2 EXPERIMENTAL

2.1 Instruments and chemicals

Melting point was recorded in °C and was measured on an Electrothermal Infrared spectra were recorded using FTS FT-IR Melting Point Apparatus. spectrophotometer and major bands (ν) were recorded in wave number (cm⁻¹). Ultraviolet (UV) absorption spectra were recorded using UV-160A spectrophotometer (SHIMADZU) and principle bands (λ_{\max}) were recorded as wavelengths (nm) and log $\mathcal E$ in chloroform solution. Nuclear magnetic resonance spectra were recorded on a FTNMR Bruker Ultra Shield 300 MHz at Department of Chemistry, Faculty of Science, Prince of Songkla University. Spectra were recorded in deuterochloroform and DMSO-d6 solution and were recorded as δ value in ppm downfield from TMS (internal standard δ 0.00). Single-crystal X-ray diffraction measurements were collected using SMART 1-K CCD diffractometer with monochromated Mo-K lpharadiation ($\lambda = 0.71073 \text{ A}^{\circ}$) using ω -scan mode and SHELXTL for structure solution and refinement. Optical rotation was measured in chloroform solution with sodium D line (590 nm) on an AUTOPOL^R II automatic polarimeter. Solvent for extraction and chromatography were distillated at their boiling poimt ranges prior to use except diethyl ether was analytical grade reagent. Quick column chromatography was perfromed on silica gel 60 GF₂₅₄ (Merck). Column chromatography was performed on silica gel (Merck) type 100 (0.063 – 0.200).

2.2 Plant material

Seeds of Xylocarpus granatum were collected from The Mangrove Reseach Station at khanom district, Nakorn Si Thammarat province, Thailand. The plant was identified by Professor Puangpen Sirirugsa, Department of Biology, Faculty of Science, Prince of Songkla University and the voucher specimen was deposited in the herbarium (collection number XG01).

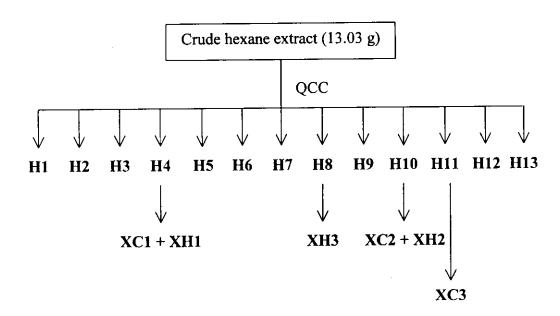
2.3 Extraction

Air-dried seeds (5.2 Kg) of *Xylocarpus granatum* were extracted with hexane, CH₂Cl₂ and MeOH, successively, at room temperature. The mixture was filtered and concentrated under reduced pressure to give the crude hexane (13.03 g), dichloromethane (68.67 g) and methanol extracts (299.98 g)

2.4 Isolation and Chemical Investigation

2.4.1 Investigation of the crude hexane extract from the seeds of X. granatum

The pale yellow viscous oil from the hexane extract (13.03 g) was purified by quick column chromatography (QCC) and eluted with gradient elution of hexane, EtOAc and MeOH which afforded twelve fractions (H1-H12).



Scheme 4 Isolation of compounds XC1, XC2, XC3, XH1, XH2 and XH3

Fraction H4, (0.3981 g) as pale yellow viscous oil, was subjected to repeated chromatography (CC) 10% EtOAc: hexane to give compounds XC1 (0.1222 g) and XH1(0.1214 g).

Fraction H8, (0.1201 g) as pale yellow viscous oil, was subjected to repeated chromatography (CC) 15% EtOAc: hexane to give compound XH3 (0.0307 g).

Fraction H10, (2.3270 g) as pale yellow viscous oil, was subjected to repeated chromatography (CC) 30% EtOAc: hexane to give compounds XC2 (1.5802 g) and XH2 (0.0138 g).

Fraction H11, (1.2430 g) as pale yellow viscous oil, was subjected to repeated chromatography (CC) 30% EtOAc: hexane to give compound XC3 (0.9580 g).

Compound XC1 (Colorless solid, mp: 263-265 °C)

 $[\alpha]_n^{26}$: -41° (c = 1.22, CHCl₃)

EIMS m/z: 438 (9) [M]⁺, 316 (94), 315 (100)

UV (CHCl₃) λ_{max} (nm): 225

IR (NaCl) V (cm⁻¹): 1741 (C=O stretching of lactone), 1709 (C=O stretching of

ketone), 1670 (C=O stretching of α,β-unsaturated carbonyl group)

 1 H NMR (CDCl₃) (δ ppm) (300 MHz): see Table 7

 13 C NMR (CDCl $_3$) (δ ppm) (75 MHz): see Table 7

DEPT (CDCl₃): see Table 4

Compound XC2 (Colorless solid, mp: 228-230 °C)

 $[\alpha]_{D}^{26}$: -40° (c = 1.31, CHCl₃)

EIMS m/z: 486 (21) [M]⁺, 349 (24), 348 (100), 330 (27), 149 (30), 135 (9), 121 (28)

UV (CHCl₃) λ_{max} (nm): 212

IR (NaCl) $V(cm^{-1})$: 3452 cm⁻¹ (O-H stretching) and 1716 cm⁻¹ (C=O stretching)

 1 H NMR (CDCl₃) (δ ppm) (300 MHz): see Table 13

 13 C NMR (CDCl₃) (δ ppm) (75 MHz): see Table 13

DEPT (CDCl₃): see Table 10

Compound XC3 (colorless oil)

 $[\mathbf{C}]_{\mathbf{D}}^{26}$: -46° (c = 0.96, CHCl₃)

UV (CHCl₃) λ_{max} (nm): 210

IR (NaCl) $V(cm^{-1})$: 3460 cm⁻¹ (O-H stretching) and 1732 cm⁻¹ (C=O stretching)

 1 H NMR (CDCl₃) (δ ppm) (300 MHz): see Table 19

 13 C NMR (CDCl₃) (δ ppm) (75 MHz): see Table 19

DEPT-135° (CDCl₃): see Table 16

Compound XH1 (colorless oil)

 $[\mathbf{\alpha}]_{\mathbf{n}}^{26}$: -44° (c = 0.02, CHCl₃)

CIMS m/z: 471 (81) [M]⁺, 439 (100), 411 (26), 375 (22), 357(15), 343 (15)

UV (CHCl₃) λ_{max} (nm): 209

IR (NaCl) V(cm⁻¹): 1733 (C=O stretching)

¹H NMR (CDCl₂) (δ ppm) (300 MHz): see Table 37

 $^{13}\mathrm{C}\ \mathrm{NMR}\ (\mathrm{CDCl_3})\ (\boldsymbol{\delta}\mathrm{ppm})\ (75\ \mathrm{MHz})$: see Table 37

DEPT-135° (CDCl₃): see Table 34

Compound XH2 (colorless oil)

 $[\mathbf{C}]_{\mathbf{D}}^{26}$: -40° (c = 0.01, acetone)

HRESIMS m/z: 781.2325 [M+Na]⁺, 759.2474 [M+H]⁺

UV (CHCl₃) λ_{max} (nm): 216

IR (NaCl) $V(cm^{-1})$: 3555 cm⁻¹ (O-H stretching) and 1733 cm⁻¹ (C=O stretching)

 1 H NMR (CDCl₃) (δ ppm) (300 MHz): see Table 43

 $^{13}\mathrm{C}\ \mathrm{NMR}\ (\mathrm{CDCl_3})\ (\boldsymbol{\delta}\mathrm{ppm})\ (75\ \mathrm{MHz})$: see Table 43

DEPT-135° (CDCl₂): see Table 40

Compound XH3 (colorless oil)

 $[\mathbf{\alpha}]_{D}^{26}$: -35° (c = 0.02, acetone)

CIMS *m/z*: 701 [M]⁺

UV (CHCl₃) λ_{max} (nm): 217

IR (NaCl) $V(cm^{-1})$: 3481 cm⁻¹ (O-H stretching) and 1736 cm⁻¹ (C=O stretching)

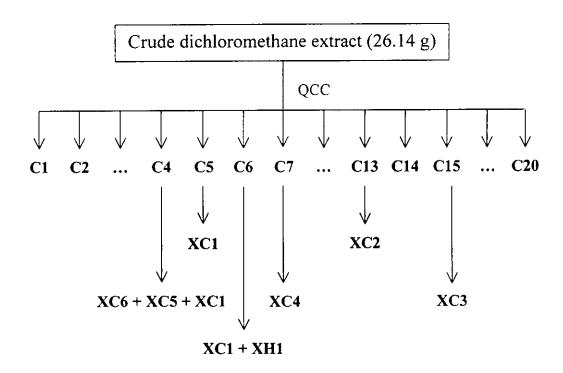
 1 H NMR (CDCl₂) (δ ppm) (300 MHz): see Table 49

 13 C NMR (CDCl₃) (δ ppm) (75 MHz): see Table 49

DEPT-135° (CDCl₃): see Table 46

2.4.2 Investigation of the crude dichloromethane extract from the seeds of X. granatum

The pale yellow viscous oil from the dichloromethane extract (26.14 g) was purified by quick column chromatography (QCC) and eluted with gradient elution of hexane, EtOAc and MeOH which afforded twenty fractions (H1-H20).



Scheme 5 Isolation of compounds XH1, XC1, XC2, XC3, XC4, XC5 and XC6

Fraction C4, (0.6145 g) as a pale yellow viscous oil was subjected to repeated chromatography (CC) 20% acetone: hexane increasing polarity with acetone to give compound XC6 (0.0644), XC5 (0.0591 g) and XC1 (0.1217 g)

Fraction C5, (0.1010 g) as a yellow viscous oil, upon crystallization from hexane: CH₂Cl₂: EtOAc give XC1 (0.0805 g)

Fraction C6, (0.4127 g) as a yellow viscous oil was subjected to repeated chromatography (CC) 10% EtOAc: hexane to gives XC1 (0.0207 g) and XH1 (0.0102 g)

Fraction C7, (1.5460 g) as a pale yellow viscous oil was subjected to repeated chromatography (CC) 10% acetone: hexane, upon crystallization from acetone: CH,Cl, give XC4 (0.0418 g)

Fraction C13, (6.9976 g) as a yellow viscous oil, upon crystallization from hexane: EtOAc give XC2 (5.6657 g)

Fraction C15, (3.4981 g) as pale yellow viscous oil, was subjected to repeated chromatography (CC) 30% EtOAc: hexane to give compound XC3 (2.5740 g)

Compound XC4 (Colorless solid, mp: 246-248 °C)

 $[\mathbf{\alpha}]_{\mathbf{D}}^{26}$: -47° (c = 0.068, acetone)

IR (NaCl) $V(cm^{-1})$: 3321 cm⁻¹ (O-H stretching)

¹H NMR (CDCl₂) (δ ppm) (300 MHz): see Table 25

 13 C NMR (CDCl₃) (δ ppm) (75 MHz): see Table 25

DEPT-135° (CDCl₂): see Table 22

Compound XC5 (Colorless oil)

$$[\mathbf{C}]_{\mathbf{D}}^{26}$$
: -75° ($c = 0.035$, CHCl₃)

FABMS m/z: 473.2 (24) [M+1]⁺, 313.2 (3), 185.1 (51), 131.1 (10), 93.0 (100)

IR (NaCl) $V(cm^{-1})$: 3426 cm⁻¹ (O-H stretching) and 1704 cm⁻¹ (C=O stretching)

 1 H NMR (CDCl₃) (δ ppm) (300 MHz): see Table 31

 13 C NMR (CDCl₃) (δ ppm) (75 MHz): see Table 31

DEPT-135° (CDCl₃): see Table 28

Compound XC6 (white solid, mp: 138-140 °C)

IR (NaCl) V(cm⁻¹): 3442 (O-H stretching)

¹H NMR (CDCl₃) (δ ppm) (300 MHz): 5.36 (2H, d, J = 5.4 Hz), 5.16 (1H, dd, J = 15.3, 8.4 Hz), 5.02 (1H, dd, J = 15.3, 8.4 Hz), 3.53 (2H, m), 2.35-2.15 (m), 2.07-1.89 (m), 1.89-1.74 (m), 1.70-1.40 (m), 1.35-1.02 (m), 1.01 (s), 0.93 (s), 0.91 (s), 0.87-0.76 (m), 0.70 (s), 0.68 (s)