

Chemical Constituents from *Rhodomyrtus tomentosa* (Aiton) Hassk. and Antibacterial Activity

Asadhawut Hiranrat

A Thesis Submitted in Partial Fulfillment of the Requirements for the Degree of Doctor of Philosophy in Organic Chemistry Prince of Songkla University

2010

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Thesis Title	Chemical Constituents from Rhodomyrtus tomentosa (Aiton) Hassk.		
	and Antibacterial Activity		
Author	Mr. Asadhawut Hiranrat		
Major Program	Organic Chemistry		

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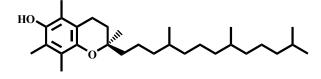
The Graduate School, Prince of Songkla University, has approved this thesis as a partial fulfillment of the requirements for the Doctor of Philosophy degree in Organic Chemistry.

(Prof. Dr. Amornrat Phongdara) Dean of Graduate School ชื่อวิทยานิพนธ์องค์ประกอบทางเกมีจากโทะและฤทธิ์ด้านแบคทีเรียผู้เขียนนายอัยฎาวุธ หิรัญรัตน์สาขาวิชาเกมีอินทรีย์ปีการศึกษา2553

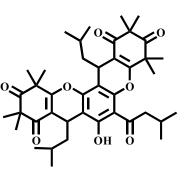
### บทคัดย่อ

โทะ (Rhodomyrtus tomentosa) เป็นพืชในวงศ์ชมพู่ (Myrtaceae) ส่วนสกัดหยาบไคคลอโร มีเทนและอะซิโตนจากใบโทะมีฤทธิ์ในการยับยั้งการเจริญของแบคทีเรีย Staphylococcus aureus ATCC 25923 และ methicillin-resistant S. aureus (MRSA) NRPC R01 ด้วยค่าความเข้มข้นต่ำสด (MIC) 31.25 และ 62.5 μg/mL ตามลำดับ การศึกษาองก์ประกอบทางเคมีของใบ ต้นและผลโทะเพื่อ หาสารที่แสดงฤทธิ์ในการยับยั้งการเจริญของเชื้อแบคทีเรีย แยกสารองค์ประกอบได้ 41 สาร เป็น สารกลุ่มเอซิลฟลอโรกลูซินอล (acylphloroglucinols) จำนวน 11 สาร คือ rhodomyrtosone A (ART2), rhodomyrtosone H (ART3), rhodomyrtosone C (ART4), rhodomyrtone (ART6), endoperoxide G3 (ART8), rhodomyrtosone B (ART9), rhodomyrtosone D (ART11), rhodomyrtosone E (ART18), rhodomyrtosone G (ART19), rhodomyrtosone F (ART20) 1162 rhodomyrtosone I (ART38) สารกลุ่มฟลาโวนอยด์ (flavonoids) จำนวน 4 สาร คือ combretol (ART7), 3,4',5',7-tetra-O-methylmyricetin (ART13), 3,3',5',7-tetra-O-methylmyricetin (ART16) และ 3,3',4',5'-tetra-O-methylmyricetin (ART17) สารกลุ่มเทอพีนอยด์ (terpenoids) จำนวน 7 สาร คือ (6R,7E,9S)-9-hydroxy-4,7-megastigmadien-3-one (ART12), loliolide (ART14), 3\beta-O-Ecoumaroylmaslinic acid (ART22), 3*β*-O-Z-coumaroylmaslinic acid (ART23), 3*β*-O-E-coumaroyl oleanolic acid (ART32), arjunolic acid (ART34) และ oleanolic acid (ART39) สารกลุ่ม สเตียรอยค์ (steroids) จำนวน 3 สาร คือ  $\beta$ -sitosterol (ART5),  $\beta$ -sitosterol glucopyranoside (ART24) และ stigmast-4-en-3-one (ART37) อนุพันธ์ของกรดแอลลาจิก (ellagic acid derivatives) จำนวน 3 สาร คือ 3,3',4-tri-O-methylellagic acid (ART10), 4-O-[B-D-glucopyranosyltetraacetate]-3,3',4'-tri-O-methylellagic acid (ART28) และ 3-O-methylellagic acid 4-O- $\alpha$ rhamnopyranoside (ART41) อนุพันธ์ของกรดฟลาเวลลางิก (flavellagic acid derivatives) งำนวน 2 สาร ก็อ 3',4'-dioxymethylene-3,4-di-O-methylflavellagic acid (ART25) และ 3,4,3',4'-tetra-Omethylflavellagic acid (ART31) สารกลุ่มอนุพันธ์เบนซีน (benzenoids) งำนวน 6 สาร ก็อ  $\alpha$ -tocopherol (ART1), trans-triacontyl-4-hydroxy-3-methoxycinnamate (AR21), trans-triacontyl-4-hydroxycinnamate (ART30), 4-hydroxy-3-methoxybenzoic acid (ART35), gallic acid (ART36) และ methyl gallate (ART40) อนุพันธ์ลิกแนน (lignans) งำนวน 1 สาร ก็อ 9,9'-O-diferuloyl-(-)-secoisolariciresinol (ART33) อนุพันธ์ของบิวไทโรแลคโทน (butyrolactone derivatives) งำนวน 1 สาร ก็อ (3a*S*\*,6a*R*\*)-3a-(hydroxymethyl)-2,2-dimethyldihydro-furo[3,4-*d*] [1,3]dioxol-4(3a*H*)-one (ART15) และน้ำตาล งำนวน 3 สาร ก็อ  $\beta$ -D-glucopyranoside pentaacetate (ART26),  $\alpha$ -D-glucopyranoside penta-acetate (ART27) และ sucrose octa-acetate (ART29) สาร 11 สาร ก็อ ART2, ART3, ART4, ART9, ART11, ART15, ART18, ART19, ART20, ART31 และ ART38 เป็นสารที่ยังไม่มีรายงานการวิจัย โครงสร้างของสารประกอบเหล่านี้ วิเคราะห์ด้วยข้อมูลทางสเปกโทรสโกปี UV IR NMR และ MS นอกจากนี้ยังได้เปรียบเทียบข้อมูล ทางสเปกโทรสโกปีกับสารที่มีรายงานการวิจัยแล้ว

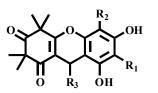
การทดสอบฤทธิ์ของสารกลุ่มเอซิลฟลอโรกลูซินอล (ART2, ART4, ART6, ART9, ART11, ART18, ART19 และ ART20) ในการด้านเชื้อแบคทีเรีย *S. aureus*, MRSA และ *Streptococcus pyogenes* DMST 101 พบว่า rhodomyrtone (ART6) และ rhodomyrtosone B (ART9) สามารถยับยั้งการเจริญของเชื้อทั้งสามสายพันธุ์ ด้วยค่าความเข้มข้นต่ำสุด (MIC) 0.39, 6.25  $\mu$ g/mL 0.39, 12.5  $\mu$ g/mL และ 0.39, 3.125  $\mu$ g/mL ตามลำดับ นอกจากนี้ยังพบว่า rhodomyrtosone G (ART19) ยับยั้งการเจริญของ *S. aureus* และ MRSA ด้วยค่าความเข้มข้นต่ำสุด (MIC) 1.56  $\mu$ g/mL และ rhodomyrtosone D (ART11) ยับยั้งการเจริญของ *S. pyogenes* ด้วยค่า ความเข้มข้นต่ำสุด (MIC) 12.5  $\mu$ g/mL



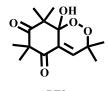
ART1



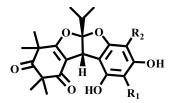
ART4



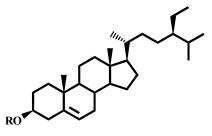
**ART6** :  $R_1$  = isovaleryl,  $R_2$  = H,  $R_3$  = isobutyl **ART9** :  $R_1$  = H,  $R_2$  = isovaleryl,  $R_3$  = isobutyl **ART19** :  $R_1$  = 2-methylbutyryl,  $R_2$  = H,  $R_3$  = isobutyl **ART38** :  $R_1$  = isovaleryl,  $R_2$  = H,  $R_3$  = phenyl



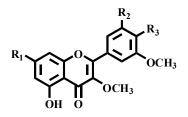
ART8



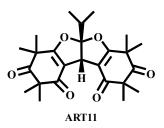
**ART2** :  $R_1 = H, R_2 = i$  sovaleryl **ART3** :  $R_1 = i$  sovaleryl,  $R_2 = H$ 

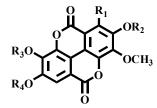


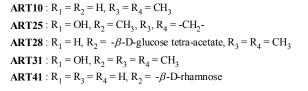
**ART5** : R = H**ART24** :  $R = -\beta$ -D-glucose

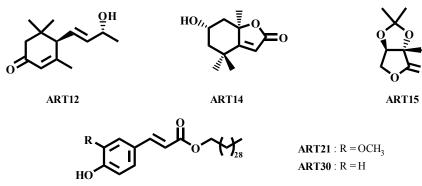


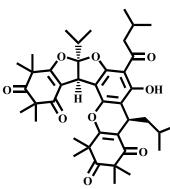
**ART7** :  $R_1 = R_2 = R_3 = OCH_3$  **ART13** :  $R_1 = R_3 = OCH_3$ ,  $R_2 = OH$  **ART16** :  $R_1 = R_2 = OCH_3$ ,  $R_3 = OH$ **ART17** :  $R_1 = OH$ ,  $R_2 = R_3 = OCH_3$ 



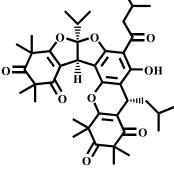








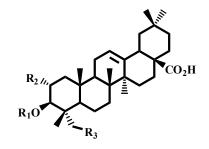




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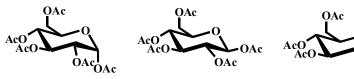
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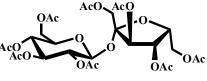


ART26

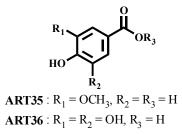
 $\begin{aligned} \mathbf{ART22} &: \mathbf{R}_1 = \textit{trans-coumaroyl}, \mathbf{R}_2 = \mathrm{OH}, \mathbf{R}_3 = \mathrm{H} \\ \mathbf{ART23} &: \mathbf{R}_1 = \textit{cis-coumaroyl}, \mathbf{R}_2 = \mathrm{OH}, \mathbf{R}_3 = \mathrm{H} \\ \mathbf{ART32} &: \mathbf{R}_1 = \textit{trans-coumaroyl}, \mathbf{R}_2 = \mathbf{R}_3 = \mathrm{H} \\ \mathbf{ART34} &: \mathbf{R}_1 = \mathrm{H}, \mathbf{R}_2 = \mathbf{R}_3 = \mathrm{OH} \\ \mathbf{ART39} &: \mathbf{R}_1 = \mathbf{R}_2 = \mathbf{R}_3 = \mathrm{H} \end{aligned}$ 



ART27

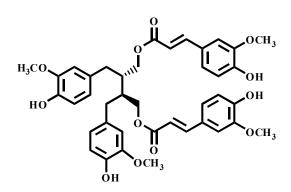






**ART40** :  $R_1 = R_2 = OH, R_3 = CH_3$ 

ART37



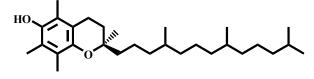
ART33

Thesis Title	Chemical Constituents from Rhodomyrtus tomentosa (Aiton) Hassk		
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Author	Mr. Asadhawut Hiranrat		
Major Program	Organic Chemistry		
Academic Year	2010		

### ABSTRACT

*Rhodomyrtus tomentosa* (Aiton) Hassk. is a flowering plant belonging to the family Myrtaceae. The preliminary study has revealed that the crude CH<sub>2</sub>Cl<sub>2</sub> and Me<sub>2</sub>CO extracts from its leaves exhibited strong antibacterial activities against Stephylococcus aureus ATCC 25923 and methicillin-resistant S. aureus NRPC R01 (MRSA) with MIC values of 31.25 and 62.5  $\mu$ g/mL, respectively. Investigation of the chemical constituents from the leaves, stems and fruits of *R. tomentosa* yielded forty one compounds. They were determined to be eleven acylphloroglucinols: rhodomyrtosone A (ART2), rhodomyrtosone H (ART3), rhodomyrtosone C (ART4), rhodomyrtone (ART6), endoperoxide G3 (ART8), rhodomyrtosone B (ART9), rhodomyrtosone D (ART11), rhodomyrtosone E (ART18), rhodomyrtosone G (ART19), rhodomyrtosone F (ART20) and rhodomyrtosone I (ART38), four flavonoids: combretol (ART7), 3,4',5',7-tetra-O-methylmyricetin (ART13), 3,3',5',7tetra-O-methylmyricetin (ART16) and 3,3',4',5'-tetra-O-methylmyricetin (ART17), seven terpenoids: (6*R*,7*E*,9*S*)-9-hydroxy-4,7-megastigmadien-3-one (ART12), loliolide (ART14), 3*β*-O-E-coumaroylmaslinic acid (ART22), 3*β*-O-Z-coumaroylmaslinic acid (ART23), 3*β-O-E*-coumaroyloleanolic acid (ART32), arjunolic acid (ART34) and oleanolic acid (ART39), three steroids:  $\beta$ -sitosterol (ART5),  $\beta$ -sitosterol glucopyranoside (ART24) and stigmast-4-en-3-one (ART37), three ellagic acid derivatives: 3,3',4-tri-O-methylellagic acid (ART10), 4-O-[\beta-D-glucopyranosyltetraacetate]-3,3',4'-tri-O-methylellagic acid (ART28) and 3-O-methyl- ellagic acid 4-O- $\alpha$ rhamnopyranoside (ART41), two flavellagic acid derivatives: 3',4'-dioxymethyleneviii 3,4-di-*O*-methyl-flavellagic acid (**ART25**) and 3,4,3',4'-tetra-*O*-methylflavellagic acid (**ART31**), six benzenoids:  $\alpha$ -tocopherol (**ART1**), *trans*-triacontyl-4-hydroxy-3-methoxycinnamate (**AR21**), *trans*-triacontyl-4-hydroxy-cinnamate (**ART30**), 4-hydroxy-3-methoxy-benzoic acid (**ART35**), gallic acid (**ART36**) and methyl gallate (**ART40**), one lignan: 9,9'-*O*-diferuloyl-(-)-secoisolariciresinol (**ART33**), one butyrolactone derivative: (3a*S*\*,6a*R*\*)-3a-(hydroxymethyl)-2,2-dimethyldihydrofuro[3,4-*d*][1,3] dioxol-4(3a*H*)-one (**ART15**) and three sugars:  $\beta$ -D-glucopyranoside penta-acetate (**ART26**),  $\alpha$ -D-glucopyranoside penta-acetate (**ART27**) and sucrose octa-acetate (**ART29**). Eleven compounds: **ART2**, **ART3**, **ART4**, **ART9**, **ART11**, **ART15**, **ART18**, **ART19**, **ART20**, **ART31** and **ART38** were newly found compounds. Their structures were elucidated on the basis of spectroscopic analyses including UV, IR, NMR, MS and by comparison of their spectroscopic data with those reported in the literature.

Some of the isolated acylphloroglucinol compounds (ART2, ART4, ART6, ART9, ART11, ART18, ART19 and ART20) were also evaluated for their antibacterial activity against three types of Gram-positive bacteria, *S. aureus*, MRSA and *Streptococcus pyogenes* DMST 101. It was found that rhodomyrtone (ART6) and rhodomyrtosone B (ART9) showed good activity to inhibit the growth of *S. aureus*, MRSA and *S. pyogenes* with MIC values of 0.39, 6.25  $\mu$ g/mL, 0.39, 12.5  $\mu$ g/mL, and 0.39, 3.125  $\mu$ g/mL, respectively. In addition, rhodomyrtosone G (ART19) inhibited the growth of both *S. aureus* and MRSA with the MIC value of 1.56  $\mu$ g/mL while rhodomyrtosone D (ART11) further showed the inhibitory activity against *S. pyogenes* with the MIC value of 12.5  $\mu$ g/mL. Interestingly, rhodomyrtone (ART6) exhibited the activity more than the standard, vancomycin (*S. aureus*: 0.60  $\mu$ g/mL, MRSA: 1.25  $\mu$ g/mL).



ART1

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ART4

R<sub>2</sub>

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 $R_3$ 

**ART6** :  $R_1$  = isovaleryl,  $R_2$  = H,  $R_3$  = isobutyl

**ART9** :  $R_1 = H$ ,  $R_2 = isovaleryl$ ,  $R_3 = isobutyl$ 

**ART38** :  $R_1$  = isovaleryl,  $R_2$  = H,  $R_3$  = phenyl

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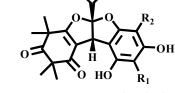
ART8

**ART19** :  $R_1 = 2$ -methylbutyryl,  $R_2 = H$ ,  $R_3 = isobutyl$ 

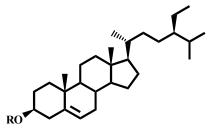
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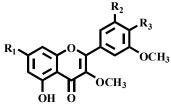
R<sub>1</sub>

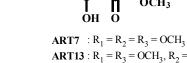


**ART2** :  $R_1 = H$ ,  $R_2 = i$  sovaleryl **ART3** :  $R_1 = i$  sovaleryl,  $R_2 = H$ 

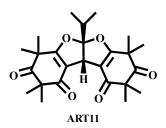


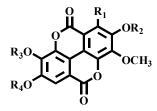
**ART5** : R = H**ART24** :  $R = -\beta$ -D-glucose



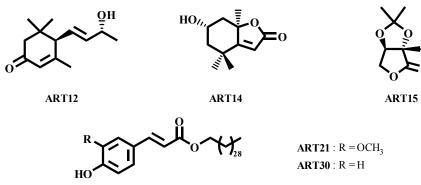


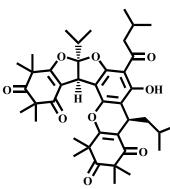
**ART13** :  $R_1 = R_3 = OCH_3$ ,  $R_2 = OH$  **ART16** :  $R_1 = R_2 = OCH_3$ ,  $R_3 = OH$ **ART17** :  $R_1 = OH$ ,  $R_2 = R_3 = OCH_3$ 

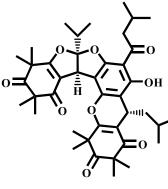




**ART10**:  $R_1 = R_2 = H$ ,  $R_3 = R_4 = CH_3$  **ART25**:  $R_1 = OH$ ,  $R_2 = CH_3$ ,  $R_3$ ,  $R_4 = -CH_2$ - **ART28**:  $R_1 = H$ ,  $R_2 = -\beta$ -D-glucose tetra-acetate,  $R_3 = R_4 = CH_3$  **ART31**:  $R_1 = OH$ ,  $R_2 = R_3 = R_4 = CH_3$ **ART41**:  $R_1 = R_3 = R_4 = H$ ,  $R_2 = -\beta$ -D-rhamnose





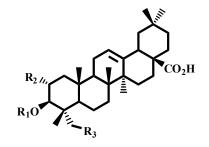


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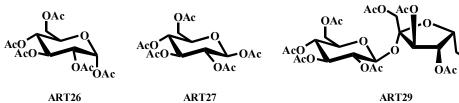
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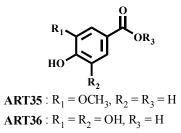
**ART22** :  $R_1 = trans$ -coumaroyl,  $R_2 = OH$ ,  $R_3 = H$ **ART23** :  $R_1 = cis$ -coumaroyl,  $R_2 = OH$ ,  $R_3 = H$ **ART32** :  $R_1 = trans$ -coumaroyl,  $R_2 = R_3 = H$ **ART34** :  $R_1 = H$ ,  $R_2 = R_3 = OH$ **ART39** :  $R_1 = R_2 = R_3 = H$ 



ART29

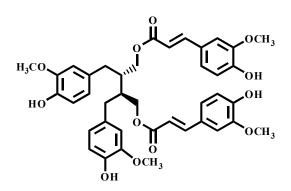
xi

OAc



**ART40** :  $R_1 = R_2 = OH, R_3 = CH_3$ 

ART37



ART33

### ACKNOWLEDGEMENTS

I wish to express my deepest and sincere gratitude to my supervisor, Associate Professor Dr. Wilawan Mahabusarakam, for her valuable instruction, expert guidance, excellent suggestion and kindness. I would also like to express my appreciation to Associate Professor Chanita Ponglimanont, my co-advisor, for correction of my thesis and her kindness. Without their help, my thesis work would not be successful.

My sincere thanks are expressed to Associate Professor Dr. Supayang Piyawan Voravuthikunchai and Mr. Surasak Limsuwan, Department of Microbiology, Faculty of Science, Prince of Songkla University for antibacterial activities testing and to Mr. J. Wai, Department of Biology, Faculty of Science, Prince of Songkla University for plant identification. I also would like to thank Assoc. Prof. Dr. Anthony R. Carroll, School of Environment, Griffith University, Queensland, Australia for his valuable instruction, expert guidance, excellent suggestion and kindness during my research visit with him at Griffith University.

I would like to acknowledge my sincere thanks to the Thailand Research Fund through the Royal Golden Jubilee Ph.D. program (Grant No. PHD/0206/2549) and the Center for Innovation in Chemistry (PERCH-CIC), Office of the Higher Education Commission, Ministry of Education for a scholarship and financial support. The CHE-RES-RG, Office of the Higher Education Commission, Ministry of Education and the Graduate School, Prince of Songkla University are gratefully acknowledged for the partly financial support.

I would like to thank Department of Chemistry, Faculty of Science, Prince of Songkla University for making available the facilities used in this research.

Finally, I am greatly indebted to my family especially my wife and my daughters for their encouragement, love, understanding and moral support.

Asadhawut Hiranrat

### THE RELEVANCE OF THE RESEARCH WORK TO THAILAND

This work is a basic research on the evaluation for utilization of Thai medicinal plants as a source of the bioactive constituents. The aims of this research are to investigate the chemical constituents of *Rhodomyrtus tomentosa* and to evaluate the antibacterial activity. In this research, we have reported on the isolation and structural elucidation of forty one compounds of eleven new and thirty known compounds from the leaves, stems and fruits of *R. tomentosa*. The crude dichloromethane and acetone extracts showed strong antibacterial activity. The isolated compounds were also evaluated for their antibacterial activity. We found that a known acylphloroglucinol, rhodomyrtone (**ART6**) and the new one, rhodomyrtosone B (**ART9**) showed strong activity. Interestingly, rhodomyrtone (**ART6**) exhibited better activity than the standard, vancomycin. This research demonstrated that *R. tomentosa* can be utilized as a potential source of the bioactive compounds.

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## LIST OF ABBREVIATIONS AND SYMBOLS

S	=	singlet
d	=	doublet
t	=	triplet
т	=	multiplet
q	=	quartet
quin	=	quintet
sext	=	sextet
hept	=	heptet
dd	=	doublet of doublet
dt	=	doublet of triplet
tt	=	triplet of triplet
td	=	triplet of doublet
ddd	=	doublet of doublet of doublet
br	=	broad
br s	=	broad singlet
μg	=	microgram
mg	=	milligram
g	=	gram
kg	=	kilogram
%	=	percentage
nm	=	nanometer
mp	=	melting point
cm <sup>-1</sup>	=	reciprocal centimeter (wave number)
δ	=	chemical shift relative to TMS
J	=	coupling constant
$\lambda_{ m max}$	=	maximum wavelength
V	=	absorption frequencies

# LIST OF ABBREVIATIONS AND SYMBOLS (Continued)

ε	=	molar extinction coefficient		
<sup>o</sup> C	=	degree celceis		
MHz	=	Megahertz		
ppm	=	part per million		
IR	=	Infrared		
UV	=	Ultraviolet		
EI-MS	=	Electron Impact Mass Spectroscopy		
HREI-MS	=	High-Resolution Electron Impact Mass Spectroscopy		
FAB-MS	=	Fast Atom Bombardment Mass Spectroscopy		
HRFAB-MS	=	High-Resolution Fast Atom Bombardment Mass Spectroscopy		
NMR	=	Nuclear Magnetic Resonance		
2D NMR	=	Two Dimentional Nuclear Magnetic Resonance		
COSY	=	Correlated Spectroscopy		
DEPT	=	Distortionless Enhancement by Polarization Transfer		
HMBC	=	Heteronuclear Multiple Bond Correlation		
HMQC	=	Heteronuclear Multiple Quantum Coherence		
NOE	=	Nuclear Overhouser Enhancement		
NOESY	=	Nuclear Overhouser Enhancement Spectroscopy		
ROESY	=	Rotating-frame Overhouser Enhancement Spectroscopy		
TMS	=	tetramethylsilane		
CDCl <sub>3</sub>	=	deuterochloroform		
CD <sub>3</sub> OD	=	tetra-deuteromethanol		
DMSO- $d_6$	=	hexa-deuterodimethylsulphoxide		
$CH_2Cl_2$	=	dichloromethane		
Me <sub>2</sub> CO	=	acetone		
MeOH	=	methanol		

# LIST OF ABBREVIATIONS AND SYMBOLS (Continued)

CC	=	column chromatography
QCC	=	quick column chromatography
TLC	=	thin-layer chromatography
MIC	=	minimum inhibition concentration

### **CHAPTER 1**

### **INTRODUCTION**

#### 1.1 Introduction

In the present time, all people around the world have extensively concerned about their health which in turn occur from many causes such as air pollutions, water pollutions, and unhealthy food consumption. These are the causes of many types of diseases including cancer, diabetes, alzheimer, hypertension, rheumatoid arthritis, inflammatory bowel disease, immune system decline, brain dysfunctions, cataracts and malaria *etc*. Thus, the new drug discovery or the development in nutrient supplement has very high competition between the suppliers within country and also abroad. Herbal plants play important roles and are mainly sources of natural remedy which are neglected for a long time in many countries. People gained knowledge to utilize the leaves, stem, bark, fruits, bulbs and roots of the plants in the folk medicines from the ancestors for their basic health care needs including the treatments of infections. Furthermore some cosmetics have herbs and some parts of plants for their active ingredients. Unsurprisingly, the medicinal plants are interesting as a source of pharmacologically active substances.

Thailand is in the tropical country abundance with many kinds of herbal plants that can promise to cure many diseases such as *Andrographis paniculata* Wall. ex Nees (ฟ้าทะลายโจร) reliefs the symptom of cold, *Curcuma longa* Linn (บมิ้นชัน) protects and heals ulcer and *Aloe barbadensis* Mill (ว่านหางจระเข้) uses as ingredients in cosmetics *etc*. Therefore Thai scientists have realized that it is necessary to conduct a research on active constituents from herbal plants which have pharmacological and biological activities.

#### **1.2** The family Myrtaceae

#### 1.2.1 Description of the Myrtaceous plants

The family Myrtaceae (Myrtle or guava family), which consists of evergreen trees or shrubs usually with essential oils-containing cavities in foliage, has at least 4,500 species, possibly more than 5,000 species, and is distributed in about 130 genera. This family is the eighth largest flowering plant family. They have a wide distribution in tropical and warm-temperate regions of the world such as Mediterranean, sub-Saharan Africa, Madagascar, tropical and temperate Asia, Australia, Pacific islands, tropical and South America (Jie and Craven, 2007). The Myrtaceous plants were conventionally classified in which the primary division into two subfamilies based on morphological of the fruits, capsular-fruited Leptospermoideae and fleshy-fruited Myrtoideae (Wilson et al., 2001). Capsularfruited genera such as Eucalyptus, Corymbia, Angophora, Leptospermum, Melaleuca, *Metrosideros* are absent from the Americas except for the monotypic Chilean genus Tepualia (Lucas et al., 2005). Fleshy-fruited genera have their greatest concentrations in eastern Australia and the Neotropics. *Eucalyptus* is a dominant, nearly ubiquitous genus in Australia and extends north sporadically to the Philippines. Eucalyptus regnans or Mountain Ash is the tallest flowering plant in the world, reaching heights of more than 100 meters (State of Victoria, 2003).

#### 1.2.2 Advantages and traditional uses of the Myrtaceous plants

Many Myrtaceae are cultivated as a popular ornamental plant in gardens, street trees or plantation trees in the tropical and subtropical areas, and some members grown for its abundant flowers and sweeten edible fruits (Jie and Craven, 2007). Furthermore, Myrtaceous plants are economically important in the spices, fruits, honeys, timbers and pharmacology industries with other economic potential beginning to be realized (bioactive compounds, vitamin-rich soft fruits *etc.*). *Eucalyptus* is one of the world's most important and most widely planted genera and is a large genus of aromatic trees comprising more than 900 species (Tian *et al.*, 2009). It is widely cultivated to provide shade and for the timber and pulp industries (Menut *et al.*, 1995;

Ghisalberti, 1996; Kim et al., 2001; Benyahia et al., 2005; Hasegawa et al., 2008; Singh et al., 2009; Tian et al., 2009). Syzygium aromatica (clove) (Charles et al., 1998; Jirovetz et al., 2006) and Pimenta dioica (allspice) (Kikuzaki et al., 2008; Nitta et al., 2009) are important in the spice industry. Myrtus comminis (Tuberoso et al., 2006), Pimenta racemosa (bay rum) (García et al., 2004), Melaleuca (cajeput), Euginia uniflora (Brazilian cherry tree) (Amorim et al., 2009) and Eucalyptus (Menut et al., 1995; Singh et al., 2009; Goodger et al., 2009) provide oils for the perfume industry, while antiseptic oils are extracted from *Eucalyptus* (Ghisalberti, 1996; Siddiqui et al., 2000; Benyahia et al., 2005; Hasegawa et al., 2008), Melaleuca alternifolia (tea tree) (Russell and Southwell, 2002), Callistemon and Leptospermum (Melching et al., 1997). Almost all fleshy-fruited Myrtaceae are edible; economically important fruits are *Psidium guajava* (guava) (Lapčík *et al.*, 2005; Reynertson *et al.*, 2008; Steinhaus et al., 2008), Syzygium jambos (rose apple), Syzygium malaccensis (Malay apple), Syzygium samaramgense (wax apple), Syzygium aqueum (water apple) (Nonaka et al., 1992) and Feijoa sellowiana (pineapple guava) (Ruberto and Tringali, 2004), with many lesser known species locally important for juice, sweets and jams, such as *Myrciaria cauliflora* (jaboticaba) (Reynertson *et al.*, 2006), Campomanesia lineatifolia (Osorio et al., 2006), Feijoa sellowiana (Ruberto and Tringali, 2004; Rossi et al., 2007; Weston, 2010) and Eugenia uniflora (pitangueira).

Furthermore, several Myrtaceous plants have also been used in folk medicinal purposes. In Hong Kong, *Baeckea frutescens* is widely used in traditional medicine for treating rheumatism and snake bites (Tsui and Brown, 1996) and as an anti-febrile in Southeast Asia and China (Fujimoto *et al.*, 1996). The leaves and stem bark of *Campomanesia xanthocarpa*, Brazilian species, are traditionally employed as a remedy for dysentery, stomach problems, fever and as anti-inflammatory agent (Markman *et al.*, 2004). Some *Kunzea* species in New Zealand have been reported to utilize the essential oils, including *K. ambigua*, for the treatment of diarrhea, cold, inflammation, and wounds (Ito *et al.*, 2004). The bark and leaves of *Melaleuca leucadendron* are used in folk medicine in Taiwan as tranquiling, sedating, evildispelling, and pain-relieving agents (Lee, 1998). In Brazil, an astringent decoction of the sun-dried skins of *Myrciaria cauliflora* (jaboticaba) has traditionally been used as

a treatment for hemoptysis, asthma, and diarrhea and gargled for chronic inflammation of the tonsils (Reynertson *et al.*, 2006). *Plinia edulis* has commonly been employed in the treatment of stomach disorders, throat infections, diabetes and also as a tonic by traditional seaside settlers of the Brazilian southeastern coast (Ishikawa *et al.*, 2008). *Rhodomyrtus tomentsa* has been used as traditional medicine for diarrhea and wound treatments in Vietnam (Tung *et al.*, 2009).

*Psidium guajava* (guava) has been claimed to be useful in a traditional medicine for the treatment of various human ailments such as wounds, ulcers, bowels and cholera. The young leaves are used as a tonic in diseases of digestive function. The decoction of young leaves and shoots is prescribed as a febrifuge, diarrheic disease and spasmolytic effect (Begum *et al.*, 2002; Oh *et al.*, 2005). The bark is valued as an astringent and as an anti-diarrhoeatic in children whereas the flowers are used to cool the body and for the treatment of bronchitis and eye sores. Furthermore, the fruit has a tonic and laxative and is good for bleeding gums (Begum *et al.*, 2002).

In New Zealand, *Leptospermum scoparium* have been employed for the treatment of fevers and pain by the Maori tribes. In addition, its honey has been reported to exhibit antimicrobial activity against *Staphylococcus aureus* and *Helicobacter pylori* (Jeong *et al.*, 2009). Traditionally, *L. recurvum* has been used to stimulate appetite and relieve stomach disorders and menstrual discomfort in Malaysia (Mustafa *et al.*, 2003).

*Pimenta racemosa* var. *ozua* is widely used in the folk medicine of the Caribbean basin for different afflictions; for example, in the Dominican Republic, the essential oil from leaves has commonly used for the local treatment of rheumatism or for toothache. The decoction of seed is not used only as a stimulant in Cuba but it is also employed for colds and influenza in Trinidad. In addition, the decoction of leaves has been used to treat abdominal pains in Haití (Fernández *et al.*, 2001). *P. dioica* (allspice) has widely used in foods as spice such as its oil has been claimed to relieve neuralgia and rheumatism (Nitta *et al.*, 2009) and acted as an antimicrobial and a digestive agent (Kikuzaki et al., 2008).

In the *Eucalyptus* genus, several species have been reported for utilization in traditional medicine such as essential oils from its leaves are acted as antiseptics for

the treatment of infection of upper respiratory tract, the common cold, influenza and sinus congestion (Siddiqui *et al.*, 2000; Hasegawa *et al.*, 2008) and for curing certain skin diseases while the gum is used in diarrhoea and as astringent in dentistry, cuts, *etc* (Siddiqui *et al.*, 2000). The leaves extract has become a potential for larvicidal and repellent properties against mosquito vectors with very eco-friendly (Nathan, 2007). The essential oil further has a therapeutic application to treat pulmonary infections by inhalation (Hasegawa *et al.*, 2008). Their barks and leaves have been believed to cure colds, influenza, toothaches, snakebites, fevers, diarrhea and other complaints (Kim *et al.*, 2001). The leaves of *E. robusta* are also used for the treatment of dysentery, malaria, and other bacterial diseases (Xu *et al.*, 1984). The essential oil from the leaves of *E. tereticornis* has long been recognized for its insecticidal properties, especially its mosquito repellent activity (Nathan, 2007).

The *Eugenia* genus has been reported to be used as several folk medicines. For examples, the infusions or decoctions of *E. uniflora* (Brazilian cherry) leaves have been used as a popular medicine for the treatment of inflammations, against rheumatic pains and fever, as hypoglycemiant, diuretic and to avoid stomach problems (Consolini and Sarubbio, 2002; Amorim *et al.*, 2009) whereas its red fruits have been used in infusions as an antihypertensive agent as well as in the treatment of digestive disorders (Consolini *et al.*, 1999). The infusions or decoction of the leaves of *E. punicifolia* have been used in the treatment of hyperglycaemic disturbs, such as diabetes mellitus (Grangeiro *et al.*, 2006).

The *Syzygium* genus has further been reported for several different uses in medicine; for examples, *S. aromaticum* buds (clove) have folk medicinally been used as diuretic, odontalgic, stomachic, tonicardiac, aromatic condiment and condiment with carminative and stimulant properties (Nassar, 2006). Traditionally the bark of *S. jambos* (syn. *E. jambos*) has been used to treat pernicious attack, amenorrhea, abdominal pain and diarrhea (Djipa *et al.*, 2000). Furthermore, in Venezuela, its leaves infusions have been used in traditional medicine as febrifuge and remedy for the relief of inflammatory pain, especially in sore throats (Ávila-Peña *et al.*, 2007). In South Africa, the barks, leaves and roots of *S. cordatum* have been claimed to use for the treatment of ailments including tuberculosis, diarrhoea, stomach and respiratory

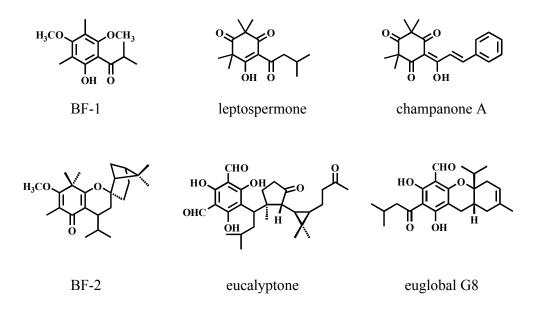
complaints (Musabayane *et al.*, 2005). The barks of *S. guineense* which distributed in Subsaharan Africa (Uganda, Swaziland and Cameroon) have been traditionally used for the treatment of stomachache and diarrhea (Djoukeng *et al.*, 2005). In Malaysia, the leaves of *S. aqueum* (watery rose apple or water apple) have been used to treat a cracked tongue whereas the root to relieve itching and to reduce swelling (Osman *et al.*, 2009).

#### 1.2.3 Unique secondary metabolites of the family Myrtaceae

Many plants in the family Myrtaceae have been phytochemically investigated and also have been reported on the isolation of the several constituents including phloroglucinols, flavonoids, anthocyanins, terpenoids, tannins, and stilbenoids. Among these the phloroglucinols, a rare natural product containing the unique structure, which mainly obtained from the plants of the family Myrtaceae, have been recently become the interesting compounds. Apart from the structural identity, they also exhibited significantly the wild ranges of biological activities.

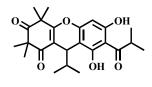
Myrtaceous plants produced a range of phloroglucinols, a major class of secondary metabolites, with different levels of methylation of the nuclear carbons and oxygens. These compounds have been classified according to a number of phloroglucinol units into monomeric phloroglucinols, dimeric phloroglucinols, trimeric phloroglucinols, tetrameric and higher phloroglucinols, and phlorotannins (Singh and Bharate, 2006).

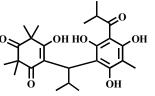
A large number of differently substituted and structurally diverse monomeric phloroglucinols have been reported to occur amongst plants as well as other natural sources, and have been shown to possess various biological activities. Monomeric phloroglucinols can be further divided into different subclasses such as acylphloroglucinols: BF-1, leptospermone and champanone A (Fujimoto *et al.*, 1996; Bonilla *et al.*, 2005) and phloroglucinol-terpene adducts: BF-2, eucalyptone and euglobal G8 (Fujimoto *et al.*, 1996; Osawa *et al.*, 1995; Umehara *et al.*, 1998) *etc.* 

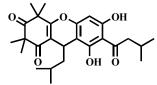


Structures of some monomeric phloroglucinols

Dimeric phloroglucinols comprised of two units of phloroglucinol joined either through a carbon-carbon linkage or by the formation of a chroman ring and have been found in many genera including *Myrtus*, *Lophomyrtus*, *Rhodomyrtus*, *Eucalyptus*, *Kunzea*, and *Corymbia etc.* such as myrtucommulone B (Shaheen *et al.*, 2006), semimyrtucommulone (Appendino *et al.*, 2002), rhodomyrtone (Dachriyanus *et al.*, 2002), dimer of jensenone (Mitaine-Offer *et al.*, 2003), bullataketals A and B (Larsen *et al.*, 2005), and corymbones A and B (Carroll *et al.*, 2008b).

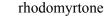


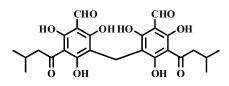




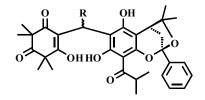
myrtucommulone B

semimyrtucommulone

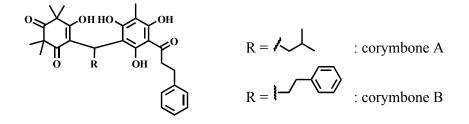




dimer of jensenone

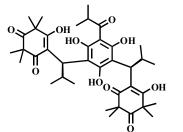


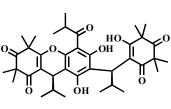
R = β-isopropyl: bullataketal A R = α-isopropyl: bullataketal B

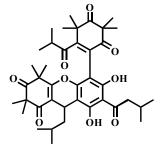


Structures of some dimeric phloroglucinols

Many genera of the Myrtaceous plants such as *Callistemon, Corymbia*, *Eucalyptus*, and *Myrtus etc.* have been reported to produce the trimeric phloroglucinols. Examples of trimeric phloroglucinols were myrtucommulone A, myrtucommulones C-I (Kashman *et al.*, 1974; Carroll *et al*, 2008a; Appendino *et al.*, 2002) and eucalyptone G (Mohamed and Ibrahim, 2007).

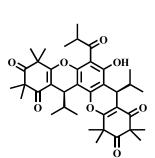




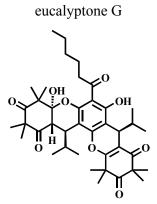


myrtucommulone A

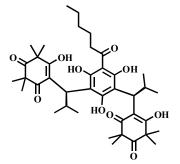
myrtucommulone C



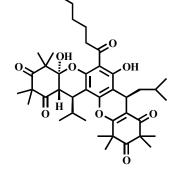
myrtucommulone E



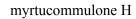
myrtucommulone G



myrtucommulone D



myrtucommulone F

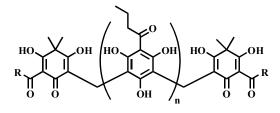


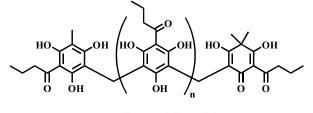
myrtucommulone I

Structures of some trimeric phloroglucinols



Tetrameric and higher phloroglucinol compounds covered the phloroglucinol derivatives bearing more than three phloroglucinol units. Based on the literature surveys, this group has not been found to isolate from the plants in the family Myrtaceae. They were reported on the isolation from the other sources such as tetraalbaspidin ABBA (dryocrassin) from ferns *Dryopteris crassirhizoma*. Tetraalbaspidin BBBB was independently isolated from the ferns *D. austriaca* and *D. aitoniana* whereas tetraflavaspidic acid BBBB, penta-albaspidin BBBBB, hexaflavaspidic acid and hexa-albaspidin BBBBB have been reported on the isolation from *D. aitoniana* (Singh and Bharate, 2006).



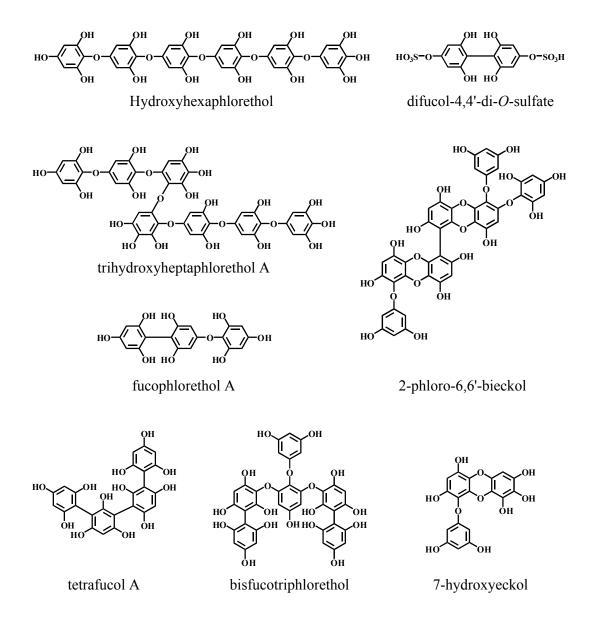


n = 2 : tetraflavaspidic acid BBBB n = 4 : hexaflavaspidic acid

Structures of some tetrameric and higher phloroglucinols

Phlorotannins consist of phloroglucinol units linked to each other in various ways, and are of wide occurrence amongst marine organisms, especially brown and red algae. This group can be classified into four subclasses: phlorotannins with an ether linkage (fuhalols and phlorethols) such as hydroxyhexaphlorethol and

trihydroxyheptaphlorethol A, with a phenyl linkage (fucols) such as difucol-4,4'-di-*O*-sulfate and tetrafucol A, with an ether and a phenyl linkage (fucophlorethols) such as fucophlorethol A and bisfucotriphlorethol, and with a dibenzodioxin linkage (eckols and carmalols) such as 7-hydroxyeckol and 2-phloro-6,6'-bieckol (Singh and Bharate, 2006).

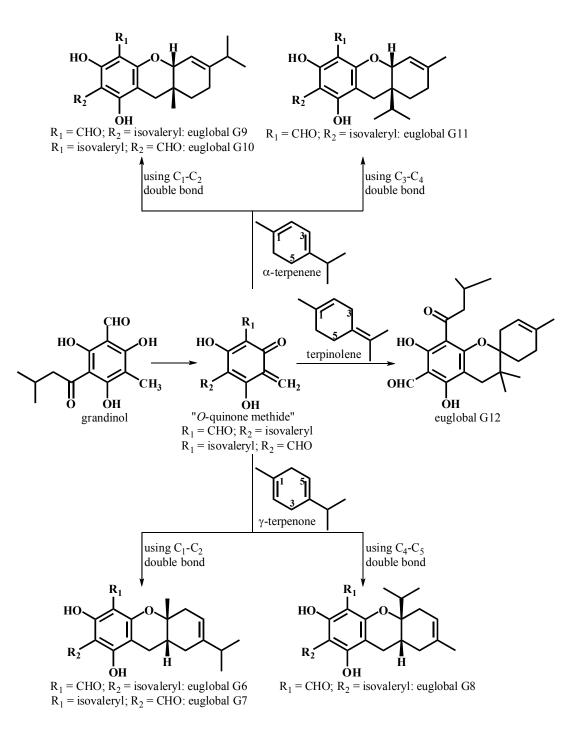


Structures of some phlorotannins

#### 1.2.4 Biosynthetic proposal of some phloroglucinols

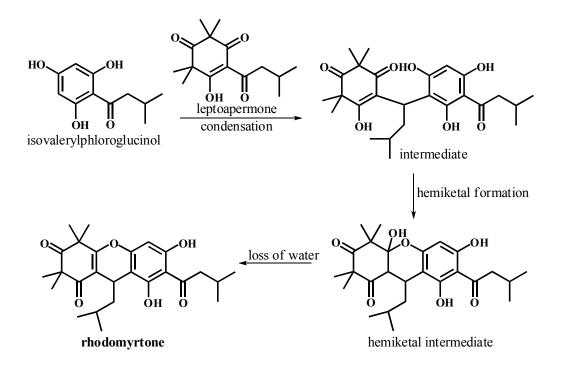
The biosynthesis of naturally myrtucommulones from *Myrtus communis* has been proposed in 1974 by Kashman and co-workers. They could be derived from the similar pathway suggested for the fern acylphloroglucinols through cyclization of linear polyketomethylene intermediates (Kashman *et al.*, 1974).

The phloroglucinol-monoterpene adducts named euglobals G6-G12 could be biogenetically formed between an appropriate monoterpene and an *O*-quinone methide derived from grandinol by Diels-Alder cycloaddition (Singh *et al.*, 1998; Umehara *et al.*, 1998).



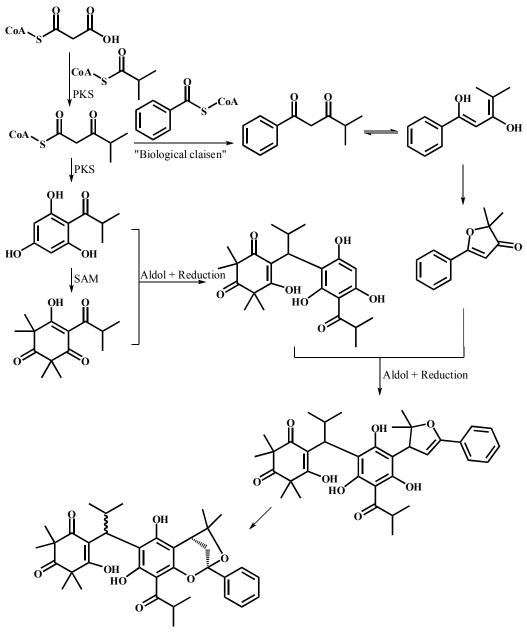
Proposed biosynthetic pathway for euglobals G6 - G12

Dachriyanus and co-workers have suggested that the biogenetically route for rhodomyrtone involved a condensation of leptospermone with the appropriate isovalerylphloroglucinol followed by hemiketal formation between phenolic hydroxyl group and a keto group with concomitant loss of water (Dachriyanus *et al.*, 2002).



Proposed biosynthetic pathway of rhodomyrtone

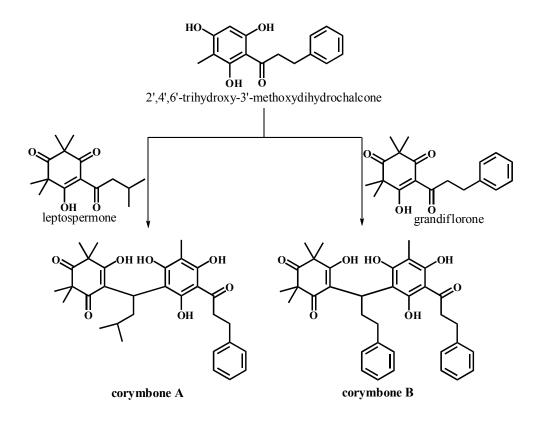
Bullataketals A and B have been biosynthetically proposed by Larsen and coworkers in year 2005. This route involved an enzymatic process, polyketide synthase (PKS) inducing the combination between isobutyryl-CoA and three malonyl-CoA units to form an isobutyrylphloroglucinol. Methylation of this phloroglucinol with *S*adenosyl methionine (SAM) could produce a  $\beta$ -triketone moiety. Aldol-like condensation of phloroglucinol and  $\beta$ -triketone units would give intermediate as a mixture of stereoisomers, followed by combination with bullatenone. Bullataketals A and B could finally be derived by acid-catalysed cyclisation (Larsen *et al.*, 2005).



bullataketals A and B

Proposed biosynthetic pathway of bullataketals A and B

The corymbone phloroglucinols could be biogenetically derived from the condensation of 2',4',6'-trihydroxy-3'-methyldihydrochalcone with the appropriate cyclohexatriones (Carroll *et al.*, 2008a).



Proposed biosynthetic pathway of corymbones A and B

## 1.2.5 Biological activities of some phloroglucinols

Some of phloroglucinols from the family Myrtaceae exhibited interesting antibacterial activities. The constituent from the leaves of *Rhodomyrtus tomentosa*, rhodomyrtone, showed significant activity to inhibit *Escherichia coli* and *Staphylococcus aureus*, but without the value and details reported (Dachriyanus *et al.*, 2002). Eucalyptone, obtained from the leaves of *Eucalyptus globulus*, has antibacterial activity against cariogenic bacteria including against *Streptococcus mutans* (MIC 6.25  $\mu$ g/mL) and *Streptococcus sobrinus* (MIC 12.5  $\mu$ g/mL). It also showed an inhibitory effect on adherent water-insoluble glucan synthesis by GTase with 97.6 and 44.0% inhibition at concentration of 100 and 10  $\mu$ g/mL, respectively, prepared from the supernatant of S. sobrinus. These data indicated that eucalyptone might be a promising natural substance for the development of a new cariostatic drug (Osawa et al., 1995). Eucalyptone G, isolated from the bark of small twigs of Eucalyptus globulus Labill, was found to be active against Bacillus subtilis, Staphylococcus aureus, and Escherichia coli (Mohamed and Ibrahim, 2007). Champanone A showed activity against *Micrococcus luteus*, *Staphylococcus aureus*, Bacillus subtilis, and Pseudomonas aeruginosa with the MIC value of 30  $\mu$ g/mL and against Streptococcus faecalis with the MIC value of 15  $\mu$ g/mL. Champanone B was active against M. luteus (MIC 30 µg/mL) and champanone C against both B. subtilis and S. faecalis (MIC 30  $\mu$ g/mL). These data indicated that the three champanones displayed mild antimicrobial activity (Bonilla et al., 2005). Myrtucommulone A which was first isolated from Myrtus communis L. showed strong antibacterial activity against Gram-positive bacteria with the concentration of  $1\gamma/mL$  (Kashman et al., 1974) whereas Appendino et al have also conducted from the leaves of Myrtus communis L. showing significant activity against multidrug-resistant (MRD) clinically relevant bacteria, Staphylococcus aureus (Appendino et al., 2002). Furthermore myrtucommulone A was also obtained from the leaves of *Callistemon* lanceolatus (Lounasmaa et al, 1977) and the seeds of Corymbia scabrida (Carroll et al, 2008b). Shaheen et al have reported that myrtucommulones D and E, isolated from the aerial parts of Myrtus communis L., exhibited significant antibacterial activity against Staphylococcus aureus (Shaheen et al, 2006). Myrtucommulone D was also isolated and identified from the seeds of Corymbia scabrida (Carroll et al, 2008b).

In 1992, Bloor have reported that 2,6-dihydroxy-4-methoxyisobutyrophenone and 4,6-dihydroxy-2-methoxyisobutyrophenone, the phloroglucinol constituents from the leaves and twigs of *Kunzea sinclairii*, showed antiviral activities against Herpes simplex Type I and Polio Type I viruses with the inhibition of cytopathic effect of either viruses of 5  $\mu$ g/disk (Bloor, 1992). Phloroglucinols from the leaves of *Eucalyptus globulus* named euglobals G6 and G7 have been reported as inhibitors of Epstein Barr Virus (EBV) activation induced by 12-*O*-tetradecanoylphorbol-13acetate (TPA). Both compounds showed *ca* 80% inhibition at 1000 mol ratio/TPA, while maintaining 70% viability of Raji cells (Singh *et al.*, 1998). In addition to euglobals G6 and G7, euglobals G8, G9 and G12 showed around 100% inhibition at 1000 mol ratio/TPA while maintaining 70% viability of the Raji cells (Umehara *et al.*, 1998).

Some of them acted as insecticidal such as 4-[1-(5,7-Dihydroxy-6-methyl-4oxo-2-phenyl-chroman-8-yl)-3-methyl-butyl]-5-hydroxy-2,2,6,6-tetramethyl-cyclohex -4-en-1,3-dione, which comprised a pair of epimers, each of which is a pair of conformers, from the hexane extract of the aerial parts of *Kunzea ambigua* and *K. baxterii.* It showed the LD<sub>50</sub> by topical application to mustard beetles, *Phaedon cochleariae*, the aphid, *Aphis fabae* and the thrips, *Thrips tabaci* of 1.0, 3.9 and 15  $\mu$ g/insect, respectively. A dose of 10  $\mu$ g/insect caused 83% mortality of houseflies, *Musca domestica.* (Khambay *et al.*, 2002). The constituents from the aerial parts of *Callistemon viminalis*, viminadione A, also exhibited insecticidal activity to houseflies, *Musca domestica*, the aphid, *Aphis fabae*, and the thrips, *Thrips tabaci* with the LD<sub>50</sub> of 1.9, 5.9 and 4.2  $\mu$ g/insect, respectively (Khambay *et al.*, 1999).

Apart from myrtucommulones A and D, the seeds of *Corymbia scabrida* have been reported on the isolation of four further trimeric phloroglucinols named myrtucommulones F-I. Myrtucommulones A, D and F-I inhibited the specific binding of [<sup>3</sup>H]3-methylhistidylTRH to HEK cell membranes expressing recombinant rat TRH receptor 2 with respectively IC<sub>50</sub> values of 39, 11, 16, 24, 31, and 16  $\mu$ M (Carroll *et al.*, 2008a). In addition, corymbones A and B obtained from the flowers of *Corymbia peltata* also exhibited rat TRH receptor 2 binding affinity with IC<sub>50</sub> values of 23 and 19  $\mu$ M, respectively (Carroll *et al.*, 2008b).

Phloroglucinols named myrtucommulones B-E were found to be more potent  $\alpha$ -glucosidase inhibitors than the clinically used standard acarbose while myrtucommulone C exhibited the highest activity among all the phloroglucinols with the IC<sub>50</sub> 35.4±1.15  $\mu$ M (Shaheen *et al.*, 2006).

Furthermore, some of them showed interesting cytotoxic activities such as BF-2 isolated from the dried leaves of *Baeckea frutescens*, against leukaemia cells (L 1210) in tissue culture with the IC<sub>50</sub> 5.0  $\mu$ g/mL (Fujimoto *et al.*, 1996).

The phloroglucinols, nemed bullataketals A and B from the leaves and twigs of *Lophomyrtus bullata*, have cytotoxic activity against the P388 mouse leukaemia cell line with the IC<sub>50</sub> 1  $\mu$ g/mL (Larsen *et al.*, 2005).

14 Genera with about 79 species of the family Myrtaceae were found in Thailand: *Acmena* (1 species), *Baeckia* (1 species), *Callistemon* (2 species), *Cleistocalyx* (2 species), *Decaspermum* (2 species), *Eucalyptus* (2 species), *Eugenia* (3 species), *Syzygium* (56 species), *Melaleuca* (1 species), *Myrtus* (1 species), *Psidium* (2 species), *Rhodamnia* (2 species), *Rhodomyrtus* (1 species) and *Tristania* (3 species) (Smitinand, 2001).

## 1.3 The Rhodomyrtus genus

#### 1.3.1 Chemical constituents of the *Rhodomyrtus* genus

The genus *Rhodomyrtus* comprises of about 20 species which widely distributed in tropical Asia, Australia and Southwest Pacific islands (Jie and Craven, 2007). Based on SciFinder Scholar database, only 8 species from the *Rhodomyrtus* plants have been reported for phytochemically investigation including *R. effussa*, *R. macrocapa*, *R. pervagata*, *R. psidioides*, *R. sericea*, *R. tomentosa*, *R. trineura* subsp. *trineura*, and *R. trineura* subsp. *capensis*. The chemical studies on this genus have concentrated on essential oils, with little work having been published on non-volatiles from *R. macrocapa* and *R. tomentosa*. The chemical constituents isolated from the *Rhodomyrtus* genus were summarized in **Table 1**.

Compounds	Structure	References
R. effussa		
Essential oil from leaves		
Globulol	8e	Brophy et. al., 1997
Viridiflorol	8f	
Spathulenol	8j	
R. macrocapa		
Fruit		
Rhodomyrtoxin	3a	Trippett, 1957
ψ-Rhodomyrtoxin	3b	Anderson et. al., 1969
Rhodomyrtoxin B	3c	Igboechi et al., 1984
Rhodomyrtoxin C	3d	
Essential oil from leaves		
$\beta$ -Caryophyllene	8h	Brophy et. al., 1997
Aromadendrene	8g	
Globulol	8e	
R. pervagata		
Essential oil from leaves		
<i>α</i> -Pinene	8b	Brophy et. al., 1997
$\beta$ -Pinene	8c	
R. psidioides		
Essential oil from leaves		
<i>α</i> -Pinene	8b	Brophy et. al., 1997
Limonene	8d	

 Table 1
 Compounds isolated from the plants of the *Rhodomyrtus* genus

# Table 1 (Continued)

Compounds	Structure	References
R. sericea		
Essential oil from leaves		
<i>a</i> -Pinene	8b	Brophy et. al., 1997
$\beta$ -Pinene	8c	
β-Caryophyllene	8h	
R. tomentosa		
Leaves		
Lupeol	8k	Hui et. al., 1975
β-Amyrin	8p	
$\beta$ -Amyrenonol	8q	
Betulin	81	
21 <i>αH</i> -Hop-22(29)-en-3 <i>β</i> ,30-diol	8n	Hui et. al., 1976
3β-Hydroxy-21αH-hop-22(29)-en-30-al	80	
Tomentosin	7a	Lui et. al., 1997
Peduculagin	7b	Lui et. al., 1998
Casuariin	7c	
Castalagin	7d	
Myricetin 3- $O$ - $\alpha$ -L-furanoarabinoside	4a	Hou et. al., 1999
Myricetin 3- $O$ - $\beta$ -D-glucoside	4b	
Myricetin 3- $O$ - $\alpha$ -L-rhamnoside	4c	
2,3-Hexahydroxydiphenyl-D-glucose	7e	
Rhodomyrtone	1	Dachriyanus et. al., 2002
Piceatannol 4- $O$ - $\beta$ -D-glucoside	6	Nojima et. al., 2007
Stems		
Friedelin	8t	Hui et. al., 1975
Lupeol	8k	
α-Amyrin	8s	

 Table 1 (Continued)

Compounds	Structure	References
R. tomentosa		
Stems		
Taraxerol	8r	Hui et. al., 1975
Betulin	8m	
Betulin-3-acetate	81	
$3\beta$ -Acetoxy-11 $\alpha$ , 12 $\alpha$ -epoxyoleanan-	8w	Hui et. al., 1976
28,13 <i>β</i> -olide		
$3\beta$ -Acetoxy- $12\alpha$ -hydroxyoleanan- $28,13\beta$ -	8v	
olide		
$3\beta$ -Acetoxy-12-oxo-oleanan-28,13 $\beta$ -olide	8u	
Flowers		
Malvidin-3-glucoside	2a	Lowry, 1976
Pelargonidin-3,5-biglucoside	2b	He et. al., 1998
Cyanidin-3-galactoside	2c	
Delphinidin-3-galactoside	2d	
Barks and Twigs		
Combretol (3,3',4',5',7-penta-O-	4d	Dachriyanus et. al., 2004
methylmyricetin)		
Aerial parts		
4,8,9,10-Tetrahydroxy-2,3,7-trimethoxy-	5a	Tung et. al., 2009
anthracene-6- $O$ - $\beta$ -D-glucopyranoside		
2,4,7,8,9,10-Hexahydroxy-3-methoxy-	5b	
anthracene-6-O-a-L-rhamnopyranoside		
Quercetin	4f	
Myricitrin	4e	
(3 <i>S</i> ,5 <i>R</i> ,6 <i>R</i> ,7 <i>E</i> ,9 <i>S</i> )-Megastiman-7-ene-	8a	
3,5,6,9-tetrol		

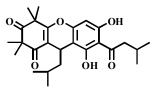
# Table 1 (Continued)

Compounds	Structure	References
R. tomentosa		
Buds		
Kaempferol 3- <i>O</i> -β-sambubioside		Phan <i>et al.</i> , 2007
Essential oil from leaves		
<i>a</i> -Pinene	8b	Brophy et. al., 1997
$\beta$ -Pinene	8c	
Aromadendrene	8g	
R. trineura subsp. trineura		
Essential oil from Leaves		
$\beta$ -Caryophyllene	8h	Brophy et. al., 1997
Caryophyllene oxide	8i	
Globulol	8e	
R. trineura subsp. capensis		
Essential oil from Leaves		
<i>a</i> -Pinene	8b	Brophy et. al., 1997
Globulol	8e	
Viridiflorol	8f	
Spathulenol	8j	

The literature surveys demonstrated that the *Rhodomyrtus* genus has been investigated and resulted in the isolation of several components including acylphloroglucinols, anthocyanins, dibenzofurans, flavonoids, naphthalenoids, stilbenoids, tannins, and terpenoids. The structures of compounds isolated from the *Rhodomyrtus* genus were summarized as following.

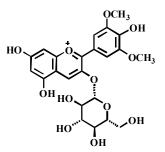
Structures of the isolated compounds from the Rhodomyrtus genus

1. Acylphloroglucinols

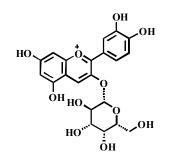


1a: rhodomyrtone

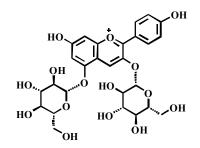
# 2. Anthocyanins



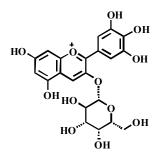
2a: malvidin-3-glucoside



2c: cyanidin-3-galactoside

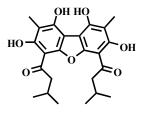


2b: pelargonidin-3,5-biglucoside

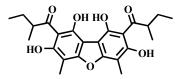


2d: delphinidin-3-galactoside

# 3. Dibenzofurans

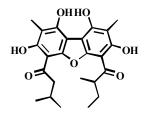


3a: rhodomyrtoxin

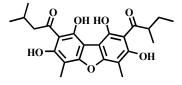


3c: rhodomyrtoxin B

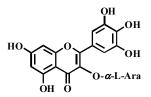
# 4. Flavonoids



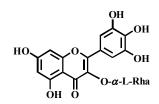
3b: *ψ*-rhodomyrtoxin



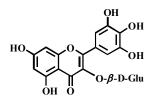
3d: rhodomyrtoxin C



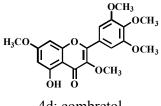
4a: myricetin 3-O- $\alpha$ -L-furanoarabinoside



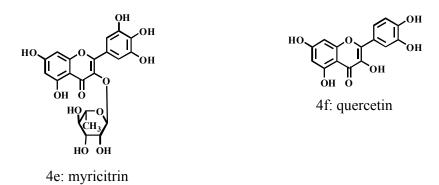
4c: myricetin 3-O- $\alpha$ -L-rhamnoside



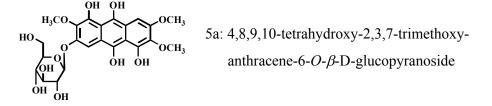
4b: myricetin 3-O- $\beta$ -D-glucoside

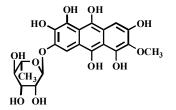


4d: combretol



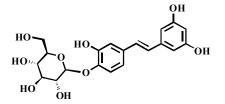
# 5. Naphthalenoids





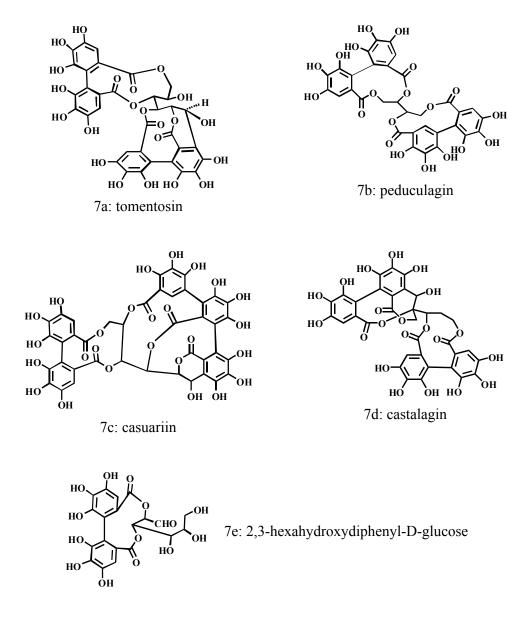
5b: 2,4,7,8,9,10-hexahydroxy-3-methoxyanthracene- $6-O-\alpha$ -L-rhamnopyranoside

6. Stilbenoids

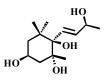


6: piceatannol 4-O- $\beta$ -D-glucoside

# 7. Tannins



- 8. Terpenoids
  - a) Nor-terpenes



8a: (3S,5R,6R,7E,9S)-megastiman-7-ene-3,5,6,9-tetrol

## b) Monoterpenes



8b:  $\alpha$ -pinene





8c: *β*-pinene

8d: limonene

# c) Sesquiterpenes

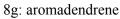


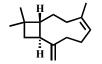
8e: globulol



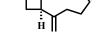
8f: viridiflorol





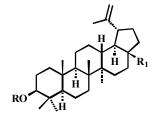


8h:  $\beta$ -caryophyllene

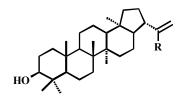


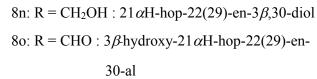
8i: caryophyllene oxide

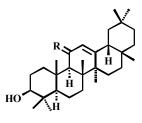
d) Triterpenes



8k: R = H;  $R_1 = CH_3$ : lupeol 8l: R = H;  $R_1 = CH_2OH$ : betulin 8m: R = Ac;  $R_1 = CH_2OH$ : betulin-3-acetate

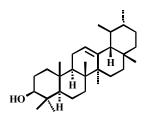




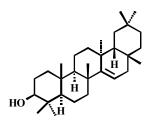


8p:  $R = H_2$ :  $\beta$ -amyrin

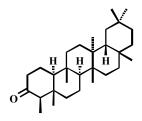
8q: R = O:  $\beta$ -amyrenonol



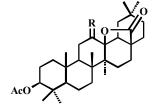
8s:  $\alpha$ -amyrin



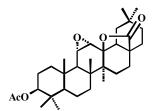
8r: taraxerol







8u: R = O :  $3\beta$ -acetoxy-12-oxo-oleanan-28,13 $\beta$ -olide 8v: R =  $\alpha$ -OH,  $\beta$ -H :  $3\beta$ -acetoxy-12 $\alpha$ -hydroxyoleanan-28,13 $\beta$ -olide



8w:  $3\beta$ -acetoxy- $11\alpha$ ,  $12\alpha$ -epoxyoleanan-28,  $13\beta$ -olide

#### 1.3.2 Biological activities of some Rhodomyrtus genus

Based on the SciFinder scholar database, only three reports have investigated and identified the active components from the *Rhodomyrtus* genus. The methanolic extract of *R. tomentosa* leaves showed significant antimicrobial activity to inhibit *Escherichia coli* and *Staphylococcus aureus*. Bioassay-guided chromatography of this fraction revealed the isolation of active compound, rhodomyrtone (Dachriyanus *et al.*, 2002). Piceatannol 4'-*O*- $\beta$ -D-glucopyranoside was purified from aq. ethanol extract of dried fruit of *R. tomentosa* showed superoxide-scavenging activity at IC<sub>50</sub> of 81.7 µg/mL and it also stimulated the proliferation of cultured human skin fibroblasts and normal human epidermal keratinocytes (Nojima *et al.*, 2007). The anthracene glycosides named 4,8,9,10-tetrahydroxy-2,3,7-trimethoxyanthracene-6-*O*- $\beta$ -D-gluco-pyranoside and 2,4,7,8,9,10-hexahydroxy-3-methoxyanthracene-6-*O*- $\alpha$ -L-rhamno-pyranoside, isolated from the aerial parts of *R. tomentosa* significantly increased the alkaline phosphatase activity, collagen synthesis, and mineralization of the nodules of MC3T3-E1 osteoblastic cells (Tung *et al.*, 2009).

#### 1.4 Rhodomyrtus tomentosa (Aiton) Hassk.

Rhodomyrtus tomentosa (Aiton) Hassk. is only one species of the genus Rhodomyrtus in Thailand, and locally named as "Toh (lne) or Pruad (m500)" (Smitinand, 2001). *R. tomentosa* has been used in the traditional medicine for a long time. The ripe fruits are eaten raw to treat diarrhea (Ong and Nordiana, 1999). The liquid that is present in this plant can be used to treat gynaecopathy including morbid leucorrhoea, menoxenia, dysmenorrhoea, endometritis, appendagitis, and pelvic inflammation (Wei, 2006). Recent studies demonstrated that its ethanolic extract showed good activity against several Gram-positive bacteria (Voravuthikunchai *et al.*, 2007). *R. tomentsa* has also been used as traditional medicine for diarrhea and wound treatments in Vietnam (Tung *et al.*, 2009).

#### 1.4.1 Description of *R. tomentosa*

*R. tomentosa* (Aiton) Hassk. (**Figure 1**) is a flowering plant in the family Myrtaceae, native to southern and southeastern Asia, from India, east to southern China, Taiwan and the Philippines, and south to Malaysia and Sulawesi including Thailand. It is an evergreen shrub growing up to 2 meters in height. The leaves are opposite, leathery with the size of 5-7 cm long and 2-3.5 cm broad. It has three-veined from base with an oval and obtuse to sharp pointed at the tip. Its leaves also have glossy green above, densely grey or rarely yellowish-hairy beneath, with a wide petiole about 4-7 mm and an entire margin. The flowers are solitary or in clusters of two or three, 2-4 cm in diameter, with five petals which are tinged white outside with purplish-pink or all pink. The fruit is edible, round with 10-15 mm long and has a

purplish black when mature. It is contained three or four-celled, capped with persistent calyx lobes, soft. Its fruit has many seeds, around 40-45 seeds, in a double row in each cell. This plant has flowers around April-May and further can be propagated by seeds. The seed dispersal is spread by humans who use this plant in landscaping and by fruit eating birds and mammals (Jie and Craven, 2007).

*R. tomentosa* grows in coasts, natural forest, riparian zones, wetlands, moist and wet forests, bog margins, from sea level up to 2,400 meters elevation. It also grows in a wide range of soil types, including salty coastal soil, but is sensitive to heavy salt spray (APIRS, 2001). Furthermore *R. tomentosa* has become an invasive species in some countries, spreading to form large, monospecific thickets that displace native flora and fauna through overcrowding and competition. Areas especially affected include Florida, Hawaii and French Polynesia. It is able to invade a range of habitats, from pine flatwoods to mangrove marshes (Winotai *et al.*, 2005).

#### 1.4.2 Nomenclature, Synonyms and Common names

The Scientific name, *Rhodomyrtus tomentosa*, is derived from the Greek "*rhodon*" meaning red and "*myrtos*" meaning myrtle which referred to rose-colored flowers that are common in plants of this genus. It has several synonyms including *Myrtus tomentosa* Aiton, *Myrtus canescens* Loureiro, and *Rhodomyrtus parviflora* Alston. This plant has many common names such as downy rose myrtle (English-Florida), downy myrtle (English-Florida), rose myrtle (English-Florida), hill gooseberry (English), Ceylon hill berry (English), hill guava (English), feijoa (French), isenberg bush (English-Hawaii), myrte-groseille (French) (Wagner *et al.*, 1999). In Thailand, it was commonly known as "Toh (โทะ)", "Pruad (พรวดใหญ่)" *etc* (Smitinand, 2001).



Figure 1 Rhodomyrtus tomentosa (Aiton) Hassk.

# 1.5 The objectives

The objectives of this research are to investigate the chemical constituents from the leaves, stems and fruits of *Rhodomyrtus tomentosa* and to evaluate the antibacterial activity of the crude extracts and the isolated compounds.

# CHAPTER 2

# EXPERIMENTAL

## 2.1 General methods

Melting points (<sup>O</sup>C) were determined on a Fisher-Johns melting point apparatus and were uncorrected. Optical rotations were measured in chloroform solution on a JASCO P-1020 polarimeter. UV spectra were recorded by a SPECORD S100 spectrophotometer (Analytikjena). The IR spectra were measured with a FTS 165 FT-IR Perkins-Elmer spectrophotometer. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were obtained from FT-NMR Bruker Avance Ultra Shield<sup>TM</sup> 300 and 500 MHz spectrometers at the Department of Chemistry, Faculty of Science, Prince of Songkla University. The spectra were recorded as  $\delta$  value in ppm downfield from TMS  $\delta$  0.00). The EI-MS, HREI-MS, FAB-MS and HRFAB-MS (internal standard mass spectra were obtained from a MAT 95 XL mass spectrometer (Thermofinigan) at the Scientific Equipment Center, Prince of Songkla University. Column chromatography (CC) was performed on silica gel 100 (70-230 Mesh, Merck) or Sephadex<sup>TM</sup> LH-20 (GE Healthcare Bio-Sciences AB) while quick column chromatography (QCC) was performed on silica gel 60H (Merck). Thin-layer chromatography (TLC), aluminum sheets of silica gel 60 F254 (20x20 cm, layer thickness 0.2 mm, Merck), and preparative TLC (silica gel 60 F<sub>254</sub>, 20x20 cm, layer thickness 0.25 mm, Merck) were used for analytical purposes. All solvents for extraction and chromatography were distilled at their boiling points prior to use.

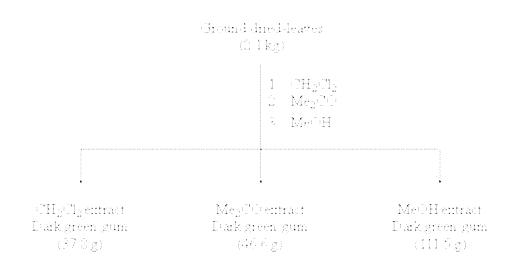
#### 2.2 Plant material

The leaves and stems of *Rhodomyrtus tomentosa* were collected in February 2007 from Singha Nakorn District, Songkhla Province. The fruits were collected in March 2008 from Khuan Khanun District, Phatthalung Province in the southern part of Thailand. The voucher specimen (A. Hiranrat 001) was identified by J. Wai and has been deposited in the herbarium of the Department of Biology, Faculty of Science, Prince of Songkla University, Thailand.

# 2.3 Chemical investigation of the leaves

## 2.3.1 Extraction and isolation

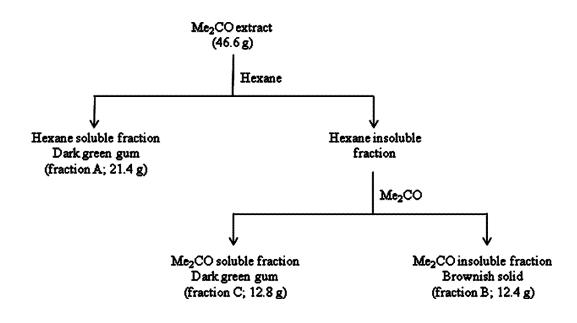
The dried ground leaves of *R. tomentosa* (2.1 kg) were successively extracted at room temperature with dichloromethane, acetone and methanol (twice for each extract time of 3 days) to give dark green gums of the dichloromethane (37.8 g), acetone (46.6 g) and methanol (111.5 g) extracts, respectively. The extract preparations were shown in **Scheme 1**.



Scheme 1 The extract preparations from the leaves of *R. tomentosa* 

#### 2.3.2 Purification of the Me<sub>2</sub>CO extract from the leaves

The Me<sub>2</sub>CO extract (46.6 g) was fractionated by dissolving in hexane to give soluble- (fraction A; 21.4 g) and insoluble (25.2 g) fractions as a dark green gum and a dark green solid, respectively. The insoluble fraction was further dissolved in Me<sub>2</sub>CO to afford the soluble fraction (fraction C; 12.8 g) as a dark green gum, and insoluble fraction (fraction B; 12.4 g) as a brownish solid (**Scheme 2**).



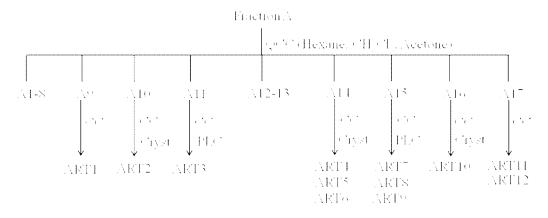
Scheme 2 Fractions obtained from the Me<sub>2</sub>CO extract utilizing its solubility

#### 2.3.2.1 Separation of fraction A

Fraction A (20.3 g) was subjected to QCC and eluted with a gradient of hexane-CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>-Me<sub>2</sub>CO and Me<sub>2</sub>CO. The eluted fractions were combined into fractions A1 to A17 on the basis of their TLC characteristics. Physical appearance and weight of each fraction are summarized in **Table 2**. After separation and purification, twelve compounds were obtained (**Scheme 3**).

Fraction	Eluent	Weight (mg)	Physical Appearance
A1	10% CH <sub>2</sub> Cl <sub>2</sub> /hexane	49.3	white powder
A2	10% CH <sub>2</sub> Cl <sub>2</sub> /hexane	199.3	white powder
A3	10-15% CH <sub>2</sub> Cl <sub>2</sub> /hexane	73.7	yellow gum
A4	15-20% CH <sub>2</sub> Cl <sub>2</sub> /hexane	60.2	yellow gum
A5	20% CH <sub>2</sub> Cl <sub>2</sub> /hexane	272.7	yellow gum
A6	20-35% CH <sub>2</sub> Cl <sub>2</sub> /hexane	153.5	yellow gum
A7	35% CH <sub>2</sub> Cl <sub>2</sub> /hexane	43.3	orange gum
A8	35-50% CH <sub>2</sub> Cl <sub>2</sub> /hexane	165.4	orange-brown gum
A9	50% CH <sub>2</sub> Cl <sub>2</sub> /hexane	177.6	brown gum
A10	50-70% CH <sub>2</sub> Cl <sub>2</sub> /hexane	1.3 g	brown gum
A11	70-85% CH <sub>2</sub> Cl <sub>2</sub> /hexane	315.4	dark green gum
A12	85% CH <sub>2</sub> Cl <sub>2</sub> /hexane	198.5	dark green gum
A13	CH <sub>2</sub> Cl <sub>2</sub>	185.4	dark green gum
A14	CH <sub>2</sub> Cl <sub>2</sub>	606.0	dark green gum
A15	5% Me <sub>2</sub> CO/CH <sub>2</sub> Cl <sub>2</sub>	2.9 g	dark green gum
A16	5-10% Me <sub>2</sub> CO/CH <sub>2</sub> Cl <sub>2</sub>	1.6 g	dark green gum
A17	10% Me <sub>2</sub> CO/CH <sub>2</sub> Cl <sub>2</sub> -Me <sub>2</sub> CO	10.0 g	dark green gum

Table 2 Physical appearance and weight of fractions obtained from QCC of fraction A



Scheme 3 Separation and purification of ART1-ART12

#### **Isolation of ART1**

Fraction A9 (177.6 mg) was subjected to CC and eluted with 15% CH<sub>2</sub>Cl<sub>2</sub> in hexane to give three fractions (A9.1-A9.3). Fraction A9.2 (122.7 mg) was further purified by CC using 15% CH<sub>2</sub>Cl<sub>2</sub> in hexane as an eluent to produce a yellow gum of **ART1** (73.5 mg).

# ART1

UV (CHCl<sub>3</sub>)  $\lambda_{max}$  nm (log $\varepsilon$ ) 246 (4.31), 291 (4.16), 383 (3.23) IR (CHCl<sub>3</sub>)  $\nu_{max}$  (cm<sup>-1</sup>) 3349, 2925, 2863, 1460, 1377, 1158, 1095 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>) See **Table 41** 

## **Isolation of ART2**

Fraction A10 (1.3 g) was subjected to CC and eluted with a gradient of 10%  $CH_2Cl_2$  in hexane to 5% Me<sub>2</sub>CO in  $CH_2Cl_2$  to give fractions A10.1-A10.8. Separation of fraction A10.6 (105.5 mg) on silica gel CC and eluted with 50%  $CH_2Cl_2$  in hexane gave four fractions (A10.61-A10.64). Crystallization of fraction A10.64 (42.5 mg), upon standing at room temperature, gave the white needles of **ART2** (14.1 mg).

## ART2

mp.	125-126 °C
$[\alpha]_{\mathrm{D}}^{29}$	-1.1 ( <i>c</i> 0.80, CHCl <sub>3</sub> )
HREI-MS $m/z$	456.2133 (calcd for $C_{26}H_{32}O_7$ 456.2148)
EI-MS <i>m</i> / <i>z</i> (% rel. int.)	456 (M <sup>+</sup> , 61), 441 (33), 414 (20), 399 (100), 372 (17),
	247 (40)
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	270 (4.39), 327 (3.42)
IR (CHCl <sub>3</sub> ) $v_{max}$ (cm <sup>-1</sup> )	3126, 2969, 2935, 2874, 1720, 1650, 1617, 1502, 1452,
	1303, 1180, 1052
$^{1}$ H (300 MHz) and $^{13}$ C NMR	$(75 \text{ MHz}) (\text{CDCl}_3)$ See Table 12

#### **Isolation of ART3**

Fraction A11 (315.4 mg) was subjected to CC and eluted with gradient solvents from 20% CH<sub>2</sub>Cl<sub>2</sub> in hexane to 40% CH<sub>2</sub>Cl<sub>2</sub> in hexane to give six fractions (A11.1-A11.6). Fraction A11.6 (166.9 mg) was further separated by silica gel CC and eluted with 3% Me<sub>2</sub>CO in hexane affording nine fractions (A11.61-A11.69). Fraction A11.63 (46.7 mg) was subjected to CC on silica gel and eluted with 3% Me<sub>2</sub>CO in hexane to give fractions A11.63A-A11.63E. Fraction A11.63D (3.2 mg) was then purified by preparative TLC using 1% Me<sub>2</sub>CO in hexane as a mobile phase to produce a white powder of **ART3** (1.0 mg).

#### ART3

$\left[\alpha\right]_{\mathrm{D}}^{29}$	-4.2 ( <i>c</i> 0.05, CHCl <sub>3</sub> )
HREI-MS $m/z$	456.2133 (calcd for C <sub>26</sub> H <sub>32</sub> O <sub>7</sub> 456.2148)
EI-MS $m/z$ (% rel. int.)	456 (M <sup>+</sup> , 50), 441 (32), 414 (20), 399 (100), 372 (15),
	247 (36)
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	244 (4.33), 271 (4.62), 325 (3.84)
IR (CHCl <sub>3</sub> ) $v_{\text{max}}$ (cm <sup>-1</sup> )	3316, 2963, 2930, 2891, 1720, 1636, 1627, 1594,
	1471, 1438

 $^{1}$ H (500 MHz) and  $^{13}$ C NMR (125 MHz) (CDCl<sub>3</sub>) See **Table 13** 

#### **Isolation of ART4, ART5 and ART6**

Fraction A14 (606.0 mg) was subjected to CC and eluted with  $CH_2Cl_2$  and  $CH_2Cl_2$ -MeOH gradient to give seven fractions (A14.1-A14.7). Fraction A14.2 (194.8 mg) was rechromatographed on CC using 3% Me<sub>2</sub>CO in hexane as an eluent to produce **ART4** (103.9 mg). Fraction A14.3 (183.6 mg) was further fractionated by silica gel CC and eluted with  $CH_2Cl_2$  to give fractions A14.31-A14.36. Compound **ART5** (34.0 mg) was crystallized from fraction A14.33 (138.1 mg) as white needles. The filtrate was further separated by CC eluting with 3% Me<sub>2</sub>CO in hexane to give seven fractions (A14.33-61-A14.33-67). White needles of **ART6** (23.0 mg) were obtained from crystallization of fraction A14.33-66 (31.4 mg) in Me<sub>2</sub>CO-hexane (1:5).

mp.	80-81 <sup>o</sup> C
$[\alpha]_{\mathrm{D}}^{29}$	-23.5 ( <i>c</i> 0.39, CHCl <sub>3</sub> )
HREI-MS $m/z$	674.3853 (calcd for C <sub>41</sub> H <sub>54</sub> O <sub>8</sub> 674.3819)
EI-MS <i>m</i> / <i>z</i> (% rel. int.)	674 (M <sup>+</sup> , 1), 617 (100), 547 (7), 419 (8)
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	263 (4.15), 306 (4.19), 348 (3.60)
IR (CHCl <sub>3</sub> ) $v_{max}$ (cm <sup>-1</sup> )	3423, 2952, 2868, 1717, 1659, 1617, 1594, 1466, 1382,
	1362, 1183, 1158, 1038
$^{1}$ H (300 MHz) and $^{13}$ C NMR	$(75 \text{ MHz}) (\text{CDCl}_3)$ See Table 14

# ART5

IR (CHCl<sub>3</sub>)  $v_{max}$  (cm<sup>-1</sup>) 3410, 2945, 2932, 1453, 1050 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>) See **Table 38** 

# ART6

mp.	188-189 <sup>o</sup> C
$\left[\alpha\right]_{\mathrm{D}}^{29}$	-9.4 ( <i>c</i> 1.0, CHCl <sub>3</sub> )
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	244 (3.95), 262 (4.01), 302 (4.14)
IR (CHCl <sub>3</sub> ) $v_{max}$ (cm <sup>-1</sup> )	3244, 2959, 2935, 1720, 1630, 1594, 1385, 1167, 1092
<sup>1</sup> H (300 MHz) and <sup>13</sup> C NMR	$(75 \text{ MHz}) (\text{CDCl}_3)$ See <b>Table 11</b>

## **Isolation of ART7, ART8 and ART9**

Fraction A15 (2.9 g) was subjected to CC eluting with CH<sub>2</sub>Cl<sub>2</sub>-MeOH gradient systems to give fractions A15.1-A15.5. Fraction A15.3 (1.1 g) was fractionated by CC eluting with 10% Me<sub>2</sub>CO in hexane to give fractions A15.3-51 to A15.3-59. Fraction A15.3-54 (66.2 mg) was subjected to CC and eluted with 5% Me<sub>2</sub>CO in hexane to give five fractions (A15.3-541-A15.3-545). A yellow solid of **ART7** (4.2 mg) was obtained from crystallization of fraction A15.3-545 (36.9 mg) in hexane-Me<sub>2</sub>CO (5:1). The filtrate was further purified by CC eluting with 5% Me<sub>2</sub>CO in hexane to produce **ART8** (0.7 mg) and **ART9** (1.3 mg) as yellowish gum.

mp.	142-144 <sup>o</sup> C
UV (MeOH) $\lambda_{max}$ nm (log $\varepsilon$ )	210 (4.15), 265 (4.19), 342 (3.60)
IR (CHCl <sub>3</sub> ) $v_{\text{max}}$ (cm <sup>-1</sup> )	3421, 1663, 1659, 1505, 1460, 1173, 1142
<sup>1</sup> H (300 MHz) and <sup>13</sup> C NMR	$(75 \text{ MHz}) (\text{CDCl}_3)$ See Table 22

# ART8

$[\alpha]_{\mathrm{D}}^{29}$	0 ( <i>c</i> 0.035, CHCl <sub>3</sub> )
EI-MS <i>m</i> / <i>z</i> (% rel. int.)	268 (M <sup>+</sup> , 15), 236 (86), 221 (32), 193 (16), 166 (100),
	140 (30), 123 (55), 96 (25), 70 (47), 69 (42)
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	243 (3.40), 294 (2.91)
IR (CHCl <sub>3</sub> ) $v_{\text{max}}$ (cm <sup>-1</sup> )	3479, 1714, 1693, 1639
$^{1}$ H (500 MHz) and $^{13}$ C NMR	(125 MHz) (CDCl <sub>3</sub> ) See <b>Table 15</b>

# ART9

$\left[\alpha\right]_{\mathrm{D}}^{29}$	-182.0 ( <i>c</i> 0.06, CHCl <sub>3</sub> )
HREI-MS $m/z$	442.2352 (calcd for C <sub>26</sub> H <sub>34</sub> O <sub>6</sub> 442.2355)
EI-MS <i>m</i> / <i>z</i> (% rel. int.)	442 (M <sup>+</sup> , 1), 428 (13), 413 (14), 386 (22), 385 (91), 330
	(8), 315 (28), 236 (46), 221 (24), 167 (38), 166 (63),
	149 (77), 123 (60), 97 (49), 70 (100), 69 (90)
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	292 (3.82), 333 (3.20)
IR (CHCl <sub>3</sub> ) $v_{\text{max}}$ (cm <sup>-1</sup> )	3372, 2975, 2952, 2868, 1717, 1653, 1622, 1468, 1385,
	1256, 1158, 1038

 $^{1}$ H (500 MHz) and  $^{13}$ C NMR (125 MHz) (CDCl<sub>3</sub>) See **Table 16** 

# **Isolation of ART10**

Fraction A16 (1.6 g) was fractionated by CC eluting with hexane-Me<sub>2</sub>CO gradient systems to give six fractions (A16.1-A16.6). Crystallization of fraction A16.5 (304.5 mg) in hexane-Me<sub>2</sub>CO (5:1) gave a yellow solid of **ART10** (15.7 mg).

UV (EtOH) $\lambda_{max}$ nm (log $\varepsilon$ )	247 (4.10), 372 (3.79)	
IR (CHCl <sub>3</sub> ) $v_{max}$ (cm <sup>-1</sup> )	3410, 1656, 1641, 1026, 995	
$^{1}$ H (300 MHz) and $^{13}$ C NMR	(125 MHz) (CDCl <sub>3</sub> +DMSO- <i>d</i> <sub>6</sub> )	See Table 26

#### **Isolation of ART11 and ART12**

Fraction A17 (10.0 g) was fractionated by CC and eluted with hexane-Me<sub>2</sub>CO in a polarity gradient to give fractions A17.1 to A17.8. Fraction A17.4 (126.1 mg) was subjected to CC and eluted with 5% Me<sub>2</sub>CO in hexane to produce **ART11** (9.9 mg) as a colorless crystal. Fraction A17.6 (210.6 mg) was subjected to CC and eluted with 10% Me<sub>2</sub>CO in hexane to produce a yellowish gum of **ART12** (5.5 mg).

# ART11

mp	160-162 <sup>o</sup> C			
HREI-MS $m/z$	428.2214 (calcd for C <sub>25</sub> H <sub>32</sub> O <sub>6</sub> 428.2199)			
EI-MS $m/z$ (% rel. int.)	428 (M <sup>+</sup> , 84), 413 (23), 385 (28), 358 (20), 330 (65),			
	315 (100), 288 (51), 287 (44), 273 (32), 260 (25), 245			
	(22), 232 (28), 217 (32), 189 (27), 96 (26), 91 (31), 69			
	(76)			
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	246 (4.08), 298 (3.24)			
IR (CHCl <sub>3</sub> ) $v_{\text{max}}$ (cm <sup>-1</sup> )	2974, 2935, 2879, 1717, 1675, 1656, 1468, 1398, 1387			
$^{1}$ H (300 MHz) and $^{13}$ C NMR (75 MHz) (CDCl <sub>3</sub> ) See <b>Table 17</b>				

# ART12

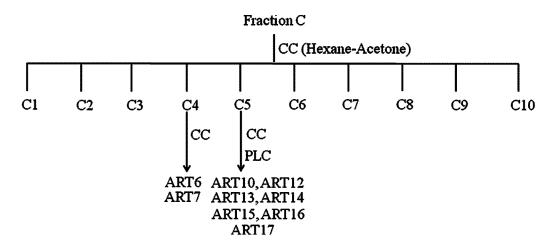
 $[\alpha]_D^{29}$ +193.5 (c 0.60, CHCl\_3)UV (CHCl\_3)  $\lambda_{max}$  nm (log $\varepsilon$ )246 (4.13)IR (CHCl\_3)  $v_{max}$  (cm<sup>-1</sup>)3406, 2971, 2940, 2875, 1656, 1460, 1375, 1251<sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl\_3)See **Table 31** 

#### 2.3.2.2 Separation of fraction C

Fraction C (11.5 g) was subjected to CC and eluted with gradient solvents from 5% Me<sub>2</sub>CO in hexane to Me<sub>2</sub>CO. The eluted fractions were combined into ten fractions (C1-C10) on the basis of their TLC characteristics. Physical appearance and weight of each fraction are summarized in **Table 3**. After separation and purification, nine compounds were obtained (**Scheme 4**).

Fraction	Eluent	Weight (mg)	Physical Appearance
C1	5-7% Me <sub>2</sub> CO/hexane	114.7	yellow gum
C2	7% Me <sub>2</sub> CO/hexane	118.0	dark green gum
C3	7% Me <sub>2</sub> CO/hexane	50.5	dark green gum
C4	7-12% Me <sub>2</sub> CO/hexane	76.7	dark green gum
C5	12-17% Me <sub>2</sub> CO/hexane	968.5	dark green gum
C6	17-25% Me <sub>2</sub> CO/hexane	340.6	dark green gum
C7	25-30% Me <sub>2</sub> CO/hexane	179.7	dark green gum
C8	30% Me <sub>2</sub> CO/hexane	846.6	dark green gum
С9	50% Me <sub>2</sub> CO/hexane	729.6	dark green gum
C10	Me <sub>2</sub> CO	7.2 g	dark green gum

 Table 3
 Physical appearance and weight of fractions obtained from QCC of fraction C



Scheme 4 Separation and purification of ART6, ART7, ART10 and ART12-ART17

#### **Isolation of ART6 and ART7**

**ART7** (15.1 mg) was obtained as a yellow solid from fraction C4 (76.7 mg) by crystallization in hexane-Me<sub>2</sub>CO (5:1). The filtrate was further separated by CC eluting with 3% Me<sub>2</sub>CO in hexane to give six fractions (C4.1-C4.6). Fraction C4.2 (33.9 mg) was further purified by CC using 3% Me<sub>2</sub>CO in hexane as an eluent to yield yellow needles of **ART6** (18.3 mg).

# Isolation of ART10, ART12, ART13, ART14, ART15, ART16, and ART17

Fraction C5 (968.5 mg) was chromatographed on CC and eluted with gradient solvents from 8% Me<sub>2</sub>CO in hexane to Me<sub>2</sub>CO to give eight fractions (C5.1-C5.8). Fraction C5.3 (89.3 mg) was further separated by CC eluting with 10% Me<sub>2</sub>CO in hexane to give fractions C5.3.1-C5.3.7. Fraction C5.3.4 (34.6 mg) was then applied to TLC plate using 1% MeOH in CH<sub>2</sub>Cl<sub>2</sub> as a mobile phase to yield a yellow gum of **ART13** (4.0 mg), a colorless gum of **ART14** (1.3 mg), a colorless gum of **ART15** (1.0 mg) and a yellowish gum of **ART12** (9.6 mg). Fraction C5.4 (25.0 mg) was also purified by TLC plate using 1% MeOH in CH<sub>2</sub>Cl<sub>2</sub> as a mobile phase to give compounds **ART16** (1.6 mg), **ART10** (1.1 mg) and **ART17** (0.9 mg) as yellow gums.

# ART13

UV (MeOH)  $\lambda_{max}$  nm (log $\varepsilon$ ) 248sh (3.23), 270 (3.38), 305sh (3.26), 345 (3.32) IR (CHCl<sub>3</sub>)  $\nu_{max}$  (cm<sup>-1</sup>) 3375, 2925, 2847, 1659, 1597, 1499, 1364, 1215, 1163, 1106 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>) See **Table 23** 

## ART14

UV (CHCl<sub>3</sub>)  $\lambda_{max}$  nm (log $\varepsilon$ ) 247 (3.72), 255*sh* (3.67) IR (CHCl<sub>3</sub>)  $\nu_{max}$  (cm<sup>-1</sup>) 3449, 2948, 2923, 1718, 1545 <sup>1</sup>H (500 MHz) and <sup>13</sup>C NMR (125 MHz) (CDCl<sub>3</sub>) See **Table 32** 

IR (CHCl<sub>3</sub>)  $v_{max}$  (cm<sup>-1</sup>) 3340, 2976, 2935, 2894, 1750, 1037 <sup>1</sup>H (500 MHz) and <sup>13</sup>C NMR (125 MHz) (CDCl<sub>3</sub>) See **Table 48** 

## ART16

UV (CHCl<sub>3</sub>)  $\lambda_{max}$  nm (log $\varepsilon$ ) 252 (3.38), 266 (3.32), 302*sh* (3.12), 355 (3.43) IR (CHCl<sub>3</sub>)  $\nu_{max}$  (cm<sup>-1</sup>) 3375, 2920, 2847, 1656, 1597, 1496, 1460, 1217, 1119 <sup>1</sup>H (500 MHz) and <sup>13</sup>C NMR (125 MHz) (CDCl<sub>3</sub>) See **Table 24** 

#### **ART17**

UV (CHCl<sub>3</sub>)  $\lambda_{max}$  nm (log $\varepsilon$ ) 252 (4.06), 268 (4.13), 308*sh* (4.05), 348 (4.11) IR (CHCl<sub>3</sub>)  $v_{max}$  (cm<sup>-1</sup>) 3255, 2920, 2852, 1656, 1638, 1579, 1460, 1160, 1126 <sup>1</sup>H (500 MHz) and <sup>13</sup>C NMR (125 MHz) (CDCl<sub>3</sub>) See **Table 25** 

# 2.3.3 Purification of the CH<sub>2</sub>Cl<sub>2</sub> extract from the leaves

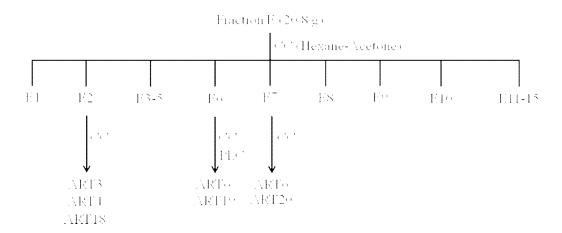
The  $CH_2Cl_2$  extract (37.8 g) was further fractionated by dissolving in MeOH to afford soluble- (fraction E; 27.9 g) and insoluble (fraction D; 9.9 g) fractions.

## 2.3.3.1 Separation of fraction E

Fraction E (20.8 g) was separated by CC and eluted with hexane-Me<sub>2</sub>CO gradient solvent systems. The eluted fractions were combined into fifteen fractions (E1-E15) on the basis of their TLC characteristics. Physical appearance and weight of each fraction are summarized in **Table 4**. After separation and purification, three compounds were additionally isolated (**Scheme 5**).

Fraction	Eluent	Weight (mg)	Physical Appearance
E1	2-5% Me <sub>2</sub> CO/hexane	1.85 g	orange gum
E2	5-10% Me <sub>2</sub> CO/hexane	1.33 g	orange gum
E3	5-10% Me <sub>2</sub> CO/hexane	382.1	yellow gum
E4	10% Me <sub>2</sub> CO/hexane	739.3	dark green gum
E5	15% Me <sub>2</sub> CO/hexane	348.0	dark green gum
E6	15-20% Me <sub>2</sub> CO/hexane	322.7	dark green gum
E7	20% Me <sub>2</sub> CO/hexane	1.42 g	dark green gum
E8	20% Me <sub>2</sub> CO/hexane	4.31 g	dark green gum
E9	30% Me <sub>2</sub> CO/hexane	858.9	dark green gum
E10	30-40% Me <sub>2</sub> CO/hexane	447.0	dark green gum
E11	40% Me <sub>2</sub> CO/hexane	349.1	dark green gum
E12	40-50% Me <sub>2</sub> CO/hexane	783.5	dark green gum
E13	50-60% Me <sub>2</sub> CO/hexane	163.4	dark green gum
E14	60-80% Me <sub>2</sub> CO/hexane	1.83 g	dark green gum
E15	Me <sub>2</sub> CO	1.40 g	dark green gum

**Table 4** Physical appearance and weight of fractions obtained from CC of fraction E



Scheme 5 Separation and purification of ART2, ART4, ART6 and ART18-ART20

#### **Isolation of ART3, ART4 and ART18**

Fraction E2 (1.33 g) was subjected to CC eluting with 2% Me<sub>2</sub>CO in hexane to give nine fractions (E2.1-E2.9). Fraction E2.5 (888.9 mg) was further rechromatographed on CC and eluted with 2% Me<sub>2</sub>CO in hexane to afford a white solid of **ART3** (54.0 mg) and a yellowish solid of **ART4** (416.4 mg). Fraction E2.7 (60.7 mg) was further purified by CC using 2% Me<sub>2</sub>CO in hexane as an eluent to produce a yellowish gum of **ART18** (22.2 mg).

#### **ART18**

$[\alpha]_{\mathrm{D}}^{25}$	-9.3 <sup>°</sup> ( <i>c</i> 0.92, CHCl <sub>3</sub> )
HREI-MS $m/z$	$688.3610$ (calcd for $C_{41}H_{52}O_9$ $688.3611$ )
EI-MS $m/z$ (% rel. int.)	688 (M <sup>+</sup> , 2), 632 (51), 630 (100), 617 (14), 561 (5), 477
	(5), 385 (3), 247 (7), 177 (4)
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	246 (4.28), 260 (4.36), 304 (4.26), 347sh (3.68)
IR (CHCl <sub>3</sub> ) $v_{max}$ (cm <sup>-1</sup> )	3135, 2975, 2945, 1720, 1656, 1617, 1500, 1452, 1303,
	1190

 $^{1}$ H (300 MHz) and  $^{13}$ C NMR (75 MHz) (CDCl<sub>3</sub>) See Table 18

### **Isolation of ART6 and ART19**

Fraction E6 (322.7 mg) was separated by CC and eluted with gradient solvents of 5% Me<sub>2</sub>CO in hexane to 10% Me<sub>2</sub>CO in hexane to afford ten fractions (E6.1-E6.10). Fraction E6.7 (65.1 mg) was then subjected to CC using 60% CH<sub>2</sub>Cl<sub>2</sub> in hexane to give fractions E6.71-E6.76. Fraction E6.71 (5.4 mg) was purified by TLC plate using 40% CH<sub>2</sub>Cl<sub>2</sub> in hexane as an eluent to produce a yellowish gum of **ART19** (1.5 mg). Compound **ART6** (8.2 mg) was obtained from fraction E6.74 (19.5 mg) as a white solid.

### ART19

$\left[\alpha\right]_{\mathrm{D}}^{29}$	+62.0 <sup>°</sup> ( <i>c</i> 0.10, CHCl <sub>3</sub> )		
HRFAB-MS $m/z$	443.2429 (calcd for C <sub>26</sub> H <sub>35</sub> O <sub>6</sub> 443.2434)		
FAB-MS $m/z$ (% rel. int.)	443 ([M+H] <sup>+</sup> , 28%), 385 (17), 185 (20), 133 (100), 93		
	(56)		
EI-MS <i>m</i> / <i>z</i> (% rel. int.)	385 ([M-C <sub>4</sub> H <sub>9</sub> ] <sup>+</sup> , 100), 367 (5), 315 (8), 297 (7)		
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	245 (4.09), 262 (4.14), 300 (4.24)		
IR (CHCl <sub>3</sub> ) $v_{max}$ (cm <sup>-1</sup> )	3365, 2970, 2955, 1717, 1655, 1622, 1465, 1382, 1249		
$^{1}\mathrm{H}$ (500 MHz) and $^{13}\mathrm{C}$ NMR	(125 MHz) (CDCl <sub>3</sub> ) See <b>Table 19</b>		

### **Isolation of ART6 and ART20**

Fraction E7 (1.42 g) was subjected to CC and eluted with gradient solvents of 5% Me<sub>2</sub>CO in hexane to 10% Me<sub>2</sub>CO in hexane to give ten fractions (E7.1-E7.10). A yellowish solid of **ART20** (15.4 mg) was obtained from crystallization of fraction E7.2 (55.2 mg) in hexane:Me<sub>2</sub>CO (5:1). Crystallization of fraction E7.5 (264.6 mg) from hexane:Me<sub>2</sub>CO (5:1) gave a white solid of **ART6** (206.9 mg).

# ART20

mp.	198-199 <sup>o</sup> C
$[\alpha]_{\mathrm{D}}^{29}$	-10.7° ( <i>c</i> 0.68, CHCl <sub>3</sub> )
HREI-MS $m/z$	$688.3610 [\text{M}]^+$ (calcd for $C_{41}H_{52}O_{9}$ , $688.3611$ )
EI-MS <i>m</i> / <i>z</i> (% rel. int.)	688 [M] <sup>+</sup> , 632, 630, 617, 561, 477, 385, 247, 177
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	246 (4.23), 262 (4.28), 306 (4.21), 350sh (3.72)
IR (CHCl <sub>3</sub> ) $v_{\text{max}}$ (cm <sup>-1</sup> )	3128, 2969, 2935, 1718, 1650, 1617, 1502, 1452, 1303
$^{1}$ H (300 MHz) and $^{13}$ C NMR	$(75 \text{ MHz}) (\text{CDCl}_3)$ See <b>Table 20</b>

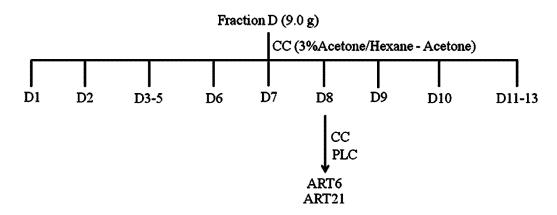
### 2.3.3.2 Separation of fraction D

Fraction D (9.0 g) was separated by CC and eluted with gradient solvents from 3% Me<sub>2</sub>CO in hexane to Me<sub>2</sub>CO. The eluted fractions were combined to give thirteen fractions (D1-D13) on the basis of their TLC characteristics. Physical appearance and

weight of each fraction are summarized in **Table 5**. After separation and purification, only two compounds were isolated (**Scheme 6**).

Fraction	Eluent	Weight (mg)	Physical Appearance
D1	3% Me <sub>2</sub> CO/hexane	959.8	white amorphous solid
D2	3% Me <sub>2</sub> CO/hexane	890.0	yellow amorphous solid
D3	3% Me <sub>2</sub> CO/hexane	452.1	yellow powder
D4	3% Me <sub>2</sub> CO/hexane	502.4	yellow powder
D5	3% Me <sub>2</sub> CO/hexane	225.6	yellow gum
D6	5% Me <sub>2</sub> CO/hexane	379.6	dark green gum
D7	5% Me <sub>2</sub> CO/hexane	558.2	dark green gum
D8	10% Me <sub>2</sub> CO/hexane	508.9	dark green powder
D9	10% Me <sub>2</sub> CO/hexane	338.5	dark green powder
D10	20% Me <sub>2</sub> CO/hexane	768.5	dark green powder
D11	20% Me <sub>2</sub> CO/hexane	868.3	dark green powder
D12	20-50% Me <sub>2</sub> CO/hexane	526.3	dark green powder
D13	50% Me <sub>2</sub> CO/hexane-	710.0	dark green powder
	Me <sub>2</sub> CO		

 Table 5
 Physical appearance and weight of fractions obtained from CC of fraction D



Scheme 6 Separation and purification of ART6 and ART21

#### **Isolation of ART6 and ART21**

Fraction D8 (508.9 mg) was chromatographed on CC and eluted with 10% Me<sub>2</sub>CO in hexane. The eluted fractions were combined into four fractions (D8.1-D8.4) based on their TLC characteristics. Fraction D8.2 (68.3 mg) was further rechromatographed on CC using 5% Me<sub>2</sub>CO in hexane as an eluent to give six fractions (D8.21-D8.26). Fraction D8.21 (19.6 mg) was further purified by TLC plate using 5% Me<sub>2</sub>CO in hexane as a mobile phase to produce a white solid of **ART6** (2.4 mg). Purification of fraction D8.22 (9.3 mg) by TLC plate using 5% Me<sub>2</sub>CO in hexane as a mobile phase to produce a white solid of **ART6** (2.4 mg). Purification of fraction D8.22 (9.3 mg) by TLC plate using 5% Me<sub>2</sub>CO in

### ART21

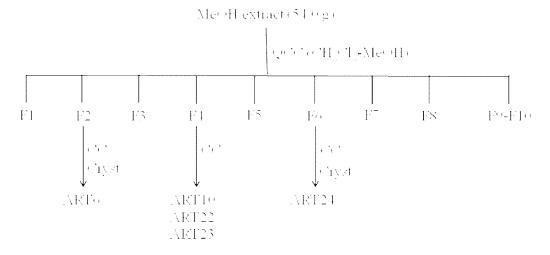
mp	75-77 <sup>o</sup> C
EI-MS <i>m</i> / <i>z</i> (% rel. int.)	614 (M <sup>+</sup> , 90), 194 (27), 120 (100), 43 (15)
UV (MeOH) $\lambda_{max}$ nm (log $\varepsilon$ )	240 (3.80), 327 (3.59)
IR (CHCl <sub>3</sub> ) $v_{\text{max}}$ (cm <sup>-1</sup> )	3445, 1720, 1642, 1619, 1505, 1460, 1173
$^{1}\mathrm{H}$ (300 MHz) and $^{13}\mathrm{C}$ NMR	$(75 \text{ MHz}) (\text{CDCl}_3)$ See Table 42

### 2.3.4 Purification of the MeOH extract from the leaves

The MeOH extract (54.0 g) was separated by QCC and eluted with the gradient solvents from  $CH_2Cl_2$  to 50% MeOH in  $CH_2Cl_2$ . The eluted fractions were combined into ten fractions (F1-F10) on the basis of their TLC characteristics. Physical appearance and weight of each fraction are summarized in **Table 6**. After separation and purification, three compounds were additionally isolated (**Scheme 7**).

Fraction	Eluent	Weight (g)	Physical Appearance
F1	CH <sub>2</sub> Cl <sub>2</sub>	0.22	dark brown gum
F2	CH <sub>2</sub> Cl <sub>2</sub>	0.61	dark brown gum
F3	CH <sub>2</sub> Cl <sub>2</sub> -2% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	0.19	dark brown gum
F4	2-5% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	2.11	dark brown gum
F5	5% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	0.74	dark brown gum
F6	10% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	0.81	dark brown gum
F7	10% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	5.14	dark brown gum
F8	15-20% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	6.34	dark brown gum
F9	20-30% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	14.50	dark brown gum
F10	30-50% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	12.43	dark brown gum

 Table 6
 Physical appearance and weight of fractions from QCC of the MeOH extract



Scheme 7 Separation and purification of ART6, ART10 and ART22-ART24

# **Isolation of ART6**

Fraction F2 (610.5 mg) was separated by CC on silica gel and eluted with 80%  $CH_2Cl_2$ in hexane. The eluted fractions were combined into eighteen fractions (F2.1-F2.18) based on their TLC characteristics. Crystallization of the combined fractions F2.10-F2.13 (175.7 mg) from a mixed solvent of hexane: Me<sub>2</sub>CO (5:1) gave **ART6** (99.7 mg) as a white solid.

#### **Isolation of ART10, ART22 and ART23**

Fraction F4 (2.11 g) was subjected to CC and eluted with the gradient solvents of 10% Me<sub>2</sub>CO in hexane to Me<sub>2</sub>CO. The eluted fractions were combined into fractions F4.1-F4.9 based on their TLC characteristics. Fraction F4.8 (457.1 mg) was rechromatographed on CC eluting with 20% Me<sub>2</sub>CO in hexane to afford ten fractions (F4.81-F4.810). Compound **ART10** (6.4 mg) was then crystallized from fraction F4.86 (14.0 mg) using hexane:Me<sub>2</sub>CO (3:2) as a solvent. Crystallization of the combined fractions F4.87-F4.89 (212.2 mg) in a mixed solvent of hexane:Me<sub>2</sub>CO (3:2) produced F4.87S (42.5 mg) as a yellow solid. The solid F4.87S was subjected to CC and eluted with gradient solvents of 2% MeOH in CH<sub>2</sub>Cl<sub>2</sub> to 10% MeOH in CH<sub>2</sub>Cl<sub>2</sub> to give fractions F4.87S1-F4.87S6. A yellow solid of **ART10** (4.3 mg) was obtained from fraction F4.87S1. Fractions 4.87S3 (4.9 mg) and F4.87S4 (15.6 mg) were combined and then purified by TLC plate using 3% MeOH in CH<sub>2</sub>Cl<sub>2</sub> as an eluent to produce **ART22** (8.7 mg) and **ART23** (4.3 mg) as white solids.

# ART22

mp272-274 °CUV (MeOH)  $\lambda_{max}$  nm (log $\varepsilon$ )211 (2.40), 229 (2.16), 315 (3.96)IR (CHCl<sub>3</sub>)  $v_{max}$  (cm<sup>-1</sup>)3359, 2929, 2858, 1696, 1610, 1455, 1169, 1021<sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>+CD<sub>3</sub>OD)See Table 33

### ART23

mp188-189 °CUV (MeOH)  $\lambda_{max}$  nm (log $\varepsilon$ )205 (4.29), 227 (2.09), 312 (3.08)IR (CHCl<sub>3</sub>)  $v_{max}$  (cm<sup>-1</sup>)3340, 2923, 2852, 1697, 1604, 1458, 1169, 1020<sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>+CD<sub>3</sub>OD)See **Table 34** 

### **Isolation of ART24**

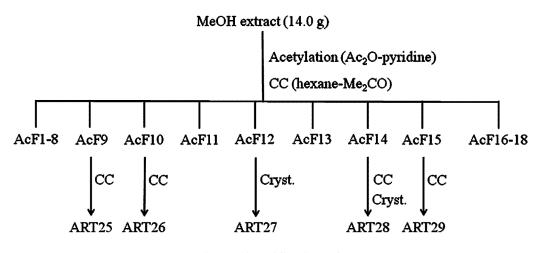
Fraction F6 (0.81 g) was separated by CC and eluted with gradient solvents from 2% MeOH in CH<sub>2</sub>Cl<sub>2</sub> to 20% MeOH in CH<sub>2</sub>Cl<sub>2</sub>. The eluted fractions were combined into seven fractions (F6.1-F6.7) based on their TLC characteristics. Fraction F6.5 (64.7 mg) was crystallized from a solvent system of hexane:Me<sub>2</sub>CO (3:2) to produce a white solid of **ART24** (25.3 mg).

### ART24

IR (CHCl<sub>3</sub>)  $v_{max}$  (cm<sup>-1</sup>) 3450, 1665, 1600, 1460, 1073 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>+CD<sub>3</sub>OD) See **Table 39** 

# 2.3.5 Acetylation of the MeOH extract from the leaves

The MeOH extract (14.0 g) was reacted with acetic anhydride in pyridine at room temperature overnight. The reaction mixture was worked up in usual manner, to give the crude product (AcF; 6.29 g) as a dark green gum. The crude AcF (6.00 g) was then subjected to CC and eluted with solvent systems from 15% Me<sub>2</sub>CO in hexane and increasing a polarity with Me<sub>2</sub>CO. The eluted fractions were combined into eighteen fractions (AcF1-AcF18) on the basis of their TLC characteristics. Physical appearance and weight of each fraction are summarized in **Table 7**. After separation and purification, five compounds were obtained (**Scheme 8**).



Scheme 8 Separation and purification of ART25-ART29

Fraction	Eluent	Weight (mg)	Physical Appearance
AcF1	15% Me <sub>2</sub> CO/hexane	147.7	pale yellow gum
AcF2	15% Me <sub>2</sub> CO/hexane	54.6	pale brown gum
AcF3	15% Me <sub>2</sub> CO/hexane	61.7	green gum
AcF4	15% Me <sub>2</sub> CO/hexane	62.9	green gum
AcF5	15% Me <sub>2</sub> CO/hexane	20.2	green gum
AcF6	20% Me <sub>2</sub> CO/hexane	54.8	green gum
AcF7	20% Me <sub>2</sub> CO/hexane	31.0	green gum
AcF8	30% Me <sub>2</sub> CO/hexane	44.9	green gum
AcF9	30% Me <sub>2</sub> CO/hexane	87.6	green gum
AcF10	30% Me <sub>2</sub> CO/hexane	110.0	pale yellow gum
AcF11	30% Me <sub>2</sub> CO/hexane	233.0	pale yellow gum
AcF12	50% Me <sub>2</sub> CO/hexane	529.5	pale yellow gum + solid
AcF13	50% Me <sub>2</sub> CO/hexane	389.1	pale yellow gum
AcF14	50% Me <sub>2</sub> CO/hexane	1.39g	pale yellow gum
AcF15	50% Me <sub>2</sub> CO/hexane	390.0	pale yellow gum
AcF16	50-70% Me <sub>2</sub> CO/hexane	345.3	brown gum
AcF17	70% Me <sub>2</sub> CO/hexane	117.0	brown gum
AcF18	Me <sub>2</sub> CO	79.6	brown gum

 Table 7 Physical appearance and weight of fractions obtained from CC of the crude AcF

# **Isolation of ART25**

Fraction AcF9 (87.6 mg) was further separated by CC eluting with 15% Me<sub>2</sub>CO in hexane to produce a pale yellow solid of **ART25** (2.7 mg).

### ART25

UV (CHCl<sub>3</sub>)  $\lambda_{max}$  nm (log $\varepsilon$ ) 248 (3.41), 307 (2.49), 320 (2.44), 365 (2.71), 380 (2.73) IR (CHCl<sub>3</sub>)  $\nu_{max}$  (cm<sup>-1</sup>) 3302, 2920, 2853, 1656, 1638, 1597, 1442, 1021 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>) See **Table 30** 

#### **Isolation of ART26**

Fraction AcF10 (110.0 mg) was subjected to CC and eluted with 15% Me<sub>2</sub>CO in hexane to yield a white solid of **ART26** (54.5 mg).

#### ART26

IR (CHCl<sub>3</sub>)  $v_{max}$  (cm<sup>-1</sup>) 3027, 2965, 1750, 1373, 1224, 1046 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>) See **Table 49** 

#### **Isolation of ART27**

Fraction AcF12 (529.5 mg) was crystallized in 15% Me<sub>2</sub>CO in hexane at room temperature to give **ART27** (232.1 mg) as a white solid.

# ART27

IR (CHCl<sub>3</sub>)  $v_{max}$  (cm<sup>-1</sup>) 3025, 2964, 1759, 1369, 1221, 1078, 1038 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>) See **Table 50** 

#### **Isolation of ART28**

Fraction AcF14 (1.39 g) was fractionated by CC and eluted with gradient solvent systems from 2% MeOH in  $CH_2Cl_2$  to 25% MeOH in  $CH_2Cl_2$  to give fractions AcF14.1-AcF14.8. Fraction AcF14.1 (95.5 mg) was recrystallized in 5% MeOH in  $CH_2Cl_2$  to produce **ART28** (10.5 mg) as a white solid.

### **ART28**

UV (EtOH)  $\lambda_{max}$  nm (log  $\varepsilon$ ) 248 (4.12), 371 (3.90) IR (CHCl<sub>3</sub>)  $\nu_{max}$  (cm<sup>-1</sup>) 1735, 1720, 1606, 1272, 1054 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>) See **Table 27** 

### **Isolation of ART29**

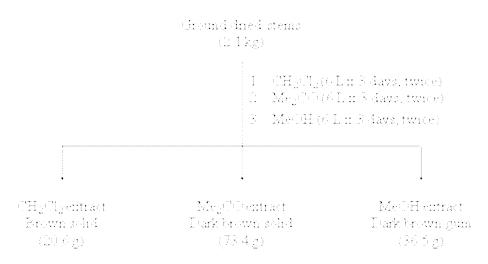
Fraction AcF15 (409.6 mg) was subjected to CC eluting with gradient manner by increasing a polarity from 2% MeOH in  $CH_2Cl_2$  to 50% MeOH in  $CH_2Cl_2$  to give eight fractions (AcF15.1-AcF15.8). Compound **ART29** (59.5 mg) was obtained as a white solid from fraction AcF15.5.

# ART29 IR (CHCl<sub>3</sub>) $v_{max}$ (cm<sup>-1</sup>) 3026, 2965, 1757, 1747, 1372, 1235, 1041 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>) See Table 51

### 2.4 Chemical investigation of the stems

#### 2.4.1 Extraction and isolation

The dried ground stems of *R. tomentosa* (2.1 kg) were successively extracted at room temperature with  $CH_2Cl_2$ ,  $Me_2CO$  and MeOH to give the  $CH_2Cl_2$ ,  $Me_2CO$  and MeOH extracts as a brown solid (20.6 g), a dark brown solid (73.4 g) and a dark brown solid (36.5 g), respectively. The extract preparations were shown in **Scheme 9**.



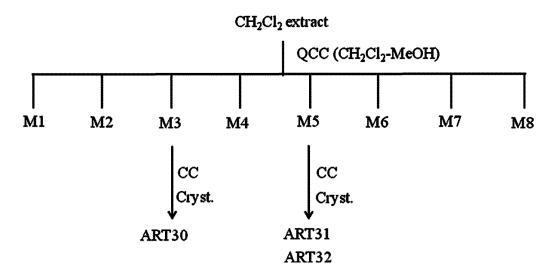
Scheme 9 The extract preparations from the stems

#### 2.4.2 Purification of the CH<sub>2</sub>Cl<sub>2</sub> extract from the stems

The  $CH_2Cl_2$  extract (20.0 g) was separated by QCC and eluted with gradient solvents from  $CH_2Cl_2$  to 30% MeOH in  $CH_2Cl_2$ . The eluted fractions were combined into eight fractions (M1-M8) on the basis of their TLC characteristics. Physical appearance and weight of each fraction are summarized in **Table 8**. After separation and purification, three compounds were isolated (**Scheme 10**).

Fraction	Eluent	Weight (g)	Physical Appearance
M1	CH <sub>2</sub> Cl <sub>2</sub>	0.25	yellow gum
M2	CH <sub>2</sub> Cl <sub>2</sub>	0.43	yellow gum
M3	CH <sub>2</sub> Cl <sub>2</sub>	3.46	yellow gum
M4	CH <sub>2</sub> Cl <sub>2</sub> -2% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	0.50	dark yellow gum
M5	2% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	6.61	dark yellow gum
M6	5% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	1.42	dark yellow gum
M7	10-15% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	3.58	dark yellow gum
M8	20-30% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	4.64	dark yellow gum

 Table 8
 Physical appearance and weight of fractions obtained from QCC of the CH<sub>2</sub>Cl<sub>2</sub> extract



Scheme 10 Separation and purification of ART30-ART32

### **Isolation of ART30**

Fraction M3 (3.46 g) was subjected to CC and eluted with 3% Me<sub>2</sub>CO in hexane to afford twelve fractions (M3.1-M3.12) based on their TLC characteristics. Fraction M3.5 (89.5 mg) was further chromatographed on CC using 3% Me<sub>2</sub>CO in hexane as an eluent to give fractions M3.51-M3.54. Upon standing at room temperature, a white solid of **ART30** (2.0 mg) was obtained from crystallization of fraction M3.53 (13.4 mg) in 3% Me<sub>2</sub>CO in hexane.

### ART30

mp	90-92 °C
EI-MS <i>m</i> / <i>z</i> (% rel. int.)	584 (M <sup>+</sup> , 13), 556 (17), 528 (15), 248 (100), 203 (75),
	166 (59), 164 (100), 147 (57)
UV (MeOH) $\lambda_{max}$ nm (log $\varepsilon$ )	234 (4.15), 327 (4.06)
IR (CHCl <sub>3</sub> ) $v_{max}$ (cm <sup>-1</sup> )	3490, 1696, 1668, 1472, 1158
$^{1}\mathrm{H}$ (300 MHz) and $^{13}\mathrm{C}$ NMR	$(75 \text{ MHz}) (\text{CDCl}_3)$ See <b>Table 43</b>

### **Isolation of ART31 and ART32**

Fraction M5 (6.61 g) was subjected to CC and eluted with a polarity gradient from 5% Me<sub>2</sub>CO in hexane to 40% Me<sub>2</sub>CO in hexane to give sixteen fractions (M5.1-M5.16) based on their TLC characteristics. Fraction M5.7 (15.2 mg) was further purified by CC using 10% Me<sub>2</sub>CO in hexane as an eluent to produce a pale yellow amorphous powder of compound **ART31** (2.0 mg). Fraction M5.12 (379.4 mg) was separated by CC eluting with 10% Me<sub>2</sub>CO in hexane to afford fractions M5.121-M5.126. Compound **ART32** (15.0 mg) was crystallized from fraction M5.125 (64.0 mg) using 10% Me<sub>2</sub>CO in hexane as a solvent.

# ART31

374.0645 (calcd. for C <sub>18</sub> H <sub>14</sub> O <sub>9</sub> , 374.0638)	
374 (M <sup>+</sup> , 100), 359 (54), 331 (28), 127 (35), 98 (31),	
69 (76)	
248 (4.15), 376 (3.90), 410sh (3.57)	
3336, 1719, 1656, 1472, 1024	
<sup>13</sup> C NMR (75 MHz) (CDCl <sub>3</sub> ) See <b>Table 29</b>	

# ART32

mp	240-242 <sup>o</sup> C
UV (MeOH) $\lambda_{max}$ nm (log $\varepsilon$ )	230 (4.09), 315 (4.25)
IR (MeOH) $v_{\text{max}}$ (cm <sup>-1</sup> )	3345, 1700, 1608, 1508, 1452, 1170
$^{1}\mathrm{H}$ (300 MHz) and $^{13}\mathrm{C}$ NMR	(75 MHz) (CDCl <sub>3</sub> +CD <sub>3</sub> OD) See Table 35

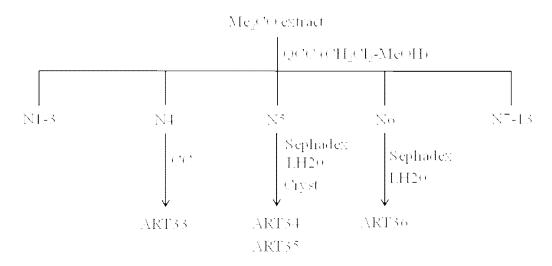
### 2.4.3 Purification of the Me<sub>2</sub>CO extract from the stems

The Me<sub>2</sub>CO extract (25.0 g) was separated by QCC and eluted with gradient solvents from  $CH_2Cl_2$  to 80% MeOH in  $CH_2Cl_2$ . The eluted fractions were combined into thirteen fractions (N1-N13) on the basis of their TLC characteristics. Physical appearance and weight of each fraction are summarized in **Table 9**. After separation and purification, four compounds were obtained (**Scheme 11**).

**Table 9**Physical appearance and weight of fractions obtained from CC of theMe2CO

Fraction	Eluent	Weight (mg)	Physical Appearance
N1	CH <sub>2</sub> Cl <sub>2</sub>	108.9	yellow solid
N2	CH <sub>2</sub> Cl <sub>2</sub>	151.3	pale green solid
N3	CH <sub>2</sub> Cl <sub>2</sub> -1% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	143.8	pale green solid
N4	1-2% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	319.1	pale green solid
N5	2-5% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	968.5	pale green solid
N6	10% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	915.4	dark brown solid
N7	10% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	79.7	brown solid
N8	10% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	185.4	brown solid
N9	10% MeOH/CH <sub>2</sub> Cl <sub>2</sub>		dark brown solid
N10	10% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	165.0	dark brown solid
N11	20% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	619.7	dark brown solid
N12	20% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	1.61g	dark brown solid
N13	40-80% MeOH/CH <sub>2</sub> Cl <sub>2</sub>		dark brown solid

extract of the stems



Scheme 11 Separation and purification of ART33-ART36

### **Isolation of ART33**

Fraction N4 (319.1 mg) was further subjected to CC and eluted with gradient solvent systems from 1% MeOH in  $CH_2Cl_2$ to 25% MeOH in  $CH_2Cl_2$  to produce a pale yellow solid of **ART33** (26.2 mg).

# ART33

 $[\alpha]_D^{29}$ -37.5 (c 0.40, MeOH)UV (MeOH)  $\lambda_{max}$  nm (log  $\varepsilon$ )216 (4.10), 230 (4.13), 290 (4.07), 327 (4.11)IR (CHCl<sub>3</sub>)  $v_{max}$  (cm<sup>-1</sup>)3422, 2938, 2849, 1700, 1684, 1516, 1270<sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>)See **Table 47** 

### **Isolation of ART34 and ART35**

Fractions N5 (968.5 mg) was further separated by CC using gradient manner from 30% Me<sub>2</sub>CO in hexane to Me<sub>2</sub>CO to give sixteen fractions (N5.1-N5.13). Fraction N5.9 (96.8 mg) was further subjected to CC on Sephadex LH-20 eluting with 60% MeOH in CH<sub>2</sub>Cl<sub>2</sub> to afford five fractions (N5.9A-N5.9E). Crystallization of fraction N5.10 (461.9 mg) using 50% Me<sub>2</sub>CO in hexane produced compound **ART34** (30.5 mg) as a white solid. Fraction N5.11 (701.0 mg) was separated by CC on Sephadex LH-20 and eluted with gradient solvent systems from 2% MeOH in CH<sub>2</sub>Cl<sub>2</sub> to 50% MeOH in  $CH_2Cl_2$  affording sixteen fractions (N5.11A-N5.11P). Fraction N5.11L (67.6 mg) was then subjected to CC on Sephadex LH-20 eluting with 60% MeOH in  $CH_2Cl_2$  to produce **ART35** (3.8 mg) as a pale yellow gum.

# ART34

mp	300-302 <sup>o</sup> C
$\left[\alpha\right]_{\mathrm{D}}^{25}$	+60.1 (c 1.0, EtOH)
IR (CHCl <sub>3</sub> ) $v_{\text{max}}$ (cm <sup>-1</sup> )	3422, 2929, 2868, 1765, 1684, 1051
<sup>1</sup> H (300 MHz) and <sup>13</sup> C NMR	$(75 \text{ MHz}) (\text{CDCl}_3 + \text{CD}_3 \text{OD})$ See <b>Table 36</b>

# ART35

UV (MeOH)  $\lambda_{max}$  nm (log  $\varepsilon$ ) 220 (3.30), 257 (3.23), 290 (2.96) IR (MeOH)  $\nu_{max}$  (cm<sup>-1</sup>) 3338, 2925, 2853, 1687, 1600, 1302, 1207 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>+CD<sub>3</sub>OD) See **Table 44** 

### **Isolation of ART36**

Fraction N6 (915.4 mg) was further subjected to CC on Sephadex LH-20 and eluted with MeOH to yield compound **ART36** (40.6 mg) as a yellowish solid.

### ART36

UV (MeOH)  $\lambda_{max}$  nm (log $\varepsilon$ ) 216 (3.78), 272 (3.54), 355 (2.45) IR (MeOH)  $v_{max}$  (cm<sup>-1</sup>) 3345, 2920, 2852, 1690, 1556, 1039 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (DMSO-*d*<sub>6</sub>) See **Table 45** 

# 2.5 Chemical investigation from the fruits

### 2.5.1 Extraction and isolation

The dried green fruits of *R. tomentosa* (161.7 g) were chopped and were successively extracted at room temperature with MeOH for three times to give the MeOH extract as a dark brown gum (9.6 g).

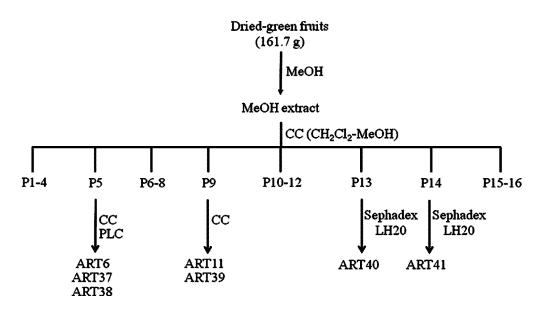
### 2.5.2 Purification of the MeOH extract from the fruits

The MeOH extract (9.3 g) was separated by CC and eluted with gradient solvents from 1% MeOH in  $CH_2Cl_2$  to 40% MeOH in  $CH_2Cl_2$ . The eluted fractions were combined into sixteen fractions (P1-P16) based on the TLC characteristics. Physical appearance and weight of each fraction are summarized in **Table 10**. After separation and purification, seven compounds were obtained (**Scheme 12**).

Fraction	Eluent	Weight (mg)	Physical Appearance
P1	1% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	284.2	yellow gum
P2	1% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	95.9	yellow gum
P3	1% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	35.5	yellow gum
P4	1% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	45.3	yellow gum
Р5	1-2% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	170.2	yellow-green gum
P6	2% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	24.8	yellow gum
P7	2% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	30.9	yellow gum
P8	5% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	48.0	yellow gum
Р9	5% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	317.3	dark green gum
P10	5% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	325.3	dark green gum
P11	5-10% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	113.7	dark green gum
P12	10% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	116.2	dark green gum
P13	10-20% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	163.5	dark green gum
P14	20% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	291.0	dark green gum
P15	20% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	263.9	dark green gum
P16	20-40% MeOH/CH <sub>2</sub> Cl <sub>2</sub>	569.1	dark green gum

 Table 10
 Physical appearance and weight of fractions obtained from CC of the

 MeOH extract



Scheme 12 Separation and purification of ART6, ART11 and ART37-ART41

# Isolation of ART6, ART37 and ART38

Fraction P5 (170.2 mg) was fractionated by CC and eluted with  $CH_2Cl_2$  to give five fractions (P5.1-P5.5). Fraction P5.1 (16.6 mg) was purified by TLC plate using 5% Me<sub>2</sub>CO in hexane as a mobile phase to produce **ART37** (1.1 mg), **ART6** (4.5 mg) and **ART38** (1.0 mg) as a pale yellow gum.

# **ART37**

UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	246 (4.08)	
IR (CHCl <sub>3</sub> ) $v_{max}$ (cm <sup>-1</sup> )	2935, 2873, 1674, 14	63, 1024
$^{1}\mathrm{H}$ (500 MHz) and $^{13}\mathrm{C}$ NMR	(125 MHz) (CDCl <sub>3</sub> )	See Table 40

# **ART38**

HR EI-MS	462.2019 (calcd. for C <sub>28</sub> H <sub>30</sub> O <sub>6</sub> , 462.2037)
EI-MS (% rel. int.)	462 (M <sup>+</sup> , 70), 434 (57), 405 (26), 385 (100), 315 (301),
	257 (18), 149 (17)
UV (CHCl <sub>3</sub> ) $\lambda_{max}$ nm (log $\varepsilon$ )	246 (3.97), 267 (4.10), 304 (4.20)
IR (CHCl <sub>3</sub> ) $v_{max}$ (cm <sup>-1</sup> )	3304, 2961, 2932, 1718, 1650, 1590, 1387, 1165, 1085
$^{1}$ H (500 MHz) and $^{13}$ C NMR	(125 MHz) (CDCl <sub>3</sub> ) See Table 21

#### **Isolation of ART11 and ART39**

Fraction P9 (317.3 mg) was subjected to CC and eluted with 30% Me<sub>2</sub>CO in hexane affording fractions P9.1-P9.7. Fraction P9.2 (88.0 mg) was further purified by CC using 20% Me<sub>2</sub>CO in hexane as an eluent to produce **ART11** (3.7 mg) and **ART39** (10.5 mg) as white solids.

### ART39

mp	284-286 <sup>o</sup> C	
IR (MeOH) $v_{\text{max}}$ (cm <sup>-1</sup> )	3406, 2925, 2858, 1685, 1654	, 1458, 1274, 1029
<sup>1</sup> H (300 MHz) and <sup>13</sup> C NMR	(75 MHz) (CDCl <sub>3</sub> +CD <sub>3</sub> OD)	See Table 37

#### **Isolation of ART40**

Fraction P13 (163.5 mg) was subjected to CC on Sephadex LH-20 using MeOH as an eluent to give four fractions (P13.1-P13.4). Fraction P13.3 (39.0 mg) was further purified by CC on Sephadex LH-20 and eluted with MeOH to produce a yellow gum of **ART40** (7.0 mg).

# ART40

UV (MeOH)  $\lambda_{max}$  nm (log $\varepsilon$ ) 218 (3.98), 274 (3.66), 356 (2.66) IR (MeOH)  $v_{max}$  (cm<sup>-1</sup>) 3328, 2925, 2853, 1706, 1687, 1563, 1439, 1248, 1039 <sup>1</sup>H (300 MHz) and <sup>13</sup>C NMR (75 MHz) (CDCl<sub>3</sub>+CD<sub>3</sub>OD) See **Table 46** 

### **Isolation of ART41**

Fraction P14 (291.0 mg) was fractionated by CC on Sephadex LH-20 and eluted with MeOH to give four fractions (P14.1-P14.4). Crystallization of fraction P14.3 (39.1 mg) using 20% MeOH in  $CH_2Cl_2$  as a solvent produced compound **ART41** (7.0 mg) as white solid.

### ART41

$$\begin{split} & [\alpha]_{D}^{25} & -33.0 \ (c \ 0.08, \ \text{MeOH}) \\ & \text{UV (MeOH)} \ \lambda_{\text{max}} \ \text{nm (log} \varepsilon) & 207 \ (3.43), 253 \ (3.68), 274 sh \ (3.53), 314 \ (3.12), 350 \\ & (3.18) \\ & \text{IR (MeOH)} \ \nu_{\text{max}} \ (\text{cm}^{-1}) & 3349, 2972, 2925, 1687, 1656, 1581, 1437, 1220, 1060 \\ ^1\text{H (300 MHz) and} \ ^{13}\text{C NMR} \ (75 \ \text{MHz}) \ (\text{CDCl}_3 + \text{CD}_3\text{OD}) & \text{See Table 28} \end{split}$$

### 2.6 Antibacterial activity

Acylphloroglucinols: **ART2**, **ART4**, **ART6**, **ART9**, **ART11**, **ART18**, **ART19** and **ART20** were tested on antibacterial activity against three strain Gram-positive bacteria: *Stephylococcus aureus* ATCC 25923, methicillin-resistant *S. aureus* NRPC R01 (MRSA) and *Streptococcus pyogenes* DMST 101, and two strain Gram-negative bacteria: *Escherichia coli* ATCC 25922 and *E. coli* O157:H7 (RIMD 05091078).

The broth micro-dilution method was used to determine the minimal inhibitory concentration (MIC) (CLSI, 2006). The bacterial suspensions ( $10^5$  cfu/mL) were added into Muller Hinton broth (MHB) supplemented with the plant extracts or pure compounds at concentrations that ranged from 0.049 to 100  $\mu$ g/mL, incubated at 37°C for 20 h (5% lysed blood was added to the MHB for *S. pyogenes*). All assays were carried out in triplicate. MIC values were recorded as the lowest concentration that produced a complete suppression of visible growth.

# CHAPTER 3

# **RESULTS AND DISCUSSION**

The preliminary study found that the crude extracts from its leaves exhibited strong antibacterial activity against *Stephylococcus aureus* and methicillin-resistant *S. aureus* (MRSA). The CH<sub>2</sub>Cl<sub>2</sub> and Me<sub>2</sub>CO extracts showed good activities with MIC values of 31.25 and 62.5  $\mu$ g/mL, respectively, whereas the CH<sub>2</sub>Cl<sub>2</sub> and Me<sub>2</sub>CO extracts from its stems displayed no activities. The leaves, stems and fruits of *R. tomentosa* were investigated for chemical constituents and 41 compounds were obtained. The structures of all compounds were determined by analyses of spectral data in association with comparison with the literatures.

#### 3.1 Structure elucidation

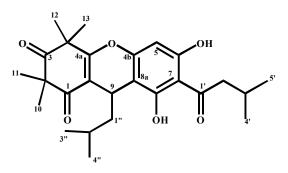
The dried ground leaves of *R. tomentosa* were successively extracted at room temperature with  $CH_2Cl_2$ ,  $Me_2CO$  and MeOH. Separation of the Me\_2CO extract yielded six new compounds: five acylphloroglucinols (ART2-ART4, ART9 and ART11) and one butyrolactone derivative (ART15) together with eleven known compounds: ART1, ART5-ART8, ART10, ART12-ART14, ART16 and ART17. Separation of the  $CH_2Cl_2$  extract yielded additionally three new acylphloroglucinols: ART18-ART20 along with one known: ART21. Separation of the MeOH extract yielded three known compounds: ART22-ART24. Addition to the above methods, the MeOH extract was further subjected to acetic anhydride acetylation in the presence of pyridine. After workup and purification, five known compounds, ART25-ART29, were also obtained.

The dried ground stems were successively extracted at room temperature with  $CH_2Cl_2$ ,  $Me_2CO$  and MeOH to give the  $CH_2Cl_2$ ,  $Me_2CO$  and MeOH extracts. Separation of the  $CH_2Cl_2$  extract yielded one new flavellagic acid derivative: **ART31** along with two known compounds: **ART30** and **ART32**. Separation of the  $Me_2CO$  extract gave four known compounds: **ART33-ART36**.

The dried green fruits were extracted at room temperature with MeOH to give the MeOH extract. Separation of this extract gave two new acylphloroglucinols: **ART11** and **ART38** together with four known compounds: **ART37**, **ART39-ART41**.

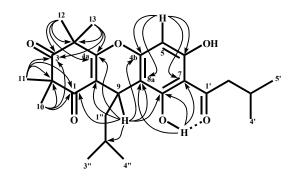
### 3.1.1 Phloroglucinols

ART6: Rhodomyrtone



**ART6** was obtained as a white solid, mp 188-189 °C with  $[\alpha]_D^{25}$  -9.4° (c = 1.0, CHCl<sub>3</sub>). The UV spectrum exhibited absorption maxima at 244, 262, and 302 nm. The IR spectrum showed absorption bands for hydroxyl, saturated carbonyl and conjugated carbonyl groups at 3244, 1720 and 1630 cm<sup>-1</sup>, respectively. The <sup>1</sup>H NMR spectrum (**Table 11**) showed the presence of a chelated hydroxyl group ( $\delta_H$  13.47, 8-OH), a free hydroxyl group ( $\delta_H$  8.22, 6-OH), an aromatic proton ( $\delta_H$  6.21, H-5), an isopentyl group ( $\delta_H$  4.31, *t*, H-9; 1.48, *obscure*, H<sub>2</sub>-1" and H-2"; 0.88, *d*, H<sub>3</sub>-3" and 0.85, *d*, H<sub>3</sub>-4"), an isovaleryl group ( $\delta_H$  3.07 and 2.97, H<sub>2</sub>-2'; 2.29, H-3' and 0.99, H<sub>3</sub>-4' and H<sub>3</sub>-5') and four singlet methyl groups of a  $\beta$ -triketone moiety ( $\delta_H$  1.57, H<sub>3</sub>-12; 1.46, H<sub>3</sub>-11; 1.44, H<sub>3</sub>-13 and 1.40, H<sub>3</sub>-10). The signals of H<sub>3</sub>-12 and H<sub>3</sub>-13 showed HMBC correlations (**Table 11**) to the carbonyl carbon C-3 ( $\delta_C$  212.17) and vinylic oxycarbon C-4a ( $\delta_C$  167.65) whereas H<sub>3</sub>-10 and H<sub>3</sub>-11 showed correlations to carbonyl carbons C-1 ( $\delta_C$  198.53) and C-3 ( $\delta_C$  212.17) supporting the presence of a  $\beta$ -triketone moiety. The methine proton H-9 and the chelated hydroxyl 8-OH exhibited the correlations with C-8 ( $\delta_C$  162.86) and C-8a ( $\delta_C$  106.41) in the HMBC experiment

indicating the location of the chelated hydroxyl at C-8. These data indicated that the isovaleryl side chain was consequently placed at C-7 ( $\delta_{\rm C}$  107.76). The <sup>13</sup>C NMR and DEPT spectra (**Table 11**) displayed twenty five signals for twenty six carbon atoms; three carbonyl carbons ( $\delta_{\rm C}$  212.17, 206.77, 198.53), nine quaternary carbons ( $\delta_{\rm C}$  167.65, 162.86, 158.76, 155.70, 114.35, 107.76, 106.41, 56.08 and 47.29), four methine carbons [ $\delta_{\rm C}$  94.81, 25.27 (2xC) and 25.17], two methylene carbons ( $\delta_{\rm C}$  53.21 and 45.85) and eight methyl carbons ( $\delta_{\rm C}$  24.78, 24.65, 26.64, 24.23, 23.57, 23.23, 22.84 and 22.78). The structure of **ART6** was then assigned as a known dimeric acylphloroglucinol named rhodomyrtone (**Table 54**), which was previously isolated from the leaves of this plant (Dachriyanus *et al.*, 2002). The assignment was fully confirmed by the HMBC experiment.



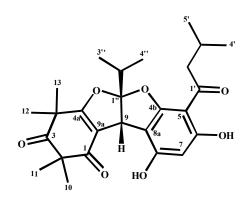
Selected HMBC correlations of ART6

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1	198.53 (C=O)		
2	56.08 (C)		
3	212.17 (C=O)		
4	47.29 (C)		
4a	167.65 (C)		
4b	155.70 (C)		
5	94.81 (CH)	6.21 ( <i>s</i> )	C-4b, C-6, C-7, C-8a, C-9,
			C-1'
6	158.76 (C)		
7	107.76 (C)		
8	162.86 (C)		
8a	106.41(C)		
9	25.27 (CH)	4.31 ( <i>t</i> ; 5.7)	C-1, C-4a, C-4b, C-8, C-8a,
			C-9a, C-1", C-2"
9a	114.35 (C)		
10	24.23 (CH <sub>3</sub> )	1.40 ( <i>s</i> )	C-1, C-2, C-3, C-12
11	24.64 (CH <sub>3</sub> )	1.46 ( <i>s</i> )	C-1, C-2, C-3, C-11
12	24.65 (CH <sub>3</sub> )	1.57 ( <i>s</i> )	C-3, C-4, C-4a, C-14
13	24.78 (CH <sub>3</sub> )	1.44 (s)	C-3, C-4, C-4a, C-13
1'	206.77 (C=O)		
2'	53.21 (CH <sub>2</sub> )	3.07 ( <i>dd</i> ; 15.6, 6.6)	C-1', C-3', C-4', C-5'
		2.97 ( <i>dd</i> ; 15.6, 6.6)	
3'	25.17 (CH)	2.29 ( <i>m</i> ; 6.6)	C-1', C-2', C-4', C-5'
4'	22.84 (CH <sub>3</sub> )	0.99 ( <i>d</i> ; 6.6)	C-2', C-3', C-5'
5'	22.78 (CH <sub>3</sub> )	0.99 ( <i>d</i> ; 6.6)	C-2', C-3', C-4'

 Table 11
 The NMR spectral data of ART6

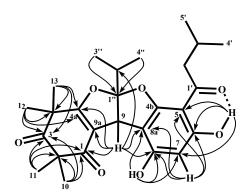
 Table 11 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1"	45.85 (CH <sub>2</sub> )	1.48 ( <i>m</i> )	
2"	25.27 (CH)	1.48 ( <i>m</i> )	
3"	23.57 (CH <sub>3</sub> )	0.88 ( <i>d</i> ; 6.6)	C-1", C-2"
4"	23.23 (CH <sub>3</sub> )	0.85 ( <i>d</i> ; 6.6)	C-1", C-2"
6-ОН		8.22 ( <i>br s</i> ; OH)	
8-OH		13.47 ( <i>br s</i> ; OH)	C-7, C-8, C-8a

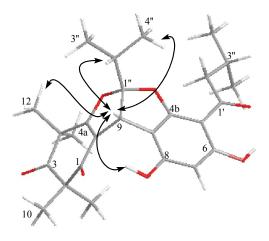


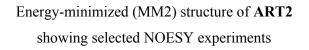
ART2 was obtained as a white solid, mp 125-126 °C. The IR spectrum displayed absorption bands of a hydroxyl (3126 cm<sup>-1</sup>), a non-conjugated carbonyl (1720 cm<sup>-1</sup>) and a conjugated carbonyl (1650 cm<sup>-1</sup>) groups. The UV spectrum exhibited absorption maxima at 270 and 327 nm. The HREI-MS spectrum showed a molecular ion peak at m/z 456.2133 corresponding to a molecular formula of  $C_{26}H_{32}O_7$  with eleven degrees of unsaturation. The <sup>13</sup>C NMR and DEPT spectra (Table 12) showed three carbonyl, ten quaternary, four methine, one methylene and eight methyl carbons. The <sup>1</sup>H NMR spectrum (Table 12) showed the resonances of four methyl groups at  $\delta_{\rm H}$  1.52 (H<sub>3</sub>-10), 1.42 (H<sub>3</sub>-11), 1.41 (H<sub>3</sub>-13) and 1.34 (H<sub>3</sub>-12). The signals of  $H_3$ -12 and  $H_3$ -13 showed HMBC correlations (Table 12) to the carbonyl carbon C-3 ( $\delta_{\rm C}$  211.14) and vinylic oxycarbon C-4a ( $\delta_{\rm C}$  179.68) whereas H<sub>3</sub>-10 and H<sub>3</sub>-11 showed correlations to carbonyl carbons C-1 ( $\delta_{\rm C}$  198.32) and C-3 ( $\delta_{\rm C}$  211.14) indicating the presence of a  $\beta$ -triketone moiety similar to rhodomyrtone (Dachriyanus *et al.*, 2002). The low field chemical shift of C-4a ( $\delta_{\rm C}$  179.68) indicated that the  $\beta$ -triketone moiety was connected to the oxygen of a furan ring (Fukuyama et al., 1998; Shaheen et al., 2006). The signals of the two hydroxyl groups ( $\delta_{\rm H}$  13.27, s, 6-OH and 9.78, s, 8-OH), an aromatic proton ( $\delta_{\rm H}$  6.11, s, H-7) and signals corresponding to an isovaleryl group ( $\delta_{\rm H}$  2.96 and 2.76, dd each, H<sub>2</sub>-2'; 2.17, m, H-3'; 1.01, d, H<sub>3</sub>-4' and  $0.99, d, H_{3}$ -5') were derived from a di-C-substituted phloroglucinol moiety with

an isovaleryl group (Bloor, 1992). The spectrum further showed signals of a methine proton ( $\delta_{\rm H}$  4.50, s, H-9) and an isopropyl group ( $\delta_{\rm H}$  2.40, hept, H-2"; 1.11, d, H<sub>3</sub>-3" and 1.09, d, H<sub>3</sub>-4"). The loss of a 43 m/z (C<sub>3</sub>H<sub>7</sub>) and 85 m/z (C<sub>4</sub>H<sub>9</sub>CO) from a molecular ion, confirmed the presence of isopropyl and isovaleryl groups. The HMBC correlations of the methine proton H-9 to C-4a, C-4b, C-8, C-8a, C-9a and C-2" as well as of the methyl protons of an isopropyl group to C-1" ( $\delta_{\rm C}$  129.36) provided evidence that the  $\beta$ -triketone was combined to a phloroglucinol moiety via a bisfuran fused-ring bearing the isopropyl group. The <sup>3</sup>J HMBC correlations of 6-OH and 8-OH to an aromatic methine carbon C-7 and of 6-OH to C-5 indicated that the aromatic proton was in between two hydroxyl groups (C-7), consequently the isovaleryl group was then placed at C-5 rather than C-7. The correlations of the methine proton (H-9) to the isopropyl protons in the NOESY experiment provided the assignment of a *cis* relative stereochemistry. **ART2**, a novel dimeric acylphloroglucinol named as rhodomyrtosone A, was thus identified as 8,10-dihydroxy-5a-isopropyl-2.2.4.4-tetramethyl-7-(3-methyl-butyryl)-5a,10b-dihydro-4*H*-benzo[*b*]benzo[4,5] furo[3,2-*d*]furan-1,3-dione.



Selected HMBC correlations of ART2



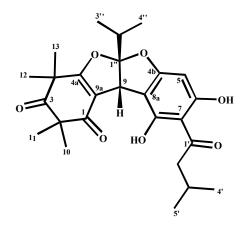


Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1	198.32 (C=O)		
2	55.12 (C)		
3	211.14 (C=O)		
4	45.62 (C)		
4a	179.68 (C)		
4b	159.76 (C)		
5	101.71 (C)		
6	166.70 (C)		
7	99.56 (CH)	6.11 ( <i>s</i> )	C-5, C-6, C-8, C-8a
8	159.64 (C)		
8a	104.21 (C)		
9	44.97 (CH)	4.50 (s)	C-4a, C-8, C-8a, C-9a, C-2"
9a	113.20 (C)		

Table 12 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
10	24.35 (CH <sub>3</sub> )	1.52 (s)	C-1, C-2, C-3, C-11
11	24.11 (CH <sub>3</sub> )	1.42 (s)	C-1, C-2, C-3, C-10
12	23.13 (CH <sub>3</sub> )	1.34 (s)	C-3, C-4, C-5, C-13
13	25.90 (CH <sub>3</sub> )	1.41 (s)	C-3, C-4, C-5, C-12
1'	203.68 (C=O)		
2'	51.51 (CH <sub>2</sub> )	2.96 ( <i>dd</i> ; 14.7, 6.6)	C-1', C-3', C-4', C-5'
		2.76 ( <i>dd</i> ; 14.7, 6.6)	
3'	25.79 (CH)	2.17 ( <i>m</i> ; 6.6)	C-2', C-4', C-5'
4'	22.75 (CH <sub>3</sub> )	1.01 ( <i>d</i> ; 6.6)	C-2', C-3', C-5'
5'	22.71 (CH <sub>3</sub> )	0.99 ( <i>d</i> ; 6.6)	C-2', C-3', C-4'
1"	129.36 (C)		
2"	35.35 (CH)	2.40 ( <i>hept</i> ; 6.9)	C-1", C-3", C-4"
3"	15.71 (CH <sub>3</sub> )	1.11 ( <i>d</i> ; 6.9)	C-1", C-2", C-4"
4"	15.65 (CH <sub>3</sub> )	1.09 ( <i>d</i> ; 6.9)	C-1", C-2", C-3"
6-OH		13.27 (s)	C-5, C-6
8-OH		9.78 (s)	C-7, C-8

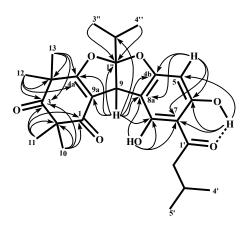
#### **ART3:** Rhodomyrtosone H



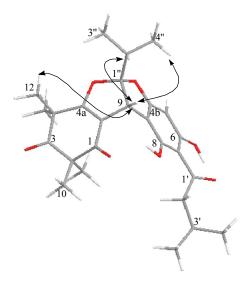
**ART3** was obtained as a white solid. The IR spectrum displayed absorption bands of a hydroxyl (3316 cm<sup>-1</sup>), a non-conjugated carbonyl (1720 cm<sup>-1</sup>) and a conjugated carbonyl (1636 cm<sup>-1</sup>) group whereas the UV spectrum exhibited the absorption bands at 244, 271 and 325 nm. A molecular formula of C<sub>26</sub>H<sub>32</sub>O<sub>7</sub> was established from a molecular ion peak at m/z 456.2133 in the HREI-MS spectrum. The <sup>13</sup>C NMR and DEPT spectra (**Table 13**) showed three carbonyl, ten quaternary, four methine, one methylene and eight methyl carbons. The <sup>1</sup>H NMR spectrum (Table 13) showed similar signals as those found in ART2 including the presence of  $\beta$ -triketone moiety [ $\delta_{\rm H}$  1.42 (H<sub>3</sub>-10), 1.36 (H<sub>3</sub>-11), 1.43 (H<sub>3</sub>-13) and 1.51 (H<sub>3</sub>-12)], two hydroxyl groups ( $\delta_{\rm H}$  14.21, s, 6-OH and 10.20, s, 8-OH), an aromatic proton ( $\delta_{\rm H}$  6.08, s, H-5) and signals corresponding to an isovaleryl group ( $\delta_{\rm H}$  3.10 and 2.97, dd each, H<sub>2</sub>-2'; 2.26, m, H-3'; 1.00, d, H<sub>3</sub>-4' and 0.98, d, H<sub>3</sub>-5'). The spectrum further showed signals of a methine proton ( $\delta_{\rm H}$  4.48, s, H-9) and an isopropyl group ( $\delta_{\rm H}$  2.39, hept, H-2"; 1.06, d, H<sub>3</sub>-3" and 1.04, d, H<sub>3</sub>-4"). The above evidences indicated that ART3 was a structural isomer of ART2 (rhodomyrtosone A) which was each differed in only the acyl side chain position. The evidences from the HMBC correlations (Table 13) of two hydroxyl groups with C-7 indicated that the isovaleryl group was then placed at C-7. The higher field hydroxyl group ( $\delta_{\rm H}$  10.20) at C-8 was confirmed

by the HMBC correlations of both this hydroxyl group and H-9 to C-8 and C-8a. Consequently, the lower field hydroxyl group ( $\delta_{\rm H}$  14.21) was then located at C-6.

The correlations of the methine proton (H-9) to the isopropyl protons in the NOESY experiment provided the assignment of a *cis* relative stereochemistry. **ART3**, a novel dimeric acylphloroglucinol named as rhodomyrtosone H, was thus identified as 7,9-dihydroxy-1a-isopropyl-3,3,5,5-tetramethyl-8-(3-methylbutanoyl)-5,6b-dihydrobenzo[*b*]benzofuro[3,2-*d*]furan-4,6(1aH,3H)-dione.



Selected HMBC correlations of ART3



Energy-minimized (MM2) structure of **ART3** showing selected NOESY experiments

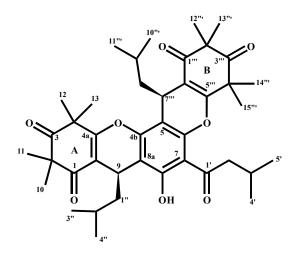
Table 13	The NMR	spectral	data	of ART3
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Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1	198.45 (C=O)		
2	55.98 (C)		
3	211.08 (C=O)		
4	45.69 (C)		
4a	179.80 (C)		
4b	163.07 (C)		
5	91.80 (CH)	6.08 (s)	C-4b, C-6, C-7, C-8a
6	168.51 (C)		
7	107.12 (C)		
8	156.54 (C)		
8a	103.79 (C)		
9	45.02 (CH)	4.48 (s)	C-4a, C-4b, C-8, C-8a, C-9a,
			C-1", C-2"
9a	113.15 (C)		

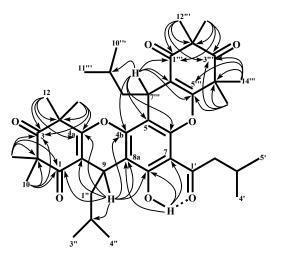
 Table 13 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
10	24.27 (CH <sub>3</sub> )	1.42 (s)	C-1, C-2, C-3, C-11
11	23.29 (CH <sub>3</sub> )	1.36 (s)	C-1, C-2, C-3, C-10
12	24.27 (CH <sub>3</sub> )	1.51 (s)	C-3, C-4, C-4a, C-13
13	25.82 (CH <sub>3</sub> )	1.43 (s)	C-3, C-4, C-4a, C-12
1'	206.32 (C=O)		
2'	51.94 (CH <sub>2</sub> )	3.10 ( <i>dd</i> ; 16.0, 6.5)	C-1', C-3', C-4', C-5'
		2.97 ( <i>dd</i> ; 16.0, 6.5)	
3'	25.12 (CH)	2.26 ( <i>m</i> ; 6.5)	C-1', C-2', C-4', C-5'
4'	22.91 (CH <sub>3</sub> )	1.00 ( <i>d</i> ; 6.5)	C-2', C-3', C-5'
5'	22.73 (CH <sub>3</sub> )	0.98 ( <i>d</i> ; 6.5)	C-2', C-3', C-4'
1"	129.33 (C)		
2"	35.23 (CH)	2.39 ( <i>hept</i> ; 6.5)	C-9, C-1", C-3", C-4"
3"	15.58 (CH <sub>3</sub> )	1.06 ( <i>d</i> ; 6.5)	C-1", C-2", C-4"
4"	15.53 (CH <sub>3</sub> )	1.04 ( <i>d</i> ; 6.5)	C-1", C-2", C-3"
6-ОН		14.21 (s)	C-5, C-6, C-7
8-OH		10.20 (s)	C-7, C-8, C8a

#### ART4: Rhodomyrtosone C



ART4 was obtained as a yellowish solid. The IR spectrum displayed absorptions of a hydroxyl (3423 cm<sup>-1</sup>), non-conjugated carbonyl (1717 cm<sup>-1</sup>) and conjugated carbonyl groups (1659 cm<sup>-1</sup>) whereas the UV spectrum exhibited maximum absorptions at 263, 306 and 348 nm. Its molecular formula of  $C_{41}H_{54}O_8$ was established on the basis of a molecular ion peak at m/z 674.3853 in its HREI-MS spectrum. The <sup>1</sup>H and <sup>13</sup>C NMR spectra (Table 14) showed the singlet signals that corresponded to two  $\beta$ -triketone moieties [ring A:  $\delta_{\rm H}$  1.64 (H<sub>3</sub>-12), 1.48 (H<sub>3</sub>-13), 1.40 (H<sub>3</sub>-10) and 1.37 (H<sub>3</sub>-11); ring B:  $\delta_{\rm H}$  1.66 (H<sub>3</sub>-14"'), 1.52 (H<sub>3</sub>-15"'), 1.44 (H<sub>3</sub>-12"') and 1.42 (H<sub>3</sub>-13"')]. Furthermore, two sets of resonances at  $\delta_{\rm H}$  4.35 (*t*, H-9), 1.50 (obscure, H<sub>2</sub>-1" and H-2"), 0.90 (d, H<sub>3</sub>-4") and 0.83 (d, H<sub>3</sub>-3") and at  $\delta_{\rm H}$  4.39 (t, H-7""), 1.50 (obscure, H<sub>2</sub>-8" and H-9"), 0.98 (d, H<sub>3</sub>-11") and 0.84 (d, H<sub>3</sub>-10") were in agreement with the resonances of two isopentyl groups. The remaining resonances were those of an isovaleryl group ( $\delta_{\rm H}$  3.23 and 3.02, dd each, H<sub>2</sub>-2'; 2.40, m, H-3'; 1.05, d, H<sub>3</sub>-4' and 1.04, d, H<sub>3</sub>-5') with their carbonyl function forming a hydrogen bonding to the hydroxyl group ( $\delta_{\rm H}$  13.50, 8-OH). **ART4** was therefore identified as 7-hydroxy-8,14diisobutyl-2,2,4,4,10,10,12,12-octamethyl-6-(3-methyl-butyryl)-4,8,12,14-tetrahydro-5,13-dioxapentaphene-1,3,9,11-tetraone and it was named as rhodomyrtosone C. The HMBC correlations confirmed the assigned structure (Table 14).



Selected HMBC correlations of ART4

 Table 14
 The NMR spectral data of ART4

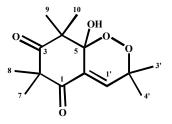
Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
1	197.38 (C=O)		
2	56.01 (C)		
3	211.62 (C=O)		
4	47.25 (C)		
4 <sup>a</sup>	166.69 (C)		
4b	152.37 (C)		
5	105.68 (C)		
6	150.50 (C)		
7	107.61 (C)		
8	160.61 (C)		
8a	107.76 (C)		
9	25.60 (CH)	4.35 ( <i>t</i> ; 5.7)	C-1, C-4a, C-4b, C-8, C-8a,
			C-9a, C-1", C-2"
9a	114.28 (C)		

 Table 14 (Continued)

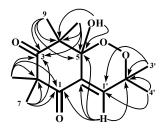
Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
10	24.87 (CH <sub>3</sub> )	1.40 (s)	C-1, C-2, C-3, C-11
11	24.36 (CH <sub>3</sub> )	1.37 (s)	C-1, C-2, C-3, C-10
12	24.69 (CH <sub>3</sub> )	1.64 ( <i>s</i> )	C-3, C-4, C-4a, C-13
13	25.03 (CH <sub>3</sub> )	1.48 (s)	C-3, C-4, C-4a, C-12
1'	204.59 (C=O)		
2'	53.89 (CH <sub>2</sub> )	3.23 ( <i>dd</i> ; 17.4, 6.6)	C-1', C-3', C-4', C-5'
		3.02 ( <i>dd</i> ; 17.4, 6.6)	
3'	24.53 (CH)	2.40 ( <i>m</i> ; 6.6)	C-1', C-2', C-4', C-5'
4'	22.64 (CH <sub>3</sub> )	1.05 ( <i>d</i> ; 6.6)	C-2', C-3', C-5'
5'	22.80 (CH <sub>3</sub> )	1.04 ( <i>d</i> ; 6.6)	C-2', C-3', C-4'
1"	45.44 (CH <sub>2</sub> )	1.50 (obscure)	
2"	25.30 (CH)	1.50 (obscure)	
3"	23.32 (CH <sub>3</sub> )	0.83 ( <i>d</i> ; 6.0)	C-1", C-2"
4"	23.28 (CH <sub>3</sub> )	0.90 ( <i>d</i> ; 6.0)	C-1", C-2"
1'''	197.45 (C=O)		
2'''	56.20 (C)		
3'''	211.39 (C=O)		
4'''	47.28 (C)		
5'''	166.76 (C)		
6'''	113.53 (C)		
7'''	25.20 (CH)	4.39 ( <i>t</i> ; 5.4)	C-4b, C-5, C-6, C-1"", C-5"",
			C-6''', C-8''', C-9'''
8'''	46.78 (CH <sub>2</sub> )	1.50 (obscure)	
9'''	25.03 (CH)	1.50 (obscure)	
10'''	23.36 (CH <sub>3</sub> )	0.84 ( <i>d</i> ; 6.0)	C-8''', C-9'''
11'''	23.78 (CH <sub>3</sub> )	0.98 ( <i>d</i> ; 6.0)	C-8''', C-10'''

 Table 14 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
12'''	24.17 (CH <sub>3</sub> )	1.44 (s)	C-1''', C-2''', C-3'''
13'''	23.81 (CH <sub>3</sub> )	1.42 (s)	C-1''', C-2''', C-3'''
14'''	25.34 (CH <sub>3</sub> )	1.66 ( <i>s</i> )	C-3''', C-4''', C-5'''
15'''	24.93 (CH <sub>3</sub> )	1.52 (s)	C-3"', C-4"'', C-5"''
6-OH		13.50 (s)	C-7, C-8, C-8a



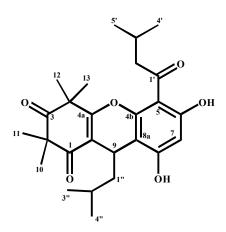
ART8 was obtained as a yellowish gum. The IR spectrum displayed absorption bands of a hydroxyl (3479 cm<sup>-1</sup>), a non-conjugated carbonyl (1714 cm<sup>-1</sup>) and a conjugated carbonyl (1693 cm<sup>-1</sup>) group whereas the UV spectrum displayed absorption bands at 243 and 294 nm. The EI-MS spectrum showed a molecular ion peak at m/z 268. The <sup>1</sup>H NMR spectral data (**Table 15**) showed the resonances of four methyl groups at  $\delta_{\rm H}$  1.39 (H<sub>3</sub>-8), 1.37 (H<sub>3</sub>-7), 1.34 (H<sub>3</sub>-9) and 1.06 (H<sub>3</sub>-10). The evidences that  $H_3$ -9 and  $H_3$ -10 showed <sup>3</sup>J HMBC correlations (**Table 15**) to carbonyl carbons C-3 ( $\delta_{\rm C}$  210.51) whereas H<sub>3</sub>-7 and H<sub>3</sub>-8 showed <sup>3</sup>J HMBC correlations to the carbonyl carbons C-3 ( $\delta_{\rm C}$  210.51) and C-1 ( $\delta_{\rm C}$  198.23) indicated the presence of a  $\beta$ -triketone moiety. A peroxide moiety with an olefinic proton ( $\delta_{\rm H}$  7.16, H-1'), two methyl groups  $(\delta_{\rm H} 1.51, {\rm H}_3-3' \text{ and } 1.39, {\rm H}_3-4')$  and a hydroxyl group  $(\delta_{\rm H} 3.62, 5-{\rm OH})$  were proposed. The low field chemical shift of H-1' ( $\delta_{\rm H}$  7.16) suggested that it was at a *peri* position to carbonyl group whereas the low field chemical shift of C-5 ( $\delta_{\rm C}$  97.46) revealed that the hydroxyl group and a peroxide were located at C-5. Finally, the HMBC correlation of H-1' to C-1 ( $\delta_{\rm C}$  198.23), C-5 ( $\delta_{\rm C}$  97.46), C-6 ( $\delta_{\rm C}$  131.74), C-3' ( $\delta_{\rm C}$  23.71) and C-4' ( $\delta_{\rm C}$  23.88) and the analysis of mass spectrum also confirmed this assignment. Thus ART8 was elucidated to be endoperoxide G3 (Crow et al., 1971).



Selected HMBC correlations of ART8

Table 15The NMR spectral data of ART8

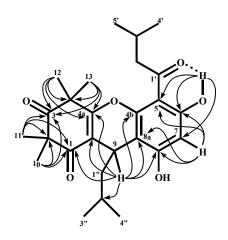
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	198.23 (C=O)		
2	55.01 (C)		
3	210.51 (C=O)		
4	51.63 (C)		
5	97.46 (C)		
6	131.74 (C)		
7	24.05 (CH <sub>3</sub> )	1.37 (s)	C-1, C-2, C-3, C-8
8	26.59 (CH <sub>3</sub> )	1.39 (s)	C-1, C-2, C-3, C-7
9	15.17 (CH <sub>3</sub> )	1.34 (s)	C-3, C-4, C-5, C-10
10	20.94 (CH <sub>3</sub> )	1.06 ( <i>s</i> )	C-3, C-4, C-5, C-9
1'	142.92 (CH)	7.16 ( <i>s</i> )	C-1, C-5, C-6, C-2'
2'	79.41 (C)		
3'	23.71 (CH <sub>3</sub> )	1.51 (s)	C-1', C-2', C-4'
4'	23.88 (CH <sub>3</sub> )	1.39 (s)	C-1', C-2', C-3'
5-OH		3.62 ( <i>s</i> )	C-4, C-5



ART9 was a yellowish gum. The IR spectrum displayed absorption bands of a hydroxyl (3372 cm<sup>-1</sup>), a non-conjugated carbonyl (1717 cm<sup>-1</sup>) and a conjugated carbonyl (1653 cm<sup>-1</sup>) group. The UV spectrum exhibited absorption bands at 292 and 333 nm. Its molecular ion peak at m/z 442.2352 in the HREI-MS spectrum corresponded to a molecular formula of  $C_{26}H_{34}O_6$  with ten degrees of unsaturation. The appearance of the proton resonances (**Table 16**) of a chelated hydroxyl group ( $\delta_{\rm H}$  13.43, 6-OH), a free hydroxyl group ( $\delta_{\rm H}$  6.40, 8-OH), an aromatic proton ( $\delta_{\rm H}$  6.23, H-7), an isopentyl group ( $\delta_{\rm H}$  4.25, *t*, H-9; 1.38, obscure, H<sub>2</sub>-1" and H-2"; 0.89, d, H<sub>3</sub>-3" and 0.87, d, H<sub>3</sub>-4"), an isovaleryl group ( $\delta_{\rm H}$  3.18 and 2.96, H<sub>2</sub>-2'; 2.37, H-3'; 1.04, H<sub>3</sub>-4' and 1.01, H<sub>3</sub>-5') and four singlet methyl groups of a  $\beta$ -triketone moiety ( $\delta_{\rm H}$  1.63, H<sub>3</sub>-12; 1.47, H<sub>3</sub>-13; 1.42, H<sub>3</sub>-11 and 1.39, H<sub>3</sub>-10) as well as its molecular ion of 442.2352 indicated that ART9 was a structural isomer of ART6 (rhodomyrtone). There were slight differences observed for the chemical shifts of the chelated hydroxyl group ( $\delta_{\rm H}$  13.43) and non-equivalent methylene protons of the isovaleryl group ( $\delta_{\rm H}$  3.18 and 2.96), consequently the isovaleryl group was placed at C-5 rather than C-7. The assignment was fully confirmed by the HMBC experiment (Table 16). ART9, named rhodomyrtosone B, was therefore identified as 6,8-dihydroxy-9-isobutyl-2,2,4,4-tetramethyl-5-(3-methylbutyryl)-4,9-dihydro-

xanthene-1,3-dione, a new naturally occurrence acylphloroglucinol. It was a

6-demethylated isomer of a synthetic 1,3-dioxo-4,9-dihydro-8-hydroxy-6-methoxy-2,2,4,4-tetramethyl-5-(3-methyl-1-oxobutyl)-9-(2-methylpropyl)-1*H*-xanthene (Bloor, 1992).



Selected HMBC correlations of ART9

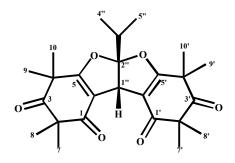
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	197.60 (C=O)		
2	56.13 (C)		
3	211.68 (C=O)		
4	47.21 (C)		
4a	166.89 (C)		
4b	153.27 (C)		
5	105.91 (C)		
6	164.32 (C)		
7	100.32 (CH)	6.23 (s)	C-5, C-6, C-8, C-9
8	159.02 (C)		

 Table 16
 The NMR spectral data of ART9

 Table 16 (Continued)

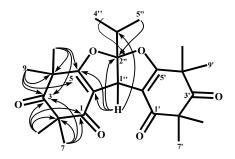
Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
8a	105.91 (C)		
9	25.08 (CH)	4.25 ( <i>t</i> ; 5.7)	C-1, C-4a, C-4b, C-8, C-8a,
			C-9a, C-1", C-2"
9a	114.54 (C)		
10	24.29 (CH <sub>3</sub> )	1.39 (s)	C-1, C-2, C-3, C-11
11	24.35 (CH <sub>3</sub> )	1.42 (s)	C-1, C-2, C-3, C-10
12	24.77 (CH <sub>3</sub> )	1.63 (s)	C-3, C-4, C-4a, C-13
13	25.36 (CH <sub>3</sub> )	1.47 (s)	C-3, C-4, C-4a, C-12
1'	203.96 (C=O)		
2'	53.56 (CH <sub>2</sub> )	3.18 ( <i>dd</i> ; 15.6, 6.5)	C-1', C-3', C-4', C-5'
		2.96 ( <i>dd</i> ; 15.6, 6.5)	
3'	24.48 (CH)	2.37 ( <i>m</i> ; 6.5)	C-1', C-2', C-4', C-5'
4'	22.89 (CH <sub>3</sub> )	1.04 ( <i>d</i> ; 6.5)	C-2', C-3', C-5'
5'	22.65 (CH <sub>3</sub> )	1.01 ( <i>d</i> ; 6.5)	C-2', C-3', C-4'
1"	46.90 (CH <sub>2</sub> )	1.38 (obscure)	
2"	24.87 (CH)	1.38 (obscure)	
3"	23.41 (CH <sub>3</sub> )	0.89 ( <i>d</i> ; 6.5)	C-1", C-2", C-4"
4"	23.10 (CH <sub>3</sub> )	0.87 ( <i>d</i> ; 6.5)	C-1", C-2", C-3"
6-OH		13.43 (s)	C-5, C-6, C-7
8-OH		6.40 ( <i>br s</i> )	

#### **ART11:** Rhodomyrtosone D

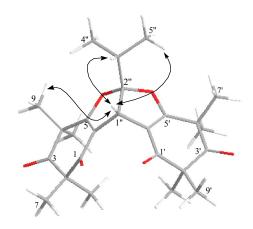


ART11 was a vellowish solid, mp 160-162 <sup>o</sup>C. The IR spectrum displayed absorption bands of a non-conjugated (1717 cm<sup>-1</sup>) and a conjugated (1675 cm<sup>-1</sup>) carbonyl functionalities. The HREI-MS spectrum showed a molecular ion peak at m/z 428.2214 corresponding to a molecular formula of C<sub>25</sub>H<sub>32</sub>O<sub>6</sub>. The <sup>13</sup>C NMR and DEPT techniques (Table 17) showed signals for four carbonyl, nine quaternary, two methine and ten methyl carbons. The <sup>1</sup>H NMR spectrum (Table 17) revealed three singlet signals of methyl groups at  $\delta_{\rm H}$ 1.27 (H<sub>3</sub>-7 and H<sub>3</sub>-7'), 1.34 (H<sub>3</sub>-8 and H<sub>3</sub>-8') and 1.44 (H<sub>3</sub>-9, H<sub>3</sub>-10, H<sub>3</sub>-9' and H<sub>3</sub>-10'). In the HMBC experiment (Table 17), these methyl groups correlated to the carbonyl carbons [ $\delta_{\rm C}$  212.10 (2xC=O) and 192.35 (2xC=O)] thus indicating the presence of two moieties of a symmetrical  $\beta$ -triketone. The resonances of a methine proton ( $\delta_{\rm H}$  4.69, s, H-1") and an isopropyl group  $(\delta_{\rm H} 2.37, hept, \text{H-3"} \text{ and } 1.02, d, \text{H}_3-4", \text{H}_3-5")$ , similar to those in ART2 (rhodomyrtosone A), were also observed. Compound ART11 was therefore 5a-isopropyl-2,2,4,4,7,7,9,9-octamethyl-7,10b-dihydro-4H,5aH-benzo[b]benzo [4,5]furo [3,2-d]furan-1,3,8,10-tetraone. The HMBC correlations of the methine proton H-1" to the C-5(5'), C-6(6') of a  $\beta$ -triketone and to C-2", C-3" of the isopropyl group along with the downfield shift of C-2" ( $\delta_{\rm C}$  128.19) confirmed the assigned structure. In addition, it was in good agreement with a molecular formula of  $C_{25}H_{32}O_6$  and the ion peaks at m/z 385, 358, 315 and 288. In the NOEDIFF spectrum, irradiation at the resonance of the methine proton H-1" resulted in enhancement of the isopropyl

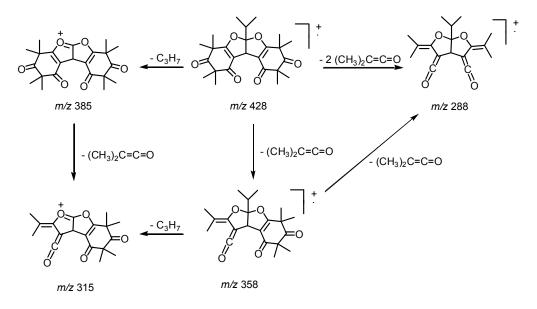
protons, indicating its *cis* stereochemistry. Thus **ART11** was a novel acylphloroglucinol named as rhodomyrtosone D.



Selected HMBC correlations of ART11



Energy-minimized (MM2) structure of **ART11** showing selected NOEDIFF experiments

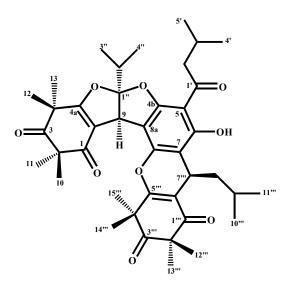


Major mass fragmentation patterns of ART11

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1 (1')	192.35 (C=O)		
2 (2')	56.48 (C)		
3 (3')	212.10 (C=O)		
4 (4')	45.18 (C)		
5 (5')	175.65 (C)		
6 (6')	113.07 (C)		
7 (7')	25.73 (CH <sub>3</sub> )	1.27 (s)	C-1 (1'), C-2 (2'), C-3 (3'),
			C-8 (8')
8 (8')	22.31 (CH <sub>3</sub> )	1.34 (s)	C-1 (1'), C-2 (2'), C-3 (3'),
			C-7 (7')
9 (9')	24.36 (CH <sub>3</sub> )	1.44 (s)	C-3 (3'), C-4 (4'), C-5 (5')
10 (10')	23.91 (CH <sub>3</sub> )	1.44 (s)	C-3 (3'), C-4 (4'), C-5 (5')
1"	46.47 (CH)	4.69 ( <i>s</i> )	C-5 (5'), C-6 (6'), C-2", C-3"
2"	128.19 (C)		
3"	34.41 (CH)	2.37 ( <i>hept</i> ; 6.9)	C-2", C-4", C-5"
4",5"	15.46 (CH <sub>3</sub> )	1.02 ( <i>d</i> ; 6.9)	C-2", C-3"

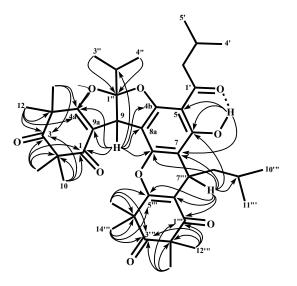
Table 17The NMR spectral data of ART11

#### **ART18:** Rhodomyrtosone E

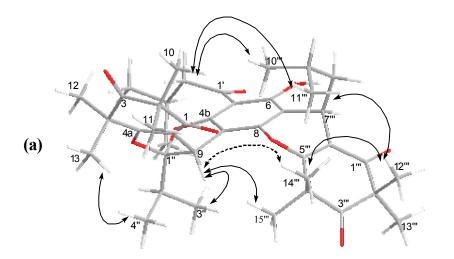


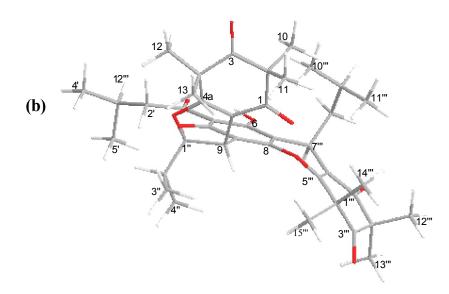
**ART18** was obtained as a yellowish gum with  $[\alpha]_D^{25}$  -9.3<sup>O</sup> (c = 0.92, CHCl<sub>3</sub>). The UV spectrum exhibited maximum absorption bands at 246, 260, 304 and 347 nm. The IR spectrum showed the absorption of a hydroxyl (3135 cm<sup>-1</sup>), a conjugated carbonyl (1656 cm<sup>-1</sup>) and a non-conjugated carbonyl (1720 cm<sup>-1</sup>) groups. A molecular ion peak at m/z 688.3610 in the HREI-MS spectrum corresponded to the molecular formula of  $C_{41}H_{52}O_{9}$ . The <sup>1</sup>H NMR spectrum (Table 18) showed the resonances of four methyl groups of a  $\beta$ -triketone moiety at  $\delta_{\rm H}$  1.47 (H<sub>3</sub>-13), 1.43 (H<sub>3</sub>-11), 1.36 (H<sub>3</sub>-12), and 1.24 (H<sub>3</sub>-10). The presence of an acylphloroglucinol moiety with an isovaleryl group was assigned from the resonances of a hydrogen bonded hydroxyl proton at  $\delta_{\rm H}$  13.62 (6-OH), non equivalent methylene protons at  $\delta_{\rm H}$  2.99 and 2.89 (H<sub>2</sub>-2'), a methine proton at  $\delta_{\rm H}$  2.23 (H-3'), and methyl protons at  $\delta_{\rm H}$ 1.03 (H<sub>3</sub>-4') and 1.01 (H<sub>3</sub>-5'). The spectrum further showed signals of a methine proton at  $\delta_{\rm H}$  4.74 (s, H-9) and the protons of an isopropyl group at  $\delta_{\rm H}$  2.37 (hept, H-2"), 1.11 (d, H<sub>3</sub>-3") and 1.09 (d, H<sub>3</sub>-4"). The HMBC correlations (Table 18) of the methine proton H-9 to C-4a, C-4b, C-8, C-8a, C-9a and C-2" as well as that of the methyl protons of the isopropyl group to C-1" ( $\delta_{\rm C}$  128.84) indicated that the  $\beta$ -triketone was combined to a phloroglucinol moiety via a bisfuran fused-ring bearing the isopropyl group as for **ART2** (rhodomyrtosone A). The resonance of additional  $\beta$ -triketone moiety with an isopentyl group was detected at  $\delta_{\rm H}$  4.28 (*t*, H-7""), 1.40 (obscure, H<sub>2</sub>-8"" and H-9""), 0.75 (*d*, H<sub>3</sub>-10""), 0.83 (*d*, H<sub>3</sub>-11""), 1.42 (*s*, H<sub>3</sub>-12""), 1.36 (*s*, H<sub>3</sub>-13""), 1.74 (*s*, H<sub>3</sub>-14""), and 1.67 (*s*, H<sub>3</sub>-15""). The <sup>3</sup>*J* HMBC correlations of the methine proton H-7"" to C-6 ( $\delta_{\rm C}$  162.53), C-8 ( $\delta_{\rm C}$  152.88), C-1"" ( $\delta_{\rm C}$  197.51) and C-5"" ( $\delta_{\rm C}$  167.70) indicated that the isopentyl  $\beta$ -triketone was linked to the acylphloroglucinol unit by C-7. The resonances of five carbonyl carbons, fifteen quaternary carbons, five methine carbons, two methylene carbons and fourteen methyl carbons that were deduced from the <sup>13</sup>C NMR and DEPT techniques (**Table 18**) were in agreement with the assigned structure.

The relative stereochemistry at C-9, C- 1" and C-7" was further studied. The methyl groups H<sub>3</sub>-14"' ( $\delta$  1.74) and H<sub>3</sub>-15"' ( $\delta$  1.67) on  $\beta$ -triketone moiety were found to resonate at the lower field than the others ( $\delta$  1.24-1.47). It was indicated that both methyl groups were lined on the deshielding zone of the carbonyl group at C-1 position (b). In addition, on the ROESY experiment (a), the methine proton H-9 showed strong correlation to methyl group H<sub>3</sub>-15" and weak correlation to methyl group H<sub>3</sub>-14". The methyl group H<sub>3</sub>-12" exhibited the correlations to both methyl groups H<sub>3</sub>-11" and H<sub>3</sub>-14"". Furthermore the methyl groups H<sub>3</sub>-10" and H<sub>3</sub>-11" of the isobutyl moiety at C-7" correlated to the methyl group H<sub>3</sub>-10. These information suggested that the isopropyl group, methine proton H-9 and isobutyl group at C-7" must be arranged in *cis-trans* configuration. The energy-minimized 3D structures generated by the MM2 force field as implemented in Chem3D® (CambridgeSoft) of **ART18** also confirmed the proposed stereochemistry. Furthermore the 3D structure showed that the non-equivalent methylene protons  $(H_2-2')$  of the isovaleryl group arranged in the similar region of environments (b), which was in agreement with the chemical shift at  $\delta_{\rm H}$  2.99 and 2.89.



Selected HMBC correlations of ART18





Energy-minimized (MM2) structure of ART18 showing

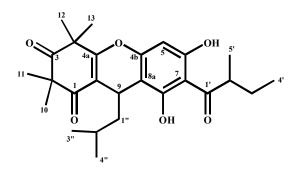
- (a) selected ROESY experiment
- (b) interaction of C=O/Me-14", Me-15" and position of CH<sub>2</sub> of the isovaleryl group

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1	192.03 (C=O)		
2	56.09 (C)		
3	212.07 (C=O)		
4	45.46 (C)		
4a	175.99 (C)		
4b	158.37 (C)		
5	103.97 (C)		
6	162.53 (C)		
7	108.73 (C)		
8	152.24 (C)		
8a	104.88 (C)		
9	45.94 (CH)	4.74 (s)	C-1, C-4a, C-4b, C-8, C-8a,
			C-9a, C-1", C-2"
9a	113.01 (C)		
10	25.82 (CH <sub>3</sub> )	1.24 (s)	C-1, C-2, C-3, C-11
11	23.84 (CH <sub>3</sub> )	1.43 (s)	C-1, C-2, C-3, C-10
12	23.84 (CH <sub>3</sub> )	1.36 (s)	C-3, C-4, C-4a, C-13
13	24.49 (CH <sub>3</sub> )	1.47 (s)	C-3, C-4, C-4a, C-12
1'	204.64 (C=O)		
2'	51.94 (CH <sub>2</sub> )	2.99 ( <i>dd</i> ; 15.0, 6.9)	C-1', C-3', C-4', C-5'
		2.89 ( <i>dd</i> ; 15.0, 6.9)	
3'	25.62 (CH)	2.23 ( <i>m</i> ; 6.9)	C-1', C-2', C-4', C-5'
4'	22.75 (CH <sub>3</sub> )	1.03 ( <i>d</i> ; 6.9)	C-2', C-3', C-5'
5'	22.66 (CH <sub>3</sub> )	1.01 ( <i>d</i> ; 6.9)	C-2', C-3', C-4'

Table 18The NMR spectral data of ART18

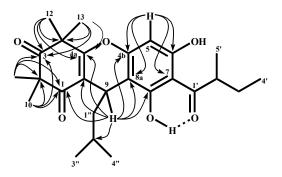
Table 18 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1"	128.84 (C)		
2"	34.87 (CH)	2.37 ( <i>hept</i> ; 6.9)	C-9, C-1", C-3", C-4"
3"	15.84 (CH <sub>3</sub> )	1.11 ( <i>d</i> ; 6.9)	C-1", C-2", C-4"
4"	15.84 (CH <sub>3</sub> )	1.09 ( <i>d</i> ; 6.9)	C-1", C-2", C-3"
1'''	197.51 (C=O)		
2'''	56.27 (C)		
3'''	212.48 (C=O)		
4'''	47.55 (C)		
5'''	167.70 (C)		
6'''	113.86 (C)		
7'''	25.26 (CH)	4.28 ( <i>t</i> ; 5.4)	C-6, C-7, C-8, C-1"", C-5"",
			C-6''', C-8''', C-9'''
8'''	45.40 (CH <sub>2</sub> )	1.40 (obscure)	
9'''	25.12 (CH)	1.40 (obscure)	
10'''	23.24 (CH <sub>3</sub> )	0.83 ( <i>d</i> ; 6.6)	C-8''', C-9''', C-11'''
11'''	23.55 (CH <sub>3</sub> )	0.75 ( <i>d</i> ; 6.6)	C-8''', C-9''', C-10'''
12'''	24.70 (CH <sub>3</sub> )	1.42 (s)	C-1''', C-2''', C-3'''
13'''	22.04 (CH <sub>3</sub> )	1.36 (s)	C-1''', C-2''', C-3'''
14'''	23.15 (CH <sub>3</sub> )	1.74 (s)	C-3''', C-4''', C-5''', C-15'''
15'''	25.26 (CH <sub>3</sub> )	1.67 (s)	C-3''', C-4''', C-5''', C-14'''
6-OH		13.62 ( <i>s</i> )	C-5, C-6, C-7



**ART19** was obtained as a yellowish gum with  $\left[\alpha\right]_{D}^{25}$  +62.0<sup>°</sup> (c = 0.10, CHCl<sub>3</sub>). The UV spectrum displayed absorption maxima at 245, 262 and 300 nm. The IR spectrum showed absorption bands of O-H stretching at 3365 cm<sup>-1</sup> and C=O stretching at 1717 and 1655 cm<sup>-1</sup>. Its  $[M+H]^+$  peak at m/z 443.2429 in the HR-FAB-MS spectrum corresponded to a molecular formula of  $C_{26}H_{34}O_6$ . The <sup>1</sup>H NMR spectrum (**Table 19**) showed the resonances of four methyl groups ( $\delta_{\rm H}$  1.55, H<sub>3</sub>-13; 1.43, H<sub>3</sub>-12; 1.40, H<sub>3</sub>-11 and 1.36, H<sub>3</sub>-10), an isopentyl group ( $\delta_{\rm H}$  4.25, H-9; 1.45, H<sub>2</sub>-1" and H-2"; 0.88, H<sub>3</sub>-3" and 0.84, H<sub>3</sub>-4"), a phloroglucinol moiety with two hydroxyl groups ( $\delta_{\rm H}$  13.00, 8-OH and 7.18, 6-OH) and an aromatic proton ( $\delta_{\rm H}$  6.07, H-5), as those of ART6 (rhodomyrtone). The side chain at C-7 was suggested to be 2-methylbutyryl group from the resonances of two methyl groups  $(1.19, d, H_3-5)$  and 0.92, t, H<sub>3</sub>-4'), a methine proton ( $\delta_{\rm H}$  3.75, sext, H-2') and methylene protons (1.84 and 1.45, *m* each, H<sub>2</sub>-3') as well as the resonance of a carbonyl carbon at  $\delta_{\rm C}$  210.82 (C-1'). The assigned structure was then confirmed by the HMBC experiment recorded in CDCl<sub>3</sub> (Table 19). The correlations of H-9 ( $\delta_{\rm H}$  4.25) to C-8, and that of the lower field hydroxyl group ( $\delta_{\rm H}$  13.00) to C-7, C-8 and C-8a were expected to confirm the position of the acyl side chain at C-7. Unfortunately, the latter correlations were not observed due to the weak signal of the hydroxyl proton. The HMBC experiment was then rerecorded in acetone- $d_6$ . The expected results were obtained. Therefore, 6,8-dihydroxy-9-isobutyl-2,2,4,4-tetramethyl-7-(2-methylbutanoyl)-4,9-dihydro-1H-

xanthene-1,3(2H)-dione was assigned for **ART19**. It was a new dimeric acylphloroglucinol named as rhodomyrtosone G.



Selected HMBC correlations of ART19

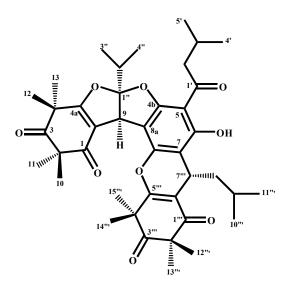
Table 19The NMR spectral data of ART19

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	197.48 (C=O)		
2	56.09 (C)		
3	212.11 (C=O)		
4	47.14 (C)		
4a	166.81 (C)		
4b	155.58 (C)		
5	94.94 (CH)	6.07 ( <i>s</i> )	C-4b, C-6, C-7, C-8a, C-1'
6	158.02 (C)		
7	107.14 (C)		
8	162.80 (C)		
8a	106.92 (C)		
9	25.23 (CH)	4.25 ( <i>t</i> ; 6.0)	C-1, C-4a, C-4b, C-8, C-8a,
			C-9a, C-1", C-2"
9a	114.25 (C)		

Table 19 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
10	24.15 (CH <sub>3</sub> )	1.36 (s)	C-1, C-2, C-3, C-11
11	24.54 (CH <sub>3</sub> )	1.40 ( <i>s</i> )	C-1, C-2, C-3, C-10
12	24.59 (CH <sub>3</sub> )	1.43 (s)	C-3, C-4, C-4a, C-13
13	24.73 (CH <sub>3</sub> )	1.55 ( <i>s</i> )	C-3, C-4, C-4a, C-12
1'	210.82 (C=O)		
2'	46.39 (CH)	3.75 ( <i>sext</i> ; 6.5)	C-1', C-3', C-4', C-5'
3'	26.95 (CH <sub>2</sub> )	1.84 ( <i>m</i> )	C-1', C-2', C-4', C-5'
		1.45 ( <i>m</i> )	
4'	11.89 (CH <sub>3</sub> )	0.92 ( <i>t</i> ; 7.0)	C-2', C-3'
5'	16.40 (CH <sub>3</sub> )	1.19 ( <i>d</i> ; 6.5)	C-1', C-2', C-3'
1"	45.89 (CH <sub>2</sub> )	1.45 ( <i>m</i> )	
2"	25.15 (CH)	1.45 ( <i>m</i> )	
3"	23.46 (CH <sub>3</sub> )	0.88 ( <i>d</i> ; 6.0)	C-1", C-2", C-4"
4"	23.16 (CH <sub>3</sub> )	0.84 ( <i>d</i> ; 6.0)	C-1", C-2", C-3"
6-OH		7.18 ( <i>br s</i> )	
8-OH		13.00 ( <i>br</i> s)	

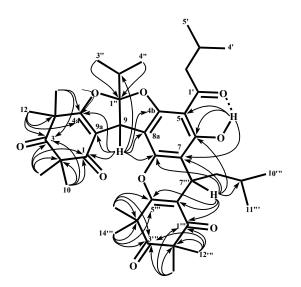
## ART20: Rhodomyrtosone F



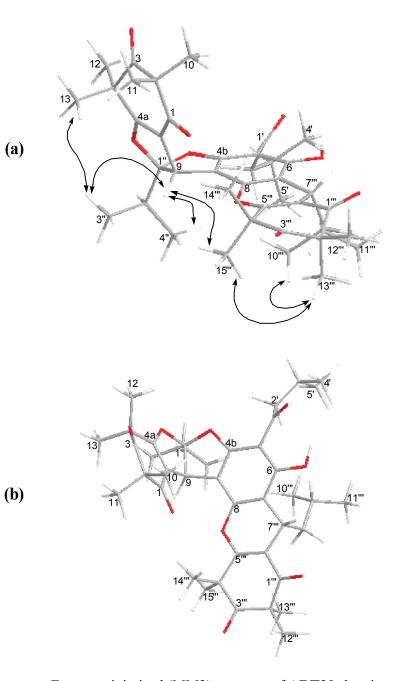
**ART20** was obtained as a pale yellow solid with mp 198-199 <sup>O</sup>C and  $[\alpha]_D^{25}$ -10.7<sup>O</sup> (c = 0.68, CHCl<sub>3</sub>). Its molecular formula of C<sub>41</sub>H<sub>52</sub>O<sub>9</sub> (M<sup>+</sup> 688.3610), its spectral data of UV, IR, MS, <sup>1</sup>H NMR (**Table 20**), <sup>13</sup>C NMR (**Table 20**) were very similar to those of **ART18** with the difference of methlyl proton signals (H<sub>3</sub>-14'''/H<sub>3</sub>-15''',  $\delta_H$  1.42/1.81 for **ART20**;  $\delta_H$  1.74/1.67 for **ART18**) and non-equivalent methylene protons (H<sub>2</sub>-2',  $\delta_H$  3.12/2.73 for **ART20**;  $\delta_H$  2.99/2.89 for **ART18**). It also showed the HMBC correlation in the same manner as **ART18** (**Table 20**). Consequently, **ART20** was suggested to be a diastereomer of **ART18**.

The relative stereochemistry at C-9, C- 1" and C-7" was further studied. The methyl group H<sub>3</sub>-15" ( $\delta_{\rm H}$  1.81) on  $\beta$ -triketone moiety was found to resonate at the lower field than the others ( $\delta_{\rm H}$  1.24-1.49). Thus indicating that this methyl group was lined in the deshielding zone of the carbonyl group at C-1 position (**a**). In addition, in the ROESY experiment (**a**), the methyl protons H<sub>3</sub>-15" showed correlations to a methine proton H-9 and a methyl group H<sub>3</sub>-13" indicating that all protons H-9, H<sub>3</sub>-13" and H<sub>3</sub>-15" were *cis*. Furthermore the methyl group H<sub>3</sub>-13" exhibited the correlation to a methyl group H<sub>3</sub>-11" of the isobutyl side chain. These information suggested that the isopropyl group, a methine proton H-9 and an isobutyl group at

C-7" must be arranged in *cis-cis* configuration. The energy-minimized 3D structures generated by the MM2 force field as implemented in Chem3D<sup>®</sup> (CambridgeSoft) of **ART20** also confirmed the proposed stereochemistry. Furthermore the 3D structure showed that the non-equivalent methylene protons (H<sub>2</sub>-2') of the isovaleryl group were arranged in the different environments (**b**), which was in agreement with the chemical shift at  $\delta_{\rm H}$  3.12 and 2.73.



Selected HMBC correlations of ART20



Energy-minimized (MM2) structure of **ART20** showing (a) selected ROESY experiment and interaction of C=O/Me-14"', Me-15"' (b) position of CH<sub>2</sub> of the isovaleryl group

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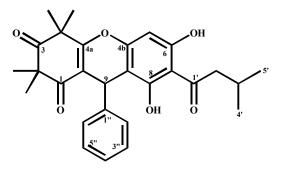
Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1	192.11 (C=O)		
2	55.74 (C)		
3	211.99 (C=O)		
4	45.42 (C)		
4a	176.89 (C)		
4b	158.71 (C)		
5	103.92 (C)		
6	162.80 (C)		
7	107.90 (C)		
8	152.10 (C)		
8a	103.75 (C)		
9	46.32 (CH)	4.81 (s)	C-1, C-4a, C-4b, C-8, C-8a,
			C-9a, C-1", C-2"
9a	113.16 (C)		
10	24.10 (CH <sub>3</sub> )	1.24 (s)	C-1, C-2, C-3, C-11
11	24.82 (CH <sub>3</sub> )	1.36 (s)	C-1, C-2, C-3, C-10
12	23.98 (CH <sub>3</sub> )	1.41 (s)	C-3, C-4, C-4a, C-13
13	24.22 (CH <sub>3</sub> )	1.49 (s)	C-3, C-4, C-4a, C-12
1'	204.68 (C=O)		
2'	52.04 (CH <sub>2</sub> )	3.12 ( <i>dd</i> ; 14.7, 6.6)	C-1', C-3', C-4', C-5'
		2.73 ( <i>dd</i> ; 14.7, 6.6)	
3'	25.95 (CH)	2.18 ( <i>m</i> ; 6.6)	C-1', C-2', C-4', C-5'
4'	22.52 (CH <sub>3</sub> )	0.99 ( <i>d</i> ; 6.6)	C-2', C-3', C-5'
5'	22.86 (CH <sub>3</sub> )	1.03 ( <i>d</i> ; 6.6)	C-2', C-3', C-4'

Table 20The NMR spectral data of ART20

Table 20 (Continued)

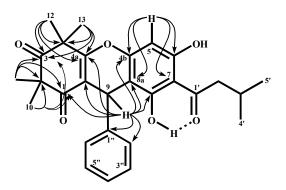
Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1"	128.86 (C)		
2"	34.86 (CH)	2.35 (hept; 6.9)	C-9, C-1", C-3", C-4"
3"	15.74 (CH <sub>3</sub> )	1.11 ( <i>d</i> ; 6.9)	C-1", C-2", C-4"
4"	15.76 (CH <sub>3</sub> )	1.08 ( <i>d</i> ; 6.9)	C-1", C-2", C-3"
1'''	198.32 (C=O)		
2'''	55.31 (C)		
3'''	212.57 (C=O)		
4'''	47.86 (C)		
5'''	165.93 (C)		
6'''	113.63 (C)		
7'''	25.44 (CH)	4.32 ( <i>dd</i> ; 5.7, 4.2)	C-6, C-7, C-8, C-1"", C-5"",
			C-6''', C-8''', C-9'''
8'''	45.27 (CH <sub>2</sub> )	1.51 ( <i>m</i> )	C-7, C-6'''
9'''	25.08 (CH)	1.40 (obscure)	
10'''	23.28 (CH <sub>3</sub> )	0.85 ( <i>d</i> ; 6.3)	C-8''', C-9''', C-11'''
11'''	23.40 (CH <sub>3</sub> )	0.69 ( <i>d</i> ; 6.3)	C-8''', C-9''', C-10'''
12'''	24.22 (CH <sub>3</sub> )	1.39 (s)	C-1''', C-2''', C-3'''
13'''	24.33 (CH <sub>3</sub> )	1.37 (s)	C-1''', C-2''', C-3'''
14'''	21.93 (CH <sub>3</sub> )	1.42 (s)	C-3''', C-4''', C-5''', C-15'''
15'''	25.83 (CH <sub>3</sub> )	1.81 (s)	C-3''', C-4''', C-5''', C-14'''
6-OH		13.64 ( <i>s</i> )	C-5, C-6, C-7

#### ART38: Rhodomyrtosone I



ART38 was obtained as a pale yellow gum. The UV spectrum exhibited absorption bands at 246, 261, and 304 nm. The IR spectrum showed absorption bands for hydroxyl, saturated carbonyl and conjugated carbonyl groups at 3304, 1718 and 1650 cm<sup>-1</sup>, respectively. Its molecular formula of C<sub>28</sub>H<sub>30</sub>O<sub>6</sub> with 14 degrees of unsaturation was established from a molecular ion at m/z 462.2017 in the HR EI-MS spectrum. The <sup>13</sup>C NMR and DEPT spectra (Table 21) showed three carbonyl, ten quaternary, eight methine, one methylene and six methyl carbons. The <sup>1</sup>H NMR spectrum (Table 21) showed resonances of four methyl groups at  $\delta_{\rm H}$  1.09 (H<sub>3</sub>-11), 1.33 (H<sub>3</sub>-10), 1.50 (H<sub>3</sub>-12) and 1.59 (H<sub>3</sub>-13). The signals of H<sub>3</sub>-12 and H<sub>3</sub>-13 showed HMBC correlations (Table 21) to the carbonyl carbon C-3 ( $\delta_{\rm C}$  211.43) and vinylic oxycarbon C-4a ( $\delta_{\rm C}$  164.58) whereas H<sub>3</sub>-10 and H<sub>3</sub>-11 showed correlations to the carbonyl carbons C-1 ( $\delta_{\rm C}$  198.85) and C-3 ( $\delta_{\rm C}$  211.43) indicating the presence of a  $\beta$ -triketone (Dachriyanus et al., 2002). The signals of the two hydroxyl groups ( $\delta_{\rm H}$  13.47, br s, 8-OH and 8.22, br s, 6-OH), an aromatic proton ( $\delta_{\rm H}$  6.17, s, H-5) and signals corresponding to an isovaleryl group ( $\delta_{\rm H}$  2.92 and 2.87, dd each, H<sub>2</sub>-2'; 2.21, m, H-3' and 0.93, d, H<sub>3</sub>-4' and H<sub>3</sub>-5') revealed a di-C-substituted phloroglucinol moiety with an isovaleryl group (Bloor, 1992). The spectrum further showed signals of a methine proton ( $\delta_{\rm H}$  5.19, s, H-9) and four aromatic protons of mono-substituted benzene ring ( $\delta_{\rm H}$  7.31, H-2"and H-6"; 7.24, H-3" and H-5" and 7.16, H-4"). The appearance of ion peak at m/z 385 in EI-MS which corresponded to the loss of a 77 m/z (C<sub>6</sub>H<sub>5</sub>) from a molecular ion confirmed the presence of the phenyl group.

The HMBC correlations of the methine proton H-9 to C-1 ( $\delta_{\rm C}$  198.85), C-9a ( $\delta_{\rm C}$  112.67) and C-4a ( $\delta_{\rm C}$  164.58) of  $\beta$ -triketone, to C-8 ( $\delta_{\rm C}$  162.01), C-8a ( $\delta_{\rm C}$  105.13) and C-4b ( $\delta_{\rm C}$  154.09) of phloroglucinol moiety, and to C-1" ( $\delta_{\rm C}$  143.44), C-2"/C-6" ( $\delta_{\rm C}$  128.52) of a benzene ring suggested that these three moieties were combined via C-9. The <sup>3</sup>*J* HMBC correlations of an aromatic H-5 to C-6 ( $\delta_{\rm C}$  159.39) and C-4b ( $\delta_{\rm C}$  154.09) confirmed the location of the aromatic proton at C-5, consequently the isovaleryl group was then placed at C-7. Therefore **ART38**, named rhodomyrtosone I, was assigned as 6,8-dihydroxy-2,2,4,4-tetramethyl-7-(2-methylbutanoyl)-9-phenyl-4,9-dihydro-1H-xanthene-1,3(2H)-dione.



Selected HMBC correlations of ART38

Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
1	198.85 (C=O)		
2	56.24 (C)		
3	211.43 (C=O)		
4	46.95 (C)		
4a	164.58 (C)		
4b	154.09 (C)		
5	95.21 (CH)	6.17 ( <i>s</i> )	C-4b, C-6, C-7, C-8a
6	159.39 (C)		
7	107.75 (C)		
8	162.01 (C)		
8a	105.13 (C)		
9	33.42 (CH)	5.19 (s)	C-1, C-4a, C-4b, C-8, C-8a,
			C-9a, C-1", C-2"
9a	112.67 (C)		
10	22.89 (CH <sub>3</sub> )	1.33 (s)	C-1, C-2, C-3, C-11
11	24.55 (CH <sub>3</sub> )	1.09 (s)	C-1, C-2, C-3, C-10
12	24.61 (CH <sub>3</sub> )	1.50 (s)	C-3, C-4, C-4a, C-13
13	24.71 (CH <sub>3</sub> )	1.59 (s)	C-3, C-4, C-4a, C-12
1'	205.92 (C=O)		
2'	53.10 (CH <sub>2</sub> )	2.92 ( <i>dd</i> ; 16.0, 6.5)	C-1', C-3', C-4', C-5'
		2.87 ( <i>dd</i> ; 16.0, 6.5)	
3'	24.88 (CH)	2.21 ( <i>m</i> ; 6.5)	C-1', C-2', C-4', C-5'
4', 5'	22.73 (CH <sub>3</sub> )	0.93 ( <i>d</i> ; 6.5)	C-2', C-3'

Table 21The NMR spectral data of ART38

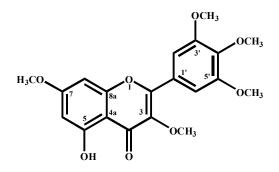
 Table 21 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
1"	143.44(C)		
2", 6"	128.52 (CH)	7.31 ( <i>dd</i> ; 7.5, 1.5)	C-9, C-3", C-4"
3", 5"	128.18 (CH)	7.24*	C-1", C-2"
4"	126.93 (CH)	7.16 ( <i>tt</i> ; 7.5, 1.5)	C-2", C-3"
6-ОН		8.22 ( <i>br</i> )	
8-OH		13.47 ( <i>br</i> )	

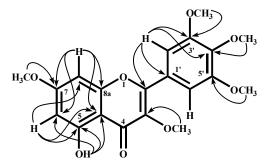
\*overlapped with solvent signal

## 3.1.2 Flavonoids

ART7: Combretol (3,3',4',5',7-Penta-O-methylmyricetin)



**ART7** was obtained as a yellowish solid, mp 142-144 <sup>o</sup>C. The IR spectrum displayed absorption bands for a hydroxyl group at 3421 cm<sup>-1</sup> and conjugated carbonyl group at 1663 cm<sup>-1</sup>. The UV spectrum exhibited the maximum absorptions at 210, 265 and 342 nm. The <sup>1</sup>H NMR spectrum (**Table 22**) showed the resonances of a chelated hydroxyl group ( $\delta_{\rm H}$  12.58, *s*, 5-OH), a vinylic methoxy group ( $\delta_{\rm H}$  3.88, *s*, 3-OCH<sub>3</sub>), and four methoxyl groups ( $\delta_{\rm H}$  3.95, *s*, 3'-, 4'- and 5'-OCH<sub>3</sub>;  $\delta_{\rm H}$  3.89, *s*, 7-OCH<sub>3</sub>). The presence of *meta* aromatic protons H-2' and H-6' was indicated from a broad singlet at  $\delta_{\rm H}$  7.37 (2H) whereas those of H-8 and H-6 was suggested from *doublet* with J = 2.1 Hz at  $\delta_{\rm H}$  6.45 and 6.37, respectively. The HMBC correlations (**Table 22**) of H-6, H-8 and 7-OCH<sub>3</sub> ( $\delta_{\rm H}$  3.89) to C-7 ( $\delta_{\rm C}$  165.58) and those of H-2' (H-6') and 3'-OCH<sub>3</sub> (5'-OCH<sub>3</sub>) to C-3' (C-5') confirmed the assignment of the methoxyl groups. The assignment of **ART7** was also in agreement with that of 3,3',4',5',7-penta-*O*-methylmyricetin or combretol (**Table 55**) which was previously obtained from the seeds and wings of *Combretum quadrangular* (Mongkolsuk *et al.*, 1966) and from the bark and twigs of *R. tomentosa* (Dachriyanus *et al.*, 2004).



Selected HMBC correlations of ART7

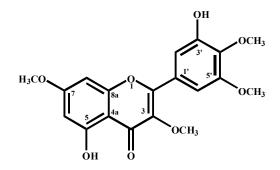
	Table 22	The NMR	spectral	data of ART7
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Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
2	155.59 (C)		
3	139.41 (C)		
4	178.77 (C=O)		
4a	106.07 (C)		
5	162.06 (C)		
6	97.93 (CH)	6.37 ( <i>d</i> ; 2.1)	C-4a, C-5, C-8
7	165.58 (C)		
8	92.26 (CH)	6.45 ( <i>d</i> ; 2.1)	C-4a, C-6, C-7, C-8a
8a	156.72 (C)		
1'	125.46 (C)		
2', 6'	106.10 (CH)	7.37 (s)	C-2, C-1', C-3' (5'), C-4'
3', 5'	153.13 (C)		
4'	140.64 (C)		
3-OCH <sub>3</sub>	60.34 (CH <sub>3</sub> )	3.88 (s)	C-3
7-OCH <sub>3</sub>	56.84 (CH <sub>3</sub> )	3.89 (s)	C-7
3', 5'-OCH <sub>3</sub>	56.35 (CH <sub>3</sub> )	3.95 (s)	C-3', C-5'
4'-OCH <sub>3</sub>	61.00 (CH <sub>3</sub> )	3.95 (s)	C-4'
5-ОН		12.58 (s)	C-4a, C-5, C-6

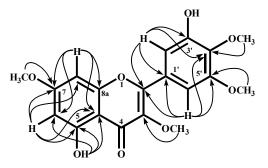
<sup>*a*</sup>300 MHz for <sup>1</sup>H and 75 MHz for <sup>13</sup>C

# ART13: 3',5-Dihydroxy-3,4',5',7-tetramethoxyflavone

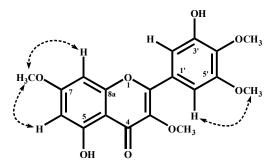
(3,4',5',7-Tetra-O-methylmyricetin)



**ART13** was obtained as a yellowish solid. The IR spectrum displayed absorption bands for a hydroxyl group at 3375 cm<sup>-1</sup> and conjugated carbonyl group at 1659 cm<sup>-1</sup> whereas the UV spectrum exhibited maximum absorptions at 248, 270, 305*sh* and 345 nm. The <sup>1</sup>H NMR spectrum (**Table 23**) which showed the resonances of two hydroxyl group ( $\delta_{\rm H}$  12.58, 5-OH; 5.95, 3'-OH), four methoxyl groups ( $\delta_{\rm H}$  4.01, 4'-OCH<sub>3</sub>; 3.95, 5'-OCH<sub>3</sub>; 3.89, 3- and 7-OCH<sub>3</sub>), and *meta* aromatic protons ( $\delta_{\rm H}$  7.36, H-2'; 7.34, H-6'; 6.46, H-8 and 6.37, H-6) indicated that it was an isomer of **ART16** and **ART17**. The NOE experiment by irradiation at the frequency of H-6 and H-8 affected the resonances of 7-OCH<sub>3</sub>, whereas irradiation at the frequency of H-2' and H-6' enhanced the methoxyl signals at C-3 and C-5'. The assigned structure was in agreement with 3',5-dihydroxy-3,4',5',7-tetramethoxyflavone or 3,4',5',7-tetra-*O*-methylmyricetin (**Table 56**) (Martos *et al.*, 1997; Datta *et al.*, 2000). The HMBC correlations (**Table 23**) fully supported the assigned structure.



Selected HMBC correlations of ART13

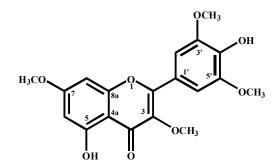


NOE correlations of ART13

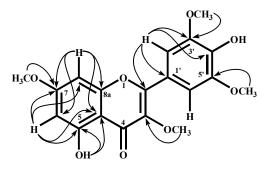
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
2	155.35 (C)		
3	139.52 (C)		
4	178.84 (C=O)		
4a	106.11 (C)		
5	162.02 (C)		
6	97.96 (CH)	6.37 ( <i>d</i> ; 2.4)	C-4a, C-5, C-7, C-8
7	165.60 (C)		
8	92.19 (CH)	6.46 ( <i>d</i> ; 2.4)	C-4a, C-6, C-7, C-8a
8a	156.76 (C)		
1'	125.96 (C)		
2'	108.59 (CH)	7.36 ( <i>d</i> ; 2.1)	C-2, C-1', C-3', C-4', C-6'
3'	149.21 (C)		
4'	137.81 (C)		
5'	152.06 (C)		
6'	105.06 (CH)	7.34 ( <i>d</i> ; 2.1)	C-2, C-1', C-2', C-4', C-5'
3-OCH <sub>3</sub>	60.32 (CH <sub>3</sub> )	3.89 ( <i>s</i> )	C-3
7-OCH <sub>3</sub>	55.82 (CH <sub>3</sub> )	3.89 ( <i>s</i> )	C-7
4'-OCH <sub>3</sub>	61.10 (CH <sub>3</sub> )	4.01 ( <i>s</i> )	C-4'
5'-OCH <sub>3</sub>	56.09 (CH <sub>3</sub> )	3.95 (s)	C-5'
5-ОН		12.58 (s)	C-4a, C-5, C-6
3'-ОН		5.95 (br s)	

Table 23The NMR spectral data of ART13

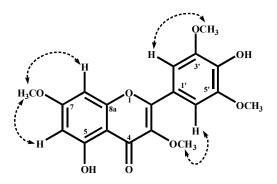
# ART16: 3,3',5',7-Tetra-O-methylmyricetin



**ART16** was obtained as a yellowish solid. The IR spectrum displayed absorption bands for a hydroxyl and conjugated carbonyl groups at 3375 and 1656 cm<sup>-1</sup>, respectively. The UV spectrum exhibited absorption bands maxima at 252, 266, 302*sh* and 355 nm. It is an isomer of **ART13** and **ART17**. The <sup>1</sup>H NMR spectrum (**Table 24**) showed the resonances corresponding to a chelated hydroxyl group ( $\delta_{\rm H}$  12.64, 5-OH), a hydroxyl group ( $\delta_{\rm H}$  5.91, 4'-OH), four methoxyl groups ( $\delta_{\rm H}$  3.99, 3'- and 5'-OCH<sub>3</sub>; 3.89, 7-OCH<sub>3</sub> and 3.86, 3-OCH<sub>3</sub>), meta aromatic protons ( $\delta_{\rm H}$  7.42, H-2' and H-6'; 6.46, H-8 and 6.37, H-6). The NOE experiment by irradiation at the frequency of H-6 and H-8 affected the resonances of 7-OCH<sub>3</sub>, whereas irradiation at the frequency of H-2' and H-6' enhanced the signals of the 3-, 3'- and 5'-OCH<sub>3</sub> groups. Therefore the structure of **ART16** was identified as 3,3',5',7-tetra-*O*-methylmyricetin (Kumari *et al.*, 1985). The assigned structure was in good agreement with the HMBC correlations (**Table 24**).



Selected HMBC correlations of ART16

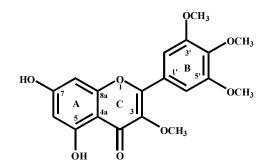


NOE correlations of ART16

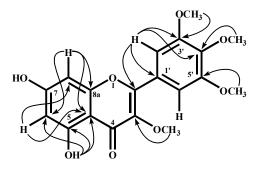
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
2	155.93 (C)		
3	139.11 (C)		
4	178.71 (C=O)		
4a	105.90 (C)		
5	162.12 (C)		
6	97.86 (CH)	6.37 ( <i>d</i> ; 1.5)	C-4a, C-5, C-7, C-8
7	165.53 (C)		
8	92.26 (CH)	6.46 ( <i>d</i> ; 1.5)	C-4, C-4a, C-6, C-7, C-8a
8a	156.70 (C)		
1'	121.44 (C)		
2', 6'	105.90 (CH)	7.42 (s)	C-2, C-1', C-3' (5'), C-4'
3', 5'	146.98 (C)		
4'	137.74 (C)		
3-OCH <sub>3</sub>	60.18 (CH <sub>3</sub> )	3.86 ( <i>s</i> )	C-3
7-OCH <sub>3</sub>	55.93 (CH <sub>3</sub> )	3.89 (s)	C-7
3', 5'-OCH <sub>3</sub>	56.56 (CH <sub>3</sub> )	3.99 (s)	C-3', C-5'
5-OH		12.64 ( <i>s</i> )	C-4a, C-5, C-6
4'-OH		5.91 (s)	

Table 24The NMR spectral data of ART16

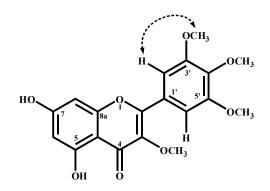
# ART17: 3,3',4',5'-Tetra-O-methylmyricetin



**ART17** was obtained as a yellowish solid. The IR spectrum exhibited absorption bands for a hydroxyl group at 3255 cm<sup>-1</sup> and a conjugated carbonyl group at 1656 cm<sup>-1</sup>. The UV spectrum displayed maximum absorptions at 252, 268, 308*sh* and 348 nm. The <sup>1</sup>H NMR spectrum (**Table 25**) showed the resonances of chelated hydroxyl group at  $\delta_{\rm H}$  12.62 (*s*, 5-OH), a hydroxyl group at  $\delta_{\rm H}$  8.67 (*br s*, 7-OH), *meta* aromatic protons at  $\delta_{\rm H}$  6.47 (*d*, H-8) and 6.35 (*d*, H-6), and at  $\delta_{\rm H}$  7.36 (*s*, H-2' and H-6'). This compound was also an isomer of **ART13** and **ART16**. The NOE experiments by irradiation at the frequency of H-2' (H-6') affected the resonances of 3'-OCH<sub>3</sub> (5'-OCH<sub>3</sub>), whereas irradiation at the frequency of H-6 and H-8 did not affect any methoxyl groups indicating that the methoxyl groups were substituted at ring C not in ring A. **ART17** was therefore identified as 3,3',4',5'-tetra-*O*methylmyricetin (Lenherr *et al.*, 1986). The assigned structure was completely confirmed by the HMBC correlations (**Table 25**).



Selected HMBC correlations of ART17



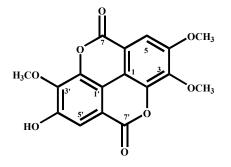
NOE correlations of ART17

Position	<b>δ</b> <sub>C</sub> (Туре)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
2	155.34 (C)		
3	139.12 (C)		
4	178.70 (C=O)		
4a	106.23 (C)		
5	162.23 (C)		
6	99.19 (CH)	6.35 ( <i>d</i> ; 1.2)	C-4a, C-5, C-8
7	163.35 (C)		
8	93.95 (CH)	6.47 ( <i>d</i> ; 1.2)	C-4a, C-6, C-7, C-8a
8a	156.91 (C)		
1'	125.54 (C)		
2', 6'	106.23 (CH)	7.36 ( <i>s</i> )	C-2, C-1', C-3' (5'), C-4'
3', 5'	153.09 (C)		
4'	140.23 (C)		
3-OCH <sub>3</sub>	60.30 (CH <sub>3</sub> )	3.87 ( <i>s</i> )	C-3
3', 5'-OCH <sub>3</sub>	56.33 (CH <sub>3</sub> )	3.94 ( <i>s</i> )	C-3', C-5'
4'-OCH <sub>3</sub>	60.94 (CH <sub>3</sub> )	3.94 ( <i>s</i> )	C-4'
5-OH		12.62 ( <i>s</i> )	C-4a, C-5, C-6
7-OH		8.67 ( <i>br s</i> )	

Table 25The NMR spectral data of ART17

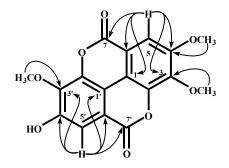
#### 3.1.3 Ellagic acid derivatives

# ART10: 3,3',4-Tri-O-methylellagic acid

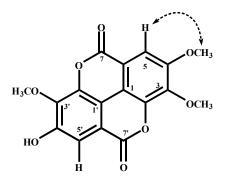


ART10 was obtained as a vellow solid, mp 298-300 °C. The UV spectrum revealed maximum absorption bands at 247 and 372 nm. The IR spectrum displayed absorption bands for a hydroxyl group (3410 cm<sup>-1</sup>) and conjugated carbonyl group (1656 cm<sup>-1</sup>), suggesting the presence of lactone functionallity. It was supported from the lactone carbonyls at  $\delta_{\rm C}$  159.06 and 158.74 in the <sup>13</sup>C NMR spectrum (**Table 26**). The <sup>13</sup>C NMR and DEPT techniques showed 17 carbon signals including two carbonyl carbons ( $\delta_c$  159.06 and 158.74), ten quaternary carbons ( $\delta_c$  154.04, 152.77, 141.81, 141.38, 141.03, 140.74, 114.03, 112.70, 112.01 and 111.68), two methine carbons ( $\delta_{\rm C}$  112.77 and 107.72) and three methyl carbons ( $\delta_{\rm C}$  61.76, 61.54 and 56.77). The <sup>1</sup>H NMR spectrum (Table 26) showed the resonances of two isolated aromatic protons at  $\delta_{\rm H}$  7.68 (s, H-5') and 7.64 (s, H-5), and three methoxyl groups at  $\delta_{\rm H}$  4.19 (3'-OCH<sub>3</sub>), 4.17 (3-OCH<sub>3</sub>) and 4.04 (4-OCH<sub>3</sub>). According to the low field chemical shift of H-5 and H-5', both aromatic protons were then located at the *peri* position to the carbonyl groups. The  ${}^{3}J$  HMBC correlations (Table 26) of H-5 and 3-OCH<sub>3</sub>  $(\delta_{\rm H} 4.17)$  to oxygen-bearing carbon C-3  $(\delta_{\rm C} 141.81)$  and the cross peak of aromatic proton H-5 to the 4-OCH<sub>3</sub> ( $\delta_{\rm H}$  4.04) in the NOESY experiment confirmed the presence of the methoxyl groups at C-3 and C-4. In addition the <sup>3</sup>J HMBC correlations of H-5' and 3'-OCH<sub>3</sub> ( $\delta_{\rm H}$  4.19) to an oxygen-bearing carbon C-3' ( $\delta_{\rm C}$  140.74) suggested the

presence of the methoxyl groups at C-3' rather than C-4'. The assigned structure of **ART10** was identical to a known 3,3',4-tri-*O*-methylellagic acid (**Table 57**) (Khac *et al.*, 1990).



Selected HMBC correlations of ART10



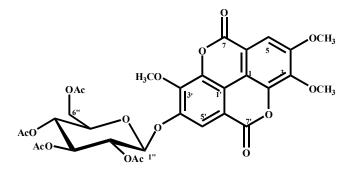
NOESY correlations of ART10

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.)	HMBC correlations
1	112.01 (C)		
2	141.38 (C)		
3	141.81 (C)		
4	154.04 (C)		
5	107.72 (CH)	7.64 ( <i>s</i> )	C-1, C-3, C-4, C-6, C-7
6	114.03 (C)		
7	159.06 (C=O)		
1'	111.68 (C)		
2'	141.03 (C)		
3'	140.74 (C)		
4'	152.77 (C)		
5'	112.77 (CH)	7.68 (s)	C-1', C-3', C-4', C-6', C-7'
6'	112.70 (C)		
7'	158.74 (C=O)		
3-OCH <sub>3</sub>	61.76 (CH <sub>3</sub> )	4.17 ( <i>s</i> )	C-3
4-OCH <sub>3</sub>	56.77 (CH <sub>3</sub> )	4.04 ( <i>s</i> )	C-4
3'-OCH <sub>3</sub>	61.54 (CH <sub>3</sub> )	4.19 ( <i>s</i> )	C-3'

Table 26The NMR spectral data of ART10

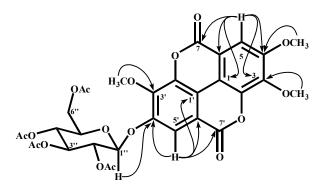
**ART28:** 4-*O*-[β-D-Glucopyranosyl-tetraacetate]-3,3',4'-tri-*O*-methyl-

ellagic acid

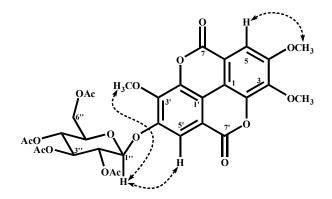


ART28 was obtained as a white powder. The IR spectrum exhibited absorption bands for a carbonyl stretching at 1735 and 1720 cm<sup>-1</sup>. Its <sup>1</sup>H NMR spectrum (Table 27) revealed that ART28 was tri-O-methylellagic acid glycoside. The presence of tri-O-methylellagic acid moiety was evidenced from the signals of two aromatic protons at  $\delta_{\rm H}$  7.92 (H-5) and 7.71 (H-5'), and three methoxyl groups at  $\delta_{\rm H}$  4.22 (3'-OCH<sub>3</sub>), 4.14 (3-OCH<sub>3</sub>) and 4.05 (4'-OCH<sub>3</sub>), as shown for **ART10**. While the presence of a glucopyranoside tetra-acetate moiety was indicated from the signals of an anomeric proton at  $\delta_{\rm H}$  5.17 (H-1"), methylene protons at  $\delta_{\rm H}$  4.26 (H<sub>2</sub>-6"), four methine protons at  $\delta_{\rm H}$  5.41 (H-2") 5.34 (H-3"), 5.16 (H-4") and 4.01 (H-5"), and four acetyl groups at  $\delta_{\rm H}$  2.19 (6"-OAc), 2.11 (4"-OAc), 2.08 (3"-OAc) and 2.06 (2"-OAc). The large coupling constant of the anomeric proton H-1" (J = 7.5 Hz) indicated an axial orientation. Accordingly, the sugar moiety was a  $\beta$ -D-glucopyranosyl tetraacetate. The  ${}^{3}J$  HMBC correlations (Table 27) of the anomeric proton H-1" to oxygen-bearing carbon C-4' ( $\delta_{\rm C}$  155.18) suggested the connection of the sugar moiety at C-4'. This suggestion was confirmed by the NOESY experiment that the anomeric proton H-1" correlated to H-5' ( $\delta_{\rm H}$  7.71) and 3'-OCH<sub>3</sub> ( $\delta_{\rm H}$  4.22). The <sup>13</sup>C NMR and DEPT techniques (Table 27) showed 29 carbon signals for 31 carbon atoms: six carbonyl carbons ( $\delta_{\rm C}$  171.14, 170.34, 169.69, 169.51, 158.82 and 158.61), ten quaternary carbons ( $\delta_{\rm C}$  155.18, 151.72, 143.42, 141.65, 141.38, 141.03, 115.17,

113.23, 112.95 and 112.76), seven methine carbons ( $\delta_{\rm C}$  113.97, 108.26, 100.33, 72.96, 72.60, 71.11 and 68.50), one methylene carbon ( $\delta_{\rm C}$  62.28) and seven methyl carbons [ $\delta_{\rm C}$  62.48, 62.20, 57.10, 20.91, and 20.83 (3xC)]. The structure of **ART28** was determined as a known 4-*O*-[ $\beta$ -D-glucopyranosyl-tetraacetate]-3,3',4'-tri-*O*-methylellagic acid (**Table 58**) (Khac *et al.*, 1990).



Selected HMBC correlations of ART28



NOESY correlations of ART28

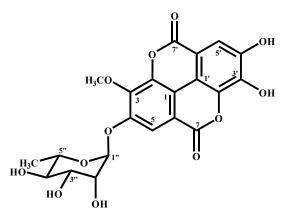
Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1	115.17 (C)		
2	141.38 (C)		
3	143.42 (C)		
4	151.72 (C)		
5	113.97 (CH)	7.92 (s)	C-1, C-3, C-4, C-6, C-7,
			C-1'
6	112.76 (C)		
7	158.61 (C=O)		
1'	113.23 (C)		
2'	141.03 (C)		
3'	141.65 (C)		
4'	155.18 (C)		
5'	108.26 (CH)	7.71 ( <i>s</i> )	C-1, C-1', C-3', C-4', C-6',
			C-7'
6'	112.95 (C)		
7'	158.82 (C=O)		
3-OCH <sub>3</sub>	62.48 (CH <sub>3</sub> )	4.14 ( <i>s</i> )	C-3
3'-OCH <sub>3</sub>	62.20 (CH <sub>3</sub> )	4.22 (s)	C-3'
4'-OCH <sub>3</sub>	57.10 (CH <sub>3</sub> )	4.05 (s)	C-4'
1"	100.33 (CH)	5.17 ( <i>d</i> , 7.5)	C-4, C-2", C-3", C-5"
2"	71.11 (CH)	5.41 ( <i>dd</i> , 9.3, 7.5)	
3"	72.60 (CH)	5.34 ( <i>t</i> , 9.3)	
4"	68.50 (CH)	5.16 ( <i>t</i> , 9.3)	
5"	72.96 (CH)	4.01 ( <i>m</i> )	
6"	62.28 (CH <sub>2</sub> )	4.26 ( <i>m</i> )	

Table 27The NMR spectral data of ART28

 Table 27 (Continued)

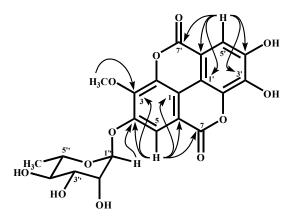
Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
2"-OAc	170.34 (C=O)		
	20.83 (CH <sub>3</sub> )	2.06 ( <i>s</i> )	2"-C=O
3"-OAc	169.69 (C=O)		
	20.83 (CH <sub>3</sub> )	2.08 (s)	3"-C=O
4"-OAc	169.51 (C=O)		
	20.83 (CH <sub>3</sub> )	2.11 (s)	4"-C=O
6"-OAc	171.14 (C=O)		
	20.91 (CH <sub>3</sub> )	2.19 (s)	6"-C=O

### **ART41:** 3-*O*-Methylellagic acid 4-*O*-α-rhamnopyranoside

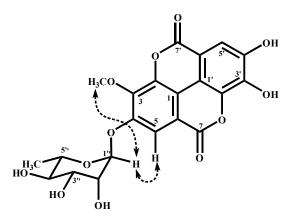


ART41 was a white powder. The IR spectrum showed a stretching of hydroxyl and carbonyl groups at 3349 and 1687 cm<sup>-1</sup>, respectively. The UV spectrum exhibited absorption bands maxima at 207, 253, 274sh, 314 and 350 nm. The <sup>1</sup>H NMR spectral data (Table 28) showed the characteristic signals of ellagic acid derivative, with the resonances of two singlet aromatic protons at  $\delta_{\rm H}$  7.85 (H-5') and 7.62 (H-5) and a methoxyl group at  $\delta_{\rm H}$  4.14 (3-OCH<sub>3</sub>). The presence of rhamnopyranosyl moiety was suggested from the resonances of an anomeric proton at  $\delta_{\rm H}$  5.55, four methine protons at  $\delta_{\rm H}$  4.14, 4.01, 3.61 and 3.47, and a methyl group at  $\delta_{\rm H}$  1.23 as doublet with the coupling constant of 5.4 Hz. The NOESY experiment exhibited the correlations of the anomeric proton H-1" to both of H-5 and 3-OCH<sub>3</sub> indicating that the rhamnose moiety was linked at C-4. The <sup>13</sup>C NMR and DEPT techniques (Table 28) showed 21 carbon signals for 22 carbon atoms including two carbonyl carbons [ $\delta_{\rm C}$  159.02 (2xC)], ten quaternary carbons ( $\delta_{\rm C}$  152.95, 146.79, 141.95, 141.84, 140.57, 136.38, 114.64, 113.22, 112.14 and 107.45), seven methine carbons ( $\delta_{\rm C}$  112.21, 112.14, 100.18, 72.41, 70.79, 70.20 and 70.04), one methylene carbon ( $\delta_{\rm C}$  62.28) and two methyl carbons ( $\delta_{\rm C}$  61.26 and 17.93). The structure of **ART41** could be assigned for three possible known ellagic acid derivatives, 4-O-methylellagic acid 3'-α-rhamnoside, 3-O-methylellagic acid 3'-O- $\alpha$ -rhamnopyranoside and 3-O-methylellagic acid 4-O- $\alpha$ -rhamnopyranoside. Further elucidation was made by comparison to the spectral data of those three related ellagic

acid derivatives (**Tables 59** and **60**) (Elkhateeb *et al.*, 2005; Kim *et al.*, 2001; El-Toumy and Rauwald, 2003). It was found to be closed to that of 3-*O*-methylellagic acid 4-*O*- $\alpha$ -rhamnopyranoside (El-Toumy and Rauwald, 2003). The HMBC experiment (**Table 28**) and the NOESY experiment completely supported the assigned structure.



Selected HMBC correlations of ART41



NOESY correlations of ART41

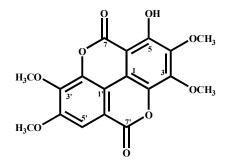
Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1	114.64 (C)		
2	141.95 (C)		
3	140.57 (C)		
4	152.95 (C)		
5	112.21 (CH)	7.62 ( <i>s</i> )	C-1, C-3, C-4, C-6, C-7
6	113.22 (C)		
7	159.02 (C=O)		
1'	112.14 (C)		
2'	136.38 (C)		
3'	141.84 (C)		
4'	146.79 (C)		
5'	112.14 (CH)	7.85 ( <i>s</i> )	C-1', C-3', C-4', C-6', C-7'
6'	107.45 (C)		
7'	159.02 (C=O)		
1"	100.18 (CH)	5.55 ( <i>br</i> s)	C-4, C-2", C-3", C-5"
2"	70.20 (CH)	4.14*	C-1", C-4"
3"	70.79 (CH)	4.01 ( <i>d</i> , 7.8)	C-1", C-2", C-5"
4"	72.41 (CH)	3.47*	C-2", C-3", C-5", C-6"
5"	70.04 (CH)	3.67 ( <i>m</i> )	C-1", C-3", C-4", C-6"
6"	17.93 (CH <sub>3</sub> )	1.23 ( <i>d</i> , 5.4)	C-4", C-6"
OCH <sub>3</sub>	61.26 (CH <sub>3</sub> )	4.14 (s)	C-3'

Table 28The NMR spectral data of ART41

\*overlapped signals

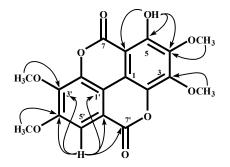
#### 3.1.4 Flavellagic acid derivatives

**ART31:** 3,4,3',4'-Tetra-*O*-methylflavellagic acid

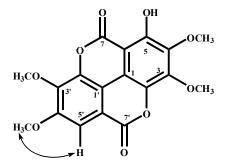


ART31 was obtained as a light yellow amorphous powder. Its molecular formula of  $C_{18}H_{14}O_9$  was established from the HR EI-MS spectrum. The UV spectrum showed the maximum absorption bands at 248, 376, and 410sh nm. The IR spectrum displayed absorption bands of a hydroxyl group at 3336 cm<sup>-1</sup> and conjugated carbonyl group at 1719 cm<sup>-1</sup>. The carbonyl carbon signals at  $\delta_{\rm C}$  163.14 (C-7) and 159.05 (C-7') in the <sup>13</sup>C NMR spectrum (Table 29) suggested the existence of lactone functionalities. The <sup>1</sup>H NMR spectrum (Table 29) exhibited the resonances of a phenolic hydroxyl proton at  $\delta_{\rm H}$  10.36, an aromatic proton at  $\delta_{\rm H}$  7.72, and four methoxyl groups at  $\delta_{\rm H}$  4.30, 4.21, 4.05 and 4.02. The aromatic proton was located at C-5', a peri-position to carbonyl group due to its low field chemical shift value ( $\delta_{\rm H}$  7.72) together with the HMBC correlations (**Table 29**) of it to C-1' ( $\delta_{\rm C}$  113.47), C-3' ( $\delta_{C}$  141.94), C-4' ( $\delta_{C}$  154.91), C-6' ( $\delta_{C}$  108.21) and C-7' ( $\delta_{C}$  159.05). The correlations of the methoxyl groups at  $\delta_{\rm H}$  4.21 to C-3' and at  $\delta_{\rm H}$  4.05 to C-4' together with the enhancement of the methoxyl group at  $\delta_{\rm H}$  4.05 by irradiation at the resonance of H-5' in the NOE experiment indicated that the methoxyl groups at  $\delta_{\rm H}$  4.21 and 4.05 were then placed at the C-3' and C-4', respectively. The evidences that the hydroxyl proton ( $\delta_{\rm H}$  10.36) showed the HMBC correlations to C-4 ( $\delta_{\rm C}$  141.38), C-5 ( $\delta_{\rm C}$  153.14) and C-6 ( $\delta_{\rm C}$  96.88) as well as the methoxyl group which resonated at  $\delta_{\rm H}$  4.02 showed correlation to C-4 ( $\delta_{\rm C}$  141.38) suggesting that the hydroxyl group was at a peri

position to C=O (C-5), and the methoxy group was at C-4. The remaining methoxy group ( $\delta_{\rm H}$  4.30) was consequently attributed to the position C-3. The <sup>13</sup>C NMR resonances were in good / positive agreement with the assigned structure, 3,4,3',4'-tetra-*O*-methylflavellagic acid. It was a new flavellagic acid derivative.



Selected HMBC correlations of ART31

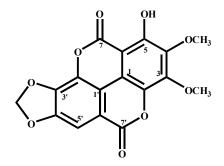


NOE correlations of ART31

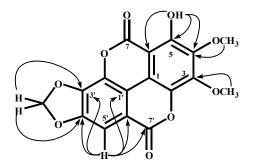
Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.)	HMBC correlations
1	112.42 (C)		
2	140.67 (C)		
3	148.40 (C)		
4	141.38 (C)		
5	153.14 (C)		
6	96.88 (C)		
7	163.14 (C=O)		
1'	113.47 (C)		
2'	140.67 (C)		
3'	141.94 (C)		
4'	154.91 (C)		
5'	108.21 (CH)	7.72 ( <i>s</i> )	C-1', C-3', C-4', C-6', C-7'
6'	113.54 (C)		
7'	159.05 (C=O)		
5-ОН		10.36 ( <i>s</i> )	C-4, C-5, C-6
3-OCH <sub>3</sub>	62.12 (CH <sub>3</sub> )	4.30 ( <i>s</i> )	C-3
4-OCH <sub>3</sub>	61.63 (CH <sub>3</sub> )	4.02 (s)	C-4
3'-OCH <sub>3</sub>	62.02 (CH <sub>3</sub> )	4.21 ( <i>s</i> )	C-3'
4'-OCH <sub>3</sub>	56.87 (CH <sub>3</sub> )	4.05 (s)	C-4'

Table 29The NMR spectral data of ART31

<sup>a</sup>recorded in CDCl<sub>3</sub>; 300 MHz for <sup>1</sup>H and 125 MHz for <sup>13</sup>C



**ART25** was obtained as a pale yellow amorphous powder. The IR spectrum exhibited absorptions for a hydroxyl group at 3302 cm<sup>-1</sup> and conjugated carbonyl group at 1656 cm<sup>-1</sup> whereas the UV spectrum displayed maximum absorptions at 248, 307, 320, 365 and 380 nm. Its <sup>1</sup>H NMR spectrum (**Table 30**) showed the resonances of a chelated hydroxyl proton ( $\delta_{\rm H}$  10.39, 5-OH), an aromatic proton ( $\delta_{\rm H}$  7.64, H-5'), two methoxyl groups ( $\delta_{\rm H}$  4.29, 3-OCH<sub>3</sub> and 4.02, 4-OCH<sub>3</sub>) and a dioxymethylene protons ( $\delta_{\rm H}$  6.30). The <sup>13</sup>C NMR and HMBC data (**Table 30**) revealed that **ART25** was an ellagic acid which comprised of 5-OH, 3-OCH<sub>3</sub>, 4-OCH<sub>3</sub> and H-5' as for **ART31**. Furthermore the HMBC experiment demonstrated that the aromatic proton H-5' ( $\delta_{\rm H}$  7.64) and the dioxymethylene protons ( $\delta_{\rm H}$  6.30) both exhibited the correlations to the carbons which resonated at  $\delta_{\rm C}$  151.38 (C-4') and 138.60 (C-3') indicating that the dioxymethylene group connected with C-3' and C-4'. The <sup>13</sup>C NMR resonances were in good agreement with the assigned structure. **ART25** was a known flavellagic acid derivative, 3',4'-dioxymethylene-3,4-di-*O*-methylflavellagic acid.



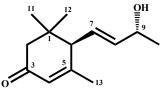
Selected HMBC correlations of ART25

Table 30	The NMR	spectral	data	of ART25
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Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.)	HMBC correlations
1	113.79 (C)		
2	*		
3	148.50 (C)		
4	141.75 (C)		
5	153.35 (C)		
6	96.65 (C)		
7	162.30 (C=O)		
1'	116.35 (C)		
2'	*		
3'	138.60 (C)		
4'	151.38 (C)		
5'	104.01 (CH)	7.64 ( <i>s</i> )	C-1', C-3', C-4', C-6', C-7'
6'	116.35 (C)		
7'	158.80 (C=O)		
5-ОН		10.39 ( <i>s</i> )	C-4, C-5, C-6
3-OCH <sub>3</sub>	62.08 (CH <sub>3</sub> )	4.29 ( <i>s</i> )	C-3
4-OCH <sub>3</sub>	61.60 (CH <sub>3</sub> )	4.03 (s)	C-4
-OCH <sub>2</sub> O-	104.01 (CH <sub>2</sub> )	6.30 ( <i>s</i> )	C-3', C-4'

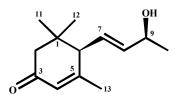
not observed

ART12: (6R,7E,9S)-9-Hydroxy-4,7-megastigmadien-3-one

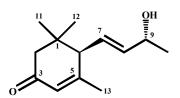


ART12 was a vellowish gum. The IR spectrum exhibited absorptions for a hydroxyl and conjugated carbonyl groups at 3406 and 1656 cm<sup>-1</sup>, respectively, whereas the UV spectrum displayed maximum absorption at 246 nm. The <sup>1</sup>H NMR spectrum (Table 31) showed resonances of geminal methyl groups H<sub>3</sub>-11 and H<sub>3</sub>-12 at  $\delta_{\rm H}$  1.04 (s) and 0.98 (s), non-equivalent methylene protons H<sub>2</sub>-2 at  $\delta_{\rm H}$  2.34 and 2.08 (each d, J = 16.5 Hz), an olefinic proton H-4 at  $\delta_{\rm H} 5.90$  (s) and a vinylic methyl protons H<sub>3</sub>-13 at  $\delta_{\rm H}$  1.90 (d, J = 1.2 Hz.). The <sup>3</sup>J HMBC correlations (**Table 31**) of H<sub>3</sub>-11 and H<sub>3</sub>-12 to C-2 ( $\delta_{\rm C}$  47.41) and C-6 ( $\delta_{\rm C}$  55.37), and of H<sub>3</sub>-13 to C-4 ( $\delta_{\rm C}$  125.86) and C-6 ( $\delta_{\rm C}$  55.37) suggested that the geminal methyl groups H<sub>3</sub>-11 and H<sub>3</sub>-12 were connected to C-1 ( $\delta_{\rm C}$  36.10) whereas a vinylic methyl group H<sub>3</sub>-13 was connected to C-5 ( $\delta_{\rm C}$  161.78). The presence of hydroxybutenyl side chain was deduced from the <sup>1</sup>H-<sup>1</sup>H COSY experiment. The result indicated the resonances at  $\delta_{\rm H}$  5.68 (dd, J = 15.3 and 6.3 Hz) and 5.55 (dd, J = 15.3 and 8.7 Hz), 4.35 (quin, J = 6.3 Hz) and 1.29 (d, J = 6.3 Hz,), corresponded to *trans* olefinic protons H-8 and H-7, methine proton H-9 and methyl group H<sub>3</sub>-10, respectively. The  ${}^{3}J$  HMBC correlation of H-7 to C-1 and C-5 indicated that the side chain was connected to C-2.

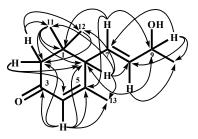
By comparison of the <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic data with the related compounds in the literatures (**Table 61**) (D' Abrosca *et al.*, 2004; Cutillo *et al.*, 2005), compound **ART12** was identical to a C-13 nor-terpene, (6R,7E,9S)-9-hydroxy-4,7-megastigmadien-3-one (Cutillo *et al.*, 2005).



(6R,7E,9R)-9-hydroxy-4,7-megastigmadien-3-one



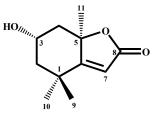
(6R,7E,9S)-9-hydroxy-4,7-megastigmadien-3-one



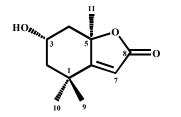
Selected HMBC correlations of ART12

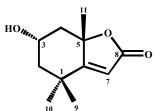
Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
1	36.10 (C)		
2	47.41 (CH <sub>2</sub> )	2.34 ( <i>d</i> ; 16.5)	C-4, C-6, C-11, C-12
		2.08 ( <i>d</i> ; 16.5)	
3	199.14 (C=O)		
4	125.86 (CH)	5.90 (s)	C-2, C-3, C-5, C-6, C-13
5	161.78 (C)		
6	55.37 (CH)	2.53 ( <i>d</i> ; 8.7)	C-1, C-2, C-4, C-5, C-7,
			C-8, C-11, C-12, C-13
7	126.59 (CH)	5.55 ( <i>dd</i> ; 15.3, 8.7)	C-1, C-6, C-8, C-9
8	138.43 (CH)	5.68 ( <i>dd</i> ; 15.3, 6.3)	C-6, C-7, C-9, C-10
9	68.32 (CH)	4.35 (quin; 6.3)	C-7, C-8, C-10
10	23.61 (CH <sub>3</sub> )	1.29 ( <i>d</i> ; 6.3)	C-8, C-9
11	27.85 (CH <sub>3</sub> )	1.04 ( <i>s</i> )	C-2, C-6, C-12
12	27.11 (CH <sub>3</sub> )	0.98 (s)	C-2, C-6, C-11
13	23.45 (CH <sub>3</sub> )	1.90 ( <i>d</i> ; 1.2)	C-4, C-5, C-6

Table 31The NMR spectral data of ART12



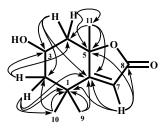
ART14 was obtained as a yellow gum. The IR spectrum showed absorptions at 3449 cm<sup>-1</sup> for a hydroxyl group and 1718 cm<sup>-1</sup> for a carbonyl group. The UV spectrum exhibited maximum absorptions at 247 and 255sh nm. The <sup>1</sup>H NMR spectrum (Table 32) displayed the signals of three methyl groups at  $\delta_{\rm H}$  1.78, 1.47 and 1.28, and an olefinic proton at  $\delta_{\rm H}$  5.70, which were assigned for H<sub>3</sub>-11, H<sub>3</sub>-9, H<sub>3</sub>-10 and H-7, respectively. The spectrum further showed a multiplet of an oxygenated methine proton H-3 ( $\delta_{\rm H}$  4.34), a doublet of doublet and doublet of triplet of nonequivalent methylene protons H<sub>2</sub>-2 ( $\delta_{\rm H}$  1.98 and 1.53) and a doublet of doublet and doublet of triplet of non-equivalent methylene protons H<sub>2</sub>-4 ( $\delta_{\rm H}$  2.46 and 1.79) indicating the presence of -CH<sub>2</sub>CH(OH)CH<sub>2</sub>- unit as evidenced from the <sup>1</sup>H-<sup>1</sup>H COSY data (Table 32). The HMBC correlations (Table 32) of H<sub>3</sub>-9 and H<sub>3</sub>-10 to C-1, C-2 and C-6 and of H<sub>3</sub>-11 to C-4, C-5 and C-6 established the six membered-ring subunit. The HMBC correlations of H-7 to C-1, C-5, C-6 and C-8 supported that the olefinic proton was at C-7. Consequently, the possible structure could be loliolide or isololiolide, two known diastereomers. In comparison, the spectral data of ART14 was close to those of loliolide rather than isololiolide (Kimura and Maki, 2002; Hattab et al., 2008). ART14 was therefore elucidated to be loliolide (Table 62) (Kimura and Maki, 2002; Hattab et al., 2008).





loliolide

isololiolide



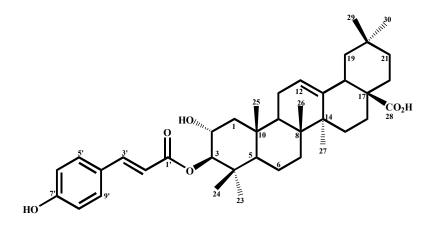
Selected HMBC correlations of ART14

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	COSY	HMBC correlations
1	36.00 (C)			
2	47.35 (CH <sub>2</sub> )	α: 1.98 ( <i>td</i> ;	H-2 <i>β</i> , H-3	C-1, C-3, C-4, C-6
		15.0, 2.5)		
		β: 1.53 (dd;		
		15.0, 3.5)		
3	66.87 (CH)	4.34 (quin; 3.0)	H-2 <i>α</i> , H-2 <i>β</i> ,	C-1, C-3, C-4, C-6,
			H-4 <i>α</i> , H-4 <i>β</i>	C-10
4	45.66 (CH <sub>2</sub> )	2.46 ( <i>td</i> ; 14.0,	H-4 <i>β</i> , H-3	C-2, C-3, C-5, C-6
		2.5)	H-4 <i>α</i> , H-3	C-3, C-11
		1.79 ( <i>dd</i> ; 14.0,		
		4.0)		
5	86.55 (C)			
6	182.45 (C)			
7	112.97 (CH)	5.70 (s)		C-1, C-5, C-6, C-8
8	172.01 (C=O)			
9	26.05 (CH <sub>3</sub> )	1.47 (s)		C-1, C-2, C-6, C-10
10	30.64 (CH <sub>3</sub> )	1.28 (s)		C-1, C-2, C-6, C-9
11	27.02 (CH <sub>3</sub> )	1.78 (s)		C-4, C-5, C-6

Table 32The NMR spectral data of ART14

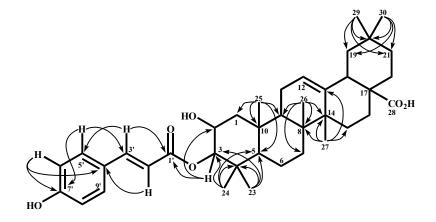
# **ART22:** $2\alpha$ -Hydroxy- $3\beta$ -*E*-coumaroyloxy-olean-12-en-28-oic acid

 $(3\beta$ -O-E-coumaroylmaslinic acid)



ART22 was obtained as a white solid. The IR spectrum showed absorption bands for a hydroxyl group at 3359 cm<sup>-1</sup> and for a carbonyl group at 1696 cm<sup>-1</sup>. The UV spectrum displayed maximum absorptions at 211, 229 and 315 nm. The <sup>1</sup>H NMR spectral data (Table 33) exhibited the presence of a *p*-coumaroyl moiety from the signals at  $\delta_{\rm H}$  7.65 (d, J = 15.9 Hz, H-3'), 7.41 (d, J = 8.1 Hz, H-5' and H-8'), 6.82 (d, J = 8.1 Hz, H-6' and H-9') and 6.33 (d, J = 15.9 Hz, H-2'). The large coupling constant of 15.9 Hz indicated E-configuration of H-2' and H-3'. The spectrum further showed the signals due to one olefinic proton ( $\delta_{\rm H}$  5.29, H-12), three methine protons ( $\delta_{\rm H}$  4.62, H-3; 3.86, H-2 and 2.84, H-18) and seven quaternary methyl groups  $(\delta_{\rm H}$  1.15, H<sub>3</sub>-27; 1.02, H<sub>3</sub>-25; 0.94, H<sub>3</sub>-23 and H<sub>3</sub>-30; 0.93, H<sub>3</sub>-24; 0.92, H<sub>3</sub>-29 and 0.78, H<sub>3</sub>-26) as well as the presence of the carboxyl carbon at  $\delta_{\rm C}$  180.66 in the <sup>13</sup>C NMR spectrum (Table 33) which were typical resonances of maslinic acid derivative. The doublet signal of an oxymethine proton H-3 ( $\delta_{\rm H}$  4.62) with the coupling constant of 9.9 Hz indicated the  $\alpha$ -axial orientation. Consequently, the oxymethine proton H-2 (dt, J = 9.9 and 5.1 Hz) was located at the  $\beta$ -axial position. The *p*-coumaroyl moiety at the C-3  $\beta$ -equatorial position was implied from the down

field shift of the oxygenated methine proton H-3 ( $\delta_{\rm H}$  4.62). This conclusion was supported by the  ${}^{3}J$  HMBC correlation (Table 33) of the H-3 to the carbonyl ester C-1' ( $\delta_{\rm C}$  168.85). The <sup>13</sup>C NMR spectrum revealed the resonance signals of two carbonyl carbons ( $\delta_{\rm C}$  180.66 and 168.85), nine quaternary carbons ( $\delta_{\rm C}$  159.18, 143.88, 126.17, 46.35, 41.75, 39.53, 39.30, 38.14 and 30.64), twelve methine carbons  $[\delta_{\rm C}$  145.42, 130.01 (2xC), 122.05, 115.83 (2xC), 114.65, 84.82, 67.36, 55.11, 47.53 and 41.17], nine methylene carbons [ $\delta_{\rm C}$  47.45, 45.90, 33.85, 32.49 (2xC), 27.62, 23.43, 22.98 and 18.30] and seven methyl carbons ( $\delta_{\rm C}$  33.01, 28.52, 25.83, 23.51, 17.71, 16.84 and 16.42). The spectral data and comparison to the previous report (Table 63) (Yagi *et al.*, 1978) indicated that ART22 was a known  $2\alpha$ -hydroxy- $3\beta$ -Ecoumaroyloxyolean-12-en-28-oic acid or  $3\beta$ -O-E-coumaroylmaslinic acid. The assigned structure was fully confirmed by the HMBC correlations.



Selected HMBC correlations of ART22

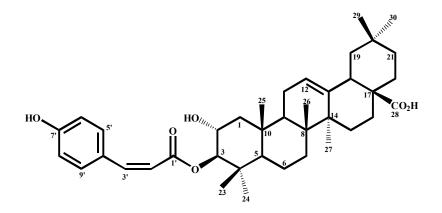
Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
1	47.45 (CH <sub>2</sub> )	2.08 (m), 1.05 (m)	
2	67.36 (CH)	3.86 ( <i>dt</i> ; 9.9, 5.1)	
3	84.82 (CH)	4.62 ( <i>d</i> ; 9.9)	C-2, C-1'
4	39.53 (C)		
5	55.11 (CH)	0.89 ( <i>m</i> )	
6	18.30 (CH <sub>2</sub> )	2.17 ( <i>m</i> ), 1.58 ( <i>m</i> )	
7	32.49 (CH <sub>2</sub> )	1.58 ( <i>m</i> )	
8	39.30 (C)		
9	47.53 (CH)	1.60 ( <i>m</i> )	
10	38.14 (C)		
11	22.98 (CH <sub>2</sub> )	1.98 ( <i>m</i> )	
12	122.05 (CH)	5.29 ( <i>br</i> s)	
13	143.88 (C)		
14	41.75 (C)		
15	27.62 (CH <sub>2</sub> )	1.26 ( <i>m</i> ), 1.08 ( <i>m</i> )	
16	23.43 (CH <sub>2</sub> )	1.90 ( <i>m</i> ), 1.65 ( <i>m</i> )	
17	46.35 (C)		
18	41.17 (CH)	2.84 ( <i>br d</i> ; 12.6)	
19	45.90 (CH <sub>2</sub> )	1.64 ( <i>m</i> )	
20	30.64 (C)		
21	33.85 (CH <sub>2</sub> )	1.80 ( <i>m</i> ), 1.58 ( <i>m</i> )	
22	32.49 (CH <sub>2</sub> )	2.18 ( <i>m</i> )	
23	28.52 (CH <sub>3</sub> )	0.94 (s)	C-3, C-4, C-5, C-24
24	17.71 (CH <sub>3</sub> )	0.93 (s)	C-3, C-4, C-5, C-23
25	16.42 (CH <sub>3</sub> )	1.02 (s)	C-1, C-5, C-9, C-10
26	16.84 (CH <sub>3</sub> )	0.78 (s)	C-7, C-8, C-9, C-14
27	25.83 (CH <sub>3</sub> )	1.15 (s)	C-8, C-13, C-14, C-15

Table 33 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
28	180.66 (C=O)		
29	33.01 (CH <sub>3</sub> )	0.92 (s)	C-19, C-20, C-21, C-30
30	23.51 (CH <sub>3</sub> )	0.94 ( <i>s</i> )	C-19, C-20, C-21, C-29
1'	168.85 (C=O)		
2'	114.65 (CH)	6.33 ( <i>d</i> ; 15.9)	C-4'
3'	145.42 (CH)	7.65 ( <i>d</i> ; 15.9)	C-1', C-5'
4'	126.17 (C)		
5', 9'	130.01 (CH)	7.41 ( <i>d</i> ; 8.1)	C-3', C-7'
6', 8'	115.83 (CH)	6.82 ( <i>d</i> ; 8.1)	C-4', C-7'
7'	159.18 (C)		

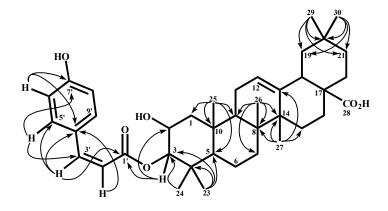
## **ART23:** $2\alpha$ -Hydroxy- $3\beta$ -Z-coumaroyloxy-olean-12-en-28-oic acid

 $(3\beta$ -O-Z-coumaroylmaslinic acid)



ART23 was obtained as a white solid. The IR spectrum exhibited absorptions at 3340 and 1697 cm<sup>-1</sup> for hydroxyl and carbonyl groups, respectively. The UV spectrum displayed absorption bands maxima at 205, 227 and 312 nm. The <sup>1</sup>H NMR spectrum (Table 34) showed the signals corresponding to a p-coumaroyl moiety  $(\delta_{\rm H} 6.90, d, J = 12.0 \text{ Hz}, \text{H-3'}; 7.65, d, J = 8.0 \text{ Hz}, \text{H-5'} \text{ and } \text{H-8'}; 6.80, d, J = 8.0 \text{ Hz},$ H-6' and H-9'; 5.91, d, J = 12.0 Hz, H-2'), an olefinic proton ( $\delta_{\rm H}$  5.32, H-12). Three methine protons ( $\delta_{\rm H}$  4.58, H-3; 3.67, H-2 and 2.84, H-18) and seven quaternary methyl groups ( $\delta_{\rm H}$  1.15, H<sub>3</sub>-27; 0.94, H<sub>3</sub>-23 and H<sub>3</sub>-30; 0.93, H<sub>3</sub>-24; 0.92, H<sub>3</sub>-29; 0.81, H<sub>3</sub>-25; and 0.78, H<sub>3</sub>-26) which belonged to a maslinic acid derivative were observed. These signals were similar to those of ART22. The slightly difference of the high-field chemical shift and the small coupling constant of olefinic protons H-2'/H-3' ( $\delta_{\rm H}$  6.90 and 5.91, J = 12.0 Hz) indicated the Z-conformation of the double bond. The <sup>13</sup>C NMR spectrum (Table 34) revealed the presence of forty carbons including two carbonyl carbons ( $\delta_{\rm C}$  180.94 and 167.65), nine quaternary carbons  $(\delta_{\rm C}$  156.87, 142.66, 129.12, 46.35, 40.72, 40.15, 38.39, 38.14 and 30.64), twelve methine carbons ( $\delta_{\rm C}$  143.23, 131.49, 121.46, 114.96, 114.00, 83.85, 66.71, 54.22,

47.53 and 40.15), nine methylene carbons ( $\delta_{\rm C}$  47.45, 45.90, 33.85, 32.49, 26.67, 23.43, 22.98 and 18.30) and seven methyl carbons ( $\delta_{\rm C}$  33.01, 29.74, 24.96, 23.51, 17.36, 16.84 and 16.42). The spectral data and comparison with the previous report (**Table 64**) (Yagi *et al.*, 1978) indicated that compound **ART23** was a known  $2\alpha$ -hydroxy-3 $\beta$ -Z-coumaroyloxy-olean-12-en-28-oic acid or  $3\beta$ -O-Z-coumaroyl-maslinic acid. The assigned structure was fully confirmed by the HMBC correlations (**Table 34**).



Selected HMBC correlations of ART23

 Table 34
 The NMR spectral data of ART23

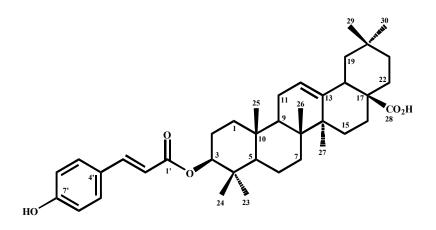
Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
1	46.65 (CH <sub>2</sub> )	2.06 ( <i>m</i> ), 1.01 ( <i>m</i> )	
2	66.71 (CH)	3.85 ( <i>m</i> )	
3	83.85 (CH)	4.58 ( <i>d</i> ; 9.6)	C-2, C-4, C-23, C-24, C-1'
4	40.15 (C)		
5	54.22 (CH)	0.90 ( <i>m</i> )	
6	16.84 (CH <sub>2</sub> )	2.17 ( <i>m</i> ), 1.56 ( <i>m</i> )	
7	31.52 (CH <sub>2</sub> )	1.58 ( <i>m</i> )	
8	38.39 (C)		
9	46.65 (CH)	1.60 ( <i>m</i> )	
10	37.24 (C)		
11	22.52 (CH <sub>2</sub> )	1.94 ( <i>m</i> )	
12	121.46 (CH)	5.32 ( <i>br</i> s)	
13	142.66 (C)		
14	40.72 (C)		
15	26.67 (CH <sub>2</sub> )	1.26 ( <i>m</i> ), 1.08 ( <i>m</i> )	
16	27.63 (CH <sub>2</sub> )	1.91 ( <i>m</i> ), 1.60 ( <i>m</i> )	
17	45.51 (C)		
18	40.15 (CH)	2.83 (br d; 12.0)	
19	44.85 (CH <sub>2</sub> )	1.64 ( <i>m</i> )	
20	30.98 (C)		
21	32.11 (CH <sub>2</sub> )	1.83 ( <i>m</i> ), 1.55 ( <i>m</i> )	
22	31.52 (CH <sub>2</sub> )	2.18 ( <i>m</i> )	
23	17.36 (CH <sub>3</sub> )	0.97 ( <i>s</i> )	C-3, C-4,C-5
24	29.74 (CH <sub>3</sub> )	0.94 (s)	C-3, C-4,C-5
25	16.64 (CH <sub>3</sub> )	0.81 (s)	C-1, C-5, C-9, C-10
26	15.63 (CH <sub>3</sub> )	0.78 (s)	C-7, C-8, C-9, C-14
27	24.96 (CH <sub>3</sub> )	1.15 (s)	C-8, C-13, C-14, C-15

<sup>a</sup>recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop)

Table 34 (Continued)

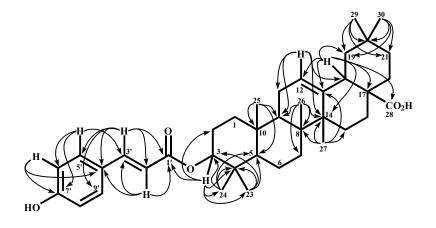
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
28	180.94 (C=O)		
29	32.84 (CH <sub>3</sub> )	0.91 (s)	C-19, C-20, C-21
30	22.62 (CH <sub>3</sub> )	0.93 (s)	C-19, C-20, C-21
1'	167.65 (C=O)		
2'	114.96 (CH)	5.91 ( <i>d</i> ; 12.0)	C-1', C-3', C-4'
3'	143.23 (CH)	6.90 ( <i>d</i> ; 12.0)	C-1', C-2', C-4', C-5' (C-9')
4'	129.12 (C)		
5', 9'	131.49 (CH)	7.65 ( <i>d</i> ; 8.0)	C-3', C-4', C-6' (C-8'), C-7'
6', 8'	114.00 (CH)	6.80 ( <i>d</i> ; 8.0)	C-4', C-5' (C-9'), C-7', C-4'
7'	156.87 (C)		

<sup>*a*</sup>recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop)



ART32 was obtained as a white solid, mp 240-242 °C. The IR spectrum exhibited absorptions at 3345 cm<sup>-1</sup> for a hydroxyl group and at 1700 cm<sup>-1</sup> for a carbonyl group. The UV spectrum displayed absorption bands maxima at 230 and 315 nm. The <sup>1</sup>H NMR spectral data (Table 35) showed the presence of a p-coumaroyl moiety from the resonances at  $\delta_{\rm H}$  7.60 (*d*, *J* = 15.9 Hz, H-3'), 7.42 (*d*, *J* = 8.7 Hz, H-5' and H-8'), 6.81 (d, J = 8.7 Hz, H-6' and H-9') and 6.28 (d, J = 15.9 Hz, H-2'). The Egeometry of p-coumaroyl moiety was assigned from the large coupling constant of olefinic protons H-2' and H-3' (J = 15.9 Hz). The spectrum further displayed the signals due to one olefinic proton ( $\delta_{\rm H}$  5.28, H-12), one methine proton ( $\delta_{\rm H}$  2.85, H-18) and seven quaternary methyl groups ( $\delta_{\rm H}$  1.16, H<sub>3</sub>-27; 0.97, H<sub>3</sub>-25; 0.94, H<sub>3</sub>-24 and H<sub>3</sub>-30; 0.91, H<sub>3</sub>-23 and H<sub>3</sub>-29 and 0.80, H<sub>3</sub>-26) which were typical resonances of oleanolic acid. The oxygenated methine proton at  $\delta_{\rm H}$  4.61 was assigned to be the H-3 position. The splitting pattern and coupling constant value of H-3 (dd, J = 10.7 and 5.3 Hz) indicated that this proton was located at the  $\alpha$ -axial orientation. The location of p-coumaroyl moiety at the C-3  $\beta$ -equatorial position of oleanolic acid implied from the down field shift of the oxygenated methine proton H-3. This conclusion was supported by the  ${}^{3}J$  HMBC correlation (Table 35) of the H-3 with the carbonyl ester

C-1' ( $\delta_{\rm C}$  171.79) of the *p*-coumaroyl moiety. The <sup>13</sup>C NMR spectrum (**Table 35**) revealed the signals of ester and carboxylic carbonyl functionalities at  $\delta_{\rm C}$  171.79 and 184.94, respectively. The remaining resonances were existence of nine quaternary carbons [ $\delta_{\rm C}$  163.13, 147.76, 130.01, 50.31, 45.62, 43.16 (2xC), 40.83 and 34.53), eleven methine carbons ( $\delta_{\rm C}$  148.63, 133.81, 126.12, 119.69, 119.03, 84.94, 59.24, 51.44 and 45.12), ten methylene carbons ( $\delta_{\rm C}$  49.84, 41.82, 37.74, 36.54, 36.44, 31.54, 27.51, 27.37, 26.91 and 22.10) and seven methyl carbons [ $\delta_{\rm C}$  36.89, 31.93, 29.70, 27.37, 20.69 (2xC) and 19.20]. The structure of **ART32** was identified as 3*β*-*O*-*E*-coumaroyloleanolic acid (**Table 65**) (Takahashi *et al.*, 1999). The HMBC experiment fully confirmed the assigned structure. Therefore **ART32** was the 3*β*-*O*-*E*-coumaroylate of **ART39** or oleanolic acid.



Selected HMBC correlations of ART32

Position	$\delta_{\rm C}$ (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	37.74 (CH <sub>2</sub> )	1.34 ( <i>m</i> ), 1.19 ( <i>m</i> )	
2	27.51 (CH <sub>2</sub> )	1.69 ( <i>m</i> )	
3	84.94 (CH)	4.61 ( <i>dd</i> ; 10.7, 5.3)	C-2, C-4, C-23, C-24, C-1'
4	43.16 (C)		
5	59.24 (CH)	0.89 ( <i>m</i> )	
6	22.10 (CH <sub>2</sub> )	1.85 ( <i>m</i> ), 1.40 ( <i>m</i> )	
7	36.44 (CH <sub>2</sub> )	1.77 ( <i>m</i> )	
8	43.16 (C)		
9	51.44 (CH)	1.60 ( <i>m</i> )	
10	40.83 (C)		
11	26.91 (CH <sub>2</sub> )	1.96 ( <i>m</i> ), 1.60 ( <i>m</i> )	
12	126.12 (CH)	5.28 ( <i>t</i> ; 3.1)	C-9, C-11, C-14, C-18
13	147.76 (C)		
14	45.62 (C)		
15	31.54 (CH <sub>2</sub> )	1.73 <i>(m)</i>	
16	27.37 (CH <sub>2</sub> )	1.89 ( <i>m</i> ), 1.69 ( <i>m</i> )	
17	50.31 (C)		
18	45.12 (CH)	2.85 ( <i>dd</i> ; 13.8, 3.9)	C-12, C-13, C-14, C-17, C-19,
			C-28
19	49.84 (CH <sub>2</sub> )	1.65 ( <i>m</i> ), 1.15 ( <i>m</i> )	
20	34.53 (C)		
21	41.82 (CH <sub>2</sub> )	1.68 ( <i>m</i> )	
22	36.54 (CH <sub>2</sub> )	1.55 ( <i>m</i> ), 1.31 ( <i>m</i> )	
23	31.93 (CH <sub>3</sub> )	0.91 (s)	C-3, C-5, C-24
24	20.69 (CH <sub>3</sub> )	0.94 (s)	C-3, C-5, C-23
25	19.20 (CH <sub>3</sub> )	0.97 (s)	C-5, C-9, C-10

<sup>*a*</sup>recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop)

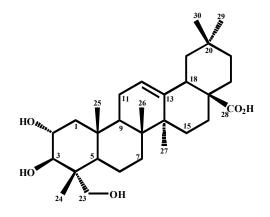
Table 35 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
26	20.69 (CH <sub>3</sub> )	0.80 (s)	C-7, C-8, C-9, C-14
27	29.70 (CH <sub>3</sub> )	1.16 (s)	C-8, C-13, C-14, C-15
28	184.94 (C=O)		
29	36.89 (CH <sub>3</sub> )	0.91 (s)	C-19, C-20, C-21, C-30
30	27.37 (CH <sub>3</sub> )	0.94 (s)	C-19, C-20, C-21, C-29
1'	171.79 (C=O)		
2'	119.03 (CH)	6.28 ( <i>d</i> ; 15.9)	C-1', C-3', C-4'
3'	148.63 (CH)	7.60 ( <i>d</i> ; 15.9)	C-1', C-2', C-4', C-5', C-9'
4'	130.01 (C)		
5', 8'	133.81 (CH)	7.42 ( <i>d</i> ; 8.7)	C-3', C-6', C-7', C-8'
6', 9'	119.69 (CH)	6.81 ( <i>d</i> ; 8.7)	C-4', C-7'
7'	163.13 (C)		

<sup>a</sup>recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop)

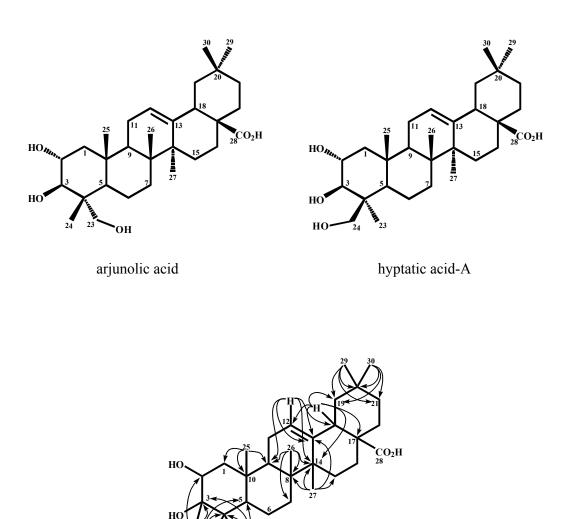
# **ART34:** $2\alpha$ , $3\beta$ , 23-Trihydroxyolean-12-en-28-oic acid

(Arjunolic acid)



ART34 was obtained as a white solid. The IR spectrum showed the presence of a hydroxyl and carboxyl group at 3422 and 1765 cm<sup>-1</sup>, respectively. The <sup>13</sup>C NMR spectrum (Table 36) confirmed the presence of a carbonyl carbon by the resonance at  $\delta_{\rm C}$  181.08. The <sup>1</sup>H NMR spectrum (**Table 36**) revealed the resonances of six tertiary methyl groups ( $\delta_{\rm H}$  1.24, 1.17, 1.00, 0.95, 0.91 and 0.82), an olefinic proton ( $\delta_{\rm H}$  5.20, H-12), four carbinolic protons ( $\delta_{\rm H}$  3.64, H-2; 3.46 and 3.26, H<sub>2</sub>-23 and 3.27, H-3) and a methine proton ( $\delta_{\rm H}$  2.76, H-18). These data indicated that ART34 was also a maslinic acid derivative. The large coupling constant (J = 9.3 Hz) between H-2 and H-3 indicated the axial-axial coupling, so both hydroxyl groups were equatorially oriented. The presence of a hydroxymethyl group attached to C-4 was evidenced from the HMBC correlations (Table 36) of  $H_2$ -23 to C-3, C-4 and C-24. These data corresponded to two possible isomers, hyptatic acid-A or arjunolic acid. The difference between both structures was the orientation of the hydroxymethyl group. In comparison with the literatures of both compounds (Lee *et al.*, 2008; Tripathi et al., 1992; Shao et al., 1995), especially the carbon resonances for C-3, C-23 and C-24 (arjunolic acid: 78.1, 66.3, 14.2; hyptatic acid-A: 84.5, 27.4, 64.7, respectively) (Table 66) and the optical rotation, the spectroscopic data and physical

properties of **ART34** was closer to that of arjunolic acid rather than hyptatic acid-A. Therefore **ART34** was identified to be  $2\alpha$ ,  $3\beta$ , 23-trihydroxyolean-12-en-28-oic acid or arjunolic acid (Tripathi *et al.*, 1992; Shao *et al.*, 1995).



Selected HMBC correlations of ART34

он

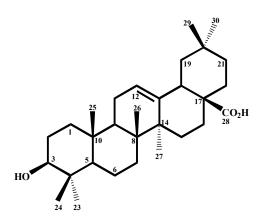
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	45.88 (CH <sub>2</sub> )	1.88 ( <i>dd</i> ; 12.6, 3.9),	
		0.81 ( <i>m</i> )	
2	68.35 (CH)	3.64 (obscure)	
3	78.86 (CH)	3.27 ( <i>d</i> ; 9.3)	C-2, C-4, C-23, C-24
4	42.55 (C)		
5	48.07 (CH)	1.55 ( <i>m</i> )	
6	18.09 (CH <sub>2</sub> )	1.31 ( <i>m</i> )	
7	32.13 (CH <sub>2</sub> )	1.47 ( <i>m</i> ), 1.19 ( <i>m</i> )	
8	39.18 (C)		
9	47.44 (CH)	1.03 ( <i>m</i> )	
10	37.94 (C)		
11	23.67 (CH <sub>2</sub> )	1.86 ( <i>m</i> )	
12	122.01 (CH)	5.20 ( <i>br s</i> )	C-9, C-11, C-13, C-14, C-18
13	143.86 (C)		
14	41.70 (C)		
15	27.55 (CH <sub>2</sub> )	1.64 ( <i>m</i> )	
16	22.91 (CH <sub>2</sub> )	1.53 ( <i>m</i> )	
17	46.31 (C)		
18	41.11 (CH)	2.76 ( <i>dd</i> ; 13.8, 4.5)	C-12, C-13, C-14, C-17,
			C-19
19	45.88 (CH <sub>2</sub> )	1.56 ( <i>m</i> ), 1.06 ( <i>m</i> )	
20	30.58 (C)		
21	33.78 (CH <sub>2</sub> )	1.27 ( <i>m</i> ), 1.11 ( <i>m</i> )	
22	32.49 (CH <sub>2</sub> )	1.68 ( <i>m</i> )	

<sup>*a*</sup>recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop)

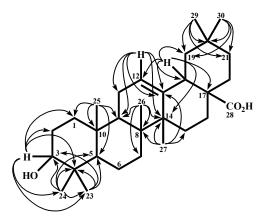
Table 36 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
23	67.98 (CH <sub>2</sub> )	a: 3.46 ( <i>d</i> ; 10.8)	C-3, C-4, C-24
		b: 3.26 ( <i>d</i> ; 10.8)	
24	12.75 (CH <sub>3</sub> )	0.70 ( <i>s</i> )	C-3, C-4, C-5, C-23
25	16.81 (CH <sub>3</sub> )	0.93 (s)	C-2, C-9, C-10
26	16.89 (CH <sub>3</sub> )	0.70 ( <i>s</i> )	C-7, C-8, C-9, C-14
27	25.77 (CH <sub>3</sub> )	1.06 ( <i>s</i> )	C-8, C-13, C-14, C-15
28	181.08 (C=O)		
29	32.95 (CH <sub>3</sub> )	0.85 (s)	C-19, C-20, C-21, C-30
30	23.45 (CH <sub>3</sub> )	0.83 (s)	C-19, C-20, C-21, C-29

<sup>*a*</sup>recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop)



ART39 was obtained as a white solid. The IR spectrum exhibited absorptions for a hydroxyl group at 3406 cm<sup>-1</sup> and for a carboxyl group at 1685 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectral data (**Table 37**) showed the presence of an olefinic proton ( $\delta_{\rm H}$  5.27, H-12), an oxygenated methine proton ( $\delta_{\rm H}$  3.21, H-3), a methine proton ( $\delta_{\rm H}$  2.82, H-18) and seven quaternary methyl groups ( $\delta_{\rm H}$  1.13, H<sub>3</sub>-27; 0.98, H<sub>3</sub>-23; 0.92, H<sub>3</sub>-30; 0.90, H<sub>3</sub>-29; 0.89, H<sub>3</sub>-25 and 0.77, H<sub>3</sub>-24 and H<sub>3</sub>-26) which were typical resonances of the pentacyclic triterpenoic acid. The oxygenated methine proton at  $\delta_{\rm H}$  3.21 was assigned to be the  $\alpha$ -axail position according to the large coupling constant value of 9.9 Hz. The <sup>13</sup>C NMR spectrum (Table 37) revealed the signals of a carbonyl carbon  $(\delta_{\rm C} 181.52)$ , seven quaternary carbons  $(\delta_{\rm C} 143.77, 46.42, 41.67, 38.71, 38.60, 37.02)$ and 30.66), five methine carbons ( $\delta_{\rm C}$  125.61, 78.98, 55.20, 47.61 and 41.11), ten methylene carbons ( $\delta_{\rm C}$  45.93, 38.41, 33.84, 32.66, 32.47, 27.67, 27.01, 23.38, 22.99 and 18.30) and seven methyl carbons ( $\delta_{\rm C}$  33.06, 25.87, 28.05, 23.55, 16.92, 15.54 and 15.29). The structure of ART39 was therefore determined to be oleanolic acid (Table 67) (Yang et al., 2009). The HMBC experiment (Table 37) completely confirmed the assigned structure.



Selected HMBC correlations of ART39

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	38.41 (CH <sub>2</sub> )	1.58 ( <i>m</i> ), 0.97 ( <i>m</i> )	
2	27.01 (CH <sub>2</sub> )	1.59 ( <i>m</i> )	
3	78.98 (CH)	3.21 ( <i>dd</i> ; 9.9, 5.1)	C-1, C-2, C-4, C-23, C-24
4	$38.60 (C)^c$		
5	55.20 (CH)	0.72 ( <i>br d</i> ; 11.1)	
6	18.30 (CH <sub>2</sub> )	1.35 <i>(m)</i>	
7	32.66 (CH <sub>2</sub> )	1.41 ( <i>m</i> )	
8	38.71 (C) <sup>c</sup>		
9	47.61 (CH)	1.52 ( <i>m</i> )	
10	37.02 (C)		
11	22.99 (CH <sub>2</sub> )	1.91 ( <i>m</i> )	
12	125.61 (CH)	5.27 ( <i>t</i> ; 3.6)	C-9, C-11, C-14, C-18
13	143.77 (C)		
14	41.67 (C) <sup>d</sup>		

<sup>*a*</sup>recored in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop); <sup>*b,c,d*</sup>may be interchangeable

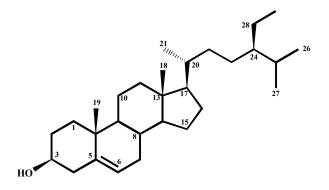
Table 37 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
15	27.67 (CH <sub>2</sub> )	1.71 ( <i>m</i> ), 1.59 ( <i>m</i> )	
16	23.38 (CH <sub>2</sub> )	1.91 ( <i>m</i> ), 1.62 ( <i>m</i> )	
17	$46.42 (C)^d$		
18	41.11 (CH)	2.82 ( <i>dd</i> ; 14.4, 4.5)	C-12, C-13, C-14, C-17, C-19
19	45.93 (CH <sub>2</sub> )	1.64 ( <i>m</i> ), 1.13 ( <i>m</i> )	
20	30.66 (C)		
21	33.84 (CH <sub>2</sub> )	2.28 ( <i>m</i> ), 1.18 ( <i>m</i> )	
22	32.47 (CH <sub>2</sub> )	1.78 ( <i>m</i> )	
23	28.05 (CH <sub>3</sub> )	0.98 (s)	C-3, C-4, C-5, C-24
24	15.54 (CH <sub>3</sub> )	0.77 (s)	C-3, C-4, C-5, C-23
25	15.29 (CH <sub>3</sub> )	$0.89(s)^b$	C-1, C-5, C-9, C-10
26	16.92 (CH <sub>3</sub> )	0.77 (s)	C-7, C-8, C-9, C-14
27	25.87 (CH <sub>3</sub> )	1.13 (s)	C-8, C-13, C-14, C-15
28	181.52 (C=O)		
29	33.06 (CH <sub>3</sub> )	$0.90(s)^b$	C-19, C-20, C-21, C-30
30	23.55 (CH <sub>3</sub> )	0.92 (s)	C-19, C-20, C-21, C-29

<sup>*a*</sup>recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop); <sup>*b,c,d*</sup>may be interchangeable

# 3.1.6 Steroids

ART5: β-Sitosterol



**ART5** was obtained as a white solid. The IR spectrum displayed an absorption band at 3410 cm<sup>-1</sup> for OH stretching. The <sup>1</sup>H NMR spectrum (**Table 38**) showed the resonances of a vinylic proton ( $\delta_{\rm H}$  5.27), an oxymethine proton ( $\delta_{\rm H}$  3.42) and six methyl groups of two singlet methyl groups ( $\delta_{\rm H}$  0.93 and 0.61), three doublet methyl groups ( $\delta_{\rm H}$  0.85, 0.76 and 0.73) and one triplet methyl group ( $\delta_{\rm H}$  0.77). The <sup>13</sup>C NMR spectrum associated with DEPT experiments (**Table 38**) revealed the resonances of 29 carbon atoms consisting of three quaternary carbons ( $\delta_{\rm C}$  139.84, 41.34 and 35.52), nine methine carbons ( $\delta_{\rm C}$  120.64, 70.55, 55.79, 55.10, 49.18, 44.86, 35.16, 30.92 and 28.19), eleven methylene carbons ( $\delta_{\rm C}$  41.07, 38.81, 36.28, 32.97, 30.40, 28.72, 27.26, 25.13, 23.31, 22.09 and 20.10) and six methyl carbons ( $\delta_{\rm C}$  18.81, 18.38, 18.04, 17.79, 10.97 and 10.86).

By comparison of the <sup>1</sup>H and <sup>13</sup>C NMR spectral data with the previous report (**Table 68**), **ART5** was assigned as  $\beta$ -sitosterol (Kim *et al.*, 2006).

Table 38	The NMR spectral data of ART5	
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Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
1	36.28 (CH <sub>2</sub> )	1.77 ( <i>m</i> ), 1.08 ( <i>m</i> )
2	28.72 (CH <sub>2</sub> )	1.93 ( <i>m</i> ), 1.63 ( <i>m</i> )
3	70.55 (CH)	3.42 ( <i>m</i> )
4	41.07 (CH <sub>2</sub> )	2.19 ( <i>m</i> )
5	139.84 (C)	
6	120.64 (CH)	5.27 ( <i>d</i> , 4.5)
7	30.40 (CH <sub>2</sub> )	1.74 ( <i>m</i> ), 1.43 ( <i>m</i> )
8	30.92 (CH)	1.95 ( <i>m</i> )
9	49.18 (CH)	0.93 ( <i>m</i> )
10	35.52 (C)	
11	20.10 (CH <sub>2</sub> )	1.54 ( <i>m</i> ), 1.48 ( <i>m</i> )
12	38.81 (CH <sub>2</sub> )	1.76 ( <i>m</i> ), 1.18 ( <i>m</i> )
13	41.34 (C)	
14	55.79 (CH)	1.03 ( <i>m</i> )
15	23.31 (CH <sub>2</sub> )	1.56 ( <i>m</i> )
16	27.26 (CH <sub>2</sub> )	1.31 ( <i>m</i> )
17	55.10 (CH)	1.11 ( <i>m</i> )
18	10.97 (CH <sub>3</sub> )	0.61 (s)
19	18.81 (CH <sub>3</sub> )	0.93 (s)
20	35.16 (CH)	1.38 ( <i>m</i> )
21	17.79 (CH <sub>3</sub> )	0.85 ( <i>d</i> ; 6.6)
22	32.97 (CH <sub>2</sub> )	1.31 ( <i>m</i> ), 1.08 ( <i>m</i> )

<sup>*a*</sup>recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop)

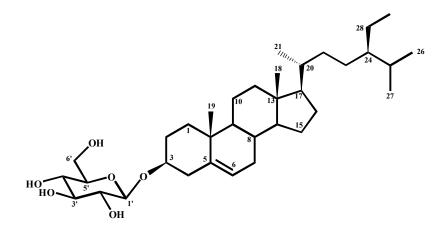
Table 38 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )
23	25.13 (CH <sub>2</sub> )	1.16 ( <i>m</i> )
24	44.86 (CH)	0.93 ( <i>m</i> )
25	28.19 (CH)	1.26 ( <i>m</i> )
26	18.38 (CH <sub>3</sub> )	0.76 ( <i>d</i> ; 6.6)
27	18.04 (CH <sub>3</sub> )	0.73 ( <i>d</i> ; 6.6)
28	22.09 (CH <sub>2</sub> )	1.25 ( <i>m</i> )
29	10.86 (CH <sub>3</sub> )	0.77 ( <i>t</i> ; 7.2)

<sup>*a*</sup>recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop)

# **ART24:** $\beta$ -Sitosterol glucopyranoside

(Daucosterol)



**ART24** was obtained as a white solid with mp 276-278  $^{\circ}$ C. The IR spectrum displayed absorption for hydroxyl group at 3450 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectrum (**Table 39**) showed a characteristic signal of sitosterol glycoside. The sitosterol moiety was deduced from the resonances of one olefinic proton ( $\delta_{\rm H}$  5.37, H-6), one oxymethine proton ( $\delta_{\rm H}$  3.60, H-3) and six methyl groups of two singlet signals ( $\delta_{\rm H}$  1.01, H<sub>3</sub>-19 and 0.69, H<sub>3</sub>-18), three doublet signals ( $\delta_{\rm H}$  0.93, H<sub>3</sub>-21, 0.84, H<sub>3</sub>-26 and 0.82, H<sub>3</sub>-27) and one triplet signal ( $\delta_{\rm H}$  0.85, H<sub>3</sub>-29). The sugar unit was assigned to a glucopyranose, which in the <sup>1</sup>H NMR spectrum, the resonance of an anomeric proton H-1' was at  $\delta_{\rm H}$  4.41, four methine protons were at  $\delta_{\rm H}$  3.44 (H-4'); 3.41 (H-3'); 3.30 (H-5'); 3.24 (H-1') and the oxygenated methylene protons were at  $\delta_{\rm H}$  3.84 and 3.75 (H<sub>2</sub>-6', J = 12.0 Hz). The <sup>13</sup>C NMR spectrum associated with DEPT experiments (**Table 39**) showed the resonances of three quaternary carbons ( $\delta_{\rm C}$  140.39, 42.44 and 36.83), fourteen methine carbons ( $\delta_{\rm C}$  122.30, 101.21, 79.31, 76.50, 75.83, 73.66, 70.28, 56.87, 56.18, 50.30, 45.98, 36.26, 32.00 and 29.27), twelve methylene carbons

 $(\delta_{\rm C}$  62.02, 39.87, 38.83, 37.37, 34.06, 32.04, 29.80, 28.35, 26.19, 24.39, 23.18 and 21.17) and six methyl carbons ( $\delta_{\rm C}$  19.87, 19.40, 19.09, 18.85, 12.03 and 11.93).

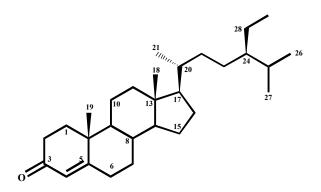
By comparison of the <sup>1</sup>H and <sup>13</sup>C NMR spectral data with the previous report (**Table 69**), **ART24** was assigned to be  $\beta$ -sitosterol glucopyranoside or daucosterol (Lendl *et al.*, 2005).

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
1	37.37 (CH <sub>2</sub> )	α: 1.08 ( <i>m</i> )
		β: 1.87 ( <i>m</i> )
2	29.80 (CH <sub>2</sub> )	a: 1.63 ( <i>m</i> )
		b: 1.91 ( <i>m</i> )
3	79.31 (CH)	3.60 ( <i>m</i> )
4	38.83 (CH <sub>2</sub> )	a: 2.27 ( <i>m</i> )
		b: 2.41 ( <i>m</i> )
5	140.39 (C)	
6	122.30 (CH)	5.37 ( <i>br d</i> , 5.1)
7	32.04 (CH <sub>2</sub> )	a: 1.43 ( <i>m</i> )
		b: 1.68 ( <i>m</i> )
8	32.00 (CH)	1.45 ( <i>m</i> )
9	50.30 (CH)	0.93 ( <i>m</i> )
10	36.83 (C)	
11	21.17 (CH <sub>2</sub> )	1.54 ( <i>m</i> ), 1.48 ( <i>m</i> )
12	39.87 (CH <sub>2</sub> )	<i>α</i> : 1.18 ( <i>m</i> )
		β: 2.03 (m)
13	42.44 (C)	
14	56.87 (CH)	1.03 ( <i>m</i> )
15	24.39 (CH <sub>2</sub> )	a: 1.07 ( <i>m</i> )
		b: 1.60 ( <i>m</i> )

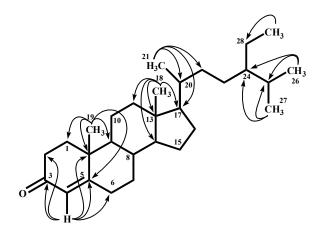
Table 39The NMR spectral data of ART24

 Table 39 (Continued)

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
16	28.35 (CH <sub>2</sub> )	a: 1.26 ( <i>m</i> )
		b: 1.89 ( <i>m</i> )
17	56.18 (CH)	1.11 ( <i>m</i> )
18	11.93 (CH <sub>3</sub> )	0.69 ( <i>s</i> )
19	19.40 (CH <sub>3</sub> )	1.01 ( <i>s</i> )
20	36.26 (CH)	1.38 ( <i>m</i> )
21	18.85 (CH <sub>3</sub> )	0.93 ( <i>d</i> ; 6.3)
22	34.06 (CH <sub>2</sub> )	a: 1.08 ( <i>m</i> )
		b: 1.31 ( <i>m</i> )
23	26.18 (CH <sub>2</sub> )	1.16 ( <i>m</i> )
24	45.98 (CH)	0.93 ( <i>m</i> )
25	29.27 (CH)	1.68 ( <i>m</i> )
26	19.87 (CH <sub>3</sub> )	0.84 ( <i>d</i> ; 6.6)
27	19.09 (CH <sub>3</sub> )	0.82 ( <i>d</i> ; 6.6)
28	23.18 (CH <sub>2</sub> )	1.25 ( <i>m</i> )
29	12.03 (CH <sub>3</sub> )	0.85 ( <i>t</i> ; 6.6)
1'	101.21 (CH)	4.41 ( <i>d</i> ; 7.5)
2'	73.66 (CH)	3.24 ( <i>m</i> )
3'	76.50 (CH)	3.41 ( <i>m</i> )
4'	70.28 (CH)	3.44 ( <i>m</i> )
5'	75.83 (CH)	3.30 ( <i>m</i> )
6'	62.02 (CH <sub>2</sub> )	a: 3.75 ( <i>dd</i> ; 12.0, 4.5)
		b: 3.84 ( <i>dd</i> ; 12.0, 3.0)



ART37 was obtained as a white solid. The IR spectrum showed a stretching of a conjugated carbonyl group at 1674 cm<sup>-1</sup> whereas the UV spectrum exhibited absorption band maxima at 246 nm. The <sup>13</sup>C NMR spectrum (Table 40) displayed the signals for 29 carbons consisting of one carbonyl carbon ( $\delta_{\rm C}$  199.61), three quaternary carbons ( $\delta_{\rm C}$  171.64, 42.42 and 38.61), eight methine carbons ( $\delta_{\rm C}$  123.75, 56.06, 55.91, 53.85, 45.88, 36.12, 35.66 and 29.21), eleven methylene carbons ( $\delta_{\rm C}$  39.66, 35.71, 33.98, 33.92, 32.95, 32.07, 28.18, 26.16, 24.19, 23.10 and 21.05) and six methyl carbons [ $\delta_{\rm C}$  19.79, 19.04, 18.70, 17.40 and 11.95 (2xC)]. The <sup>1</sup>H NMR spectrum (Table 40) showed the same characteristic signals of  $\beta$ -sitosterol except for the downfield shift of olefinic proton ( $\delta_{\rm H}$  5.73, H-4) and the absence of oxygenated methine proton at C-3. Two  $sp^2$  carbons at  $\delta_c$  123.75 (C-4) and 171.64 (C-5) and the down field chemical shift of C-5 associated with the presence of a carbonyl carbon at  $\delta_{\rm C}$  199.61 indicating the existence of the conjugated carbonyl functionality at C-3. In the HMBC spectrum (Table 40), the olefinic proton H-4 exhibited the correlations with C-2 ( $\delta_{\rm C}$  33.92), C-3 ( $\delta_{\rm C}$  199.61), C-5 ( $\delta_{\rm C}$  171.64), C-6 ( $\delta_{\rm C}$  32.95) and C-10  $(\delta_{\rm C}$  38.61) suggesting the presence of a double bond at C-4 and C-5. On the basis of its spectral data as well as comparison with the literature (Table 70), compound ART37 was therefore determined to be stigmast-4-en-3-one.



Selected HMBC correlations of ART37

Table 40	The NMR s	spectral data	of <b>ART37</b>
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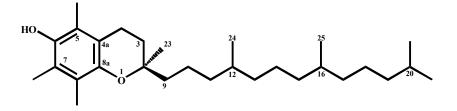
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	35.71 (CH <sub>2</sub> )	1.67 ( <i>m</i> ), 1.54 ( <i>m</i> )	
2	33.92 (CH <sub>2</sub> )	2.50 ( <i>m</i> ), 2.28 ( <i>m</i> )	
3	199.61 (C=O)		
4	123.75 (CH)	5.73 (s)	C-2, C-3, C-5, C-6, C-10
5	171.64 (C)		
6	32.95 (CH <sub>2</sub> )	2.40 ( <i>m</i> ), 2.25 ( <i>m</i> )	
7	32.07 (CH <sub>2</sub> )	1.85 ( <i>m</i> ), 1.01 ( <i>m</i> )	
8	35.66 (CH)	1.71 ( <i>m</i> )	
9	53.85 (CH)	0.92 ( <i>m</i> )	
10	38.61 (C)		
11	21.05 (CH <sub>2</sub> )	1.50 ( <i>m</i> ), 1.40 ( <i>m</i> )	
12	39.66 (CH <sub>2</sub> )	2.04 ( <i>m</i> ), 1.15 ( <i>m</i> )	
13	42.42 (C)		
14	55.91 (CH)	1.00 ( <i>m</i> )	
15	24.19 (CH <sub>2</sub> )	1.29 ( <i>m</i> ), 1.23 ( <i>m</i> )	
16	28.18 (CH <sub>2</sub> )	1.32 ( <i>m</i> ), 1.27 ( <i>m</i> )	

Table 40 (Continued)

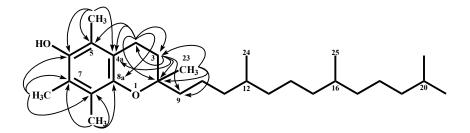
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
17	56.06 (CH)	1.11 ( <i>m</i> )	
18	11.95 (CH <sub>3</sub> )	0.71 ( <i>s</i> )	C-12, C-13, C-14, C-17
19	17.40 (CH <sub>3</sub> )	1.18 (s)	C-1, C-5, C-9, C-10
20	36.12 (CH)	2.01 ( <i>m</i> )	
21	18.70 (CH <sub>3</sub> )	0.92 ( <i>d</i> ; 6.3)	C-17, C-20, C-22
22	33.98 (CH <sub>2</sub> )	2.39 ( <i>m</i> )	
23	26.16 (CH <sub>2</sub> )	1.17 ( <i>m</i> )	
24	45.88 (CH)	0.93 ( <i>m</i> )	
25	29.21 (CH)	1.26 ( <i>m</i> )	
26	19.79 (CH <sub>3</sub> )	0.85 ( <i>d</i> ; 6.9)	C-24, C-25, C-27
27	19.04 (CH <sub>3</sub> )	0.84 ( <i>d</i> ; 6.6)	C-24, C-25, C-26
28	23.10 (CH <sub>2</sub> )	1.29 ( <i>m</i> )	
29	11.95 (CH <sub>3</sub> )	0.83 ( <i>d</i> ; 6.6)	C-24, C-28

#### 3.1.7 Benzenoids

**ART1:**  $\alpha$ -Tocopherol



**ART1** was obtained as a yellow gum. The UV spectrum displayed absorption maxima at 246, 291 and 383 nm. The IR spectrum showed absorption bands of O-H stretching at 3349 cm<sup>-1</sup>. The <sup>1</sup>H and <sup>13</sup>C NMR spectral data (**Table 41**) displayed the resonances corresponding to three aryl methyl groups [ $\delta_{\rm H}$  2.16 and 2.11 (2xCH<sub>3</sub>);  $\delta_{\rm C}$ 12.23, 11.79 and 11.30], four secondary methyl groups [ $\delta_{\rm H}$  0.86 (2xCH<sub>3</sub>), 0.85 and 0.84;  $\delta_{\rm C}$  22.75, 22.65, 19.77 and 19.68] and a tertiary methyl group ( $\delta_{\rm H}$  1.23;  $\delta_{\rm C}$  23.82). The protons resonated at  $\delta_{\rm H}$  2.60 (*t*) and 1.78 (*m*) which coupled to each other were assigned for methylene protons H<sub>2</sub>-4 and H<sub>2</sub>-3, respectively. The proton resonances at  $\delta_{\rm H}$  1.02-1.60 and carbon resonances at  $\delta_{\rm C}$  39.84, 39.39, 37.47 (2xC), 37.44, 37.41, 24.83, 24.47, and 21.06 corresponded to nine methylene groups. The <sup>13</sup>C NMR spectrum (**Table 41**) further showed the resonances of oxygenated aromatic carbon at  $\delta_{\rm C}$  145.57 and 144.55 together with oxygenated quaternary carbon at  $\delta_{\rm C}$  74.53. **ART1** was thus determined as  $\alpha$ -tocopherol. The HMBC experiment (**Table 41**) was in agreement with the assigned structure.



Selected HMBC correlations of ART1

 Table 41
 The NMR spectral data of ART1

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
2	74.53 (C)		
3	31.58 (CH <sub>2</sub> )	1.78 ( <i>m</i> , 6.9)	C-2, C-4, C-4a, C-9
4	20.78 (CH <sub>2</sub> )	2.60 ( <i>t</i> , 6.9)	C-2, C-3, C-4a, C-5, C-8a
4a	117.36 (C)		
5	118.49 (C)		
6	144.55 (C)		
7	122.62 (C)		
8	121.04 (C)		
8a	145.57 (C)		
9	39.84 (CH <sub>2</sub> )	1.47-1.60 ( <i>m</i> )	C-2
10	21.06 (CH <sub>2</sub> )	1.02-1.48 ( <i>m</i> )	
11	$37.47 (CH_2)^a$	1.02-1.48 ( <i>m</i> )	
12	$32.82 (CH)^{b}$	1.02-1.48 ( <i>m</i> )	
13	$37.47 (CH_2)^a$	1.02-1.48 ( <i>m</i> )	
14	24.47 $(CH_2)^c$	1.02-1.48 ( <i>m</i> )	
15	$37.44 (CH_2)^a$	1.02-1.48 ( <i>m</i> )	
16	32.72 (CH) <sup>b</sup>	1.02-1.48 ( <i>m</i> )	
17	$37.41 (CH_2)^a$	1.02-1.48 ( <i>m</i> )	
18 a.b.c.d.e 1	24.83 $(CH_2)^c$	1.02-1.48 ( <i>m</i> )	

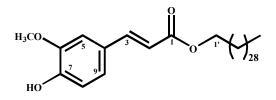
*a,b,c,d,e* may be interchangeable in the same sign

 Table 41 (Continued)

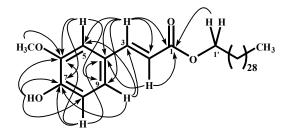
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
19	39.39 (CH <sub>2</sub> )	1.02-1.48 ( <i>m</i> )	
20	28.00 (CH)	1.47-1.60 ( <i>m</i> )	
21	$22.75 (CH_3)^d$	0.86 ( <i>d</i> ; 6.6)	C-19, C-20
22	$22.65 (CH_3)^d$	0.86 ( <i>d</i> ; 6.6)	C-19, C-20
23	23.82 (CH <sub>3</sub> )	1.23 (s)	C-2, C-3, C-9
24	19.77 (CH <sub>3</sub> ) <sup>e</sup>	0.85 ( <i>d</i> ; 6.6)	C-11, C-12, C-13
25	19.68 (CH <sub>3</sub> ) <sup>e</sup>	0.84 ( <i>d</i> ; 6.6)	C-15, C-16, C-17
5-CH <sub>3</sub>	11.30 (CH <sub>3</sub> )	2.11 (s)	C-4a, C-5, C-6
7-CH3	12.23 (CH <sub>3</sub> )	2.16 ( <i>s</i> )	C-6, C-7, C-8
8-CH <sub>3</sub>	11.79 (CH <sub>3</sub> )	2.11 (s)	C-7, C-8, C-8a

*a,b,c,d,e* may be interchangeable in the same sign

ART21: trans-Triacontyl-4-hydroxy-3-methoxycinnamate



ART21 was obtained as a white powder. The IR spectrum displayed absorption bands for a hydroxyl and a carbonyl groups at 3445 and 1720 cm<sup>-1</sup>, respectively. The UV spectrum exhibited absorptions maxima at 240 and 327 nm. The EI-MS spectrum displayed a molecular ion peak at m/z 614 which was 30 mass units higher than ART30. The <sup>1</sup>H NMR spectrum (Table 42) exhibited the resonances of aromatic protons at  $\delta_{\rm H}$  7.04 (*d*; *J* = 1.8 Hz), 6.92 (*d*; *J* = 8.4 Hz) and 6.84 (*dd*; *J* = 8.4 and 1.8 Hz) corresponding to H-5, H-8 and H-9, respectively, which indicated the presence of 1,2,4-trisubstituted benzene ring. A hydroxyl proton resonated at  $\delta_{\rm H}$  5.86 and a methoxyl group resonated at  $\delta_{\rm H}$  3.93 (s) were proposed to be substituted at ortho position. The NOESY experiment exhibited the cross peak between the methoxyl group and the aromatic proton at  $\delta_{\rm H}$  7.04 (H-5) indicating that the methoxyl substituent was then positioned at the C-6. The spectrum further showed the two trans-olefinic protons at  $\delta_{\rm H}$  7.61 (d, J = 15.9 Hz) and 6.29 (d, J = 15.9 Hz) together with the 30 carbons side chain signals seen as signals of oxygenated methylene protons at  $\delta_{\rm H}$  4.18 (t, J = 6.9 Hz, H<sub>2</sub>-1'), metheylene protons at  $\delta_{\rm H}$  1.69 (m, J = 6.9 Hz, H<sub>2</sub>-2') and 1.25 (*br s*, H<sub>2</sub>-3' - H<sub>2</sub>-29') and one methyl group at  $\delta_{\rm H}$  0.88 (*t*, *J* = 6.9 Hz, H<sub>3</sub>-30'). These data corresponded to the feruloyl moiety with long chain hydrocarbons of 30 carbons. The connectivity of the 30 carbons side chain to an ester group was evidenced from the  ${}^{3}J$  HMBC correlation (Table 42) of H<sub>2</sub>-2' ( $\delta_{\rm H}$  4.19) to the ester carbonyl carbon ( $\delta_{\rm H}$  167.35). ART21 was thus designated as *trans*-triacontyl-4hydroxy-3-methoxy-cinnamate (**Table 71**) (Boonyaratavej *et al.*, 1992). The HMBC correlations completely confirmed the assigned structure.



Selected HMBC correlations of ART21

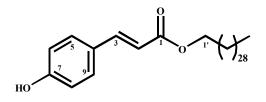
ý H<sub>3</sub>CO CH<sub>3</sub>  $J_{28}$ но

NOESY correlations of ART21

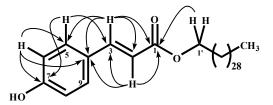
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	167.35 (C=O)		
2	115.74 (CH)	6.29 ( <i>d</i> ; 15.9)	C-1, C-4
3	144.59 (CH)	7.61 ( <i>d</i> ; 15.9)	C-1, C-2, C-4, C-5, C-9
4	127.10 (C)		
5	109.33 (CH)	7.04 ( <i>d</i> ; 1.8)	C-3, C-6, C-7, C-9
6	147.90 (C)		
7	146.76 (C)		
8	114.70 (CH)	6.92 ( <i>d</i> ; 8.4)	C-4, C-6, C-7
9	123.02 (CH)	6.84 ( <i>dd</i> ; 8.4, 1.8)	C-3, C-5, C-7
1'	64.61 (CH <sub>2</sub> )	4.19 ( <i>t</i> ; 6.6)	C-1, C-2', C-3'
2'	28.78 (CH <sub>2</sub> )	1.69 ( <i>m</i> )	
3'-29'	31.92, 29.54,	1.27 ( <i>br s</i> )	
	29.34, 29.30,		
	26.00, 22.68		
	(CH <sub>2</sub> )		
30'	14.09 (CH <sub>3</sub> )	0.88 ( <i>t</i> ; 6.3)	C-28', C-29'
7 <b>-</b> OH		5.86 (s)	C-6, C-7, C-8
6-OCH <sub>3</sub>	55.95 (CH <sub>3</sub> )	3.93 (s)	C-6

Table 42The NMR spectral data of ART21

ART30: trans-Triacontyl-4-hydroxycinnamate



ART30 was obtained as white powder, mp 90-92 °C. The IR spectrum displayed absorptions for a hydroxyl and carbonyl groups at 3490 and 1696 cm<sup>-1</sup>, respectively, whereas the UV spectrum showed maximum absorption bands at 234 and 327 nm. The EI-MS spectrum exhibited the molecular ion peak at m/z 584 which was 30 mass units less than ART21. The <sup>1</sup>H NMR spectral data (Table 43) showed the resonance of aromatic proton as AA'BB' of H-5/H-8 at  $\delta_{\rm H}$  7.43 (d, J = 8.7 Hz) and H-6/H-9 at  $\delta_{\rm H}$  6.84 (d, J = 8.7 Hz), and two doublets of *trans* olefinic protons H-3 and H-2 at  $\delta_{\rm H}$  7.62 and 6.30 (J = 15.9 Hz), respectively, which corresponded to a p-coumaroyl moiety. The presence of long chain hydrocarbons was implied from the mass and signals of a triplet of oxygenated methylene proton H<sub>2</sub>-1' at  $\delta_{\rm H}$  4.18 (J = 6.9 Hz), multiplet of metheylene protons H<sub>2</sub>-2' at  $\delta_{\rm H}$  1.69 (J = 6.9 Hz) and H<sub>2</sub>-3' - H<sub>2</sub>-29' at  $\delta_{\rm H}$  1.25 (br s) as well as a triplet of methyl protons H<sub>3</sub>-30' at  $\delta_{\rm H}$  0.88 (J = 6.9 Hz). The ester linkage was confirmed by the HMBC correlations (Table 43) of H<sub>2</sub>-1' and H-3 to the ester carbonyl carbon ( $\delta_{\rm H}$  167.41). The ion peak at m/z 421, which was in agreement with  $C_{30}H_{61}$ , suggested that the long chain hydrocarbon contained thirty carbons. ART30 was then identified as trans-triacontyl-4hydroxycinnamate (Table 72) (Wandji et al., 1990).

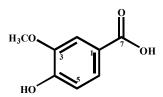


Selected HMBC correlations of ART30

Table 43The NMR spectral data of ART30

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	167.41 (C=O)		
2	115.96 (CH)	6.30 ( <i>d</i> ; 15.9)	C-1, C-3, C-4
3	144.09 (CH)	7.62 ( <i>d</i> ; 15.9)	C-1, C-2, C-4, C-5, C-9
4	127.50 (C)		
5, 9	129.89 (CH)	7.43 ( <i>d</i> ; 8.7)	C-3, C-7
6, 8	115.84 (CH)	6.84 ( <i>d</i> ; 8.7)	C-4, C-5, C-9, C-7
7	157.49 (C)		
1'	64.61 (CH <sub>2</sub> )	4.18 ( <i>t</i> ; 6.9)	C-1, C-2', C-3'
2'	28.77 (CH <sub>2</sub> )	1.69 ( <i>m</i> ; 6.9)	C-1', C-3', C-4'
3'-29'	31.91, 29.58,	1.25 ( <i>br s</i> )	
	29.53, 29.34,		
	29.29, 25.99,		
	22.67 (CH <sub>2</sub> )		
30'	14.07 (CH <sub>3</sub> )	0.88 ( <i>t</i> ; 6.9)	C-28', C-29'

ART35: 4-Hydroxy-3-methoxybenzoic acid



**ART35** was obtained as a pale yellow solid. The IR spectrum showed the stretching of a hydroxyl and carbonyl groups at 3338 and 1687 cm<sup>-1</sup>, respectively. The UV spectrum displayed absorption bands maxima at 220, 257 and 290 nm. The <sup>1</sup>H NMR spectrum (**Table 44**) exhibited signals of a methoxyl group at  $\delta_{\rm H}$  3.93 and ABX signals of 1,2,4-trisubstituted benzene at  $\delta_{\rm H}$  7.61 (*dd*; *J* = 8.1 and 1.8 Hz, H-6), 7.56 (*d*; *J* = 1.8 Hz, H-2) and 6.89 (*d*; *J* = 8.1 Hz, H-5). The downfield shifts of the aromatic protons H-2 and H-6 suggested that both protons were adjacent to the electron withdrawing group which was suggested as carboxyl group due to the carbon signal at  $\delta_{\rm C}$  168.81 (**Table 44**). The methoxyl group was placed at C-3 which was *ortho* position to C-2 by irradiation at the resonance of this methoxyl group resulted in enhancement of the signal of H-2. It was thus concluded that **ART35** was 4-hydroxy-3-methoxybenzoic acid.

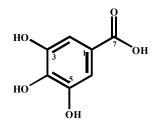
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult., $J_{ m Hz}$ )	HMBC correlations
1	121.78 (C)		
2	112.43 (CH)	7.56 ( <i>d</i> ; 1.8)	C-1, C-3, C-4, C-6, C-7
3	146.65 (C)		
4	150.48 (C)		
5	114.43 (CH)	6.89 ( <i>d</i> ; 8.1)	C-1, C-3, C-4
6	124.25 (CH)	7.61 ( <i>dd</i> ; 8.1, 1.8)	C-2, C-4, C-7
		l	

**Table 44**The NMR spectral data of **ART35** 

7	168.81 (C=O)			
3-OCH <sub>3</sub>		3.93 (s)		
$a_{\text{recorded in CDCl} \perp \text{CD OD (1 dron)}}$				

recorded in CDCl<sub>3</sub>+CD<sub>3</sub>OD (1 drop)

ART36: Gallic acid

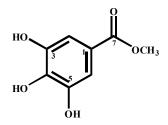


**ART36** was obtained as a pale yellow solid. The IR spectrum exhibited absorptions for a hydroxyl stretching at 3345 cm<sup>-1</sup> and for a carbonyl stretching at 1690 cm<sup>-1</sup>whereas the UV spectrum displayed maximum absorption bands at 216, 272 and 355 nm. The <sup>1</sup>H NMR spectrum (**Table 45**) showed only the singlet resonance of the aromatic proton at  $\delta_{\rm H}$  6.90. The <sup>13</sup>C NMR spectrum (**Table 45**) exhibited five signals for seven carbon atoms: one carbonyl ( $\delta_{\rm C}$  167.96), four quaternary [ $\delta_{\rm C}$  145.84 (2xC), 138.43 and 120.89] and two methine [ $\delta_{\rm C}$  109.16 (2xC)] carbons. The above data suggested that **ART36** was gallic acid.

Table 45The NMR spectral data of ART36

Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.)	HMBC correlations
1	109.16 (C)		
2	145.84 (C)		
3	138.43 (C)		
4	145.84 (C)		
5	109.16 (CH)	6.90 ( <i>s</i> )	C-1, C-3, C-4, C-6, C-7
6	120.89 (C)		
7	167.96 (C=O)		

**ART40:** Methyl gallate



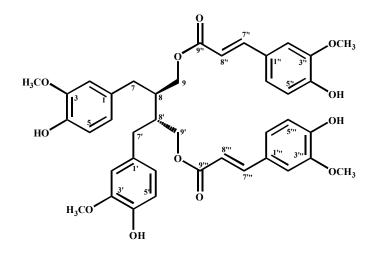
**ART40** was obtained as a yellow gum. The IR spectrum showed a stretching of a hydroxyl group at 3328 cm-1 and a carbonyl group at 1706 cm<sup>-1</sup> whereas the UV spectrum displayed maximum absorptions at 218, 274 and 356 nm. The <sup>1</sup>H NMR spectrum (**Table 46**) showed two singlet signals of an aromatic proton at  $\delta_{\rm H}$  7.19 and methoxyl group at  $\delta_{\rm H}$  3.88. The methoxyl protons exhibited the correlation to the carbonyl carbon ( $\delta_{\rm C}$  167.11) in the HMBC experiment (**Table 46**) indicating the presence of a methyl ester, therefore **ART40** was determined as methyl gallate. The carbon signals of one carbonyl ( $\delta_{\rm C}$  167.11), four quaternary [ $\delta_{\rm C}$  144.85 (2xC), 137.43 and 121.00], two methine [ $\delta_{\rm C}$  109.41 (2xC)] and one methyl ( $\delta_{\rm C}$  51.67) carbons (**Table 46**) were in agreement with the assigned structure.

Table 46The NMR spectral data of ART40

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.)	HMBC correlations
1	121.00 (C)		
2,6	109.41 (CH)	7.19 (s)	C-1, C-3(C-5), C-4, C-7
3,5	144.85 (C)		
4	137.34 (C)		
7	167.11 (C=O)		

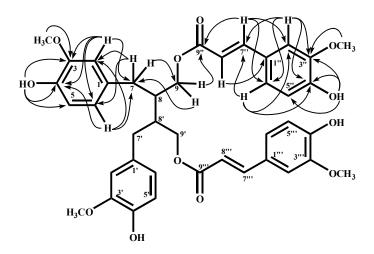
# 3.1.8 Lignans

ART33: 9,9'-O-Diferuloyl-(-)-secoisolariciresinol



**ART33** was obtained as a white solid with mp 276-278  $^{\circ}$ C. The IR spectrum showed the stretching of a hydroxyl group (3409 cm<sup>-1</sup>), carbonyl ester (1702 cm<sup>-1</sup>), double bond (1637 cm<sup>-1</sup>), and aromatic ring (1602, 1513 cm<sup>-1</sup>). The UV spectrum exhibited maximum absorptions at 216, 230, 290 and 327 nm. The <sup>1</sup>H NMR spectrum (**Table 47**) indicated the presence of a phenolic hydroxyl proton at  $\delta_{\rm H}$  5.89 (*s*); two olefinic protons at  $\delta_{\rm H}$  7.59 (*d*, *J* = 15.9 Hz) and 6.29 (*d*, *J* = 15.9 Hz) with *trans*- onfiguration; aromatic protons at  $\delta_{\rm H}$  7.07 (*dd*, *J* = 8.1 and 1.5 Hz), 6.91 (*d*, *J* = 8.1 Hz) and 7.02 (*d*, *J* = 1.5 Hz) and methoxyl protons at  $\delta_{\rm H}$  3.93 (*s*). The above data in association with the HMBC experiment (**Table 47**) indicated that the partial substructure was a feruloyl group. The remaining part of the spectrum showed a hydroxyl proton signal [ $\delta_{\rm H}$  5.48 (*s*)]; an aromatic ABX proton signals [ $\delta_{\rm H}$  6.62 (*dd*, *J* = 8.1 and 1.5 Hz), 6.81 (*d*, *J* = 8.1 Hz) and 6.53 (*d*, *J* = 1.5 Hz)] and a methoxyl group signal [ $\delta_{\rm H}$  3.78 (*s*)], suggesting the presence of a 4-hydroxy-3-methoxyphenyl group. In addition, the spectrum also showed the resonance signals for two sets of methylene protons [ $\delta_{\rm H}$  4.40 (*dd*, *J* = 11.4 and 5.7 Hz) and 4.22 (*dd*, *J* = 11.4 and 5.7 Hz); and  $\delta_{\rm H}$  2.70 (*dd*, *J* = 15.0, 7.0 Hz) and 2.76 (*dd*, *J* = 15.0, 7.0 Hz)] and one methine proton [ $\delta_{\rm H}$  2.22 (*m*)]. The <sup>1</sup>H–<sup>1</sup>H COSY, HMQC and HMBC correlations suggested that -CH<sub>2</sub>CH(CH<sub>2</sub>O-)- moiety was an additional partial substructure. The connectivity between the feruloyl, the 4-hydroxy-3-methoxyphenyl, and the -CH<sub>2</sub>CH(CH<sub>2</sub>O-)- moieties was established from the HMBC correlations (**Table 47**). The correlations of the oxygenated methylene protons H-9 ( $\delta_{\rm H}$  4.40 and 4.22) with the carbonyl ester C-9" at  $\delta_{\rm C}$  167.20 confirmed the connection of the feruloyl to -CH<sub>2</sub>CH(CH<sub>2</sub>O-)-moiety via ester linkage. The <sup>3</sup>*J* HMBC correlations of the methylene protons H<sub>2</sub>-7 ( $\delta_{\rm H}$  2.70 and 2.76) to C-2 ( $\delta_{\rm C}$  111.31) and C-6 ( $\delta_{\rm C}$  121.74) and of the aromatic protons H-2 ( $\delta_{\rm H}$  6.53) and H-6 ( $\delta_{\rm H}$  6.62) to C-7 ( $\delta_{\rm C}$  35.31) indicated that 4-hydroxy-3-methoxyphenyl group was directly connected to -CH<sub>2</sub>CH(CH<sub>2</sub>O-)- moiety. The chemical shifts of H-8 ( $\delta_{\rm H}$  2.22) and C-8 ( $\delta_{\rm C}$  35.31) implied that of **ART33** could be a symmetrical dimeric compound.

Upon comparison of the spectral data and an optical rotation to the reported data, **ART33** was as a known lignan named 9,9'-O-diferuloyl-(-)-secoisolariciresinol (**Table 73**) (Fuchino *et al.*, 1995).



Selected HMBC correlations of ART33

Position	δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
1,1'	131.71 (C)		
2,2'	111.31 (CH)	6.53 ( <i>d</i> , 1.5)	C-1(C-1'), C-3(C-3'),
			C-4(C-4'), C-6(C-6'),
			C-7(C-7')
3,3'	146.47 (C)		
4,4'	143.96 (C)		
5,5'	114.15 (CH)	6.81 ( <i>d</i> , 8.1)	C-1(C-1'), C-3(C-3'),
			C-4(C-4'), C-6(C-6')
6,6'	121.74 (CH)	6.62 ( <i>dd</i> , 8.1, 1.5)	C-1(C-1'), C-4(C-4'),
			C-7(C-7')
7,7'	35.31 (CH <sub>2</sub> )	2.76 ( <i>dd</i> , 15.0, 7.0)	C-1(C-1'), C-2(C-2'),
		2.70 ( <i>dd</i> , 15.0, 7.0)	C-6(C-6'), C-8(C-8'),
			C-9(C-9')
8,8'	40.21 (CH)	2.22 ( <i>m</i> )	C-1(C-1'), C-7(C-7'),
			C-9(C-9')
9,9'	64.47 (CH <sub>2</sub> )	4.40 ( <i>dd</i> , 11.4, 5.7)	C-7(C-7'), C-8(C-8'),
		4.22 ( <i>dd</i> , 11.4, 5.7)	C-9"(C-9"")
3,3'-OCH <sub>3</sub>	55.77 (CH <sub>3</sub> )	3.78 (s)	C-3(C-3')
4,4' <b>-</b> OH		5.48 (s)	C-3(C-3'), C-4(C-4'),
			C-5(C-5')
1",1"	126.87 (C)		
2",2"	109.51 (CH)	7.02 ( <i>d</i> , 1.5)	C-1"(C-1""), C-3"(C-3"")
			C-4"(C-4""), C-6"(C-6"")
			C-7"(C-7"")
3",3"'	146.79 (C)		
4",4"'	148.09 (C)		

Table 47The NMR spectral data of ART33

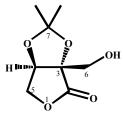
5",5"	114.74 (CH)	6.91 ( <i>d</i> , 8.1)	C-1"(C-1""), C-3"(C-3""),
			C-4"(C-4"")

# Table 47 (Continued)

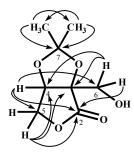
Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
6",6"	123.06 (CH)	7.07 ( <i>dd</i> , 8.1, 1.5)	C-2"(C-2""), C-4"(C-4""),
			C-7"(C-7"")
7",7"	145.11 (CH)	7.59 ( <i>d</i> , 15.9)	C-1", C-2", C-6", C-8", C-9"
8",8"	115.25 (CH)	6.29 ( <i>d</i> , 15.9)	C-1"(C-1""), C-9"(C-9"")
9",9""	167.20 (C=O)		
3",3""-OCH3	55.98 (CH <sub>3</sub> )	3.93 (s)	C-3"(C-3"")
4",4""-OH		5.89 (s)	C-3"(C-3""), C-4"(C-4""),
			C-5"(C-5"")

# 3.1.9 Miscellaneous compounds

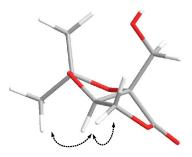
**ART15:**  $(3aS^*, 6aR^*)$ -3a-(hydroxymethyl)-2,2-dimethyldihydrofuro [3,4-*d*][1,3]dioxol-4(3a*H*)-one



**ART15** was obtained as colorless oil. The IR spectrum displayed absorption bands for a hydroxyl group at 3340 cm<sup>-1</sup> and for a carbonyl group at 1750 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectrum (**Table 48**) showed the resonances of an oxymethine proton at  $\delta_{\rm H}$  4.83 attributed to H-4, two non-equivalent oxymethylene protons at  $\delta_{\rm H}$  4.45 and 4.39 attributed to H-5 and at  $\delta_{\rm H}$  3.96 and 3.93 attributed to H-6. The spectrum further exhibited the signals of two singlet methyl groups at  $\delta_{\rm H}$  1.49 and 1.43, assigned for H<sub>3</sub>-8 and H<sub>3</sub>-9, respectively. The five membered-ring lactone was evidenced from the <sup>3</sup>J HMBC correlations (**Table 48**) of H-4 and H<sub>2</sub>-5 to the carbonyl functionality C-2 ( $\delta_{\rm C}$  175.60). The hydroxymethyl group was then placed at C-3 ( $\delta_{\rm C}$  85.53) due to the <sup>3</sup>J correlations of H-4 to C-6 ( $\delta_{\rm C}$  61.41) and that of H-6 to C-2 ( $\delta_{\rm C}$  175.60). The two methyl groups were assigned to be connected with the deoxygenated quaternary carbon C-7 ( $\delta_{\rm C}$  113.51) according to its HMBC correlations (**Table 48**). The relative stereochemistry was deduced from the cross peaks in the NOESY experiment between H-4 and H<sub>2</sub>-5 $\alpha$  and H<sub>3</sub>-8, but not to H<sub>2</sub>-5 $\beta$ , H<sub>2</sub>-6 and H<sub>3</sub>-9. These data indicated that the H-4 and the hydroxymethyl group was a *trans*  stereorelatives. The <sup>13</sup>C NMR spectral data associated with DEPT experiments (**Table 48**) revealed the resonances of eight carbons: a carbonyl carbon ( $\delta_{\rm C}$  175.60), two quaternary carbons ( $\delta_{\rm C}$  113.51 and 85.53), a methine carbon ( $\delta_{\rm C}$  78.09), two methylene carbons ( $\delta_{\rm C}$  70.23 and 61.41) and two methyl carbons ( $\delta_{\rm C}$  26.99 and 26.33). The above conclusion confirmed the assigned structure of **ART15** as a new butyrolactone derivative, (3a*S*\*,6a*R*\*)-3a-(hydroxymethyl)-2,2-dimethyldihydro-furo[3,4-*d*][1,3]dioxol-4(3a*H*)-one.



Selected HMBC correlations of ART15

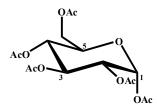


Energy-minimized (MM2) structure of **ART15** showing selected NOESY experiment

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
2	175.60 (C=O)		
3	85.53 (C)		
4	78.09 (CH)	4.83 ( <i>d</i> ; 3.5)	C-2, C-5, C-6
5	70.23 (CH <sub>2</sub> )	4.45 ( <i>d</i> ; 11.0)	C-2, C-3, C-4
		4.39 ( <i>dd</i> ; 11.0, 3.5)	
6	61.41 (CH <sub>2</sub> )	3.96 ( <i>d</i> ; 11.5)	C-2, C-3, C-4
		3.93 ( <i>d</i> ; 11.5)	C-2, C-3, C-4
7	113.51 (C)		
8	26.99 (CH <sub>3</sub> )	1.49 ( <i>s</i> )	C-7, C-9
9	26.33 (CH <sub>3</sub> )	1.43 (s)	C-7, C-8

Table 48The NMR spectral data of ART15

**ART26:**  $\alpha$ -D-glucopyranoside penta-acetate

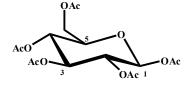


ART26 was obtained as a white solid from the acetylation of the methanolic extract of the leaves. The IR spectrum displayed absorption band for a carbonyl group at 1750 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectrum (**Table 49**) revealed that it was an acetyl derivative of glucose with an anomeric proton H-1 resonated at  $\delta_{\rm H}$  6.32 (d, J = 3.0 Hz) while five acetyl groups resonated at  $\delta_{\rm H}$  2.19, 2.10, 2.04, 2.03 and 2.02. The remaining signals corresponded to four methine protons ( $\delta_{\rm H}$  5.47, t, J = 10.2 Hz, H-3; 5.14, m, H-4,; 5.09, m, H-2 and 4.13, m, H-5) and methylene protons at  $\delta_{\rm H}$  4.27 and 4.09 (each m, H<sub>2</sub>-6). The  $\beta$ -anomeric proton was indicated from a coupling constant value of 3.6 Hz. The <sup>13</sup>C NMR spectrum (Table 49) displayed the resonances of five carbonyl carbons ( $\delta_{\rm C}$  170.54, 170.14, 169.58, 169.34 and 168.67), five methine carbons [ $\delta_{\rm C}$  88.97, 69.78 (2xC), 69.15 and 67.85], one methylene carbon ( $\delta_{\rm C}$  61.42) and five methyl carbons ( $\delta_c$  20.76, 20.58, 20.55, 20.45 and 20.34). The HMBC experiment (Table 49) also confirmed the structural assignment. Furthermore, ART26 was clearly identified to be  $\alpha$ -D-glucopyranoside penta-acetate by comparison of its <sup>1</sup>H NMR data to an authentic compound, which was prepared from acetylation of  $\alpha$ -D-glucose.

Position	δ <sub>C</sub> (Type)	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	HMBC correlations
1	88.97 (CH)	6.32 ( <i>d</i> ; 3.0)	C-2, C-3, C-5, 1-C=O
2	69.15 (CH)	5.09 ( <i>dd</i> ; 10.2, 3.0)	C-1, C-3, 2-C=O
3	69.78 (CH)	5.47 ( <i>t</i> ; 10.2)	C-2, C-4, 3-C=O
4	67.85 (CH)	5.14 ( <i>t</i> ; 10.2)	C-3, C-5, C-6, 4-C=O
5	69.78 (CH)	4.13 ( <i>m</i> )	C-1, C-4
6	61.42 (CH <sub>2</sub> )	4.27 ( <i>m</i> ), 4.09 ( <i>m</i> )	C-4, C-5, 6-C=O
1-OAc	168.67 (C=O)		
	20.76 (CH <sub>3</sub> )	2.19 (s)	1-C=O
2-OAc	169.58 (C=O)		
	20.55 (CH <sub>3</sub> )	2.04 (s)	2-C=O
3-OAc	170.14 (C=O)		
	20.34 (CH <sub>3</sub> )	2.02 (s)	3-C=O
4-OAc	169.34 (C=O)		
	20.45 (CH <sub>3</sub> )	2.03 (s)	4-C=O
6-OAc	170.54 (C=O)		
	20.58 (CH <sub>3</sub> )	2.10 (s)	6-C=O

Table 49The NMR spectral data of ART26

**ART27:**  $\beta$ -D-glucopyranoside penta-acetate

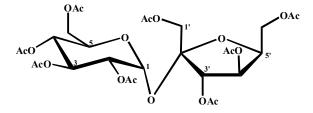


ART27, a white solid, was obtained from the acetylation of the methanolic extract of the leaves. The IR spectrum exhibited absorption at 1759 cm<sup>-1</sup> for a carbonyl stretching. The <sup>1</sup>H NMR spectrum (Table 50) showed the resonances of an anomeric proton H-1 ( $\delta_{\rm H}$  5.72, d, J = 9.3 Hz), five acetyl groups [ $\delta_{\rm H}$  2.11, 2.08, 2.03 (2xCH<sub>3</sub>) and 2.01], four methine protons ( $\delta_{\rm H}$  5.23, t, J = 9.3 Hz, H-3; 5.13, t, J = 9.3 Hz, H-4; 5.11, *t*, *J* = 9.3 Hz, H-2 and 3.85, *ddd*, *J* = 9.3, 4.5 and 2.1 Hz, H-5) and methylene protons ( $\delta_{\rm H}$  4.30, dd, J = 12.6 and 4.5 Hz and 4.10, dd, J = 12.6 and 2.1 Hz, H<sub>2</sub>-6). The chemical shift and a coupling constant of J = 9.3 Hz of H-1 suggested that it was an  $\alpha$ -anomeric proton. Therefore,  $\beta$ -D-glucopyranoside penta-acetate was assigned. The <sup>13</sup>C NMR spectrum (Table 50) displayed the resonances of five carbonyl carbons  $(\delta_{\rm C}$  170.55, 170.04, 169.35, 169.20 and 168.91), five methine carbons  $(\delta_{\rm C}$  91.65, 72.74, 72.67, 70.67 and 67.72 ), one methylene carbon ( $\delta_{\rm C}$  61.41) and five methyl carbons [ $\delta_c$  20.76, 20.65 and 20.52 (3xC)]. The HMBC experiment (Table 50) completely confirmed the assigned structure. These data was in agreement with those of the authentic compound. Therefore, ART27 was clearly identified to be  $\beta$ -D-glucopyranoside penta-acetate.

Position	δ <sub>C</sub> (Type)	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	HMBC correlations
1	91.65 (CH)	5.72 ( <i>d</i> ; 9.3)	C-3, C-5, 1-C=O
2	70.67 (CH)	5.11 ( <i>t</i> ; 9.3)	C-1, C-3, 2-C=O
3	72.74 (CH)	5.23 ( <i>t</i> ; 9.3)	C-1, C-2, C-4, 3-C=O
4	67.72 (CH)	5.13 ( <i>t</i> ; 9.3)	C-3, C-5, C-6, 4-C=O
5	72.67 (CH)	3.85 ( <i>ddd</i> ; 9.3, 4.5, 2.1)	C-1, C-3, C-4, C-6
6	61.41 (CH <sub>2</sub> )	4.30 ( <i>dd</i> , 12.6, 4.5)	C-4, C-5, 6-C=O
		4.10 ( <i>dd</i> , 12.6, 2.1)	
1-OAc	168.91 (C=O)		
	20.76 (CH <sub>3</sub> )	2.11 (s)	1-C=O
2-OAc	169.35 (C=O)		
	20.52 (CH <sub>3</sub> )	2.03 (s)	2-C=O
3-OAc	170.04 (C=O)		
	20.52 (CH <sub>3</sub> )	2.01 (s)	3-C=O
4-OAc	169.20 (C=O)		
	20.52 (CH <sub>3</sub> )	2.03 (s)	4-C=O
6-OAc	170.55 (C=O)		
	20.65 (CH <sub>3</sub> )	2.08 (s)	6-C=O

Table 50The NMR spectral data of ART27

**ART29:** sucrose octa-acetate



ART29 was obtained as a white solid. The IR spectrum exhibited absorptions at 1757 and 1747 cm<sup>-1</sup> for a carbonyl stretching. The <sup>13</sup>C NMR spectrum (Table 51) displayed the resonances of twelve oxygenated carbon signals of one quaternary at  $\delta_{\rm C}$  103.94, eight methine carbons at  $\delta_{\rm C}$  89.87, 79.06, 75.64, 74.94, 70.21, 69.56, 68.44 and 68.15, and three methylene carbons at  $\delta_{\rm C}$  63.57, 62.81 and 61.71. The spectrum further showed the presence of eight acetyl groups [ $\delta_{\rm C}$  170.62, 170.42, 170.03 (2xC), 169.96, 169.83, 169.59, 169.44, and 20.61 (8xCH<sub>3</sub>)]. These data indicated that ART29 was a disaccharide octa-acetate. The <sup>1</sup>H NMR spectrum (Table 51) revealed the characteristic signals of  $\alpha$ -D-glucose [ $\delta_{\rm H}$  5.69 (d, J = 3.6 Hz, H-1), 5.44 (*t*, *J* = 10.5 Hz, H-3), 5.08 (*t*, *J* = 10.5 Hz, H-4), 4.87 (*dd*, *J* = 10.5 and 3.6 Hz, H-2), 4.14 (*m*, H<sub>2</sub>-6), 4.28 (*m*, H<sub>2</sub>-6) and 4.26 (*m*, H-5)] and fructose [ $\delta_{\rm H}$  5.45 (*d*; J = 5.7 Hz, H-3'), 5.37 (*t*; *J* = 5.7 Hz, H-4'), 4.31 (*dd*; *J* = 11.7 and 4.5 Hz, H<sub>2</sub>-6'), 4.24 (*m*, H<sub>2</sub>-6'), 4.21 (m, H-5') and 4.18 (s, H<sub>2</sub>-1')]. The resonances of eight acetyl protons [ $\delta_{\rm H}$  2.18, 2.12 (2xCH<sub>3</sub>), 2.11, 2.10 (2xCH<sub>3</sub>), 2.05 and 2.02] were also observed. The HMBC experiment (Table 51) clearly confirmed the assigned structure. Its <sup>1</sup>H NMR data was in agreement with the data of authentic compound. ART29 thus was confirmed as sucrose octa-acetate.

δ <sub>C</sub> (Type)	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	HMBC correlations
89.87 (CH)	5.69 ( <i>d</i> , 3.6)	C-3, C-5, C-2'
70.21 (CH)	4.87 ( <i>dd</i> , 10.5, 3.6)	C-3, C-4, 2-C=O
69.56 (CH)	5.44 ( <i>t</i> , 10.5)	C-1, C-2, C-4, 3-C=O
68.15 (CH)	5.08 ( <i>t</i> , 10.5)	C-3, C-5, C-6, 4-C=O
68.44 (CH)	4.26 ( <i>m</i> )	
61.71 (CH <sub>2</sub> )	4.14 ( <i>m</i> ), 4.28 ( <i>m</i> )	
170.03 (C=O)		
20.61 (CH <sub>3</sub> )	2.10 (s)	2-C=O
169.59 (C=O)		
20.61 (CH <sub>3</sub> )	2.02 (s)	3-С=О
169.44 (C=O)		
20.61 (CH <sub>3</sub> )	2.05 (s)	4-C=O
170.62 (C=O)		
20.61 (CH <sub>3</sub> )	2.10 ( <i>s</i> )	6-C=O
62.81 (CH <sub>2</sub> )	4.18 ( <i>s</i> )	C-2', C-3', 1'-C=O
103.94 (C)		
75.64 (CH)	5.45 ( <i>d</i> ; 5.7)	C-1', C-4', C-5', 3'-C=O
74.94 (CH)	5.37 ( <i>t</i> ; 5.7)	C-3', C-5', C-6', 4'-C=O
79.06 (CH)	4.21 ( <i>m</i> )	
63.57 (CH <sub>2</sub> )	4.31 ( <i>dd</i> ; 11.7, 4.5)	C-4', C-5', 4'-C=O
	4.24 ( <i>m</i> )	
170.03 (C=O)		
20.61 (CH <sub>3</sub> )	2.12 (s)	1'-C=O
1		

3'-C=O

4'-C=O

Table 51The NMR spectral data of ART29

Position

6

2-OAc

3-OAc

4-OAc

6-OAc

1'-OAc

3'-OAc

4'-OAc

169.83 (C=O)

20.61 (CH<sub>3</sub>)

169.96 (C=O)

20.61 (CH<sub>3</sub>)

2.18 (s)

2.11 (s)

1' 2' 3' 4' 5'

6'-OAc	170.42 (C=O)		
	20.61 (CH <sub>3</sub> )	2.12 ( <i>s</i> )	6'-C=O

## 3.2 Antibacterial activity of some of the isolated phloroglucinols

The extracts from the leaves and stems of *R. tomentosa* were tested on antibacterial activity against *S. aureus* and MRSA. The results indicated that the crude CH<sub>2</sub>Cl<sub>2</sub> and Me<sub>2</sub>CO extracts from its leaves exhibited strong activity against *S. aureus* and MRSA with MIC values of 31.25 and 62.5  $\mu$ g/mL, respectively, The CH<sub>2</sub>Cl<sub>2</sub> and Me<sub>2</sub>CO extracts from its stems were found to show no activities (**Table 52**).

Some of the isolated phloroglucinols were also evaluated for their antibacterial activity against S. aureus, MRSA, S. pyogenes and E. coli. The results indicated that rhodomyrtosones A-G and rhodomyrtone exhibited activities against three types of Gram-positive bacteria, S. aureus, MRSA and S. pyogenes (Table 53). Rhodomyrtosone B (ART9), rhodomyrtosone D (ART11), rhodomyrtone (ART6) and rhodomyrtosone G (ART19) were able to inhibit the growth of S. aureus with MIC values of 6.25, 12.5, 0.39 and 1.56  $\mu$ g/mL and MRSA with MIC values of 12.5, 25, 0.39 and 1.56  $\mu$ g/mL, respectively. Rhodomyrtosone B (**ART9**), rhodomyrtosone D (**ART11**) and rhodomyrtone (ART6) further showed the inhibitory activity against S. pyogenes with MIC values of 3.125, 12.5 and 0.39  $\mu$ g/mL, respectively. Compound **ART19** was not tested against S. pyogenes due to insufficient amount. Rhodomyrtosones A (ART2), C (ART4), E (ART18) and F (ART20) showed no activity at the concentration 100  $\mu g/mL$ . Rhodomyrtone (ART6), which is the most active compound, provided stronger antibacterial activity than the reference antibiotic, vancomycin. No activity was observed for rhodomyrtosones A-F (ART2, ART9, ART4, ART11, ART18 and ART20) and rhodomyrtone (ART6) against Gram-negative bacteria tested, E. coli ATCC 25922 and E. coli O157:H7 (RIMD 05091078), at the concentration 100  $\mu g/mL$ .

extract	MIC (µ	<i>ı</i> g/mL)
extract	S. aureus	MRSA
Leaves		
CH <sub>2</sub> Cl <sub>2</sub> extract	31.25	31.25
Me <sub>2</sub> CO extract	62.5	62.5
Stems		
CH <sub>2</sub> Cl <sub>2</sub> extract	NA	NA
Me <sub>2</sub> CO extract	NA	NA
vancomycin	0.60	1.25

 Table 52
 Antibacterial activity of the extracts from R. tomentosa

NA = no activity

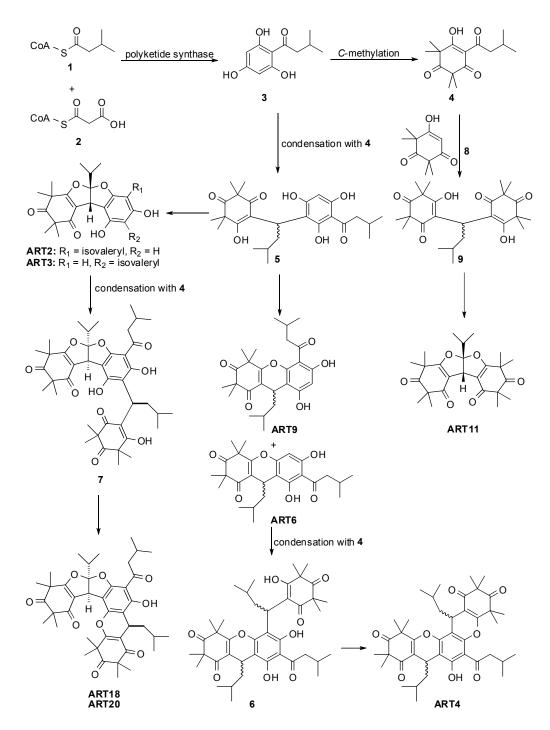
 Table 53
 Antibacterial activity of acylphloroglucinols isolated from R. tomentosa

Compound		MIC (µg/mL)				
Compound	S. aureus	MRSA	S. pyogenes			
ART2	>100	>100	>100			
ART4	>100	>100	>100			
ART6	0.39	0.39	0.39			
ART9	6.25	12.5	3.125			
ART11	12.5	25	12.5			
ART18	>100	>100	>100			
ART19	1.56	1.56	-			
ART20	>100	>100	>100			
vancomycin	0.60	1.25	-			
penicillin G	-	-	0.015			

- = not tested

### 3.3 Biosynthetic proposal of some of the isolated phloroglucinols

The biosynthetic pathway of ART2-4, ART9, ART11, and ART18-ART20 can now be proposed. Isovalerylphloroglucinol (3) was produced from condensation of isovaleryl-CoA (1) (Mahmud et al., 2002; Bode et al., 2009) with three units of malonyl-CoA (2) by a polyketide synthase (Paniego et al., 1999) and was further C-methylated (Birch et al., 1966) to give leptospermone (4). An intermediate 5 was then formed by condensation of 3 and 4. This compound is not known and has not been isolated from our extract. However, it is closely related to 4-cyclohexene-1,3dioxo-5-hydroxy-2,2,6,6-tetramethyl-4-{1-[2,6-dihydroxy-4-methoxy-3-(3-methyl-1oxobutyl)phenyl]-3-methylbutyl}(Bloor, 1992). Cyclisation of an hydroxyl group of the isovaleryl-phloroglucinol moiety with a carbonyl group of the  $\beta$ -triketone moiety forming the hemiketal intermediate and then dehydration (Dachriyanus et al., 2002) gave ART9 (rhodomyrtosone B) and ART6 (rhodomyrtone). The Aldol-like condensation of ART6 (rhodomyrtone) or ART9 (rhodomyrtosone B) with leptospermone (4) giving trimeric acylphloroglucinol (3), following by formation of benzopyran produced ART4 (rhodomyrtosone C). Oxidation of an isobutyl group followed by formation of bisfuran via cyclisation and dehydration of 5 could give ART2 (rhodomyrtosone A) and ART3 (rhodomyrtosone H). Further reacted of ART3 (rhodomyrtosone A) with 4 finally gave ART18 (rhodomyrtosone E) and ART20 (rhodomyrtosone F). ART11 (rhodomyrtosone D), a symmetric molecule containing two  $\beta$ -triketone moieties, may be obtained by oxidation of an isobutyl group of the intermediate 9, followed by formation of bisfuran. The intermediate 9 can be plausibly derived by condensation of leptospermone (4) and syncapic acid (8). The biosynthetic pathway of **ART19** (rhodomyrtosone G) would proceed in the same manner as for ART6 (rhodomyrtone), with 2-methylbutyryl-CoA as a precursor.



Proposed biosynthetic relationships of ART2-4, ART6, ART9, ART11,

# ART18 and ART20

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APPENDIX

Position	$\delta_{\rm C}$	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	Position	$\delta_{\rm C}$	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )
1	198.56		11	24.58	1.44 (s)
2	56.05		12	24.58	1.56 ( <i>s</i> )
3	212.16		13	24.72	1.42 (s)
4	47.23		1'	206.75	
4a	167.65		2'	53.18	3.03 ( <i>dd</i> ; 15.5,
					6.8)
					2.97 ( <i>dd</i> ; 15.5,
					6.8)
4b	155.63		3'	25.15	2.28 ( <i>m</i> )
5	94.74	6.19 ( <i>s</i> )	4'	22.81	0.98 ( <i>d</i> ; 6.3)
6	158.70		5'	22.74	0.98 ( <i>d</i> ; 6.3)
7	107.63		1"	45.82	1.48 ( <i>m</i> )
8	162.84		2"	25.10	1.48 ( <i>m</i> )
8a	106.30		3"	23.16	0.87 ( <i>d</i> ; 5.7)
9	25.19	4.30 ( <i>t</i> ; 5.5)	4"	22.53	0.84 ( <i>d</i> ; 5.7)
9a	114.26		6-OH		8.08 (s)
10	24.21	1.39 (s)	8-OH		13.39 (s)

**Table 54** The NMR spectral data of **rhodomyrtone** 

Position	$\delta_{ m C}$	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	Position	$\delta_{\rm C}$	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$
2	155.53		1'	125.40	
3	139.34		2', 6'	106.00	7.35 (s)
4	178.70		3', 5'	153.06	
4a	106.00		4'	140.54	
5	161.96		3-OCH <sub>3</sub>	60.27	3.861 (s)
6	97.87	6.34 ( <i>d</i> ; 2.2)	7-OCH <sub>3</sub>	55.79	3.864 (s)
7	165.51		3', 5'-OCH <sub>3</sub>	56.28	3.932 (s)
8	92.18	6.43 ( <i>d</i> ; 2.2)	4'-OCH <sub>3</sub>	60.95	3.935 (s)
8a	156.65		5-ОН		12.56 (s)
	1				1

**Table 55** The NMR spectral data of combretol

 Table 56
 The NMR spectral data of 3',5-dihydroxy-3,4',5',7-tetramethoxyflavone

Position	δ <sub>C</sub>	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	Position	$\delta_{\rm C}$	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )
2	155.3		2'	108.6	7.28 ( <i>d</i> ; 2.0)
3	139.7		3'	149.2	
4	178.8		4'	137.8	
4a	106.1		5'	152.0	
5	162.0		6'	105.1	7.20 ( <i>d</i> ; 2.0)
6	98.0	6.39 ( <i>d</i> ; 2.0)	3-OCH <sub>3</sub>	60.3	
7	165.6		7-OCH3	55.8	3.87, 3.86,
8	92.2	6.75 ( <i>d</i> ; 2.0)	4'-OCH <sub>3</sub>	61.1	3.82, 3.77
8a	156.8		5'-OCH <sub>3</sub>	56.1	
1'	126.0				

Position	$\delta_{\mathrm{C}}$	$\delta_{\mathrm{H}}$ (mult.)	Position	δ <sub>C</sub>	$\delta_{\rm H}$ (mult.)
1	112.30		3'	139.90	
2	141.50		4'	152.52	
3	140.50		5'	111.58	7.51 (s)
4	153.70		6'	112.05	
5	107.49	7.60 ( <i>s</i> )	7'	158.00	
6	114.50		3-OCH <sub>3</sub>	61.16	4.02 ( <i>s</i> )
7	157.50		4-OCH <sub>3</sub>	56.80	3.98 (s)
1'	111.00		3'-OCH <sub>3</sub>	60.89	4.04 ( <i>s</i> )
2'	141.00				

 Table 57 The NMR spectral data of 3,3',4-tri-O-methylellagic acid

Table 58 The NMR spectral data of 4-O-[β-D-glucopyranosyltetraacetate]-3,3',4'-tri-O-methylellagic acid

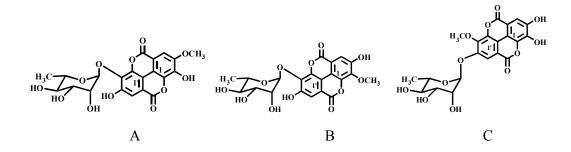
Position	$\delta_{\rm C}$	$\delta_{\mathrm{H}}$ (mult.)		Position	$\delta_{\rm C}$	$\delta_{\rm H}$ (mult.)
1	114.98			3'	140.90	
2	141.30			4'	154.60	
3	141.56			5'	107.65	7.56 (s)
4	151.28			6'	114.98	
5	113.34	8.00 ( <i>s</i> )		7'	158.10	
6	112.36			3-OCH <sub>3</sub>	62.06	4.02 (s)
7	158.10			3'-OCH <sub>3</sub>	61.80	3.96 ( <i>s</i> )
1'	112.17			4'-OCH <sub>3</sub>	56.64	3.36 ( <i>s</i> )
2'	141.23		130			

recorded in  $C_6D_6$  for <sup>1</sup>H and CDCl<sub>3</sub> for <sup>13</sup>C

Table 59 The <sup>1</sup>H NMR spectral data of 4-O-methylellagic acid 3'-α-rhamnoside(A), 3-O-methylellagic acid 3'-O-α-rhamnopyranoside (B) and 3-O-<br/>methyl ellagic acid 4-O-α-rhamnopyranoside (C)

Position	$\mathbf{A}^{b}$	$\mathbf{B}^{b}$	$\mathbf{C}^{c}$
1 USICION	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )
5	7.20 ( <i>s</i> )	7.46 ( <i>s</i> )	7.54 (s)
5'	7.30 ( <i>s</i> )	7.47 (s)	7.71 ( <i>s</i> )
1"	5.52 (br s)	5.73 ( <i>d</i> , 1.6)	5.49 ( <i>d</i> , 1.2)
2"	4.20 (br s)	4.34 ( <i>dd</i> , 1.6, 3.4)	4.02 ( <i>br</i> )
3"	3.96 ( <i>dd</i> , 2.5, 9.8)	4.05 ( <i>dd</i> , 3.4,9.6)	3.84 ( <i>dd</i> , 9.4, 3.0)
4"	3.41 ( <i>t</i> , 9.8)	3.46 ( <i>t</i> , 9.6)	3.34 ( <i>t</i> , 9.4)
5"	4.25 ( <i>dd</i> , 6.1, 9.8)	4.46 ( <i>dq</i> , 9.6, 6.2)	3.56 ( <i>m</i> )
6"	1.14 ( <i>d</i> , 6.1)	1.21 ( <i>d</i> , 6.2)	1.15 ( <i>d</i> , 6.0)
OCH <sub>3</sub>	3.85 (s)	4.12 ( <i>s</i> )	4.05 (s)

<sup>a</sup>recorded in CDCl<sub>3</sub>+DMSO-*d*<sub>6</sub>; <sup>b</sup>recorded in CD<sub>3</sub>OD; <sup>c</sup>recorded in DMSO-*d*<sub>6</sub>



# Table 60 The <sup>13</sup>C NMR spectral data of 4-O-methylellagic acid 3'-α-rhamnoside(A), 3-O-methylellagic acid 3'-O-α-rhamnopyranoside (B) and 3-O-<br/>methylellagic acid 4-O-α-rhamnopyranoside (C)

Position	A <sup>b</sup>	$\mathbf{B}^{b}$	$\mathbf{C}^{c}$
rosition	δ <sub>C</sub>	$\delta_{\rm C}$	$\delta_{ m C}$
1	113.1	112.3	115.12
2	138.2	142.3	137.07
3	152.0	143.5	140.74
4	154.0	154.4	153.31
5	107.6	114.0	112.23
6	115.2	113.3	113.84
7	161.0	161.8	159.52
1'	113.1	110.6	111.99
2'	143.8	143.6	137.07
3'	138.2	140.3	142.63
4'	154.0	153.7	147.51
5'	112.6	115.2	112.83
6'	115.2	114.6	108.14
7'	160.8	161.8	159.60
1"	103.7	102.9	100.66
2"	71.9	72.0	70.61
3"	72.1	72.2	70.80
4"	73.6	73.9	72.53
5"	71.9	71.7	70.83
6"	17.9	17.9	18.60
OCH <sub>3</sub>	57.1	61.6	61.62

<sup>a</sup>recorded in CDCl<sub>3</sub>+DMSO-*d*<sub>6</sub>; <sup>b</sup>recorded in CD<sub>3</sub>OD; <sup>c</sup>recorded in DMSO-*d*<sub>6</sub>

D		Α		В
Position	$\delta_{\rm C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	$\delta_{\rm C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
1	36.0		36.0	
2	47.4	2.32 ( <i>d</i> ; 16.8)	47.4	2.34 ( <i>d</i> ; 17.0)
		2.06 ( <i>d</i> ; 16.8)		2.08 ( <i>d</i> ; 17.0)
3	199.0		199.0	
4	125.7	5.89 (s)	125.7	5.89 (s)
5	161.7		161.7	
6	55.3	2.54 ( <i>d</i> ; 8.2)	55.3	2.52 ( <i>d</i> ; 8.6)
7	126.6	5.54 ( <i>dd</i> ; 15.5, 8.2)	126.6	5.53 ( <i>dd</i> ; 16.0, 8.8)
8	138.5	5.56 ( <i>dd</i> ; 15.5, 5.6)	138.5	5.67 ( <i>dd</i> ; 16.0, 6.0)
9	68.2	4.34 ( <i>m</i> )	68.2	4.35 ( <i>m</i> )
10	23.5	1.28 ( <i>d</i> ; 6.6)	23.5	1.30 ( <i>d</i> ; 6.4)
11	27.8	1.03 (s)	27.7	1.03 (s)
12	27.0	0.97 ( <i>s</i> )	26.9	0.97 (s)
13	23.4	1.88 ( <i>d</i> ; 1.2)	23.4	1.90 (s)

Table 61 The NMR spectral data of (6R,7E,9R)-9-hydroxy-4,7-megastigmadien-3-one (A) and (6R,7E,9S)-9-hydroxy-4,7-megastigmadien-3-one (B)

Position		Loliolide	Isololiolide	
rosition	$\delta_{ m C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	
1	36.3			
2	47.7	1.97 ( <i>ddd</i> ; 14.5, 3.0, 2.3)	2.03 ( <i>ddd</i> ; 12.9, 4.4, 2.2)	
		1.53 ( <i>dd</i> ; 14.7, 3.7)	1.33 ( <i>dd</i> ; 12.1,12.1)	
3	67.2	4.33 (quin; 3.4)	4.13 ( <i>tt</i> ; 11.6, 4.5)	
4	46.0	2.46 ( <i>ddd</i> ; 14.0, 3.2, 2.3)	2.53 ( <i>ddd</i> ; 11.5, 4.0, 2.2)	
		1.78 ( <i>dd</i> ; 13.5, 3.7)	1.51 ( <i>dd</i> ; 11.7, 11.5)	
5	87.1			
6	182.8			
7	113.3	5.69 (s)	5.71 (s)	
8	172.3			
9	26.9	1.47 (s)	1.26 (s)	
10	31.1	1.27 (s)	1.31 (s)	
11	27.4	1.78 (s)	1.58 (s)	

 Table 62
 The NMR spectral data of loliolide and isololiolide

Position	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	Position	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )
2	3.98 (sext; 10.0, 10.0, 4.0)	27	1.20 (s)
3	4.70 ( <i>d</i> ; 10.0)	29	0.96 (s)
12	5.36 ( <i>t</i> ; 4.0)	30	0.96 (s)
18	2.97 ( <i>dd</i> ; 14.0, 4.0)	2'	6.29 ( <i>d</i> ; 16.0)
23	0.96 (s)	3'	7.63 ( <i>d</i> ; 16.0)
24	0.96 (s)	5', 9'	7.38 ( <i>d</i> ; 8.0)
25	1.03 (s)	6', 8'	6.88 ( <i>d</i> ; 8.0)
26	0.84 ( <i>s</i> )		

**Table 63** The NMR spectral data of *3β-O-E*-coumaroylmaslinic acid

**Table 64** The NMR spectral data of **3β-O-Z-coumaroylmaslinic acid** 

Position	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	Position	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )
2	3.84 (sext; 10, 10, 4.0)	27	1.17 (s)
3	4.61 ( <i>d</i> ; 10.0)	29	0.93 (s)
12	5.32 ( <i>t</i> ; 4.0)	30	0.98 (s)
18	2.95 ( <i>dd</i> ; 14.0, 4.0)	2'	5.82 ( <i>d</i> ; 12.0)
23	0.98 (s)	3'	6.82 ( <i>d</i> ; 12.0)
24	0.98 (s)	5', 9'	7.66 ( <i>d</i> ; 8.0)
25	0.93 (s)	6', 8'	6.83 ( <i>d</i> ; 8.0)
26	0.84 ( <i>s</i> )		
1			

recorded in CDCl<sub>3</sub>+pyridine- $d_5$  (10:1)

Position	$\delta_{ m C}$	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	Position	δ <sub>C</sub>	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
1	37.4		20	30.3	
2	23.1		21	36.5	
3	79.8	4.90 ( <i>dd</i> ; 11.4,	22	32.3	
		5.5)			
4	37.5		23	27.6	7 quaternary Me
					(23-27, 29 and
					30)
5	54.9		24	14.7	0.86 (s), 0.96 (s),
6	17.8		25	16.5	0.98 (s, 2xCH <sub>3</sub> ),
7	32.5		26	16.7	1.01 (s), 1.02 (s),
8	39.0		27	25.5	1.30 (s)
9	47.2		28	179.3	
10	37.5		29	33.5	
11	23.4		30	23.1	
12	121.6	5.49 ( <i>br</i> s)	1'	166.4	
13	144.1		2'	115.0	6.71 ( <i>d</i> ; 16.1)
14	41.4		3'	144.0	8.04 ( <i>d</i> ; 16.1)
15	27.6		4'	125.4	
16	23.0		5', 8'	129.9	7.67 ( <i>d</i> ; 8.4)
17	45.9		6', 9'	116.0	7.18 ( <i>d</i> ; 8.4)
18	41.3		7'	160.6	
19	45.7				

**Table 65** The NMR spectral data of *3β-O-E*-coumaroyloleanolic acid

recorded in pyridine- $d_5$ 

Position	arjunolic acia <sup>a</sup>			hyptatic acid-A <sup>b</sup>		
rosition	$\delta_{\rm C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	δ <sub>C</sub>	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )		
1	47.3		46.7			
2	68.7	4.21 (overlap)	68.1	3.79 ( <i>ddd</i> ; 12.0, 10.0,		
				5.0)		
3	78.1	4.20 (overlap)	84.5	3.39 ( <i>d</i> ; 10.0)		
4	43.5		39.1			
5	48.3		55.7			
6	18.6		18.4			
7	33.6		32.2			
8	40.0		48.1			
9	47.7		47.3			
10	38.4		37.7			
11	24.2		22.6			
12	123.2	4.23 (overlap)	121.6	5.25 ( <i>br</i> s)		
13	144.5		144.2			
14	42.1		46.0			
15	28.3		24.9			
16	24.8		22.3			
17	47.9		46.4			
18	41.2	3.26 ( <i>dd</i> ; 10.5, 4.0)	41.4	2.88 ( <i>dd</i> ; 12.0, 5.0)		
19	45.9		42.9			
20	29.1		30.2			
21	35.6		33.6			
22	32.9		32.8			
23	66.3	a: 4.19 ( <i>d</i> , 10.4)	27.4	a: 4.04 ( <i>d</i> ; 11.0)		
		b: 3.71 ( <i>d</i> , 10.4)		b: 3.06 ( <i>d</i> ; 11.0)		

 Table 66
 The NMR spectral data of arjunolic acid and hyptatic acid-A

<sup>a</sup>recorded in pyridine-d<sub>5</sub>; <sup>b</sup>recorded in CD<sub>3</sub>OD

Position	arjunolic acia <sup>a</sup>		hy	ptatic acid-A <sup>b</sup>
rosition	$\delta_{\rm C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	$\delta_{\mathrm{C}}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
24	14.2		64.7	1.24 (s)
25	17.6		16.0	0.82 (s)
26	17.2		16.3	1.00 (s)
27	28.8		23.4	1.17 (s)
28	180.5		180.2	
29	34.6		32.5	0.95 (s)
30	24.8		22.7	0.91 (s)

<sup>*a*</sup>recorded in pyridine-*d*<sub>5</sub>; <sup>*b*</sup>recorded in CD<sub>3</sub>OD

Position	$\delta_{\rm C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	Position	$\delta_{ m C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
1	38.5		16	23.4	
2	27.1		17	46.6	
3	79.0	3.21 ( <i>m</i> )	18	41.0	
4	38.7		19	45.8	
5	55.2		20	30.7	
6	18.3		21	33.8	
7	32.6		22	32.4	
8	38.9		23	28.1	0.98 (s)
9	47.6		24	15.5	0.75 (s)
10	37.0		25	15.3	0.89 (s)
11	22.9		26	16.8	0.74 ( <i>s</i> )
12	122.6	5.26 ( <i>t</i> ; 3.0)	27	26.0	1.11 (s)
13	143.5		28	179.1	
14	41.6		29	33.0	0.91 (s)
15	27.6		30	23.6	0.96 (s)
	1		1	1	1

 Table 67
 The NMR spectral data of oleanolic acid

recoded in CDCl<sub>3</sub>

Position	$\delta_{ m C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	Position	$\delta_{ m C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
1	37.2		16	28.2	
2	31.9		17	56.0	
3	71.8	3.53 <i>(m)</i>	18	12.0	0.69 (s)
4	42.3		19	19.0	1.01 ( <i>m</i> )
5	140.7		20	36.1	
6	121.7	5.35 ( <i>d</i> , 5.2)	21	18.8	0.97 ( <i>d</i> , 6.5)
7	31.9		22	33.9	
8	31.7		23	26.1	
9	50.1		24	45.8	
10	36.5		25	29.1	
11	21.1		26	19.4	0.92 ( <i>d</i> , 6.5)
12	39.8		27	19.8	0.81 ( <i>d</i> , 6.6)
13	42.3		28	23.1	
14	56.7		29	11.9	0.84 ( <i>t</i> , 7.5)
15	24.3				

**Table 68** The NMR spectral data of  $\beta$ -sitosterol

recorded in CDCl<sub>3</sub>

Position	$\delta_{ m C}$	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )	Position	$\delta_{ m C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
1	36.8	α: 1.09 ( <i>t</i> ; 13)	17	55.5	1.07-1.17 ( <i>m</i> )
		β: 1.88 (d; 13)	18	11.1	0.70 ( <i>s</i> )
2	29.0	a: 1.57-1.66 ( <i>m</i> )	19	18.5	1.03 (s)
		b: 1.90-1.97 ( <i>m</i> )	20	35.6	1.34-1.42 ( <i>m</i> )
3	78.4	3.57-3.65 ( <i>m</i> )	21	18.0	0.94 ( <i>d</i> ; 6.4)
4	38.1	a: 2.24-2.32 ( <i>m</i> )	22	33.4	a: 1.00-1.07 ( <i>m</i> )
		b: 2.39-2.45 ( <i>m</i> )			b: 1.32-1.39 ( <i>m</i> )
5	139.9		23	25.4	1.15-1.22 ( <i>m</i> )
6	121.4	5.36-5.39 ( <i>m</i> )	24	45.4	0.91-0.98 ( <i>m</i> )
7	31.4	a: 1.53-1.59 ( <i>m</i> )	25	28.6	1.64-1.72 ( <i>m</i> )
		b: 1.95-2.03 ( <i>m</i> )	26	18.1	0.83 ( <i>d</i> ; 6.5)
8	31.4	1.43-1.50 ( <i>m</i> )	27	18.9	0.85 ( <i>d</i> ; 6.5)
9	49.7	0.90-0.99 ( <i>m</i> )	28	22.5	1.21-1.32 ( <i>m</i> )
10	36.2		29	11.1	0.86 ( <i>t</i> ; 8.3)
11	20.5	1.45-1.55 ( <i>m</i> )	1′	100.6	4.41 ( <i>d</i> ; 7.8)
12	39.3	<i>α</i> : 1.19 ( <i>t</i> ; 11.5)	2'	73.1	3.22 ( <i>t</i> ; 8.3)
		β: 2.00-2.06 (m)	3'	76.1	3.38-3.44 ( <i>m</i> )
13	41.8		4'	69.7	3.35-3.42 ( <i>m</i> )
14	56.3	0.97-1.07 ( <i>m</i> )	5'	75.6	3.27-3.31 ( <i>m</i> )
15	23.7	a: 1.04-1.13 ( <i>m</i> )	6'	61.1	a: 3.73 ( <i>dd</i> ; 12.0,
		b: 1.57-1.64 ( <i>m</i> )			5.1)
					b: 3.86 ( <i>dd</i> ; 12.0,
					2.6)
16	27.7	a: 1.24-1.32 ( <i>m</i> )			
		b: 1.83-1.92 ( <i>m</i> )			

**Table 69** The NMR spectral data of  $\beta$ -sitosterol glucopyranoside (daucosterol)

Position	$\delta_{\rm C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	Position	$\delta_{\rm C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
1	35.7		16	28.1	
2	33.9		17	56.1	
3	198.9		18	12.0	0.72 ( <i>s</i> )
4	123.6	5.74 ( <i>d</i> ; 2.2)	19	17.4	1.19 (s)
5	171.0		20	36.1	
6	32.9		21	18.7	0.93 ( <i>d</i> ; 6.6)
7	32.1		22	34.0	
8	35.7		23	26.0	
9	53.8		24	45.8	
10	38.6		25	29.1	
11	21.0		26	19.8	0.84 ( <i>d</i> ; 6.8)
12	39.5		27	19.2	0.82 ( <i>d</i> ; 6.8)
13	42.4		28	23.1	
14	55.9		29	11.4	0.85 ( <i>t</i> ; 7.2)
15	24.1				

 Table 70 The NMR spectral data of stimast-4-en-3-one

Position	$\delta_{\rm C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	Position	$\delta_{ m C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )
1	167.39		1'	64.59	4.12 <i>(t)</i>
2	114.65	6.23 ( <i>d</i> ; 15.8)	2'	31.94	
3	146.71	7.61 ( <i>d</i> ; 15.8)	3'-29'	29.68,	1.26 (27xCH <sub>2</sub> )
4	127.11			25.98,	
5	109.35	6.94 ( <i>s</i> )		22.70	
6	147.84		30'	14.12	0.89 ( <i>t</i> )
7	144.67		7-OH		5.84 (s)
8	122.99	7.15 ( <i>d</i> )	6-OCH <sub>3</sub>	55.94	3.93 (s)
9	115.72	7.15 ( <i>d</i> )			

 Table 71
 The NMR spectral data of *trans*-triacontyl-4-hydroxy-3-methoxy-cinnamate

**Table 72** The NMR spectral data of *trans*-triacontyl-4-hydroxycinnamate

Position	$\delta_{\rm C}$	$\delta_{ m H}$ (mult.; $J_{ m Hz}$ )	Position	$\delta_{ m C}$	$\delta_{\mathrm{H}}$ (mult.; $J_{\mathrm{Hz}}$ )
1	169.3	-	2'	26.1	
2	110.5	6.29 ( <i>d</i> ; 16.0)	3'-29'	31.9,	1.25 ( <i>br</i> )
3	144.4	7.63 ( <i>d</i> ; 16.0)		30.8,	
4	127.5			29.6,	
5,9	129.9	7.43 ( <i>d</i> ; 8.0)		29.4,	
6, 8	115.9	6.85 ( <i>d</i> ; 8.0)		29.3,	
7	152.3			22.6	
1'	64.6	4.19 ( <i>t</i> ; 7.5)	30′	14.0	0.93 ( <i>t</i> ; 6.7)

Position	$\delta_{ m C}$	$\delta_{\rm H}$ (mult.; $J_{\rm Hz}$ )
1,1′	131.7	
2,2'	111.3	6.53 ( <i>d</i> , 1.8)
3,3'	146.5	
4,4'	144.0	
5,5'	114.2	6.81 ( <i>d</i> , 8.1)
6,6'	121.8	6.61 ( <i>dd</i> , 8.1, 1.8)
7,7′	35.3	2.75 ( <i>dd</i> , 14.1, 7.5)
		2.70 ( <i>dd</i> , 14.1, 7.5)
8,8′	40.2	2.22 ( <i>m</i> )
9,9′	64.5	4.39 ( <i>dd</i> , 11.4, 5.7)
		4.22 ( <i>dd</i> , 11.4, 5.5)
3,3′-OCH <sub>3</sub>	55.8	3.77 <i>(s)</i>
1'',1'''	126.9	
2'',2'''	109.5	7.01 ( <i>d</i> , 1.8)
3'',3'''	146.8	
4′′,4′′′	148.1	
5'',5'''	115.2	6.91 ( <i>d</i> , 8.2)
6'',6'''	123.1	7.06 ( <i>dd</i> , 8.2, 1.8)
7'',7'''	145.2	7.59 ( <i>d</i> , 15.9)
8'',8'''	114.8	6.28 ( <i>d</i> , 15.9)
9′′,9′′′	167.3	
3'',3'''-OCH <sub>3</sub>	56.0	3.92 (s)

 Table 73 The NMR spectral data of 9,9'-O-diferuloyl-(-)-secoisolariciresinol

## VITAE

Name	Mr. Asadhawut Hiranrat
Student ID	4910230011

# **Education Attainment**

Degree	Name of Institution	Year of Graduation
B.Sc.(Chemistry)	Prince of Songkla University	1998
M.Sc.(Organic Chemistry)	Prince of Songkla University	2001

#### **Scholarship Awards during Enrolement**

- Thailand Research Fund through the Royal Golden Jubilee Ph.D. Program (Grant No. PHD0206/2549)
- 2. Center for Innovation in Chemistry (PERCH-CIC), Office of the Higher Education Commission, Ministry of Education
- 3. Prince of Songkla University Graduate Studies Grant
- CHE-RES-RG, Office of the Higher Education Commission, Ministry of Education

## List of Publication and Proceeding

## **Publications**

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 Surasak Limsuwan, Erik N. Trip, Thijs R. H. M. Kouwen, Sjouke Piersma, Asadhawut Hiranrat, Wilawan Mahabusarakam, Supayang Piyawan Voravuthikunchai, Maarten van Dijl Jan, Oliver Kayser. 2009. "Rhodomyrtone: a new candidate as natural antibacterial drug from *Rhodomyrtus tomentosa*", *Phytomedicine*, 16 (6-7), 645-651.

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- Asadhawut Hiranrat and Wilawan Mahabusarakam. "Chemical Constituents of *Rhodomyrtus tomentosa* (Aiton) Hassk." The Sixth Regional IMT-GT Uninet Conference 2008, The Gurney Resort Hotel & Residences, Penang, Malaysia, August 28-30, 2008. (international; poster presentation)
- 3. **Asadhawut Hiranrat** and Wilawan Mahabusarakam. "New Acylphloroglucinols from the Leaves of *Rhodomyrtus tomentosa*" RGJ-Ph.D. Congress X, Jomtien Palm Beach Resort, Pattaya, Chonburi, Thailand, April 3-5, 2009. (international; oral presentation)
- 4. **Asadhawut Hiranrat** and Wilawan Mahabusarakam. "New Acylphloroglucinols from the Leaves of *Rhodomyrtus tomentosa*" PERCH-CIC Congress VI, Jomtien Palm Beach Resort, Pattaya, Chonburi, Thailand, May 3-6, 2009. (international; oral presentation)
- Asadhawut Hiranrat, Wilawan Mahabusarakam, Surasak Limsuwan and Supayang Piyawan Voravuthikunchai. "Chemical Constituents from the Leaves of *Rhodomyrtus tomentosa*" การสัมมนาวิชาการ "วิทยาศาสตร์ชีวภาพโมเลกุล-ปัจจุบันและ อนาคต", Faculty of Science, Mahidol University, Bangkok, Thailand, August 26, 2009. (meeting; poster presentation)

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