinson and Company, Spark USA.

- Sabouraud's Dextrose Agar (SDA)

- Difco, Bacto Dickinson and Company, Spark USA.

- Tryptic Soy Broth (TSB)

- Difco, Bacto Dickinson and Company, Spark USA.

### (3) Antibiotic

#### Antibiotic

- Oxacillin paper disc

#### Source

- Oxoid Limited, England.

- Oxacillin sodium salt

- Fluka, Sigma-Aldrich, China

- Amphotericin B

- Sigma, Sigma-Aldrich, Germany

- Ketoconazole

- Sigma, Sigma-Aldrich, Germany

### (4) Enzyme

#### Enzyme

- Tyrosinase enzyme

#### Source

- Sigma, Sigma-Aldrich, Germany

(5) Chemical

Chemical

-  $\text{NaH}_2\text{PO}_2 \cdot 2\text{H}_2\text{O}$

-  $\text{Na}_2\text{HPO}_4$

- Kojic acid

- Dimethyl sulfoxide (DMSO)

Source

- MAY & BAKER Limited Dagenham

England

- MAY & BAKER Limited Dagenham

England

- Sigma-Aldrich, Germany

- Sigma-Aldrich, Germany



All solvents for extraction and isolation of compound were commercial grade and redistilled prior to use.

### 3.1.4 General techniques and equipments

#### Analytical Thin-Layer Chromatography (TLC)

Technique	:	One dimension, ascending
Absorbent	:	Silica gel 60 F <sub>254</sub> (E. Merck) precoated plate
Layer thickness	:	0.2 mm
Distance	:	7x5* cm (*= depend on amount of samples)
Detection	:	Ultraviolet light at the wavelength of 254 and 365 nm

#### Quick and Flash Column Chromatography

Adsorbent	:	Silica gel 60 particle size 0.040-0.063 nm (230-400 mesh ASTM) (Merck, Germany) and Flash Silica gel particle size 0.040-0.063 nm (Silicycle chemical division, Canada)
Packing method	:	Dry and Wet Packing
Sample loading	:	The sample was dissolved in a small amount of organic solvent, mixed with a small quantity of adsorbent, dried and then placed gently on surface of the adsorbent on column for dry packing and the sample was dissolved in a small amount of eluent and then applied gently on surface of adsorbent in column for wet packing.
Detection	:	Fractions were examined by TLC observing under Ultraviolet light at the wavelengths of 254 and 365 nm.

#### Ultraviolet (UV) Absorption Spectra

Isolated compound was dissolved in methanol or chloroform before measured Ultraviolet absorption. UV spectra were obtained on a Spectronic Genesys 6 UV-Visible Spectrophotometer, Thermo Scientific, Thermo Electron Corporation (Department of Pharmacognosy and Pharmaceutical Botany, Faculty of Pharmaceutical Sciences, Prince of Songkla University).

#### **Infrared (IR) Absorption Spectra**

IR (KBr disc) spectra were obtained from a Perkin Elmer FT-IR Spectrum One spectrometer (Department of Pharmaceutical Chemistry, Faculty of Pharmaceutical Sciences, Prince of Songkla University).

#### **Mass Spectra (MS)**

Electron Impact Mass Spectra (EIMS) were measured on a Thermofinnigan MAT 95 XL mass spectrometer, low resolution electron ionization mass spectrometry (Scientific Equipment Center, Prince of Songkla University).

#### **Proton-1 and Carbon-13 Nuclear Magnetic Resonance ( $^1\text{H}$ and $^{13}\text{C}$ -NMR) Spectra**

$^1\text{H}$  and  $^{13}\text{C}$  spectra were obtained with a Fourier Transform NMR Spectrometer ( $^1\text{H}$ -NMR 500 MHz and  $^{13}\text{C}$ -NMR 125 MHz), model UNITY INNOVA, Varian (Scientific Equipment Center, Prince of Songkla University) was used for identification of M-1, M-2 and M-3, and Bruker Avance DPX-300 FT-NMR spectrometer ( $^1\text{H}$ -NMR 300 MHz and  $^{13}\text{C}$ -NMR 75 MHz) (Faculty of Pharmaceutical Sciences, Chulalongkorn University) was used for identification of H-1, H-2 and H-3.

#### **Absorbance of Anti-tyrosinase activity**

Absorbance at 492 nm was performed on a DTX 880 Microplate Reader (Multimode Detector) (Pharmaceutical Laboratory Service Center, Faculty of Pharmaceutical Sciences, Prince of Songkla University).

### **Melting Point**

Melting points were obtained on a BUCIH MIA-21 melting point apparatus (Department of Pharmacognosy and Pharmaceutical Botany, Faculty of Pharmaceutical Sciences, Prince of Songkla University).

### **Optical rotation**

Optical rotations were measured on a Perkin Elmer 341 polarimeter (Pharmaceutical Research Instrument Center, Faculty of Pharmaceutical Sciences, Chulalongkorn University).

## **3.2 Methods**

### **3.2.1 Plant extraction**

Plant materials were chopped and blended into small pieces. They were extracted repeatedly with ethanol at room temperature 3 days (x3 times). The filtrates were pooled and evaporated under reduced pressure at temperature not exceeding 40 °C to yield the ethanol extract and were kept at 4 °C until examination.

### **3.2.2 Anti-tyrosinase activity**

Anti-tyrosinase activity was performed by using Dopachrome method (Sritularak, 1998a and 1998b). Dopachrome is one intermediate in melanogenesis, this method used the L-Dopa as a substrate. The oxidation of L-Dopa can be to dopachrome, which showed red and detected by visible light at 492 nm.

- Preparation of sample and positive control

1 mg of sample or positive control dissolved in 5 mL of DMSO (concentration is 200  $\mu\text{g}/\text{mL}$ ), kojic acid and water extract of *Artocarpus lakoocha* wood were used to positive control.

- Preparation of phosphate buffer pH 6.8

156 mg of  $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$  and 142 mg of  $\text{Na}_2\text{HPO}_4$  were mixed in 100 mL water distillation.

- Preparation of 0.85 mM L-Dopa and 203.3 unit/mL tyrosinase

0.85 mg of L-Dopa dissolved in 5 mL of phosphate buffer pH 6.8.

0.5 mg of tyrosinase enzyme 2033 unit/mg dissolved in 5 mL of phosphate buffer pH 6.8.

This method was tested in 96 well plate and detected with microplate reader at 492 nm. Each well of 96 well plate are including control, blank control, test sample and blank sample.

A. Control	20 $\mu\text{L}$ tyrosinase solution (203.3 unit/mL)
	140 $\mu\text{L}$ phosphate buffer pH 6.8
	20 $\mu\text{L}$ dimethylsulfoxide
B. Blank control	160 $\mu\text{L}$ phosphate buffer pH 6.8
	20 $\mu\text{L}$ dimethylsulfoxide
C. Test sample	20 $\mu\text{L}$ tyrosinase solution (203.3 unit/mL)
	140 $\mu\text{L}$ phosphate buffer pH 6.8
	20 $\mu\text{L}$ sample or standard solution
D. Blank sample	160 $\mu\text{L}$ phosphate buffer pH 6.8
	20 $\mu\text{L}$ sample or standard solution

The solution was mixed and then pre-incubated at 25  $^{\circ}\text{C}$ , 10 min, and 20  $\mu\text{L}$  of 0.85 mM L-Dopa was added in all well then optical density (OD) was detected with microplate reader at 492 nm and incubated at 25  $^{\circ}\text{C}$ , 20 min and OD was detected with microplate reader after that.

The % of tyrosinase inhibition was calculated, from equation as below, the absorbance value before incubation will be subtracted from the absorbance value after incubation then all values instead of equation.

$$\% \text{ tyrosinase inhibitory} = [((A-B) - (C-D)) / (A-B)] \times 100$$

### **3.2.3 Screening of anti-microbial activity of Thai medicinal plant extracts (follow to Lorian (2005) and Kummee and Intaraksa (2008)) by Agar Disc Diffusion method**

#### **3.2.3.1 Preparation of sample and antibiotic**

The plant extracts which showed % tyrosinase inhibition more than 50 % were selected for screening of anti-microbial activity. A sterile paper disc (diameter 6 mm) was impregnated with sample (10  $\mu$ L). The concentration of each sample was 200 mg/mL (2 mg/disc), by dissolving in DMSO. Control disc was similarly prepared by using DMSO as a solvent control. The positive controls (antibiotics) were including 1  $\mu$ g/disc oxacillin for testing of *S. aureus*, *S. epidermidis* and *P. acnes*, 25  $\mu$ g/disc Amphotericin B for testing of *C. albicans* and 25  $\mu$ g/disc ketoconazole for testing of *T. rubrum* and *T. mentagrophytes*.

#### **3.2.3.2 Preparation of microbial and testing**

**Table 3.2 Condition for culturing and testing of each microorganism**

Microbial	Temp. (°C)	Time (hr.)	Media
Bacteria			
- <i>S. aureus</i>	35-37	18-24	Mueller Hinton Agar (MHA) or Broth (MHB)
- <i>S. epidermidis</i>	35-37	18-24	Mueller Hinton Agar (MHA) or Broth (MHB)
- <i>P. acnes</i> <sup>A</sup>	35-37	72	Brain Heart Infusion Agar (BHIA) or Broth (BHIB)
Yeast			
- <i>C. albicans</i>	35-37	48	Sabouraud's Dextrose Agar (SDA)
Fungi			
- <i>T. rubrum</i>	30	7-10 days	Sabouraud's Dextrose Agar (SDA)
- <i>T. mentagrophytes</i>	30	7-10 days	Sabouraud's Dextrose Agar (SDA)

A = culturing in anaerobe condition

A bacterium or yeast from stock was streaked and incubated follow under condition in table 3.2. A colony from culture agar plate was taken to suspend in 0.85% NaCl solution. The cell suspension was dilution with 0.85% NaCl solution to achieve optical density (OD) with 0.08-0.1 by spectrophotometer at 625 nm (approximately to  $10^8$  CFU/mL) for bacterial and 0.11-0.13 by spectrophotometer at 530 nm (approximately to  $10^6$  CFU/mL) for yeast. A sterile cotton swab was dipped in the inoculum and excess was removed by rotation the swab several times against the inside wall of the tube above the fluid level. The surface of medium agar plate was inoculated by streaking the swab over the surface. Streaking was repeated three times, and each time the plate was rotated  $60^\circ$ . This ensured an even distribution of inoculum. The paper discs of sample, positive control and solvent were placed on the surface of bacterium inoculate follow under condition in table 3.2.

A fungus from stock was cultured and incubated follow under condition in table 3.2. A colony from culture agar slant was taken to suspend in 0.85% NaCl solution. The cell suspension was dilution with 0.85% NaCl solution to achieve optical density (OD) with 0.11-0.13 by spectrophotometer at 530 nm (approximately to  $10^6$  CFU/mL). The agar medium is maintained

in molten state at 45° C. One mL of inoculum was mixed with medium and added to sterile plate, cool agar medium. The contents are thoroughly mixed and allowed to solidify. The paper discs of sample, positive control and solvent were placed on the surface of fungi inoculate follow under condition in table 3.2.

### **3.2.3.3 Measurement of the result**

The inhibition zone or clear zone diameters were then measured by venires caliper.

### **3.2.4 Determination of minimum inhibitory concentration (MIC)**

The minimum inhibitory concentration was defined the lowest of compound to inhibit the growth of microorganisms. The samples shown clear zone were selected for determination of minimum inhibitory concentration by Broth micro-dilution method follow to Lorian, 2005, Phongpaichit *et al.*, 2006 and Niyomkam, 2007.

#### **3.2.4.1 Preparation of microbial for testing**

The inoculum was prepared and adjusted, as noted 3.2.3.2, to contain  $10^8$  CFU/mL, by adjusting the turbidity of 0.85% NaCl to match OD with 0.08-0.1 of spectrophotometer at 625 nm. It was then diluted 1: 100 in MHB for aerobic bacteria and BHIB for anaerobic bacteria to contain  $10^6$  CFU/mL.

#### **3.2.4.2 Preparation of sample, controls and testing**

The sample was dissolved in DMSO and diluted with MHB to the concentration of 4 mg/mL, Oxacillin sodium salt as a positive control, was diluted in sterile water distillation

(water was filtered through 0.45 micron sterile filter paper) to a concentration 256 µg/mL. The growth control is 2% DMSO in media and contamination control is only media.

The test was performed in 96-well plate two-fold dilutions were prepared directly in wells, as follow: 50 µL of the working solution of positive control or sample was added to well 1 and 2 of the dilution series. To each remaining well, 50 µL of MHB or BHIB is added. With a sterile pipette, 50 µL was transferred from well 2 to well 3. After thorough mixing, 50 µL was transferred (with a separate pipette for this and each succeeding transfer) to well 4. This process was continued to the next well until to the last well, from which 50 µL was removed and discarded. The solvent control was added with 50 µL of 2% DMSO. The contamination control and growth control were added with 100 and 50 µL MHB or BHIB, respectively. The stock solution of oxacillin sodium salt was diluted with MHB or BHIB to give the concentration of 128 to 0.0625 µg/mL. The stock solution of the sample and pure compound were diluted with MHB or BHIB to give the concentration of 2000 to 0.9765 µg/mL and 256 to 0.125 µg/mL, respectively. The 50 µL of the adjusted inoculum was added to positive control, sample, and growth control wells. The final concentration of bacteria in each well was  $5 \times 10^5$  CFU/mL. The cultures were then incubated follow under condition in table 3.2.

#### **3.2.4.3 Measurement of the result**

The lowest concentration that turbidity was not observed of microbial was the taken as the MIC, and confirmed with colorimetric method using Alamar blue as indicator, started by adding 5 µL of 1% Alamar blue in sterile water distillation every well and incubated 5-10 hr. The pink color showed growth of microbial, while that still blue color did not show any growth of microbial.

#### **3.2.5 Minimum bactericidal concentration (MBC)**

The minimum bactericidal concentration was defined as the lowest concentration of compound to kill microorganisms. The incubation mixture that showed positive result of



inhibitory effect (MIC) were streaked on Tryptic Soy Agar (TSA) for aerobic bacteria and BHIA for anaerobic bacterial then incubated follow under condition in table 3.2. The lowest concentration that did not show any growth was taken as the MBC.

### 3.3 Isolation of pure compounds

Thai medicinal plant, which showed high potential activities of anti-tyrosinase and anti-microbial, was selected for phytochemical investigation. The root of *Artocarpus integer* was selected. The dried roots of *Artocarpus integer* (2.88 kg) were chopped and blended into small pieces. They were extracted with hexane three times (3 days, each). The filtrates were pooled and evaporated under reduced pressure at temperature not exceeding 40 °C to yield a hexane extract (9 g, 0.332 % base on dried weight of root).

The marc (after extracted with hexane) was extracted three times with dichloromethane (3 days, each). Removal of organic solvent gave a dichloromethane extract (53.95 g, 1.873 % base on dried weight of root).

The marc (after extracted with hexane and dichloromethane, respectively) was extracted three times with ethyl acetate (3 days, each). Removal of organic solvent gave an ethyl acetate extract (31.80 g, 1.104 % base on dried weight of root).

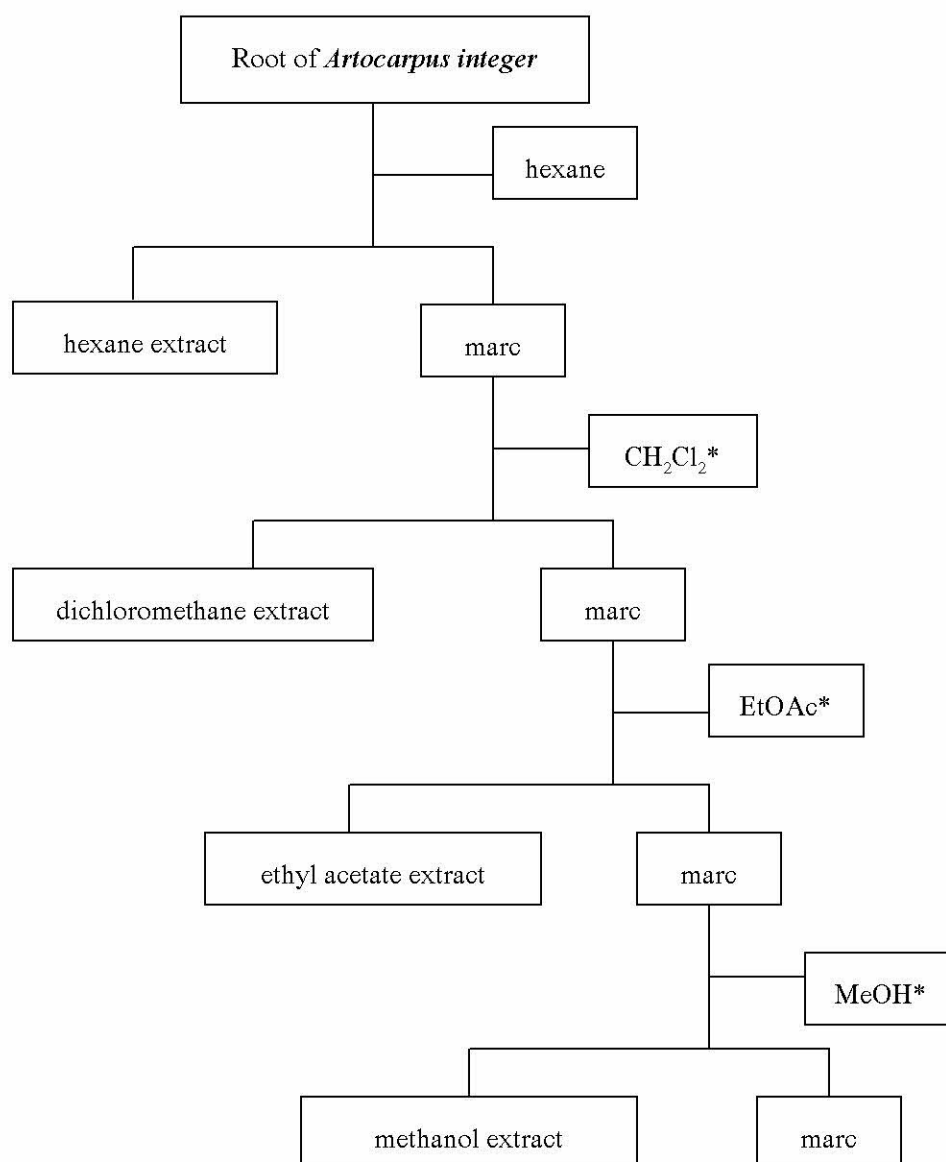
The marc (after extracted with hexane, dichloromethane and ethyl acetate, respectively) was extracted three times with methanol (3 days, each). Removal of organic solvent gave a methanol extract (86.21 g, 2.993 % base on dried weight of root). The separation of *Artocarpus integer* shown on scheme 3.1

These crude extracts were tested anti-tyrosinase activity (follow item 3.2.2) and TLC pattern was checked, the methanol extract showed high potential activity (see table 4.5) and the hexane extract showed different TLC pattern. Therefore, we have been isolated of the chemical constituents from methanol and hexane extracts.

#### 3.3.1 Isolation of chemical compound from methanol extract

The methanol extract (10 g) was dissolved in a small amount of methanol, triturated with silica gel 60 (Merck) and dried under the vacuum. It was then fractionated by quick column chromatographic technique using a sintered glass-filter column (size 500 mL) of silica gel 60 (Merck) (length 15 cm;  $\approx$  200 g). Elution was performed in a polarity gradient manner with dichloromethane, ethyl acetate and methanol as the solvents. The ratio and volume of solvents used in this column chromatography are summarized in Table 3.3. The isolation of pure compounds from methanol extract, are summarized in scheme 3.2.

The eluates were examined by TLC and fractions with similar chromatographic pattern were combined to 11 fractions, as shown in table 3.4.



**Scheme 3.1 Solvent extraction of *Artocarpus integer* root**

\* ( $\text{CH}_2\text{Cl}_2$  = dichloromethane; EtOAc = ethyl acetate and MeOH = methanol)

**Table 3.3 Ratio and volume of solvents in fourteen portions of quick column chromatography for isolation pure compound from methanol extract**

Portion	Ratio (%) of CH <sub>2</sub> Cl <sub>2</sub> *:	Volume of solvents (mL)		
	EtOAc*: MeOH*	CH <sub>2</sub> Cl <sub>2</sub> *	EtOAc*	MeOH*
1	100: 0: 0	500	0	0
2	90: 10: 0	225	25	0
3	80: 20: 0	200	50	0
4	70: 30: 0	175	75	0
5	60: 40: 0	150	100	0
6	50: 50: 0	125	125	0
7	40: 60: 0	100	150	0
8	30: 70: 0	75	175	0
9	20: 80: 0	50	200	0
10	10: 90: 0	25	225	0
11	0: 100: 0	0	250	0
12	0: 90: 10	0	225	25
13	0: 80: 20	0	200	50
14	0: 70: 30	0	525	225

\* (CH<sub>2</sub>Cl<sub>2</sub> = dichloromethane; EtOAc = ethyl acetate and MeOH = methanol)

**Table 3.4 Combination of fractions from quick column chromatography of methanol extract (10 g)**

Portion of quick coloumn	Fractions	Weight (g)
1	A	0.0459
2	B	0.0412
3-4	C	0.4400
5	D	0.3500
6	E	0.7000
7	F	0.5200
8	G	1.4400
9	H	1.2700
10-12	I	1.7300
13	J	1.9100
14	K	0.8300

### 3.3.1.1 Isolation of compound M-1

The fraction C (440 mg) was further purified by wet column chromatography. A 5 cm diameter of glass column was filled 30 cm length of Silica gel 60. The column was eluted with a solvent isocratic of hexane: dichloromethane: ethyl acetate in ratio 2.5:1:1, given to 5 fractions are including C-1, C-2, C-3, C-4 and C-5.

The fraction C-3 (246 mg) was further purified by wet column chromatography. A 3 cm diameter of glass column was filled 30 cm length silica gel 60. The column was eluted

with a solvent isocratic of dichloromethane 100%, given to 6 fractions are including C-3-1, C-3-2, C-3-3, C-3-4, C-3-5 and C-3-6.

The fraction C-3-3 (M-1) yielded a pure compound was obtained as yellow crystal amount 85 mg (0.00295 % based on dried weight of root). It was subsequently identified as Artocarpin.

### 3.3.1.2 Isolation of compounds M-2 and M-3

The fraction D (350 mg) was further purified by wet column chromatography. A 3 cm diameter of glass column was filled 30 cm length of silica gel 60. The column was eluted with a solvent isocratic of hexane: dichloromethane: ethyl acetate in ratio 2.5:1:1, given to 8 fractions are including D-1, D-2, D-3, D-4, D-5, D-6, D-7 and D-8.

The fraction D-3 (246 mg) was further purified by wet column chromatography. A 1 cm diameter of glass column was filled 20 cm length silica gel 60. The column was eluted with a solvent isocratic of hexane: dichloromethane: ethyl acetate in ratio 1:4:3, given to 3 fractions are including D-3-1, D-3-2 and D-3-3.

The fraction D-3-1 (87.7 mg) was further purified by wet column chromatography. A 1 cm diameter of glass column was filled 15 cm length silica gel 60. The column was eluted with a solvent isocratic of hexane: dichloromethane: ethyl acetate in ratio 1:4:3, given to 2 fractions are including D-3-1a, D-3-1b.

The fraction D-3-1a (M-2) yielded a pure compound was obtained as yellow powder amount 22 mg (0.00076 % based on dried weight of root). It was subsequently identified as Cudraflavone C.

The fraction D-5 (M-3) yielded a pure compound was obtained as white crystal amount 23 mg (0.00079 % based on dried weight of root). It was subsequently identified as Artocarpanone.

### 3.3.2 Isolation of chemical compound from hexane extract

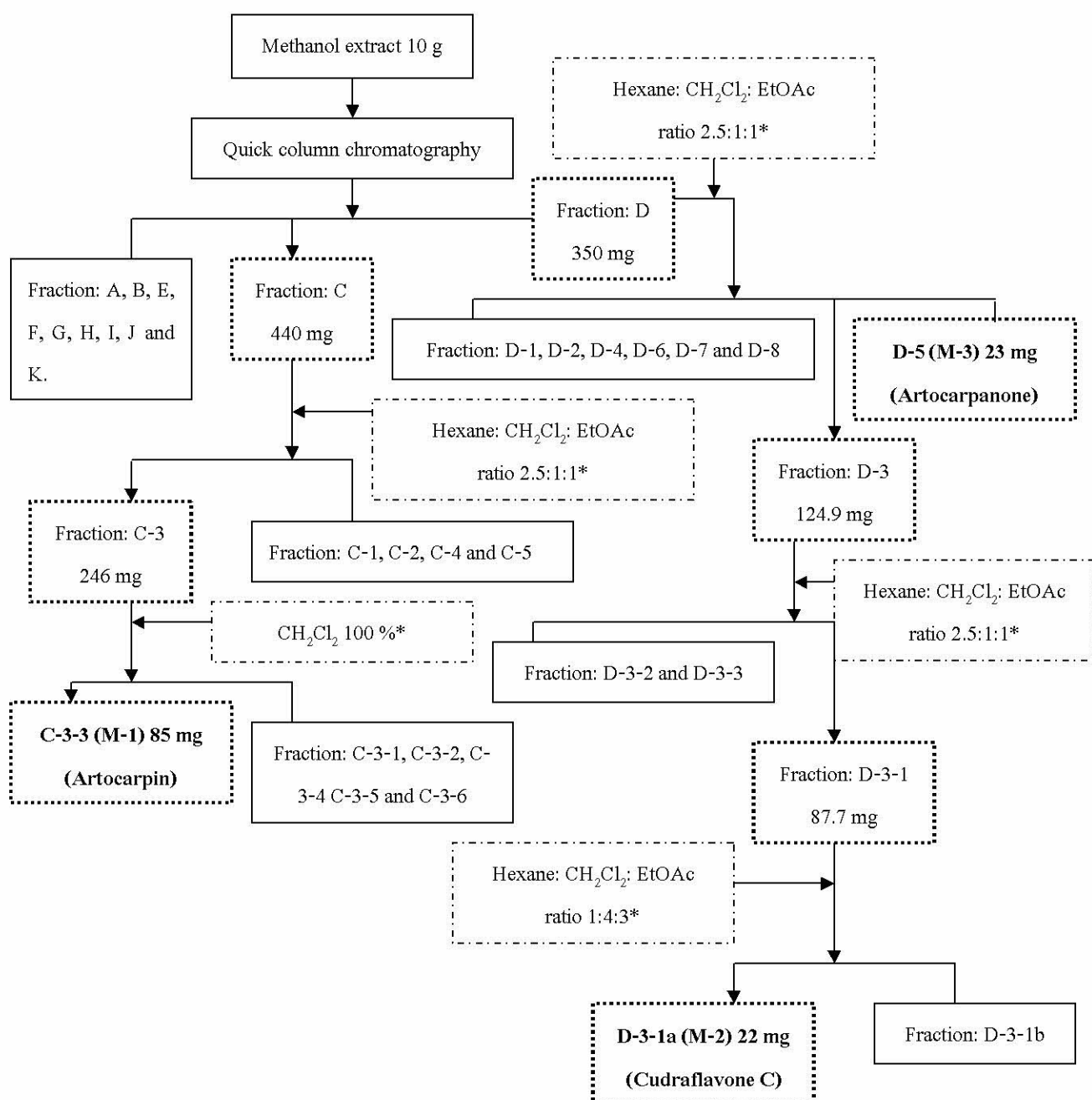
### 3.3.2.1 Isolation of compounds H-1, H-2 and mixture H-3

The hexane extract (2.014 g) was further purified by dry column chromatography. A 3 cm diameter of glass column was filled 20 cm length of flash silica gel. The column was eluted with a solvent isocratic of hexane: dichloromethane in ratio 6:4, given to 10 fractions are including CH-1, CH-2, CH-3, CH-4, CH-5, CH-6, CH-7, CH-8, CH-9 and CH-10. The isolation of pure and mixture compounds from hexane extract are summarized in scheme 3.3.

The fraction CH-1 (H-1) yielded a pure compound was obtained as colorless wax amount 13.4 mg (0.00046 % based on dried weight of root). It was subsequently identified as long chain alkene.

The fraction CH-2 (H-2) yielded a pure compound was obtained as colorless needles amount 14.6 mg (0.00050 % based on dried weight of root). It was subsequently identified as Lanosterol.

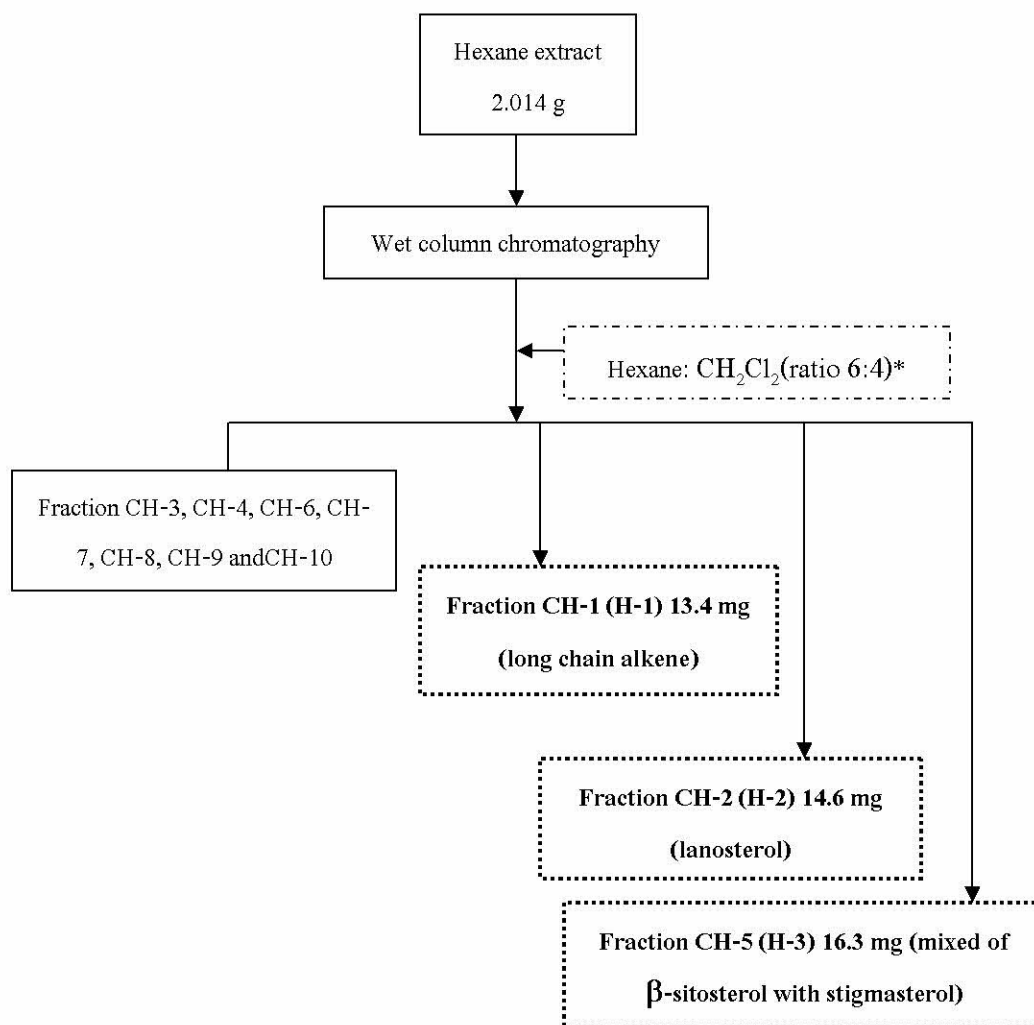
The fraction CH-5 (H-3) yielded a mixture compounds was obtained as colorless needles amount 16.3 mg (0.00056 % based on dried weight of root). It was subsequently identified as the mixture of  $\beta$ -sitosterol and stigmasterol.





### Scheme 3.2 Isolation of pure compounds from methanol extract

\* ( $\text{CH}_2\text{Cl}_2$  = dichloromethane; EtOAc = ethyl acetate and MeOH = methanol)



### Scheme 3.3 Isolation of pure and mixture compounds from hexane extract

\* ( $\text{CH}_2\text{Cl}_2$  = dichloromethane)

## CHAPTER 4

### RESULTS AND DISCUSSIONS

#### 4.1 Screening of Anti-tyrosinase and Anti-microbial activities from Thai medicinal plants

Among 77 samples of Thai medicinal plant extract (as shown in Table 4.1a) were investigated for anti-tyrosinase activity, 11 samples showed the tyrosinase inhibition more than 50 %, including *Averrhoa bilimbi* (juice), *Sauropus changiana* (leaves), *Ficus racemosa* (wood), *Cudrania javanensis* (wood), *Hydnophytum formicarum* (rhizome<sup>w</sup>), *Solanum ferox* (branch), *Artocarpus integer* (stem bark), *Artocarpus integer* (wood), *Artocarpus integer* (root bark), *Artocarpus integer* (root). The root extract of *A. integer* showed the highest tyrosinase inhibition (as shown in Table 4.1b). Therefore, these samples were further screening for anti-microbial activity by agar disc diffusion method. The samples showed inhibition zone, were selected in order to testing the minimum inhibitory concentration (MIC) by broth micro-dilution method and minimum bactericidal concentration (MBC) as following by item 3.2.3, 3.2.4 and 3.2.5, respectively.

The results of disc diffusion test (Table 4.2a and 4.2b) showed that the extract from *C. javanensis* was the most active against five microbes, as *Staphylococcus aureus*, *S. epidermidis*, *P. acnes*, *Trichophyton rubrum* and *T. mentagrophytes*. All extracts were inactive against *Candida albicans*. The MIC and MBC (Table 4.3 and 4.4) were examined by testing with *S. aureus*, *S. epidermidis*, and *P. acnes*. The results showed that the extracts of root and wood of *A. integer* were active against these microbes more than other samples. MIC values of root and wood extract against *S. aureus*, *S. epidermidis* and *P. acnes* were 3.91, 1.95 and 0.98 µg/mL, respectively. The root extract showed anti-microbial activity with MBC values of 1000, 3.91 and 62.5 µg/mL, respectively and the wood extract showed MBC values of 250, 2000 and 62.5 µg/mL, respectively against three microbes mentioned above.

The root of *A. integer* showed the potential of anti-tyrosinase and anti-microbial activities. Thus, this plant was selected for phytochemical investigation.

**Table 4.1a Percentage of tyrosinase inhibition of 77 samples from Thai medicinal plants**

No.	Family	Scientific name	Plant part	% tyrosinase inhibition ± SD at 200 µL/mL
1	Acanthaceae	<i>Barleria lupulina</i>	Leaf	22.1±1.42
2	Acanthaceae	<i>Andrographis paniculata</i>	Leaf	17.15 ± 0.42
3	Acanthaceae	<i>Clinacanthus nutans</i>	Leaf	22.89 ± 2.81
4	Acanthaceae	<i>Thunbergia laurifolia</i>	Leaf	6.96 ± 2.67
5	Anacardiaceae	<i>Spondias cytherea</i>	Leaf	10.17 ± 1.48
6	Annonaceae	<i>Desmos chinensis</i>	Stem bark	34.3 ± 0.84
7	Annonaceae	<i>Desmos chinensis</i>	Leaf	-1.96 ± 3.4
8	Averrhoaceae	<i>Averrhoa bilimbi</i>	Fruit	35.63 ± 3.98
9	<b>Averrhoaceae</b>	<i>Averrhoa bilimbi</i>	<b>Juice</b>	<b>61.23 ± 1.55</b>
10	Capparidaceae	<i>Crateva murvala</i>	Branch	5.24 ± 1.37
11	Capparidaceae	<i>Cleoma viscosa</i>	Leaf	31.47 ± 4.87
12	Caesalpinaceae	<i>Caesalpinia sappan</i>	Wood	20.27 ± 2.48
13	Compositae	<i>Chromolaena odoratum</i>	Leaf	8.63 ± 2.38
14	Compositae	<i>Saussurea lappa</i>	Rhizome	-20.48±4.76
15	Combretaceae	<i>Combretum quadrangulare</i>	Leaf	6.72±0.17
16	Euphorbiaceae	<i>Antidesma</i> sp.	Leaf	39.37±1.10
17	Euphorbiaceae	<i>Bridelia ovata</i>	Leaf	8.83±0.23
18	Euphorbiaceae	<i>Baccaurea macrophylla</i>	Fruit	29.01±4.95
19	Euphorbiaceae	<i>Euphorbia tirucalli</i>	Aerial	13.95±2.03
20	Euphorbiaceae	<i>Phyllanthus acidus</i>	Fruit	46.43±4.89
21	Euphorbiaceae	<i>Phyllanthus acidus</i>	Root	42.48±0.92
22	<b>Euphorbiaceae</b>	<i>Sauropus changiana</i>	<b>Leaf</b>	<b>55.79±3.87</b>
23	Guttiferae	<i>Garcinia cowa</i>	Leaf	17.24±0.00
24	Guttiferae	<i>Garcinia atroviridis</i>	Leaf	20.69±0.00
25	Lauraceae	<i>Cinnamomum iners</i>	Branch	22.13±4.21
26	Labiatae	<i>Ocimum tenuiflorum</i>	Leaf	-10.59±1.02

**Table 4.1a Percentage of tyrosinase inhibition of 77 samples from Thai medicinal plants  
(continued)**

No.	Family	Scientific name	Plant part	% tyrosinase inhibition ± SD at 200 µL/mL
27	Labiatae	<i>Ocimum americanum</i>	Leaf	21.18±2.04
28 <sup>w</sup>	Labiatae	<i>Ocimum basilicum</i>	Leaf	-3.17±5.50
29	Leguminosae	<i>Acacia concinna</i>	Leaf	4.41±0.11
30	Lythraceae	<i>Punica granatum</i>	Peel (young)	32.77±2.98
31	Lythraceae	<i>Punica granatum</i>	Peel (old)	27.08±2.08
32	Meliaceae	<i>Melia toosenden</i>	Stem bark	19.29±8.29
33	Meliaceae	<i>Aglaiia andamanica</i>	Leaf	0.00±0.00
34	Melastomataceae	<i>Melastoma malabathricum</i>	Leaf	11.58±8.1
35	Moraceae	<i>Artocarpus heterophyllus</i>	Leaf	35.41±5.6
36	Moraceae	<i>Artocarpus altitis</i>	Stem bark	3.75±1.37
37	Moraceae	<i>Artocarpus altitis</i>	Leaf	8.23±1.48
38	Moraceae	<i>Artocarpus altitis</i>	Wood	20.13±1.80
<b>39</b>	<b>Moraceae</b>	<i>Artocarpus integer</i>	<b>Stem bark</b>	<b>50.01±1.79</b>
<b>40</b>	<b>Moraceae</b>	<i>Artocarpus integer</i>	<b>Wood</b>	<b>80.02±3.22</b>
<b>41</b>	<b>Moraceae</b>	<i>Artocarpus integer</i>	<b>Root bark</b>	<b>82.60±0.76</b>
<b>42</b>	<b>Moraceae</b>	<i>Artocarpus integer</i>	<b>Root</b>	<b>90.57±2.93</b>
43	Moraceae	<i>Streblus asper</i>	Leaf	17.49±1.63
44	Moraceae	<i>Streblus asper</i>	Branch	13.85±0.37
45	Moraceae	<i>Streblus asper</i>	Wood	12.34±2.83
46	Moraceae	<i>Streblus asper</i>	Stem bark	41.56±1.12
47	Moraceae	<i>Ficus benamina</i>	Leaf	18.47 ± 0.50
48	Moraceae	<i>Ficus benamina</i>	Branch	12.64 ± 2.31
49	Moraceae	<i>Ficus benamina</i>	Wood	30.13 ± 1.63
<b>50</b>	<b>Moraceae</b>	<i>Ficus racemosa</i>	<b>Wood</b>	<b>56.41 ± 2.22</b>
51	Moraceae	<i>Ficus religiosa</i>	Leaf	-12.03 ± 6.90

**Table 4.1a Percentage of tyrosinase inhibition of 77 samples from Thai medicinal plants**  
(continued)

No.	Family	Scientific name	Plant part	% tyrosinase inhibition ± SD at 200 µL/mL
52	Moraceae	<i>Ficus religiosa</i>	Branch	3.10 ± 2.69
53	Moraceae	<i>Ficus religiosa</i>	Stem bark	1.67 ± 2.89
54	Moraceae	<i>Ficus religiosa</i>	Wood	6.44 ± 3.09
<b>55<sup>m</sup></b>	<b>Moraceae</b>	<b><i>Cudrania javanensis</i></b>	<b>Wood</b>	<b>77.86 ± 2.41</b>
56	Malvaceae	<i>Gossypium arboretum</i>	Aerial	-17.10 ± 0.84
57	Mimosaceae	<i>Albizia procera</i>	Stem bark	40.73 ± 3.98
58	Mimosaceae	<i>Mimosa</i> sp.	Aerial	39.28 ± 1.07
59	Mimosaceae	<i>Mimosa</i> sp.	Flower	22.1 ± 1.42
60	Piperaceae	<i>Piper nigrum</i>	Leaf	2.84 ± 2.46
61	Piperaceae	<i>Piper chaba</i>	Leaf	8.57 ± 0.21
62	Rubiaceae	<i>Hydnophytum formicarum</i>	Rhizome	45.24 ± 4.12
<b>63<sup>w</sup></b>	<b>Rubiaceae</b>	<b><i>Hydnophytum formicarum</i></b>	<b>Rhizome</b>	<b>53.71 ± 1.78</b>
<b>64</b>	<b>Rutaceae</b>	<b><i>Citrus ichangensis</i></b>	<b>Peel</b>	<b>52.46 ± 2.50</b>
65	Saururaceae	<i>Houttuynia cordata</i>	Aerial	-25.24 ± 1.82
66	Sapindaceae	<i>Nephelium lappaceum</i>	Peel	23.48 ± 3.35
67	Sapindaceae	<i>Cardiospermum halicacabum</i>	Stem	-14.32 ± 4.26
68	Salvadoraceae	<i>Azima sarmentosa</i>	Root	8.83 ± 0.23
<b>69</b>	<b>Solanaceae</b>	<b><i>Solanum ferox</i></b>	<b>Branch</b>	<b>50.87 ± 3.77</b>
70	Solanaceae	<i>Capsicum frutescens</i>	Fruit	15.44 ± 3.12
71	Simaroubaceae	<i>Eurycoma longifolia</i>	Root	37.17 ± 4.79
72	Taccaceae	<i>Tacca leontopetaloides</i>	Stem	25.75 ± 4.58
73	Verbenaceae	<i>Vitex trifolia</i>	Fruit	32.64 ± 1.20
74	Verbenaceae	<i>Vitex trifolia</i>	Leaf	37.35 ± 1.81
75	Verbenaceae	<i>Vitex</i> sp.	Leaf	21.74 ± 4.04
76	Umbelliferae	<i>Conioselinum univittatum</i>	Rhizome	-25.20 ± 0.71
77	Zingiberaceae	<i>Languas galangal</i>	Rhizome	19.23 ± 3.85

**Table 4.1b Percentage of tyrosinase inhibition of Kojic acid and *Artocarpus lakoocha* extract**

Positive control	% tyrosinase inhibition± SD at 200 µL/mL
Kojic acid	89.57 ± 2.15
<i>Artocarpus lakoocha</i> extract <sup>w</sup>	90.14 ± 1.46

Table 4.1a and 4.1b; <sup>w</sup> = water extract; <sup>m</sup> = methanol extract

**Table 4.2a Anti-bacterial activity of Thai medicinal plants by disc diffusion method**

Sample	Part	Inhibition zone [mm] ± SD		
		<i>S. aureus</i> ATCC 25923	<i>S. epidermidis</i> TISTR 517	<i>P. acnes</i> DMST 14916
<i>Ficus racemosa</i>	Wood	-	-	13.23±0.46
<i>Averrhoa bilimbi</i>	Juice	7.70±0.00	-	17.00±0.72
<i>Hydnophytum formicarum</i>	Root	8.10±0.00	8.55±0.15	16.88±0.60
<i>Citrus ichangensis</i>	Peel	-	-	8.77±0.40
<i>Solanum ferox</i>	Branch	7.60±0.00	-	11.23±0.70
<i>Sauropus changiana</i>	Leaf	-	-	-
<i>Artocarpus integer</i>	Stem bark	8.87±0.12	9.60±0.00	15.67±0.42
<i>Artocarpus integer</i>	Wood	8.13±0.15	10.60±0.00	15.95±0.68
<i>Artocarpus integer</i>	Root bark	11.57±0.40	13.83±0.12	15.97±0.85
<i>Artocarpus integer</i>	Root	9.10±0.00	10.67±0.09	15.25±0.05
<i>Cudrania javanensis</i>	Wood	13.37±0.40	12.67±0.12	21.55±0.64
DMSO		-	-	-
Oxacillin		20.13±0.12	21.80±0.00	36.12±0.97

- = Did not show inhibition zone;

Positive control = Oxacillin;

Negative control = DMSO

Table 4.2b Anti-fungal activity of Thai medicinal plants by disc diffusion method

Sample	Part	Inhibition zone [mm] $\pm$ SD		
		<i>C. albicans</i> TISTR 5779	<i>T. mentagrophytes</i>	<i>T. rubrum</i>
<i>Ficus racemosa</i>	Wood	-	-	-
<i>Averrhoa bilimbi</i>	Juice	-	-	-
<i>Hydnophytum formicarum</i>	Root	-	-	-
<i>Citrus ichangensis</i>	Peel	-	-	8.40 $\pm$ 0.00
<i>Solanum ferox</i>	Branch	-	-	-
<i>Sauropus changiana</i>	Leaf	-	-	-
<i>Artocarpus integer</i>	Stem bark	-	-	-
<i>Artocarpus integer</i>	Wood	-	6.83 $\pm$ 0.20	-
<i>Artocarpus integer</i>	Root bark	-	-	-
<i>Artocarpus integer</i>	Root	-	6.60 $\pm$ 0.17	-
<i>Cudrania javanensis</i>	Wood	-	7.82 $\pm$ 0.32	8.60 $\pm$ 0.52
DMSO		-	-	-
Amphotericin B		18.13 $\pm$ 0.32	NT	NT
Ketoconazole		NT	38.63 $\pm$ 0.86	30.47 $\pm$ 0.81

- = Did not show inhibition zone;

Positive control = Amphotericin B and Ketoconazole;

Negative control = DMSO

NT = No Test

Table 4.3 MIC of Thai medicinal plants against *S. aureus*, *S. epidermidis* and *P. acnes*

Sample	Part	MIC ( $\mu\text{g/mL}$ )		
		<i>S. aureus</i> ATCC 25923	<i>S. epidermidis</i> TISTR 517	<i>P. acnes</i> DMST 14916
<i>Ficus racemosa</i>	Wood	NT	NT	250
<i>Averrhoa bilimbi</i>	Juice	1000	NT	500
<i>Hydnophytum formicarum</i>	Root	1000	250	250
<i>Citrus ichangensis</i>	Peel	NT	NT	500
<i>Solanum ferox</i>	Branch	1000	NT	250
<i>Artocarpus integer</i>	Stem bark	1000	250	62.50
<i>Artocarpus integer</i>	Wood	3.91	1.95	0.98
<i>Artocarpus integer</i>	Root bark	62.50	31.25	31.25
<i>Artocarpus integer</i>	Root	3.91	1.95	0.98
<i>Cudrania javanensis</i>	Wood	1000	500	250
Oxacillin		0.5	0.125	0.0625

NT = No Test



**Table 4.4** MBC of Thai medicinal plants against *S. aureus*, *S. epidermidis* and *P. acnes*

Sample	Part	MBC ( $\mu\text{g/mL}$ )		
		<i>S. aureus</i> ATCC 25923	<i>S. epidermidis</i> TISTR 517	<i>P. acnes</i> DMST 14916
<i>Ficus racemosa</i>	Wood	NT	NT	1000
<i>Averrhoa bilimbi</i>	Juice	2000	NT	1000
<i>Hydnophytum formicarum</i>	Root	2000	2000	1000
<i>Citrus ichangensis</i>	Peel	NT	NT	1000
<i>Solanum ferox</i>	Branch	2000	NT	1000
<i>Artocarpus integer</i>	Stem bark	2000	2000	1000
<i>Artocarpus integer</i>	Wood	250	2000	62.50
<i>Artocarpus integer</i>	Root bark	1000	62.50	1000
<i>Artocarpus integer</i>	Root	1000	3.91	62.50
<i>Cudrania javanensis</i>	Wood	2000	2000	1000

NT = No Test

#### 4.2 Extraction of root from *Artocarpus integer*

The dried roots of *A. integer* (2.88 kg) were extracted with hexane, dichloromethane, ethyl acetate and methanol. The dry weight, % yield and % tyrosinase inhibition of these crude extracts as shown in Table 4.5. Methanol and hexane extracts were selected for isolation of the chemical constituents.

**Table 4.5 Dry weight, % yield, % tyrosinase inhibition of *A. integer* crude extracts**

Crude extract	Dry weight (g)	% yield	% tyrosinase inhibition at 200 $\mu\text{g/mL}$
Hexane	9.55	0.332	14.86
Dichloromethane	53.95	1.873	53.25
Ethyl acetate	31.8	1.104	85.68
Methanol	86.21	2.993	88.77

#### 4.3 Structure determination of isolated compounds

Three pure compounds were isolated from methanol extract and were identified as M-1 (Artocarpin), M-2 (Cudraflavone C) and M-3 (Artocarpanone). The other two pure compounds and one mixture compound were isolated from hexane extract and were identified as H-1 (long chain alkene), H-2 (Lanosterol) and H-3 (mixture of  $\beta$ -Sitosterol and Stigmasterol). These structures of the isolated compounds were identified by physical properties (such as melting point) and spectroscopic data including UV, IR, NMR and MS data and confirmed by comparison with previously reports.

### 4.3.1 Structure determination of compound M-1

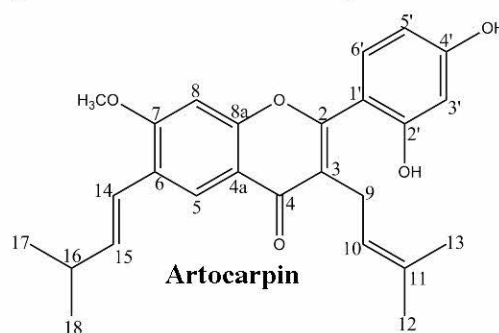
The compound M-1 was obtained as a yellow crystal. The UV-visible spectrum was found maximum wavelengths ( $\lambda_{\text{max}}$ ) at 278 and 322 nm (Figure A1) (279 and 316 nm (Cunha and Socorro, 1994)). The melting point and optical rotation showed at 136-140 °C (137-141 °C (Sritularak, 1998a)) and  $[\alpha]_D^{28}$  (0.06 g/100 mL, methanol) -6.45, respectively. In IR-Spectrum (KBr)  $\nu_{\text{max}}$  (Figure A2) showed at 3436 (OH), 1643 (C=O), 1542-1620 (C=C, aromatic ring)  $\text{cm}^{-1}$ , respectively (3417, 1647, 1615 and 1563  $\text{cm}^{-1}$  (Sritularak, 1998a)). The EI mass spectrum (Figure A3) showed as molecular ion peak at  $m/z$  436 corresponding to  $\text{C}_{26}\text{H}_{28}\text{O}_6$ .

The  $^1\text{H}$  NMR spectra data showed total 25 protons, including 5 proton groups from 14 signals, as aromatic protons 4 signal ( $\delta$  6.33 (s), 6.48 (s), 6.50 (dd,  $J = 2.2$  Hz) and 7.19 (d,  $J = 9.03$  Hz)), olefin proton 3 signals ( $\delta$  5.14 (m), 6.69 (dd,  $J = 7.07, 16.23$ ), 6.55 (dd,  $J = 1.22, 16.35$ ), methyl proton 3 signals ( $\delta$  1.09 (S), 1.44 (S) and 1.62 (S)) methane proton 2 signals ( $\delta$  3.11 (d,  $J = 6.59$  Hz), 2.46 (m), a chelated hydroxyl proton to carbonyl group 1 signal ( $\delta$  13.50) and proton of methoxyl group ( $\delta$  3.86), respectively.

The  $^{13}\text{C}$  NMR spectra data showed total 26 carbons, including 6 carbon groups from 26 signals, as carbonyl 1 signal ( $\delta$  182.2), aromatic carbon 14 signals ( $\delta$  89.4, 103.8, 104.9, 108.3, 109.7, 112.5, 120.8, 131.5, 155.1, 156.0, 158.6, 158.9, 159.3 and 162.9), methane carbon 2 signals ( $\delta$  24.3, 33.0), olefin carbon 4 signals ( $\delta$  115.5, 121.5, 133.3 and 142.6), methyl carbon 3 signals ( $\delta$  17.6, 22.6 and 25.6) and methoxyl carbon 1 signal ( $\delta$  55.9), respectively.

From the spectra,  $^1\text{H}$  NMR (Figure A4),  $^{13}\text{C}$  NMR (Figure A5), DEPT (Figure A6.1 and A6.2), COSY (Figure A7), HMBC (Figure A8) and HMQC (Figure A9), physical property data and comparison with previously reports (Sritularak, 1998a; Cunha and Socorro, 1994), compound M-1 was identified as Artocarpin.

The complete  $^1\text{H}$  and  $^{13}\text{C}$  NMR assignments and the HMBC correlation are summarized in Table 4.6



**Table 4.6**  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of Compound M-1 (in Chloroform- $d_3$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively)) and Artocarpin (in DMSO- $d_6$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively)) with long-range correlations in HMBC spectrum

Position	Compound M-1		Artocarpin ( Sritularak, 1998a)		Correlation with proton
	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity, $J$ in Hz)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity, $J$ in Hz)	
2	159.3	-	162.9	-	H-9
3	120.8	-	121.1	-	-
4	182.2	-	182.7	-	H-8 and H-9
4a	104.9	-	105.0	-	5-OH and H-8
5	158.6	-	158.9	-	5-OH, H-8 and H-14
6	109.7	-	109.1	-	5-OH and H-8
7	162.9	-	163.4	-	7- OCH <sub>3</sub> ,H-8 and H-14
8	89.4	6.33 (s)	91.0	6.65 (s)	-
8a	156.0	-	158.6	-	-
9	24.3	3.11 (d, 6.59)	24.5	3.02 (d, 6.7)	H-10
10	121.5	5.14 (t)	122.3	5.05 (t)	H-9, H-12 and H-13
11	133.3	-	132.1	-	-
12	17.6	1.44 (s)	18.2	1.38 (s)	H-10 and H-13
13	25.6	1.62 (s)	26.3	1.55 (s)	H-10 and H-12
14	115.5	6.55 (dd, 1.22, 16.25)	116.7	6.50 (d, 16.5)	H-15
15	142.6	6.69 (dd, 7.07, 16.23)	142.0	6.64 (dd, 6.9, 16.5)	H-14, H-16,H-17 and H-18 H-15, H-17 and H-18
16	33.0	2.46 (m)	33.4	2.44 (m)	H-15, H-16, H-17 and H-18
17	22.6	1.09 (d, 6.84)	23.5	1.06 (d, 6.6)	H-15, H-16, H-17 and H-18 H-3'
18	22.6	1.09 (d, 6.84)	23.5	1.04 (d, 6.6)	H-6'
1'	112.5	-	111.7	-	H-5'
2'	155.1	-	157.3	-	H-6'
3'	103.8	6.48 (s)	103.5	6.45 (d ,1.8)	H-5'
4'	158.9	-	161.3	-	-

**Table 4.6**  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of Compound M-1 (in Chloroform- $d_3$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively)) and Artocarpin (in DMSO- $d_6$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively)) with long-range correlations in HMBC spectrum (continued)

Position	Compound M-1		Artocarpin ( Sritularak, 1998a)		Correlation with proton
	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity, $J$ in Hz)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity, $J$ in Hz)	
5'	108.3	6.50 (d, 2.2)	107.6	6.36 (dd, 8.1, 2.1)	-
6'	131.5	7.19 (d, 9.03)	132.9	7.12 (d, 8.4)	-
5-OH	-	13.50 (s)	-	13.89 (s)	-
7-OCH <sub>3</sub>	55.9	3.86 (s)	56.2	3.89 (s)	-

#### 4.3.2 Structure determination of compound M-2

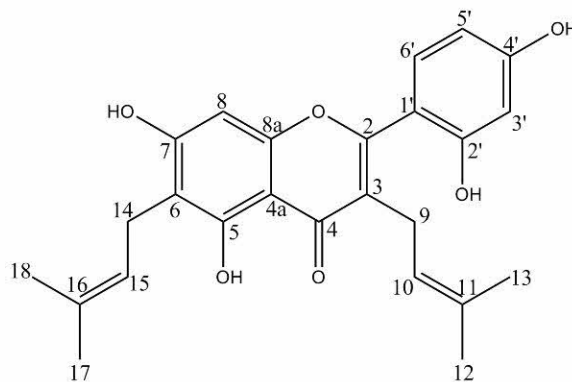
The compound M-2 was obtained as a yellow powder. The UV-visible spectrum was found maximum wavelengths ( $\lambda_{\text{max}}$ ) at 262 and 315 nm (Figure B1) (274 and 306 nm (Sritularak, 1998a)). The optical rotation showed at  $[\alpha]_{\text{D}}^{28}$  (0.02 g/100 mL) +30.00. In IR-Spectrum (KBr)  $\nu_{\text{max}}$  (Figure B2) showed at 3401 (OH), 1643 (C=O), 1566-1623 (C=C, aromatic ring)  $\text{cm}^{-1}$ , respectively (3321, 1651, 1626 and 1575  $\text{cm}^{-1}$  (Sritularak, 1998a)). The EI mass spectrum (Figure B3) showed as molecular ion peak at  $m/z$  422 corresponding to  $\text{C}_{25}\text{H}_{26}\text{O}_6$ .

The  $^1\text{H}$  NMR spectra data showed total 22 proton, including 6 proton groups from 12 signals, as aromatic proton 4 signals ( $\delta$  6.29 (s), 6.37 (dd,  $J$  = 8.30, 2.19 Hz), 6.39 (dd,  $J$  = 2.2 Hz) and 7.04 (d,  $J$  = 8.06 Hz)), olefin proton 2 signals ( $\delta$  5.07 (m), 5.23 (m)), methane proton 2 signals ( $\delta$  3.06 (d,  $J$  = 2.84 Hz), 3.30 (m) and methyl proton 4 singles ( $\delta$  1.37 (S), 1.56 (S), 1.65 (S) and 1.76 (S)), respectively.

The  $^{13}\text{C}$  NMR spectra data showed total 25 carbons, including 6 carbon groups from 25 signals, as carbonyl 1 signal ( $\delta$  183.65), aromatic carbon 14 signals ( $\delta$  93.67, 103.77, 105.20, 107.95, 112.39, 113.54, 121.70, 132.36, 157.64, 157.68, 159.88, 161.73, 163.17 and 163.25), olefin carbon 4 signals ( $\delta$  122.94, 123.64, 131.88 and 132.55), methane carbon 2 signals ( $\delta$  22.27 and 24.88) and methyl carbon 4 signals ( $\delta$  17.61, 17.89, 25.81 and 25.93), respectively.

From the spectra,  $^1\text{H}$  NMR (Figure B4),  $^{13}\text{C}$  NMR (Figure B5), DEPT (Figure B6.1 and B 6.2), COSY (Figure B7), HMBC (Figure B8) and HMQC (Figure B9), physical property data and comparing with previously reports (Sritularak, 1998a), compound M-2 was identified as Cudraflavone C.

The complete  $^1\text{H}$  and  $^{13}\text{C}$  NMR assignments and the HMBC correlation are summarized in Table 4.7



**Cudraflavone C**

**Table 4.7**  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of Compound M-2 (in Methanol- $d_4$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively)) and Cudraflavone C (in DMSO- $d_6$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively)) with long-range correlations in HMBC spectrum

Position	Compound M-2		Cudraflavones C (Sritularak, 1998a)		Correlation with proton
	$\delta_c$ (ppm)	$\delta_H$ (ppm) (multiplicity, $J$ in Hz)	$\delta_c$ (ppm)	$\delta_H$ (ppm) (multiplicity, $J$ in Hz)	
2	163.17	-	162.3	-	H-9 and H-6'
3	121.70	-	120.5	-	-
4	183.65	-	182.3	-	-
4a	105.20	-	104.1	-	H-9
5	159.88	-	159.1	-	H-14
6	112.39	-	111.3	-	-
7	163.25	-	162.6	-	H-14
8	93.67	6.29 (s)	93.5	6.34 (s)	-

**Table 4.7  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of Compound M-2 (in Methanol- $d_4$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively)) and Cudraflavone C (in DMSO- $d_6$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively)) with long-range correlations in HMBC spectrum (continued)**

Position	Compound M-2		Cudraflavones C (Sritularak, 1998a)		Correlation with proton
	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity, $J$ in Hz)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity, $J$ in Hz)	
8a	157.64	-	156.3	-	H-8
9	24.88	3.06 (d, 2.84)	24.5	2.97 (d, 6.6)	H-10
10	122.94	5.07 (t)	122.5	5.02 (t)	H-9, H-10, H-12 and H-13
11	132.55	-	131.9	-	H-12 and H-13
12	17.89	1.37 (s)	18.2	1.36 (s)	H-13
13	25.81	1.56 (s)	26.3	1.54 (s)	H-12
14	22.27	3.30 (m)	21.9	3.22 (d, 6.9)	H-15
15	123.64	5.23 (t)	123.2	5.17 (t)	H-14, H-15, H-17 And H-18
16	131.88	-	131.4	-	H-14, H-17 and H-18
17	17.61	1.76 (s)	18.6	1.73 (s)	H-18
18	25.93	1.65 (s)	26.3	1.62 (s)	H-17
1'	113.54	-	111.9	-	H-3' and H-5'
2'	157.68	-	157.2	-	H-3' and H-6'
3'	103.77	6.39 (d, 2.2)	103.5	6.42 (d, 2.1)	H-5'
4'	161.73	-	161.2	-	H-3'
5'	107.95	6.37 (dd, 8.30, 2.19)	107.6	6.34 (m)	H-3' and H-6'
6'	132.36	7.04 (d, 8.06)	132.0	7.06 (d, 8.04)	-
5-OH	-	-	-	13.31 (s)	-

### 4.3.3 Structure determination of compound M-3

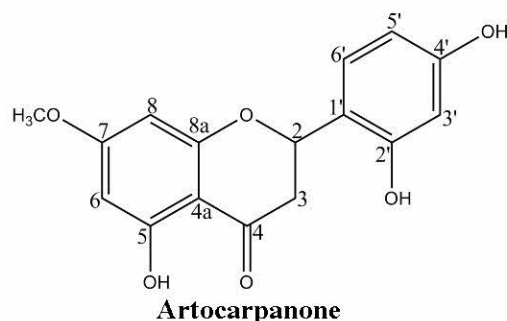
The compound M-3 was obtained as a white crystal. The UV-visible spectrum was found maximum wavelengths ( $\lambda_{\max}$ ) at 204 and 286 nm (Figure C1). The melting point and optical rotation showed at 156-160 °C (220-212 °C (Wei *et al*, 2005)) and  $[\alpha]_D^{28}$  -23.58 (0.02 g/100 mL, methanol). In IR-Spectrum (KBr)  $\nu_{\max}$  (Figure C2) showed at 3436 (OH), 1626 (C=O), 1468-1525 (C=C, aromatic ring)  $\text{cm}^{-1}$ , respectively (3450, 1640, and 1615  $\text{cm}^{-1}$  (Wei *et al*, 2005)). The EI mass spectrum (Figure C3) showed as molecular ion peak at  $m/z$  302 corresponding to  $\text{C}_{16}\text{H}_{14}\text{O}_6$ .

The  $^1\text{H}$  NMR spectra data showed total 11 protons, including 3 proton groups from 9 signals, as aromatic proton 5 signals ( $\delta$  6.01 (d,  $J = 2.29$  Hz), 6.04 (d,  $J = 2.29$  Hz), 6.32 (dd,  $J = 2.29$  Hz), 6.33 (dd,  $J = 8.23, 2.29$  Hz), and 7.22 (d,  $J = 8.01$  Hz)), methane proton 3 signals ( $\delta$  2.71 (dd,  $J = 2.57, 17.14$  Hz), 3.09 (dd,  $J = 13.4, 17.15$  Hz) and 5.62 (dd,  $J = 2.97, 13.04$  Hz), and proton of methoxyl group ( $\delta$  3.79), respectively.

The  $^{13}\text{C}$  NMR spectra data showed total 16 carbons, including 6 carbon groups from 16 signals, as carbonyl 1 signal ( $\delta$  198.92), aromatic carbon 14 signals ( $\delta$  94.87, 95.67, 103.48, 104.04, 107.81, 117.77, 128.90, 157.68, 159.88, 165.22, 165.25 and 169.46), olefin carbon 4 signals ( $\delta$  122.94, 123.64, 131.88 and 132.55), methane carbon 2 signals ( $\delta$  43.15 and 76.15) and methoxyl carbon 1 signals ( $\delta$  56.23), respectively.

From the spectra  $^1\text{H}$  NMR (Figure C4),  $^{13}\text{C}$  NMR (Figure C5), DEPT (Figure C6.1 and C6.2), COSY (Figure C7), HMQC (Figure C8) and HMBC (Figure C9)), physical property data and comparing with previously reports (Wei *et al*, 2005), compound M-3 was identified as Artocarpanone.

The complete  $^1\text{H}$  and  $^{13}\text{C}$  NMR assignments and the HMBC correlation are summarized in Table 4.8





**Table 4.8**  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of Compound M-3 (in Methanol- $d_4$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively)) and Artocarpanone (in Acetone- $d_6$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (400 and 100 MHz, respectively)) with long-range correlations in HMBC spectrum

Position	Compound M-3		Artocarpanone (Wei <i>et al.</i> , 2005)		Correlation with proton
	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity, $J$ in Hz)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity, $J$ in Hz)	
2	76.15	5.62 (dd, 13.04, 2.97)	75.9	5.75 (dd, 14.0, 3.0)	H-2
3	43.15	3.09 (dd, 17.15, 13.04)	43.0	3.21 (dd, 17.0, 3.0)	H-3
		2.71 (dd, 17.14, 2.51)		2.76 (dd, 17.0, 3.0)	H-3
4	198.92	-	198.5	-	-
4a	104.04	-	104.1	-	-
5	159.88	-	160.0	-	-
6	95.67	6.01 (d, 2.29)	95.7	6.02 (d, 2.2)	H-6
7	169.46	-	169.1	-	-
8	94.87	6.04 (d, 2.29)	94.8	6.05 (d, 2.2)	H-8
8a	165.22	-	165.1	-	-
1'	117.77	-	117.1	-	H-3'
2'	157.68	-	156.7	-	-
3'	103.48	6.33 (d, 2.29)	103.9	6.47 (dd, 2.0)	-
4'	165.25	-	165.4	-	-
5'	107.81	6.32 (dd, 8.23, 2.29)	108.3	6.43 (dd, 8.0, 2.0)	H-5'
6'	128.90	7.22 (d, 8.01)	129.4	7.32 (d, 8.0)	H-6'
5-OH	-	-	-	12.17	-
7-OCH <sub>3</sub>	56.23	3.79	56.6	3.85	

#### 4.2.4 Structure determination of compound H-1

The compound H-1 was obtained as wax. The UV-visible spectrum was maximum wavelength ( $\lambda_{\max}$ ) at 240 nm (Figure D1). The optical rotation showed at  $[\alpha]_D^{28}$  -28.57 (0.04 g/100 mL). In IR-Spectrum (KBr)  $\nu_{\max}$  (Figure D2) showed at 1374, 828 and 719  $\text{cm}^{-1}$  (tri-substituted of methine group), other group as 2971, 2849 and 1462  $\text{cm}^{-1}$  (a methylene group of alkane).

The  $^1\text{H}$  NMR 300 MHz (Figure D4) showed 3 proton groups, as olefin protons ( $\delta$  5.14 (dd)), methylene and methyl protons ( $\delta$  2.06-0.86). This may suggestion that the compound H-1 is a long chain alkene.

The  $^{13}\text{C}$  NMR 75 MHz (Figure D5) showed 3 carbon groups, as olefin carbon 2 signals ( $\delta$  124.3 and 134.3 ppm), methane and methyl 6 signals ( $\delta$  14.0-39.7 ppm).

The EI mass spectrum (Figure D3) occurred fragmentation also with alkene (Field, 1968) showed straight-chain 1-olefin.

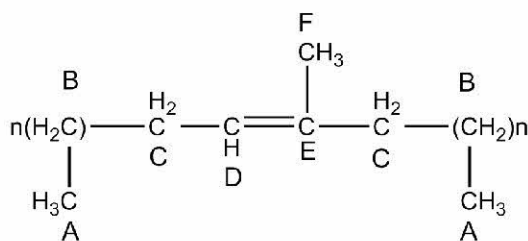


Figure 4.1 The estimation of structure H-1. (related to table 4.9)

Table 4.9 The estimation of  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR. (related figure 4.1)

Protons	$\delta$ ppm	Carbons	$\delta$ ppm
A	0.88-0.92	A	14.04
B	1.28	B	22-31
C	2.06	C	22-31
D	5.14	D	124.33
E	-	E	135.00
F	1.58	F	22-31

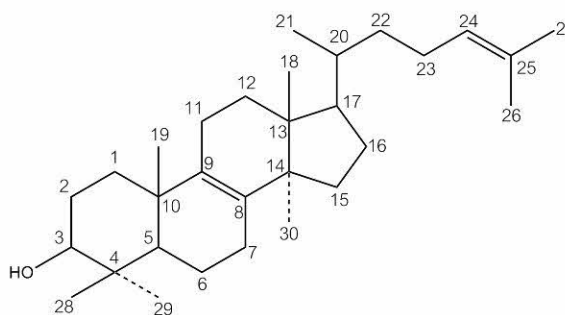
#### 4.2.5 Structure determination of compound H-2

The compound H-2 was obtained as colorless needles. The UV-visible spectrum was found maximum wavelengths at 222, 240 and 322 nm (Figure E1). The melting point showed 130-133 °C (138-140 °C (Marienthal and Franklin, 1995)). In IR-Spectrum (KBr)  $\nu_{\max}$  (Figure E2) showed at 3436  $\text{cm}^{-1}$  (OH). The EI mass spectrum (Figure E3) showed as molecular ion peak at  $m/z$  426 corresponding to  $\text{C}_{30}\text{H}_{50}\text{O}$ .

The  $^1\text{H}$  NMR spectra data (Figure E4) showed 4 proton groups as olefin protons ( $\delta$  5.12 (t)), methane proton ( $\delta$  3.25), methane and methyl protons ( $\delta$  0.77-2.19).

The  $^{13}\text{C}$  NMR spectra data (Figure E5) showed with 30 signals, including 3 carbon groups, as olefin carbon 3 signals ( $\delta$  125.25, 130.83 and 134.05), methane and methyl carbon ( $\delta$  15.43, 15.67, 17.65, 18.15, 18.57, 18.95, 21.44, 24.77, 24.97, 25.70, 26.41, 27.94, 28.04, 28.10, 30.91, 30.97, 35.44, 35.78, 35.85, 37.21, 38.93, 44.13, 49.66, 50.65 and 79.00).

From the NMR spectroscopic data, chromatograms seem to be Lanosterol. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectrum were compared with previously reports (Emmons, 1989; Sawai *et al*, 2006). The complete  $^1\text{H}$  and  $^{13}\text{C}$  NMR assignments are summarized in Table 4.10.



**Lanosterol**

**Table 4.10**  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of Compound H-2 (in  $\text{CDCl}_3$ -*d*;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (300 and 75 MHz, respectively)) and Lanosterol (in  $\text{CDCl}_3$ -*d*;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively))

Position	Compound H-2		Lanosterol (Emmons, 1989)	
	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)
1	35.44	-	35.55	-
2	27.94	-	27.81	-
3	79.00	3.25 (dd)	78.96	3.24 (dd)
4	38.93	-	38.85	-
5	50.65	1.15 (m)	50.35	1.05 (m)
6	18.15	1.53 (m)	18.22	1.63 (m)
7	26.41	-	26.46	-
8	134.05	-	134.35	-
9	134.05	-	134.35	-
10	37.21	-	36.97	-
11	21.44	-	20.97	-
12	30.97	-	30.94	-
13	44.13	-	44.43	-
14	49.66	-	49.76	-
15	30.91	-	30.82	-
16	28.10	-	28.18	-
17	50.65	1.49 (m)	50.35	1.48 (m)
18	15.67	0.77 (s)	15.71	0.68 (s)
19	18.95	0.97 (s)	19.11	0.98 (s)
20	35.78	1.38 (m)	36.24	1.39 (m)
21	18.57	0.87 (d)	18.61	0.91 (d)
22	35.85	-	36.33	-
23	24.97	-	24.89	-
24	125.25	5.12 (t)	125.22	5.10 (t)

**Table 4.10**  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of Compound H-2 (in  $\text{CDCl}_3-d$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (300 and 75 MHz, respectively)) and Lanosterol (in  $\text{CDCl}_3-d$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (500 and 125 MHz, respectively) (continued)

Position	Compound H-2		Lanosterol (Emmons, 1989)	
	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)
25	130.83	-	130.91	-
26	25.70	1.70 (s)	25.72	1.68 (d or s)
27	17.65	1.62 (s)	17.61	1.60 (d or s)
28	28.04	1.02 (s)	27.93	1.00 (s)
29	15.43	0.81 (s)	15.40	0.81 (s)
30	24.77	0.86 (s)	24.23	0.87 (s)

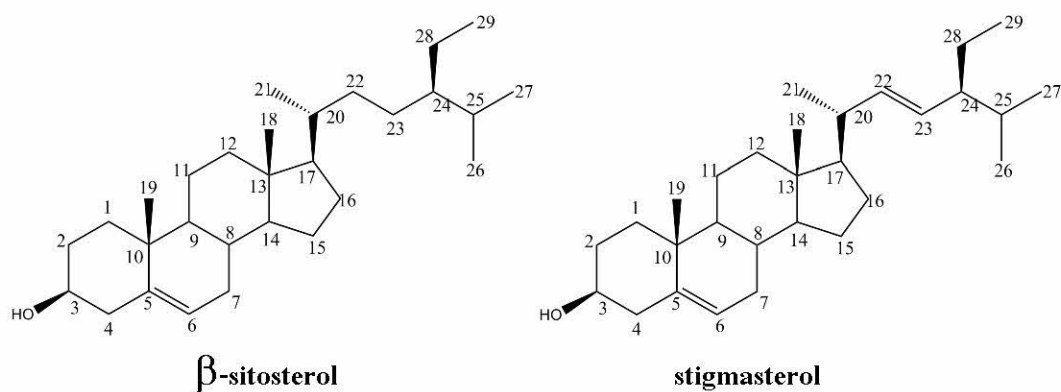
#### 4.2.6 Structure determination of mixture H-3

The mixture H-3 was obtained as colorless needles. The UV-visible spectrum was found maximum wavelengths at 278, 298 and 326 nm (Figure G). In IR-Spectrum (KBr)  $\nu_{\text{max}}$  (Figure F3) showed at  $3435\text{ cm}^{-1}$  (OH).

The  $^1\text{H}$  NMR (Figure F1) showed olefin proton 3 signals ( $\delta$  5.36 (m), 5.17 (dd), 5.02 (dd)), methine proton (3.54 (m)), methane and methyl protons (the remaining proton signals were 0.70-2.30). The integration steps of H-6, H-22 and H-23 were approximately in ratio of 0.893:0.217:0.217. Thus, H-3 was a mixture of  $\beta$ -sitosterol and stigmasterol in ratio of 7.5:2.5.

The  $^{13}\text{C}$  NMR spectrum (Figure F2) showed with 34 signals, including 3 carbon groups, as olefin carbon 4 signals ( $\delta$  121.68, 129.38, 138.28 and 140.77), methane and methyl carbon ( $\delta$  11.97, 12.21, 19.79, 21.04, 21.09, 21.20, 23.10, 24.29, 24.36, 28.88, 29.21, 31.68, 31.86, 31.92, 33.98, 36.14, 36.51, 37.27, 39.70, 39.80, 40.43, 42.32, 45.88, 50.17, 51.23, 56.00, 56.10, 56.78, 56.88 and 71.80).

The spectra were compared with previously reports of  $\beta$ -sitosterol and stigmasterol (Elgendy and Al-Ghamdy, 2007; Subhadhirasakul and Pechpongs, 2005; Sritularak, 1998a)



**Table 4.11**  $^{13}\text{C}$  NMR spectrum data of mixture H-3 (in  $\text{CDCl}_3$ -*d*;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (300 and 75 MHz, respectively)),  $\beta$ -sitosterol and stigmasterol (in  $\text{CDCl}_3$ -*d*;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (300 and 75 MHz, respectively))

Position	$\beta$ -sitosterol		stigmasterol		Mixture H-3	
	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)
1	36.2	-	39.0	-	36.14, 37.27	-
2	39.8	-	39.4	-	39.80, 39.70	-
3	71.8	3.50 (m)	70.5	3.52 (m)	71.80	3.54 (m)
4	42.3	-	50.6	-	42.32, 51.23	-
5	140.8	5.35 (s)	140.4	5.34 (s)	140.77	5.36 (d)
6	121.7	-	120.6	-	121.68	-
7	31.5	-	31.9	-	31.92	-
8	29.2	-	31.2	-	29.21, 31.86	-
9	45.8	-	49.5	-	45.88, 50.17	-
10	34.0	-	40.0	-	33.98, 40.43	-
11	21.1	-	20.6	-	21.09	-
12	31.9	-	36.7	-	31.92, 37.27	-
13	42.2	-	41.5	-	42.32	-
14	56.1	-	55.2	-	56.10, 56.00	-
15	24.3	-	24.8	-	24.29, 24.36	-
16	23.1	-	23.7	-	23.10	-
17	56.8	-	56.2	-	56.88, 56.78	-

**Table 4.11**  $^{13}\text{C}$  NMR spectrum data of mixture H-3 (in  $\text{CDCl}_3$ -*d*;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (300 and 75 MHz, respectively)),  $\beta$ -sitosterol and stigmasterol (in  $\text{CDCl}_3$ -*d*;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (300 and 75 MHz, respectively)) (continued)

Position	$\beta$ -sitosterol		stigmasterol		Mixture H-3	
	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{H}}$ (ppm) (multiplicity)
18	19.1	0.69 s	20.4	0.69 s	19.79, 21.09	0.70, 0.71 s
19	19.4	1.01 s	20.5	1.01 s	21.04, 21.20	1.02 s
20	28.3	-	31.0	-	28.88, 31.68	-
21	12.0	0.87 d	11.7	1.02 d	12.21, 11.97	0.93, 1.09 d
22	37.3	-	128.5	5.00 dd	37.27, 129.31	5.02 dd
23	36.5	-	137.7	5.14 dd	36.51, 138.04	5.17 dd
24	50.1	-	41.7	-	50.17, 42.32	-
25	31.7	-	35.9	-	31.86, 36.14	-
26	18.3	0.93 d	18.4	0.79 d	18.78, 18.98	0.94, 0.79 d
27	18.7	0.93 d	18.8	0.84 d	19.04, 19.38	0.94, 0.81d
28	26.1	-	28.3	-	26.16, 28.23	-
29	11.9	0.84 dd	11.2	0.80 t	12.04, 11.85	0.83 dd, 0.80 t

#### 4.3 Anti-tyrosinase activity of pure compounds

Tyrosinase inhibitory activity of pure compounds were showed in table 4.12. The Artocarpanone (M-3) was the most potent anyi-tyrosinase activity with  $\text{IC}_{50}$  44.56  $\mu\text{g}/\text{mL}$ .

Artocarpanone is a compound of flavonoid group and related 4-substituted resorcinols, suggested that compounds with the 4-substituted resorcinol skeleton (see the figure 4.2) have potent tyrosinase inhibitory activity (Kim and Uyama, 2005). However, the additional group (such as hydroxyl and prenyl) at the 3 position (see the figure 4.2) also somewhat affected the activity (Kim and Uyama, 2005; Shimizu *et al*, 2000). Thus, Artocarpin (M-1) and Cudraflavone C (M-2) did not showed tyrosinse inhibition.

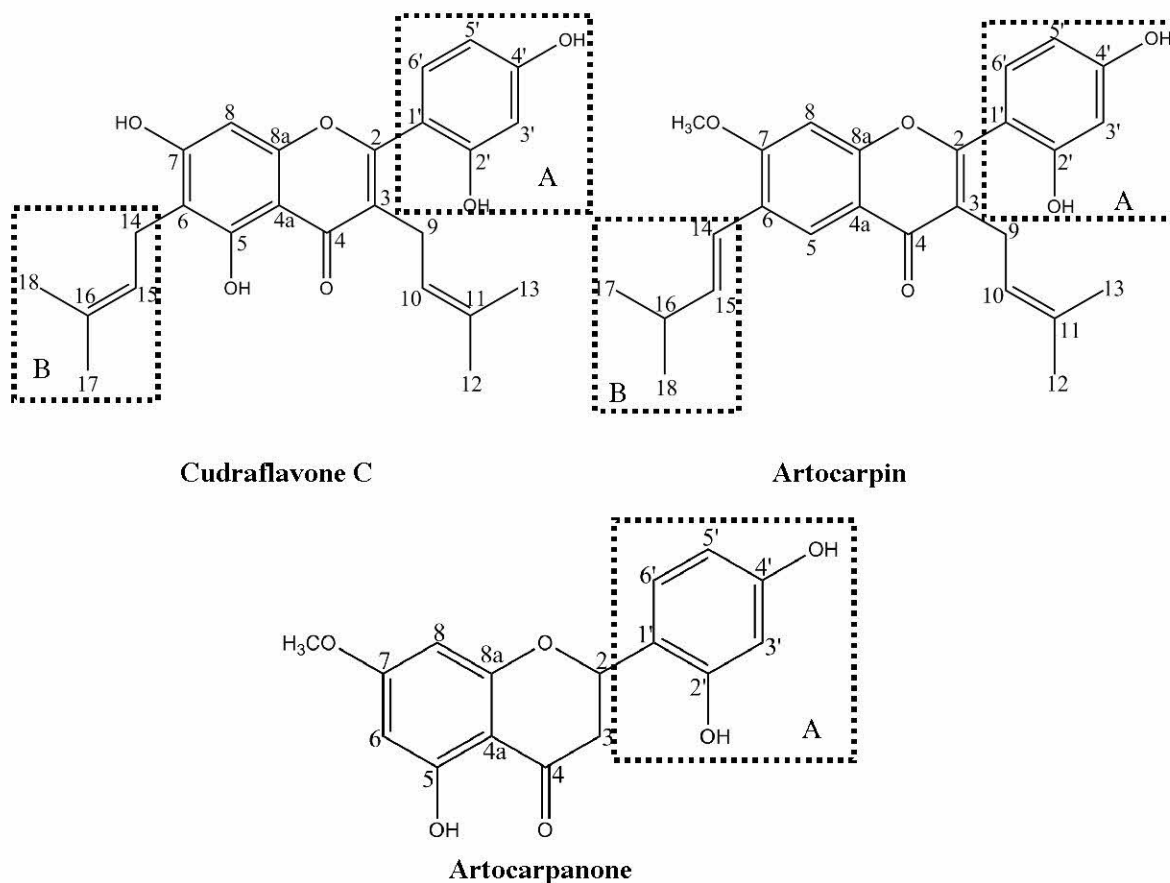
**Table 4.12 IC<sub>50</sub> of pure compounds on tyrosinase inhibitory activity**

Compound	IC <sub>50</sub> (µg/mL)
Artocarpin (M-1)	> 200
Cudraflavone C (M-2)	> 200
Artocarpanone (M-3)	44.56
Long chain alkene (H-1)	> 200
Lanosterol (H-2)	> 200
Mixture of β-sitosterol and stigmasterol (H-5)	> 200
Kojic acid	31.43
<i>Artocarpus lakoocha</i> <sup>W</sup>	6.41
<i>Artocarpus integer</i> <sup>E</sup>	10.81

W = water wood extract; E = ethanol root extract

In the other hand, H-1 (long chain alkene), H-2 (mixture of β-sitosterol and stigmasterol) and H-3 (lanosterol), may be not showed functional group or side chain for interaction with active site of tyrosinase. Thus, these compounds did not show tyrosinase inhibition.





**Figure 4.2** The chemical structures of Cudraflavone C, Artocarpin and Artocarpanone

The A box part: 4-substituted resorcinol skeleton

The B box part: long chain aliphatic group

#### 4.4 Anti-microbial activity of pure compounds

Artocarpin (M-1) and Cudraflavone C (M-2) showed strong inhibitory effect against *Propionibacterium acnes*, *Staphylococcus aureus* and *S. epidermidis*. Artocarpin and Cudraflavone C are flavonoid compounds. The substituents of flavonoid as long chain aliphatic group at 6 or 8 positions and hydroxyl group at 2' and 4' positions (see figure 4.2) are important for anti-bacterial activity (Cushnie and Lamb, 2005). The MIC and MBC values of compounds showed in Table 4.13

**Table 4.13 MIC and MBC of pure compounds against *P. acnes*, *S. aureus* and *S. epidermidis***

Compound	<i>P. acnes</i>		<i>S. aureus</i>		<i>S. epidermidis</i>	
	MIC	MBC	MIC	MBC	MIC	MBC
	µg/mL		µg/mL		µg/mL	
M-1	2	32	2	32	4	32
M-2	2	32	2	32	4	32
M-3	32	64	64	256	64	128
H-1	128	128	128	>256	128	256
H-2	128	128	128	256	128	256
H-3	128	128	128	256	128	256
Oxacillin	0.062	-	0.5	-	0.5	-

Even though, Artocarpin did not show tyrosinase inhibitory activity. In previously report, it showed cytotoxic activity and inhibitory activity on melanin biosynthesis on B16 melanoma cells (Arung *et al*, 2008). Cudraflavone C (M-2) has the similar skeleton to Artocarpin, so it might be interested for studies of anti-melanogenesis and cytotoxic. By the way, Artocarpanone has been reported that it has inhibitions of both tyrosinase activity and melanin production in B16 melanoma cells (Arung *et al*, 2006). Therefore, Artocarpin, Cudraflavone C and Artocarpanone are interesting to further study for whitening agent and anti-acne.

## CHAPTER 5

### CONCLUSION

In this investigation, the crude extracts from family Moraceae showed the potential inhibitory effects of anti-tyrosinase and anti-microbial activities, such as *Artocarpus integer*, *Ficus racemosa* and *Cudrania javanesis*. The root of *A. integer* was selected for phytochemical investigation because it showed the highest potential of anti-tyrosinase and anti-microbial activities. Five pure compounds and one mixture compound were isolated from the root of *A. integer*.

Three pure compounds as Artocarpin (M-1), Cudraflavone C (M-2) and Artocarpanone (M-3) were isolated from methanol extract. The other two pure compounds and one mixture compound as Long chain alkene (H-1), Lanosterol (H-2) and mixed of  $\beta$ -Sitosterol with Stigmasterol (H-3), they were isolated from hexane extract.

Artocarpanone showed the potential of anti-tyrosinase activity. While Artocarpin and Cudraflavone C showed anti-microbial activity. Whereas, the other compounds as Long chain alkene, Lanosterol and the mixture of  $\beta$ -Sitosterol and Stigmasterol showed low activities on anti-tyrosinase and anti-microbial.

Thus, Artocarpin, Cudraflavone C and Artocarpanone are interesting for further study in order to provide possibilities for the development of new whitening and anti-acne agents from *A. integer*.

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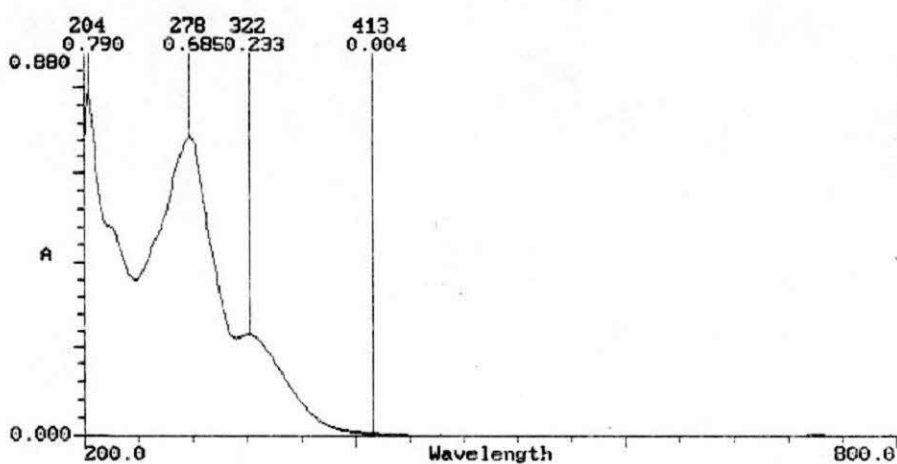
Wetwitayaklung, P. 1994. Chemical constituents from the heartwood of *Artocarpus lakoocha* Roxb. **Master's thesis**, Department of Pharmacognosy, Graduate School, Chulalongkorn University.

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**APPENDIX**

TEST SETUP  
GENESYS 6 v2.001 2M9M008001

Scanning 23:11 4May10  
 Test Name C-3-3  
 Measurement Mode Absorbance  
 Start Wavelength 200.0nm  
 Stop Wavelength 800.0nm  
 Sample Positioner Auto 6  
 Scan Speed Fast  
 Interval 1.0nm  
 Cell Correction Off  
 ID# (0=OFF) 1  
 Auto Print On  
 Auto Save Data Off



ID#: 1  
 Smoothing [On]

Wavelength	Abs	
204.0	0.790	Peak
278.0	0.685	Peak
322.0	0.233	Peak
413.0	0.004	Peak

Figure A1 UV-Visible spectrum of M-1 in Methanol



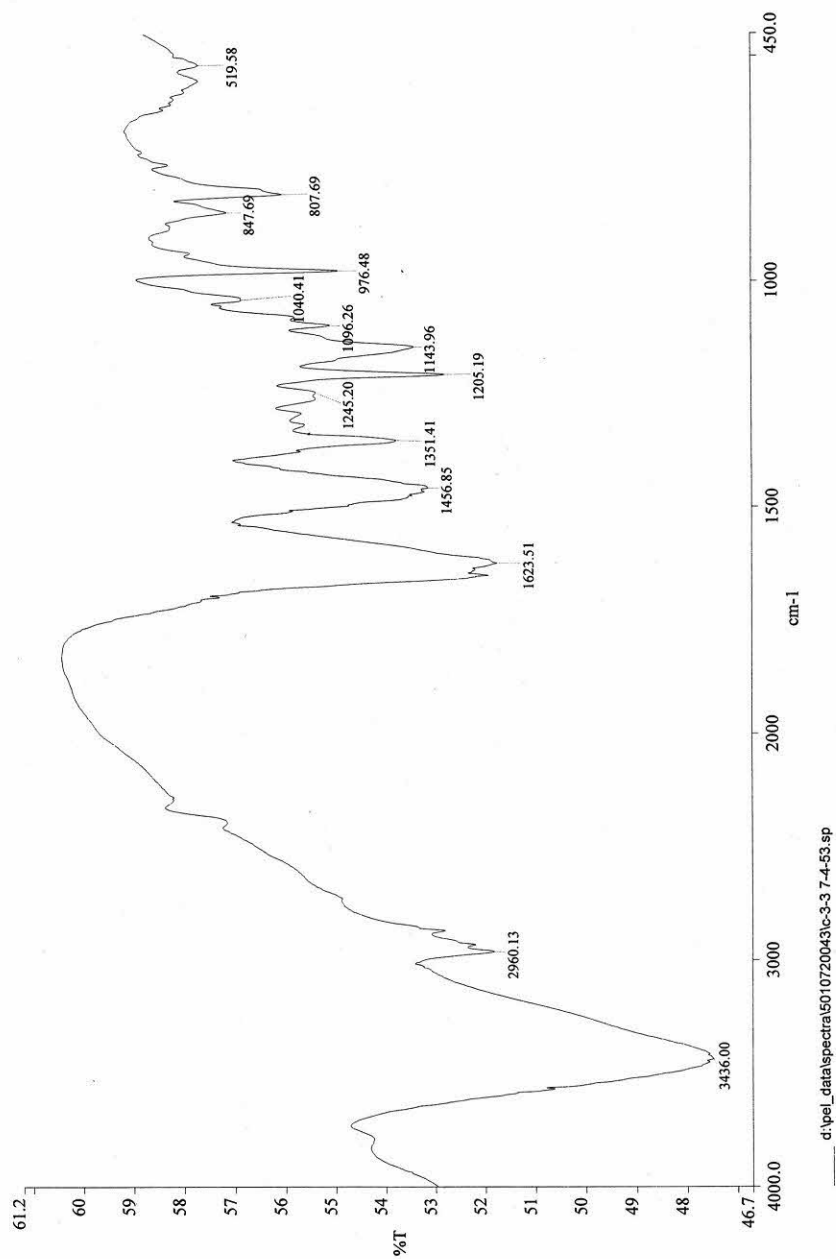


Figure A2 IR spectrum of compound M-1 (KBr disc)

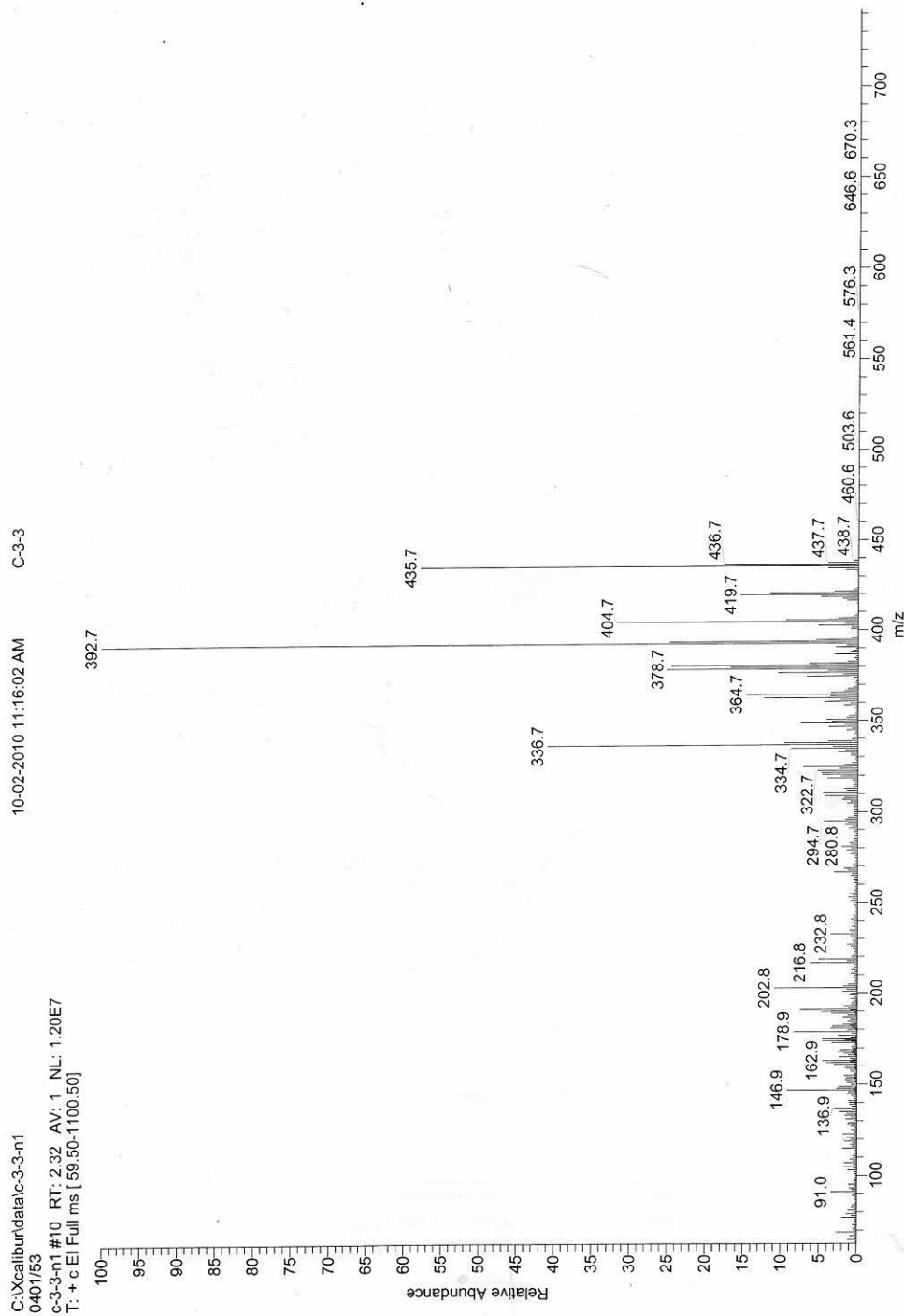
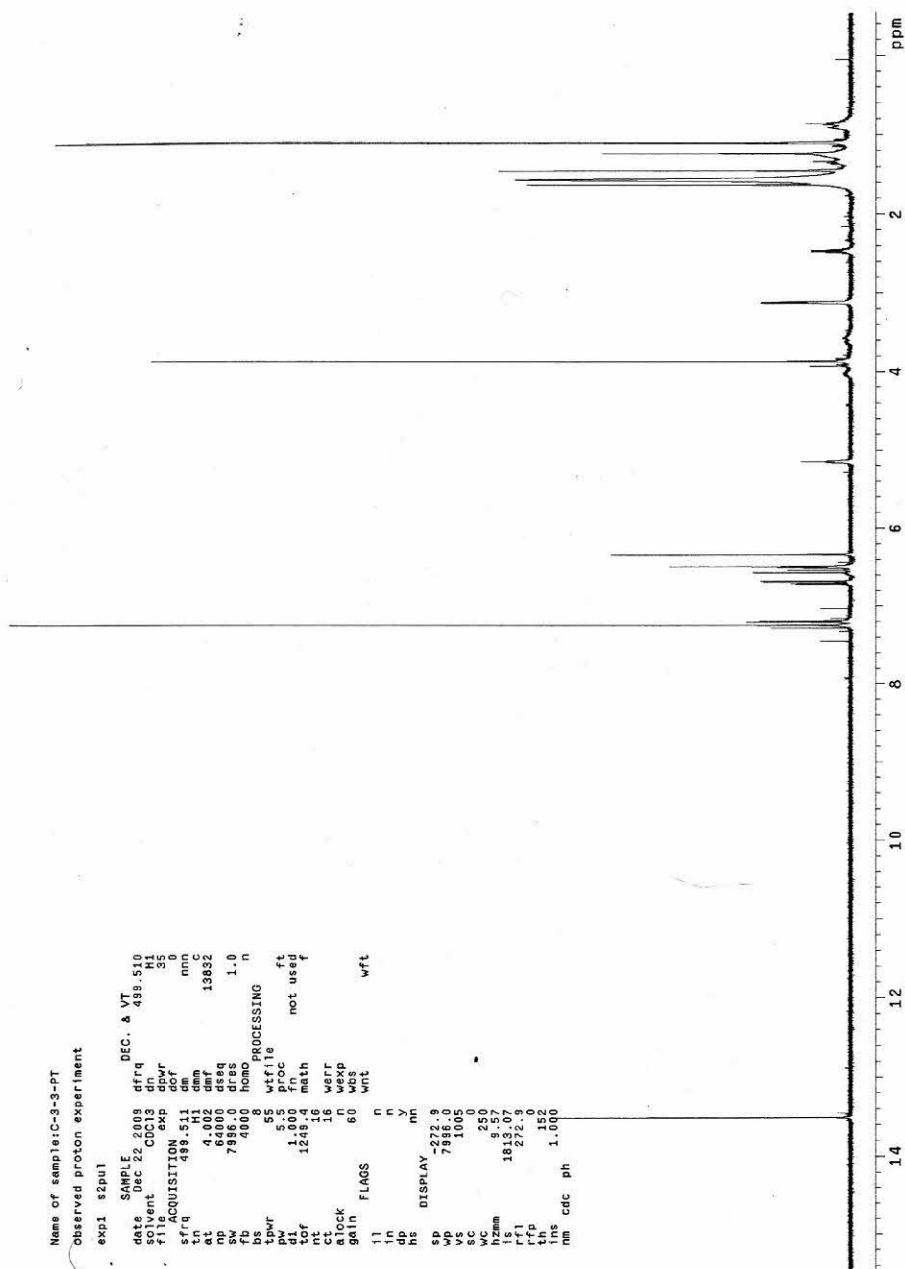
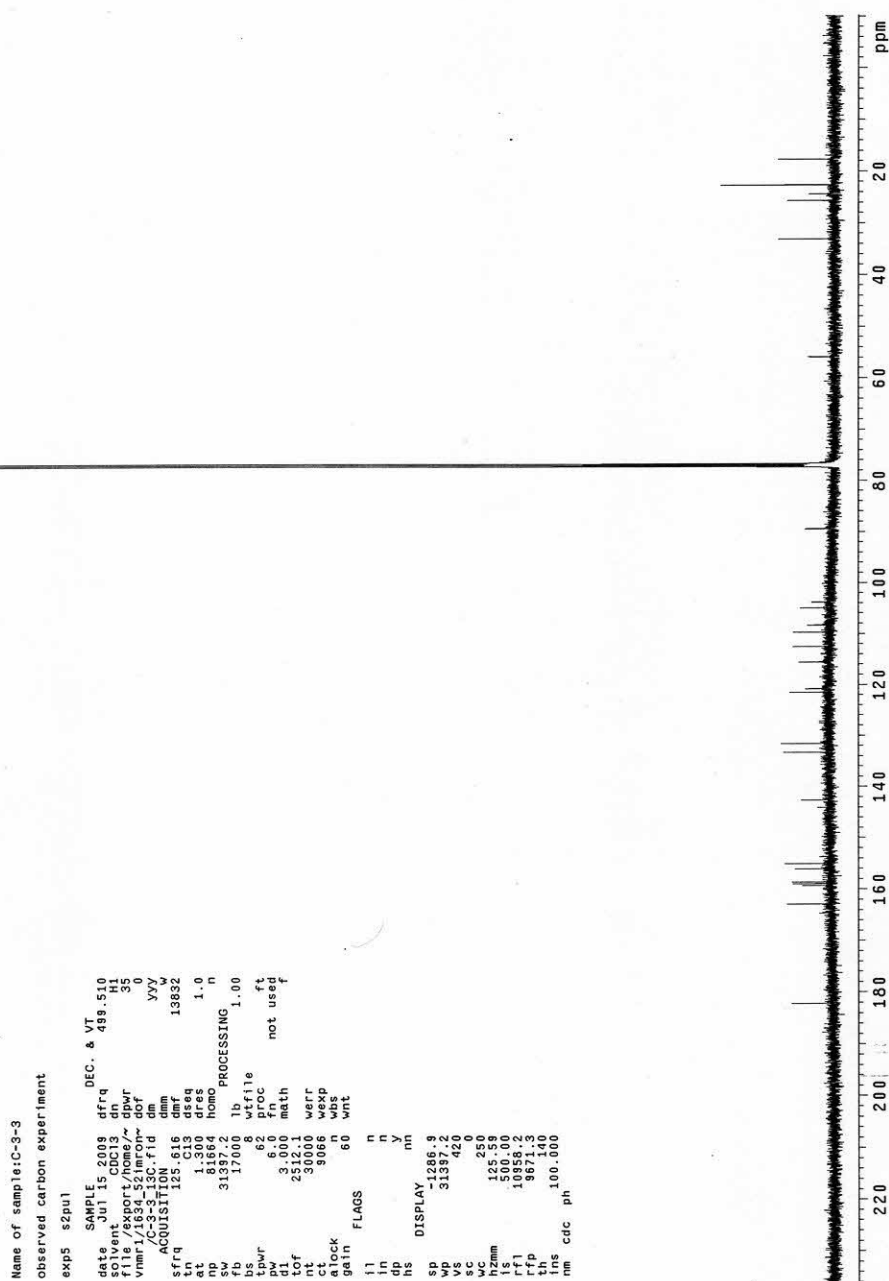


Figure A3 EI mass spectrum of compound M-1

Figure A4 500 MHz  $^1\text{H}$  spectrum of compound M-1 (in chloroform  $-d$ )

Figure A5 125 MHz  $^{13}\text{C}$  spectrum of compound M-1 (in chloroform  $-d$ )

```

Name of sample:C-3-3
DEPT90 experiment
CH only
exp2 dept
SAMPLE
date: Jul 21 2000          DEC. & VT 439.510
solvent: cdc13            dn 38
file: cdc13                dpwr 38
ACQUISITION exp 0
sfreq 125.611             nmv/
                                11768
at 1.301                  dmf
np 58816                   dseq
sw 22605.3                 dres 1.0
bs 12000                   homo PROCESSING n
ss 8                        lb 1.00
tpwr s4                    wtfile 1.00
ft 2000                     proc
tof -1739.0                math not used
nt 2000                     werr
ct 1531                     wexp
gain 60                     was
ppv1 10.5                   wnt
mult 14.61
satdly 1.0
ll n
ln y
dp y
hs nh
SP DISPLAY 1142.7
wp 22605.3
vs 3820
sc 0
sz 25
hzmm 90.42
ls 500.00
rfl 1905.8
tpp 17913.6
ins 100.000
al cdc ph
  
```

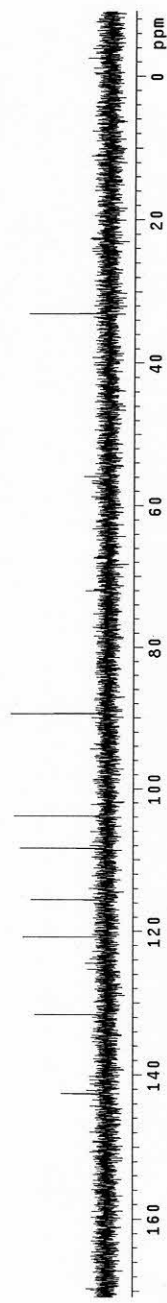
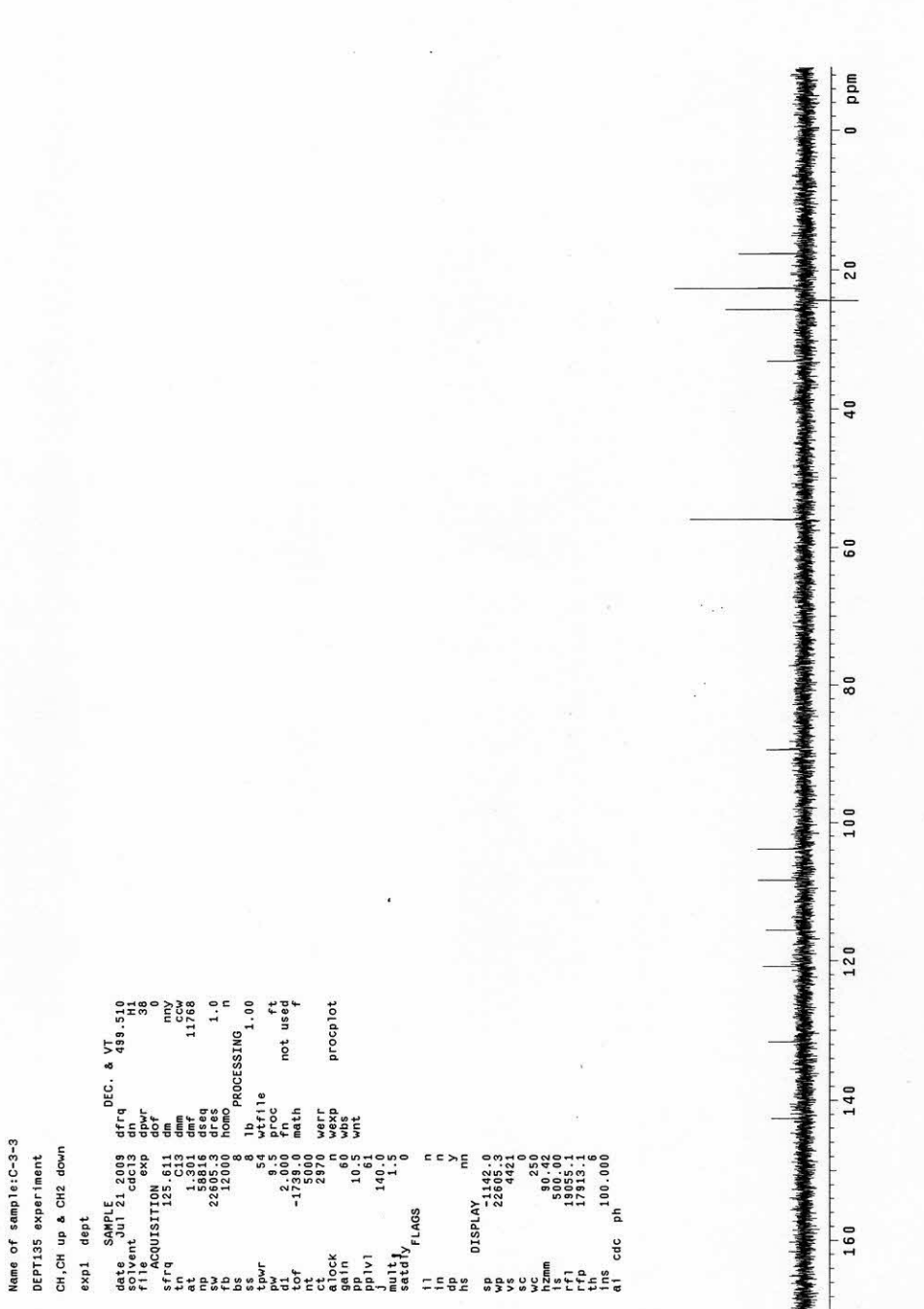


Figure A6.1 DEPT 90 spectrum of compound M-1 (in chloroform -d)



```

Name of sample: C-3-3
DEPT135 experiment
CH, CH up & CH2 down
exp1 dept
SAMPLE
date   JUL 21 2009      DEC. & VT
time   12.50           499.510
file   c3exp          36
dpr    0
dof    0
$frq   125.611        nny
at     1.303          dm
np     58816          dseq
sw     22605.3        dres
bs     12000          homo
ss     8             lb
tpwr   54            wtfile
pt     2.00          proc
tof    -1739.0       math
nt     5000         warr
ct     2870         werr
gain   60           wexp
pp     10.5         wnt
pplvl  140.0
mult   1.0
sctfy  1.0
-----
ll     n
lp     y
dp     y
hs     nn
$P     DISPLAY
wp     1142.0
vs     22605.3
v8     4421
sc     0
sc     0
mc     80.40
mm     500.00
rf1    19055.1
rfp    17913.1
lms    100.000
at     cdc ph
    
```

Figure A6.2 DEPT 135 spectrum of compound M-1 (in chloroform -d) (continued)

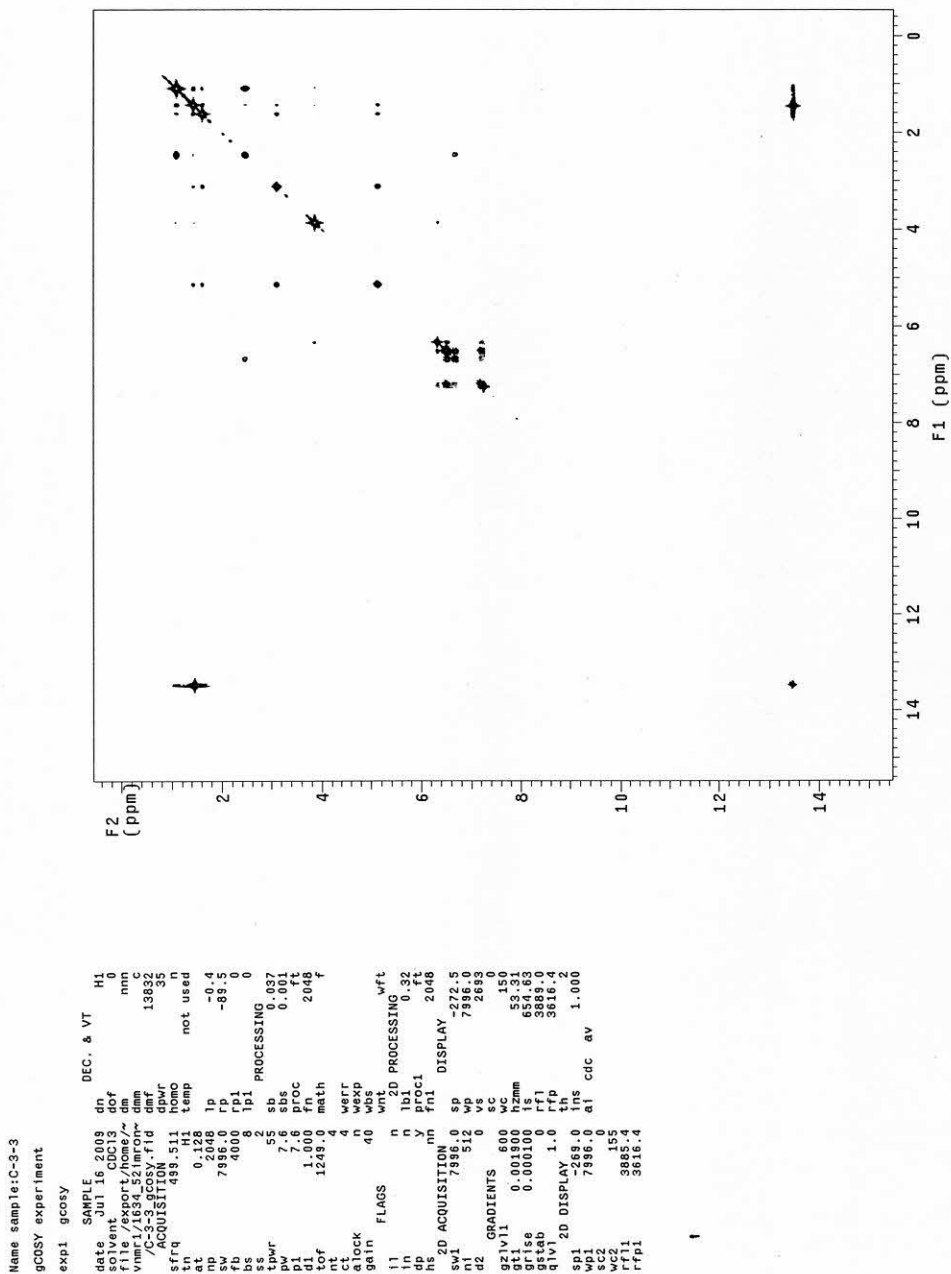


Figure A7 COSY spectrum of compound M-1 (in chloroform -d)

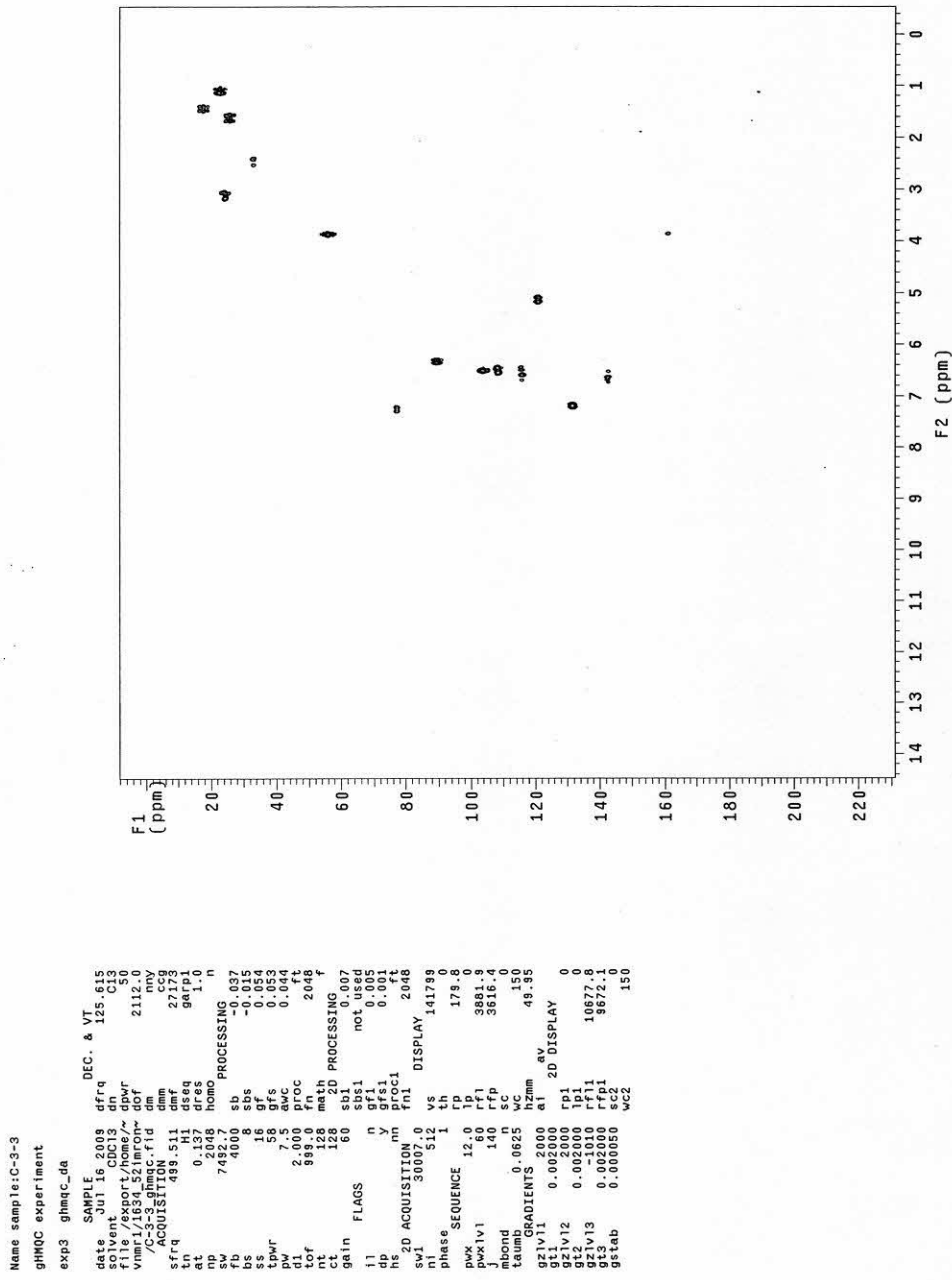


Figure A8 HMOC spectrum of compound M-1 (in chloroform -d)



Name: sample:C-3-3  
 gHMBC experiment  
 using ghmasc pulse sequence  
 exp2 ghmasc\_da

```

SAMPLE DEC. & VT
date Jul 17 2009 dfrq 125.615
solvent CDCl3 dnmr C13
file ACQUISITION sap ddr 2112.0
sfrq 489.511 dm nnn
tn 0.151 dmp 27.59
ns 40048 dsag 949.1
sw 7482.7 dras 1.0
fb 40000 homo PROCESSING n
ss 16 sb -0.087
spwr 58 sbs -0.051
pw 7.5 gf 0.038
d1 10.0 gfs 0.038
dd 589.0 gfd 0.084
nt 256 proc 2048
ct 240 fn 2048
gain flags 50 math PROCESSING f
tl n sbl 0.007
dp y sbs1 not used
hs 20 ACQUISITION nm gfl 0.005
prfl 30007 0 prcl 0.041
nl 512 fn1 2048
phase sequence 1 us DISPLAY 141798
pwx sequence 12.0 ts 141798
pwxlv1 160 fp 179.8
j 140 lp 0
mbond 3882.8
teumms 0.0625 rfl 3819.0
GRADIENTS SC 150
g2lv11 2000 wc 150
gt1 0.002000 hzmm 49.95
g2lv12 0.002000 at 20 DISPLAY
g2lv13 -1010 rp1 0
gt3 0.002000 rp1 10637.4
gstab 0.000050 rfl1 3872.4
          sc2
          wc2 150
  
```

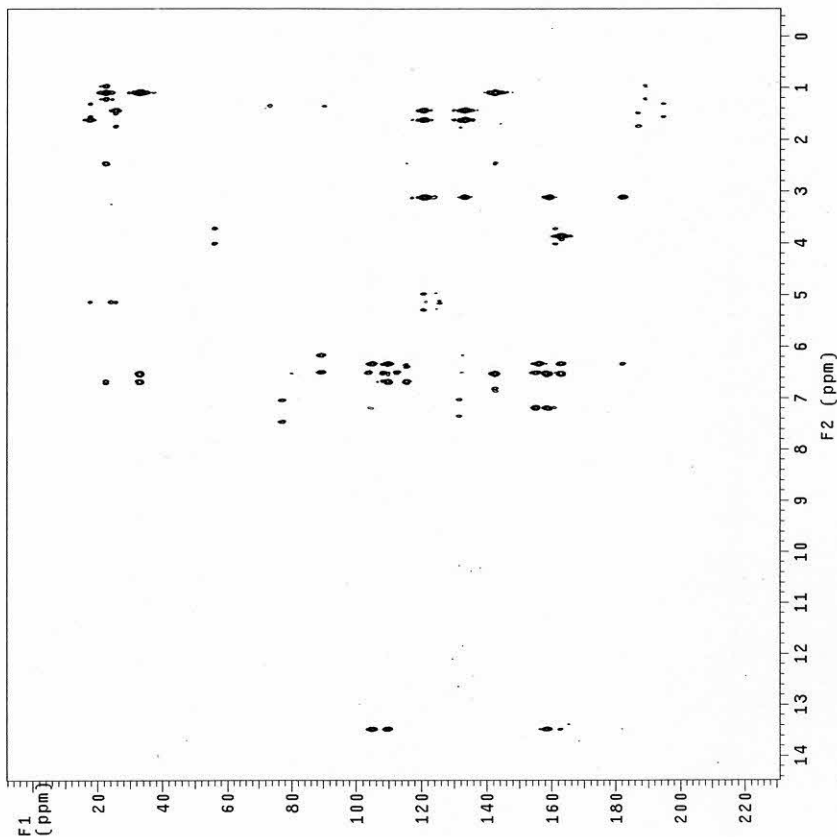
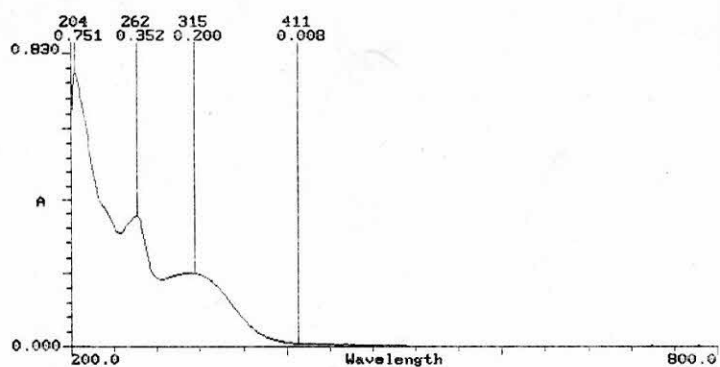


Figure A9 HMBC spectrum of compound M-1 (in chloroform -d)

TEST SETUP  
GENESYS 6 v2.001 2M9M008001

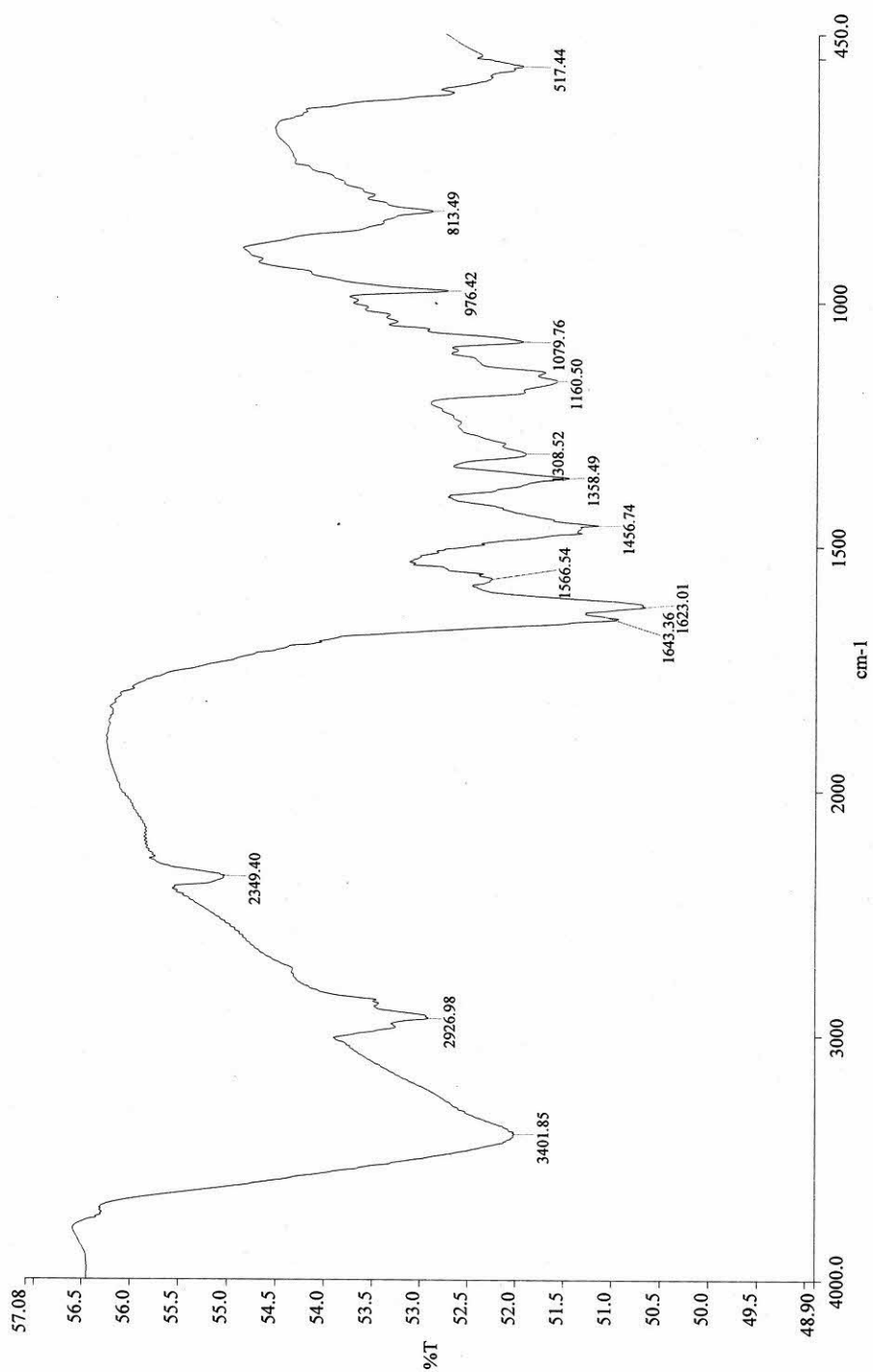
Scanning 23:18 4May10  
Test Name D-3-1a  
Measurement Mode Absorbance  
Start Wavelength 200.0nm  
Stop Wavelength 800.0nm  
Sample Positioner Auto 6  
Scan Speed Fast  
Interval 1.0nm  
Cell Correction Off  
ID# (0=OFF) 1  
Auto Print On  
Auto Save Data Off



ID#: 1  
Smoothing [On]  
Wavelength Abs

Wavelength	Abs	
204.0	0.751	Peak
262.0	0.352	Peak
315.0	0.200	Peak
411.0	0.008	Peak

Figure B1 UV-Visible spectrum of M-2 in Methanol



d:\pel\_data\spectra\c-3-1a.001

Figure B2 IR spectrum of compound M-2 (KBr disc)

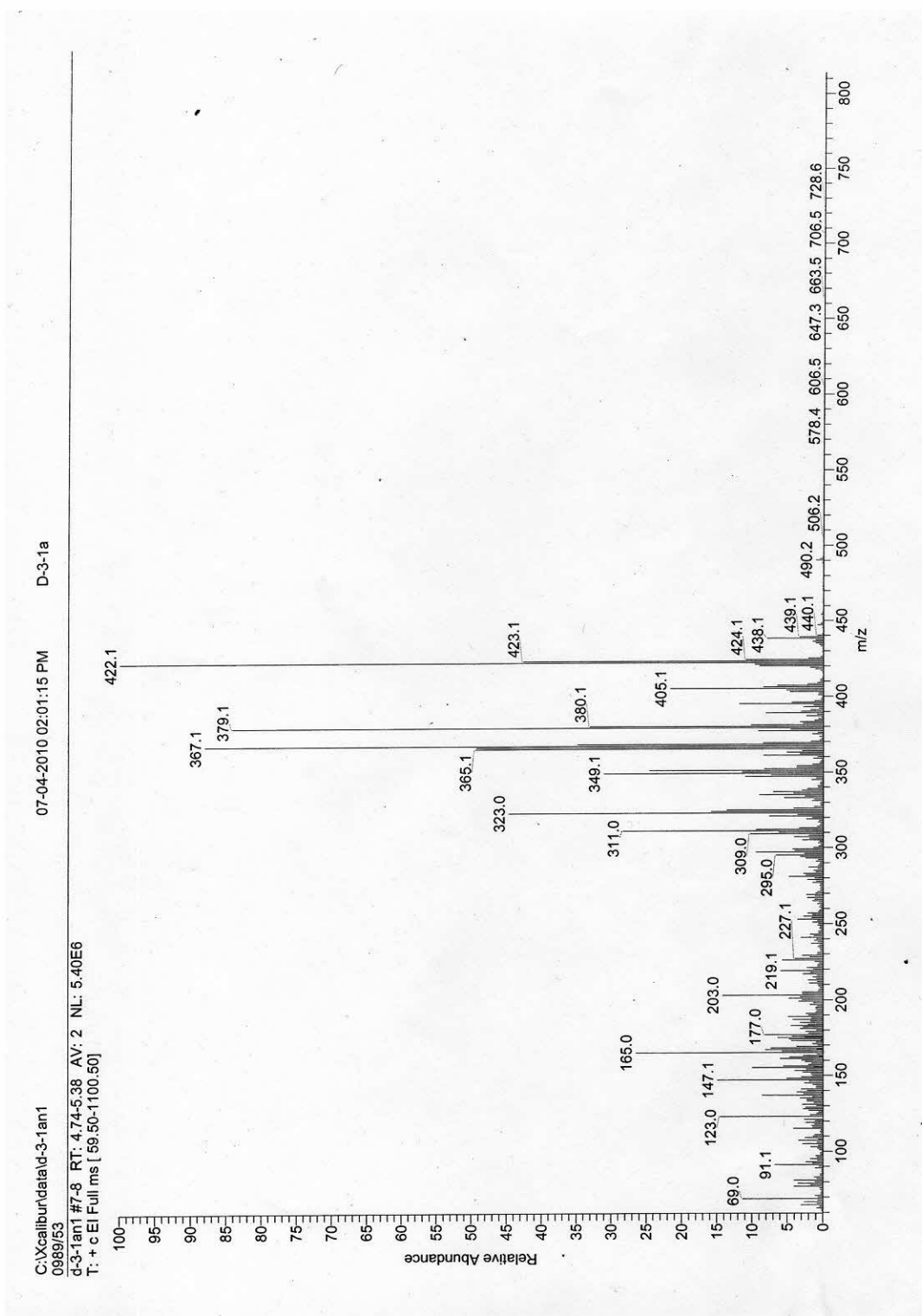


Figure B3 EI mass spectrum of compound M-2

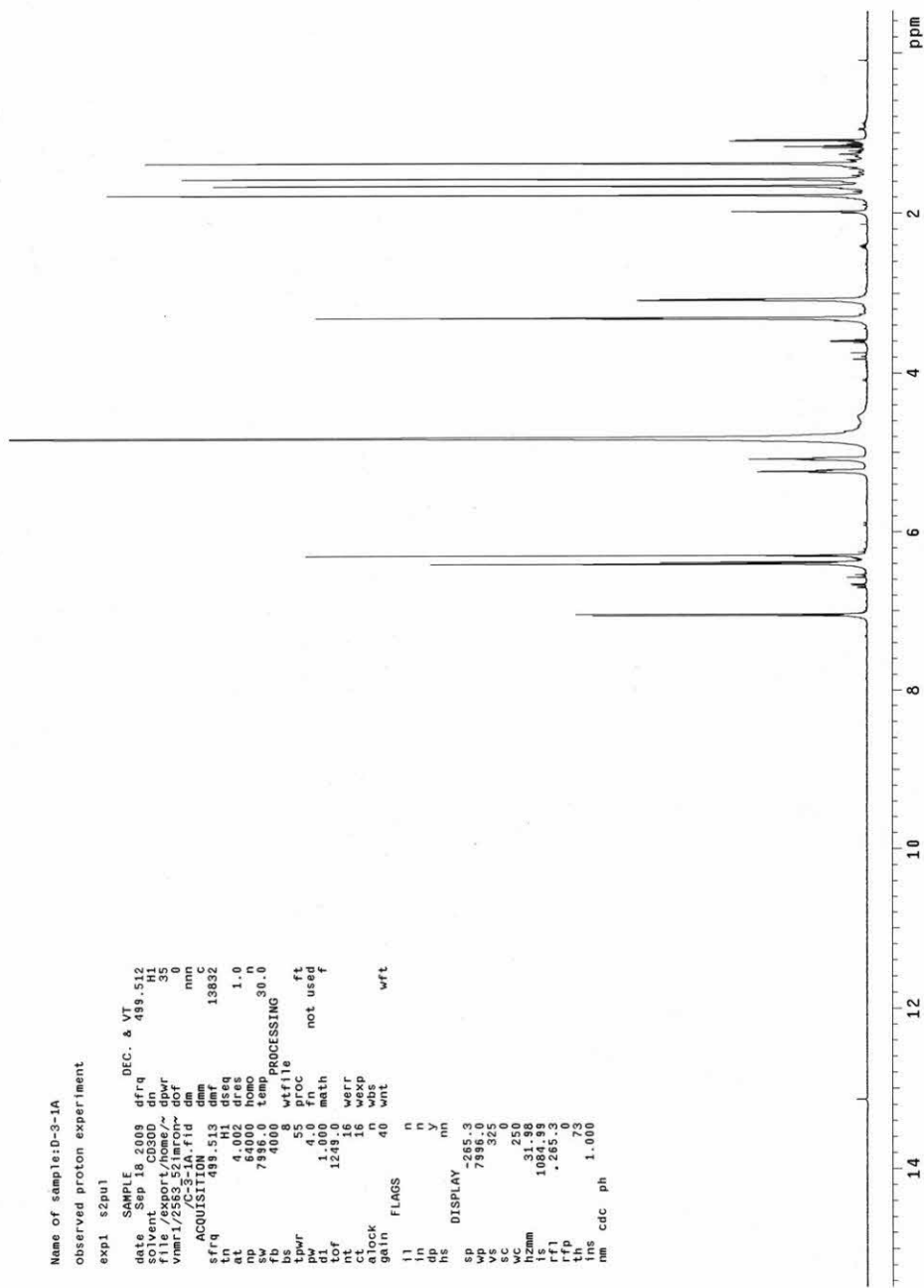


Figure B4 500 MHz <sup>1</sup>H spectrum of compound M-2 (in methanol -d<sub>4</sub>)

```

Name of sample: D-3-1A
observed carbon experiment
exp4 s2put
SAMPLE DEC. & VT
date Sep 28 2009 dfrq 499.512
dir /mnt/CDROM/11
file /export/home/~dpcwr
vmr1/2696.521mron~ dof 38
/D-3-1A-13C.fid dm yyy
ACQUISITION dmm 11768
sfreq 125.616 dseq
at 1.308 dres 1.0
np 8212 homo 30.0
fb 17000 temp 30.0
bs 8 lb 1.00
tpwr 54 wtfile
dl 3.000 proc ft
tof 2512.0 math not used
nt 10000
clock 556 werr
gain wbp
gain 60 wnt
FLAGS n
ll y
dp y
hs mn
SP DISPLAY
sp -1110.0
vs 3139.418
wc 0
wmm 250
ts 500.00
rf1 7264.4
rfp 6154.5
th 133
tms 100.000
nm cdc ph

```

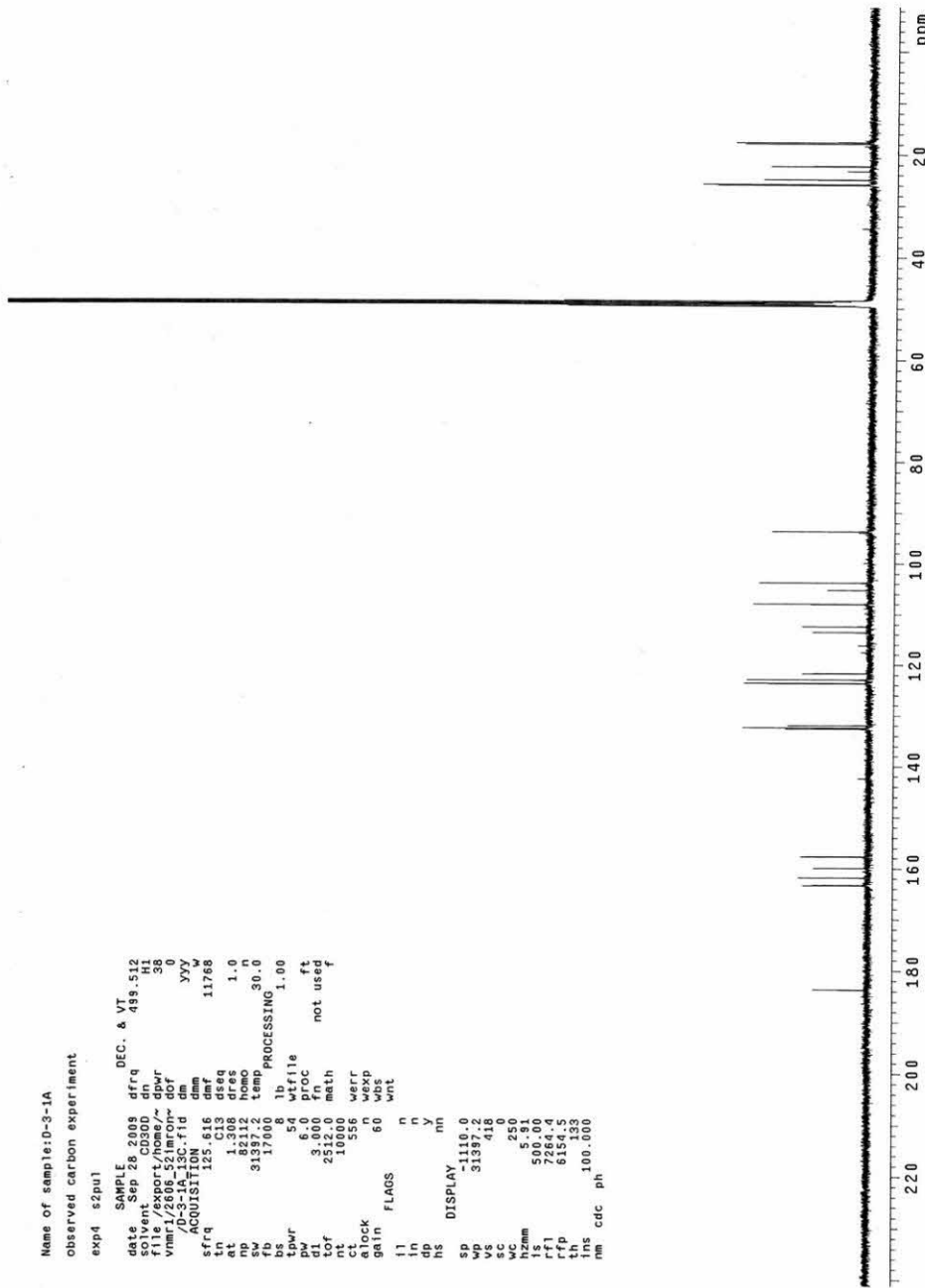


Figure B5 125 MHz <sup>13</sup>C spectrum of compound M-2 (in methanol -d<sub>4</sub>)

Name of sample: D-3-1A  
 DEPT 90 experiment  
 CH only

```

exp5  dept
SAMPLE      DEC. & VT
date       Sep 29 2009  dfrq      499.512
solvent    CD300  dn      H1
file       /export/home/~ dpwr      35
nmr1/2606_22imrn~ dor      nm0
/0-~      DEPT-90  dnm      nm0
ACQUISITION dm      ccv      13832
sfrq      125.612  dmf
tn        C13  deeq      1.0
nt        1.00  hres      30.0
sv        53504  hmc0
wb        20581.4 temp      30.0
bs        11000  lb  PROCESSING
ss        8      vrf1le      1.00
tpwr      62  proc      ft
pw        11.5  fn      not used
di        -20200  math
nt        256  wexp      procp1ot
ct        256  wexp
atlock    n  wbs
gain      6  wnt
pplv1     7.33
J         140.0
mult      1.0
sadtly   FLAGS
ll        n
ln        n
hp        y
hs        mh
SP        -266.4
wp        20551.2
SC        1212
WC        250
hzmh      82.33
rf1       16891.8
rff       16625.5
th        100.000
d1  cdc  ph
  
```

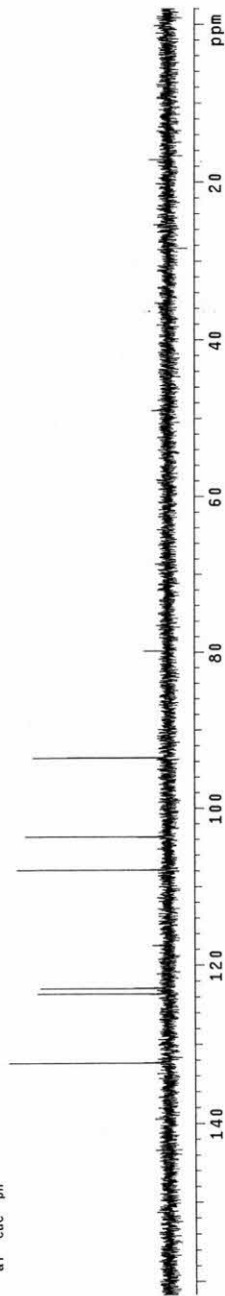


Figure B6.1 DEPT 90 spectrum of compound M-2 (in methanol -d<sub>4</sub>)





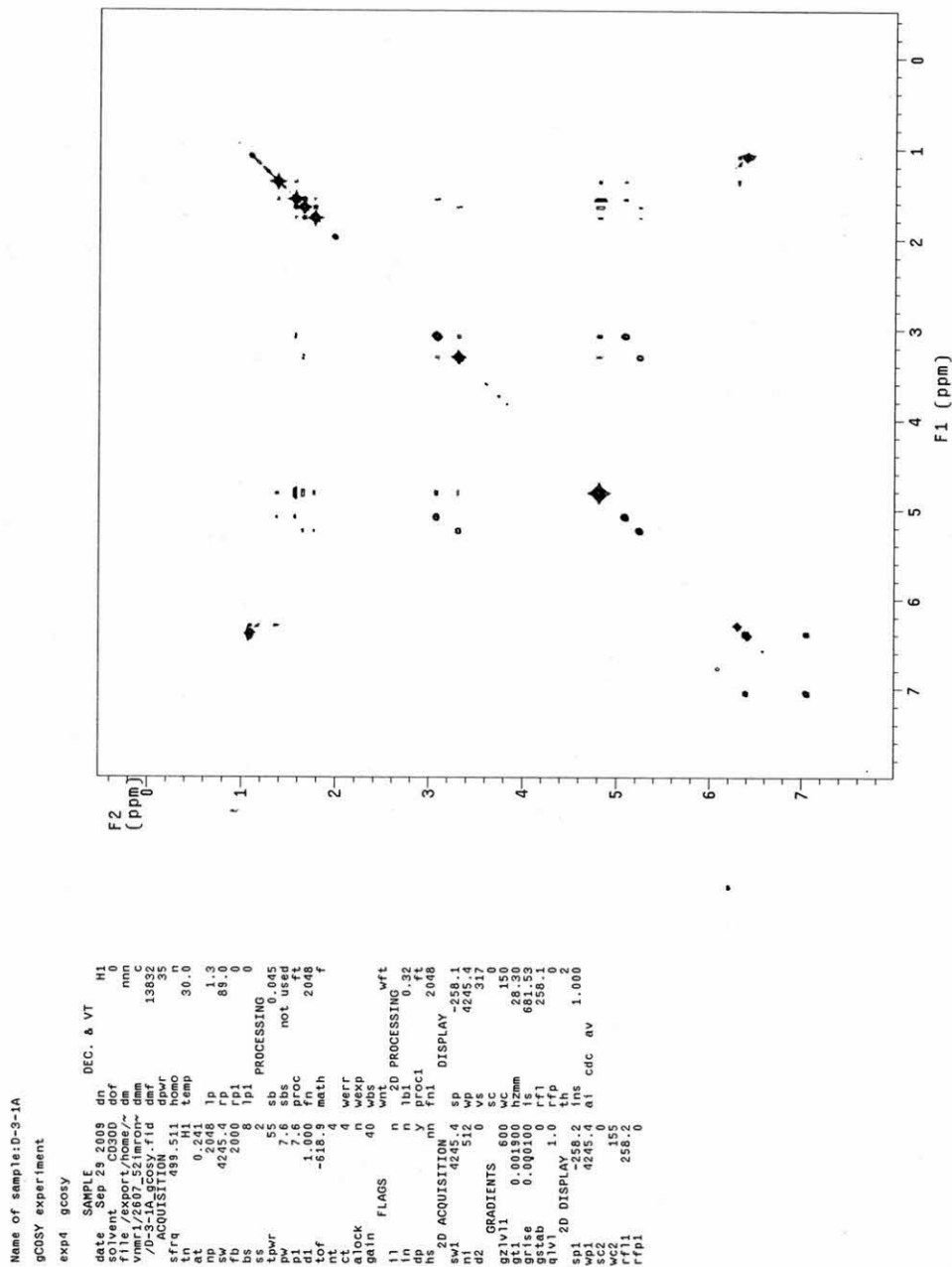


Figure B7 COSY spectrum of compound M-2 (in methanol -d<sub>4</sub>)

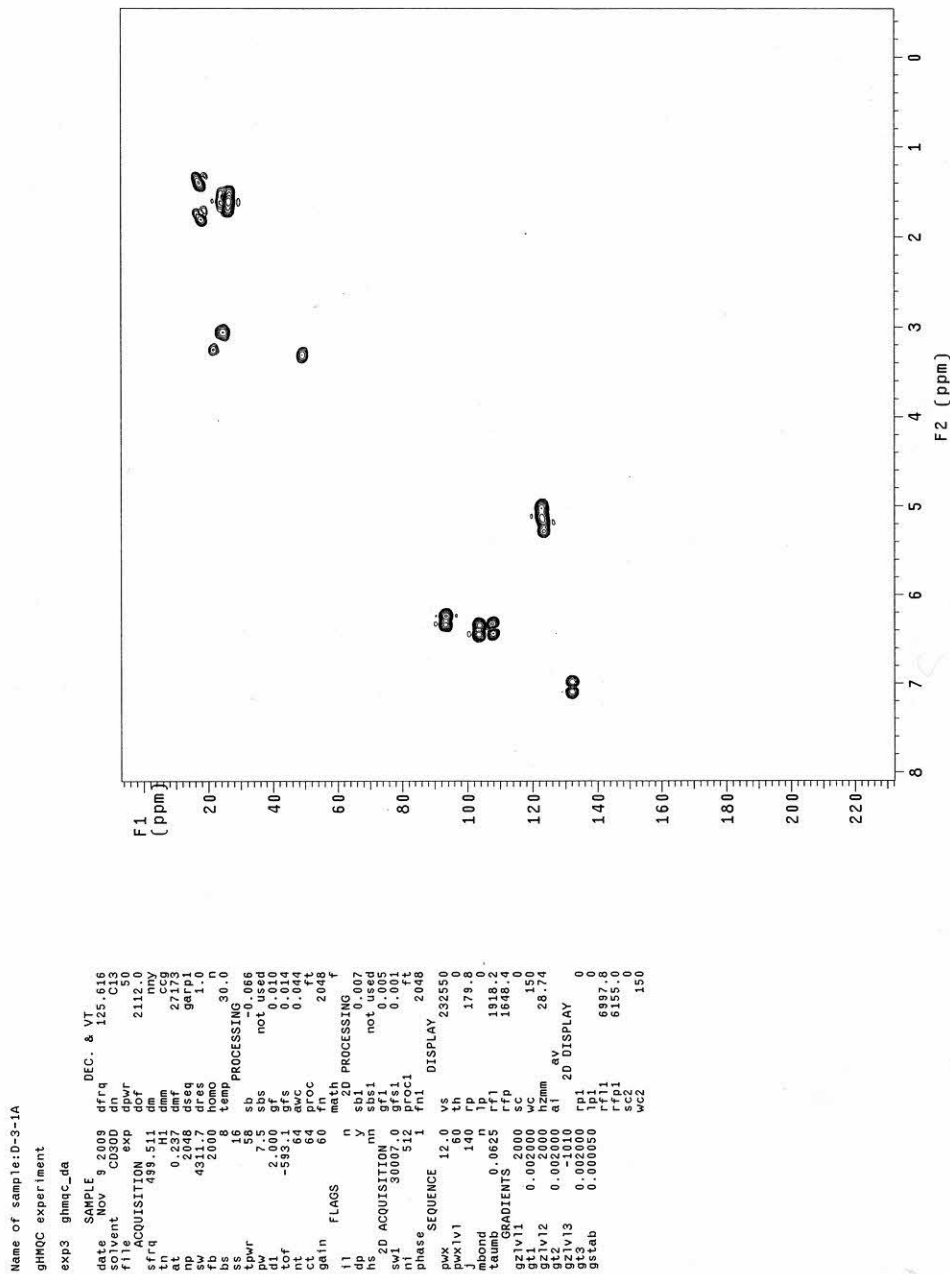


Figure B8 HMQC spectrum of compound M-2 (in methanol -d<sub>4</sub>)

Name of sample: D-3-1A  
 gHMBC experiment  
 using ghmqc pulse sequence  
 exp2 ghmqc\_da

SAMPLE DEC. & VT  
 date Nov 9 2009 dfrq 125.618  
 solvent CDCl<sub>3</sub> CS  
 F1 ACQUISITION SVP DOF 2112.0  
 sfrq 489.511 dm  
 ln 0.237 dm  
 lc 2712  
 nd 2048 ds84  
 sw 4311.7 dres 1.0  
 fb 2000 homo 30.0  
 ss 16 temp PROCESSING  
 spwr 58 sb  
 pw 7.5 sbs  
 dl 2.000 gfs  
 dtf -33544 swc  
 nt 64  
 ct 32 proc ft  
 gain 60 fnh 2048  
 l1 flags n mzd0 PROCESSING f  
 dp y sb1 0.007  
 hs 2D ACQUISITION nm sb51 not used  
 sw1 3000.0 gfs1 0.005  
 nt1 512 proc1 0.074  
 phase SEQUENCE 1 fn1 2048  
 pwx SEQUENCE 12.0 us  
 pwx1v) 140 th 141788  
 j 140 fp 179.8  
 mbond 1988 0  
 tau GRADIENTS 0.0625 rfp 1688.4  
 g21v11 2000 scf 0  
 g1 150  
 g21v12 0.002000 wczmm 3.26  
 g21v13 -1010 d1 2D DISPLAY  
 g3 0.002000 rfp1 0  
 gstab 0.000050 lpl 6887.0  
 rfp1 8155.0  
 sc2 150  
 wcz2

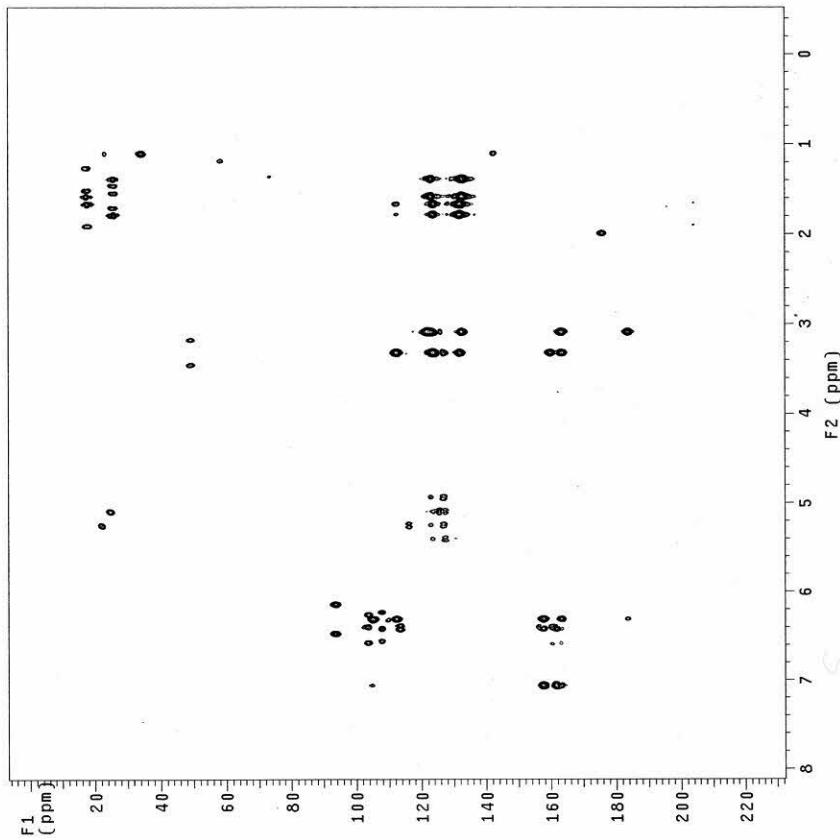
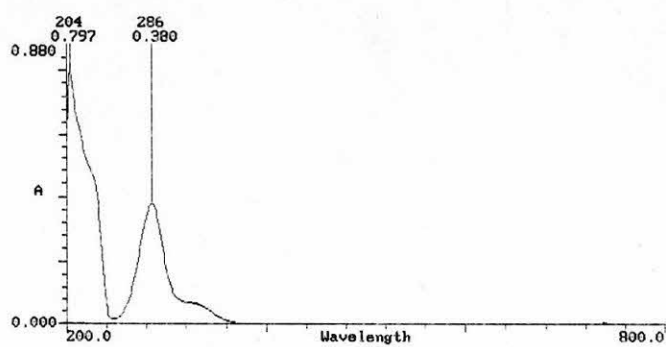


Figure B9 HMBC spectrum of compound M-2 (in methanol -d<sub>4</sub>)

TEST SETUP  
GENESYS 6 v2.001 2M9M008001

Scanning 23:04 4May10  
Test Name D-5  
Measurement Mode Absorbance  
Start Wavelength 200.0nm  
Stop Wavelength 800.0nm  
Sample Positioner Auto 6  
Scan Speed Fast  
Interval 1.0nm  
Cell Correction Off  
ID# (D=OFF) 1  
Auto Print On  
Auto Save Data Off



ID#: 1  
Smoothing [On]  
Wavelength Abs

Wavelength	Abs	
204.0	0.797	Peak
286.0	0.380	Peak

Figure C1 UV-Visible spectrum of M-3 in Methanol

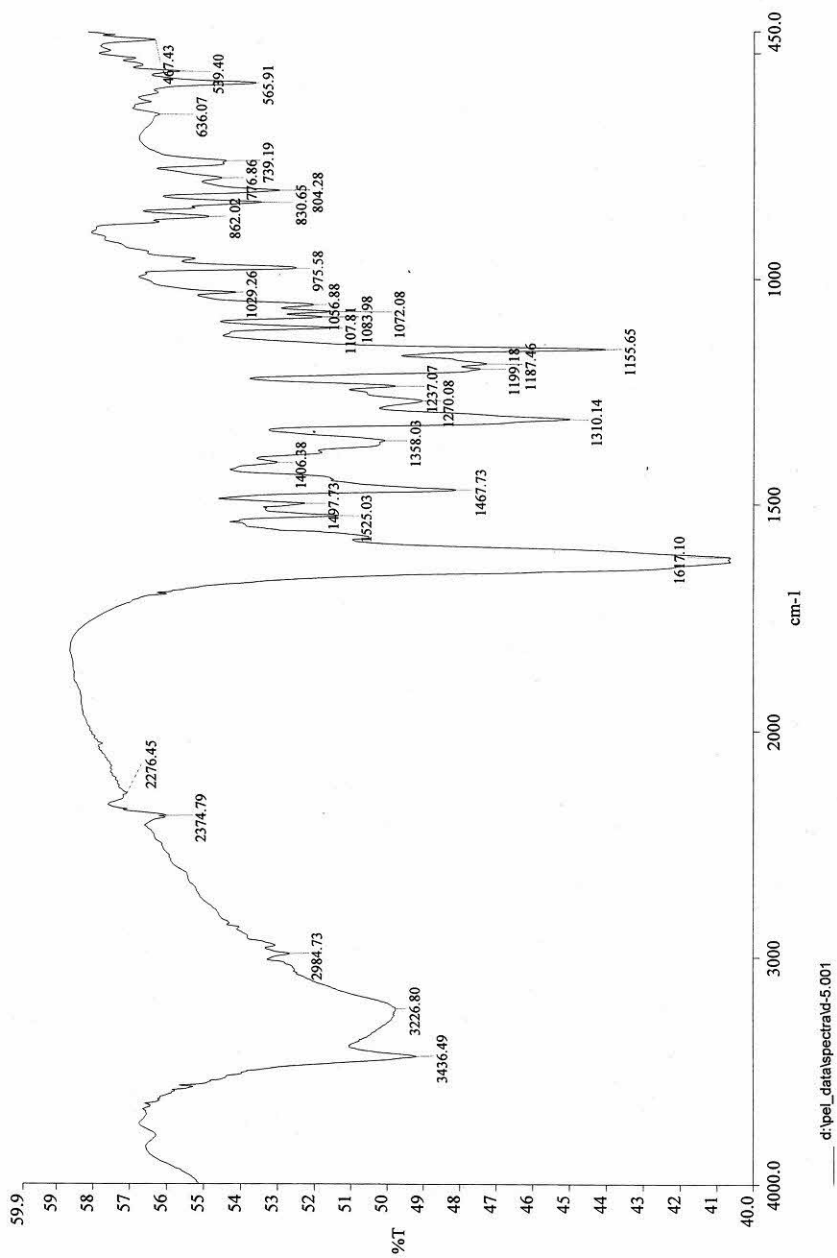
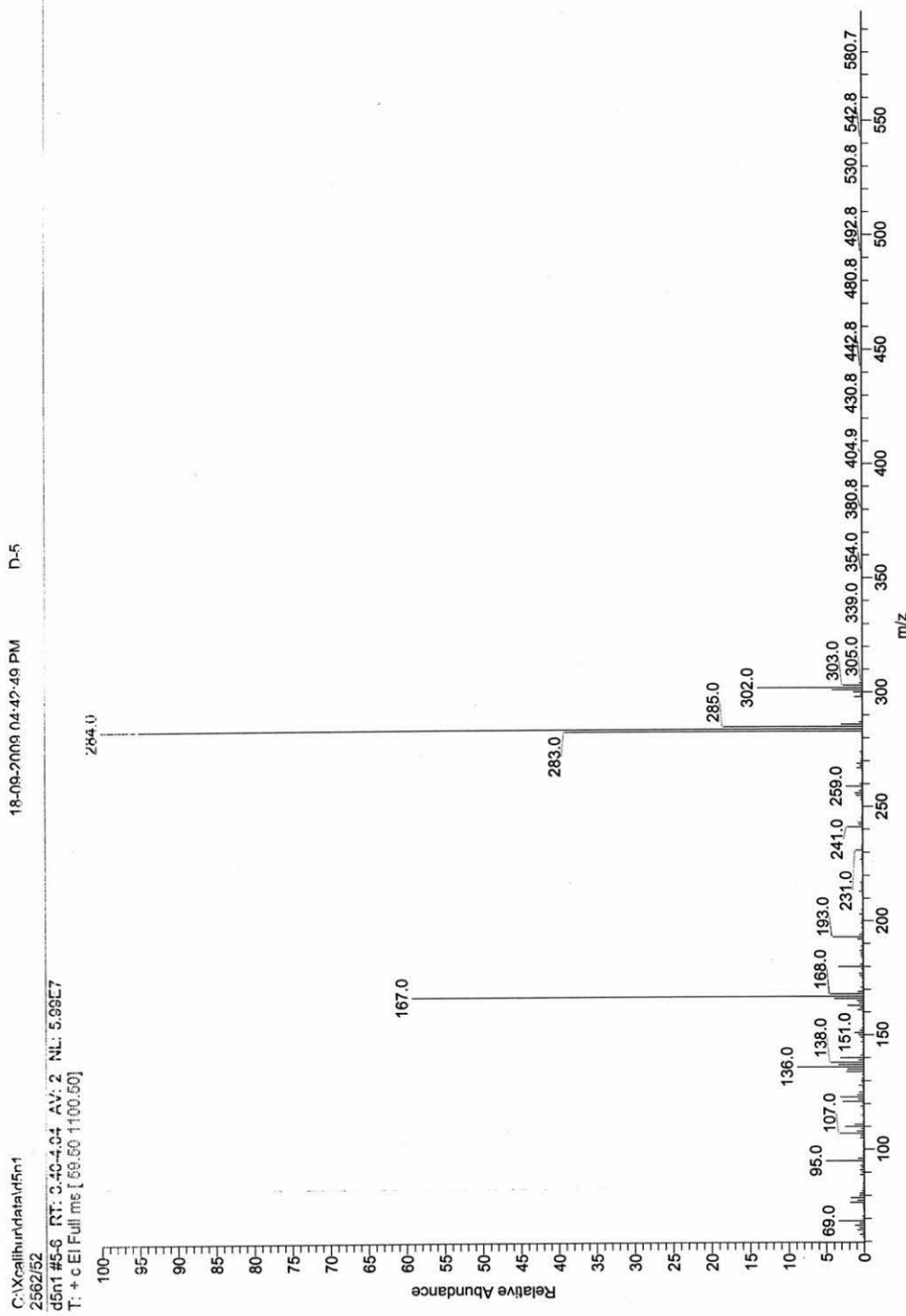


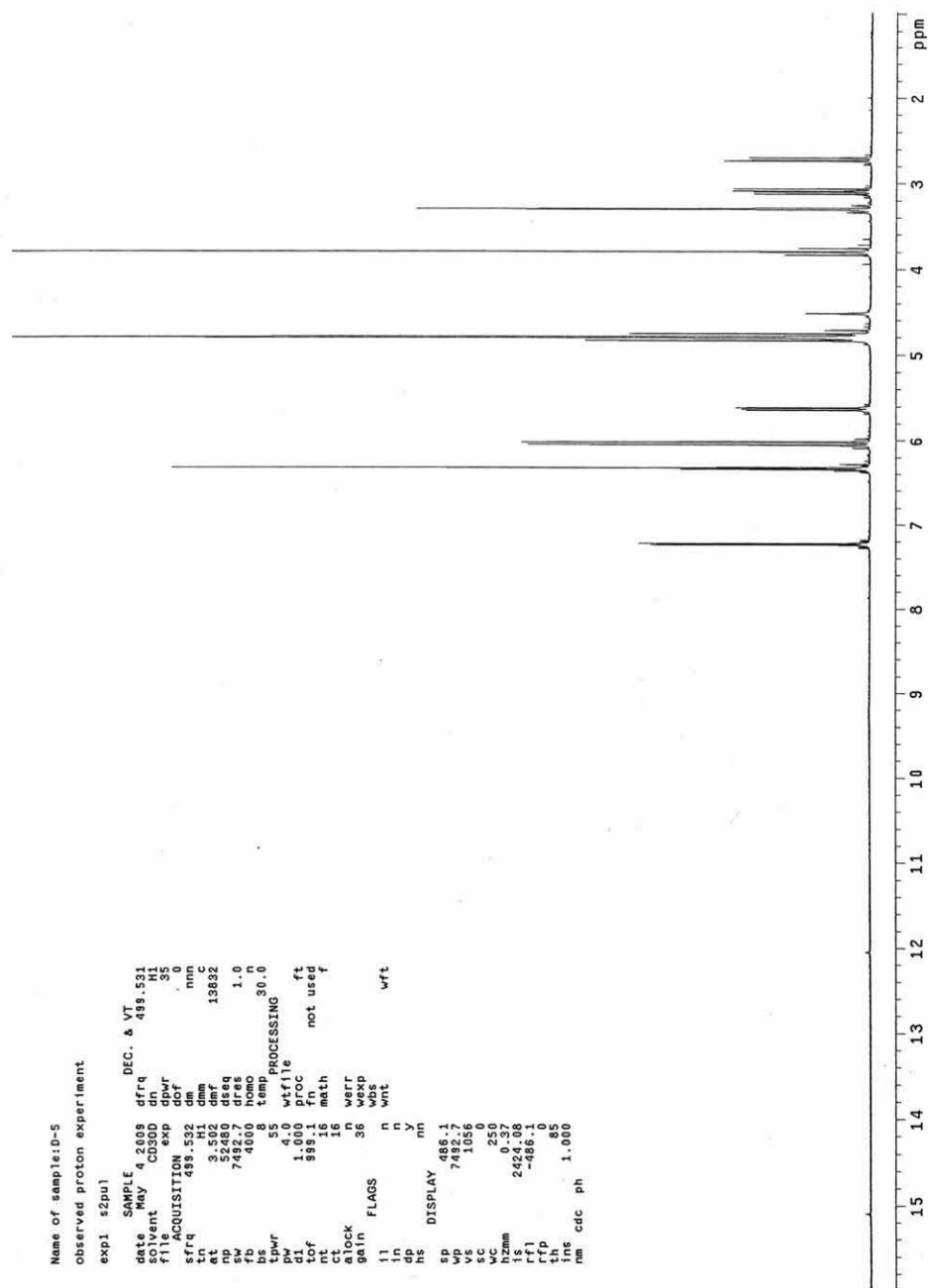
Figure C2 IR spectrum of compound M-3 (KBr disc)

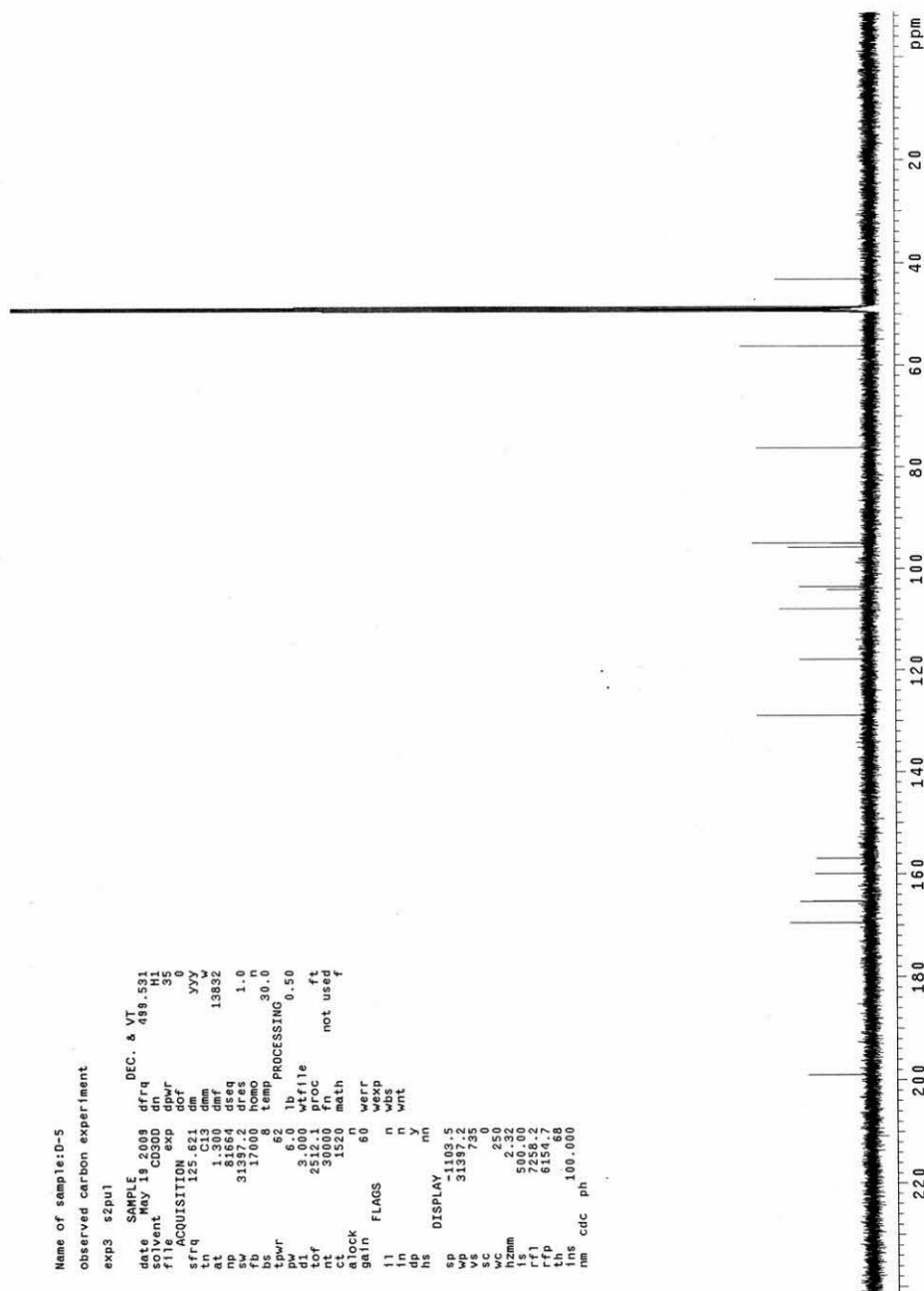


C:\Xcalibur\data\d5m1  
256252  
d5m1 #5-6 RT: 3.40-4.04 AV: 2 NL: 5.98E7  
T: +c EI Full ms [ 59.50 1100.50]

18-09-2009 04:42:49 PM D-5

Figure C3 EI mass spectrum of compound M-3

Figure C4 500 MHz  $^1\text{H}$  spectrum of compound M-3 (in methanol- $d_4$ )



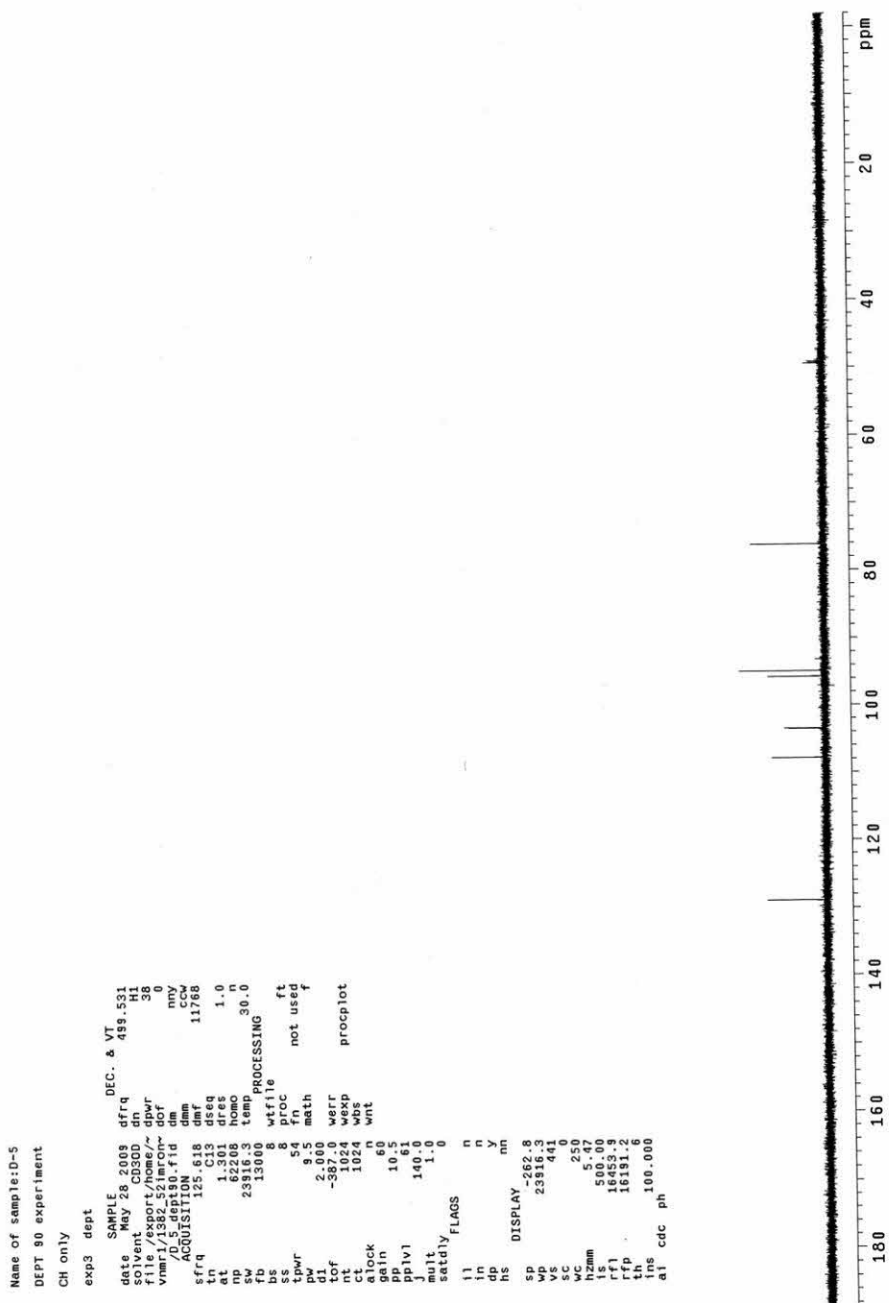
```

Name of sample:D-5
observed carbon experiment
exp3 szpul

SAMPLE
date May 16 2008          DEC. & VI
solvent CD3OD           489.531
file CD3OD              dnf
dpwr 35
ACQUISITION
sfrq 125.021           77y
at 1.500              dm
np 81664              dseq
sw 31337.2            ores
sb 170.0              temp
bs 8                  temp
tpwr 62               30.0
pw 6.0               lb
ds 8.000             wf file
dd 2512              fproc
nt 30000             fnot used
ct 1520              math
alock n
gain 60              warr
flags n wbs
il n
in n
dp n
ns y
DISPLAY
sp -1103.5
wp 3139.752
sc 750
wc 250
hzmm 2.32
f 7258.2
rfl 6154.7
th
ins cdc
nm 100.000
ph
    
```

Figure C5 125 MHz <sup>13</sup>C spectrum of compound M-3 (in methanol-d<sub>4</sub>)





```

Name of sample: D-5
DEPT 90 experiment
CH only
exp3 dept
SAMPLE DEC. & VT
date May 28 2009 dfrq 459.931
file ent/export/home/~ dprw
vnmr1/1382.52/Incon~ ddf 38
/B_5_dept90_fid dm nny
ACQUISITION dmr 11768
sfreq 105.618 dseq
in C13 dres 1.0
at 1.301 dres
np 62209 homo
23209 temp
fb 13000 PROCESSING
bs 8 wtf file
ss 8 proc ft
tpwr 5 math not used
g1 2.060 math
tof -387.0 werr
nt 102 wexp
atlock 104 wnt
gain 60 n
pp1v1 10.5
pp1v2 140.0
mult 1.0
satdly/flags
l1 n
in n
dp y
hs DISPLAY nm
SP -262.8
wp 23916.3
vs 471
vc 250
hzzmm 5.47
ls 1500.00
rfp 16191.2
th 100.000
ins
at cdc ph
    
```

Figure C6.1 DEPT 90 spectrum of compound M-3 (in methanol-*d*<sub>4</sub>)

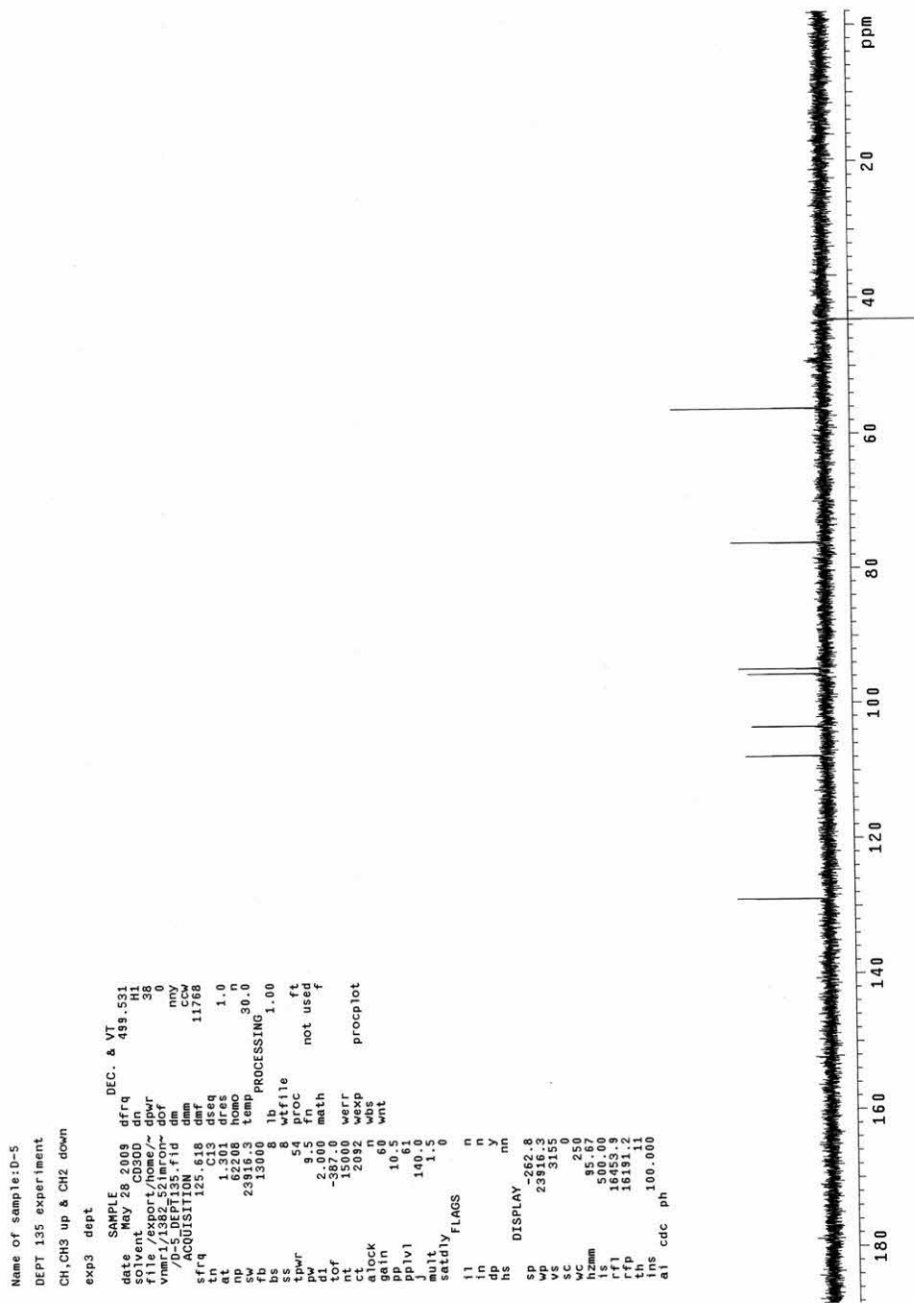


Figure C6.2 DEPT 135 spectrum of compound M-3 (in methanol-*d*<sub>4</sub>)

```

Name of sample:D-5
gcossy experiment
exp3 gcossy
SAMPLE DEC. & VT
date May 29 2009 dn H1
solvent MeSO-d6 dof
nu1 1382.52 hmc
nu2 1382.52 hmc
nu3 1382.52 hmc
nmr1 /D-5_gcossy.fid dmf 35
ACQUISITION dpr 30.0
$frq 499.530 homo
at 0.215 temp
np 2048 lp 17.1
sw 4766.7 rp -0.2
bb 3000 p1 0
ss 2 p2 0
PROCESSING
tpwr 55 sb 0.051
pw 7.6 sbs 0.001
dl 1.000 fnc 2048
tof -833.3 math f
nt 4 werr
clock n wexp
gain 36 was
FLG1 n wntD PROCESSING wft
fl n lbiD 0.32
dp y procl
hs y procl
SWH ACQUISITION mm fml DISPLAY 2048
nu1 4766.7 sp -401.8
nu2 4765.12 wp 4268.2
nu3 1897
d2 0 vs
GRADIENTS SC 0
g2lv11 600
g1lv11 0.001900 hzmm 15.0
grlsc 10.44
grlsc 0.000100 ls 2424.08
gstab 2468.6
qiv 1.0 rfl 1646.2
spt 20 DISPLAY 1.0 ttp 1646.2
wp1 -174.0 ins
sc2 3856.0 at cdc ev 1.000
rf1 150
rf11 3875.6
rfp1 3158.5
    
```

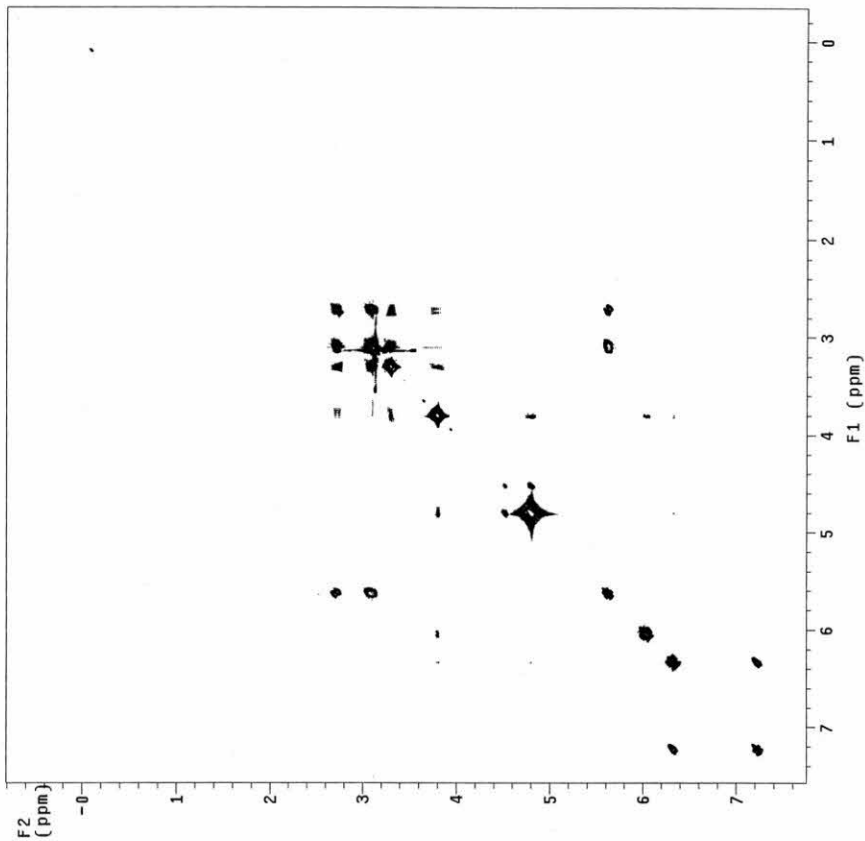


Figure C7 COSY spectrum of compound M-3 (in methanol-*d*<sub>4</sub>)

```

Name of sample:D-5
ghMQC experiment
exp3 ghmqc_da
SAMPLE DEC. & VT
date May 29 2009 dfrq 125.620
solvent CD3OD dn C13
file_/export/home/~ dpwr 211250
vimr/~/D-S-ghmqc-fid of
/~/D-S-ghmqc-fid nm
ACQUISITION C9
sfrq 499.530 dmf 27173
tn 0.211 dseq 94791
np 2048 homo 1
sw 4767.0 temp 30.0
fb 3000 sb PROCESSING 058
ss 16 sbs -0.054
tpwf 58 gf 0.020
pw 7.5 gfs 0.026
tof -833.0 proc 0.064
nt 64 fn 2048
ct 64 meth f
gain FLAG 60 sb1 0.007
1) n sb1 not used
dp y gf1 0.005
hs 2D ACQUISITION nm gfs1 0.001
sw1 30007.0 fn1 DISPLAY 2048
n1 512
phase 1 vs 366517
pwx sequence 12.0 tn 0
pwxlv1 160 lp 179.8
j 140 rfl 4425.3
mbond n rfp 3606.0
temp GRADIENTS wc 150
g2lv11 2000 hzmm 31.78
g1 0.002000 at 2D DISPLAY 0
g2v12 0.002000 rp1 0
g2lv13 -1010 lp1 17019.1
g3 0.002000 rfl1 16132.0
g3tab 0.000050 sc2 150
wc2 150

```

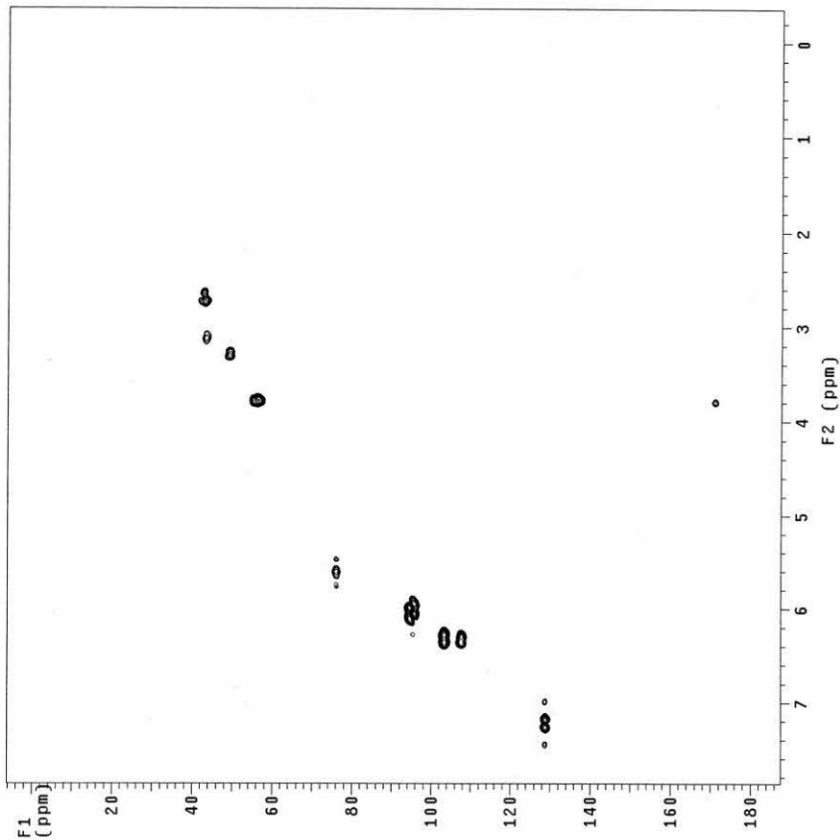


Figure C8 HMQC spectrum of compound M-3 (in methanol-*d*<sub>4</sub>)

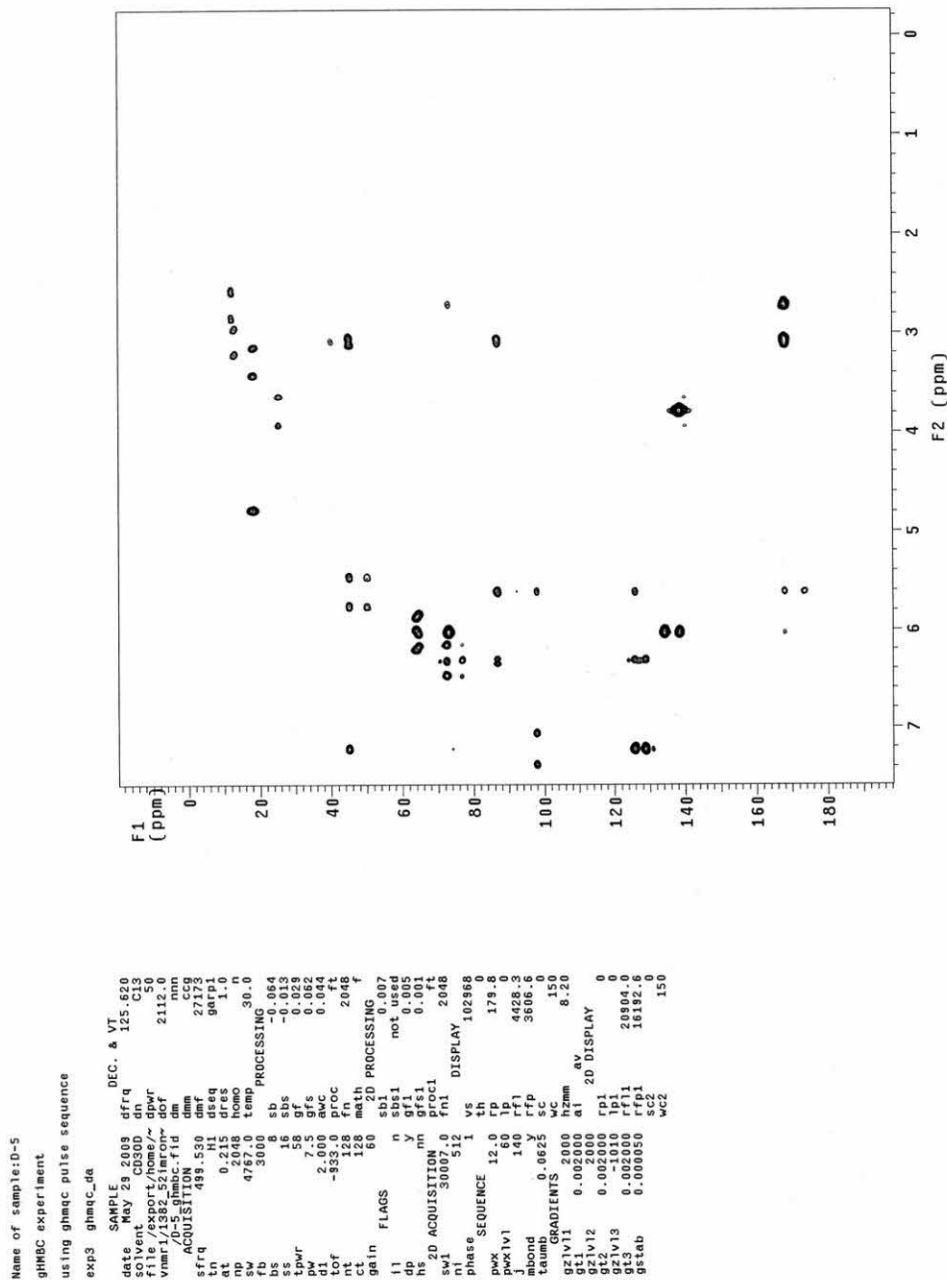
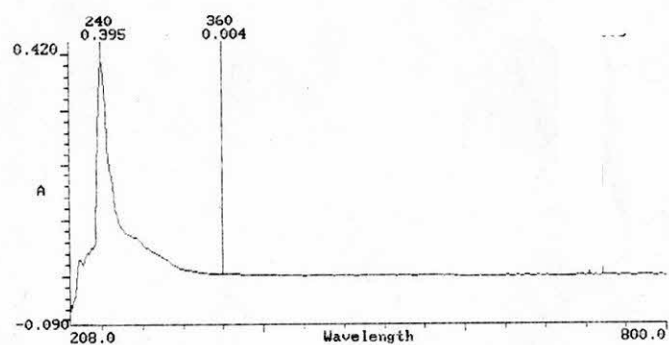


Figure C9 HMBBC spectrum of compound MI-3 (in methanol-*d*<sub>4</sub>)

TEST SETUP  
GENESYS 6 v2.001 2M9M008001

Scanning 21:54 8Apr10  
Test Name CH-1  
Measurement Mode Absorbance  
Start Wavelength 200.0nm  
Stop Wavelength 800.0nm  
Sample Positioner Auto 6  
Scan Speed Fast  
Interval 2.0nm  
Cell Correction Off  
ID# (0=OFF) 1  
Auto Print On  
Auto Save Data Off



ID#: 1  
Smoothing [On]  
Wavelength Abs  
-----  
240.0 0.395 Peak  
360.0 0.004 Peak

Figure D1 UV-Visible spectrum of H-1 in Chloroform

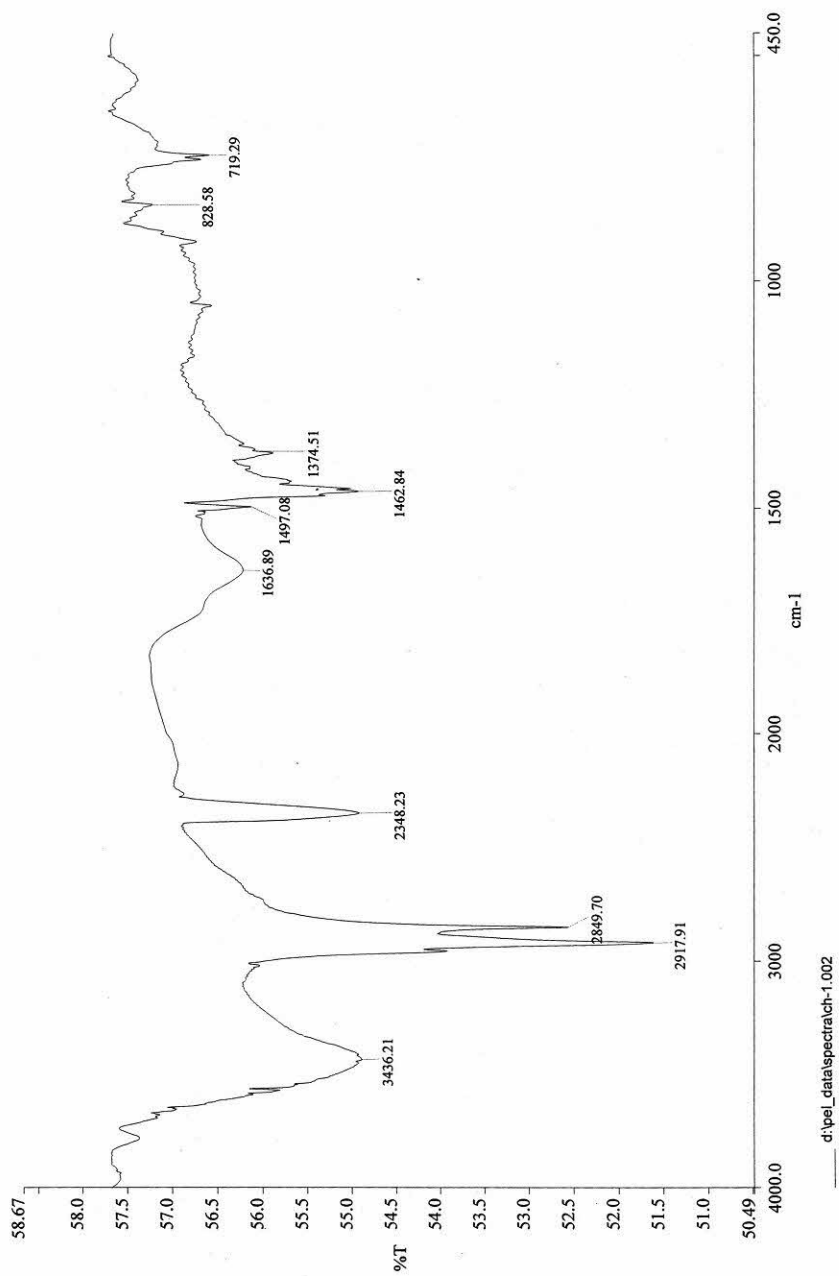


Figure D2 IR spectrum of compound H-1 (KBr disc)

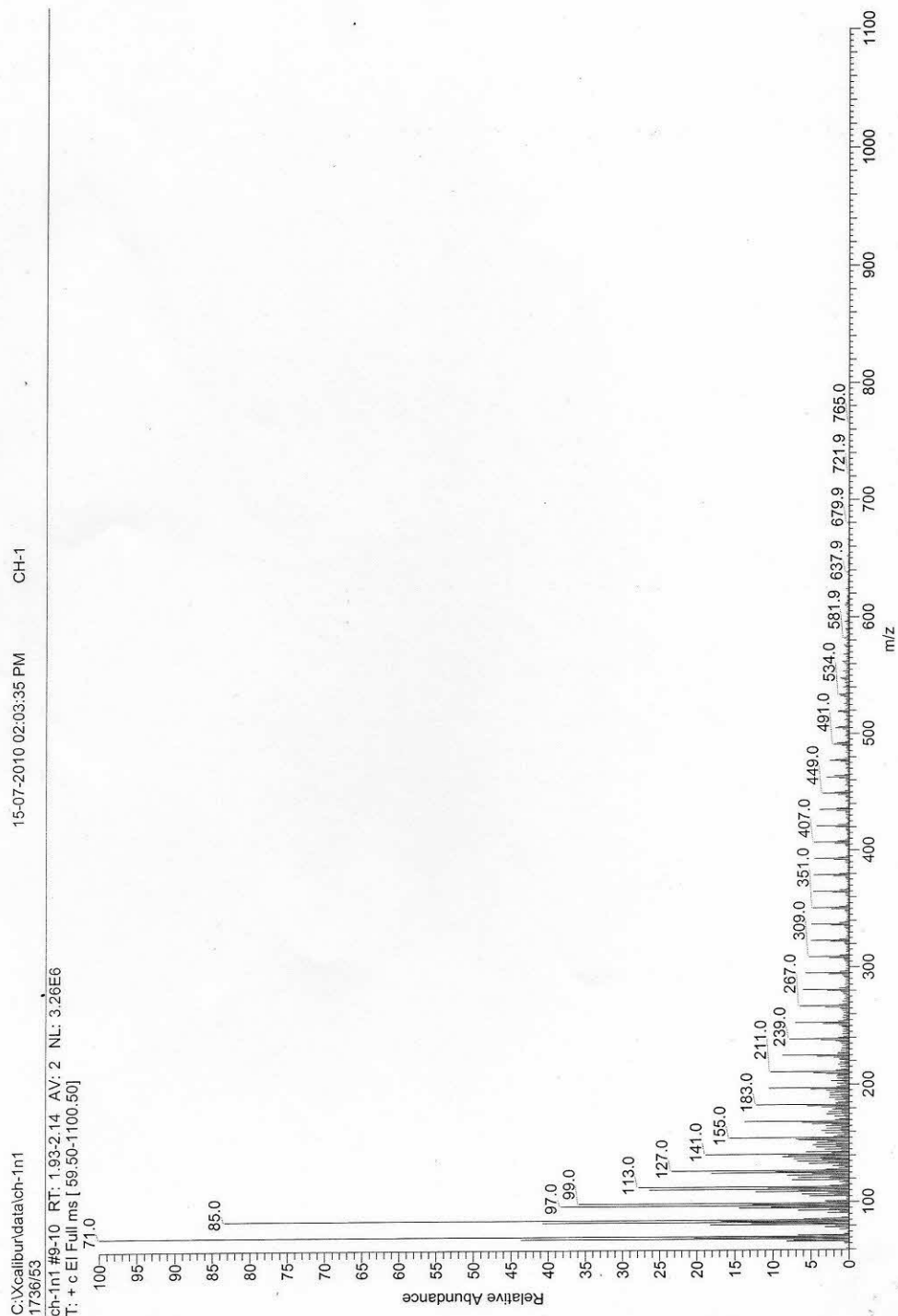


Figure D3 EI mass spectrum of compound H-1



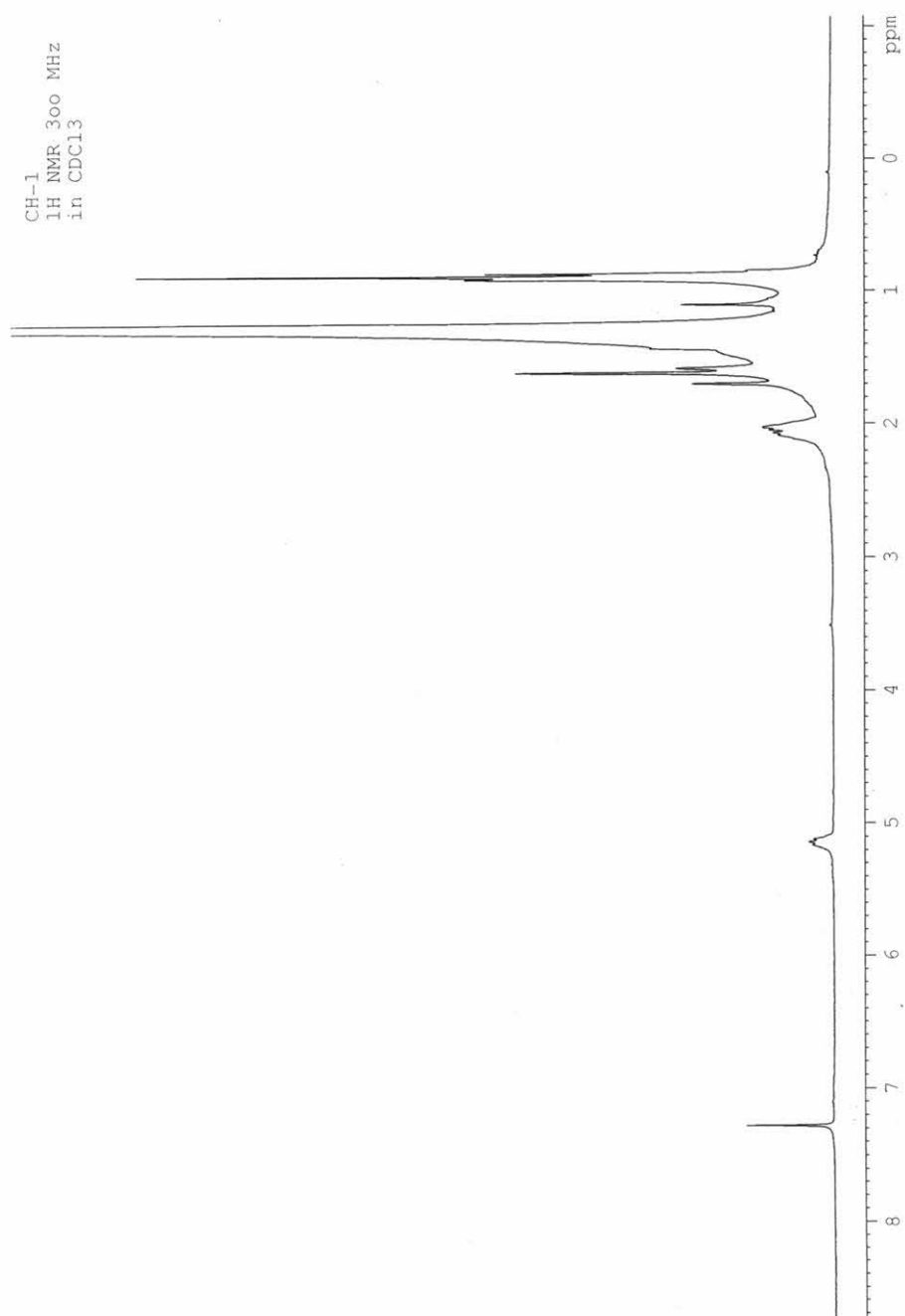
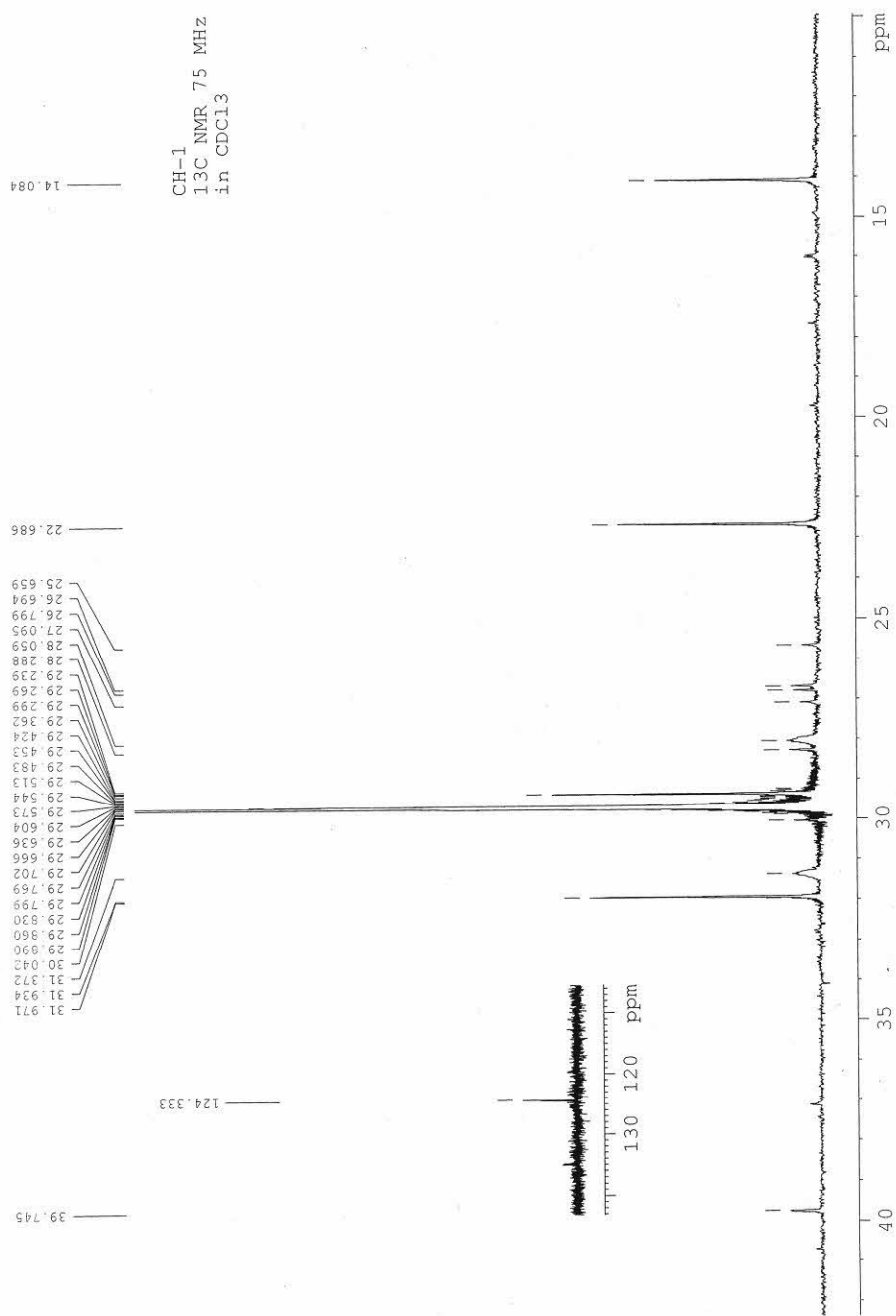
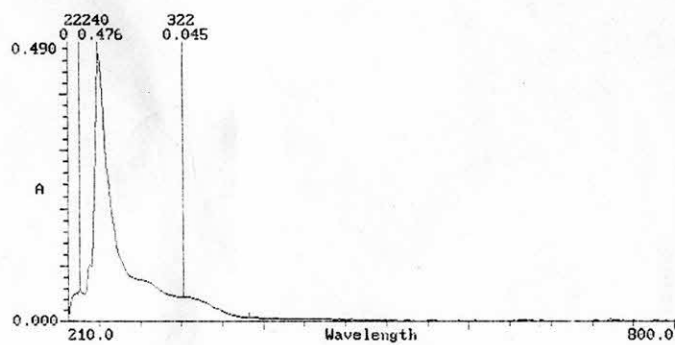


Figure D4 300 MHz <sup>1</sup>H spectrum of compound H-1 (in chloroform -d)

Figure D5 75 MHz  $^{13}\text{C}$  spectrum of compound H-1 (in chloroform  $-d$ )

TEST SETUP  
GENESYS 6 v2.001 2M9M008001

Scanning 21:40 8Apr10  
Test Name CH-2  
Measurement Mode Absorbance  
Start Wavelength 210.0nm  
Stop Wavelength 800.0nm  
Sample Positioner Auto 6  
Scan Speed Fast  
Interval 2.0nm  
Cell Correction Off  
ID# (0=OFF) 1  
Auto Print On  
Auto Save Data Off



ID#: 1  
Smoothing [On]

Wavelength	Abs	
240.0	0.476	Peak
222.0	0.057	Peak
322.0	0.045	Peak

Figure E1 UV-Visible spectrum of H-2 in Chloroform

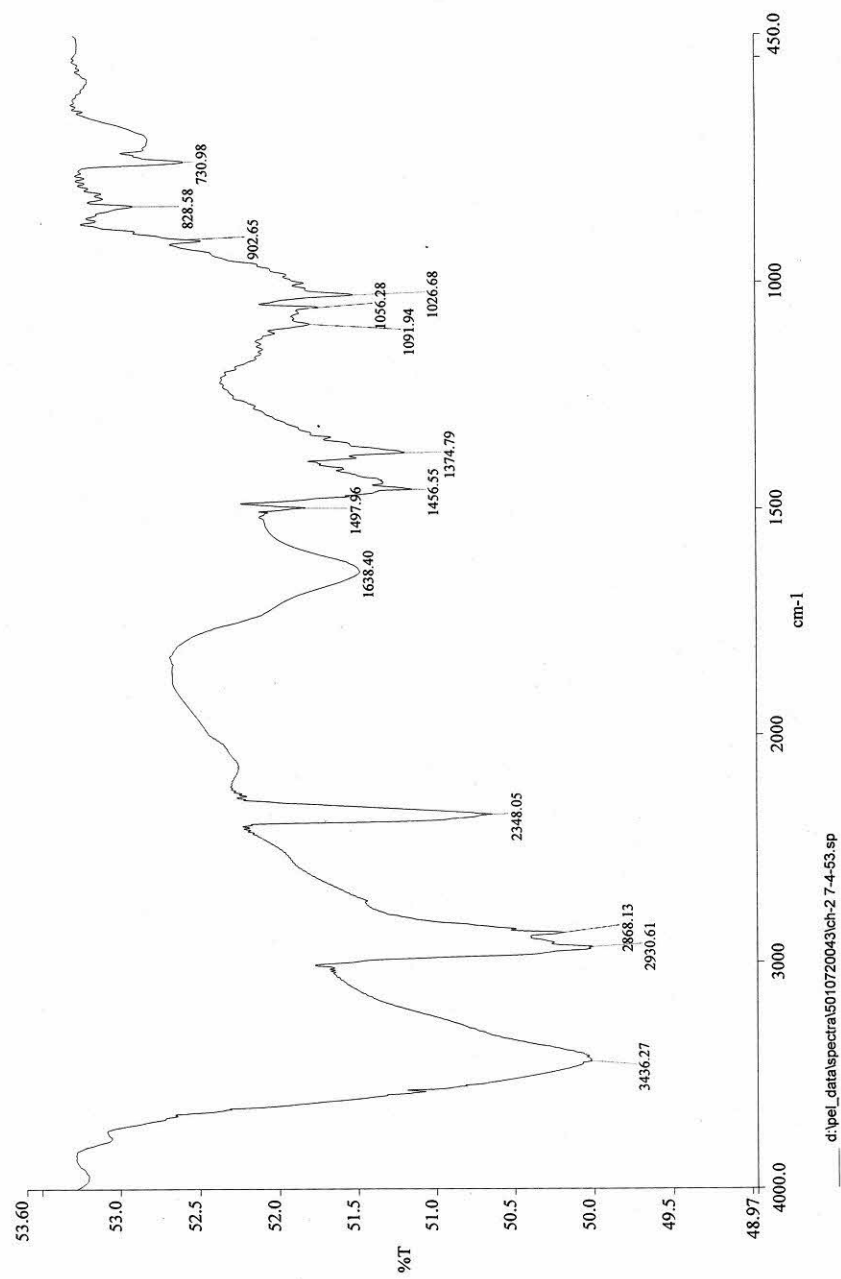


Figure E2 IR spectrum of compound H-2 (KBr disc)

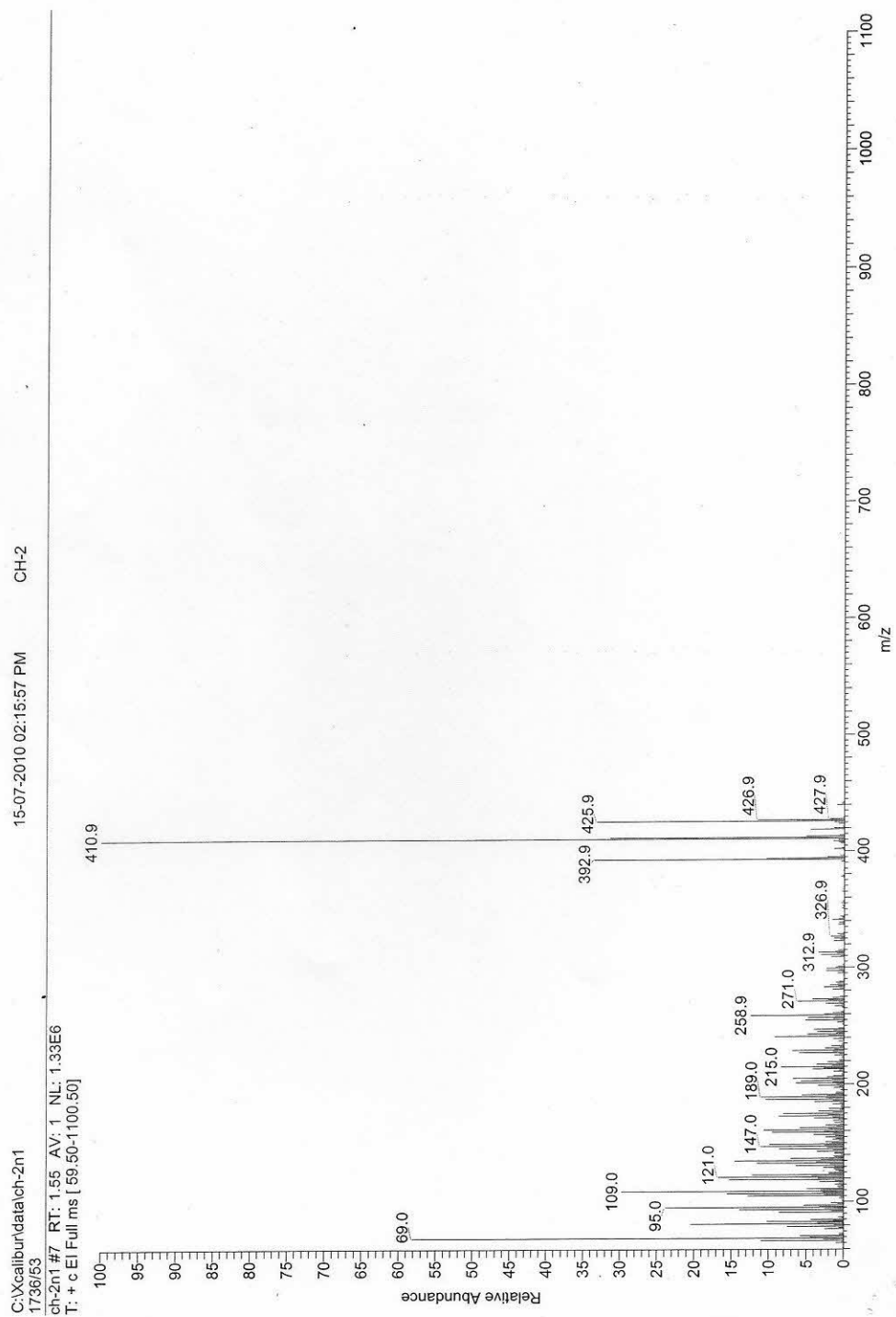


Figure E3 EI mass spectrum of compound H-2

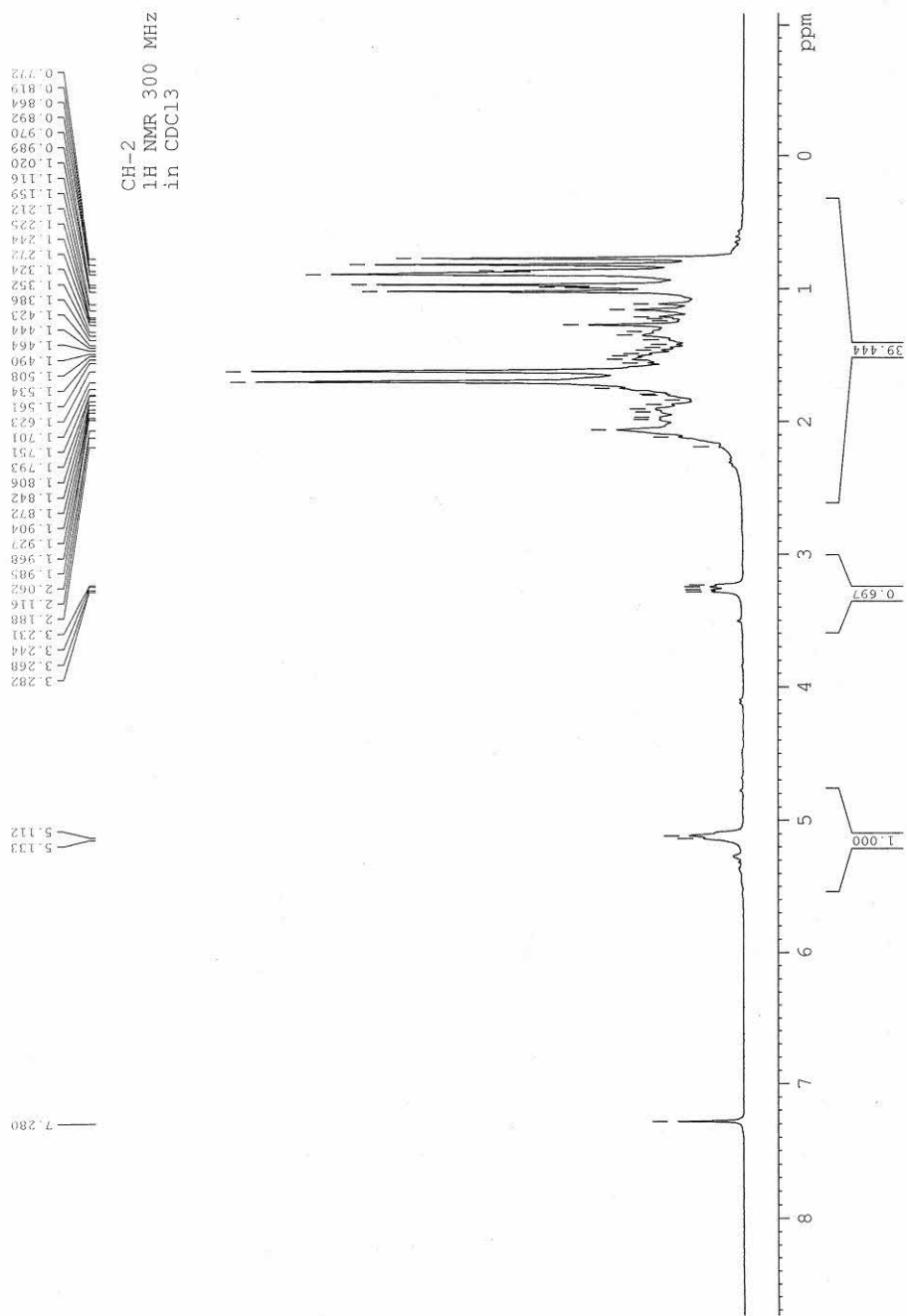
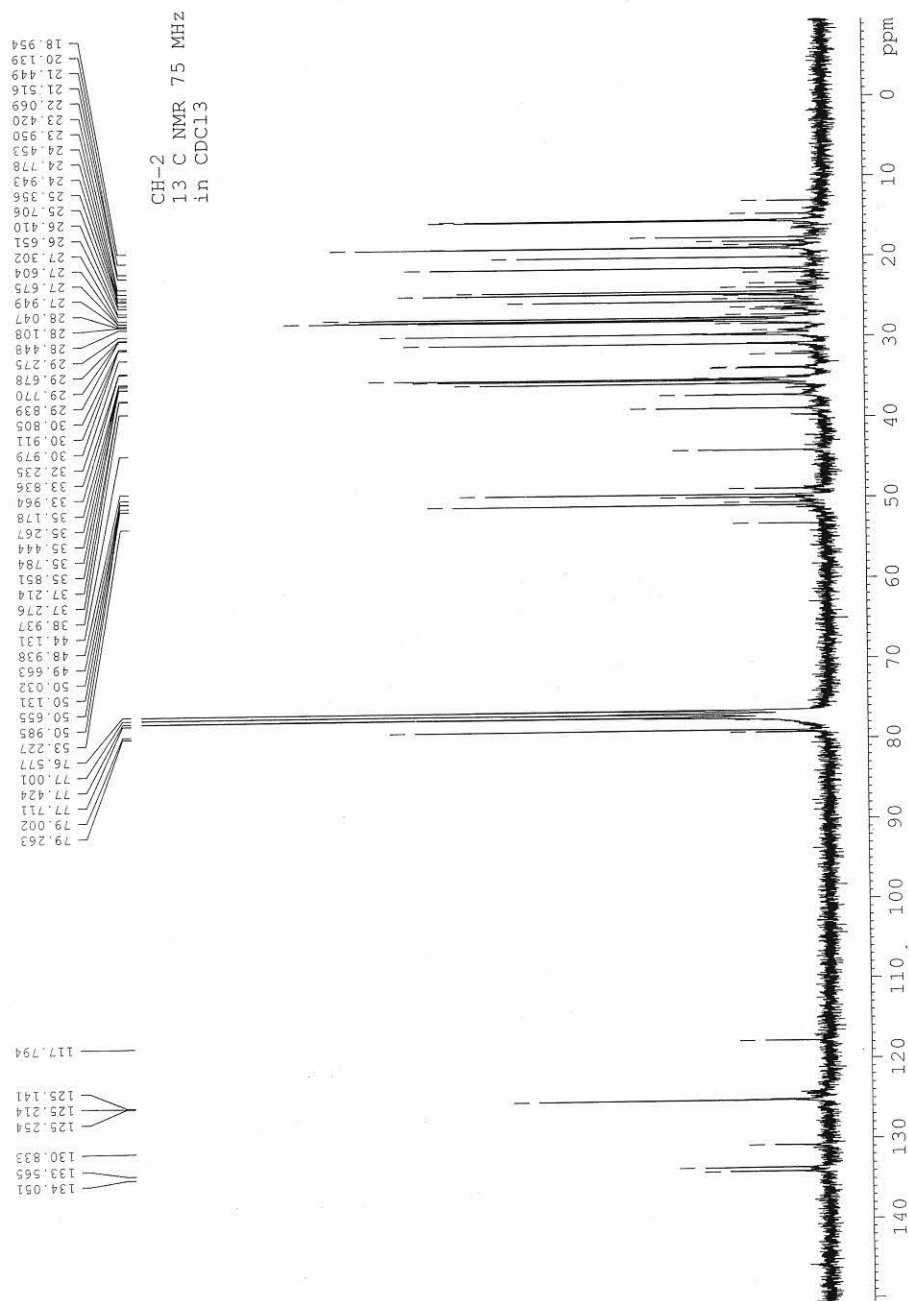
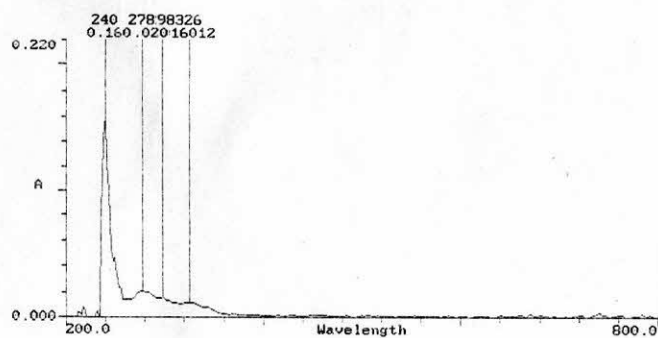


Figure E4 300 MHz <sup>1</sup>H spectrum of compound H-2 (in chloroform -d)

Figure E5 75 MHz  $^{13}\text{C}$  spectrum of compound H-2 (in chloroform -d) (continued)

TEST SETUP  
GENESYS 6 v2.001 2M9M008001

Scanning 21:12 8Apr10  
Test Name CH-5  
Measurement Mode Absorbance  
Start Wavelength 200.0nm  
Stop Wavelength 800.0nm  
Sample Positioner Auto 6  
Scan Speed Fast  
Interval 2.0nm  
Cell Correction Off  
ID# (0=OFF) 1  
Auto Print On  
Auto Save Data Off



ID#: 1  
Smoothing [On]  
Wavelength Abs

Wavelength	Abs	Peak
240.0	0.160	Peak
278.0	0.020	Peak
298.0	0.016	Peak
326.0	0.012	Peak

Figure F1 UV-Visible spectrum of H-3 in Chloroform



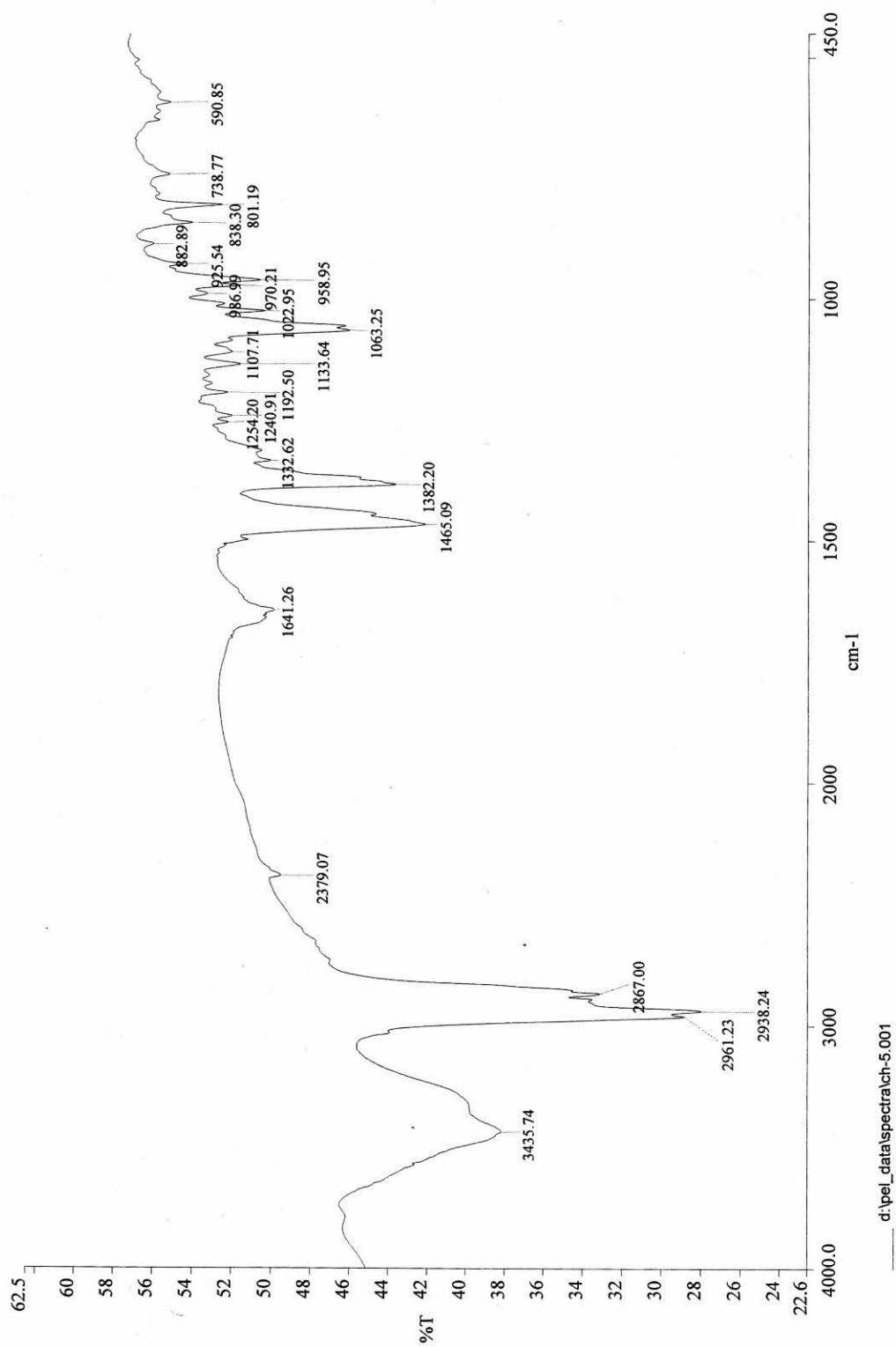


Figure F2 IR spectrum of compound H-3 (KBr disc)

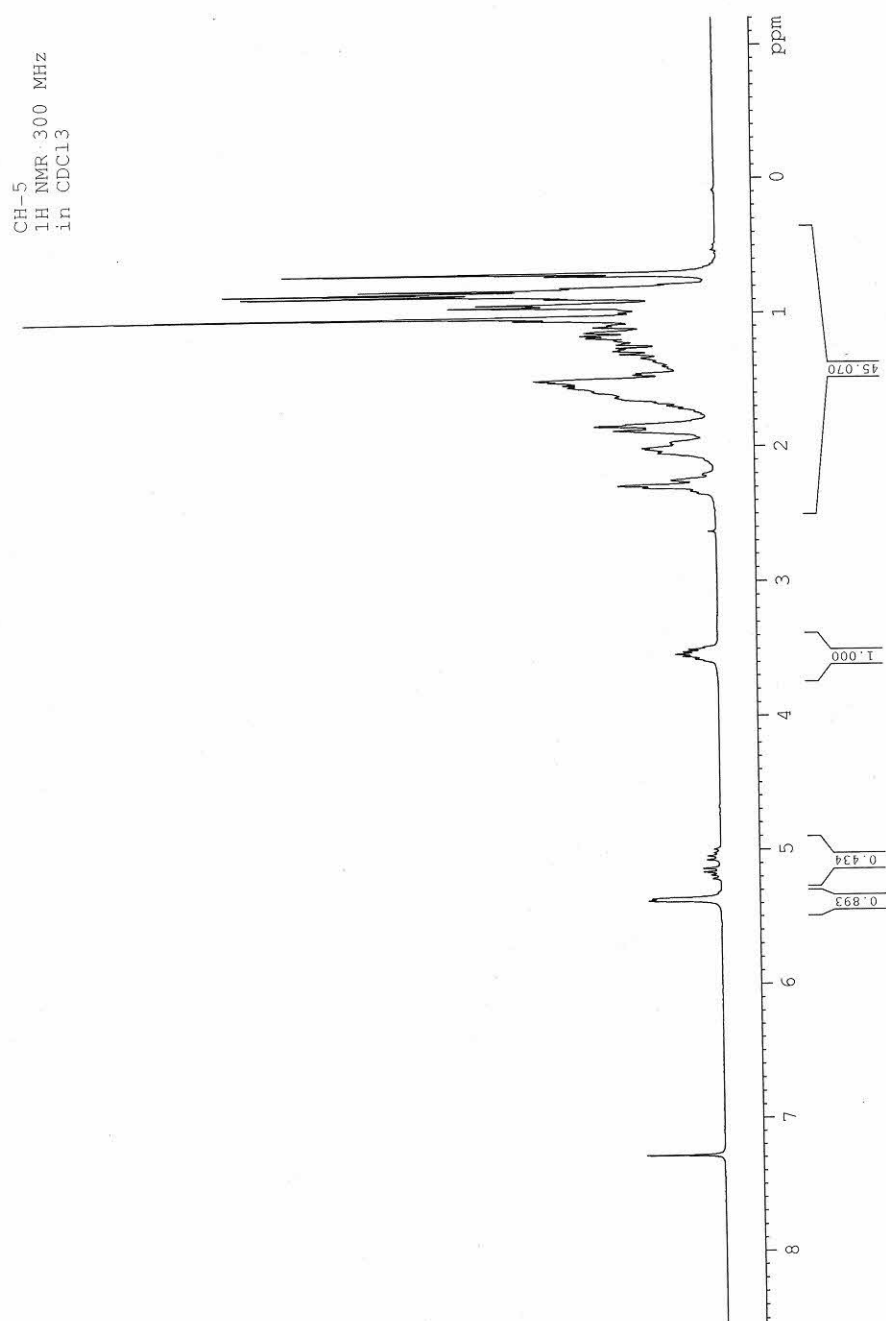
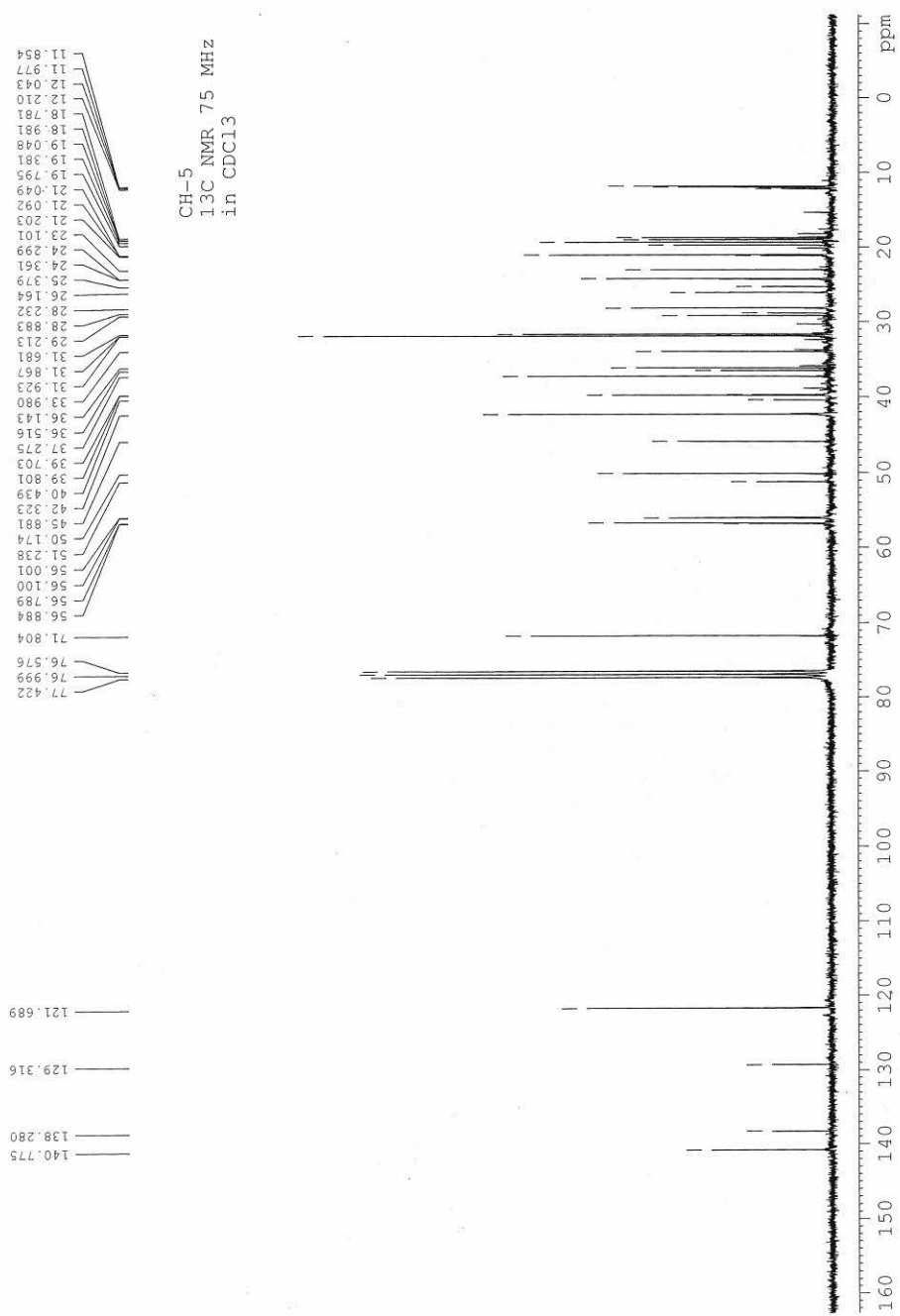


Figure F3 300 MHz <sup>1</sup>H spectrum of compound H-3 (in chloroform -d)

Figure F4 75 MHz  $^{13}\text{C}$  spectrum of compound H-3 (in chloroform -d)

## VITAE

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**Student** 5010720043

### Education Attainment

Degree	Name of Institution	Year of Graduation
Bachelor of Science (Chemistry)	Yala Rajabhat University	2006

### List of Publication and Proceedings

Meechai, I., Puripattavong, J. and Dej-adisai, S. 2010. Anti-tyrosinase and anti-microbial activities from the root of *Artocarpus integer*. Thailand Research Symposium 2010 Proceedings, 26-31 August, 2010 at Bangkok Convention Centre, Central World, Bangkok, Thailand (poster presentation).

Meechai, I., Puripattavong, J. and Dej-adisai, S. 2010. Screening of anti-tyrosinase activity from Thai medicinal plants. NRCT-JSPS Core University Program on Natural Medicine in Pharmaceutical Sciences *The 9<sup>th</sup> Joint Seminar* Natural Medicine Research for the Next Decade: New Challenges and Future Collaboration Proceedings, 8-9 December, 2010. Faculty of Pharmaceutical Sciences, Chulalongkorn University, Bangkok, Thailand (poster presentation).